

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 78 experiments for

(no ligands specified)

2 metals : As(III), As(V)

(no references specified)

(no experimental details specified)

\*\*\*\*\*

e- HL Electron (442)  
Electron;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) EMF none 25°C 0.0 U T 1924SCa (312) 1

K=11.87(234 mV)

K: 0.5As<sub>2</sub>O<sub>3</sub>(s)+3H+3e=As(s)+1.5H<sub>2</sub>O. K=10.70(45 C;225 mV). K(HAsO<sub>2</sub>+3H+3e=As(s)+2H<sub>2</sub>O)=12.55(247.5 mV)

\*\*\*\*\*

Cl- HL Chloride CAS 7647-01-0 (50)  
Chloride;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) EMF NaCl 25°C 5.00M U I 1988PEa (4477) 2

B(As(OH)<sub>2</sub>Cl)=-2.34

B(As(OH)Cl<sub>2</sub>)=-5.2

-----  
As(III) sol none 25°C 0.0 U 1957ARb (4478) 3

I=0 corr. K(As(OH)<sub>3</sub>+H+L=As(OH)<sub>2</sub>L+H<sub>2</sub>O)=-1.07, K(As(OH)<sub>2</sub>L+H+L=As(OH)L<sub>2</sub>+H<sub>2</sub>O)=-3.47, K(As(OH)L<sub>2</sub>+H+Cl=AsL<sub>3</sub>+H<sub>2</sub>O)=-4.20. Also Kd values

-----  
As(III) sol oth/un 25°C var U 1940GHa (4479) 4

K(0.5As<sub>2</sub>O<sub>3</sub>(s)+3H+3L=AsL<sub>3</sub>+1.5H<sub>2</sub>O)=-10.5

\*\*\*\*\*

F- HL Fluoride CAS 7644-39-3 (201)  
Fluoride;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) sp NaClO<sub>4</sub> 22°C 1.00M U 1976IVa (6738) 5

K(As(OH)<sub>2</sub>+F)=3.51

K(As(OH)<sub>2</sub>+F)=3.52 (solubility)

\*\*\*\*\*

MoO<sub>4</sub>-- H<sub>2</sub>L Molybdate (443)  
Molybdate;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) g1 NaClO<sub>4</sub> 25°C 3.00M C 1975PEa (8712) 6

B(8,5,2)=60.92  
 B(10,6,2)=75.25  
 B(11,6,2)=80.73  
 B(12,6,2)=84.07

B(p,q,r): pH+qB+rC=HpBqCr; B=MOo4 2-; C=HAsO4-

\*\*\*\*\*

OH- HL Hydroxide (57)  
 Hydroxide;

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

As(III) sol oth/un 22°C C TIH 1999PBa (10998) 7

Ks(0.5AsO3+H2O=As(OH)3)=-3.08

Medium: 0-0.99 mole fraction H2O/HOAc. As2O3 is arsenolite. Data for 60 and 90C and H2O/acetone mixtures. K(As(OH)3+4H2O=As(OH)3.4H2O)=0.36

-----  
 As(III) sp NaClO4 22°C 1.00M U 1976IVa (10999) 8

K(As(OH)3+H=As(OH)2+H2O)=-1.17

\*Ks(As(OH)3(s)+H)=-0.82

\*\*\*\*\*

S-- H2L Sulfide CAS 7783-06-4 (705)  
 Sulfide;

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

As(III) sol oth/un 25°C 0.10M M T H 1992EAa (14307) 9

Ks(As2S3(s),H3AsO3)=-11.9

Ks(As2S3(s),H2As3S6)=-5.0

K(As2S3(s),H3ASO3)=K(.5As2S3(s)+3H2O=H3AsO3+1.5H2S);

K(As2S3(s),H2As3S6)=1.5As2S3(s)+1.5H2S=H2As3S6+H); 25-90 C

-----  
 As(III) oth oth/un 25°C 0.0 U 1990SAa (14308) 10

Ks(As2S3)=-219.14

Ks(As2S2(orpiment))=-180.43

Calculated from thermodynamic data. Ks(As2S3): As2S3(s)+20H2O=2HAsO4+3SO4+3OH. Ks(orpiment): As2S2(s)+16H2O=2HAsO4+2SO4+3OH.

-----  
 As(III) sol oth/un 25°C var M T 1990WEa (14309) 11

K(As2S3,H3AsO3)=23.11

K(As2S3,H2As3S6)=-3.61

K(As2S3,H3AsO3)=K(0.5As2S3(s)+3H2O=H3AsO3+1.5HS+1.5H);

K(As2S3,H2As3S6)=K(1.5As2S3(s)+1.5HS+0.5H=H2As3S6; Constants at I=0

-----  
 As(III) oth none 25°C dil C T 1989SRf (14310) 12

K(HAs3S6+H=H2As3S6)=6.56

Critical eval. of lit. data for the solubility of As2S3 in sulfide media.

Ks(As2S3+HS=0.67HAs3S6+0.33H)=-2.82; Ks(As2S3+HS+0.33H=0.67H2As3S6)=1.55

-----  
 As(III) gl NaCl 22°C 1.00M U 1977VIa (14311) 13

B(As3S6)=-134.40

B(As2S5)=-92.0

-----  
As(III) sol oth/un 0°C var U 1964PCa (14312) 14  
K(0.5As2L3(s)+0.5L=HAsL2)=1.0  
K(AsL2+H)=3.7  
K(0.5As2L3(s)+3H2O=1.5H2S(aq)+As(OH)3)=-12.6; K(0.5As2L3(s)+0.5H2L=HAsL2)=-5.3  
-----

As(III) gl KCl ? 1.0M U 1960ASc (14313) 15  
K(3As2L3(s)+3H2L=6H+2As3L6)=-33.19. K(2As2L3(s)+2H2O=3H+As3L6+HAsO2)=-27.43.  
K(As2L3(s)+4H2O=2HAsO2+3H2L)=-21.68  
-----

As(III) sol oth/un 20°C var U 1956BLa (14314) 16  
K(As2L3+2OH=AsL2+AsL(OH)2)=2.15. K(As2L3(s)+2L=AsL2+AsL3)=12.94  
\*\*\*\*\*  
SO4-- H2L Sulfate CAS 7664-93-9 (15)  
Sulfate;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
As(III) con mixed 25°C ? U 1963GRc (15998) 17  
K(AsO+HL)=1.08  
Medium: H2SO4. Also by freezing point  
-----

As(III) con mixed 25°C ? U 1961BGa (15999) 18  
K(As(HL)4+H)=2.8  
Medium: H2SO4  
\*\*\*\*\*  
CH4O L Methyl alcohol CAS 67-56-1 (597)  
Methanol; CH3.OH  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
As(III) EMF alc/w 20°C 100% U 1971GSa (17875) 19  
K(As+2L=As(L')2+2H) > 1  
K(As(L')2+L')=14.20  
K(2As(L')3+L'=As2(L')7)=4.74  
Medium: MeOH, 1 M Me4NCl. Method: H electrode. L'=H-1L  
\*\*\*\*\*  
C2H6O2 L Ethyleneglycol CAS 107-21-1 (924)  
1,2-Dihydroxyethane (Ethane-1,2-diol); HO.CH2.CH2.OH  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
As(III) gl oth/un 25°C 0.10M U 1957RLa (22134) 20  
K(As(OH)4+L)=-1.15  
\*\*\*\*\*  
C3H8O2 L Propyleneglycol CAS 57-55-6 (2025)  
Propan-1,2-diol; CH3.CH(OH).CH2(OH)  
-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	oth/un	25°C	0.10M	U				1957RLa (27666)	21
									$K(\text{As}(\text{OH})_4+\text{L})=-1.00$	

\*\*\*\*\*

C3H8O3		L	Glycerol						CAS 56-81-5 (2707)	
Propane-1,2,3-triol; HO.CH2.CH(OH).CH2.OH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	oth/un	25°C	0.10M	U				1957RLa (27716)	22
									$K(\text{As}(\text{OH})_4+\text{L})=0.06$	

\*\*\*\*\*

C4H6O6		H2L	L-Tartaric acid						CAS 87-69-4 (92)	
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	vlt	KCl	25°C	0.10M	U				1980ETa (31195)	23
									$K(\text{As}(\text{OH})_2+\text{L})=6.62$	

\*\*\*\*\*

C4H10O2		L							CAS 5341-95-7 (3575)	
meso-Butan-2,3-diol; CH3.CH(OH).CH(OH).CH3										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	oth/un	25°C	0.10M	U				1957RLa (34668)	24
									$K(\text{As}(\text{OH})_4+\text{L}=\text{As}(\text{OH})_2\text{H}-2\text{L})=-0.89$	

DL- or meso- not stated

\*\*\*\*\*

C4H10O3		L							CAS 623-39-2 (3577)	
3-Methoxypropan-1,2-diol; CH2(OH).CH(OH).CH2.OCH3										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	oth/un	25°C	0.10M	U				1957RLa (34705)	25
									$K(\text{As}(\text{OH})_4+\text{L}=\text{As}(\text{OH})_2\text{H}-2\text{L})=-0.18$	

\*\*\*\*\*

C5H10O4		L	Deoxy-Ribose						CAS 533-67-5 (7470)	
2-Deoxy-D-ribose, 2-Deoxy-D-erythro-pentose;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	KCl	25°C	0.10M	U				1979HUa (40325)	26
									$K(\text{H}_2\text{AsO}_3+\text{L})=2.89$	

\*\*\*\*\*

C5H10O5		L	D-Arabinose						CAS 10323-20-3 (3606)	
D-Arabinose;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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As(III) gl KCl 25°C 0.10M U 1960ATa (40332) 27  
K(As(OH)4+2L=As(H-2L)2)=1.28

\*\*\*\*\*  
C5H10O5 L D-Xylose CAS 58-86-6 (3607)  
D-Xylose;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) gl KCl 25°C 0.10M U 1959ATa (40360) 28  
K(As(OH)4+2L=As(H-2L)2)=0.74

\*\*\*\*\*  
C5H10O5 L L-Arabinose CAS 5328-37-0 (1616)  
L-Arabinose

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) gl KCl 25°C 0.10M U 1960ATa (40365) 29  
K(As(OH)4+2L=As(H-2L)2)=1.24

-----  
As(III) gl oth/un 25°C 0.10M U 1957RLa (40366) 30  
K(AsO(OH)2+H2L=AsOL)=0.20

\*\*\*\*\*  
C5H11NS2 HL CAS 147-84-2 (2126)  
Diethyldithiocarbamic acid; (CH3.CH2)2N.CSSH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) sp non-aq ? 100% U M 1968SRg (41346) 31  
K(AsAL+2HL=AsL3+H2A)=7.93

Medium: CCl4. H2A=dithizone.

\*\*\*\*\*  
C5H12O4 H2L Pentaerythritol CAS 115-77-5 (3028)  
Pentaerythritol; C(CH2.OH)4

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) gl KCl 25°C 0.10M U 1960ARA (41659) 32  
K(As(OH)4+L=As(H-2L)2)=0.94

-----  
As(III) gl oth/un 25°C 0.10M U 1957RLa (41660) 33  
K(As(OH)4+L=As(H-2L)2)=0.00

\*\*\*\*\*  
C6H3N3O7 HL Picric acid CAS 88-89-1 (593)  
2,4,6-Trinitrophenol; HO.C6H2(NO2)3

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) sol alc/w 25°C 50% C I K1=1.44 1983Bwb (42092) 34  
Kso((C6H5)4AsL)=-5.94

Method: spectrophotometry. Data for 20-100% MeOH/H2O

Cation is tetraphenylarsonium.

\*\*\*\*\*

C6H6O2 H2L Catechol CAS 120-80-9 (534)  
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) gl KCl 25°C 0.10M U 1959ARa (43724) 35  
 $K(As(OH)_4 + H_2L = As(OH)_2L) = 2.24$   
 $K(As(OH)_4 + 2H_2L = AsL_2) = 2.71$

-----  
As(III) gl oth/un 25°C 0.10M U 1957RLa (43725) 36  
 $K(AsO(OH)_2 + H_2L = AsOL) = 2.04$

\*\*\*\*\*

C6H6O3 H3L Pyrogallol CAS 87-66-1 (696)  
1,2,3-Trihydroxybenzene; C6H3(OH)3

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) gl KCl 25°C 0.10M U 1959ARa (43948) 37  
 $K(As(OH)_4 + H_3L = As(OH)_2HL) = 2.81$   
 $K(As(OH)_4 + 2H_3L = As(HL)_2) = 3.09$

\*\*\*\*\*

C6H6O8S2 H4L Tiron CAS 149-45-1 (104)  
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)2.C6H2(SO3H)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) gl KCl 25°C var U I 1964ATa (44405) 38  
 $K(H_3AsO_3 + 2H_2L = AsL_2 + H) = -8.186 + 9.162SQRTI / (1 + 0.553SQRTI) = -1.61I$

\*\*\*\*\*

C6H8O6 H2L Ascorbic acid CAS 50-81-7 (285)  
Ascorbic acid (Vitamin C);

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) vlt oth/un 25°C 0.10M U 1972ETa (45625) 39  
 $K(H + L + As(OH)_2) = 18.84$

Medium: Na2SO4

\*\*\*\*\*

C6H9NO6 H3L NTA CAS 139-13-9 (191)  
Nitriilotriethanoic acid; N(CH2.COOH)3

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(III) vlt oth/un 25°C 0.10M U 1973ETa (46702) 40  
 $K(As(OH)_2 + H + L) = 15.58$

Medium: Na2SO4. Using a glass electrode, K=15.33

\*\*\*\*\*

C6H1007 HL Glucuronic acid CAS 6556-12-3 (599)  
D-Glucuronic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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As(III) gl KCl 25°C 0.10M M K1=2.46 B2=2.79 1987PLb (48417) 41  
\*\*\*\*\*

C6H1205 L L-Rhamnose CAS 634-74-2 (3659)  
6-Deoxy-L-mannose;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
As(III) gl KCl 25°C 0.10M U 1960ATa (49505) 42  
K(As(OH)4+2L=As(H-2L))=0.68  
\*\*\*\*\*

C6H1206 L D-Fructose CAS 57-48-7 (1561)  
D-Fructose

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
As(III) gl KCl 45°C 0.10M U T H 1968APd (49535) 43  
K(As(OH)4+L=As(OH)2H-2L)=0.703  
K=0.779(15 C), 0.739(25 C), 0.724(35 C). DH=-4.3 kJ mol<sup>-1</sup>, DS=0  
-----

As(III) gl KCl 25°C 0.10M U 1960ATa (49536) 44  
K(As(OH)4+2L=As(H-2L)2)=1.08  
-----

As(III) gl oth/un 25°C ? U 1957RLa (49537) 45  
K(AsO(OH)2+H2L=AsOL)=0.77  
\*\*\*\*\*

C6H1206 L D-Galactose CAS 59-23-4 (1559)  
D-Galactose

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
As(III) gl oth/un 25°C 0.10M U 1957RLa (49557) 46  
K(AsO(OH)2+H2L=AsOL)=0.29  
\*\*\*\*\*

C6H1206 L D-Glucose CAS 492-62-6 (1560)  
D-Glucose

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
As(III) gl oth/un 25°C 0.10M U 1957RLa (49575) 47  
K(AsO(OH)2+H2L=AsOL)=0.16  
\*\*\*\*\*

C6H1206 L D-Mannose CAS 3458-28-4 (1562)  
D-Mannose

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
 As(III) gl KCl 25°C 0.10M U 1959ARa (49598) 48  
 $K(\text{As}(\text{OH})_4+\text{L}=\text{As}(\text{OH})_2\text{H}-2\text{L})=2.22$   
 $K(\text{As}(\text{OH})_4+2\text{L}=\text{As}(\text{H}-2\text{L})_2)=2.97$   
 -----

As(III) gl oth/un 25°C 0.10M U 1957RLa (49599) 49  
 $K(\text{AsO}(\text{OH})_2+\text{H}_2\text{L}=\text{AsOL})=0.36$   
 -----

\*\*\*\*\*  
 C6H12O6 L Sorbose CAS 87-79-6 (930)  
 L(-)-Sorbose;  
 -----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

As(III) gl KCl 25°C 0.10M U 1960ATa (49611) 50  
 $K(\text{As}(\text{OH})_4+2\text{L}=\text{As}(\text{H}-2\text{L})_2)=1.08$   
 -----

\*\*\*\*\*  
 C6H12O7 HL Gluconic acid CAS 526-95-4 (904)  
 D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH2(CHOH)4.COOH  
 -----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

As(III) gl KCl 25°C 0.10M M K1=1.60 B2=2.29 1987PLb (49698) 51  
 -----

\*\*\*\*\*  
 C6H14O6 L D-Mannitol CAS 69-65-8 (3664)  
 D-Mannitol;  
 -----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

As(III) gl KNO3 20°C 0.10M M 1980MBc (51067) 52  
 $K(\text{As}(\text{OH})_3+\text{H}_2\text{L}=\text{As}(\text{OH})\text{L})=0.20$   
 $K'(\text{As}(\text{OH})_3+\text{H}_2\text{L}=\text{As}(\text{OH})_2\text{L}+\text{H})=-8.20$ . For L=D-sorbitol, K=0.46, K'=-7.96;  
 L=D-dulcitol, K=0.30, K'=-8.30; L=D-glucose, K=-0.96, K'=-9.21.  
 -----

As(III) oth KCl 25°C 0.10M U 1970ATb (51068) 53  
 $K(\text{As}(\text{OH})_4+\text{L}=\text{As}(\text{OH})_2\text{H}-2\text{L})=1.07$   
 -----

Method: optical rotary dispersion  
 $K(\text{As}(\text{OH})_2\text{H}-2\text{L}+\text{H}=\text{As}(\text{OH})_2\text{H}-1\text{L})=8.44$ ,  $K(\text{AsH}(\text{OH})_4+\text{L}=\text{As}(\text{OH})_2\text{H}-1\text{L})=0.38$   
 -----

As(III) gl KCl 25°C 0.10M U 1959ARa (51069) 54  
 $K(\text{As}(\text{OH})_4+\text{L}=\text{As}(\text{OH})_2\text{H}-2\text{L})=0.85$   
 -----

\*\*\*\*\*  
 C7H6O3 H2L CAS 139-85-5 (881)  
 3,4-Dihydroxybenzaldehyde, protocatechuic aldehyde; C6H3(OH)2.CHO  
 -----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

As(III) gl KCl 25°C 0.10M U 1968A0a (54353) 55  
 $K(\text{As}(\text{OH})_4+\text{H}_2\text{L})=2.96$   
 -----

\*\*\*\*\*  
 C7H10N2 L CAS 1122-58-3 (492)  
 -----



4-(N,N-Dimethylamino)pyridine; C5H4N.N(CH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)		nmr non-aq	25°C	100%	U			K1=3.33	1992PWb (56629)	56
Medium: CDCl3; metal salt:EtN(CH2)2NEtAs+CF3SO3-; other data for adducts with other arsenium cations and Lewis bases.										
*****										
C8H10O2		L						CAS 7138-28-5	(3199)	
Phenylethane-1,2-diol; C6H5.CH(OH).CH2.OH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	oth/un	25°C	0.10M	U				1957RLa (60834)	57
K(AsO(OH)2+H2L)=-0.64										
*****										
C10H16N2O8		H4L		EDTA				CAS 60-00-4	(120)	
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	vlt	oth/un	20°C	0.10M	U				1972EVa (73593)	58
K(As(OH)2+H+L)=19.3										
-----										
As(III)	EMF	oth/un	24°C	0.10M	U				1972EVa (73594)	59
K(As(OH)2+H+L)=19.6										
Medium: 0.1 M Na2SO4										
*****										
C12H22O11		L		Turanose				CAS 547-25-1	(2701)	
3-O-D-Glucopyranosyl-D-fructose;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	KCl	25°C	0.10M	M			K1=0.80	1987PLb (82864)	60
*****										
C12H22O11		L		alpha-Lactose				CAS 5989-81-1	(2486)	
4-D-Beta-D-Galactopyranosyl-alpha-D-glucose;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	KCl	25°C	0.10M	M			K1=0.64	1987PLb (82870)	61
*****										
C12H22O11		L		Maltose				CAS 6363-53-7	(2705)	
4-O-alpha-D-Glucopyranosyl-D-glucose, Maltobiose;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	KCl	25°C	0.10M	M			K1=0.40	1987PLb (82877)	62
*****										
C12H22O11		L		Cellobiose				CAS 528-50-7	(2697)	

4-0-beta-D-Glucopyranosyl-D-glucose;

---

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	KCl	25°C	0.10M	M			K1=0.41	1987PLb (82884)	63
*****										
C12H22O11		L		Melibiose				CAS 66009-10-7	(2699)	
6-0-D-Galactopyranose-D-glucose;										

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	KCl	25°C	0.10M	M			K1=0.82	1987PLb (82888)	64
*****										
C12H24O11		L		Maltitol				CAS 585-88-6	(2709)	
4-0-alpha-D-Glucopyranosyl-D-glucitol;										

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	KCl	25°C	0.10M	M			K1=1.5	1988HLA (83681)	65
*****										
C12H27O4P		L						CAS 126-73-8	(2432)	
Tri-n-butyl phosphate; (C4H9O)3PO										

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	sp	oth/un	?	?	U	M			1973RGA (84118)	66
*****										
C14H22N2O8		H4L		CDTA				CAS 482-54-2	(200)	
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;										

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	vlt	KCl	25°C	0.10M	U				1980ETA (88586)	67
*****										
Polymer								K(As(OH)2+H+L)=20.67	(4200)	
Polyvinyl alcohol;										

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
As(III)	gl	oth/un	25°C	0.10M	U				1957RLA (108380)	68
*****										
e-		HL		Electron				K(As(OH)4+L=As(OH)2H-2L)=-0.15	(442)	
Electron;										

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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As(V) oth none 25°C 0.0 U 1952LAB (313) 69  
K=18.9(559 mV)

K:  $\text{H3AsO4} + 2\text{H} + 2\text{e} = \text{HAsO2} + 2\text{H2O}$ . From thermodynamic data.  $\text{K}(\text{AsO4} + 2\text{H2O} + 2\text{e} = \text{AsO2} + 4\text{OH}) = -22.9(-670 \text{ mV})$ .  $\text{K}(\text{As(s)} + 3\text{H} + 3\text{e} = \text{AsH3(g)}) = -30.8(-600 \text{ mV})$

\*\*\*\*\*

ClO4- HL Perchlorate CAS 7001-90-3 (287)  
Perchlorate;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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As(V) cal oth/un 25°C dil U H 1972CJa (6147) 70  
Kso(Ph4AsL(s)=Ph4As+L)=-8

DH(Kso)=45.6 kJ mol<sup>-1</sup>, DS(Kso)=0. Kso=-8.4 to -7.9

\*\*\*\*\*

F- HL Fluoride CAS 7644-39-3 (201)  
Fluoride;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

As(V) kin oth/un 65°C 9.0M U 1969LJb (6739) 71  
K(HAsF6+H2O=HAsF5OH+HF)=-0.9

medium:9-15 M H2SO4

-----  
As(V) EMF KCl ? 1.0M U 1961DGa (6740) 72  
K(H3AsO4+HF=AsO3F+2H+H2O)=-6.2

K(H2AsO4+F=AsO3F+H2O)=-0.75

\*\*\*\*\*

MoO4-- H2L Molybdate (443)  
Molybdate;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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As(V) gl NaClO4 25°C 0.63M U 1987BKb (8713) 73  
K(14H+AsO4+9MoO4=H3AsMo9O34)=100.43; K(15H+AsO4+9MoO4=H4AsMo9O34)=103.43  
K(11H+2AsO4+6MoO4=HAs2Mo6O26)=79.65. Other data also given.

-----  
As(V) ISE NaCl ? 2.00M U 1973COa (8714) 74  
K'=69.7

K(H+As2Mo6O26(6-))=ca.5.9

K(H+HAs2Mo6O26(5-))=2.6

K': 2HAsO4-- + 6L + 10H=As2Mo6O26(6-) + 6H2O. In 1 M LiCl at 3 C:

K(H+A)=5; K(H+HA)=3.9; K(H+H2A)=2.9; K(H+H3A)=1.9 where A=As4Mo12O50(8-)

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C6H6O2 H2L Catechol CAS 120-80-9 (534)  
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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As(V) gl KNO3 20°C 0.10M U 1977VBb (43726) 75  
K(H2AsO4+2H2L=As(OH)2L2)=0.81



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#### EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES  
I Data with various BACKGROUNDS  
H Data for THERMOCHEMICAL quantities  
M Data for TERNARY Complexes

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END