

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 914 experiments for
(no ligands specified)

Metal : Al+++

(no references specified)

(no experimental details specified)

e- HL Electron (442)

Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
Al+++	vlt	oth/un	hi	100%	U				1971BSe	(301)	1
								$K(Al + 2Al(l)=3Al+) = 8.1$			

Medium: Na₃AlF₆(l); units of B?; 1015 C

Al+++	oth	none	25°C	0.0	U			1952LAb	(302)	2
								$K(Al+3e=Al(s)) = -84.3 (-1660 \text{ mV})$		

AsO ₄ ---	H3L	Arsenate	CAS	7778-39-4	(1557)
Arsenate;					

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	oth	oth/un	25°C	0.0	U			1990SAa	(1128)	3
								$*K(AlAsO_4(s)+H=Al+HAsO_4) = -4.70$		

Calculated from thermodynamic data.

Al+++	sol	oth/un	22°C	var	U			1956CHc	(1129)	4
								$K_{so}(Al(l)) = -15.80$		

B04H4-	HL	Borate	CAS	10043-35-3	(991)
Borate; B(OH)4-					

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	nmr	oth/un	25°C	var	C			2004TSa	(1301)	5
At 25 C by 27Al nmr. Medium: 0.02-0.2 m B(OH)3, pH 8.95.										
$K(Al(OH)_4+B(OH)_3=Al(OH)_3BO(OH)_2+H_2O) = 1.62$.										

Al+++	sol	oth/un	50°C	var	C T H			2004TSa	(1302)	6
Solubility of gibbsite or boehmite in 0.02-0.2 m B(OH)3, pH 8.95, 50-200 C										
$K(Al(OH)_4+B(OH)_3=Al(OH)_3BO(OH)_2+H_2O) = 1.58$ (50 C), 1.46 (78), 1.25 (200).										

Al+++	sol	none	22°C	0.0	U			1961SBc	(1303)	7
								$K_{so}(Al(OH)_3L_3) = -22.92$		

By spectrophotometry K1=7.62?, B2=14.64?, B3=20.0? B6=38.54?

Br- HL Bromide CAS 10035-10-6 (19)
Bromide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	con	non-aq	-78°C	100%	U	T		1977GPa (1717) 8 K(2AlBr3=AlBr4+AlBr2)=-6.6		
Medium: CH3Br										
Al+++	EMF	non-aq	210°C	100%	U	T	H	1973TDa (1718) 9 K(2Al4=Al2L7+L)=-4.40		
Medium: (Na,Al)Br. DH(K)=20.1 kJ mol-1; K=-4.30(225 C), -4.15(240 C) m units										
Al+++	nmr	oth/un	26°C	var	U	HM		1972J0b (1719) 10 K(2AlCl3L=AlCl4+AlCl2L2)=-0.5 K'=-0.1 K"(2AlCl3L=AlCl2L2+AlL4)=-0.4		
K': 2AlCl2L2=AlCl3L+AlCl3L. DH(K)=0.2 kJ mol-1; DH(K')=-1.5; DH(K")=-0.3										
Al+++	con	non-aq	21°C	100%	U	I		1972SVa (1720) 11 K(Al2L6=AlL2+AlL4)=-22.14		
Medium: n-heptane. In benzene: K=-16.12										

Al+++	con	non-aq	25°C	100%	U			1964WEa (1721) 12 K3=4.9 K4=3.3 K(2AlBr3=Al2Br6)=0.0		
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Medium: PhNO2

C03--	H2L	Carbonate	CAS 465-79-6 (268)		
Carbonate;					

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	3.00M	C	I		1987HSa (3141) 13 B(-4,2,1)=-16.61 B(-5,3,1)=-18.39		
B(p,q,r); pH+qAl+rCO2(g)=HpAlq(CO2(g))r										
Al+++	gl	NaCl	25°C	0.60M	C			1981Fa (3142) 14 B(-4,2,1)=-20.41 B(-5,3,1)=-22.74		
B(p,q,r): pH + qAl + rCO2(g) = HpAlq(CO2)r										

C6N6Co--- H3L Cyanocobaltate (5470)
Hexacyanocobaltate; [Co(CN)6]---

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	con	none	25°C	0.00	U			K1=4.30	1971KKF (3485) 15	

By kinetics, ($K_{1\text{out}}/K_1$)=-1.3

C6N6Fe--- H3L Ferricyanide (2491)
Hexacyanoferrate (III); Fe(III)(CN)6---

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ kin KN03 25°C 0.01M C K1=4.30 1983KLa (3629) 16

Method: stopped flow, by conductivity measurement.

Cl- HL Chloride CAS 7647-01-0 (50)

Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp non-aq 25°C 100% C IH K1=1.73 B2= 2.20 1998UKb (4443) 17

Medium: DMF, 0.2 M Et4NC1O4. Also data for DMA, 0.2 M Bu4NC1O4.

By calorimetry, DH(K1)=5.1 kJ mol-1, DH(B2)=24.4, DH(B3)=34.

Al+++ EMF non-aq 40°C 100% U 1985KOa (4444) 18

K(A12C17+Cl=2A1C14)=-17.0

Medium: N-1-butylpyridinium Cl. In 1-Me-3-ethylimidazolium Cl, K=-17.1

Al+++ con non-aq -78°C 100% U T H 1977GPa (4445) 19

K(2A1C13=A1C14+A1C12)=-2.48

Medium: CH2Cl2, at -78 and 0 C. DH=-15 kJ mol-1, DS=-125 J K-1 mol-1

Al+++ oth oth/un 550°C 100% U 1974CHb (4446) 20

B4=5.60(x units)

Medium: (K,A1)Cl. Method: Raman

Al+++ EMF non-aq 300°C 100% U 1974IKa (4447) 21

K(A12L6+L)=7.5

K(2A1L4=A12L7+L)=-5.5

Medium: (K,A1)Cl; x units. 300-450 C

Al+++ EMF non-aq 300°C 100% U 1974IKa (4448) 22

K(A12L6+L)=9.5

K(2A1L4=A12L7+L)=-5.5

Medium: (Cs,A1)Cl; x units. 300-450 C

Al+++ EMF non-aq 300°C 100% U 1973BBC (4449) 23

K(2A1L4=A12L7+L)=-7.83

Medium: (K,A1)Cl(51.7% KCl); m units

Al+++ EMF non-aq 175°C 100% U T H 1973BJc (4450) 24

K(A1L3+A1L4)=4.38

Medium: (Na,A1)Cl;m units. DH=-54.4 kJ mol-1,DS=-38 J K-1 m-1(200 C), K=4.11 (200 C),3.53(250 C),3.00(300 C),2.60(355 C). K4=10.8(200 C)

Al+++	EMF non-aq 450°C 100% U	1973SSi (4451) 25
	B4=25	
	K(A12L6+L)=5.70	
	K(2A1L4=A12L7+L)=-6.24	
Medium:	(Na,Al)Cl	

Al+++	EMF non-aq 175°C 100% U T H	1973TDa (4452) 26
	K(2A1L4=A12L7+L)=-5.50	
Medium:	(Na,Al)Cl. DH(B)=28.0 kJ mol-1. K=-5.32(190 C), -5.00(210 C), -4.87(225 C), -4.75(240 C) m units	

Al+++	EMF non-aq 175°C 100% U	1972FKb (4453) 27
	K4=11.5	
	K(A1L3+A1L4)=4.4	
Medium:	(Na,Al)Cl	

Al+++	oth non-aq 300°C 100% U I	1972KOa (4454) 28
	K(Co(A12C17)2+4Cl)=6.5	
Medium:	molten (LiAl)Cl. K: (Co(A12C17)2+4Cl)=CoCl4+2A12C17. K=1.2(in (NaAl)Cl); 19.1((in KAl)Cl); 18.7(in (RbAl)Cl); 19.7(in (CsAl)Cl)	

Al+++	EMF non-aq 300°C 100% U T	1972KOa (4455) 29
	K(A12L6+2A1L4=2A12L7)=1	
Medium:	(xK,(1-x)Al)Cl; 0.35<x<0.45	

Al+++	oth non-aq 170°C 100% U	1971Ra (4456) 30
	K(A12L6+2A1L4=2A12L7)=2.1	
Medium:	(K,Al)Cl; m units. 170-240 C. Method: Raman. Error in abstract ?	

Al+++	oth oth/un ? var U	1971SCc (4457) 31
	K2=0.5	
	K3=-2.7	
Method:	ionophoresis	

Al+++	EMF non-aq 175°C 100% U T	1971TMa (4458) 32
	K(2A1L4=A12L7+L)=-7.1	
Medium:	(Na,Al)Cl; K=-6.3(250 C), -5.7(300 C), -5.3(350 C), -5.0(400 C)	

Al+++	con non-aq 25°C 100% U	1970MLa (4459) 33
	K=5.0	
Medium:	CH3COCl. K: A1L3+CH3COL=CH3CO++A1L4	

Al+++	oth non-aq 289°C 100% U	1969JSb (4460) 34
	K(2A1L4=A12L7+L)=-3.52	
Medium:	KAlCl4; m units. Method: gas chromatography	

Al+++	con non-aq 25°C 100% U	1966WIa (4461) 35
	K3=4.45	
	K4=3.04	
	K(2A1L3+A1L2)=3.04	
	K(A12L5+L)=2.95	

Medium: PhNO₂

F- HL Fluoride CAS 7644-39-3 (201)
Fluoride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
Al+++	nmr	R4N.X	5°C	0.60M	C			K1=6.42 K3=3.99 K4=2.50 K5=0.84	2000BTa	(6696)	36

Method: 19F nmr and potentiometry. Medium: NMe₄Cl

In 3M KCl at 25 C: K1=6.35, K2=5.25, K3=4.11.

Al+++	sp	oth/un	23°C	0.10M	U			1994KGa	(6697)	37
								Keff=10.7		

Method: spectrophotometric using pyrocatechol violet. Tris buffer adjusted to a pH=5.34 with HCl

Al+++	nmr	oth/un	21°C	var	U			1991NMa	(6698)	38
								K(Al(NDP)+F)=4.1 K(Al(NDP)F+F)=3.1 K(Al(NDP)F ₂ +F)=1.8		

K(Al+NDP)=7.8, K(AlF+NDP)=5.8, K(AlF₂+NDP)=4.0, K(AlF₃+NDP)=2.2. NDP=nucleoside diphosphates: guanosine diphosphate or adenosine diphosphate.

Al+++	ISE	non-aq	185°C	100%	M			K1=6.8 B3=19.7 B4=24.92 B5=28.79	1988JHa	(6699)	39

Medium: molten KSCN. K1=mol-1 kg, B2=mol-2 kg² etc.

Al+++	ISE	KNO ₃	25°C	0.10M	C			K1=4.92 B2= 8.46	1988YYa	(6700)	40

Method: fluoride ion selective electrode

Al+++	sol	oth/un	50°C	var	M	M		1987SMe	(6701)	41
								K(AlF ₃ (s)=Al+3F)=-21.2		

Also mixed hydroxo-fluoro complexes. Solubility using pH and pF electrodes

Al+++	ISE	KNO ₃	25°C	0.10M	C	M	K1=6.40 K3=3.86, K4=2.7	B2=11.64	1987YHa	(6702)	42

K(AlA+F)= 5.41(H₃A=NTA), 5.53(H₃A=HEDTA), 4.95(H₄A=EDTA), 3.14(H₄A=CDTA)

Al+++	gl	NaNO ₃	25°C	0.10M	M	TI		1986C0a	(6703)	43
								B(Al(OH)F)=15.58 B(Al(OH)F ₂)=20.0 B(Al(OH) ₃ F)=30.01 K(AlOH+F)=6.13		

K(AlOH+F)=5.97(35 C), 5.88(50 C); K(AlOH+2F)=10.36, 10.07(35 C), 9.88(50 C); K(Al(OH)₄+F)=-3.73, -3.34(35 C); I=1.0 M, K(Al(OH)₃F+OH)=-3.43

Al+++ ISE KN03 25°C 0.10M C M K1=6.40 B2=11.64 1986YUa (6704) 44
K3=3.86
K4=2.75
K(Al(edta)+F)=4.95
K(Al(cdtta)+F)=3.14

Method: F ion-selective electrode. K(AlHA+F)=5.7, K(AlA+F)=4.5,
K(AlFA+F)=3.9. H3A is citric acid.

Al+++ gl KN03 25°C 0.50M M 1985TZb (6705) 45
K(Al(OH)4+L=Al(OH)3L+OH)=-2.20

Al(OH)2L2 may form, but no Ga complexes, detected.

Al+++ oth none 25°C 0 U 1984DYa (6706) 46
K(Al+F=Al(OH)F+H)=1.45
K(Al+2F=Al(OH)F2+H)=6.21
K(Al+F=Al(OH)2F+2H)=-4.76
K(Al+3F=Al(OH)F3+H)=-0.77

Recalc. of lit. data. K(Al+F=Al(OH)3F+3H)=-10.97.

Al+++ gl R4N.X 25°C 0.20M U I K1=6.46 B2=11.44 1982KMa (6707) 47
B3=15.16
B4=17.83
B5=19.29
B6=20.46

Al+++ cal NaClO4 25°C 0.5M C 1975VKb (6708) 48
DH(Al+L)=2.68 kJ/mol
DH(Al+3L)=5.31 kJ/mol
DH(Al+2L)=4.9 kJ/mol

Also data for 35 C

Al+++ EMF non-aq 210°C 100% U M 1973TDA (6709) 49
K(2AlCl4+L=Al2Cl6L+2Cl)=-3.5
K(2AlBr4+L=Al2Br6L+2Br)=-4.0

Medium: (Na,Al)Cl and (Na,Al)Br, m units

Al+++ ISE KN03 25°C 0.10M U I K1=6.45 B2=11.66 1971AMB (6710) 50
K3=3.79
K4=3.18

K1=6.51, K2=5.29, K3=3.76, K4=3.05 (I=0.05); 6.32, 5.16, 3.85, 3.30 (I=0.2);
6.14, 5.09, 3.93, 3.68 (I=0.5); At I=0 (corr): K1=6.69, K2=5.35, K3=5.68, K4=2.75

Al+++ ISE KN03 37°C 0.10M U TI K1=6.49 B2=11.73 1971AMB (6711) 51
K3=3.86
K4=3.38

K1=6.71, K2=5.26, K3=3.92, K4=3.29 (I=0.05); 6.39, 5.17, 3.86, 3.38 (I=0.2);
6.29, 5.09, 3.84, 3.43. At I=0 (corr): K1=6.68, K2=5.34, K3=3.94, K4=3.29

Al+++ EMF NaClO4 25°C 1.0M U H 1971WTa (6712) 52

$$*K(Al+HF=AlF+H)=3.15$$

$$*K(AlF+HF=AlF_2+H)=1.99$$

Method: quinhydrone electrode. By calorimetry: DH(*K1)=3.10 kJ mol-1,
DS=128 J K-1 mol-1

Al+++	nmr	oth/un	-15°C	var	U	1970MWa	(6713)	53
						K(2AlF=Al+AlF2)=-0.8		
						K(2AlF2=AlF+AlF3)=-0.9		
						K(2AlF3=AlF2+AlF4)=-1.4		

Medium: Al(NO₃, F). Method: nmr

Al+++	ISE	R4N.X	25°C	0.01M	U	I	K1=6.65	B2=12.09	1969BAa	(6714)	54
							K3=3.92				
							K4=2.38				

Medium: NH₄NO₃. K1=6.40, K2=5.19, K3=3.91, K4=2.42(I=0.1); K1=6.29, K2=4.97, K3=3.73, K4=2.50(I=0.3); I=0(corr): K1=6.98, K2=5.62, K3=4.05, K4=2.38

Al+++	oth	oth/un	782°C	100%	U	1969RLa	(6715)	55
						K5.K6=1.22		

Medium: molten (Li,Al)F. Method: combination of thermodynamic data

Al+++	oth	oth/un	25?°C	var	U	K1=6.08	B2=11.10	1964BSc	(6716)	56
Method: refractometry.										

Al+++	oth	non-aq	999°C	100%	U	1962RBa	(6717)	57
						K5K6=1.7		

Method freezing point, ca.1000 C. Medium: Na₃AlF₆(l), ionic fraction units

Al+++	oth	non-aq	930°C	100%	U	1960BGe	(6718)	58
						K5.K6=1.45		

Method: freezing point. Medium: Na₃AlF₆(liquid). 1008-930 C

Al+++	oth	non-aq	999°C	100%	U T H	1960FFb	(6719)	59
						K5K6=1.05		

In liquid Na₃AlF₆, 1000-1090 C. DH(K5K6)=-92.0KJ mol-1. K5K6=0.96(1030 C), 0.82(1075 C), 0.80(1090 C). Method: density

Al+++	oth	non-aq	999°C	100%	U	1960R0a	(6720)	60
						K5K6=0.75		

Method: freezing point. Medium: Na₃AlF₃(l), mole fraction units

Al+++	oth	non-aq	930°C	100%	U	1959BGh	(6721)	61
						K5K6=1.22		

Method:freezing point. Medium: liquid Na₃AlF₆,930-1008 C. Ion fraction units
K5K6=1.4 to 1.5 in x units

Al+++	sp	oth/un	?	var	U	K1=6.4	1959BSg	(6722)	62

Al+++	EMF	KNO ₃	25°C	0.53M	U	I	K1=6.16	B2=11.21	1959KGa	(6723)	63
							K3=3.91				

K4=2.71
K5=1.46?

In NH4NO3 K3=3.57, K4=2.64, K5=1.46, K6=0.04?

Al+++ cal oth/un 25°C 0.07M U H 1959KGa (6724) 64
DH(K1)=4.4 kJ mol-1, DS=141 J K-1 mol-1. DH(K2)=3.9, DS=115; DH(K3)=0.8,
DS=80.3; DH(K4)=0.17, DS=51.9; DH(K5)=-1.5, DS=21

Al+++ cal none 25°C 0.0 U H 1959SCe (6725) 65
DH(K1)=4.9 kJ mol-1, DH(B2)=8.2, DH(B3)=9.1, DH(B4)=9.0, DH(B5)=9.5, DH(B6)=-5.2

Al+++ sol oth/un 25°C var U B2=9.06 1957TVa (6726) 66

Al+++ EMF none 25°C 0.0 U K1=7.00 1955PAa (6727) 67

Al+++ cal none 25°C 0.0 U H 1955PAa (6728) 68
At I=0 corr: DS(K1)=160J K-1 mol-1, DS(K2)=130, DS(K3)=84, DS(K4)=54, DS(K5)
=-8.4, DS(K6)=-25. Values also at I=0.07 M

Al+++ cal oth/un 25°C var U H 1953LJa (6729) 69
DH(K1)=4.8 kJ mol-1, DS=130 J K-1 mol-1; DH(K2)=3.3, DS=110; DH(K3)=0.8, DS=
75; DH(K4)=1.2, DS=54; DH(K5)=-3.1, DS=21; DH(K6)=-6.5, DS=-13

Al+++ sp KN03 ? 0.10M U K1=6.32 1950KLb (6730) 70

Al+++ gl oth/un 18°C var U 1949LAa (6731) 71
B6=ca.27
Ks(A12F6(s)=A1+AlF6)=-9.4

Al+++ EMF KN03 25°C 0.53M U K1=6.13 B2=11.15 1943B0a (6732) 72
K3=3.85
K4=2.74
K5=1.63
K6=0.47

B6=19.84. Method: quinhydrone electrode and redox

HPO3-- H2L Phosphite CAS 13598-36-2 (6305)
Phosphite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ nmr NaClO4 25°C 1.0M U K1=2.01 B2= 3.24 1999MHa (7502) 73
B(A12L2)=2.21

Method: Al nmr.

Al+++ nmr NaClO4 25°C 1.0M U 1999MHa (7503) 74
K(Al+HL)=2.01
K(A1HL+HL=A1H2L2)=1.23
K(2A1HL=A12H2L2)=2.21

Method: nmr. L is H2PO3-.

Al+++	nmr	NaNO ₃	20°C	0.10M	C	K1=6.11	1991FWa	(7504)	75
Method: ³¹ P nmr.									
H ₂ PO ₂ -	HL	Hypophosphite	CAS	6303-21-5	(6304)				
Hypophosphite;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference ExptNo
Al+++	nmr	NaClO ₄	25°C	1.0M	U	H	K1=2.01 K3=1.26 B(A12L2)=1.11	B2= 3.29	1999MHa (7634) 76
Method: Al nmr. DH(K1)=9.3 kJ mol-1, DS=69 J K-1 mol-1. DH(K2)=7.1, DS(K2)=48. DH(K3)=-0.3, DS(K3)=26. DH(A12L2)=46, DS(A12L2)=175.									
Al+++	nmr	NaClO ₄	25°C	1.0M	U	H	K1=2.01 K3=1.26 K(2A1L=A12L2)=1.11	B2= 3.29	1999MHa (7635) 77
Method: nmr. DH(K1)=9.3 kJ mol-1, DS=69 J K-1 mol-1. DH(K2)=7.1, DS=48; DH(K3)=-0.3, DS=26; DH(A12L2)=46, DS=175.									
Al+++	nmr	NaNO ₃	20°C	0.10M	C	K1=2.38	1991FWa	(7636) 78	
Method: ³¹ P nmr.									
I-	HL	Iodide	CAS	10034-85-2	(20)				
Iodide;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference ExptNo
Al+++	con	non-aq	-78°C	100%	U	I	K(2AlI ₃ =AlI ₂ +AlI ₄)=-8.15	1977GPa (7883) 79	
Medium: CH ₃ I									
MoO ₄ --	H ₂ L	Molybdate					(443)		
Molybdate;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference ExptNo
Al+++	gl	NaCl	25°C	0.60M	C			19890Ha (8710) 80	
B(A1H6L6)=50.95									
Al+++	EMF	oth/un	25°C		U			1971GLa (8711) 81	
B6=ca.19									
OH-	HL	Hydroxide					(57)		
Hydroxide;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference ExptNo
Al+++	sol	oth/un	50°C	var	C T H			2004TSa (10887) 82	

$K_s(Al(OH)_3 + OH = Al(OH)_4) = -0.83$
 Solubility of gibbsite or boehmite in NH₃/NH₄Cl at 50-200 C. At 78 C:
 $K's(AlOOH + OH + H_2O = Al(OH)_4) = -0.88$; at 150 C, $K's = -0.45$, at 200 C, -0.11

Al+++ gl KCl 25°C 0.10M C I 2001DJa (10888) 83
 *K1=-4.81
 *B(3,4)=-13.82
 *B3=-14.17
 *Ks(Al(OH)₃)=10.38

Medium: 0.10 M LiCl, 0.005 M CTAB. Kw=-13.12. In 0.10 M LiCl, 0.001 M Tiron, Kw=-13.40. In 0.010 M LiCl, Kw=-13.50

Al+++ sol NaCl 100°C 0.10M C 2001PBA (10889) 84
 *Kso(AlOOH)=4.20
 *B4=-21.16

Solubility of boehmite in 0.1-5.0 m NaCl at 100-290 C, using a Pt/H₂ conc cell. At I=0, *Kso=3.46. $K_s(AlOOH + 2H_2O = Al(OH)_4 + H) = -12.76$ (-13.02 at I=0)

Al+++ sol NaCl 25°C 0.03M C 2001PBB (10890) 85
 *B2=-10.73
 *B3=-15.46
 *B4=-22.78

Calc from solubility of synth boehmite in 0.03 m NaCl using a Pt/H₂ conc'n cell. Data for at 100-290 C. At 100 C, *B2=-6.76, *B3=-10.95, *B4=-16.48.

Al+++ gl NaCl 37°C 0.15M C 2000GKb (10891) 86
 *B4=-21.031

Al+++ gl KCl 30°C 3.0M C 2000STc (10892) 87
 *B(2,2)=-6.68
 *B(3,6)=-20.90
 *B(13,32)=-104.45
 *B(13,35)=-117.78

Al+++ gl NaClO₄ 25°C 1.0M C I 1999CIb (10893) 88
 B(1,1)=-5.48
 B(1,2)=-10.3
 B(2,2)=-8.0
 B(3,4)=-13.47

B(13,32)=-104.8, B(pq):pAl+qH₂O=ALp(OH)(q-r)+qH

Al+++ sp NaClO₄ 25°C 8.0M U 1998SCa (10894) 89
 *B(4,3)=-11.7
 *B(4,2)=-25.9
 *B(5,4)=-35.4
 *B(5,3)=-17.0

Medium: 8 M NaClO₄. *B(6,5)=-45.1, *B(6,4)=-31.1, *B(6,3)=-16.9, *B(7,5)=-40.7.

Al+++ gl NaCl 25°C 0.01M C 19970Wa (10895) 90

$$*K_{so}(Al(OH)_3) = 10.35 \text{ (fresh)}$$

$$*K_{so}(Al(OH)_3) = 9.0 \text{ (48 h)}$$

Fresh indicates precipitate immediately after flashing reactants (10mM Al and 30 mM NaOH) together ($I=0.015$ M).

Al+++	oth none	400°C	0.00	C T	1995ANa (10896)	91
					K _{so} (AlO _{1.5}) = -5.666	

$$K(Al(OH)_3 = Al(OH)_4 + H) = -3.359$$

From literature data on solubility of corundum (AlO_{1.5}) in H₂O and KOH. Data for 400-700 C at 2000 bar.

Al+++	gl oth/un	25°C	0.10M	C	1995DJa (10897)	92
					*K ₁ = -5.62	

$$*B_2 = -9.74$$

$$K(3Al + 4H_2O = Al_3(OH)_4 + 4H) = -13.7$$

Medium: LiCl

Al+++	sol NaCl	25°C	0.01M	U T	1994SHa (10898)	93
					K(Al(OH) ₃ (s) + 3H = Al + 3H ₂ O) = 8.02	

Gibbsite (Al(OH)₃(s)) solubility measurements. Constant at $I=0$

Al+++	sol NaCl	25°C	0.10M	C	1994WPa (10899)	94
					*K _{so} = 8.31	
					*K _{so} = 7.74 ($I=0$)	
					Gibbsite solubility study using H electrode. Data for 50-100 C, 0-5 M NaCl extrapolated to 0 and 25 C. At $I=0.1$ M *K ₁ = -5.31, *K ₂ = -5.8, *K ₃ = -6.8, *K ₄ = -5.4	
Al+++	sol oth/un	150°C	var	M TI	1993BKa (10900)	95
					*K ₁ = -2.1	
					*B ₂ = -4.8	
					*B ₃ = -7.8	
					*B ₄ = -13.6	

Boehmite solubility study at 150, 200, 250 C and P=10 bars.

$$K(AlO(OH)(s) + 3H = Al + 3H_2O) = 1.53$$

Al+++	sol none	170°C	0.0	C T	1993CDC (10901)	96
					K(AlOOH(s) + 3H = Al + 2H ₂ O) = 0.81	
					K(AlOOH(s) + 2H = AlOH + H ₂ O) = -0.95	
					K(AlOOH + H = Al(OH) ₂) = -3.78	
					K(AlOOH + H ₂ O = Al(OH) ₃) = -6.76	

AlOOH is Boehmite. Data for 90-350 C. $K(AlOOH + 2H_2O = Al(OH)_4 + H) = -11.75$.

$$*K_1(Al) = -1.765, *K_2 = -2.82, *K_3 = -2.98, *K_4 = -4.99.$$

Al+++	EMF NaCl	25°C	0.10M	C TIH	1993PWa (10902)	97
					*K ₁ = -5.31	

Data at $I=0.1, 0.3, 1.0$ and 5.0 M. 25-125 C. $DH(*K_1) = 54.4$ kJ mol⁻¹.

H electrode. Using Pitzer evaluation. $*K_1(I=0) = -4.95$

Al+++	sol none	80°C	0.0	C	1992NLa (10903)	98
					K(Al(OH) ₃ (s) + 3H = Al + 3H ₂ O) = 5.00	

Al(OH)3(s) is gibbsite.

Al+++ sol NaCl 50°C 0.10M C TIH 1992PWa (10904) 99

$$*K_s(Al(OH)_3 + 3H) = 6.919$$

Gibbsite solubility study using H electrode. I=0.1-5.0 M(NaCl), 30,50,70 C.
Pitzer ion interaction treatment of data. DH(*Kso)=-101.8 kJ mol⁻¹

Al+++ EMF oth/un 25°C var C TIH 1992WEa (10905) 100

$$K(Al(OH)_4 = Al(OH)_3(s) + OH) = 1.143$$

I=0.01-5.0 M, 6.4-80 C. Gibbsite solubility studies. DH(K)=-22.5 kJ mol⁻¹

Al+++ gl NaCl 25°C 3.0M U 1991MBa (10906) 101

$$*B(2,2) = -7.53$$

$$*B(2,4) = -16.50$$

$$*B(3,4) = -13.44$$

Al+++ gl NaNO₃ 25°C 3.0M C 1991MBe (10907) 102

$$*B(2,2) = -7.55$$

$$*B(2,4) = -16.41$$

$$*B(3,4) = -13.24$$

Al+++ gl NaCl 25°C 0.60M C 1990MOa (10908) 103

$$*B(13,32) = -105.5$$

Al+++ gl NaNO₃ 25°C 0.50M C 1989DJa (10909) 104

$$*K_1 = -5.65$$

$$*B(2,2) = -7.03$$

$$*B(2,4) = -15.65$$

$$*B(3,4) = -12.60$$

Al+++ gl NaCl 25°C 0.60M C 19890Ha (10910) 105

$$K(Al + 3H_2O = Al(OH)_3(s) + 3H) = -10.5$$

Al+++ gl NaCl 25°C 0.60M C 19890Ha (10911) 106

$$*K_{so}(Al(OH)_3) = 10.49 \text{ (4 h)}$$

Precipitate aged for 4 hours.

Al+++ cal none 25°C 0.0 C 1988HHc (10912) 107

$$B_4 = 32.76$$

Heat capacity measurements on NaAl(OH)₄ solutions and Al+++ solutions
at 10-55 C. At 0 C: B₄=33.71; at 50 C: B₄=32.19.

Al+++ gl NaCl 25°C 3.00M C 1987HSa (10913) 108

$$*K_1 = -5.52$$

$$*B(3,4) = -13.96$$

$$*B(13,32) = -113.35$$

Al+++ sol oth/un 505°C var M M 1987SMe (10914) 109

$$B(1,3,3) = 30.69$$

$$B(1,2,2) = 20.80$$

$$B(1,1,2)=16.24$$

$B(1,y,x): Al(OH)_4+yF+xH=Al(OH)(4-x)Fy+xH_2O$. Solubility using pH and pF elec.

Al+++	gl	NaCl	37°C	0.15M	C	K1=8.577 B(ALL3)=26.138 B(ALL4)=29.044 B(Al3L11)=91.025 B(Al6L15)=149.278	1987VBa (10915) 110
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Constants also reported for data restricted to pH<8

Al+++	nmr	none	20°C	0.0	U	*K1=-4.93	1985AEa (10916) 111
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Al+++	gl	NaNO ₃	25°C	0.10M	C	*B1=-5.33 *B2=-10.91 *B(3,4)=-13.13 *B(13,32)=-107.47	1985BSa (10917) 112
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Al+++	EMF	NaCl	25°C	0.60M	C	*Kso(Al(OH) ₃)=11.2 (20 h)	19850Sb (10918) 113
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Precipitate aged for 20 h.

Al+++	dis	oth/un	25°C	var	U T H	*K1=-5.00 *B4=-22.20	1984CMc (10919) 114
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20-70 C. DH(*K1)=49.4 kJ mol⁻¹; DH(*B4)=177.6 kJ mol⁻¹

Al+++	oth	none	25°C	0	U	*K1=-5.0 *K2=-5.3 *K3=-5.9 *K4=-6.0	1984DYa (10920) 115
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Recalc. of lit. data. B4=-22.24; *Kso(Gibbsite)=8.11, Ks(Al(OH)₃=Al(OH)₃)=-8.13, Ks(Al(OH)₃+2H)=3.12, Ks(Al(OH)₃+H)=-2.22, Ks(Al(OH)₃=Al(OH)₄+H)=-14.13

Al+++	gl	NaCl	25°C	0.60M	C	*B4=-23.46	19830Sb (10921) 116
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Al+++	oth	none	80°C	0.0	C T H	Ks(AlOOH(s)+OH=Al(OH) ₄)=-0.70	1981CHa (10922) 117
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Calculations based on literature solubility data for boehmite, AlOOH.
Ks(AlOOH(s)+2H₂O=Al(OH)₄+H)=-13.30; DH=51.9 kJ mol⁻¹. Data for 80-300 C.

Al+++	gl	NaCl	25°C	0.60M	C	*K1=-5.52 *B(3,4)=-13.57 *B(13,32)=-109.2	19810Fa (10923) 118
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Al+++	gl	oth/un	25°C	6.0M	U		1980BCa (10924) 119
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$*K_1=-5.5$
 $*B_3=-10.1$
 $*B(13,36)=-105$

Additional method: ^{27}Al nmr. Medium: 0.5 M AlCl_3 .

Polymer is $\text{Al}_1\text{O}_4(\text{OH})_{28}$

Al+++	sol	none	25°C	0.0	U	1979MHa (10925) 120
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$*K_{so}=8.11$
 $*K_s(\text{Al}(\text{OH})_3+2\text{H})=3.12$
 $*K_s(\text{Al}(\text{OH})_3+\text{H})=-2.02$
 $K(\text{Al}(\text{OH})_4+\text{H})=14.0$

Al+++	oth	oth/un	20°C	?	U	1979STa (10926) 121
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$K_{so}(\text{Al}(\text{OH})_3)=-32.60$
 $K(\text{Al}(\text{OH})_3(s)=\text{Al}(\text{OH})_2+\text{OH})=-12.6$
 $*B_2=-8.30$

Medium: seawater. Method: Tyndallometry

Al+++	oth	none	25°C	0.0	U	1977VLa (10927) 122
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$B_3=33.96$

Al+++	kin	none	25°C	0.0	U	1975TUa (10928) 123
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$*K_1=-5.17$
 $*B(2,2)=-6.95$
 $*B(13,32)=-100.7$

Further data available for NaCl concentrations of 0.003 to 0.06M

Al+++	sp	NaClO_4	25°C	0.10M	U I	$K_1=9.10$ $B_2=17.65$ 1974NBe (10929) 124 $B_3=25.75$
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$K_1=9.40$, $B_2=18.27$, $B_3=26.75(I=0.3)$. $K_1=9.70$, $B_2=18.88$, $B_3=27.62(I=0.5)$.

$K_1=10.4$, $B_2=20.42$, $B_3=30.16(I=1)$ Error in abstract?

Al+++	sol	R4N.X	25°C	U	I	1973CHc (10930) 125
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$K_{so}(\text{Al}(\text{OH})_3(s)=\text{Al}+3\text{OH})=-30.55$

Medium: NH_4Cl . In LiCl , $K_{so}=-33.15$. In NaCl , $K_{so}=-30.75$. In KCl , $K_{so}=-30.36$. In CaCl_2 , $K_{so}=-31.00$

Al+++	sol	oth/un	23°C	0.10M	U	1972IYa (10931) 126
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$K_{so}(\text{Al}(\text{OH})_3(s)=\text{Al}+3\text{OH})=-32.94$

$*K_s(\text{Al}(\text{OH})_3(s)+\text{H}_2\text{O}=\text{Al}(\text{OH})_4+\text{H})=-12.23$ at 20 °C

Al+++	gl	KCl	20°C	0.10M	U	$K_1=12.3$ 1972SKa (10932) 127 $B_3=32.0$ $B_4=36.6$
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Al+++	sol	oth/un	30°C	U	1971DBa (10933) 128
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$*K_s(\text{AlL}_3(s)+\text{H})=2.59$
 $K(\text{Al}(\text{OH})_3(s)=\text{Al}(\text{OH})_3)=-3.92$
 $K_s(\text{Al}(\text{OH})_3+\text{H}_2\text{O}=\text{Al}(\text{OH})_4)=-12.62$
 $*K_{so}=11.40$

 Al+++ EMF KCl 62°C 1.00M U H 1971MBc (10934) 129
 *B(2,2)=-5.90
 *B(3,4)=-10.74
 DH(*B(2,2))=76.6 kJ mol-1 ,DS=113 J K-1 mol-1. DH(*B(3,4))=129.3, DS=176;
 DH(*B(14,34))=1100, DS=1577

 Al+++ EMF KCl 99°C 1.00M U T 1971MBc (10935) 130
 *B(2,2)=-4.81
 *B(3,4)=-8.20
 *B(14,34)=-67.9
 At 150 C: *B(2,2)=-3.95, *B(3,4)=-7.01, *B(14,34)=-55.7

 Al+++ gl oth/un 25°C 0.07M U IH 1971VPa (10936) 131
 *K1(Al+H2O=AlOH+H)=-5.20
 DH(*K1)=11.13 kJ mol-1. *K1=-5.11(I=0.0025), -5.23(I=0.01), -5.20(I=0.02),
 -5.30(I=0.03), -5.30(I=0.05)

 Al+++ gl none 25°C 0.00 U H 1971VPa (10937) 132
 *K1=-4.99
 *DH(*K1)=11.42 kJ mol-1. Data also 10 C, 15 C, 20 C, 30 C, 35 C, 40 C

 Al+++ sol none 25°C 0.00 U T H 1970BSd (10938) 133
 Ks(Al(OH)3(s)+OH=Al(OH)4)=-1.3
 DH(Ks)=34.7 kJ mol-1. Ks=-1.05(35 C), -0.78(50 C), -0.63(60 C)(gibbsite)

 Al+++ gl diox/w 25°C 55% U 19700Ha (10939) 134
 *B(2,2)=-6.95
 *B(3,2)=-10.02
 Medium: 55% w/w dioxan/H2O, 3 M LiClO4

 Al+++ gl NaNO3 30°C 2.00M U T 1969CBc (10940) 135
 *B(8,4)=-27.0
 *B(16,7)=-52.7
 At 50 C, *B(4,8)=-25.5, *B(7,16)=-48.5

 Al+++ gl none 30°C 0.00 U 1969GFa (10941) 136
 *K1=-4.61
 *B(2,2)=-7.44

 Al+++ kin none 15°C 0.00 U 1969GFa (10942) 137
 *K1=-5.11
 *B(2,2)=-8.03

 Al+++ oth none 50°C 0.0 U T K1=9.4 1969HEa (10943) 138
 B4=32.3
 Method:Literature.K1=9.5(60C),10.0(100C),10.8(150C),11.9(200C),13.1(250C)
 14.7(300C),B4=32.2(60C+100C),33.0(150C),34.6(200C),36.7(250C),39.5(300C)

 Al+++ oth none 60°C 0.0 U T 1969HEa (10944) 139

*Kso=6.17

Method: Estimated data. *Kso=4.62(100 C), 3.16(150 C), 2.15(200 C), 1.36(250 C), 0.76(300 C) (gibbsite)

Al+++ sp NaClO₄ 25°C 0.10M U K1=9.02 B2=17.59 1969NNc (10945) 140
B3=25.73

Al+++ sol oth/un 25°C 0.00 U 1969RPa (10946) 141
K(Al(OH)₃(s)=Al(OH)₂+OH)=-15.0
*Ks(Al(OH)₃(s)+OH)=-14.40

Al+++ sol oth/un 25°C U T 1969SBF (10947) 142
Kso(Al(OH)₃(s)=Al+3OH)=-33.7

Kso=-33.9(20 C), -33.4(30 C)

Al+++ oth oth/un 25°C dil U 1968HCA (10948) 143
*K1=-4.5

Medium: AlCl₃ dil. Method: Dissociation field effect relaxation

Al+++ kin NaClO₄ 25°C 1.00M U 1968SRc (10949) 144
*K1=-4.31

Al+++ gl oth/un ? U 1967FSb (10950) 145
Kso(Al(OH)₃(s)=Al+3OH)=-31.8

Al(OH)₃: gibbsite

Al+++ gl NaClO₄ 25°C 2.00M U 1965AVa (10951) 146
*B(2,2)=-7.07
*B(13,32)=-104.5

Al+++ con none 25°C 0.0 U 1965NTa (10952) 147
*K1=-4.5

Al+++ gl NaClO₄ 25°C 3.00M U 1964BIB (10953) 148
*B(7,17)=-48.8
*B(13,34)=-97.6

Al+++ sol none 25°C 0.0 U 1964PCa (10954) 149
K(Al(OH)₃(s)+OH=Al(OH)₄)=-0.68
*Kso=9.57
*Ks(Al(OH)₃+2H=AlOH+2H₂O)=5.27
*K(Al(OH)₃+H=Al(OH)₂+H₂O)=1.01
Kso(Al(OH)₃(s)=M+3OH)=-32.43, *K1(Al+H₂O=AlOH+H)=-4.3

Al+++ gl NaClO₄ 50°C 3.0M U 1964PCa (10955) 150
*B(17,7)=-48.8
*B(34,13)=-97.6

Al+++ con none 25°C 0.0 U 1963FPc (10956) 151
*K1(Al+H₂O=AlOH+H)=-5.02

Al+++	sol none	25°C	0.00	U	1963RAa (10957) 152 *Ks=-12.45 for boehmite *Ks=-13.84 for bayerite *Ks=-14.57 for gibbsite
*Ks: Al(OH)3(s) + H2O=Al(OH)4 + H					
Al+++	sol none	25°C	0.00	U	1962FPa (10958) 153 Kso(Al(OH)3(s)=Al+3OH)=-33.5
Al(OH)3:gibbsite					
Al+++	con none	25°C	0.0	U	1960FRa (10959) 154 *K1(Al+H2O=AlOH+H)=-5.02 Kso(Al(OH)3=Al+3OH)=-33.51
Gibbsite. Also pH and solubility					
Al+++	gl oth/un	25°C	dil	U	1960G0b (10960) 155 *Ks(Al(OH)3+OH)=-12.6 *Ks(Al(OH)3+2H=AlOH+2H2O)=5.73 Ks(Al(OH)3=AlOH+2OH)=-22.28
Al+++	sol none	25°C	0.0	U	1958GTa (10961) 156 Ks(Al(OH)3(s)+OH=Al(OH)4)=-0.5 *Ks(Al(OH)3+H=Al(OH)2+H2O)=0.4 *Kso(Al(OH)3+3H=Al+3H2O)=9.04 Kso(Al(OH)3=Al+3OH)=-32.96
Al+++	gl none	?	0.0	U	1957MOa (10962) 157 Kso=-33.45
Al+++	oth none	25°C	0.0	U	1956DPa (10963) 158 *Kso=9.66(Al(OH)3 amorphous) *Kso=8.55(alpha-Al2O3) *Kso=7.98(AlOOH, boehmite) *Kso=6.48(Al(OH)3,bayerite)
*Kso(Al(OH)3(s)+3H=Al+3H2O)=5.70(hydrargillite);					*K(Al+4H2O=Al(OH)4+4H)=20.3
Al+++	oth none	25°C	0.0	U	1955KEa (10964) 159 *K1(Al+H2O=AlOH+H)=-4.96 *B(2,2)=-7.55
*B(2,2): 2Al+2H2O=Al2(OH)2+2H. Method: freezing point etc.					
Al+++	EMF oth/un	20°C	var	U T	1955SCc (10965) 160 Kso(Al(OH)3)=-32.90 *Ks(Al(OH)3+OH)=-12.74
Kso=-31.72, *Ks=-12.87(30 C). Method: Sb electrode					
Al+++	gl oth/un	20°C	0.60M	U I	1954FAa (10966) 161 *K1(Al+H2O=AlOH+H)=-5.97 *B(2,2)=-8.24

Medium: Ba(NO₃)₂. In 0.12 M Ba(NO₃)₂ *K1=-5.74, *B(2,2)=-8.06

Al+++ gl none 25°C 0.0 U T 1954STa (10967) 162
*K1(Al+H₂O=AlOH+H)=-4.98

*K1=-5.28(15 °C), -5.15(20 °C)

Al+++ gl none 25°C 0.0 U 1953IYa (10968) 163
*K1(Al+H₂O=AlOH+H)=-5.10

Al+++ gl oth/un 22°C var U 1953KFa (10969) 164
K_{so}(Al(OH)₃)=-29.92

Al+++ EMF none 18°C 0.0 C I 1950AFa (10970) 165
K_{so}(Al(OH)₃)=-31.7

Method: H electrode. By solubility, dil. soln., B₃=26.96

Al+++ gl oth/un 18°C 0.01M U I 1949LAa (10971) 166
*K1(Al+H₂O=AlOH+H)=-4.60
K_{so}(Al(OH)₃(s)=Al+3OH)=-33.8

At I=0 corr: *K_s(Al(OH)₃+OH)=-13, *K_{so}(Al(OH)₃)=-34.0

Al+++ oth oth/un 20°C var U 1943CFa (10972) 167
K_s(Al(OH)₃(s)+OH=Al(OH)₄)=1.82
*K=-12.19

Al+++ oth oth/un ? var U 1943KTa (10973) 168
K_{so}=ca.-32

Al+++ cal oth/un 20°C 30% U 1942RWa (10974) 169
Medium: 30% w/w NaOH. DH(K_s(Al(OH)₃(s)+OH))=15.9 kJ mol⁻¹
At 77.3 °C: DH=22.8. Al(OH)₃ as hydrargillite

Al+++ gl oth/un 25°C dil U 1938OKa (10975) 170
K_{so}(M(OH)₃(s)=M+3OH)=-31.7
*K_s(Al(OH)₃+OH)=-11.92

Al+++ con oth/un 25°C var U 1934MAa (10976) 171
K₄=2.78
*K₄=-11.22

Al+++ oth oth/un 18°C var U 1933FMa (10977) 172
K_{so}=-12.2(fresh)
K_{so}=-13.8(aged)

Al+++ EMF oth/un rt var C 1930TRa (10978) 173
*K_s(Al(OH)₃(s)+OH)=-12.98

Al+++ sol oth/un 18°C 0.62M U 1929FRa (10979) 174
*K_s(Al(OH)₃+OH)=-12.44(fresh)
*K_s=-13.89(after 24 h)

Al+++	con	oth/un	25°C	?	U	T	1920HEa (10980) 175 Ks(Al(OH)3+OH=Al(OH)4)=1.54
Al(OH)3	fresh.	At 25 C:	K=1.85, or K=0.18(25 C, crystalline)				
Al+++	oth	oth/un	15°C	var	U		1920KO _b (10981) 176 Ks(Al(OH)3(s)+OH=Al(OH)4)=1.60 *Ks=-12.40
Al+++	kin	oth/un	100°C	0.01M	U		K1=9.49 1913KUa (10982) 177 *K1(Al+H2O=AlOH+H)=-2.88
Al+++	sol	oth/un	19°C	var	U		1911SLa (10983) 178 Ks(Al(OH)3+OH=Al(OH)4)=-0.74
By solubility	K=-0.48						
Al+++	oth	oth/un	25°C	var	U	I	1908DEa (10984) 179 *K1(Al+H2O=AlOH+H)=-4.29
Al+++	kin	oth/un	77°C	var	U	T	1899LEa (10985) 180 *K1(Al+H2O=AlOH+H)=-4.12
*K1=-3.4(99.7 C). At 25 C, I=0 corr: *K1=4.85	*****	*****	*****	*****	*****	*****	*****
P04---		H3L	Phosphate		CAS 7664-38-2	(176)	
Phosphate;							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values Reference ExptNo
Al+++	gl	KCl	25°C	0.20M	C	M	K1=13.50 2001LEa (13081) 181 B(AlHL)=17.60 B(AlH2L)=19.65 B(AlH-1L)=8.37 B(Al2L)=17.42
B(Al2H-2L)=11.05, B(Al2H-3L)=6.9. For citrate: B(AlH2(cit)L)=28.46, B(AlH(cit)L)=25.02, B(Al(cit)L)=19.68, B(AlH-1(cit)L)=12.03. Also 31P nmr.							
Al+++	gl	KCl	25°C	0.20M	C		1996AKa (13082) 182 B(AlHL)=17.60 B(Al2L)=16.65 B(Al2H-1L)=14.21 B(Al2H-3L)=7.42
Al+++	gl	NaClO ₄	25°C	3.00M	C	I	1996CIa (13083) 183 B(1,1,1)=0.13 B(1,2,2)=1.04 B(3,5,2)=-0.81 B(3,6,3)=0.23
B(p,q,r): pAl+rH3PO ₄ =AlpH-q(H3PO ₄)r+qH. B(3,8,3)=-6.11, B(3,6,4)=2.62, B(3,8,5)=0.98, B(3,6,1)=-9.60. Values at I=0.0 M calculated.							

Al+++	gl	NaClO ₄	25°C	3.0M	M	I	1996CIa (13084) 184
							K(Al+H ₃ L=AlH ₂ L+L)=0.13
							K(Al+2H ₃ L=AlHL+2H)=1.04
							K(3Al+2H ₃ L=Al ₃ HL ₂ +5H)=-0.92
							K(3Al+3H ₃ L=Al ₃ H ₃ L ₃ +6H)=0.15
K(3Al+3H ₃ L=Al ₃ HL ₃ +8H)=-6.17, K(3Al+4H ₃ L=Al ₃ H ₆ L ₄ +6H)=2.69. At I=0: K(Al+H ₃ L=AlH ₂ L+H)=0.6, K(3Al+2H ₃ L=Al ₃ HL ₂ +5H)=0.3, K(3Al+3H ₃ L=Al ₃ H ₃ L ₃ +6H)=2.5							

Al+++	gl	NaCl	37°C	0.15M	U		K ₁ =15.32 1991DEa (13085) 185
							B(AlHL)=17.79
							B(AlH ₂ L)=20.93
							B(Al ₂ L)=18.72
							B(Al ₂ H-2L)=12.58

Al+++	nmr	NaNO ₃	20°C	0.10M	C	K ₁ =17.26	1991FWa (13086) 186
Method: 31P nmr.							

Al+++	gl	NaCl	37°C	0.15M	C	K ₁ =15.660 B ₂ =20.88 1990DFA (13087) 187	
							B(AlHL)=19.072
							B(AlH ₂ L)=22.247
							B(AlH-2L ₂)=15.796
							B(AlH-3L ₂)=6.667

Al+++	gl	NaCl	25°C	0.15M	C		1988JVa (13088) 188
							B(AlHL)=23.25
							B(AlH ₂ L)=26.18
							B(AlHL ₂)=37.95

Al+++	con	oth/un	25°C	0.06M	U		1978RPa (13089) 189
							K(Al+H ₂ PO ₄)=3.06

Al+++	gl	NaClO ₄	25°C	0.10M	U	M	1975RMa (13090) 190
							K(Al+HPo ₄)=9.17
							K(Al+citrate+HPo ₄)=19.29
							K(Al+NTA+HPo ₄)=23.89
							K(Al+Cys+HPo ₄)=15.66

Al+++	ix	R4N.X	?	0.20M	U		1974FGc (13091) 191
							K(2Al+H ₃ L=Al ₂ HL+2H)=-1.96

Al+++	sol	none	25°C	0.0	U		1961TGa (13092) 192
Ks(K ₃ Al ₁₅ H-10H ₂ L ₈ (H ₂ O) ₁₈)(taranakite)=-22.5 ? Ks((NH ₄) ₃ Al ₁₅ H-10H ₂ L ₈ (H ₂ O) ₁₈)=19.3 ?. L-Al complex neglected							

Al+++	sol	none	25°C	0.0	U		1959LPb (13093) 193
							Ks(Al(H ₂ L)(OH) ₂)=-30.5
Ks(Al(H ₂ L)H-2(H ₂ O) ₂)(variscite)=-2.48							

Al+++	sol	oth/un	25°C	var	U		1957TVa (13094) 194
							K _{so} (All)=-10.41

Al+++ sol none ? 0.0 U 1955KJa (13095) 195
Ks(Al(H2L)(OH)2)=-28.0

Al+++ sol oth/un 19°C var U 1951ZHa (13096) 196
Kso(AlL)=-18.24

Al+++ sol NaCl ? 0.05M U 1950CJa (13097) 197
Kso(Al(H2L)(OH)2)=-29.55

Al+++ con oth/un 18°C 0.10M U 1931BDb (13098) 198
K(Al+H2L)=3 ?
K(AlH2L+H2L)=2.3
K(Al(H2L)2+H2L)=2.3
K(AlHL+H)=2.1

Also quinhydrone electrode. K(AlHL(H2L)+H)=2.1 and others

P207---- H4L Pyrophosphate CAS 2466-09-3 (198)
Diphosphate; from (HO)2PO.O.PO(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.20M C K1=13.74 B2=19.77 1996AKa (13560) 199
B(AlHL)=17.03
B(AlH2L)=18.69
B(AlH-1L)=7.41
B(AlHL2)=25.64

Al+++ gl NaCl 25°C 0.15M C K1=14.30 1988JVa (13561) 200
B(AlHL)=19.20
B(AlH2L)=22.79

P3010---- H5L CAS 10380-08-2 (1001)
Tripolyphosphate; from (HO)2PO.O.PO(OH).O.PO(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.20M C K1=13.15 B2=19.14 1996AKa (13836) 201
B(AlHL)=16.65
B(AlH2L)=18.07
B(AlH-1L)=6.53
B(AlHL2)=24.43

Al+++ gl NaCl 25°C 0.15M C K1=17.31 1988JVa (13837) 202
B(AlHL)=20.98
B(AlH-1L)=11.72

Al+++ gl KN03 35°C 0.10M U 1980KHc (13838) 203
K(AlL+thr)=7.55
K(AlL+ala)=7.51

				K(AlL+pro)=8.63	
				K(AlL+val)=7.90	
K(AlL+gly)=7.97. For tyrosine: K(AlL+HA)=7.95, *K(AlL(HA))=-7.15.					
K(AlL+Hgly-gly)=4.18, *K(AlL(Hgly-gly))=-5.53. Data for other aminoacids.					

P309---	H3L			CAS 13566-25-1 (235)	
Cyclotrimetaphosphate;					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo

Al+++	cal	oth/un	25°C	0.10M	C H K1=3.05 1983GGb (13944) 204
Medium:	0.10 M	HCl.	DH(K1)=19.6	kJ mol-1,	DS(K1)=124 J K-1 mol-1.

P4012----	H4L			CAS 13598-74-8 (234)	
Cyclotetrametaphosphate;					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo

Al+++	cal	oth/un	25°C	0.10M	C H K1=3.29 1983GGb (13996) 205
Medium:	0.10 M	HCl.	DH(K1)=37.8	kJ mol-1,	DS(K1)=190 J K-1 mol-1.

S--	H2L	Sulfide			CAS 7783-06-4 (705)
Sulfide;					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo

Al+++	vlt	oth/un	25°C	0.72M	C I K(Al+HL)=13.0 1999AVb (14306) 206
Method:	determination of free S-- by cathodic stripping voltammetry.				
Medium:	seawater, pH 8.0, S=35.	Also data for S=21 and 10.5.			

SCN-	HL	Thiocyanate			CAS 463-56-9 (106)
Thiocyanate;					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo

Al+++	sp	oth/un	25°C	7.5M	C K1=0.18 2001SZA (14800) 207
Method:	Raman spectroscopy.	Medium: NaSCN.			

Al+++	cal	non-aq	25°C	100%	C IH K1=2.1 B2= 3.60 1996TSa (14801) 208 K3=1.4 K4=1.0
Medium:	N,N-Dimethylformamide, 0.20 M Et4NC1O4.	Also data at 0.4 M Et4NC1O4			
DH(K1)=9.0	kJ mol-1,	DH(K2)=13,	DH(K3)=0,	DH(K4)=12.	

Al+++	sp	non-aq	25°C	100%	U IH K1=2.66 1985PWa (14802) 209
Medium:	dimethylsulphoxide.	K1 extrapolated to I = 0.0			

Al+++	sp	none	22°C	0.0	U T K1=0.42 1963VMA (14803) 210

S04-- H2L **Sulfate** CAS 7664-93-9 (15)
Sulfate:

Al+++ EMF none 15°C 0.0 U T H K1=1.75 B2=2.25 1970SPd (15977) 222
At 5 C: K1=1.60; 25 C: K1=1.90, K2=0.80; 35 C: K1=2.08, K2=1.05.
DH(K1)=25 kJ mol-1, DH(K2)=46

Al+++ nmr oth/un 25°C var U 1969AGa (15978) 223
B(Al+HL)=0.5

Method:N.M.R.

Al+++ cal none 25°C 0.0 U H K1=3.01 B2=4.90 1969IEa (15979) 224
DH(K1)=9.6 kJ mol-1, DS=89.5 J K-1 mol-1; DH(K2)=3.3, DS=47.2

Al+++ sol oth/un 20°C var U T 1969SBF (15980) 225
Kso(Al(OH)2.5(L)0.25)=-29.5
Kso(Al(OH)2.5(L)0.25)=-29.3(25 C), -29.1(30 C)

Al+++ sol none 0.0 U 1969SIa (15981) 226
Kso=ca.33 (fresh)
Ks(Al(OH)2.5(L)0.25)=-28.6

Al+++ oth oth/un var U K1=2.57 1969SMi (15982) 227
Method: coagulation

Al+++ con oth/un 25°C 0.0 U K1=3.73 1965NTa (15983) 228

Al+++ kin oth/un 25°C 0.10M U 1963BLa (15984) 229
K(Al(aq)+L(aq)=Al(H2O)L)=1.28
K(Al(H2O)L=AlL)=-1 to -2

Method: pressure jump

Al+++ EMF NaClO4 25°C 0.60M U I K1=1.30 B2=2.30 1962BWa (15985) 230
Method: Pb electrode. At I=0 corr. K1=3.2, K2=1.0

Al+++ sp oth/un 30°C 0.0 U K1=2.04 1962NAc (15986) 231

SiO3-- H2L Silicate CAS 7699-41-4 (747)
Silicate; SiO2(OH)2--

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.10M C T H 1996PSb (17194) 232
K(Al+H4SiO4=AlH3SiO4+H)=-2.38

Data for 90 and 150 C. DH(Al+H4SiO4)=66.66 kJ mol-1, DS(Al+H4SiO4)=
177 J K-1 mol-1.

Al+++ nmr oth/un RT 4.0M C 1995LMd (17195) 233
Ks(Al(OH)2SiO2(OH)(s)+2OH=HSiO3+Al(OH)4)=-3.0
Method: 27Al nmr. Medium: 4.0 M NaOH.

Al+++ gl NaClO4 25°C 0.0 M 1994FLa (17196) 234

$$K(Al+H2L=AlHL+H)=-2.50$$

Al+++ sol NaCl 25°C 0.01M U T 1994SHa (17197) 235

$$*Ks(imogolite)=13.04$$

*Ks(imogolite): $K(Al2SiO_3(OH)_4(s)+6H=2Al+Si(OH)_4+3H_2O)$. Constant at I=0

Al+++ sol NaNO₃ 25°C 0.10M U 1986MKa (17198) 236

$$*Ks(kaolinite)=7.42$$

*Ks(kaolinite): $Al_2Si_2O_5(OH)_4(s)+6H=2Al+2Si(OH)_4+3H_2O$. Constants at I=0 corr

Al+++ oth none 60°C 0.0 U T 1969HEa (17199) 237

$$*Ks(Al_2Si_2O_5(OH)_4+6H)=4.75$$

Method: estimated data.(kaolinite,Al₂Si₂O₅(OH)₄).

$$*Ks=2.27(100\text{ }^{\circ}\text{C}); -0.12(150\text{ }^{\circ}\text{C}); -1.72(200\text{ }^{\circ}\text{C}); -2.98(250\text{ }^{\circ}\text{C}); -4.02(300\text{ }^{\circ}\text{C})$$

Al+++ oth none 60°C 0.0 U T 1969HEa (17200) 238

$$*Ks(Al_2Si_2O_5(OH)_4+6H)=5.63$$

Method: estimated data.(dickite,Al₂Si₂O₅(OH)₄).

$$*Ks=3.10(100\text{ }^{\circ}\text{C}); 0.69(150\text{ }^{\circ}\text{C}); -0.90(200\text{ }^{\circ}\text{C}); -2.13(250\text{ }^{\circ}\text{C}); -3.09(300\text{ }^{\circ}\text{C})$$

Al+++ oth none 60°C 0.0 U T 1969HEa (17201) 239

$$*Ks(Al_2Si_2O_5(OH)_4+6H)=8.03$$

Method: estimated data.(halloysite,Al₂Si₂O₅(OH)₄).

$$*Ks=5.21(100\text{ }^{\circ}\text{C}); 2.50(150\text{ }^{\circ}\text{C}); 0.67(200\text{ }^{\circ}\text{C}); -0.75(250\text{ }^{\circ}\text{C}); -1.89(300\text{ }^{\circ}\text{C})$$

Al+++ oth none 150°C 0.0 U T 1969HEa (17202) 240

Method:estimated data.*Kso($K_0.6Mg_0.25Al_2.3Si_3.5O_10(OH)_2+8H$)=0.81,6.85(60 °C), 3.82(100 °C),-1.23(200 °C),-2.87(250 °C),-4.29(300 °C),(illite)

Al+++ oth none 150°C 0.0 U T 1969HEa (17203) 241

Method:est.data. *Ks($KFe_3AlSi_3O_10(OH)_2+10H$)=10.8. *Ks=18.7(60 °C),14.7(100 °C) 8.0(200 °C), 5.8(250 °C), 3.9(300 °C),(annite).

Al+++ oth none 150°C 0.0 U T 1969HEa (17204) 242

$$*Kso=-2.37$$

Method:estimated data. *Kso(montmorillonite):($K_0.33Al_2.33Si_3.67O_10(OH)_2(s)+12.7H$).Also at 60-300 °C: 3.00(60 °C); 0.31(100 °C); -4.17(200 °C); -6.9(300 °C).

Al+++ gl oth/un 400°C dil U I 1961HMa (17205) 243

$$K=3.0$$

K: Na-feldspar(s)+H=Na-mica,paragonite(s)+3SiO₂(s)+Na. Also other equilibria

Al+++ oth oth/un 200°C var U 1959HEa (17206) 244

$$K=4.9$$

By chemical analysis. P=1000 atm. Data also for mica to kaolinite etc.

$$K(1.5K\text{-feldspar}+H=0.5K\text{-mica}+3SiO_2(s)+K)=3.55(300C), 2.7(400C), 2.1(500C).$$

CH₂O₂ HL Formic acid CAS 64-18-6 (37)

Methanoic acid; H.CO₂H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
Al+++	kin	KNO ₃	25°C	0.01M	C			K1=1.28	1983KLa (17588)	245	
Method: stopped flow, by conductivity measurement.											
Al+++	gl	NaNO ₃	25°C	1.0M	U			B2=2.02	1976KIB (17589)	246	
Al+++	gl	NaNO ₃	25°C	1.00M	U			K1=1.3	1975KIB (17590)	247	
Al+++	oth	oth/un	25°C	1.00M	U			K1=0.56	B2=1.76	1973TRc (17591) 248	
Al+++	ix	oth/un	25°C	1.0M	U			K1=1.78	1962TSa (17592)	249	

CH4O			L	Methyl alcohol		CAS	67-56-1	(597)			
Methanol; CH ₃ .OH											
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
Al+++	EMF	alc/w	20°C	100%	U				1964GUa (17874)	250	
K(2Al+3H-1L=Al ₂ (H-1L) ₃)=42.0											
K(Al ₂ (H-1L) ₃ +H-1L)=11.1											
K(Al(H-1L) ₂ +H-1L)=10.5											
K(Al(H-1L) ₃ +H-1L)=5.5											
Method: H electrode. Medium: MeOH, 1.0 M Me4NCl											

CH503P			H2L			CAS	13590-71-1	(1752)			
Methylphosphonic acid; CH ₃ .PO ₃ H ₂											
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
Al+++	gl	KCl	25°C	0.20M	C			K1=6.48	B2=12.3	1996AKa (18122)	251
B(AlH-1L)=2.33											
B(AlH-2L)=-3.91											
B(AlH-1L2)=5.8											

CH606P2			H4L	Medronic acid		CAS	1984-15-2	(2384)			
Methanediphosphonic acid; CH ₂ (PO ₃ H ₂) ₂											
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
Al+++	gl	KCl	25°C	0.10M	U			K1=14.08	B2=23.01	1967KLa (18274)	252
K(Al+HL)=9.05											
K(Al+2HL)=13.66											

C2H2O3			HL	Glyoxylic acid		CAS	298-12-4	(1142)			
Glyoxylic acid; OHC.COOH											
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
Al+++	gl	KCl	25°C	0.10M	U			K1=13.5	B2=22.80	1975SDa (18416)	253

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)
 Ethanedioic acid; (COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ EMF NaCl 25°C 0.60M C M 2003BTa (18777) 254
 B(ALLF)=11.53
 B(ALLF2)=15.67
 B(ALL2F)=15.74
 B(ALL2F2)=19.09

Method: quinhydrone electrode and fluoride ISE.

Al+++ gl NaClO4 25°C 1.0M C I K1=6.03 B2=11.05 1999C1b (18778) 255
 B3=15.02
 B(2,2,2)=5.14; B(336)=-21.41
 B((7,17,3))=-43.2; B(353)=-0.87
 B(2,2,3)=-9.47

At I=0 (by SIT): K1=8.3, B2=14.1, B3=17.2

B(pqr):pAl+qH2O+rL=ALp(OH)(q-r)Lr+qH

Al+++ sol oth/un 80°C var U M K1=8.3 1994FHa (18779) 256
 B3=18.1

Alternatively data may be explained with formation of Al(OH)2L and Al(OH)L

Al+++ sol NaCl 25°C 0.60M C 1986BHc (18780) 257
 B(-9,3,1,0)=-21.87
 B(-4,1,1,1)=-5.61

B(-9,3,1,0):3Al+H2L=Al3(OH)7L(s)+9H; B(-4,1,1,1):Na+Al+H2L=NaAl(OH)2L(s)

Al+++ gl NaCl 25°C 0.60M C M 1985S0a (18781) 258
 B(-1,1,1)=1.40
 B(-2,1,1)=1.43
 B(-4,1,2)=1.85
 B(-6,1,3)=1.26

B(-9,3,3)=-4.28; B(-10,2,4)=-4.62. B(p,q,r) for pH+qAl+r(H2L)=HpAlq(H2L)r

Al+++ gl NaCl 37°C 0.15M C K1=5.02 B2=9.33 1982JCa (18782) 259
 B3=12.41
 B(AlHL)=6.63
 B(AlH-1L2)=7.07
 B(AlH-1L3)=9.52

Al+++ nmr none 25°C 0.0 U K1=7.18 B2=13.49 1977JBb (18783) 260
 B3=17.53

Al+++ vlt NaClO4 25°C 1.00M U K1=4.90 1970GMi (18784) 261

Al+++ gl NaClO4 25°C 1.00M U K1=4.85 1970GMi (18785) 262

Al+++ EMF NaClO₄ 25°C 1.00M U K1=6.06 B2=11.09 1968BCa (18786) 263
B3=15.12

Al+++ dis NaClO₄ 20°C 0.10M U 1963STc (18787) 264
B3=15.60

Al+++ gl oth/un ? ? U K1=7.26 B2=12.11 1957BDc (18788) 265
K3=1.31

Al+++ gl KN0₃ 32°C 1.0M U K2=5.45 1957DSa (18789) 266
K3=3.69

Al+++ gl none 18°C 0.0 U B2=13 1949LAb (18790) 267
B3=16.3

C2H3N0₄ HL CAS 625-75-2 (2968)

Nitroacetic acid; O2N.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ kin oth/un 18°C 0.20M U K1=0.48 1949PEa (19206) 268
Medium: Ba(NO₃)₂

C2H4O2 HL Acetic acid CAS 64-19-7 (36)

Ethanoic acid; CH₃.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sol oth/un 60°C 0.02M M TI K1=2.33 B2=4.10 1994BCa (19881) 269
Measurements at 60-200 C

Al+++ EMF NaCl 25°C 0.10M C TIH K1=2.02 B2=3.5 1994PBa (19882) 270
Measurements at 25-150 C and I=0.1-1.0 M. Pitzer formalism and equations.
DH(K1)=17 kJ mol⁻¹; DH(B2)=40 kJ mol⁻¹

Al+++ sol oth/un 80°C var U K1=2.9 B2=4.8 1991FEa (19883) 271
Gibbsite solubility measurements.Constants at I=0

Al+++ gl NaCl 25°C 0.60M C 1989MOa (19884) 272
K(2Al+HL=Al₂(OH)₂L+3H)=-7.98

Al+++ gl NaNO₃ 25°C 1.00M U K1=1.4 1975Kib (19885) 273

C2H5N0₂ HL Glycine CAS 56-40-6 (85)
2-Aminoethanoic acid; H₂N.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 37°C 0.15M C 2002DCa (21487) 274
B(Al₂H-2L)=0.361

Al+++ gl KCl 25°C 0.20M C K1=5.91 1997KSa (21488) 275
B(AlH-1L)=1.08
B(Al2H-1L)=4.35

Al+++ gl NaNO3 25°C 0.50M C 1989DJa (21489) 276
B(Al2H-2L2)=6.56
B(AlH-3L)=-7.53

Al+++ oth NaClO4 35°C 0.01M U 1984YSa (21490) 277
B3=19.40

Method: paper electrophoresis.

Al+++ gl KN03 35°C 0.10M U M 1980KHb (21491) 278
B(AlL(tripolyphosphate))=7.97

C2H5N02 HL Acetohydroxamic CAS 546-88-3 (2766)
Acetohydroxamic acid; N-Hydroxyacetamide; CH3.CO.NHOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.20M C K1=8.15 B2=15.77 1995FKa (21804) 279
B3=21.5
B(AlH-1L2)=10.40
B(AlH-2L2)=1.04

C2H7NS HL CAS 60-23-1 (588)
2-Aminoethanethiol; H2N.CH2.CH2.SH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.10M C K1=11.74 B2=22.94 1995LMa (22487) 280
B(ALHL)=17.60

C2H703P H2L CAS 71778-99-9 (1978)
Ethylphosphonic acid; CH3.CH2.PO3H2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.20M C K1=6.63 B2=12.1 1996AKa (22566) 281
B(AlH-1L)=2.34
B(AlH-2L)=-3.93
B(AlH-1L2)=6.1

C2H8N2 L Ethylenediamine CAS 107-15-7 (23)
1,2-Diaminoethane; H2N.CH2.CH2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ oth non-aq ? 100% U M 1962BBa (23128) 282

$$K(4Al_2A_6L=Al_8A_{24}L_4)=6.2$$

Method: freezing point. Medium: benzene. HA=isopropylalcohol

C2H8O7P2 H4L HEDPA CAS 2809-21-4 (436)

1-Hydroxyethane-1,1-diphosphonic acid; CH₃.C(OH)(PO₃H₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO ₃	25°C	0.10M	C			K1=19.12 B(AlH-1L)=13.93 B(AlH-2L)=2.71 B(AlH3L2)=44.28	2002GKc (23356)	283

Al+++	gl	KNO ₃	25°C	0.10M	C		K1=22.7 B(AlHL)=27.2 B(AlH2L)=29.1 B(Al2HL)=38.7 B(AlH2L2)=43.1	1998LDa (23357)	284
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B(AlH-1L)=17.6

Al+++	gl	KCl	25°C	0.10M	U		K1=15.29 K(Al+H-1L)=21.37 K(Al+2H-1L))=25.87 K(2Al+H-1L))=27.25 K(2Al+L)=19.33	1967KLa (23358)	285
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C2H9N06P2 H4L IDPA CAS 32545-63-4 (1335)

Imino-N,N-bis(methylenephosphonic acid); HN(CH₂PO₃H₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO ₃	25°C	0.1M	C		B2=20.60 K(AlL(OH)+H)=6.92	1985MMa (23451)	286	

C2H16N504Co HL (231)

Pentaammineoxalatocobalt(III); Co(NH₃)₅(HC₂O₄)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++	sp	NaClO ₄	28°C	0.30M	U		K1=1.74	1974NDa (23472)	287
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C3H4O4 H2L Malonic acid CAS 141-82-2 (79)

Propanedioic acid; CH₂(COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++	EMF	NaCl	25°C	0.0	C	TIH	K1=7.49	B2=12.62	1998RPb (24383)	288
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Method: Pt/H₂ electrode. Calculated from data for 0.10-1.0 m NaCl, 5-75 C.

DH(K1)=19 kJ mol⁻¹, DS(K1)=208 J K⁻¹ mol⁻¹; DH(B2)=29, DS(B2)=340.

Al+++	sol none	35°C 0.00	C T	B2=11.3	1995FYa (24384) 289
Method: atomic absorption and ion chromatography.					
At 80 C, B2=14.5					
Al+++	gl KCl	25°C 0.10M	C	K1=6.711 B2=11.53 K3=2.58 *K(A1L2)=-6.68	1993PTa (24385) 290

Al+++	gl NaCl	37°C 0.15M	C	K1=6.264 B2=11.111 B3=13.3	1982JCa (24386) 291

Al+++	gl NaClO4	25°C 1.00M	U T	K(Al+HL)=3.58 K(Al+2HL)=5.99 K(Al+3HL)=8.58	1974TGa (24387) 292
At 35 C: K(Al+HL)=4.04, K(Al+2HL)=6.32, K(Al+3HL)=8.98					
Al+++	gl NaClO4	30°C 0.20M	U	K1=5.24 B2=9.40	1967AMa (24388) 293

Al+++	gl oth/un	35°C	?	K3=4.06 B3=15.84	1958DBb (24389) 294

C3H6N02Cl	HL			(8169)	
3-Chloroalanine;					

Metal	Mtd Medium Temp Conc	Cal Flags	Lg K values	Reference	ExptNo
Al+++	gl KN03	25°C 0.10M	C	K1=4.05 B2= 6.05	1981TMe (24759) 295
Also data for the schiff based formed with pyridoxal.					

C3H6N2O3	H2L			(7445)	
2-(Hydroxyimino)propanohydroxamic acid; CH3C(:NOH)CONHOH					

Metal	Mtd Medium Temp Conc	Cal Flags	Lg K values	Reference	ExptNo
Al+++	gl NaCl	37°C 0.15M	C TI	2000GKb (24824) 296	
B(AlHL)=17.370 B(AlH2L2)=34.461 B(AlH3L3)=50.130 B(AlH2L3)=43.968					
B(AlHL3)=36.176, B3=26.940, B(Al3H-1L2)=27.322. At 25C, 0.1 M KN03: B(AlHL)=17.885, B(AlH2L2)=34.996, B3=26.471, B(AlH3L3)=51.036.					

C3H6O2	HL	Propionic acid	CAS 79-09-4	(35)	
Propanoic acid; CH3.CH2.COOH					

Metal	Mtd Medium Temp Conc	Cal Flags	Lg K values	Reference	ExptNo
Al+++	gl NaCl	25°C 0.60M	C	1989MOa (24980) 297	

$$K(2Al+HL=Al_2(OH)_2L+3H)=-8.038$$

 Al+++ gl NaNO₃ 25°C 1.00M U K1=1.7 1975KIb (24981) 298

 Al+++ gl NaClO₄ 25°C 1.00M U T K1=1.78 B2=3.4 1975TRa (24982) 299

Values also at 35 C, 45 C

C3H6O₃ HL CAS 81598-26-7 (2521)
 3-Hydroxypropanoic acid; HO.CH₂.CH₂.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Al+++ sp oth/un ? ? U 1972PKa (25259) 300
 $K(Al+3HL)=7.38$

 C3H6O₃ HL L-Lactic acid CAS 79-33-4 (82)
 L-2-Hydroxypropanoic acid; CH₃.CH(OH).COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Al+++ gl NaCl 25°C 0.60M C 1990MOa (25390) 301
 $K(2Al+HL=Al_2L(OH)+3H)=-6.86$
 $K(2Al+2HL=Al_2H-4L_2+6H)=-16.79$

$$K(13Al+4HL=Al_{13}L_4(OH)_{32}+36H)=-106.9$$

 Al+++ gl NaCl 25°C 0.60M C K1=2.36 B2=4.42 1986MSb (25391) 302
 $K_3=1.37$
 $K(AlL_2=AlH-1L_2+H)=-3.45$

 Al+++ gl NaClO₄ 25°C 1.00M U T K1=4.26 B2=4.8 1975TRa (25392) 303
 Values also at 35 C, 45 C

 Al+++ sp oth/un ? ? U 1972PKa (25393) 304
 $K(Al+HL)=4.46$
 $K(Al+2HL)=6.51$

 Al+++ EMF NaNO₃ 20°C 0.20M U K1=2.38 B2=4.56 1971HUb (25394) 305
 $B_3=6.66$

 Al+++ sp oth/un ? ? U 1970PKd (25395) 306
 $B_3=3.79$
 $K(Al+L+2OH=AlOH(H-1L))=8.49$
 $K(Al+L+3OH=Al(OH)_2(H-1L))=24.6$

 Al+++ sp oth/un ? ? U K1=0.85 B2=2.92 1969PKc (25396) 307
 $K(Al+2H-1L)=23.05$

C3H7N₂O₂ HL Alanine CAS 56-41-7 (86)
 2-Aminopropanoic acid; H₂N.CH(CH₃).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaNO ₃	25°C	0.50M	C				1989DJa (26137)	308
								B(Al ₂ H-2L ₂)=7.23		
								B(AlH-3L)=-7.70		
Al+++	gl	KNO ₃	35°C	0.10M	U	M			1980KHb (26138)	309
								B(AlL(tripolyphosphate))=7.51		

C ₃ H ₇ N ₀ 2		HL					(6927)			
N-Methylacetohydroxamic acid; CH ₃ .CO.N(OH)CH ₃										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K ₁ =8.69 B ₂ =16.21 2000FEC (26620)	310	
								B ₃ =22.41		

C ₃ H ₇ N ₀ 2		HL					(7502)			
Propanohydroxamic acid; C ₂ H ₅ CONHOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K ₁ =7.97 B ₂ =15.59 2000FEC (26630)	311	
								B(AlH-1L ₂)=10.14		
								B(AlH-2L ₂)=0.3		

C ₃ H ₇ N ₀ 2S		H ₂ L	Cysteine				CAS 52-90-4 (96)			
2-Amino-3-mercaptopropanoic acid; H ₂ N.CH(CH ₂ .SH)COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO ₄	25°C	0.10M	U	M		K ₁ =6.43 1975RMa (26751)	312	
								B(AlL(citrate))=14.90		
								B(AlL(NTA))=18.89		
								K(Al+L+HPO ₄)=15.66		

C ₃ H ₇ N ₀ 3		HL	Serine				CAS 56-45-1 (49)			
2-Amino-3-hydroxypropanoic acid; H ₂ N.CH(CH ₂ .OH)COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	37°C	0.15M	C				2002DCa (27114)	313
								B(Al ₂ H-2L)=0.198		
Al+++	gl	KCl	25°C	0.20M	C			K ₁ =5.66 1997KSa (27115)	314	
								B(AlH-1L)=0.62		
								B(Al ₂ H-1L)=3.75		
Al+++	oth	NaClO ₄	35°C	0.10M	C			K ₁ =6.90 B ₂ =12.20 1986SGd (27116)	315	
								B ₃ =16.04		

Method: electrophoresis

C3H8N05P H3L 3-Phosphono-Ala CAS 20263-06-3 (1509)

2-Amino-3-phosphonatopropanoic acid; (H2O3P)CH₂.CH(NH₂).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	1.00M	C	M		K1=13.12 K(AlL+H)=3.22	1985SMd (27350)	316

ternary complexes with pyridoxal-5'-phosphoric acid

C3H8N05P H3L CAS 23052-80-4 (1508)

3-Amino-3-phosphonatopropanoic acid; (H2O3P)(NH₂)CH.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	1.00M	C			K1=10.93 K(AlL+H)=4.19	1989MSb (27361)	317

C3H8N05P H3L Glyphosate CAS 1071-83-6 (1617)

N-(Phosphonomethyl)glycine; H2O3P.CH₂.NH.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KN03	25°C	0.1M	C			K1=13.70 B(AlHL)=16.18 B(AlHL2)=27.76	1985MMa (27401)	318

C3H8N06P H3L Phosphoserine CAS 17885-08-4 (1865)

Serine dihydrogenphosphate, O-Phosphoserine; NH₂.CH(CH₂.OP03H₂).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=11.48 B(AlHL)=14.88 B(AlH-1L)=5.44 B(AlH-2L)=-2.63	1998KLb (27463)	319
Al+++	gl	KCl	25°C	0.10M	U			K1=8.50 B2=12.76	1997ZTa (27464)	320
Al+++	gl	KN03	25°C	0.10M	C			K1=4.79 B2= 8.81 B3=10.68	1981TMe (27465)	321

Also data for the schiff based formed with pyridoxal.

C3H8N202 HL Ala-hydroxamic CAS 16707-85-0 (1582)

2-Amino-N-hydroxypropanamide, Alanine hydroxamic acid; CH₃.CH(NH₂).CO.NH.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			B2=16.7	1989FSa (27576)	322

$$\begin{aligned}B(\text{A1HL}) &= 14.35 \\B(\text{A12L2}) &= 22.21 \\B(\text{A12H-1L2}) &= 17.59 \\B(\text{A12H-2L2}) &= 12.63\end{aligned}$$

Also B(A12H-3L2)=5.85; B(A12H-4L2)=-2.44; B(A1H-1L2)=9.62; B(A1H-2L2)=-0.16.

C3H8N2O2 HL (6666)
beta-Alaninehydroxamic acid; NH₂.CH₂.CH₂.CO.NHOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			B2=19.87 B(AlHL)=16.72 B(AlH2L2)=32.07 B(AlHL2)=27.04 B(AlH-1L2)=10.74	1995FKa (27606)	323

$$B(AlH_2L2) = 0.04.$$

C3H8O10P2 H5L (6577)
2,3-Diphospho-D-glyceric acid; H₂O₃P₀.CH₂.CH(COOH)OP₀H₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			B2=12.46 B(AlHL)=13.12 B(AlH2L2)=24.42 B(AlHL2)=18.78	1990SKc (27802)	324

C3H9N3O2 HL CAS 471915-95-4 (8549)
2,3-Diamino-N-hydroxypropanamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=11.7 B(AlHL2)=24.2	2002ECA (27982)	325

C3H9O6P HL CAS 17181-54-3 (7537)
1,3-Dihydroxypropyl-2-phosphoric acid: HOCH₂CH(OP(OH)₂)CH₂OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO4	25°C	0.10M	C			K1=5.30 B(Al2H-1L2)=11.72 B(Al3H-5L2)=-3.42 B(Al3H-6L2)=-10.03 B(Al3H-7L2)=-18.05	2003CCa (28029)	326

Additional method: 31P nmr. B(Al3H-1L3)=20.53. Fixed values: B(AlHL)=8.5, B(AlH2L)=9.4, B(AlHL2)=14.0.

C3H9O6P H2L CAS 57-03-4 (2984)

2,3-Dihydroxypropylphosphoric acid, Glycerol 1-phosphate; HO.CH₂.CH(OH).CH₂.OPO₃H₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	g1	NaClO4	25°C	0.10M	C			K1=5.87 B(Al2H-1L2)=11.80 B(Al3H-5L2)=-2.98 B(Al3H-6L2)=-9.51 B(Al3H-7L2)=-17.30	2003CCa (28046)	327

Additional method: ^{31}P nmr. $B(\text{Al}^3\text{H}-\text{L}^3)=20.63$. Fixed values:

$B(AlHL) = 8.5$, $B(AlH2L) = 9.4$, $B(AlHL2) = 14.0$.

C₃H₁₀N₀P H₂L (1986)

1,1-Dimethyl-1-aminomethylphosphonic acid; H2N.C(CH₃)₂.PO₃H₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	g1	KCl	25°C	0.10M	U			K1=11.42 K(Al+HL)=5.59 K(Al+2HL)=9.61 K(Al+3HL)=13.07	1969DMd (28073)	328

* * * * * H₂SiO₃ Silicic acid CAS 2802-51-5 (420)

C₄H₂O₄ HZL Squaric acid
3,4-Dihydroxy-3-cyclobutene-1,2-dione

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	ix	R4N.X	25°C	1.00M	U			K1=1.85 B2=2.74	1972CSb (28635)	329
Medium: NH4ClO4										

Al+++ ix NaClO4 25°C 0.30M II K1=2 83 1969Twa (28636) 330

C4H4O4 H2I Maleic acid CAS 110-16-7 (111)

cis-Butenedioic acid: HOOC-CH=CH-COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaClO4 25°C 0.04M M K1=5.48 B2= 8.78 1993MYa (29046) 331
Medium: 0.01 M HClO4, 0.005 M Al(ClO4)3.

* * * * * C4HEN303 * * * * * III * * * * * C Aminoacetoxylic * * * * * CAS 873-83-6 (6212)

C4H5N3O2 HL 6-A
4-Amino-3,6-dihydroxypyrimidine:

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl04	30°C	0.10M	U			K1=13.32 B2=25.27	1986JDa (29421)	332

C4H6O3		HL					CAS	600-18-0 (5474)		
2-Ketobutanoic acid; CH ₃ .CH ₂ .CO.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C		K1=2.09 B2=3.65	1982KMc (29746)	333

C4H6O3		HL		Acetoacetic aci	CAS	541-50-4	(5475)		
3-Ketobutanoic acid; CH ₃ .CO.CH ₂ .COOH									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C		K1=1.21 B2=1.84	1982KMc (29750)	334

C4H6O4		H2L		Succinic acid	CAS	110-15-6	(112)		
1,4-Butanedioic acid; HOOC.CH ₂ .CH ₂ .COOH									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	NaCl	37°C	0.15M	C		K1=3.753 B(AlHL)=7.154 B(Al2H-2L)=-1.103 B(Al3H-1L2)=9.114 B(Al3H-3L2)=1.385	1998VBa (29928)	335
B(Al4H-6L3)=-4.44, B(Al8H-12L4)=-12.16									
Al+++	gl	KCl	25°C	0.20M	C		K1=3.63 B(AlHL)=7.03 B(AlH-1L)=-0.53 B(AlH-2L)=-5.55	1997KSa (29929)	336
Al+++	gl	NaCl	37°C	0.15M	C		K1=4.17 B(AlHL)=7.18 B(AlH-2L)=-4.65 B(Al2H-3L)=-4.94	1990FDa (29930)	337
Al+++	gl	NaCl	37°C	0.15M	U		K1=3.91 B(Al2H-3L)=-5.23 B(Al3H-2L2)=5.34	1987VBe (29931)	338
Al+++	gl	NaClO ₄	25°C	0.50M	U		K1=3.2 B(AlHL)=6.60 B(AlH-1L)=4.2	1984CDa (29932)	339
Al+++	gl	NaCl	37°C	0.15M	C		B(AlH-1L)=-0.25 B(Al2H-1L)=2.88 B(AlH-2L)=-5.19	1982JCa (29933)	340
Al+++	gl	NaClO ₄	25°C	1.00M	U		K(Al+HL)=3.84 K(Al+2HL)=5.93 K(Al+3HL)=8.98	1974TGa (29934)	341

At 35 C: K(Al+HL)=3.78, K(Al+2HL)=6.40, K(Al+3HL)=9.52

C4H604 H2L Me-Malonic Acid CAS 516-15-2 (816)
Methylpropanedioic acid; HOOC.CH(CH3).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.60M	C				1990MOb (30114)	342
								K(Al+H2L=AlL+2H)=-2.213		
								K(Al+2H2L=AlL2+4H)=-5.73		
								K(Al+3H2L=AlL3+6H)=-11.19		

C4H604S H2L Thiodiacetic CAS 123-93-3 (140)
2,2'-Thiodiglycolic acid, Thiodiethanoic acid; HOOC.CH2.S.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO4	25°C	0.50M	U			K1=1.93	1972NAb (30210)	343
								B(AlL(OH))=12.34		

C4H604S H3L Thiomalic acid CAS 70-49-5 (109)
2-Mercaptosuccinic acid, 2-Sulfanyl-1,4-butanedioic acid; HOOC.CH(SH).CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=8.63	1997KSa (30318)	344
								B(AlHL)=12.99		
								B(AlH-1L)=4.05		

C4H605 H2L Malic acid CAS 617-48-1 (393)
2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	37°C	0.15M	C			K1=4.519	2001VBa (30578)	345
								B(AlHL)=7.032		
								B(AlHL2)=10.981		
								B(AlH-1L)=1.268		
								B(Al2H-2L)=0.564		

B(Al2H-3L)=-3.054, B(Al2H-3L2)=1.779, B(Al2H-4L2)=-4.462,
B(Al2H-1L3)=12.789, B(Al3H-4L4)=10.132, B(Al4H-5L4)=10.537.

Al+++	sp	oth/un	23°C	0.10M	U				1994KGa (30579)	346
								Keff=11.6		

Method: spectrophotometric using pyrocatechol violet. Tris buffer adjusted
to a pH=5.34 with HCl

Al+++	gl	NaCl	37°C	0.15M	C			K1=4.37 B2=8.17	1990FDa (30580)	347
								B(AlH-1L2)=4.11		

Al+++	gl	NaNO ₃	25°C	0.50M	M	M	1989MAa (30581) 348	
					B(-3,1,1)=-10.5			
					K(2AlH-2L=Al2H-4L2)=-22.2			
B(p,q,r):	pH+qM+rH2L.	K(UO ₂ +Al+2H2L=UO ₂ AlH-4L2+8H)=-8.06						

Al+++	gl	NaCl	37°C	0.15M	U	K1=4.60 B2= 7.62	1987VBe (30582) 349	
					B(AlHL)=6.87			
					B(AlHL2)=11.31			
					B(AlH-1L2)=4.31			
					B(Al2H-2L2)=5.59			
B(Al2H-3L2)=2.50								

Al+++	gl	NaClO ₄	25°C	0.01M	U	1976MPb (30583) 350		
					K(Al+H2L=AlH-1L+3H)=-5.39			
					K(AlH-2L+H)=4.72			
					K(AlH-3L+H)=7.60			

Al+++	EMF	KNO ₃	20°C	0.20M	U	K1=5.34 B2=9.32	1969PVc (30584) 351	

Al+++	sp	NaClO ₄	29°C	1.0M	U	K1=3.32	1965MNa (30585) 352	
*****	*****	*****	*****	*****	*****	*****	*****	
C4H6O5	H2L	Diglycolic acid	CAS	110-99-6	(243)			
Di(carboxy)methyl ether, 2,2'-Oxydiethanoic acid; HOOCH ₂ OCH ₂ COOH								

Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo	

Al+++	gl	NaClO ₄	25°C	0.10M	U	TIH	K1=3.43 B2=6.42	1979SDc (30851) 353

Al+++	gl	NaClO ₄	25°C	0.50M	U		K1=3.16 B2=5.25	1972NAd (30852) 354
*****	*****	*****	*****	*****	*****	*****	*****	
C4H6O6	H2L	DL-Tartaric acid	CAS	133-37-9	(94)			
DL-Tartaric acid, DL-2,3-Dihydroxybutanedioic acid; HOOCH(OH).CH(OH).COOH								

Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo	

Al+++	gl	NaNO ₃	25°C	0.50M	M	M	1989MAa (31009) 355	
						B(-4,1,1)=-7.8		
						K(2AlH-2L=Al2H-4L2)=-18.5		
B(p,q,r):	pH+qM+rH2L.	K(UO ₂ +Al+2H2L=UO ₂ AlH-4L2+8H)=->-12						
*****	*****	*****	*****	*****	*****	*****	*****	
C4H6O6	H2L	L-Tartaric acid	CAS	87-69-4	(92)			
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOCH(OH).CH(OH).COOH								

Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo	

Al+++	gl	NaCl	37°C	0.15M	C	K1=3.788	2000DDb (31185) 356	
						B(AlH-1L)=1.165		
						B(Al2H-1L2)=7.976		
						B(Al2H-2L2)=5.347		
						B(Al2H-3L2)=1.009		

B(Al2H-4L2)=-5.101

Al+++ sp oth/un 23°C 0.10M U 1994KGa (31186) 357
Keff=11.7

Method: spectrophotometric using pyrocatechol violet. Tris buffer adjusted to a pH=6.00 with HCl

Al+++ gl NaCl 25°C 0.60M C 1990M0d (31187) 358
B(1,-2,1)=-3.44
B(2,-5,2)=-6.30
B(2,-6,2)=-8.91
B(2,-7,2)=-13.12

B(2,-8,2)=-18.95. B(p,q,r): pAl+qH+rH2L=AlpHq(H2L)r

Al+++ gl KN03 25°C 0.10M C B2=7.65 1984MMb (31188) 359
B(AlH-1L)=1.18
*K(AlH-1L)=-5.15
*K(AlL2)=-3.72
K(AlH-3L2+2H)=12.67

Al+++ gl NaClO4 25°C 0.10M U K1=5.62 B2=9.95 1972MRc (31189) 360
Meso Tartaric acid: K1=5.32, K2=4.45.

Al+++ oth oth/un 20°C 0.0 U K1=6.35 1967FRa (31190) 361
K(Al+HL)=3.43
K(AlOH+L=AlH-1L)=9.05
K(Al(OH)2+L=AlH-2L)=10.92
K(Al(OH)3+L=AlH-2LOH)=8.37

Method: optical rotation. K(Al(OH)4+L=Al(H-2L)(OH)2)=8.89, B(Al2L)=2, K(Al+AlH-1L)=3, K(Al+H-2L)=4, K(AlL+H2L)=0.66, K(AlH-1L+H2L)=1.12 plus others

Al+++ oth oth/un 20°C ? U 1967PTa (31191) 362
K(AlOH+L)=1.8

Method: refraction

Al+++ dis NaClO4 20°C 0.10M U B2=9.56 1963STc (31192) 363

C4H7N04 H2L Aspartic acid CAS 56-84-8 (21)
Aminobutanedioic acid; H2N.CH(CH2.CO0H).CO0H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.20M C K1=7.87 1997KSa (31811) 364
B(AlHL)=11.76
B(AlH-1L)=3.30
B(AlH-2L)=-2.32

Al+++ gl NaClO4 25°C 0.50M U 1984CDa (31812) 365
K(Al+HL)=2.16
K(Al(OH)+HL)=3.03

Al+++ gl KN03 35°C 0.10M U M 1980KHB (31813) 366
B(AlHL(tripolyphosphate))=8.29

Al+++ gl NaClO4 25°C 0.10M U K1=16.29 B2=30.69 1972SSe (31814) 367
K3=11.50

C4H7N04 H2L IDA CAS 142-73-4 (118)
Iminodiethanoic acid; HN(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	25°C	0.10M	U			K1=8.10	1997YSA (32195)	368

Al+++ gl NaClO4 25°C 0.10M U T K1=8.62 B2=16.12 1981DSa (32196) 369
At 35 C: K1=8.42, B2=16.62; 45 C: 8.28, 15.49

Al+++ gl NaClO4 25°C 0.50M U K1=8.10 B2=15.07 1971LnB (32197) 370

C4H7N05 H2L (1234)
N-Hydroxyiminodiethanoic acid; HO.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KN03	25°C	0.10M	C			K1=5.1	1987AKa (32426)	371

C4H8N02C1 HL (8170)
3-Chloro-2-aminobutanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KN03	25°C	0.10M	C			K1=4.13 B2= 6.33	1981TMe (32468)	372

Also data for the schiff based formed with pyridoxal.

C4H8N203 HL Asparagine CAS 70-47-3 (17)
2-Aminobutanedioic acid 4-amide; H2N.CH(CH2.CO.NH2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=5.50	1997KSa (32681)	373

C4H8N203 HL Gly-Gly CAS 556-50-3 (54)
Glycyl-glycine; H2N.CH2.CO.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KN03	35°C	0.10M	U				1980KHB (33017)	374

B(AlHL(tripolyphosphate))=4.18

C4H8N2O4 H2L (6369)
 N(1)-Hydroxyasparagine, aspartyl-beta-hydroxamic acid; H2N.CH(CH₂.CO.NHOH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			B2=19.26 B(AlHL)=16.27 B(AlH2L2)=31.76 B(AlHL2)=26.80 B(AlH-1L2)=10.36	1995FKa	(33132) 375

C4H8O2 HL CAS 107-92-6 (1118)

n-Butanoic acid; CH₃.CH₂.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++	gl	NaNO ₃	25°C	1.00M	U			K1=1.6	1975KIb	(33328) 376
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C4H8O2S HL CAS 627-04-3 (3007)

S-Ethylthioethanoic acid; CH₃.CH₂.S.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++	gl	diox/w	30°C	50%	U			K1=3.45	1956IFa	(33403) 377
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C4H9N02S HL Methylcysteine CAS 1187-84-4 (84)

2-Amino-3-methylmercaptopropanoic acid; H2N.CH(CH₂.S.CH₃).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++	dis	NaClO ₄	35°C	0.10M	U	M		K1=7.68	1995TKa	(34094) 378
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Method: Paper electrophoresis; Ternary complexes with NTA.

C4H9N03 HL Threonine CAS 72-19-5 (48)

2-Amino-3-hydroxybutanoic acid; H2N.CH(CH(OH).CH₃).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++	gl	NaCl	37°C	0.15M	C				2002DCa	(34287) 379
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B(Al2H-2L)=-0.177

Al+++	gl	KCl	25°C	0.20M	C			K1=5.51	1997KSa	(34288) 380
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B(AlH-1L)=0.94

Al+++	oth	NaClO ₄	35°C	0.10M	C			K1=7.94	1986SGd	(34289) 381
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B3=18.94

Method: electrophoresis

Al+++	gl	KNO ₃	35°C	0.10M	U	M			1980KHb	(34290) 382
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B(AlL(tripolyphosphate))=7.55

C4H100 L Ether CAS 60-29-7 (3573)
Diethyl ether (ethyl ether, ethoxyethane); C2H5.O.C2H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ oth oth/un 20°C 0.07M U K1=1.34 1965PBc (34651) 383

Method: mass spectrograph. Medium: AlI3

C4H11N3O2 HL CAS 471915-94-3 (8550)
2,4-Diamino-N-hydroxybutanamide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.20M C B2=19.81 2002ECa (35175) 384
B(AlHL)=17.3
B(AlHL2)=27.2
B(AlH2L2)=34.0

C5H202F6 HL HFA CAS 1522-22-1 (195)
1,1,1,5,5,5-Hexafluoropentane-2,4-dione; F3C.CO.CH2.CO.CF3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ nmr non-aq 25°C 100% U H 1964PCa (35920) 385
Method:NMR, medium:CHCl3. DG(trans-AlL3=cis-AlL3)=3.8 kJ mol-1, DH=1.0, DS=-8

C5H5NO2 HL CAS 13161-30-3 (5582)
1-Hydroxypyridin-2(1H)-one, 2-Hydroxypyridine 1-oxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.20M C B2=12.5 2000FEc (36751) 386

Al+++ gl KCl 25°C 0.10M U K1=8.16 B2=15.54 1993LMc (36752) 387
K3=6.05

C5H5NO2 HL CAS 16867-04-2 (2316)
2,3-Dihydroxypyridine, 3-Hydroxypyridin-2(1H)-one; C5H3N(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 25°C 0.60M U T K1=8.59 B2=16.34 1999DBa (36779) 388
B3=23.11
B(AlH-1L3)=13.85
At 37 °C, K1=8.19, B2=16.03, B3=21.77, B(AlH-1L3)=13.0

Al+++ kin NaClO4 34°C 0.10M C 1979BMb (36780) 389
K(Al+H2L=AlL+2H)=-1.85

Method: stopped-flow.

C5H5N03 H2L CAS 99110-85-7 (2195)
1,4-Dihydroxy-2-pyridinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C			B2=21.20 B3=25.16 B(AlHL)=16.5 B(AlHL2)=27.22 B(AlH2L2)=32.64	1992CMc (36841)	390

C5H6O5 H2L CAS 642-93-3 (5476)
3-Methyl-2-oxobutanedioic acid HOOC.CO.CH(CH3).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C			K1=6.16 B2=10.80	1982KMc (37479)	391

C5H8O2 HL Acetylacetone CAS 123-54-6 (164)
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	dis	NaClO4	25°C	0.10M	C			K1=8.7	1986SNa (37910)	392

Method: rate of distribution of volatile ligand between aqueous phase and inert gas phase. K(H+L)=9.17 assumed.

Al+++	oth	NaClO4	25°C	0.10M	C	I	T	K1=8.2 B2=15.7 B3=21.4	1982SLc (37911)	393
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IUPAC evaluation. I=0 corr.: K1=8.6, B2=16.5, B3=22.3

Al+++	gl	oth/un	30°C	0.0	U			K1=8.6 B2=16.5 K3=5.8	1955IFa (37912)	394
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C5H9N02 HL Proline CAS 147-85-3 (44)
Pyrrolidine-2-carboxylic acid; C4H8N.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO3	35°C	0.10M	U	M			1980KHb (38601)	395

B(All(tripolyphosphate))=8.63

C5H9N03 HL Hydroxyproline CAS 51-35-4 (416)
4-Hydroxy-2-pyrrolidinecarboxylic acid; C4H7N(OH)(COOH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO3	35°C	0.10M	U	M			1980KHb (38719)	396

B(All(tripolyphosphate))=7.83

C5H9N04 H2L Glutamic acid CAS 56-86-0 (22)
2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 37°C 0.15M C 2003DBa (39059) 397
B(AlH2L)=15.015
B(Al2H-1L)=6.680
B(AlHL2)=18.850
B(AlH2L2)=23.425

B(Al3H-4L2)=3.810, B(Al3H-4L3)=11.220.

Al+++ gl KCl 25°C 0.20M C K1=7.29 1997KSa (39060) 398
B(AlHL)=10.88
B(AlH-1L)=2.55
B(Al2L)=9.46

Al+++ gl NaClO4 25°C 0.50M U 1984CDa (39061) 399
K(Al+HL)=2.30
K(Al(OH)+HL)=3.50

Al+++ gl KN03 35°C 0.10M U M 1980KHb (39062) 400
B(AlHL(tripolyphosphate))=7.97

Al+++ gl NaClO4 25°C 0.10M U K1=15.12 B2=29.40 1972SSe (39063) 401
K3=9.20

C5H9N04 H2L MIDA CAS 4408-64-4 (190)
N-Methyliminodiethanoic acid; CH3.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaClO4 25°C 0.50M C K1=7.55 1984NAa (39238) 402

C5H10N07P H4L PMIDA CAS 5994-61-6 (2433)
N-(Phosphonomethyl)iminodiethanoic acid; H2O3P.CH2.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.10M U K1=14.7 1980VRa (39666) 403
K(Al+HL)=7.3

C5H10N203 HL Glutamine CAS 56-85-9 (18)
2-Aminopentanedioic acid 5-amide; H2N.CH(CH2.CH2.CO.NH2)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.20M C K1=5.61 1997KSa (39811) 404
B(AlH-1L)=1.33

C5H10N204		HL		CAS 1955-67-5 (6736)				
2-Aminopentanoic-5-hydroxamic acid; HOOC.CH(NH2).CH2.CH2.CO.NOH								

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo
Al+++	gl	KCl	25°C	0.20M	C		B2=19.89 B(AlHL)=16.65 B(AlH2L2)=32.79 B(AlHL2)=27.62 B(AlH-1L2)=9.9	1995FKa (40077) 405
B(AlH-2L2)=1.10								

C5H10O4		HL					(7178)	
2,5-Dihydroxypentanoic acid; HOCH2CH2CH2CHOHCOOH								

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo
Al+++	gl	NaCl	25°C	0.10M	C		K1=2.04 *K(AlL)=-3.14 *K(AlH-1L)=-2.25	1994BHa (40324) 406

C5H11N02		HL	Valine		CAS 72-18-4 (43)			
2-Amino-3-methylbutanoic acid; H2N.CH(CH(CH3)2)COOH								

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo
Al+++	gl	KNO3	35°C	0.10M	U M			1980KHb (40687) 407

C5H11N02S		HL	Methionine		CAS 63-68-3 (42)			
2-Amino-4-(methylthio)butanoic acid; H2N.CH(CH2.CH2.S.CH3)COOH								

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo
Al+++	oth	oth/un	25°C	0.10M	C		K1=7.00 B2=11.50	1998TEb (41077) 408
Method: electrophoresis. Medium: 0.1 M HClO4.								

C5H11N02S		H2L	Penicillamine		CAS 52-66-4 (350)			
DL-2-Amino-3-mercaptopropanoic acid; (CH3)2C(SH)CH(NH2)COOH								

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo
Al+++	oth	NaClO4	35°C	0.10M	C		K1=11.50 B2=15.05	1996TKb (41252) 409
Method: paper electrophoresis.								

C5H11O8P		H2L	Ribose-5-phosph		CAS 4300-28-1 (2756)			
Ribose-5-phosphoric acid, Ribofuranoside 5 Phosphoric acid;								

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo

Al+++	gl	KCl	25°C	0.20M	C	K1=5.63 B(AlH-1L)=1.69 B(AlH-2L)=-4.83	1996AKa (41420)	410	

C6H3N3O7		HL	Picric acid	CAS	88-89-1	(593)			
2,4,6-Trinitrophenol; HO.C6H2(NO2)3									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo	
Al+++	sp	oth/un	21°C	0.40M	U		K1=1.05 B3=3.12	1955BKa (42091)	411

C6H4N2O5		HL		CAS	50-28-5	(505)			
2,4-Dinitrophenol; HO.C6H3(NO2)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo	
Al+++	sp	oth/un	21°C	0.40M	U		K1=0.89	1955BKa (42223)	412
Medium:0.2-0.7(some EtOH)									

C6H4N2O6		H2L		CAS	7659-29-2	(2694)			
1,2-Dihydroxy-3,5-dinitrobenzene; (HO)2.C6H2(NO2)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo	
Al+++	gl	KCl	25°C	0.10M	C		B2=21.80 B3=31.68	2004GAa (42263)	413

C6H5NC12		L	Dichloroaniline	CAS	554-00-7	(761)			
2,4-Dichloroaniline; H2N.C6H3(Cl)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo	
Al+++	sp	diox/w	25°C	100%	U			1976BSa (42347)	414
K(AlCl3+L)=2.14									

C6H5NC12		L	Dichloroaniline	CAS	95-76-1	(759)			
3,4-Dichloroaniline; H2N.C6H3(Cl)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo	
Al+++	sp	diox/w	25°C	100%	U			1976BSa (42352)	415
K(AlCl3+L)=2.83									

C6H5N02		HL	Picolinic acid	CAS	98-98-6	(391)			
2-Pyridine-carboxylic acid; C5H4N.COOH									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo	

Al+++ nmr NaCl 25°C 1.5M C K1=4.11 B2= 7.90 2000Lkb (42494) 416
Method: 27Al nmr.

Al+++ gl NaCl 37°C 0.15M C K1=4.35 B2=8.48 1990FDA (42495) 417
B(AlH-1L2)=4.15
B(AlH-2L)=-3.19

Al+++ gl KN03 25°C 0.15M U K1=4.487 B2=8.419 1988JJa (42496) 418
B(Al(OH)L2)=17.589
B(Al(OH)2L)=21.73

Al+++ gl NaClO4 25°C 0.50M C K1=4.62 B2=8.60 1986Mnb (42497) 419
B3=12.3

Al+++ gl KN03 25°C 0.15M U K1=4.497 B2= 8.27 1985JJa (42498) 420
B(AlH-1L2)=17.668
B(Al2H-3L2)=39.27

C6H5N03 H2L CAS 609-71-2 (5910)

2-Hydroxypyridine-3-carboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 25°C 0.60M C K1=12.48 1999MTb (42722) 421
B(AlHL)=18.02
B(AlH2L2)=34.73
B(AlH3L3)=49.92
B(AlH2L3)=43.0

Confirmed by H-nmr. By spectrophotometry: B(AlHL)=17.96

C6H5N03 H2L CAS 874-24-8 (4356)

3-Hydroxypyridine-2-carboxylic acid; C5H3N.(OH)(COOH)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 25°C 0.60M C K1=11.89 B2=21.13 1999MTb (42747) 422
B(AlHL)=16.91
B(AlH2L2)=32.62
B(AlHL2)=27.2
B(AlH3L3)=46.91

B(AlH2L3)=41.2, B(AlHL3)=34.7, B3=27.04. H-nmr also used.

By spectrophotometry: B(AlHL)=16.89

C6H5N03 H2L CAS 10128-71-9 (8910)

3-Hydroxypyridine-4-carboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 25°C 0.60M C K1=10.84 B2=19.50 2002DYa (42758) 423
B3=26.09

$B(AlHL)=15.97$
 $B(AlH2L2)=30.78$
 $B(AlHL2)=25.47$
 $B(AlH3L3)=44.05$, $B(AlH2L3)=38.69$, $B(AlHL3)=32.67$.

C6H5N03 H2L CAS 609-70-1 (8911)

4-Hydroxypyridine-3-carboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.60M	C			B2=20.5 B3=25.76 $B(AlHL)=18.19$ $B(AlH2L2)=35.15$ $B(AlHL2)=28.7$	2002DYa (42775)	424

$B(AlH3L3)=50.76$, $B(AlH2L3)=42.92$, $B(AlHL3)=34.77$.

C6H5N04 H2L 3-Nitrocatechol CAS 6665-98-1 (2685)

1,2-Dihydroxy-3-nitrobenzene; O2N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	M			K1=14.17 B2=26.30 B3=35.81	1986HAb (42855)	425

C6H5N04 H2L 4-Nitrocatechol CAS 3316-09-4 (890)

1,2-Dihydroxy-4-nitrobenzene; O2N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C			K1=13.89 B2=26.33 B3=37.08	2004GAa (42911)	426
Al+++	gl	KCl	25°C	0.10M	C			K1=13.75 B2=25.44 B3=34.38 $B(AlH-1L2)=17.93$	1997DSa (42912)	427

Al+++	gl	KCl	25°C	0.10M	M			K1=13.74 B2=25.39 B3=34.31	1984HAd (42913)	428

C6H5N2O2C1 L CAS 635-22-3 (763)

3-Nitro-4-chloroaniline; H2N.C6H3(C1)(NO2)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	diox/w	25°C	100%	U				1976BSa (42977)	429
								$K(AlCl3+L)=1.67$		

C6H6NC1 L o-Chloroaniline CAS 95-51-2 (3088)

2-Chloroaniline (1-amino-2-chlorobenzene); C1.C6H4.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sol	NaNO ₃	55°C	0.10M	C	T		K1=<-2.37	1998YFa (43199)	430
At 80 °C, K1<-2.50.										

C6H6N06P		H2L						CAS 330-13-2 (5865)		
4-Nitrophenylphosphoric acid; NO ₂ .C ₆ H ₄ .O.PO.(OH)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=4.80	1996AKa (43246)	431
B(AlH-1L)=1.19										
B(AlH-2L)=-5.5										

C6H6N202		HL						(8281)		
3-Hydroxy-2-amidocarboxypyridine, Hydroxypicolinamide;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO ₃	25°C	0.10M	C			K1=7.54 B2=13.79	1990ARa (43373)	432

C6H6N202		L	m-Nitroaniline		CAS 99-09-2	(464)				
3-Nitroaminobenzene; H ₂ N.C ₆ H ₄ .NO ₂										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	diox/w	25°C	100%	U				1976BSa (43387)	433
K(AlCl ₃ +L)=2.35										

C6H6N202		L	p-Nitroaniline		CAS 100-01-6	(465)				
4-Nitroaminobenzene; H ₂ N.C ₆ H ₄ .NO ₂										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	diox/w	25°C	100%	U				1976BSa (43404)	434
K(AlCl ₃ +L)=1.52										

C6H6N203		H2L						CAS 2504-83-8 (1141)		
Imidazolylpyruvic acid; C ₃ H ₃ N ₂ .CH ₂ .CO.CO ₂ H										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	U			K1=9.0 B2=16.10	1975SDa (43451)	435

C6H6O2		H2L	Catechol		CAS 120-80-9	(534)				
1,2-Dihydroxybenzene, pyrocatechol; HO.C ₆ H ₄ .OH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

Al+++ gl KCl 25°C 0.20M C M K1=16.20 B2=29.26 1993Kaa (43715) 436
 B3=37.95
 B(Al2H-2L2)=24.05

B(AlL)=26.92, B(AlL2)=35.2, B(AlL2A)=37.12. H2A=salicylic acid

 Al+++ sp KCl 25°C 0.10M C K1=16.22 1989SMa (43716) 437

 Al+++ gl KCl 25°C 0.10M C K1=16.89 B2=30.55 1985KPa (43717) 438
 K3=8.98
 *K(AlL)=-6.07
 *K(AlL2)=-8.10

 Al+++ gl KN03 25°C 0.10M C 1984MMb (43718) 439
 K(Al+H2L=AlL+2H)=-6.08
 K(AlL+H2L=AlL2+2H)=-9.18
 K(AlL2+H2L=AlL3+2H)=-13.52
 K(AlL2+H)=6.03

 Al+++ gl NaCl 25°C 0.60M U 19830Sa (43719) 440
 B(-2,1,1)=-6.337
 B(-4,1,2)=-15.44
 B(-6,1,3)=-28.62
 B(-5,1,2)=-23.45

B(-9,3,3)=-29.91. B(p,q,r): pH+qAl+r(H2L)

 Al+++ gl KN03 25°C 0.20M U K1=15.31 B2=27.67 1982H0b (43720) 441
 K3=7.74

 Al+++ gl KCl 25°C 0.20M U K1=16.27 B2=29.75 1970G0a (43721) 442
 K3=9.00

 Al+++ gl KN03 20°C 0.10M U K1=16.9 B2=30.50 1969HBa (43722) 443
 K3=8.9

 Al+++ gl KN03 ? 0.20M U K1=16.56 B2=32.20 1964DMa (43723) 444
 K3=13.65

C6H6O3 H3L Pyrogallol CAS 87-66-1 (696)
 1,2,3-Trihydroxybenzene; C6H3(OH)3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Al+++ gl KCl 22°C 0.20M U K1=24.50 B2=44.55 1970G0b (43946) 445
 K3=13.40

 Al+++ gl KN03 ? 0.20M U 1967DMa (43947) 446
 K(Al+HL)=14.3
 K(AlHL+HL)=13.5
 K(Al(HL)2+HL)=11.9

C6H6O3 HL Isomaltol CAS 3420-59-5 (5885)
1-(3-Hydroxy-2-furanyl)ethanone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.15M	C			K1=5.66 K2=4.92	1989LCa	(44031) 447

C6H6O3 HL Maltol CAS 118-71-8 (2442)
3-Hydroxy-2-methyl-4H-pyran-4-one:

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ nmr NaCl 25°C 1.50M C T H 2002YPa (44073) 448
 Method: 27Al and 170 nmr measurements at 0-65 C.
 $DH(Al+HL=AlL+H)=22 \text{ kJ mol}^{-1}$; $DH(AlL+HL=AlL2+H)=28 \text{ kJ mol}^{-1}$.

Al+++ gl NaCl 25°C 0.60M C 1988Ho (44075) 450
 $B(-1,1,1) = -0.130$
 $B(-2,1,2) = -0.956$
 $B(-3,1,3) = -2.669$
 $B(-4,2,2) = -7.203$

$$B(p,q,r): pH + q Al + r (HL) \rightarrow H_p Al_q (HL)_r$$

Al+++ g1 KNO₃ 25°C 0.10M U K1=7.7 B2=15.25 1969CBb (44076) 451
K3=6.65

C6H6O4 HL Kojic acid CAS 501-30-4 (1800)
5-Hydroxy-2-(hydroxymethyl)-4H-pyran-4-one:

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 25°C 0.60M C 1988Ho (44194) 452
 $B(-1,1,1) = -0.371$
 $B(-2,1,2) = -1.499$
 $B(-3,1,3) = -3.564$
 $B(-4,2,2) = -7.656$

$$B(p,q,r): pH + q Al + r (HL) \rightarrow HpAlq(HL)r$$

Al+++ sp KCl 25°C 0.10M C K1=7.66 1987PEa (44195) 453

Al+++ EMF KC1 21°C 0.10M U K1=7.7 B2=14.2 19590Kb (44196) 454
B3=19.5

Method: H electrode

C6H6O5S H3L CAS 7134-09-0 (3687)
3,4-Dihydroxybenzenesulfonic acid; (HO)₂C₆H₃S(O₂H)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
Al+++	gl	KNO ₃	20°C	0.10M	U			K1=16.6 K3=9.3	B2=29.90	1969HBb (44280)	455

C6H6O8S2 H4L Tiron CAS 149-45-1 (104)
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)₂.C6H₂(SO₃H)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
Al+++	gl	KCl	25°C	0.20M	C	M		K1=15.75 B3=38.97 B(AlAL)=22.48 B(AlA2L)=34.27	B2=29.10	2002FCa (44397)	456

A is acetohydroxamic acid.

Al+++	gl	KNO ₃	25°C	0.10M	C				1988YYa (44398)	457
								K(Al+H2L=A1L+2H)=-3.11		
								K(A1L+H2L=A1L2+2H)=-6.26		
								K(A1L2+H2L=A1L3+2H)=-9.9		

Al+++	gl	KCl	30°C	0.10M	U	TIH		K1=15.48 K3=12.72	B2=30.09	1980BDe (44399)	458
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Data for I=0.20 and 0.30 M. Data at 40 C. DH and DS values.
At I=0, K1=16.00, K2=15.18, K3=13.00.

Al+++	kin	NaClO ₄	34°C	0.10M	C				1979BMb (44400)	459
								K(Al+H2L=A1L+2H)=-2.62		

Method: stopped-flow.

Al+++	gl	KNO ₃	20°C	0.10M	U			K1=16.7 K3=9.7	B2=30.30	1969HBa (44401)	460
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Al+++	gl	NaNO ₃	25°C	0.20M	U			K1=16.65	B2=30.25	1968ASa (44402)	461
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Al+++	con	KNO ₃	?	0.10M	U			K1=16.81		1965DMA (44403)	462
								By glass electrode: K1=16.79, K2=16.58, K3=14.34			

Al+++	gl	oth/un	25°C	0.0	U			K1=19.02 K3=2.4	B2=31.10	1957NAd (44404)	463
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C6H7N L Aniline CAS 62-53-3 (583)
Aminobenzene, aniline; C6H₅.NH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++	sol	NaNO ₃	80°C	0.10M	C			K1=<-0.93		1998YFa (44867)	464
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C6H7NO₂ HL CAS 19365-01-6 (6771)

1-Methyl-3-hydroxy-2-pyridinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C			K1=9.41	B2=17.79	1992CMc	(45023) 465
								B3=25.10			

C6H7NO2 HI CAS 17184-19-9 (5888)

3-Hydroxy-2-methylpyridin-4(1H)-one:

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 25°C 0.15M C I K1=11.87 B2=22.54 1989CNa (45047) 466
B3=32.05

Data also at I=0.6 M(NaCl): K1=11.43, B2=21.73, B3=30.41

C6H7O4P H2L CAS 701-64-4 (5866)

Phenyl phosphoric acid; C₆H₅O₃PO(OH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ g1 KCl 25°C 0.20M C K1=5.29 1996AKa (45230) 467
B(AlH-1L)=1.51
B(AlH-2L)=-4.5

C6H8O6 H3L Tricarballylic CAS 99-14-9 (1620)
1,3,2-Dihydro-1,1-dimethyl-2,2,2-trifluoro-3,3,3-trifluoropropyl 1,1,1-trifluoro-2-hydroxy-3,3,3-trifluoropropyl ester

1,2,3-Propanetricarboxylic acid; HOOC.CH₂.CH(COOH).CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

CCUSOC H2I Acetophenone acid CAS 50-81-7 (285)

C₆H₈O₆ H₂L
Ascorbic acid (Vitamin C)

Metal Mtd. Medium Temp. Conc. Gal. Flags. Ig. K values Reference ExptNo

Aliq. -51 NaCl -25°C 0.6M G 1002ONa (45631) 468

Al+++ g1 NaCl 25°C 0.60M C B(-1,1,1)=-2.59
B(-6,3,1)=-18.38
B(-9,3,4)=-24.19

$$B(p,q,r); \quad pH + qAl + rHL = Hp(Al)q(HL)r$$

Note: L-ascorbic acid is here defined as H-

C6H8O7 H3L Isocitric acid CAS 1637-73-6 (2527)

2-Hydroxy-3-carboxypentanedioic acid; HOOC.CH(OH).CH(COOH).CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.10M C K1=6.905 1995PTa (45731) 470
B(AlHL)=9.55
B(AlH-1L)=3.06
B(Al2H-3L2)=4.08

An alternative model gave K1=6.96, B(AlHL)=9.37, B(AlH-1L)=2.92,
B(Al3H-4L3)=10.36.

C6H8O7 H3L Citric acid CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.20M C K1=7.85 B2=12.73 2001LBa (46012) 471
B(AlHL)=10.18
B(AlH-1L)=4.27
B(AlH-2L)=-1.77
B(AlH-1L2)=7.81

B(AlH-2L2)=0.4, B(Al3H-4L3)=16.34.

Al+++ cal oth/un 25°C 0.04M C H 1997WDa (46013) 472
DH(Al+L=Al(OH)L+H)=41.6 kJ mol-1; DH(Al(OH)L+3OH=Al(OH)4+L)=-29.2 kJ mol-1
DH(Al+4OH=Al(OH)4)=-43.4 kJ mol-1.

Al+++ sp oth/un 23°C 0.10M U 1994KGa (46014) 473
Keff=9.4

Method: spectrophotometric using pyrocatechol violet. Tris buffer adjusted
to a pH=5.34 with HCl

Al+++ gl NaCl 37°C 0.15M C K1=8.34 B2=13.56 1990FDa (46015) 474
B(AlHL)=10.78
B(AlH-1L)=5.30
B(AlH-1L2)=7.99
B(AlH-2L)=-0.59

Al+++ gl NaNO3 25°C 0.50M M M 1989MAa (46016) 475
K(Al+H3L=AlH-1L+4H)=-6.6
K(2AlH-1L=Al2H-2L2)=-14.8

K(UO2+Al+2H4L=AlUO2H-2L2+8H)=-8.21

Al+++ gl NaCl 37°C 0.15M C K1=8.246 B2=13.068 1989VBa (46017) 476
B(AlHL)=10.502
B(AlH-2L)=6.777
B(AlH-2L2)-0.209
B(Al2H-2L2)=12.694

Al+++ gl NaCl 25°C 0.60M C 19880Hb (46018) 477
B(-4,1,1)=-8.48
B(-5,1,1)=-14.71
B(-16,3,3)=-47.11

$$B(p, q, r) = pH + qAl + rH_3L = HqAlq(H_3L)r$$

Al+++	gl	NaCl	37°C	0.15M	U	K1=8.34	B2=13.69	1987FDb (46019)	478
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B(AlHL)=10.79
 B(AlH-1L)=5.29
 B(AlH-2L)=-0.53
 B(AlH-1L2)=8.21

Al+++	gl	NaCl	37°C	0.15M	U	K1=8.25	B2=13.07	1987VBe (46020)	479
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B(AlHL)=10.50
 B(AlH-2L2)=-0.21
 B(Al2H-2L2)=12.69
 B(Al3H-4L3)=15.08

$$B(AlH-1L2)=6.78$$

Al+++	gl	KCl	25°C	0.10M	C	K1=8.10	B2=12.90	1986Gpc (46021)	480
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B(AlHL)=10.81
 K(AlL2=AlH-1L2+H)=-6.10
 K(Al+L+HL)=11.14
 B(AlHL2)=16.84

$$K(AlH-1L2=AlH-2L2+H)=-7.17$$

Al+++		NaCl	25°C	0.15M	U	K1=8.0	B2=13.00	1986MAa (46022)	481
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K(Al+HL)=4.7
 K(AlHL=AlL+H)=-2.5
 K(AlL=AlH-1L+H)=-3.4

25-37 C. From a survey of literature data

Al+++	kin	NaNO ₃	205°C	var	U	K1=10.72		1984LKa (46023)	482
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K(Al+HL)=6.56
 K(Al+H2L)=2.91

Al+++	gl	KNO ₃	25°C	0.10M	C	K1=7.98		1984MMb (46024)	483
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K(AlL+H)=2.94
 K(AlL=AlH-1L+H)=-3.31

Al+++	gl	NaCl	25°C	0.60M	C			19830Sd (46025)	484
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B(-2,1,1)=-2.68
 B(-3,1,1)=-4.925
 B(-6,1,2)=-12.53
 B(-13,3,3)=-21.77

$$B(p, q, r): pH + qAl + rH_3L = AlqHp(H_3L)r$$

Al+++	gl	NaCl	25°C	0.60M	C			19830Se (46026)	485
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B(-2,1,1)=-2.68
 B(-3,1,1)=-4.925
 B(-6,1,2)=-12.53
 B(-13,3,3)=-21.77

$$B(p, q, r): pH+qAl+r(H_3L)=HpAlq(H_3L)r$$

Al+++	g1	NaCl	37°C	0.15M	C	K1=7.87 B(AlHL)=10.12 B(AlH-1L)=4.64 B(AlH-1L2)=8.8	1982JAc (46027) 486
Al+++	g1	NaCl	25°C	0.12M	C	K(Al+H-1L)=18.0 K(AlL(OH)2+2H)=18.4	1981RMb (46028) 487
Al+++	g1	NaClO4	25°C	0.10M	U	M	K1=8.65 B(AlL(Cys))=14.90 K(Al+L+HPo4)=19.29
Al+++	g1	NaClO4	33°C	0.25M	U		1961PPa (46030) 489 K(Al+H3L=AlL+3H)=-4.7 K(AlH-1L+H)=3.5 K(Al(OH)H-1L+H=AlH-1L)=6.8

C6H9NO6 H3L CAS 41035-84-1 (4367)
N-Carboxymethyl-L-aspartic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=2.24 B(AlHL)=6.15 B(AlH-1L)=-1.91	1997KSa (46374)	490

C6H9N06 H3L NTA CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH₂.COOH)₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.60M	C			K1=11.097 B(AlH-1L)=5.67 B(AlH-2L)=-2.52 B(Al2H-2L2)=13.16	1990OHb (46686)	491

Al+++ gl NaClO₄ 25°C 0.50M C K1=10.80 1984Na (46687) 492

Al+++ gl NaClO₄ 25°C 0.10M U T K1=10.53 B2=19.08 1981DSa (46688) 493
 At 35 C: K1=10.30, B2=18.57; 45 C: 10.18, 18.32

Al+++ gl KNO₃ 35°C 0.10M U K1=11.61 1980Khb (46689) 494

Al+++ sp NaClO₄ 25°C 0.20M U K1=11.37 1967BD**b** (46691) 496
By glass electrode: K(AlL+H)=1.90, K(AlLOH+H)=5.09, K(AlL(OH)₂+H)=8.28

Al+++	dis	NaClO4	20°C	0.10M	U	T	K1=9.5	1963STc (46692) 497
Al+++	gl	KCl	20°C	0.10M	U	K1=>10	1948SBa (46693) 498	
K(AlLOH+H)=5.8 K(AlL(OH)2+H)=8.6								

C6H9N3O2		HL	Histidine	CAS	71-00-1 (1)			
2-Amino-3-(4'-imidazolyl)propanoic acid; H2N.CH(CH2.C3H3N2)COOH								
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg	K values	Reference ExptNo
Al+++	gl	NaCl	37°C	0.15M	C			2002DCa (47529) 499
B(Al2H-2L)=1.163								
Al+++	gl	NaClO4	25°C	0.50M	U			1984CDa (47530) 500
K(Al(OH)+HL)=3.62 K(Al(OH)+L)=8.45								

C6H1004S2		H2L		CAS	7244-02-2 (438)			
1,2-Bis(carboxymethylthio)ethane; HOOC.CH2.S.CH2.CH2.S.CH2.COOH								
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg	K values	Reference ExptNo
Al+++	gl	NaClO4	25°C	0.50M	C	K1=2.05		1985NAb (48234) 501
B(AlHL)=12.46								

C6H1008		H2L	Saccharic acid	CAS	87-73-0 (1191)			
D-2,3,4,5-Tetrahydroxy-1,6-hexanedioic acid, Glucaric acid; HOOC.(CHOH)4.COOH								
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg	K values	Reference ExptNo
Al+++	gl	KNO3	25°C	0.10M	C			1984MMb (48467) 502
B(AlH-1L)=1.57 *K(AlH-1L)=-3.76								

C6H11N05		H2L	HIMDA	CAS	93-62-9 (192)			
N-(2-Hydroxyethyl)iminodieethanoic acid; HO.CH2.CH2.N(CH2.COOH)2								
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg	K values	Reference ExptNo
Al+++	gl	NaClO4	25°C	0.50M	C	K1=7.49		1984NAa (48687) 503
Al+++	gl	KNO3	35°C	0.10M	U	K1=9.33		1980KHb (48688) 504

C6H11N05		H2L			(1233)			
N-Hydroxyimino-2,2'-dipropionic acid; HO.N(CH(CH3)COOH)2								
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg	K values	Reference ExptNo

Al+++ gl KN03 25°C 0.10M C K1=5.9 1987AKa (48839) 505

C6H12N204 H2L CAS 4726-83-4 (5911)
N,N-Dihydroxyhexanediamide; HN(OH).CO.(CH₂)₄.CO.NH(OH)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaNO₃ 25°C 0.10M C K1=14.20 1989EHa (49332) 506

C6H12O6 HL CAS 498-43-1 (5803)
3-Deoxy-D-ribohexanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 25°C 0.10M C K1=1.97 1994BHa (49529) 507
*K(ALL)=-2.83
K(A1H-1L=A1H-3L+2H)=-9.17

C6H12O7 HL Gluconic acid CAS 526-95-4 (904)
D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH₂(CHOH)₄.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaNO₃ 25°C 0.10M C 1995E0a (49695) 508
B(A1H-1L)=-0.84
B(A1H-3L)=-10.70

Al+++ gl NaCl 25°C 0.10M C K1=2.01 1994BHa (49696) 509
*K(ALL)=-2.89
K(A1H-1L=A1H-3L+2H)=-9.30

Al+++ gl KN03 25°C 0.10M C K1=1.98 1984MMb (49697) 510
*K(ALL)=-2.87
K(A1H-1L=A1H-3L+2H)=-9.29

C6H13N02 HL Leucine CAS 61-90-5 (47)
2-Amino-4-methylpentanoic acid; H2N.CH(CH₂.CH(CH₃)₂).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ oth NaClO₄ 35°C 0.10M C K1=7.92 B2=14.01 1986SGd (50058) 511
B3=18.90

Method: electrophoresis

C6H13N02 HL CAS 4312-93-0 (4386)
Hexanohydroxamic acid; CH₃.CH₂.CH₂.CH₂.CH₂.CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.20M C K1=8.32 B2=16.17 2000FEC (50226) 512

$$B(AlH-1L2)=11.26$$

C6H13N04 HL Bicine CAS 150-25-4 (2124)
 N,N-Bis(2-hydroxyethyl)glycine; (HO.CH2.CH2)2N.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO ₃	25°C	0.10M	C				1984MMb (50340)	513
								$K(Al+HL)=3.38$		
								$K(AlHL=AlH-2L+3H)=-13.72$		

C6H13O9P H2L CAS 59-56-3 (3049)
 alpha-D-Glucose-1-phosphoric acid; Glucopyranose-1-phosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=5.60 B2=10.41	2001CRA (50620)	514
								$B(AlH2L)=ca.9.40$		
								$K(AlHL)=8.50$		
								$B(AlHL2)=14.04$		
								$B(Al2H-1L2)=11.08$		

$B(Al2H-2L2)=5.73$, $B(Al2H-3L2)=0.84$, $B(Al2H-4L2)=-5.09$, $B(AlH-3L)=-11.80$.

Additional methods: ¹³C, ²⁷Al and ³¹P nmr.

C6H14N402 HL Arginine CAS 74-79-3 (40)
 2-Amino-5-guanidopentanoic acid; H2N.CH((CH₂)₃.NH.C(:NH)(NH₂)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	oth/un	25°C	?	U T			K1=6.63 B2=12.86	1960PED (51001)	515
7 C:				K1=6.67, K2=6.38						

C6H15N302 HL CAS 52760-35-7 (6670)
 Lysine hydroxamic acid; H2N.(CH₂)₄.CH(NH₂)CO.NHOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			B2=21.28	2002ECa (51424)	516
								$B(AlH2L)=24.21$		
								$B(AlH2L2)=36.94$		
								$B(AlHL2)=29.84$		

C6H15N303 L (6613)
 1,3,5-Triamino-1,3,5-trideoxy-cis-inositol,5-Amino-5-deoxy-streptamine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C			K1=11.8 B2=18.8	1993HGa (51447)	517
								$B(AlHL2)=25.3$		
								* $K(AlL2)=-8.1$		

$*K(AlH-1L2)=-8.9$
 $*K(AlH-2L2)=-9.1$
 $*K(AlH-3L2)=-9.7$

C6H15O15P3 H6L Ins(1,2,6)P3 CAS 28841-62-5 (6479)
D-myoinositol 1,2,6-trisphosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	R4N.X	25°C	0.10M	C			K1=13.72 B(AlHL)=19.18 B(Al2L)=19.72 B(AlH-1L)=6.10	1995MBb (51533)	518

Medium: 0.10 M (n-Bu)4NBr.

C6H15O15P3 H6L CAS 88269-39-0 (8168)
D-myoinositol-1,4,5-triphosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	R4N.X	25°C	0.10M	C			K1=13.37 B(AlHL)=18.98 B(AlH-1L)=5.84 B(AlH-2L)=-1.82	1995MBb (51543)	519

Medium: 0.10 M (n-Bu)4NBr.

C6H16O6P2 H4L CAS 4721-22-6 (3708)
Hexane-1,6-diphosphonic acid; H2O3P(CH2)6PO3H2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	U			K1=14.66	1967KLa (51791)	520

C7H4N2O7 H2L CAS 609-99-4 (400)
3,5-Dinitrosalicylic acid; (O2N)2.C6H2(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KN03	30°C	0.10M	C			K1=7.53 K2=13.24 K3=4.24	1996MMa (52461)	521
Al+++	gl	NaClO4	25°C	0.10M	U			K1=8.81 K2=15.39 K3=4.34	1979LTc (52462)	522

C7H5N04 H2L Quinolinic acid CAS 89-00-9 (567)
2,3-Pyridinedicarboxylic acid; C5H3N.(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	nmr	NaCl	25°C	1.33M	U			K1=3.72 B2= 7.10	2001LKc (52620)	523

Method: 27Al NMR spectroscopy.

Al+++ gl NaClO4 25°C 0.50M C K1=4.50 B2=7.00 1986Mnb (52621) 524

C7H5N04 H2L CAS 499-80-9 (566)
2,4-Pyridinedicarboxylic acid; C5H3N.(COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaClO4 25°C 0.50M C K1=4.35 B2=7.30 1986Mnb (52649) 525

C7H5N04 H2L CAS 100-26-5 (2528)
2,5-Pyridinedicarboxylic acid, Isocinchomeric acid; C5H3N.(COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaClO4 25°C 0.50M U K1=3.95 B2=7.24 1970NAb (52664) 526

C7H5N04 H2L Dipicolinic aci CAS 449-83-2 (418)
2,6-Pyridinedicarboxylic acid; C5H3N.(COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp NaCl 25°C 1.00M U K1=11.37 1993NRa (52751) 527
K(Al+HL)=7.44

Al+++ gl NaClO4 25°C 0.50M U K1=4.87 B2=8.32 1968NAc (52752) 528
By spectrophotometry: K1=4.85

C7H5N05 H2L Nitrosalicylic CAS 85-38-1 (1416)
2-Hydroxy-3-nitrobenzoic acid; HO.C6H3(NO2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KN03 30°C 0.10M C K1=10.22 B2=18.20 1996MMa (52972) 529
K3=6.87
K(AlL3+H)=6.56

C7H5N05 H2L Nitrosalicylic CAS 96-97-9 (148)
2-Hydroxy-5-nitrobenzoic acid; HO.C6H3(NO2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.10M C K1=10.65 B2=19.81 2004GAa (53035) 530
B3=25.74

Al+++ gl KCl 30°C 0.10M C K1=10.91 B2=19.67 1996MMa (53036) 531
K3=5.36
K(AlL3+H)=6.24

Al+++ gl NaClO₄ 25°C 0.10M U K₁=11.11 B₂=19.73 1979LTc (53037) 532
K₃=6.13

Al+++ sp oth/un 25°C 0.10M C 1979PTb (53038) 533
K(Al+HL=AlL+H)=1.11

Al+++ sp NaClO₄ 29°C 1.00M U M 1976DDa (53039) 534
K(Co(en)₂(NH₃)L+Al)=-0.45

C7H503Br HL CAS 85-55-4 (1194)
5-Bromosalicylic acid; Br.C₆H₃(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp NaClO₄ 29°C 1.00M U M 1976DDa (53307) 535
K(Co(en)₂(NH₃)L+Al)=-0.72

C7H602 HL Tropolone CAS 533-75-5 (3129)
2-Hydroxycyclohepta-2,4,6-trien-1-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 25°C 0.60M C 19900Ha (53665) 536
K(Al+3HL=AlL₃(s)+3H)=11.21
K_{so}=-31.05

C7H602 HL Benzoic Acid CAS 65-85-0 (462)
Benzene carboxylic acid; C₆H₅.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 25°C 0.60M C 19910Ha (53821) 537
B(-1,1,1)=-2.67
B(-3,2,1)=-7.446

B(p,q,r); pH+qAl+rHL=HpAlq(HL)r

Al+++ gl NaClO₄ 25°C 0.50M U 1970NLa (53822) 538
B(AlL(OH))=12.1

C7H603 H2L Salicylic acid CAS 69-72-7 (14)
2-Hydroxybenzoic acid, Salicylic acid; HO.C₆H₄.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp oth/un 23°C 0.10M U 1994KGa (54139) 539
Keff=4.8

Method: spectrophotometric using pyrocatechol violet. Tris buffer adjusted to a pH=5.34 with HCl

Al+++ gl NaCl 25°C 0.60M C T 19830Sc (54140) 540

$B(-2,1,1)=-3.052$
 $B(-4,1,2)=-8.39$
 $B(-5,1,2)=-15.99$
 $B(-6,1,2)=-25.31$

$B(p,q,r); pH+qAl+r(H_2L)=HpAlq(H_2L)r$

Al+++ gl NaCl 25°C 0.12M C T K1=13.7 B2=26.70 1981RMB (54141) 541
 K3=10.73
 $K(AlH-2L+2H)=10.40$
 $K(AlH-3L+H)=9.37$

Al+++ sp NaClO4 29°C 1.00M U M 1976DDa (54142) 542
 $K(Co(en)2(NH3)L+Al)=-0.46$

Al+++ kin NaClO4 25°C 0.10M U 1975SVa (54143) 543
 $K(Al+HL=AlL+H)=0.067$

Al+++ gl KN03 20°C 0.10M U K1=12.9 B2=23.20 1969HBb (54144) 544
 K3=6.6

Al+++ EMF NaNO3 18°C 1.0M U 1961C0b (54145) 545
 $K(Al+3HL=AlHL2+H2L)=4.5$

Method: quinhydrone electrode

Al+++ sp NaClO4 27°C 0.20M U I 1959DAa (54146) 546
 $K(Al+HL=AlL+H)=-0.18$

$K=0.26(I=0), 0.06(I=0.02), 0.01(I=0.05), -0.13(I=0.10)$. Recalculated values

Al+++ sp oth/un 27°C ->0 U K1=14.11 1959DAa (54147) 547

C7H6O3 H2L CAS 99-06-9 (1370)
 3-Hydroxybenzoic acid; HO.C6H4.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 25°C 0.60M C 19910Ha (54375) 548
 $B(-1,1,1)=-2.59$
 $B(-3,2,1)=-7.453$

$B(p,q,r); pH+qAl+rHL=HpAlq(HL)r$

C7H6O4 H3L CAS 303-38-8 (1398)
 2,3-Dihydroxybenzoic acid; C6H3(OH)2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.10M C K1=10.62 B2=19.20 2004GAa (54461) 549
 $B(Al2H-3L2)=10.30$
 $B(AlH-1L2)=13.50$

Al+++ gl KCl 25°C 0.20M C T K1=10.32 B2=18.26 1993KAA (54462) 550

$B(AlH-1L2)=11.56$
 $B(AlH-2L2)=1.74$
 $B(Al2H-2L2)=13.62$
 $B(Al2H-3L2)=8.87$

C7H6O4 H3L Resorcylic acid CAS 89-86-1 (876)
 2,4-Dihydroxybenzoic acid, *b*-Resorcylic acid; C6H3(OH)2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C		T	K1=8.71 B2=15.03 B(Al2H-2L2)=9.1 B(AlH-1L2)=7.21	1993KAA (54514)	551

C7H6O4 H3L CAS 409-79-9 (1115)
 2,5-Dihydroxybenzoic acid; C6H3(OH)2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C		T	K1=9.74 B2=17.17 B(Al2H-2L2)=11.5 B(AlH-1L2)=9.97	1993KAA (54580)	552

Al+++	gl	NaClO4	25°C	0.50M	C		T		1985CDa (54581)	553
								K(Al+HL)=10.4 K(Al+2HL)=18.15		

Al+++	sp	NaClO4	20°C	0.09M	U	TIH			1971SGa (54582)	554
								K(Al+H2L=AlHL+H)=-3.96		
								K(Al+H2L=AlHL+H)(I=0.01)=-3.49, (I=0.20)=-4.47. DH=22.7 kJ mol-1, DS=173		

C7H6O4 H3L *g*-Resorcylic ac CAS 303-07-1 (1624)
 2,6-Dihydroxybenzoic acid; C6H3(OH)2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C		T	K1=12.79 B2=23.67 B(Al2H-2L2)=17.2 B(AlH-1L2)=16.46	1993KAA (54603)	555

C7H6O4 H3L Protocatechuic CAS 99-50-3 (875)
 3,4-Dihydroxybenzoic acid; C6H3(OH)2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=16.47 B2=29.38 B3=38.35 B(AlHL)=21.09 B(Al2H-2L2)=25.03	1993KAA (54657)	556

Al+++	sp	none	25°C	0	U	K1=15.33	1990SMd (54658) 557
Al+++	gl	NaClO ₄	25°C	0.50M	C	K1=16.1 K(Al+HL)=7.8	1985CDa (54659) 558
Al+++	gl	KCl	25°C	0.10M	C	K1=16.87 K3=8.76 *K(AlL)=-5.77 *K(AlL ₂)=-8.39 K(Al+H ₂ L=AlH ₂ L)=2.85	1985KPa (54660) 559

K(AlL+H)=4.66

Al+++	gl	KNO ₃	25°C	0.20M	U	K1=15.03 K3=9.91	1982HOb (54661) 560
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C7H6O5 H4L CAS 610-02-6 (3725)
2,3,4-Trihydroxybenzoic acid; (HO)₃.C₆H₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C		K1=9.10 B(AlH-1L2)=10.26 B(AlH-2L2)=2.26 B(AlH-2L3)=6.79 B(Al ₂ H-2L2)=11.09	1993KAa (54719)	561

B(Al₂H-3L2)=6.49, B(Al₂H-4L2)=-0.69. Ligand as H₂L

C7H6O5 H4L Gallic acid CAS 149-91-7 (446)
3,4,5-Trihydroxybenzoic acid; C₆H₂(OH)₃.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.60M	C			19820Sa (54748)	562
							B(-5,3,1)=-12.52 B(-9,4,3)=-20.25		
							B(p,q,r); pH+qAl+r(H ₃ L)=HpAlq(H ₃ L)r. K(Al+H ₃ L+4H ₂ O=AlL.H ₂ O(s)+3H)=-6.2		
Al+++	gl	NaCl	25°C	0.60M	C			19810Sa (54749)	563
							B(-2,1,1)=-4.933 B(-3,1,1)=-9.43 B(-6,1,2)=-21.98 B(-9,1,3)=-37.69		

B(p,q,r): pH + qAl + rH₃L = HpAlqH₃Lr. B(-8,2,3)=-22.65, B(-9,2,3)=-27.81,
B(-10,2,3)=-32.87, B(-11,2,3)=-39.56.

C7H6O6S H3L CAS 5965-83-3 (399)
5-Sulfosalicylic acid, 2-Hydroxy-5-sulfobenzoic; H₃O₃S.C₆H₃(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Al+++ gl NaClO4 25°C 0.50M C K1=11.8 B2=21.2 1985CDa (54930) 564

Al+++ sp NaClO4 25°C 0.10M U 1977PTa (54931) 565
K(Al+HL=AlL+H)=0.88

Al+++ sp NaClO4 29°C 1.00M U M 1976DDa (54932) 566
K(Co(en)2(NH3)L+Al)=-0.30

Al+++ gl KN03 20°C 0.10M U K1=12.3 B2=22.00 1969HBb (54933) 567
K3=5.8

Al+++ gl NaClO4 30°C 0.20M U K1=12.20 B2=22.21 1967AMa (54934) 568

Al+++ sp NaClO4 31°C 0.20M U TI 1963NAa (54935) 569
K(Al+HL=AlL+H)=0.23

K=0.79(I=0.02), 0.55(I=0.05), 0.37(I=0.10). Recalculated values

At I=0.1 M: K=0.22(15 C), 0.41(28 C), 0.53(40 C); DH=20.5 kJ mol⁻¹, DS=58

Al+++ sp NaClO4 25°C 0.10M U K1=12.91 B2=22.92 1960BSb (54936) 570
By glass electrode K1=13.20, K2=9.63, K4=6.06

Al+++ sp oth/un ? 0.20M U TI 1957NAa (54937) 571
K(Al+HL=AlL+H)=0.19

K=0.36(I=0.02), 0.31(I=0.05). Recalculated values

C7H609S2 H3L CAS 56507-30-3 (2659)

3,5-Disulfosalicylic acid; (HO3S)2.C6H2(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaClO4 25°C 0.50M C T K1=11.51 B2=20.19 1978LAa (55091) 572

C7H7N02 HL CAS 495-18-1 (184)

Benzohydroxamic acid; C6H5.CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.20M C K1=7.57 B2=14.60 2000FEc (55494) 573

C7H7NS L Thiobenzamide CAS 2227-79-4 (1660)

Thiobenzamide; C6H5.CS.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp non-aq 25°C 100% U 1977SWa (55702) 574

K(AlCl3+L)=2.20

Medium: Et2O

C7H8N2O2 L CAS 99-52-5 (470)

2-Methyl-4-nitro-aminobenzene; CH3.C6H3(NO2).NH2

C7H9N02 H2L DHB (8381)
 3,4-Dihydroxybenzylamine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C				2002A0a (56411)	581
B(-2,1,1)=-5.58 B(-4,1,2)=-13.74 B(-6,1,3)=-25.51 B(-7,1,3)=-35.55										
B(-8,1,3)=-46.39, B(-9,1,3)=-56.79, B(-6,2,2)=-19.58 B(p,q,r) defined for the protonated ligand, H3L+: pH+qAl+rH3L=HpAlq(H3L)r										

C7H9N02 HL CAS 30652-11-0 (2458)

3-Hydroxy-1,2-dimethylpyridin-4(1H)-one; (OH)(CH3)(O:)C5H2N.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO3	25°C	0.10M	C			K1=12.20 B2=23.25	2004SGc (56423)	582
								B3=32.62		
Al+++	gl	KCl	25°C	0.10M	C			K1=12.20 B2=23.25	1994MRa (56424)	583
								K3=9.37		
Al+++	gl	KCl	25°C	0.10M	C			K1=12.20 B2=23.25	1992CMb (56425)	584
								K3=9.37		
Al+++	gl	NaCl	25°C	0.15M	C I			K1=11.91 B2=22.83	1989CNa (56426)	585
								B3=32.35		

Data also at I=0.6 M(NaCl): K1=11.57, B2=22.01, B3=30.90

C7H9N03 HL CAS 157070-43-4 (7154)

3-Hydroxy-5-methyl-2-(N-methylformamido)furan;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.60M	C			K1=7.28 B2=13.42	1994LSc (56445)	586
K3=4.83 B(-1,1,1)=0.19 B(-2,2,1)=-0.76 B(-3,3,1)=-3.02										
B(-7,3,3)=-10.43, B(-4,2,1)=-12.08. B(p,q,r): pH+qHL+rM=Hp(HL)qMr.										

C7H1007P2 H4L CAS 2809-26-9 (8731)

1-Phenyl-1-hydroxymethylene-1,1-diphosphonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO3	25°C	0.10M	C			K1=18.50	2002GKc (56762)	587

B(AlH-1L)=13.75

$$B(AlH_2L) = 3.57$$

$$B(AlH_3L_2) = 43.41$$

C7H11NO6P2 H4L CAS 4712-06-5 (4470)

Amino(phenyl)methylenediphosphonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	U			K1=18.97 K(Al+HL)=12.50 B(Al2L)=23.09 K(Al+HL+L)=19.68	1969DMd (56938)	588

C7H11O9P H5L (5041)
2-Phosphonobutane-1,2,4-tricarboxylic acid; HOOCCH₂CH₂C(P(=O)(H)₂)(COOH).CH₂COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	g1	NaNO ₃	25°C	0.50M	C				1999SEa (57023)	589
								B(AlH-6L2)=18.42		
								B(AlH-6L1)=16.06		

C7H12O2 HL CAS 98-89-5 (2793)
Cyclohexanecarboxylic acid, Hexahydrobenzoic acid; C6H11.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	g1	NaCl	25°C	0.60M	C				19910Ha (57227)	590
							B(-1,1,1)=-3.48			
							B(-3,2,1)=-8.04			

$$B(p,q,r) : pH + qAl + rH_2O \rightarrow H_pAl_q(OH_r)r$$

C8H6O4 H2L Phthalic acid CAS 88-99-3 (113)
Benzene-1,2-dicarboxylic acid; C6H4(COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 25°C 0.60M C K1=2.94 B2=4.97 1988HBa (58943) 591
 $B(-2,2,1)=-2.50$
 $B(-4,3,1)=-8.47$
 $B(-2,2,2)=-0.07$

B(p,q,r): pH+qAl+rL=HpAlqLr

Al+++ gl NaClO4 25°C 0.50M U K1=3.18 B2=6.32 1970NL_a (58944) 592

C8H606 H4L (6671)

2,3-Dihydroxybenzene-1,4-dicarboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KC1 25°C 0.20M C K1=3.90 1993KAa (59077) 593
 B(AlH-1L)=0.96
 B(AlH-2L2)=-0.36
 B(AlH-3L2)=-8.11
 B(Al2H-3L2)=1.36
 B(Al3H-5L3)=2.27, B(Al3H-6L3)=-2.11. Ligand as H2L ?

C8H8N02Cl HL CAS 2153-11-9 (4570)
 N-Chloroacetyl-N-phenylhydroxylamine; Cl.CH2.CO.N(C6H5).OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	mixed	30°C	50%	U			K1=6.94 B2=14.52 K3 = 6.62	1971GSc (59284)	594

Medium: 50% acetone/H2O, 0.5 M NaClO4

C8H8N202 H2L (3821)
 1-(2'-Hydroxyphenyl)-4-oxo-2,3-diazabut-1-ene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	alc/w	19°C	28%	U				1963H0c (59324)	595

Medium: 28% EtOH, 0.025 M, acetate buffer

C8H8O2 HL 2-Acetylphenol CAS 118-93-4 (1888)
 2-Hydroxyacetophenone; HO.C6H4.CO.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.60M	C			K1=7.34 B(AlH-1L)=1.97	1999MTa (59456)	596

By spectrophotometry: K1=7.30

C8H8O3 H2L CAS 614-75-5 (4475)
 2-Hydroxyphenylethanoic acid; HO.C6H4.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.60M	C			K1=8.06 B(AlH-1L)=3.11	1999MTa (59714)	597

Confirmed by H-nmr

C8H8O3 HL Mandelic Acid CAS 611-72-3 (80)
 2-Phenyl-2-hydroxyethanoic acid; C6H5.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	EMF oth/un	?	?	U				K1=13.91 B2=26.90	19680Sb (59810)	598

K3=11.98

C8H8O4 HL CAS 520-45-6 (4478)
3-Acetyl-2-hydroxy-6-methylpyran-4-one, Dehydroethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl diox/w 35°C 50% U K1=5.39 B2=10.32 1971MAa (60082) 599
Medium: 50% dioxan, 0.1 M NaClO4

C8H8O5 H2L CAS 5629-08-3 (679)
7-Oxy-bicyclo[2.2.1]-hept-5-ene-2,3-dicarboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaCl 37°C 0.15M C 1988HYa (60124) 600
B(AlHL)=9.279
B(Al+L=Al(OH)L+H)=1.565
B(AlHL2)=13.585

C8H9NOS HL CAS 4822-44-0 (3240)
N-(Mercaptoacetyl)aniline (thioglycolanilide); C6H5.NH.CO.CH2.SH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl diox/w 30°C 50% U K1=6.94 B2=13.00 1973ABB (60158) 601
Medium: 0.1 M NaClO4

C8H9N02 L CAS 1849-49-6 (5907)
5'-Deoxypyridoxal

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 1.00M C K1=6.15 B2=12.19 1989MSb (60246) 602

C8H9N02 HL CAS 4746-61-6 (4512)
Glycolanilide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl diox/w 30°C 50% U K1=8.48 B2=16.17 1973ABB (60250) 603
Medium: 50% dioxan, 0.1 M NaClO4

C8H9N02 HL (2591)
N-Phenyl-N-acetohydroxamic acid; CH3.CO.N(OH)C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.20M C K1=7.84 B2=15.10 2000FEC (60281) 604
B(AlH-1L2)=10.77

Al+++ gl mixed 30°C 50% U K1=9.20 B2=17.30 1971GSc (60282) 605
K3=7.53

Medium: 50% acetone/H2O, 0.5 M NaClO4

C8H9N02S HL CAS 6310-11-8 (4576)

3-Mercaptoacetamidophenol; HS.CH2.CO.NH.C6H4.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl oth/un 17°C ? U K1=6.72 B2=13.01 1973KPd (60382) 606

C8H9N03 HL Pyridoxal CAS 65-22-5 (110)
3-Hydroxy-5-(hydroxymethyl)-2-methyl-4-pyridinecarboxaldehyde;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KN03 25°C 0.10M C K1=2.45 1981TMe (60424) 607

C8H9N04 H2L (4520)
Dehydroethanoic acid oxime;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl diox/w 35°C 50% U 1971MAa (60488) 608
K(Al+HL)=5.11
K(Al+2HL)=9.57

Medium: 50% dioxan, 0.01 M NaClO4

C8H10N06P H3L Codecarboxylase CAS 41468-25-1 (2555)
Pyridoxal-5-phosphoric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 1.00M C M K1=10.48 1985SMd (60701) 609
K(All+H)=4.23

ternary complexes with 2-amino-3-phosphonopropionic acid

C8H1009 H4L CAS 137172-86-2 (6612)
SS-Oxydisuccinic acid; O(CH(COOH)CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.10M C K1=8.43 1992MMa (60902) 610
K(All+H)=3.42
*K(All)=-5.31
*K(AlH-1L)=-6.43
K(Al+HL)=5.87

C8H1009 H4L CAS 84852-72-2 (6611)

meso-Oxydisuccinic acid; O(CH(COOH)CH₂.COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C			K1=9.17 K(AlL+H)=3.37 *K(AlL)=-5.30 *K(AlH-1L)=-6.96 K(Al+HL)=6.57	1992MMa (60914)	611

C8H10O10 H4L (5894)
1-Hydroxy-3-oxapentane-1,2,4,5-tetracarboxylic acid;
HO.CH(COOH).CH(COOH).O.CH(COOH).CH₂(COOH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C			K1=7.63 K(AlL+H)=2.98 K(AlH-1L+H)=5.05	1989MMd (60926)	612

C8H11N02 H2L Dopamine CAS 579-59-9 (251)
2-(3',4'-Dihydroxyphenyl)ethylamine; (HO)₂.C₆H₃.CH₂.CH₂.NH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K(Al+HL)=15.63 K(AlHL+HL)=12.98 K(AlH ₂ L ₂ +HL)=8.95	1989KSd (61075)	613

At pH 7: K_{1eff}=8.01, K_{2eff}=5.36, K_{3eff}=1.33

C8H11N02 HL CAS 30652-12-1 (5889)
3-Hydroxy-2-methyl-1-ethylpyridin-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.15M	C			K1=11.75 B2=22.52 1989CNa (61090) 614 B3=32.17		

C8H11N03 H2L Noradrenaline CAS 138-65-8 (253)
Norepinephrine, 3,4-Dihydroxyphenylethanolamine; (HO)₂C₆H₃.CH(CH₂.NH₂).OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K(Al+HL)=15.60 K(AlHL+HL)=12.99 K(AlH ₂ L ₂ +HL)=9.27 B(AlHL)=25.33	1989KSd (61160)	615

B(AlH₂L₂)=48.08, B(AlH₃L₃)=67.05. At pH 7: K_{1eff}=8.31, K_{2eff}=5.70, K_{3eff}=1.98

C8H12N207 H3L (9050)
Aspartyl-aspartic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=7.54 B(AlHL)=11.09 B(AlH2L)=14.42 B(AlH-1L)=3.01	2003KFa	(61471) 616

C8H1207P2 H4L (7244)
1-Hydroxy-2-phenylethane-1,1-diphosphonic acid; HO.C(PO(OH)2)2.CH2C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO3	25°C	0.10M	C			K1=18.74 B(AlH-1L)=13.26 B(AlH-2L)=2.57 B(AlH3L2)=44.74	2002GKc	(61738) 617

C8H15N304 HL Gly-Ala-Ala CAS 6491-25-4 (6783)
Glycyl-alanyl-alanine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO3	25°C	0.10M	C			K1=21.85	1983IMb	(62249) 618

C8H16N204 H2L CAS 38937-66-5 (5912)
N,N-Dihydroxyoctanediamide; HN(OH).CO.(CH2)6.CO.NH(OH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaNO3	25°C	0.10M	C			K1=14.59	1989EHa	(62538) 619

C8H19N05 L Bis-tris CAS 6976-37-0 (2827)
Bis-(2-hydroxyethyl)imino-tris(hydroxymethyl)methane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sol	NaCl	25°C	0.10M	U T				1990WPa	(63054) 620

K=2.6 at 50 C. K: Al(OH)4+H2L=Al(OH)2L+2H2O

C9H6NOCl HL CAS 130-16-5 (1268)
5-Chloro-8-hydroxyquinoline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	diox/w	25°C	60%	U				1973SCd	(63658) 621

B3=31.83

Medium: 60% dioxan, 0.1 M NaClO4

C9H6N04IS H2L Ferron CAS 547-91-1 (275)

7-Iodo-8-hydroxyquinoline-5-sulfonic acid; (HO)(HO3S)C9H4NI

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
Al+++	sp	NaCl	25°C	1.0M	C			K1=8.0	1988NMa (63776)	622	
Medium pH 1.8-3.5											
Al+++	sp	KCl	25°C	0.14M	U	I		K1=7.9 K3=7.2	1982GTa (63777)	623	
In Al(III)-ferron-CTMAC solution:								K1=8.5, K2 < 5.8, K3 > 10.8			
Al+++	sp	oth/un	?	dil	U			B2=12.5	1971BRf (63778)	624	
Al+++	gl	KNO3	28°C	0.10M	U			K1=6.76	B2=13.76	1971LSb (63779)	625
Al+++	gl	KCl	25°C	0.10M	U			K1=7.6 K3=5.6	B2=14.7	1961LSa (63780)	626
								K(Al(OH)L2+H)=5.0			

C9H6N203 HL CAS 5437-99-0 (3865)

5-Nitro-8-hydroxyquinoline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	diox/w	25°C	60%	U				1973SCd (63860)	627
Medium: 60% dioxan, 0.1 M NaClO4								B3=21.42		

C9H6O7 H4L CAS 609-98-3 (4591)

2-Hydroxybenzene-1,3,5-tricarboxylic acid; HO.C6H2(COOH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	EMF	NaCl	25°C	0.10M	U				1971PPe (64004)	628
								K(Al+H2L)=4.97		

C9H6O7 H4L CAS 54176-76-0 (4592)

5-Hydroxybenzene-1,2,4-tricarboxylic acid; HO.C6H2(COOH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	EMF	NaCl	25°C	0.10M	U				1971PPe (64008)	629
								K(Al+H2L)=4.40		

C9H7NO HL Oxine CAS 148-24-3 (504)

8-Hydroxyquinoline (8-quinolinol);

By fluorescence, $K(Al+HL=AlL+H)=3.590.$

C9H8N203S HL CAS 292149-08-7 (8799)

2-(4,5-Dihydro-2-thiazolyl)-4-nitrophenol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp alc/w 25°C 73% C 2000LTb (64810) 638
 $K(Al+HL=AlL+H)=4.65$

Medium: 73.2% v/v MeOH, 2.4% DMF, 24.4% H₂O, 0.10 M NaClO₄.

C9H8N203S HL CAS 292149-08-7 (8798)

2-(4,5-Dihydro-2-thiazolyl)-5-nitrophenol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp alc/w 25°C 73% C 2000LTb (64812) 639
 $K(Al+HL=AlL+H)=4.08$

Medium: 73.2% v/v MeOH, 2.4% DMF, 24.4% H₂O, 0.10 M NaClO₄.

C9H8N203S H2L CAS 148292-08-4 (7219)

Nordesferriferrithiocin; (HO)C₅NH₃.C₃NSH₃(COOH)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp KCl 25°C 0.10M C B2=22.0 1996LHa (64814) 640
 $K(AlL2+H)=6.8$
 $K(AlHL2+H)=3.7$

C9H8N204S H2L CAS 15851-62-4 (3886)

7-Amino-8-hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp NaNO₃ 25°C 0.10M C K1=8.52 1995SOa (64823) 641

C9H8O4 H3L Caffeic acid CAS 331-39-5 (6037)

3-(3,4-Dihydroxyphenyl)propenoic acid; (HO)₂C₆H₃.CH:CH.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.10M C 2002AOa (64917) 642
 $B(-2,1,1)=-4.88$
 $B(-3,1,1)=-9.45$
 $B(-4,1,1)=-15.53$
 $B(-6,1,2)=-22.24$

$B(-7,1,2)=-30.73$, $B(-9,1,3)=-39.23$.

$B(p,q,r): pH+qAl+rH3L=HpAlq(H3L)r$

Al+++ sp none 25°C 0 U K1=15.06 1990SMd (64918) 643

C9H9NOS HL CAS 101821-30-1 (8796)
2-(4,5-Dihydro-2-thiazolyl)phenol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	alc/w	25°C	73%	C				2000LTb (65028)	644

$$K(Al+HL=AlL+H)=3.75$$

Medium: 73.2% v/v MeOH, 2.4% DMF, 24.4% H₂O, 0.10 M NaClO₄.

By fluorescence, K(Al+HL=AlL+H)=3.50.

C9H9N3O2S2 HL Sulfathiazole CAS 72-14-0 (8357)
4-Amino-N-2-thiazolyl-benzenesulfonamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	alc/w	25°C	50%	C			K2=6.70	1999GAa (65132)	645

Medium: 50% EtOH/H₂O, 0.10 M NaNO₃.

C9H11N02 HL Phenylalanine CAS 63-91-2 (2)
2-Amino-3-phenylpropanoic acid; H₂N.CH(CH₂.C₆H₅).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO ₃	35°C	0.10M	U	M			1980KHb (65923)	646

$$B(AlL(tripolyphosphate))=7.55$$

C9H11N03 H₂L Tyrosine CAS 60-18-4 (4)
2-Amino-3-(4-hydroxyphenyl)propanoic acid; HO.C₆H₄.CH₂.CH(NH₂).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO ₃	35°C	0.10M	U	M			1980KHb (66210)	647

$$B(AlHL(tripolyphosphate))=7.95$$

C9H11N04 HL CAS 95215-59-1 (8724)
1-(2'-Carboxyethyl)-2-methyl-3-hydroxy-4-pyridinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO ₃	25°C	0.10M	C			K1=13.04 B2=24.06	2002SGb (66302)	648

C9H11N04 H₃L DOPA CAS 59-92-7 (5)
2-Amino-3-(3,4-dihydroxyphenyl)propanoic acid; H₂NCH(CH₂C₆H₃(OH)₂).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C				1989KSd (66392)	649

$$K(Al+HL)=16.03$$

$$K(AlHL+HL)=13.21$$

K(AlH₂L₂+HL)=9.12
At pH 7: K_{1eff}=8.08, K_{2eff}=5.36, K_{3eff}=1.17

Al+++ gl NaCl 25°C 0.12M U K1=19.60 1978RMc (66393) 650

C9H12N06P H3L PhosphoTyrosine CAS 41863-47-2 (5813)
Phosphotyrosine; 4-((OH)2P(O).O)C6H4.CH2.CH(NH2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++	gl	KCl	25°C	0.20M	C		K1=10.37 B(AlHL)=14.42 B(AlH-1L)=4.08		1998KLb (66552)	651
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C9H13N02 L (7151)
1,2-Diethyl-3-hydroxy-4-pyridinone

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++	gl	KCl	25°C	0.10M	C		K1=13.42 B2=25.06 K3=8.48	1994MRa (66793)	652
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C9H13N03 H2L (-)Adrenaline CAS 51-43-4 (252)
4-(1-Hydroxy-2-(methylamino)ethyl)-1,2-dihydroxybenzene,
Epinephrine;CH₃NHCH(OH)C₆H₃(OH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++	gl	KCl	25°C	0.20M	C		K(Al+HL)=15.57 K(AlHL+HL)=13.03 K(AlH ₂ L ₂ +HL)=9.25 B(AlHL)=25.81	1989KSd (66857)	653
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B(AlH₂L₂)=49.08, B(AlH₃L₃)=68.57. At pH 7:K_{1eff}=8.22, K_{2eff}=5.68, K_{3eff}=1.90

C9H13N209P H3L UMP-5 CAS 58-97-9 (2948)

Uridine-5'-monophosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++	gl	KCl	25°C	0.20M	C		K1=11.18 B2=17.49 B(AlHL)=14.83	1996AKa (66969)	654
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C9H14N308P H2L CMP-5 CAS 63-37-6 (1243)
Cytidine-5'-monophosphoric acid, Cytidilic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++	gl	KCl	25°C	0.20M	C		K1=6.08 B2=9.7 B(AlHL)=9.0	1996AKa (67248)	655
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B(AlH-1L)=0.39

C9H14O7P2 H2L CAS 445253-97-4 (8732)
[(Dimethoxyphosphinyl)hydroxyphenylmethyl]phosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	KNO ₃	25°C	0.10M	C		B2=12.21 B(AlH-3L)=-11.91 B(AlHL2)=16.30 B(AlH-1L2)=6.06 B(AlH-2L2)=-0.78	2002GKc (67375)	656

C9H18N2O4 H2L CAS 18992-11-5 (5913)
N,N-Dihydroxynonanediamide; HN(OH).CO.(CH₂)₇.CO.NH(OH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	NaNO ₃	25°C	0.10M	C		K1=15.55	1989EHa (67937)	657

C9H24N3O9P3 H6L NOTPH CAS 83843-39-3 (224)
1,4,7-Triazacyclononane-N,N',N"-tris(methylenephosphonic acid);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	KNO ₃	25°C	1.00M	U		K1=18.6 K(Al+HL)=12.6 K(Al+H2L)=10.3 K(Al+H3L)=7.6	1990BSd (68310)	658

C10H6O8 H4L Pyromellitic Ac CAS 89-05-4 (519)
Benzene-1,2,4,5-tetracarboxylic acid; C₆H₂.(COOH)₄

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.10M	C		K1=4.44 B(AlHL)=8.44 B(AlH-1L)=-0.25	1998NPa (68507)	659

C10H8O5S H2L CAS 16223-97-7 (2392)
1,2-Dihydroxynaphthalene-4-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.60M	C			1983oSb (69806)	660

B(-2,1,1)=-5.343
B(-3,1,1)=-11.24
B(-4,1,2)=-13.115
B(-5,1,2)=-21.15

B(p,q,r); pH+qAl+rH2L=HpAlqH2Lr. B(-6,1,3)=-24.47

C10H8O5S H3L DHNSA (877)
2,3-Dihydroxynaphthalene-6-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO4	25°C	0.50M	C		K1=15.7	B2=29.1	1985CDa (69834)	661
Al+++	gl	NaNO3	25°C	0.10M	U		K1=16.48	B2=29.82	1984NHa (69835)	662
							B3=39.12			

C10H8O5S2 H4L Chromotropic ac CAS 148-25-4 (1875)
1,8-Dihydroxynaphthalene-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaNO3	25°C	0.10M	U		K1=17.18	B2=30.10	1990HWa (69920)	664
Al+++	gl	NaClO4	25°C	0.50M	C		K1=12.9	B2=22.5	1985CDa (69921)	665
Al+++	kin	NaClO4	25°C	0.50M	C				1981BMg (69922)	666

Method: stopped-flow.

Al+++	gl	KNO3	20°C	0.10M	U		K1=17.1	B2=29.80	1969HBb (69923)	667
Al+++	gl	NaClO4	30°C	0.20M	U		K1=17.16	B2=30.41	1967AMa (69924)	668
Al+++	con	KNO3	?	0.10M	U		K1=17.22		1965DMA (69925)	669
By glass electrode:										
Al+++	gl	oth/un	25°C	0.01M	U				1957JAc (69926)	670

$$K(Al+H_2L=AlL+2H)=-3.87$$

C10H9NO HL 8-OH-Quinaldine CAS 826-81-3 (998)
2-Methyl-8-hydroxyquinoline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	alc/w	?	100%	U		K1=8?		19620Ba (70043)	671
Medium:	EtOH									
C10H9NO		HL						CAS 5541-67-3 (999)		
5-Methyl-8-hydroxyquinoline;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

Al+++ gl diox/w 25°C 50% U B2=20.52 1978THc (70063) 672
 B3=30.53
 B(AlHL)=15.08
 B(AlHL2)=25.24

C10H9N03S H2L CAS 49608-51-7 (8280)
 4,5-Dihydro-2-(2-hydroxyphenyl)-4-thiazolecarboxylic acid,
 Deazademethylferrithiocin;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	alc/w	25°C	75%	C				2000LTb (70167)	673
								K(Al+H2L=AlL+2H)=3.38		

Medium: 75% v/v MeOH/H2O, 0.10 M NaClO4.

By fluorescence, K(Al+H2L=AlL+2H)=3.26

Al+++ gl KN03 25°C 0.10M C K1=12.22 B2=22.82 1990ARa (70168) 674

C10H10N203S H2L CAS 76045-30-2 (7218)
 Desferriferrithiocin,
 2-(3-Hydroxypyridin-2-yl)-4-methyl-4,5-dihydrothiazole-4-carboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	KCl	25°C	0.10M	C			B2=23.6	1996LHa (70558)	675
								K(AlL2+H)=6.6		
								K(AlHL2+H)=3.3		

Al+++ gl KN03 25°C 0.10M C B2=22.2 1990ARa (70559) 676

C10H13N04 HL CAS 137528-47-3 (8725)
 1-(3'-Carboxypropyl)-2-methyl-3-hydroxy-4-pyridinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KN03	25°C	0.10M	C			K1=13.03 B2=22.97	2002SGb (71755)	677
								B3=31.27		
								B(AlHL2)=27.34		
								B(AlH2L2)=30.26		

C10H14N507P H2L AMP-5 CAS 18422-05-4 (842)
 Adenosine-5'-monophosphoric acid, 5-Adenylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=6.14	1996AKa (72439)	678
								B(AlH-1L)=1.90		

Al+++ gl KCl 25°C 0.20M U K1=6.17 B2=10.35 1991KSb (72440) 679
 B(AlH-1L)=2.02

C10H14N5O8P H3L GMP-5 CAS 85-32-5 (2947)
Guanosine-5'-monophosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=11.66 B(ALHL)=14.91	1996AKa (72585)	680

C10H15N2O8P H2L TMP-5 CAS 365-07-1 (2949)
Thymidine-5'-monophosphoric acid, Thymidylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=11.92 B(ALHL)=15.41 B(ALH-1L)=5.6	1996AKa (72698)	681

C10H15N5O10P2 H3L ADP CAS 20398-34-9 (2181)
Adenosine-5'-diphosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	U			K1=7.82 B(ALHL)=10.98 B(ALH-1L)=2.94 B(ALH-1L2)=5.01	1991KSb (72976)	682

Al+++ gl NaCl 25°C 0.15M U K1=10.03 1987JVa (72977) 683
B(ALH-1L)=4.18

C10H16N2O8 H4L EDDS CAS 52759-67-8 (1100)
1,2-Diaminoethane-N,N'-di-1,4-butanedioic acid; (CH₂.NH.CH(COOH)CH₂.COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	dis	KNO ₃	20°C	0.10M	U			K1=14.8	1968MJa (73111)	684

Method: paper electrophoresis

C10H16N2O8 H4L EDTA CAS 60-00-4 (120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	cal	none	25°C	0.0	C	H			19900Ba (73567)	685

Medium: pH 8.7. DH(K1)=48.96 kJ mol⁻¹.

Al+++	gl	KNO ₃	25°C	0.50M	C	M			1989TBa (73568)	686
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K(ALL+H)=2.41
*K(ALL)=-5.81

					K(AlL+F)=4.8 K(AlL+S)=10.3	
Al+++	gl	KNO ₃	25°C	0.50M C	M	1986TBa (73569) 687
					K(AlL+H)=2.41 *K(AlL)=-5.81 K(AlL+F)=4.8 K(Al(OH)L+HS=AlLS)=10.3	
Al+++	gl	NaCl	25°C	0.12M C		K1=15.3 1981RMb (73570) 688
					*K(AlL)=-5.73 *K(AlH-1L)=-9.82	
Al+++	gl	KNO ₃	35°C	0.10M U	K1=16.95	1980KHb (73571) 689
Al+++	vlt	oth/un	?	1.08M U	K1=15.9	1969SVd (73572) 690
Medium:	K ₂ SO ₄ ,	pH=2.7				
Al+++	ISE	KNO ₃	25°C	0.10M U	K1=16.5 K(Al+HL)=3.4 K(AlL+OH)=8.0	1967ABb (73573) 691
Al+++	sp	NaClO ₄	25°C	0.20M U	K1=16.01	1967BDb (73574) 692
By glass electrode:	K(AlL+H)=2.63,	K(AlLOH+H)=5.87,	K(AlL(OH) ₂ +H)=10.31			
Al+++	ISE	KNO ₃	20°C	0.10M U T H	T K1=16.7 K(AlL+H)=2.77	1966MCa (73575) 693
K1=16.84(30 C), 17.26(40 C).	DH(K1)=50.2 kJ mol-1,	DS=489 J K-1 mol-1				
Al+++	dis	NaClO ₄	20°C	0.10M U	B(AlL(OH))=25.04	1963STc (73576) 694
Medium:	KClO ₄					
Al+++	cal	KNO ₃	20°C	0.10M U H		1958SRa (73577) 695
DH(K1)=52.6 kJ mol-1,	DS=487 J K-1 mol-1					
Al+++	gl	KNO ₃	15°C	0.10M U	K1=16.11 K(AlL+H)=2.0 K(AlLOH+H)=5.16	1956STa (73578) 696
Al+++	vlt	KNO ₃	20°C	0.10M U	T K1=16.13 K(Al+HL)=8.4 K(AlLOH+H)=5.89 K(AlL(OH) ₂ +H)=9.97	1954SGa (73579) 697
C10H16N5O13P3	H4L	ATP			CAS 56-65-5 (403)	
Adenosine-5'-triphosphoric acid;						
Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values	Reference ExptNo

Al+++ nmr oth/un 25°C ? U 1994DBa (74697) 698
 K1eff=2.4
 K2eff=2.1

At pH 7.4

Al+++ gl KCl 25°C 0.20M U K1=7.92 B2=12.47 1991KSb (74698) 699
 B(AlHL)=11.30
 B(AlH-1L)=2.46
 B(AlH-1L2)=4.84

Al+++ gl NaCl 25°C 0.15M U 1987JVa (74699) 700
 B(AlHL)=12.47

C10H18N2O7 H3L HEDTA CAS 150-39-0 (392)
 N-(Hydroxyethyl)diaminoethane-N,N',N'-triethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO ₃	35°C	0.10M	U			K1=15.68	1980Kb (75329)	701
Al+++	ISE	KNO ₃	25°C	0.10M	U			K1=14.4 K(AlL+H)=2.4 K(AlL+OH)=9.3	1967ABb (75330)	702

Al+++ sp NaClO₄ 25°C 0.20M U K1=13.96 1967BD_b (75331) 703
 By glass electrode: K(AlL+H)=2.14, K(AlLOH+H)=4.89, K(AlL(OH)₂+H)=9.19

Al+++ ISE KNO₃ 20°C 0.10M U T H K1=12.43 1966MCa (75332) 704
 K(AlL+H)=5.08

K1=12.6(30 C), 12.9(40 C). DH(K1)=37.6 kJ mol⁻¹, DS=372 J K⁻¹ mol⁻¹

C10H18O L (6695)
 4-tert-Butylcyclohexanone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	nmr	non-aq	25°C	100%	U	T			1993HMb (75586)	705

Medium: C₆D₆. At 15 C: K=2.39; 37 C: K=1.81

C10H20N2O4 H2L CAS 5578-84-7 (5914)
 N,N-Dihydroxydecanediamide; HN(OH).CO.(CH₂)₈.CO.NH(OH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaNO ₃	25°C	0.10M	C			K1=15.29	1989EHa (75798)	706
C10H22N2O3		L	Cryptand 2,1		CAS 31249-95-3	(835)				

4,7,13-Trioxa-1,10-diazacyclopentadecane (Trioxa(2,1)cryptand);

K1 measured by spectrophotometry

C11H806S H3L CAS 6407-91-6 (1994)
1-Hydroxy-7-sulfo-2-naphthoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO ₄	25°C	0.50M	C			K1=13.88 B2=23.52 B(AlH-1L2)=15.22 B(AlH-2L2)=6.54	1988LK _a (77238)	714

K1 measured by spectrophotometry

C11H806S H3L CAS 15509-36-1 (2658)
3-Hydroxy-7-sulfo-2-naphthoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	NaClO ₄	25°C	0.10M	U	I		K1=11.934	1980LPf (77247)	715

In 0.1 M KCl: K1=11.316

C11H807S H4L CAS 6407-90-5 (2683)
1,7-Dihydroxy-4-sulfo-2-naphthoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO ₄	25°C	0.50M	C			K1=14.95 B2=22.14 B3=28.21 B(AlHL2)=30.80 B(AlHL)=20.00	1982LAa (77262)	716
Al+++	gl	NaClO ₄	25°C	0.50M	C			K1=14.92 B2=22.14 B3=28.21 B(AlHL)=20.00 B(AlHL2)=30.80	1982LKc (77263)	717

C11H807S H4L CAS 6470-93-5 (8345)
3,5-Dihydroxy-7-sulfo-2-naphthoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO ₄	25°C	0.50M	C			K1=16.00 B2=24.09 B3=30.85 B(AlHL)=20.65 B(AlHL2)=32.44	1982LAa (77268)	718

C11H809S2 H4L CAS 67097-84-1 (1995)
1-Hydroxy-4,7-disulfo-2-naphthoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO ₄	25°C	0.50M	C			K1=12.38 B2=21.55	1988LK _a (77274)	719

$$B(AlH-1L2)=13.66$$

K1 measured by spectrophotometry

C11H809S2 H4L CAS 67097-83-0 (1618)

3-Hydroxy-5,7-disulfo-2-naphthoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO4	25°C	0.50M	C			K1=10.81 B2=19.26	1978LAa (77293)	720

C11H9N02 HL CAS 92609-55-3 (4827)

5-Acetyl-8-hydroxyquinoline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	diox/w	25°C	60%	U			B3=26.64	1973SCd (77327)	721

Medium: 60% dioxan, 0.1 M NaClO4

C11H9N04 H2L CAS 4321-82-7 (4829)

3-Acetyl-4-hydroxycoumarin oxime;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	diox/w	35°C	50%	U			K(Al+HL)=4.32 K(Al+2HL)=7.69	1971MAa (77412)	722

Medium: 50% dioxan, 0.01 M NaClO4

C11H9N302 H2L PAR CAS 1141-59-9 (636)

4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	vlt	NaClO4	25°C	0.10M	U			K(Al+2HL)=23.52	1975TBc (77526)	723

Al+++ sp NaClO4 20°C 0.10M U K1=11.5 1967SNb (77527) 724

C11H10N206S H3L (7533)

7-(2-Carboxymethyl)amino-8-hydroxyquinoline-5-sulfonic acid; HOOCCH2NHC9H4N(OH)HSO3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaNO3	25°C	0.10M	M			K1=11.23 B2=22.17 K(Al+HL)=10.52 K(Al+2HL)=21.17	1996SOa (77682)	725

C11H11N06 H3L CAS 1147-65-5 (425)

N-(2'-Carboxyphenyl)iminodiethanoic acid; HOOC.C6H4.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaNO ₃	25°C	0.10M	M			K1=11.88 B2=22.34 K(Al+HL)=10.91 K(Al+2HL)=21.20	1996S0a (81106)	737

C12H13N03		HL					(1054)			
4-Dimethylamino-benzylideneypyruvic acid; (CH ₃) ₂ N.C ₆ H ₄ .CH:CH.CO.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	NaClO ₄	25°C	0.50M	C			K1=2.563	1984MTa (81191)	738

C12H14N4O2S		L	Sulfadimidine		CAS 57-68-1	(6167)				
2-(4-Aminobenzolsulfamido)-4,6-dimethylpyrimidine;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	alc/w	25°C	50%	C			K1=8.55 B2=15.00	1999GAa (81366)	739
Medium: 50% EtOH/H ₂ O, 0.10 M NaNO ₃ .										

C12H14O14		H6L					CAS 111451-17-3	(5895)		
3,6-Dioxaoctane-1,2,4,5,7,8-hexacarboxylic acid; (CH ₂ (COOH).CH(COOH).O.CH(COOH))-) ₂										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C			K1=8.56	1989MMd (81415)	740
K(AlL+H)=5.73 K(AlHL+H)=3.23 K(AlL+Al)=1.2 K(AlH-1L+H)=4.95										
B(Al2H-1L)=10.96										

C12H17N3O10		H4L	Asp-Asp-Asp				CAS 6445-	(6445)		
Aspartyl-aspartyl-aspartic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=8.45	2003KFa (81735)	741
B(AlHL)=12.33 B(AlH2L)=15.51 B(AlH-1L)=3.00										

C12H18N2O5S		H2L					CAS 80459-15-0	(1595)		
2-Nitroso-5-(N-propyl-3-sulfopropylamino)phenol;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO ₃	25°C	0.10M	C			K1=7.47 B2=14.41	1988YSc (81805)	742

C12H19N02 HL (5890)

3-Hydroxy-2-methyl-1-hexylpyridin-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.15M	C			K1=11.51 B2=22.49 B3=31.71	1989CNa (81976)	743

C12H20N2O8 H4I BDTA CAS 868-43-9 (1742)

C12H22N2O8 1142 BBTA CAS 60-16-0
DL-2,3-Diaminobutane-N,N,N',N'-tetraethanoic acid;
(HOOC.CH₂)₂N.CH(CH₃).CH(CH₃).N(CH₂.COOH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ oth KNO₃ 20°C 0.10M U K1=18.5 1965JMb (82282) 744

Method: electrophoresis

C12H20N2O8 H4L CAS 22968-57-6 (3992)

meso-2,3-Diaminobutane-N,N,N',N'-tetraethanoic acid;

$$(\text{HOOC} \cdot \text{CH}_2)_2 \text{N} \cdot \text{CH}(\text{CH}_3) \cdot \text{CH}(\text{CH}_3) \cdot \text{N}(\text{CH}_2 \cdot \text{COOH})_2$$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al^{+++} 8th KNO_3 20°C 0-10M II $K=16.5$ 1965 JMB (82383) 745

Method: electrophoresis

C12H22O12 HL Lactobionic acid CAS 96-82-2 (2487)

4-O-Beta-D-Galactopyran

(e) *Food & Galactopygeus*, *angustifrons* (Gmelin),

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
Al+++	gl	NaNO ₃	25°C	0.10M	C					1995E0a (82927)	746
										B(AlH-3L)=-11.98	

C12H26N2O4 L Cryptand 2,2 CAS 23978-55-4 (925)

4,7,13,16-Tetraoxa-1,10-diazacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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Al+++ gl R4N-X 25°C 0-05M u K1=9.4 1999BDh (83812) 747

Medium: Et4NC104

C12H26O4S HL SDS CAS 151-21-3 (2522)
Dodecyl sulfate; CH₃(CH₂)₁₁OSO₃H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ dis NaNO₃ 25°C 0.10M C M 1994BSb (83979) 748

REDACTED (EX-17) (EX-17) - 2003

Read (REVIEW REVIEW) 2003

At pH 6.88. A=Ferrioxamine B.

C12H27N3O3 L (6685)

1,3,5-Trideoxy-1,3,5-tris(dimethylamino)-cis-inositol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C			K1=14.23 B2=26.25 B(AlHL2)=30.50	1995HKb (84067)	749
Other models also considered., e.g. K1=14.21, B2=26.23, B(AlH-1L)=8.71										
Al+++	gl	KCl	25°C	0.10M	C			K1=16.74 B2=30.25 B(AlHL2)=34.37	1995HKb (84068)	750
Other models also considered., e.g. K1=16.62, B2=30.22, B(AlH-1L)12.53										
Al+++	gl	KCl	25°C	0.10M	U			K1=14.3 B2=26.4 B(AlH-1L)=8.9 B(AlHL2)=30.6	1992KHa (84069)	751

C13H8O3 HL CAS 719-41-5 (3397)

1-Hydroxyxanthone (1-Hydroxy-9-xanthenone)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	alc/w	25°C	50%	U			K1=10.37	1968GDb (84494)	752
Medium: 50% EtOH, 0.1 M NaClO4										

C13H9N03		H2L						(6878)		
2-Hydroxy-1-carbazole carboxylic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	25°C	?	U				1993TPa (84596)	753
K(Al+HL=AlL+H)=0.97										
K(AlLOH+H)=6.2										
K(AlL(OH)2+2H)=11.9										
K(Al2LOH+3H=AlL+Al)=11.8										
K(Al3L(OH)4+4H=AlL+2Al)=12.7. Method: fluorescence spectroscopy										

C13H9N3O5 HL TAN CAS 1147-56-4 (4030)

1-(1',3'-Thiazol-2'-ylazo)-2-hydroxynaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	vlt	NaClO4	25°C	0.10M	U			B2=14.28	1975TBc (84614)	754

C13H9N3O8S3		H3L						(28467-51-8)		(898)
2-(2-Thiazolylazo)chromotropic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

 Al+++ sp NaClO₄ 25°C 0.10M U 1982PRa (84663) 755
 K(Al+H₂L=AlL+2H)=-2.70

 C13H10N2O4 HL CAS 13245-57-3 (4983)
 N-4-Nitrobenzoyl-N-phenylhydroxylamine; O₂N.C₆H₄.CO.N(C₆H₅).OH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Al+++ gl mixed 30°C 50% U K1=8.27 B2=15.89 1971GSc (84882) 756
 K3=6.73

 Medium: 50% v/v acetone/H₂O, 0.5 M NaClO₄

 C13H10N2O4 HL CAS 2029-61-0 (178)
 N-Phenyl-2-nitrobenzohydroxamic acid; O₂N.C₆H₄.CO.N(C₆H₅).OH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Al+++ gl mixed 30°C 50% U K1=6.60 B2=14.07 1971GSc (84895) 757
 K3=6.73

 Medium: 50% v/v acetone/H₂O, 0.5 M NaClO₄

 C13H10N2O4 HL CAS 17120-18-2 (220)
 N-Phenyl-3-nitrobenzohydroxamic acid; O₂N.C₆H₄.CO.N(C₆H₅).OH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Al+++ gl mixed 30°C 50% U K1=8.32 B2=16.07 1971GSc (84907) 758
 K3=6.77

 Medium: 50% v/v acetone/H₂O, 0.5 M NaClO₄

 C13H11N L 3-Stilbazole (6869)
 (3-Pyridyl)styrene; C₅H₄N.CH:CH.C₆H₅

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Al+++ sp non-aq 25°C 100% U M 1993IWa (85006) 759
 K(AlACl+L)=-2.54 (L is trans)
 K(AlACl+L)=-2.62 (L is cis)

 Medium:Dichloroethane. H₂A:Tetraphenylporphyrin

 C13H11N L 4-Stilbazole (6868)
 (4-Pyridyl)styrene; C₅H₄N.CH:CH.C₆H₅

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Al+++ sp non-aq 25°C 100% U M 1993IWa (85007) 760
 K(AlACl+L)-1.98 (L is trans)
 K(AlACl+L)=-1.98 (L is cis)

 Medium:Dichloroethane. H₂A:Tetraphenylporphyrin

B(-2,1,1) determined by spectrophotometry.

Al+++ gl NaClO4 25°C 0.10M M T K1=14.11 B2=26.69 1989C0a (86710) 767

Al+++ gl oth/un 25°C 0.00 U 1988RCa (86711) 768
B(AlH2L2)=12.88

Al+++ sp NaClO4 20°C 0.10M M K1=14.11 B2=26.92 1987C0a (86712) 769
B(Al(OH)L2)=34.1

Al+++ EMF oth/un ? 0.10M U K1=11.31 B2=17.37 1972GBc (86713) 770

Al+++ sp NaClO4 rt 0.10M U 1971NOc (86714) 771
K(Al+2H2L)=11.5

Al+++ sp oth/un 25°C 0.10M U 1968BNa (86715) 772
K(Al+2HL)=12.06

C14H807S H3L (4037)

1,4-Dihydroxyanthraquinone-2-sulfonic acid, quinizarin-2-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KN03 25°C 0.05M C K1=9.04 1993PFa (86777) 773
B(AlH-1L)=4.33
B(AlH-2L)=-1.47
B(Al2H-1L)=8.89
B(Al2H-3L)=0.80

C14H10N02F3 HL CAS 530-28-9 (2574)

N-(3-Trifluoromethylphenyl)-2-aminobenzoic acid; HOOC(C6H4)NH(C6H4)CF3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl mixed 22°C 90% U K1=5.95 1982GKb (86896) 774
Medium: 90% DMF/H2O

C14H1007S H5L CAS 30782-99-1 (5045)

1,2,5,10-Tetrahydroxyanthracene-3-sulfonic acid (Leucoalizarin red S)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp NaClO4 ? 0.10M U 1971NPb (86934) 775
K(Al+H3L)=7.9
K(Al+H4L)=6.3

C14H11N508S2 H5L CAS 1105-53-9 (5084)

1,5-Bis(2-hydroxy-5-sulfophenyl)-3-cyanoformazan;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaNO₃ 20°C 0.10M U K1=16.40 1971SEa (87017) 776

C14H11O2NF₂S HL CAS 51679-49-3 (2928)
N-((3-Difluoromethylthio)phenyl)anthranilic acid;HOOC(C₆H₄).NH.(C₆H₄).S.CHF₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	mixed	22°C	90%	U			K1=6.32	1982GKb (87026)	777
Medium: 90% DMF/H ₂ O										

C14H11O2NF ₂ S	HL							CAS 51679-50-4 (2929)		
N-((4-Difluoromethylthio)phenyl)anthranilic acid;HOOC(C ₆ H ₄).NH.(C ₆ H ₄).S.CHF ₂										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	mixed	22°C	90%	U			K1=6.12	1982GKb (87031)	778
Medium: 90% DMF/H ₂ O										

C14H16N20S5	H2L							CAS 390426-77-4 (8803)		
1-n-Butyl-5-sulfo-8-hydroxyquinoline-7-carboxamide;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	NaClO ₄	25°C	0.01M	C			K1=8.4 B2=16.20	2001LAa (87905)	779
Medium: 0.01 M HClO ₄ . Method: spectrophotometric titration.										

C14H17N5O3	HL	Pipemidic acid	CAS 51940-44-4 (2535)							
8-Ethyl-5,8-dihydro-5-oxo-2-(1-piperazinyl)pyrido[2,3-d]pyrimidine-6-carboxylic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	25°C	0.05M	U				2000MPa (88055)	780
K _{1eff} =6.07										
Medium: 0.05 M chloroethanoate, pH=5.5. Method: spectrofluorimetry.										
For Cinoxacin, K _{1eff} =5.601-Ethyl-3-carboxy-6,7-methylenedioxy-4-cinnolone										

C14H22N208	H4L	CDTA	CAS 482-54-2 (200)							
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO ₃	35°C	0.10M	U			K1=19.18	1980KHb (88572)	781
Al+++	ISE	KNO ₃	25°C	0.10M	U			K1=18.9 K(Al+HL)=3.4 K(AlL+OH)=6.3	1967ABb (88573)	782
Al+++	sp	NaClO ₄	25°C	0.20M	U			K1=18.50	1967BAC (88574)	783

By glass electrode: $K(AlL+H)=2.29$, $K(AlL(OH)+H)=7.82$

Al+++ ISE KN03 20°C 0.10M U T H K1=18.63 1966MCa (88575) 784
 $K(AlL+H)=2.59$

$K1=18.8(30\text{ }^{\circ}\text{C}), 19.15(40\text{ }^{\circ}\text{C})$; $DH(K1)=-46.0\text{ kJ mol}^{-1}$, $DS=510\text{ J K}^{-1}\text{ mol}^{-1}$

Al+++ dis NaClO4 20°C 0.10M U 1963STc (88576) 785
 $B(AlL(OH))=26.61$

Medium: KC1O4

Al+++ vlt KN03 20°C 0.10M U K1=17.63 1954SGa (88577) 786
 $K(AlLOH+H)=7.58$
 $K(AlL+H)=3.93$

C14H23N3O8 H4L NODASA CAS 210217-93-9 (8716)

1,4,7-Triazacyclononane-1-succinic acid-4,7-diethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ nmr oth/un 20°C 0.12M C K1=18.50 2002AMb (89008) 787
Medium: 0.02 M Al(NO3)3, pH 2.0. Method: 27Al nmr.

C14H23N3O10 H5L DTPA CAS 67-43-6 (238)
Diethylenetriamine-pentaethanoic acid; HOOC.CH2.N(CH2.CH2.N(CH2.COOH)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ ISE KN03 25°C 0.10M C M 1996YHa (89144) 788
 $K(AlL+H)=5.18$
 $K(AlHL+F)=5.3$
 $K(AlL+F)=2.9$

Method: Fluoride ISE.

Al+++ gl KN03 35°C 0.10M U K1=20.66 1980KHz (89145) 789

Al+++ ISE KN03 25°C 0.10M U K1=18.7 1967ABb (89146) 790
 $K(Al+HL)=4.3$
 $K(AlL+OH)=6.6$

Al+++ ISE KN03 20°C 0.10M U T H K1=18.4 1966MCa (89147) 791
 $K(AlL+H)=4.63$

$K1=18.51(25\text{ }^{\circ}\text{C}), 18.62(30\text{ }^{\circ}\text{C}), 18.80(40\text{ }^{\circ}\text{C})$. At 25 °C: $DH(K1)=33\text{ kJ mol}^{-1}$, $DS=472$

C14H24N2O10 EGTA CAS 67-42-5 (349)
Ethylene glycol-O,O'-bis(2-aminoethyl ether)-N,N,N',N'-tetraethanoic acid; H4L

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl NaClO4 25°C 0.20M U K1=13.90 1967BDb (89837) 792
 $K(AlL+H)=3.97$

$$K(AlOH+H)=5.20$$

$$K(Al(OH)_2+H)=8.42$$

K1 by spectrophotometry

C14H25N3O7 H3L (5397)
1-Oxa-4,7,10-triazacyclododecane-4,7,10-triethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C			K1=12.5 K(Al+H)=3.28	1993DSa (90079)	793

C14H28N2O4 L Cryptand 2,1,1 CAS 31250-06-3 (836)
1,10-Diaza-4,7,13,18-tetraoxabicyclo[8.5.5]eicosane (2,1,1);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	R4N.X	25°C	0.05M	U			K1=12.9	1999BDb (90345)	794

Medium: Et4NC1O4

C15H10N3OCl1 HL CAS 16195-35-0 (27)
5-(4-Chlorophenylazo)-8-hydroxyquinoline; Cl.C6H4.N:N.C9H5N.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	25°C	0.10M	U			B3=11.91	1978K1a (90946)	795

C15H10N3O5C1S H3L (7520)
7-[(2-Hydroxy-5-chlorophenyl)azo]-8-hydroxyquinoline-5-sulfonic acid; C6H3Cl(OH)N=NC9H4N(OH)(SO3H)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	kin	KNO3	25°C	0.10M	C			K1=23.3	1996PKa (90952)	796

C15H1007 H5L Quercetin CAS 117-39-5 (5101)
3,5,7-Trihydroxy-2-(3',4'-dihydroxyphenyl)-1-benzopyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	non-aq	25°C	100M	C			K1eff=-2.34	2001ADb (91020)	797

Medium: MeOH, 0.2 M acetate buffer, pH 5.0. K1eff: Al+HnL=All

C15H10010S H5L CAS 141896-20-0 (8182)
2-(3,4-Dihydroxyphenyl)-3,5-dihydroxy-7-(sulphooxy)-4H-1-benzopyran-4-one, Quercetin-7-sulfonic;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Al+++ sp non-aq 25°C 100M C 2001ADb (91028) 798

K1eff=-1.73

Medium: MeOH, 0.2 M acetate buffer, pH 5.0. K1eff: Al+HnL=All

C15H10010S H5L Quercetin S F CAS 25001-18-7 (1520)
3,5,7,3',4'-Pentahydroxy-5'-sulfoflavone; (HO)3(O)C9H20.C6H2(SO3H)(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp NaClO4 20°C 0.10M U K1=5.11 1989K0a (91031) 799

Al+++ sp NaClO4 20°C 0.10M U 1976KTb (91032) 800

B(AlH4L)=7.56

B(A12H2L)=20.9

C15H11N02 H2L (430)
2-(2'-Hydroxyphenyl)-8-hydroxyquinoline; HO.C6H4.C9H5N.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl diox/w 25°C 50% U K1=19.8 B2=34.70 1974CCb (91056) 801

C15H11N30 HL PAN CAS 85-85-8 (572)
1-(2-Pyridylazo)-2-naphthol; C5H4N.N:N.C10H6.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ vlt NaClO4 25°C 0.10M U B2=21.90 1975TBc (91206) 802

Al+++ vlt alc/w 25°C 50% U K1=12.86 1973TBa (91207) 803

Medium: 50% EtOH, 0.06 M (H/Na)ClO4

C15H11N30 HL CAS 4312-09-8 (989)
5-Phenylazo-8-hydroxyquinoline; C6H5.N:N.C9H5N.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp oth/un 25°C 0.10M U K1=4.37 1978KIa (91266) 804

C15H11N304S H2L (5130)
7-Phenylazo-8-hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp NaNO3 25°C 0.10M C K1=7.53 1995SOa (91334) 805

C15H11N305S H3L CAS 111248-75-0 (8411)
5-(2'-Hydroxy-5'-phenylazo)-8-quinolinol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	RT	dil	C				1985IBa (91340)	806
								K1eff=6.73 B2eff=11.20 B3eff=15.64		
Medium: Britton and Robinson buffer, pH 6.6										
C15H11O3Cl		H2L					CAS	654637-45-3	(9237)	
7,8-Dihydroxyflavylium chloride;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	none	25°C	0.0	C			K1=9.11	2003MMa (91401)	807

C15H11O5Cl		H4L	Luteolinidin				CAS	1154-78-5	(9239)	
2-(3,4-Dihydroxyphenyl)-5,7-dihydroxy-1-benzopyrilium chloride;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	none	25°C	0.0	C			K1=6.81	2003MMa (91403)	808

C15H12O2		HL					CAS	1214-47-7	(951)	
3-Phenyl-1-(2'-hydroxyphenyl)-2-propen-1-one, 2'-hydroxychalkone;										
C6H5.CH:CH.CO.C6H4.OH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	diox/w	30°C	60%	U			K1=11.35 B2=21.05	1975KKc (91576)	809

C15H14O6		H4L	Catechin				CAS	154-23-4	(2737)	
3,3',4',5,7-Pentahydroxyflavone; (HO)3(O)C9H6O.C6H3(OH)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C				2002IIb (91815)	810
								K(Al+H4L=AlH2L+2H)=-5.57		
								K(AlH2L=AlH2L(OH)+H)=-5.76		
								K(AlH2L+H4L=Al(H2L)2+2H)=-8.3		
For (-)-epicatechin, K(Al+H4L=AlH2L+2H)=-5.75, K(AlH2L=AlH2L(OH)+H)=-6.0, K(AlH2L+H4L=Al(H2L)2+2H)=-8.2.										
Al+++	gl	NaClO4	25°C	0.50M	C			K1=18.3	1985CDa (91816)	811
								K(Al+HL)=12.4		

Al+++	gl	KCl	25°C	0.10M	C				1985KPa (91817)	812
								K(Al+H2L)=17.10		
								K(AlH2L+H2L)=13.89		
								K(Al(H2L)2+H2L)=9.93		
								K(Al(OH)H2L+H)=5.98		

$K(A1(H2L)2=A1(OH)(H2L)2+H)=-8.22$. For the epicatechin dimer H8L constants for AlH6L, Al(H6L)2 and Al(OH)H6L given in Austral. J Chem., (1985) 38, 879

C15H14O7 H5L CAS 970-73-0 (1796)
Epigallocatechin;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C			2002IIb (91819)	813
							$K(A1+H5L=A1H3L+2H)=-5.75$		
							$K(A1H3L=A1H3L(OH)+H)=-5.23$		
							$K(A1H3L+H5L=A1(H3L)2+2H)=-7.16$		

C15H15N02 HL CAS 61-68-7 (2927)
N-(2,3-Dimethylphenyl)anthranilic acid; HOOC(C6H4).NH.(C6H3)(CH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	mixed	22°C	90%	U		K1=6.56	1982GKb (91829)	814

Medium: 90% DMF/H2O

C16H11N205C1S HL (7535)
2-[(2-Hydroxy-5-chlorophenyl)azo]-1-hydroxynaphthalene-4-sulfonic acid; HO.C6H3C1N=NC10H5N(OH)HSO3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	kin	KNO3	25°C	0.10M	C		K1=15.26	1996PKa (92769)	815

C16H11N306S H3L (1047)
7-(4-Carboxyphenylazo)-8-hydroxyquinoline-5-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	sp	NaCl	25°C	0.10M	U		K1=9.86 B2=16.84	19840Fa (92850)	816

C16H11N307S H3L CAS 116946-37-3 (1598)
7-Hydroxy-((4-carboxyphenyl)azo)-8-hydroxy-5-quinolinesulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	gl	NaCl	22°C	0.10M	C		K1=7.49	1988BEa (92852)	817

C16H11N3010S2 H4L Chromotrope 2B CAS 548-80-1 (896)
2-((4-Nitrophenyl)azo)chromotropic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Al+++	sp	NaClO4	25°C	0.10M	U			1982PRa (92861)	818
							$K(A1+H2L=A1L+2H)=-4.70$		

C16H12N205S H3L SolochromeVio R CAS 94205-83-1 (4093)
1-(2'-Hydroxy-5'-sulfophenylazo)-2-naphthol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	25°C	0.0	U			K1=18.4 B2=31.6	1962CRa (93021)	819

C16H12N208S2 H4L Chromotrope 2R CAS 4197-07-3 (2604)
2-(Benzeneazo)-chromotropic acid, Acid Red 29

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO4	25°C	0.10M	U			K1=16.70 B2=25.70	1975MPa (93059)	820

Al+++ gl KCl 20°C 0.10M U K1=18.41 1964PCa (93060) 821

C16H12N2011S3 H5L CAS 548-81-2 (5180)
2-(4'-Sulfophenylazo)chromotropic acid,
2-(4-sulfophenylazo)-1,8-dihydroxyphthalene-3,6-diHSO3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO4	25°C	0.10M	U			K1=10.00 B2=13.80	1975MPa (93091)	822

C16H13N2010AsS2 H5L Thorin I CAS 3688-92-4 (2609)
1-((2-Arsenophenyl)azo)-2-hydroxy-3,6-naphthalylsulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	25°C	?	U				1968GSe (93184)	823

K(?)=10.5

C16H13N307S H4L (1596)
8-Hydroxy-7-((2-hydroxy-5-carboxyphenyl)azo)-5-quinoline sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	NaCl	22°C	0.10M	C			K1=14.15	1988BEa (93287)	824

C16H13O3Cl H2L CAS 125653-94-3 (9238)
7,8-Dihydroxy-4-methylflavylium chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	none	25°C	0.0	C			K1=8.30	2003MMa (93393)	825

C16H13O5Cl H3L (261)
3',4',7-Trihydroxy-3-methoxyflavylium chloride; (HO)2C6H3.C9H4O(OH)(OCH3)Cl

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.10M	C				2002A0a (93898)	832
							B(-2,1,1)=-3.91			
							B(-3,1,1)=-8.17			
							B(-4,1,1)=-13.79			
							B(-6,1,2)=-19.28			
							B(-7,1,2)=-27.65, B(-9,1,3)=-34.01.			
							B(p,q,r): pH+qAl+rH3L=HpAlq(H3L)r			

C16H24O14			H4L				CAS 61696-54-6 (6104)			
1,4,7,10,13,16-Hexaoxacyclooctadeca-2,3,11,12-tetracarboxylic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	R4N.X	25°C	0.05M	C	M	K1=9.0		1998TSb (94490)	833
							B(AlHL)=13.1			
							B(AlH2L)=17.3			
Medium: 0.05 M Et4NC104. Also ternary complexes, MAlH-nL, where M=Na, K, Cs, Ca, Sr, Ba.										

C16H28N4O8			H4L	DOTA			CAS 60239-18-1 (1017)			
1,4,7,10-Tetraazacyclododecane-1,4,7,10-tetraethanoic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaNO3	25°C	0.20M	C		K1=17.0		1995KKa (94876)	834

C16H32N2O5		L	Cryptand 2,2,1		CAS 31364-42-8 (837)					
1,10-Diaza-4,7,13,16,21-pentaoxabicyclo[8,8,5]tricosane (2,2,1);										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	R4N.X	25°C	0.05M	U		K1=11.3		1999BD _b (95175)	835
Medium: Et4NC104										

C17H14N2O7S			H4L		(1597)					
4-Hydroxy-((2-hydroxy-5-carboxyphenyl)azo)-naphthalenesulfonic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	22°C	0.10M	C		K1=15.27		1988BEa (95936)	836

C17H16N2O2			HL				CAS 65840-98-4 (8454)			
3-(2-Hydroxy-5-methylphenyl)-5-(4-methoxyphenyl)pyrazole;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	diox/w	27°C	70%	C		K1=10.55	B2=20.80	1994SNa (96027)	837
							K3=7.45			

Medium: 70% v/v dioxane/H₂O, 0.10 M NaClO₄.

C17H18N3O3F		HL	Ciprofloxacin	CAS 189257-90-7	(7142)		
1-Cyclopropyl-6-fluoro-1,4-dihydro-4-oxo-7[1-piperazinyl]-3-quinoline carboxylic acid;							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values
Al+++	gl	KCl	25°C	0.20M	C		Reference ExptNo
							1996TBc (96223) 838
							B(AlH2L2)=29.33
							B(AlH2L3)=35.8
							B(AlH3L3)=43.26

C18H20N2O6		H4L		CAS 10328-28-6	(3501)		
Ethylenedinitriolo-N,N'-bis(2'-hydroxyphenyl)-N,N'-diethanoic acid;							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values
Al+++	gl	KCl	25°C	0.10M	C		Reference ExptNo
							K1=25.78
							K(A1L+H)=3.73

C18H20N2O6		H4L	EHPG	CAS 10328-28-6	(429)		
N,N'-Ethylene-bis-(2-(2'-hydroxyphenyl))glycine; (HOOCCH(C6H4OH)NHCH2.) ₂							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values
Al+++	sp	oth/un	25°C	0.10M	C		Reference ExptNo
							2003YFc (97420) 840
							K1eff=9.38
Method: UV difference spectrophotometry. Medium: 0.10 M HEPES, pH 7.4.							
Al+++	gl	NaCl	25°C	0.12M	C		K1=24.48
							1981RMb (97421) 841

C18H20N3O4F		HL	Ofloxacin	CAS 82419-36-1	(7789)		
a-Fluoro-3-methyl-10-(4-methyl-1-piperazinyl)-7-oxo-2,3-dihydro-7H-pyrido-1,4-benzo xazine-6-COOH;							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values
Al+++	gl	KCl	25°C	0.10M	C	I	Reference ExptNo
							K1=10.37
							B(AlHL)=16.40
Medium: 0.10 M LiCl, 0.001 M Triton. In 0.10 M LiCl, 0.005 M CTAB,							
							K1=11.56, B(Al2H-2L)=3.6.

C18H22N4O4		H2L		CAS 2444-14-6	(3502)		
N,N'-Bis(2-pyridylmethyl)diaminoethane-N,N'-diethanoic acid;							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values
Al+++	gl	NaCl	25°C	0.16M	C		Reference ExptNo
							K1=10.85
							K(A1+L=A1L(OH)+H)=6.37

$$K(AlL(OH)+H=AlL)=4.48$$

C18H26N206P2 H4L CAS 53431-86-0 (5266)
Ethylenebis(imino(2-hydroxyphenyl)methylene(methyl)phosphinic acid);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	EMF	oth/un	?	?	U			K1=20.0 K(Al+HL)=15.36	1970DMc (97673)	844

C18H30N4O12 H6L TTHA CAS 869-52-3 (694)
Triethylenetetraaminehexaethanoic acid;((HOOC.CH2)2N.CH2.CH2.N(CH2.COOH).CH2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	R4N.X	25°C	0.10M	C			K1=20.23 K(AlL+H)=5.97 K(AlL+Al)=9.55 K(Al2L(OH)+H)=4.68 K(Al2L(OH)2+2H)=9.87	1998ACc (98007)	845

Medium: N(CH₃)₄NO₃.

Al+++	ISE	KNO ₃	25°C	0.10M	C	M			1996YHa (98008)	846
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Method: Fluoride ISE.

Al+++	gl	KNO ₃	35°C	0.10M	U			K1=18.74	1980KHb (98009)	847
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Al+++	gl	KNO ₃	25°C	0.10M	U			K1=19.7 K(AlL+H)=5.85 K(Al2L+2OH)=15.9	1970HAa (98010)	848
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By ion-selective electrode (Hg): B(Al₂L)=28.6. By redox: B(Al₂L)=28.9

C18H32N4O8 H4L TETA CAS 60239-22-7 (1019)
1,4,8,11-Tetraazacyclotetradecane-1,4,8,11-tetraethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaNO ₃	25°C	0.20M	C			K1=16.3	1995KKa (98188)	849

C18H36N206 L Cryptand 2,2,2 CAS 23978-09-8 (514)
1,10-Diaza-4,7,13,16,21,24-hexaoxabicyclo[8.8.8]hexacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	R4N.X	25°C	0.05M	U			K1=10.6	1999BD _b (98513)	850

Medium: Et₄NC₁₀O₄

C19H12O8S H4L Pyrogallol red CAS 85531-30-2 (638)
Pyrogallolsulfonephthalein;

Al+++ gl KCl 25°C 0.20M U K1=25.12 B2=47.39 1970G0a (99103) 857
K3=20.74

C19H15N L (6870)
(4-Phenyl-3-pyridyl)styrene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	non-aq	25°C	100%	U	M			1993IWa (99120)	858
								K(AlACl+L)=-2.02 (L is trans)		
								K(AlACl+L)=-2.52 (L is cis)		

Medium:Dichloroethane. H2A:Tetraphenylporphyrin

C19H15N08 H4L Alizarin Comp. CAS 3952-78-1 (671)
(3,4-Dihydroxy-2-anthraquinonyl-methyl)iminodiethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	diox/w		20%	U				1973INa (99129)	859
								K(Al+HL)=14.3		
								B(Al2L)=25.3		

Medium: 20% dioxan, 0.1 M

C19H19N706 H3L Folic acid CAS 75708-92-8 (194)
Pteroylglutamic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO3	30°C	0.10M	U	I		K1=4.65 B2=8.85 1970NDa (99283)	860	
								K3=4.10		

I=0: K1=5.80, K2=4.70, K3=4.65. I=0.01: K1=5.25, K2=4.55, K3=4.50.

I=0.05: K1=4.80, K2=4.28, K3=4.15

C19H39N706 H3L TETMAHA (7468)
1,4,8,11-Tetraazacyclotetradecane-N,N',N"-tris(N-methylacetohydroxamic acid);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO3	25°C	0.10M	C			K1=21.02 1999GGa (99501)	861	
								B(AlHL)=30.72		
								B(AlH2L)=35.60		
								B(AlH3L)=40.71		
								B(AlH4L)=44.17		

C20H11N09S2 H3L CAS 65501-73-7 (8982)
6-Hydroxy-5-dibenzo[a,j]phenoxazone-8,11-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	KCl	RT	1.0M	C				1980NLa (99533)	862

$$K(Al+HL=AlL+H)=-0.39$$

At pH 5-6.5, $K_{1\text{eff}}=4.77$. Data for solutions with Septonex and cetylpyridinium surfactants

C20H13N307S H3L Eriochrome Bl T CAS 1787-61-7 (997)
1-(1-Hydroxy-2-naphthylazo)-6-nitro-2-naphthol-4-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	20°C	0.10M	U				1980PKa (99559)	863

$$K(Al+2HL)=12.51$$

Medium: Na₂SO₄

Al+++ gl NaClO₄ 25°C 0.10M U K1=9.56 B2=13.66 1975MPa (99560) 864

C20H14N205S H3L Solochrome 6B CAS 3564-14-5 (3507)
1-(1-Hydroxy-2-naphthylazo)-2-naphthol-4-sulfonic acid, Mordant Black3, Eriochrome blue-black B;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO ₄	25°C	0.10M	U			K1=11.58 B2=20.97	1975MPa (99644)	865

C20H14N205S H3L EriochrBluBlk R CAS 2538-85-4 (3508)
3-Hydroxy-4-(2-hydroxy-1-naphthylazo)naphthalene-1-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	25°C	0.10M	U				1967NNc (99686)	866

$$K(Al+2HL)=41.97$$

$$K(AlOH+2HL)=40.62$$

C20H24N206 H4L HBED CAS 3625-89-6 (2208)
N,N'-Di-(2-hydroxybenzyl)-diaminoethane-N,N'-diethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	25°C	0.10M	C				2003YFc (99987)	867

$$K_{1\text{eff}}=8.88$$

Method: UV difference spectrophotometry. Medium: 0.10 M HEPES, pH 7.4.

Al+++ gl NaCl 25°C 0.12M C K1=24.78 1981RMB (99988) 868

C20H35N5010 H5L (6545)
1,4,7,10,13-Pentaazacyclopentadecane-N,N',N",N''',N""-pentaethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaNO ₃	25°C	0.20M	C			K1=16.1	1995KKa (100531)	869

C21H21012 H5L CAS 50986-17-9 (7770)
3-O-beta-D-Glucopyranosyldelphinidin ion;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp none 25°C 0.00 U 1997EFa (101191) 870
 $K(Al+H2L=AlL+2H)=-3.53$

C21H22010 L G-Rubrofusarin CAS 63174-98-1 (7067)
2-Methyl-5,6-dihydroxy-6-O-B-D-galactosyl-8-methoxy-naphtho-pyrone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp NaClO₄ 25°C 1.00M C K1=8.91 1995PDA (101213) 871

C22H1409 H5L CAS 4431-00-9 (3513)
Aurintricarboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp oth/un ? ? U K1=16.6 1972PKa (101491) 872

C22H18N4O14As2S2 H8L Arsenazo III CAS 1668-00-4 (1148)
2,7-Bis(2'-arsonophenylazo)chromotropic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sp oth/un 25°C 0.10M U 1975MIA (101609) 873
 $K(Al+H4L=AlH2L+2H)=-1.17$
 $K(AlOH+H5L=Al(OH)H3L+2H)=0.47$

C22H18011 H8L CAS 989-51-5 (2270)
Epigallocatechin gallate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl KCl 25°C 0.10M C 2002IIb (101680) 874
 $K(Al+H8L=AlH6L+2H)=-4.47$
 $K(AlH6L=AlH5L(OH)+H)=-4.74$

C22H23N208Cl H2L Aureomycin CAS 56235-18-8 (3515)
Chlorotetracycline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ gl oth/un 20°C 0.01M U K1=7.2 1956ARd (101758) 875

C22H24N208 H2L Tetracycline CAS 60-54-8 (2201)
Tetracycline;

$$\begin{aligned}B(\text{Al3-H2L2}) &= 13.44 \\B(\text{Al4-H3L5}) &= 29.07 \\B(\text{Al4-H4L5}) &= 25.30\end{aligned}$$

B(Al1-HL) determined by spectrophotometry, the other values by potentiometry
 $\text{B(Al4-H5L5)} = 20.67$ species formed at higher alkali addition rate.

C23H18O9S H4L Eriochrome cyan CAS 3564-18-9 (433)
4'-Hydroxy-3,3'-dimethyl-2''-sulfofuchsone-5,5'-dicarboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	25°C	0.10M	U			K1=13.66 B(AlHL)=18.25 B(AlH2L)=22.29	1975EPa (102625)	884
Al+++	sp	R4N.X	25°C	0.10M	U			K(Al+H2L=AlHL+H)=1.9 K(AlHL+2HL)=8.1	1973NNb (102626)	885

Medium: $(\text{Na}, \text{NH}_4)\text{Cl}$

C24H29N3012S3 H6L (7355)
1,2,3-Tris((2-hydroxy-5-sulfobenzyl)amino)propane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.16M	C			K1=22.8	1997C0a (103014)	886

C24H42N6O12 H6L (6546)
1,4,7,10,13,16-Hexaazacyclooctadecane-N,N',N'',N''',N'''',N'''''-hexaethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaNO ₃	25°C	0.20M	C			K1=22.09	1995KKa (103370)	887

C25H2009 H5L CAS 2947-64-0 (4166)
4',3''-Dihydroxy-3,3',4'''-trimethylfuchsone-5,5',5'''-tricarboxylic acid, Chromoxane violet R

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	?	0.10M	U			K1=10.42	1967LMF (103603)	888

C25H48N6O8 H3L Desferrioxamine CAS 70-51-9 (2488)
Desferrioxamine B; NH₂.((CH₂)₅.NOH.CO.C₂H₄.CO.NH)2.(CH₂)₅.NOH.CO.CH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KCl	25°C	0.20M	C			K1=23.9 B(AlHL)=33.8 B(AlH2L)=36.6	2000FEc (103800)	889

$$K(Al+H_2L)=16.3$$

$$K(Al+HL)=23.0$$

* $K(AlH_2L)=-2.8$, * $K(AlHL)=-9.9$.

Al+++	gl	KCl	25°C	0.10M	C	$K_1=24.50$	1989EHa (103801)	890
						$K(Al+HL)=24.14$		
						$K(AlHL+H)=1.18$		
						$K(AlL+H)=9.43$		

C26H25N09S H4L Semi-Xylenol O (426)
3-(N,N-Di(carboxymethyl)aminomethyl)-2-cresolsulfonephthalein;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	oth/un	25°C	0.10M	U			$K_1=16.7$	1979MUa (103942)	891

C26H33N3012S3 H6L (7354)
1,1,1-Tris((2-hydroxy-5-sulfobenzyl)amino)methyl)ethane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaCl	25°C	0.16M	C			$K_1=22.5$	1997COa (104061)	892

C26H48N6010 H4L CAS 207388-25-8 (7648)

Triethylenetetramine-N,N,N',N'',N''',N'''-hexaethanoic acid NN-bis(butanamide);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	R4N.X	25°C	0.10M	C			$K_1=14.59$	1998ACc (104304)	893

$$K(AlL+H)=4.40$$

$$K(AlL(OH)+H)=7.53$$

$$K(AlL+Al)=5.58$$

$$K(Al2L(OH)+H)=3.16$$

Medium: $N(CH_3)_4NO_3$. $K(Al2L(OH)2+2H)=6.92$.

C27H29N011 L Adriamycin CAS 25316-40-9 (2407)
Doxorubicin;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO ₃	25°C	0.05M	C				1993PFa (104452)	894

$$B(AlHL)=16.70$$

$$B(AlH-1L)=7.45$$

$$B(AlH-2L)=-0.46$$

C27H30N4018S3 H9L TRENCAMS CAS 252906-93-7 (7599)
3,3',3"-[Nitrilotris(2,1-ethanediyliminocarbonyl)]tris(4,5-dihydroxybenzenesulfonic

acid);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	NaClO4	25°C	0.10M	C			K1=39.97 B(AlHL)=46.27 B(AlH3L)=54.79	2002BBd (104479)	895

K1 by spectrophotometry using competitive reaction with edta, pH 7.2.

C27H30O16 H4L Rutin CAS 153-18-4 (4169)

3,3',4',5,7-Pentahydroxyflavone-3-beta-rutinoside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	non-aq	25°C	100M	C				2001ADb (104505)	896
								K1eff=-1.92		

Medium: MeOH, 0.2 M acetate buffer, pH 5.95. K_{1eff}: A1+HnL=A1L

C27H31O16Cl H3L Cyanin CAS 2611-67-8 (9240)

2-(3,4-Dihydroxyphenyl)-3,5-bis(beta-D-glucopyranosyloxy)-7-hydroxy-1-benzopyrili um chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ sn none 25°C 0.0 C K1-6 74 2003MMa (104512) 897

C27H44O L Vitamin D3 CAS 67-97-0 (6103)
7-Dihydrocholesterol Cholecalciferol

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Al+++ g1 a1c/w 25°C 70% C K1=12.4 B2=24.40 2003MYC (104613) 898

Medium: 70% v/v EtOH/H₂O 0.10 M KNO₃

medium: 70% v/v EtOH/H₂O, 0.1% w/v KNO₃.

C₃₀H₂₇N₃O₁₅ H₆I Enterobactin CAS 28384-96-5 (2259)

Enterobactin: cyclo-((OH)C₆H₃(OH)CO-NH-CH-CO-CH₂)₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Alum. 50% KCl 25.86 g 12M C 1000 mL Ba²⁺ (105182) 800

Al⁺⁺⁺ Sp KCl 25°C 0.10M C
 K(AlL+H)=5.15
 K(AlHL+H)=3.4
 K(AlH₂L+H)=2.6

C30H45N4O6P3 H3L CAS 182250-11-9 (8686)

Tris(4-(phenylphosphinato)-3-methyl-3-azabutyl)amine;

Metal Mtd Medium Temp Conc CaL Flags Lg K values Reference ExptNo

Metal Area Medium Temp Zone Cut Flags Eg R Values Reference Expire

AI+++ 0.1M NaCl 25 °C 0.16M C 1996ERC (105321) 900

$$K(Al+H_3L)=0.93$$

$$K(Al+2H_3L)=3.45$$

Method: ^{27}Al nmr. Medium pH 1.5.

C31H32N2013S H6L Xylenol orange CAS 63721-85-5 (432)
5,5'-Bis-N,N-bis(carboxymethyl)aminomethyl-4'-hydroxy-3,3'-dimethylfuchsone-2"-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	KNO ₃	25°C	0.10M	C	M	K1=18.18 K(AlL+H)=7.45 K(AlHL+H)=5.14 K(AlL ₂ +2Mg)=4.03 K(AlL ₂ +2Ba)=1.12	B2=23.40	1998GBa (105450)	901
$K(AlL_2+Ba+Mg)=2.79$										
Al+++	gl	NaClO ₄	25°C	0.10M	U		B2=27.0 K(Al+HL)=16.3 K(Al(OH)HL+H=AlHL)=5.8 K*(Al ₂ L)=13.4		1981MYa (105451)	902
$K^*(Al_2L): K((Al(OH))_2L+2H=Al_2L+2H_2O)$										
Al+++	sp	NaClO ₄	20°C	0.10M	U		K(2Al+H ₃ L)=13.64 K(Al+H ₂ L+H ₃ L)=11.79		1969NNb (105452)	903

C32H37N09S H4L SemiMeThymolBlu (427)
3-(N,N-Di(carboxymethyl)-aminomethyl)thymolsulfonephthalein;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	oth/un	25°C	0.10M	U		K1=17.9 K(Al+HL)=9.3 K(AlL+H)=3.5 K(Al(OH)L+H)=7.6		1979MUa (105663)	904
$*****$										

C36H33N7015S3 H6L O-TREN SOX CAS 169209-69-2 (7370)
Tris-N-(2-aminoethyl-(8-hydroxyquinoline-5-sulphonato-7-carboxamido))amine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	NaClO ₄	25°C	0.10M	C		K1=21.40		2002BBd (106239)	905
Method: spectrophotometry using competitive reaction with edta, pH 7.2.										
Al+++	sp	NaClO ₄	25°C	0.01M	C				2001LAa (106240)	906

$$B(AlHL)=24.8$$

$$B(Al_2HL)=30.4$$

$$B(Al_3HL)=34.3$$

Medium: 0.01 M HClO₄. Method: spectrophotometric titration.

C37H44N2013S H6L MeThymol Blue (428)
3,3'-Bis(N,N-di(carboxymethyl)aminomethyl)thymolsulfonephthalein;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	25°C	0.10M	C				1997ASa (106583)	907
								K _{1eff} =6.84		
								K _{2eff} =6.61		

Medium: 0.10 M acetate buffer, pH 5.0.

Al+++	gl	NaClO ₄	25°C	0.10M	U				1982MYa (106584)	908
								B(AlHL)=29.14		
								B(Al2L)=26.8		

Al+++	sp	NaClO ₄	22°C	1.0M	U				1967LMg (106585)	909	
K(2Al+2H4L=AlH3L+AlH4L+H)=3.88(?),											
K(2Al+4H4L=Al(H3L) ₂ +Al(H4L) ₂ +2H)=6.02(?)											

C40H47N3010 H7L CAS 86728-01-0 (5503)
Bis(3-(((2-hydroxy-5-methylbenzyl)amino)methyl)-2-hydroxy-5-methylbenzyl)amine-triethanoic acid

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	gl	oth/un	25°C	0.10M	U			K ₁ =15.29	1983YMa (106785)	910
								K(AlH-1L+H)=6.19		
								K(AlH-2L+H)=8.23		
								K(AlH-3L+H)=10.02		
								K(AlL+H)=3.05		

Polymer Fulvic acid (1523)
Fulvic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	oth/un	22°C	?	U	H			1993LMd (108176)	911
								K _{1eff} =4.6 (type I site)		
								K _{2eff} =3.5 (type I site)		

pH=3.5; method: Synchronous scan fluorescence spectroscopy. Fulvic acid from northern coniferous forest. K=5.0 and 4.2 (type II site); 5.3 (type III)

Polymer L (3532)
Human transferrin;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Al+++	sp	KNO ₃	25°C	0.10M	C				1994HCa (108204)	912
								K _{eff} (Al+HC ₀₃ L)=13.72		
								K _{eff} (Al+AlHC ₀₃ L)=12.72		

		Keff(Al+L)=7.6
At pH 7.4 in 0.1M N-(2-hydroxyethyl)piperazine-N'-2-ethanesulfonic acid, (HEPES) and 5mM HCO ₃		

Al+++	sp oth/un 25°C 0.10M U	1990HSb (108205) 913
		Keff(Al+L)=13.5
		Keff(AlL+L)=12.5
Medium: 0.1 M N-(2-Hydroxyethyl)piperazine-N'-ethanesulfonic acid and 5 mM NaHCO ₃ , pH 7.4.		

Al+++	sp oth/un 25°C 0.10M U	1987MSc (108206) 914
		K _{1eff} =12.9
		K _{1eff} =12.3
Medium: 0.1 M Tris buffer, pH 7.4; 0.027 M HCO ₃ -.	By competition with the Al-NTA complex.	
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EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES
I Data with various BACKGROUNDS
H Data for THERMOCHEMICAL quantities
M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC

END