

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 247 experiments for

(no ligands specified)

Metal : B(III)

(no references specified)

(no experimental details specified)

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e- HL Electron (442)  
Electron;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) oth none 25°C 0.0 U 1952LAB (360) 1  
K=-44.1(-870 mV)

K: B(OH)3+3H+3e=B(s)+3H2O. From thermodynamic data

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BrO- HL Hypobromite (870)  
Hypobromite;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) sp NaCl 25°C 0.50M U 1987BBa (2389) 2  
K(B(OH)4+HL=B(OH)3L)=1.83

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CO3-- H2L Carbonate CAS 465-79-6 (268)  
Carbonate;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) sp NaCl 25°C 0.70M C K1=11.44 1998MBa (3144) 3  
K(B(OH)3+HCO3=B(OH)2CO3)=2.6

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Cl- HL Chloride CAS 7647-01-0 (50)  
Chloride;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) con non-aq 20°C 100% U 1960BGF (4510) 4  
K(BCl3POCl3=POCl2+BCl4)=-6.7

Medium: POCl3(liquid)

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ClO- HL Hypochlorite CAS 7790-92-3 (869)  
Hypochlorite;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) sp NaCl 25°C 0.50M U 1987BBa (5993) 5  
K(B(OH)4+HL=B(OH)3L)=2.25

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F- HL Fluoride CAS 7644-39-3 (201)  
Fluoride;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) gl oth/un 25°C 0.20M U 1979MMd (6741) 6

K3'=6.43

K4'=2.45

K3': 3HF+H3BO3=H+B(OH)F3+2H2O, K4': 4HF+H3BO3=H+BF4+3H2O.

-----  
B(III) ISE NaCl 25°C 1.0M U 1973MPb (6742) 7

K(B(OH)3+F)=-0.36

K(B(OH)3+2F+H=BF2(OH)2)=7.06

K(B(OH)3+3F+2H=BF3OH)=13.69

Kn(B(OH)4+nF=B(OH)(4-n)Fn+nOH)=-5.3(n=1); -11.6(n=2); -18.7(n=3); -27.1(n=4)

-----  
B(III) ISE NaNO3 25°C 1.0M U T H 1971GHg (6743) 8

K(B(OH)3+F)=-0.30

K(B(OH)3+2F=BF2(OH)2+OH)=-6.27

K(B(OH)3+3F=BF3OH+2OH)=-14.23

K(B(OH)3+4F=BF4+3OH)=-21.6

DH(K4)=147.7 kJ mol<sup>-1</sup>, DS=313.8 J K<sup>-1</sup> mol<sup>-1</sup>. At 35 C: values are -0.27, -6.2, -13.4, -20.8

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B(III) nmr non-aq -61°C 100% U H 1965BPa (6744) 9

K(BF4+BF3=B2F7)=2.68

K(B2F7+BF3=B3F10)=0.32

Other methods:partial pressure BF3,infrared spectra. Medium: CH2Cl2

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B(III) con non-aq 20°C 100% U 1961CKa (6745) 10

K4=-2.89

Medium: liquid HF, I=0 corr

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B(III) sol non-aq 0°C 100% U M 1961CKa (6746) 11

K(AgBF4(s)=Ag+BF4)=-2.53

K4=2.11

Medium: liquid HF, I=0 corr.

-----  
B(III) ISE oth/un 25°C var U 1959RDa (6747) 12

Ks=-22.85

Method: H, Pb and quinhydrone electrode. Ks: KBF4(s)+3H2O=B(OH)3(s)+2H+4F+K  
By solubility K(B(OH)3(s)+3H+4F=BF4+3H2O)=20.0

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B(III) dis non-aq 0°C 100% U 1958MHb (6748) 13

K4=6.6

Medium: liquid HF

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B(III) sol none 25°C 0.0 U T H 1958RKb (6749) 14

Ks(KBF4(s)=K+BF4)=-2.86

$K_s = -3.79(0\text{ C}), -2.11(50\text{ C}), -1.54(70\text{ C})$ .  $DH(K_s) = 59.0\text{ kJ mol}^{-1}$ .  $K_s(\text{CsBF}_4(s)) = -3.35(0\text{ C}), -2.37(25\text{ C}), -1.25(60\text{ C})$ .  $DH = 60$

-----  
 B(III) sol none 25°C 0.0 U T HM 1958Rkb (6750) 15  
 $K(\text{CsBF}_4(s) = \text{Cs} + \text{BF}_4) = -2.37$   
 I=0 corr.  $K_s = -3.35(0\text{ C}), -1.25(60\text{ C})$ .  $DH(K_s) = 59\text{ kJ mol}^{-1}(25\text{ C})$   
 -----

B(III) EMF oth/un 15°C var U 1955RUa (6751) 16  
 $K(\text{BF}_2(\text{OH})_2 + \text{HF} = \text{BF}_3\text{OH} + \text{H}_2\text{O}) = 3.57$   
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B(III) kin oth/un 25°C var U 1951WAa (6752) 17  
 $K = 1.96$   
 $K' = 2.64$   
 K:  $\text{BF}_2(\text{OH})_2 + \text{HF} = \text{BF}_3\text{OH} + \text{H}_2\text{O}$ .  $K'$ :  $\text{BF}_3\text{OH} + \text{HF} = \text{BF}_4 + \text{H}_2\text{O}$   
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B(III) oth oth/un 20°C 0.02M U T 1948RSa (6753) 18  
 $K(\text{BF}_3\text{OH} + \text{HF} = \text{H}_2\text{O} + \text{BF}_4) = 2.57$   
 Method: chemical analysis, Medium: HBF<sub>4</sub>.  $K = 2.32(60, 75\text{ C}), 2.14(90\text{ C})$   
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B(III) oth oth/un 25°C 0.0 U 1948WAa (6754) 19  
 $K(\text{BF}_3\text{OH} + \text{HF} = \text{H}_2\text{O} + \text{BF}_4) = 2.64$   
 Methods: chemical analysis, kinetics  
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B(III) oth oth/un 25°C var U T H 1946RYa (6755) 20  
 $K(\text{BF}_3\text{OH} + \text{HF} = \text{BF}_4 + \text{H}_2\text{O}) = 2.55$   
 Method: chemical analysis.  $K = 2.26(80\text{ C}), 2.14(100\text{ C})$ .  $DH(K) = -13.5\text{ kJ mol}^{-1}$ ,  
 $DS = 4.6\text{ J K}^{-1}\text{ mol}^{-1}(25\text{C})$   
 -----

B(III) EMF none 18°C 0.0 U 1936RBa (6756) 21  
 $K(\text{BF}_4 + 3\text{H}_2\text{O} = \text{B}(\text{OH})_3 + 3\text{H} + 4\text{F}) = -19.4$   
 \*\*\*\*\*  
 H<sub>2</sub>O L Water CAS 7732-18-5 (6115)  
 Water  
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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) nmr non-aq 36°C 100% U 1971CBc (7586) 22  
 $K(\text{BF}_4 + \text{L}) = -0.5$   
 $K(\text{B}(\text{Ph})_4 + \text{L}) = -1$   
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NH<sub>3</sub> L Ammonia CAS 7664-41-7 (414)  
 Ammonia  
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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) gl oth/un 23°C var U 1965RPa (9095) 23  
 $K(\text{H}_2\text{NBF}_3 + \text{H}) = 12$   
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NH<sub>3</sub> L Hydroxylamine; CAS 5470-11-1 (1808)  
 Hydroxylamine; NH<sub>2</sub>.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	gl	KCl	10°C	0.04M	U			K(F3BONH2+H)=7.52	1965RIa (9259)	24
*****										
NO2- Nitrite;			HL	Nitrite				CAS 7782-77-6	(635)	
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	oth	none	25°C	0.0	C			K(H3BO3+NO2=B(OH)3NO2)=-0.49	1991MIa (9356)	25
By pH titration, K=-0.61.										
*****										
OH- Hydroxide;			HL	Hydroxide				(57)		
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	gl	NaClO4	25°C	0.10M	C			K(B(OH)3=B(OH)4+H)=-9.02 K(B(OH)3=B3O3(OH)4+H)=-7.20	2000KAa (11012)	26
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B(III)	nmr	KCl	21°C	0.10M	U	M		K(BAL2+L)=-8.7 K(BCL2+L)=-7.9 K(BDL2+E)=0.748	1994LGA (11013)	27
Method:NMR. A:C6H4F, C:C6H3ClF; D:C6H4F; E:4-(hydroxymethyl)imidazole										
*****										
O2-- Peroxide; -0.0-			H2L	Peroxide				CAS 7772-84-1	(2813)	
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	gl	KNO3	25°C	0.10M	U			K(B(OH)3+H2L=B(OH)3HL+H)=-7.7 K(B(OH)3HL+H2L=B(OH)2(HL)2)=0.3 K(B(OH)3+H2L=B(OH)2HL)=-2	1987PTa (12648)	28
-----										
B(III)	gl	none	25°C	0.0	U			K(Bi(OH)4+H2L=Bi(OH)3HL)=1.32 I=0 corr. K(Bi(OH)3HL+H2L=Bi(OH)2(HL)2+H2O)=0.21	1956ANb (12649)	29
-----										
B(III)	vlt	KNO3	25°C	0.50M	U			K(Bi(OH)4+H2L=Bi(OH)3HL)=1.52	1955KEb (12650)	30
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B(III)	gl	oth/un	25°C	0.20M	U			K(Bi(OH)4+H2L=Bi(OH)3HL)=1.5	1953EDb (12651)	31
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B(III) dis oth/un 18°C var U T 1923MEa (12652) 32  
K(Bi(OH)<sub>4</sub>+H<sub>2</sub>L=Bi(OH)<sub>3</sub>HL)=1.48

K=1.62(θ C)

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S04-- H2L Sulfate CAS 7664-93-9 (15)  
Sulfate;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) con mixed 25°C ? U 1961BGa (16000) 33  
K(B(HL)<sub>4</sub>+H)=0.85

medium: H<sub>2</sub>SO<sub>4</sub>. K(average)=0.7

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CH<sub>4</sub>O L Methyl alcohol CAS 67-56-1 (597)  
Methanol; CH<sub>3</sub>.OH

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) EMF alc/w 20°C 100% U 1964GUa (17876) 34  
K(B(H-1L)<sub>3</sub>+H-1L)=5.62  
K(B(H-1L)<sub>4</sub>+H=B(H-1L)<sub>3</sub>+L)=10.98

Method: H electrode. Medium: MeOH, 1.0 M Me<sub>4</sub>NC<sub>1</sub>

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C<sub>2</sub>H<sub>2</sub>O<sub>4</sub> H2L Oxalic acid CAS 144-62-7 (24)  
Ethanedioic acid; (COOH)<sub>2</sub>

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) nmr KNO<sub>3</sub> 25°C 0.10M U H 1994PRb (18802) 35  
K(B(OH)<sub>4</sub>+H<sub>2</sub>L=B(OH)<sub>2</sub>O<sub>2</sub>L)=8.20

DH=-46 kJ mol<sup>-1</sup>, DS=4 J mol<sup>-1</sup> K<sup>-1</sup>

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B(III) gl KNO<sub>3</sub> 21°C 0.10M U 1977RBb (18803) 36  
K(H<sub>3</sub>BO<sub>3</sub>+HL=B(OH)<sub>2</sub>L+H<sub>2</sub>O)=0.35

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B(III) gl KNO<sub>3</sub> 25°C 0.10M C 1975FPb (18804) 37  
K(Ph(B(OH)<sub>2</sub>+H<sub>2</sub>L=PhB(OH)L+H)=0.51. Metal is phenylboronic acid.

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C<sub>2</sub>H<sub>4</sub>O<sub>2</sub> HL Acetic acid CAS 64-19-7 (36)  
Ethanoic acid; CH<sub>3</sub>.COOH

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) gl none 25°C 0.0 M 1991MIa (19897) 38  
B(H<sub>3</sub>BO<sub>3</sub>+L=B(OH)<sub>3</sub>L)=-0.42

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C<sub>2</sub>H<sub>4</sub>O<sub>3</sub> HL Glycolic acid CAS 79-14-1 (33)  
2-Hydroxyethanoic acid; HO.CH<sub>2</sub>.COOH

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

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B(III) nmr KNO3 25°C 0.10M U H 1994PRb (20492) 39  
K(B(OH)4+HL=B(OH)2O2H-1L)=5.11  
K(B(OH)2O2H-1L+HL=B(OH)4(H-1L)2+4H)=1.0  
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B(III) gl KNO3 21°C 0.09M U I 1977RBb (20493) 40  
K(H3BO3+L=B(OH)2H-1L+H2O)=0.17  
In 0.21 M NaNO3 K(H3BO3+L=B(OH)2H-1L+H2O)=0.54  
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C2H6O2 L Ethyleneglycol CAS 107-21-1 (924)  
1,2-Dihydroxyethane (Ethane-1,2-diol); HO.CH2.CH2.OH  
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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) nmr KNO3 25°C 0.10M U HM 1994PRb (22135) 41  
K(B(OH)4+L=B(OH)2O2H-2L)=0.15  
K(B(OH)2O2H-2L+L=B(OH)4(H-2L)2+4H)=-0.74, K(B(OH)2O2H-1A+L=B(OH)4H-1AH-2L)=-1.19  
A=2-Hydroxypropanoic acid (lactic acid)  
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B(III) cal NaNO3 25°C 1.0M C H 1985ARb (22136) 42  
DH(B(OH)4+L)=-5.8 kJ mol<sup>-1</sup>, DS=-15 J K<sup>-1</sup> mol<sup>-1</sup>.  
-----

B(III) gl oth/un 35°C .025M U T H 1967CBc (22137) 43  
K'(B(OH)4+L)=0.27  
K''(B(OH)4+2L)=-0.05  
Medium:0.025 M borax. K'=0.52(0 C),0.46(13 C),0.33(25 C); DH=-11.3 kJ mol<sup>-1</sup>  
DS=-33.4 J K<sup>-1</sup> mol<sup>-1</sup>; K''=0.14(0 C),0.08(13 C),0.06(25 C), DH=-8.36, DS=-25.1  
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B(III) gl KCl 25°C var U I 1967NEb (22138) 44  
K(B(OH)4+2L=B(H-2L)2)=-0.007+1.334(SQRT I)  
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B(III) gl oth/un 25°C 0.10M U 1957RLa (22139) 45  
K(B(OH)4+L)=0.27  
K(B(OH)4+2L)=-1.0  
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C3H4O4 H2L Malonic acid CAS 141-82-2 (79)  
Propanedioic acid; CH2(COOH)2  
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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) gl KNO3 25°C 0.10M C M 1976LPa (24390) 46  
K(PhB(OH)2+H2L=PhB(OH)L+H)=-1.59. PhB(OH)2 is phenylboronic acid.  
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C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)  
L-2-Hydroxypropanoic acid; CH3.CH(OH).COOH  
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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) nmr KNO3 25°C 0.10M U HM 1994PRb (25401) 47  
K(B(OH)4+HL=B(OH)2O2H-1L)=5.86  
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$K(B(OH)2O2H-1L+HL=B04(H-1L)2+4H)=1.79$ ,  $K(B(OH)2O2H-2A+HL=B04H-2AH-1L+4H)=4.5$   
 $A=1,2$ -Dihydroxyethane.  $K(B(OH)2O2H-2B+HL=B04H-2BH-1L)=4.9$ .  $B=Propan-1,2$ -diol

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 B(III) gl KNO3 25°C 0.10M U 1984PSb (25402) 48  
 $K(H3BO3+H2L=(HO)2BHL+H)=-2.74$   
 $K((HO)2BHL+H2L=BL2+2H2O)=1.82$

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 B(III) gl KNO3 25°C 0.10M U 1984PSd (25403) 49  
 $K(B(OH)3+L=B(OH)2H-2L+H+H2O)=-2.75$   
 $K(B(OH)2H-2L+L=B(H-2L)2+2H2O)=1.82$

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 B(III) sp NaCl ? 3.00M U 1970LNc (25404) 50  
 $K(B(OH)3+HL=B(OH)2L)=0.40$   
 $K(B(OH)3+L=B(OH)2H(-1)L)=0.78$   
 $K(B(OH)3+2L=B(H-1L)2+OH)=0.78$

Method: infrared spectra

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C3H8O2 L Propyleneglycol CAS 57-55-6 (2025)  
 Propan-1,2-diol; CH3.CH(OH).CH2(OH)

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 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

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 B(III) nmr KNO3 25°C 0.10M U HM 1994PRb (27667) 51  
 $K(B(OH)4+L=B(OH)2O2H-2L)=0.45$   
 $K(B(OH)2O2H-2L+L=B04(H-2L)2+4H)=-0.09$ ,  $K(B(OH)2O2H-1A+L=B04H-1AH-2L)=-0.49$   
 $A=2$ -Hydroxypropanoic acid (lactic acid)

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 B(III) cal NaNO3 25°C 1.0M C H 1985ARb (27668) 52  
 $DH(B(OH)4+L)=-9.3$  kJ mol<sup>-1</sup>,  $DS=-22$  J K<sup>-1</sup> mol<sup>-1</sup>.  $DH(B(OH)4L+L)=-38.9$ ,  
 $DS=-138$ .

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 B(III) gl oth/un 35°C 0.02M U T H 1967CBd (27669) 53  
 $K(B(OH)4+L)=0.53$   
 $K'(B(OH)4+2L)=0.37$   
 Med.:0.025 borax. $K=0.8(0$  C),0.64(13 C),0.61(25 C);  $K'=0.92(0$  C),0.78(13 C),  
 0.59(25 C).  $DH(K)=-12.5$  kJ mol<sup>-1</sup>,  $DS=-29.3$  J K<sup>-1</sup> mol<sup>-1</sup>;  $DH(K')=-30$ ,  $DS=-92$

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 B(III) gl oth/un 25°C 0.10M U 1957RLa (27670) 54  
 $K(B(OH)4+L)=0.49$   
 $K(B(OH)4+2L)=0.21$

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C3H8O2 L Dihydroxypropan CAS 504-63-2 (130)  
 Propane-1,3-diol; HO.CH2.CH2.CH2.OH

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 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

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 B(III) gl oth/un 35°C 0.02M U T H 1967CBd (27691) 55  
 $K(B(OH)4+L)=-0.02$   
 $K'(B(OH)4+2L)=-1.25$   
 Med.:0.025 borax. $K=0.45(0$  C),0.25(13 C),0.1(25 C);  $K'=-0.7(0$  C),-0.92(13 C),

-0.96(25 C). DH(K)=-19.2 kJ mol<sup>-1</sup>, DS=-62.7 J K<sup>-1</sup>, mol<sup>-1</sup>, DH(K')=-25.9, DS=-66.8  
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C3H8O2S HL 1-Thioglycerol CAS 96-27-5 (1848)  
3-Mercapto-1,2-propanediol HS.CH2.CH(OH).CH2.OH

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) gl KCl 25°C 0.10M U 1964ATb (27709) 56  
K(H3BO3+HL=B(OH)2(H-1L)+H)=-7.79; K(H3BO3+2HL=B(H-1L)2+H)=-6.12

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C3H8O3 L Glycerol CAS 56-81-5 (2707)  
Propane-1,2,3-triol; HO.CH2.CH(OH).CH2.OH

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) gl KCl 25°C 0.10M M K1=1.39 1986LHa (27717) 57

B(III) cal NaNO3 25°C 1.0M C H 1985ARb (27718) 58  
DH(B(OH)4+L)=-10.2 kJ mol<sup>-1</sup>, DS=-9.6 J K<sup>-1</sup> mol<sup>-1</sup>. DH(B(OH)4L+L)=-28.9,  
DS=-92.

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B(III) gl oth/un 35°C 0.02M U T H 1967CBd (27719) 59

K(B(OH)4+L)=1.10

K'(B(OH)4+2L)=1.62

Med.:0.025 borax. K=1.36(0 C),1.24(13 C),1.15(25 C); K'=1.99(0 C),1.84(13 C)  
,1.76(25 C). DH(K)=-12.1 kJ mol<sup>-1</sup>, DS=-16.7 J K<sup>-1</sup> mol<sup>-1</sup>, DH(K')=-16.3, DS=-20.9

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B(III) gl KCl 25°C var U 1967NEb (27720) 60

K(B(OH)4+2L)=1.584+0.730sqrtI

-----  
B(III) gl oth/un 25°C 0.10M U 1957RLa (27721) 61

K(B(OH)4+L)=1.21

K(B(OH)4+2L)=1.62

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B(III) oth KCl 25°C 0.10M U 1956ANa (27722) 62

K(B(OH)4+L)=1.56

K(B(OH)4+2L)=1.91

Method: quinhydrone electrode.

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C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)  
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) nmr KNO3 25°C 1.50M U 1994PRa (31196) 63

Keff(B(OH)4+L=B(OH)2O2H-1L+4H)=0.7, Keff(B(OH)2O2H-1L+L=B04(H-1L)2+4H)=-0.92  
At pH 11.5

-----  
B(III) gl NaNO3 21°C 0.10M U I 1977RBb (31197) 64

K(H3BO3+L=B(OH)2H-1L+H2O)=0.92



In 0.2 M NaNO3  $K(H_3BO_3+L=B(OH)_2H-1L+H_2O)=0.65$

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B(III) oth oth/un 25°C ? U 1969KPa (31198) 65  
 $K(B(OH)_3+L)=0.70$   
 $K(B(OH)_3+HL)=1.18$

Method: optical rotatory dispersion

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B(III) sol oth/un 22°C ? U 1967SBg (31199) 66  
 $K(B(OH)_4+HL=BOL+OH)=6.97$   
 $K(2B(OH)_4+HL=B2O3L+OH)=14.07$

-----  
B(III) gl oth/un 20°C ? U 1965FSa (31200) 67  
 $K(B(OH)_3+H_2L=BH-1L)=0.77$   
 $K(B(OH)_3+HL=BH-2L)=1.60$   
 $K(B(OH)_3+L=BH-2LOH)=0.61$   
 $K(B(OH)_4+L=BH-2L(OH)_2)=0.77$

\*\*\*\*\*  
C4H6O6 H2L meso-Tartaric CAS 147-73-9 (91)  
meso-2,3-Dihydroxybutanedioic acid;  $HOOC.CH(OH).CH(OH).COOH$

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) nmr KNO3 25°C 1.50M U 1994PRa (31425) 68  
 $K_{eff}(B(OH)_4+L=B(OH)_2O_2H-1L+4H)=0.15$ ,  $K_{eff}(B(OH)_2O_2H-1L+L=B_2O_4(H-1L)_2+4H)=-0.96$ . At pH 11.5

\*\*\*\*\*  
C4H10O2 L Butane-2,3-diol CAS 513-85-9 (3576)  
Butane-2,3-diol;  $CH_3.CH(OH).CH(OH).CH_3$

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) gl oth/un 35°C .025M U T H 1967CBd (34666) 69  
 $K(B(OH)_4+L=B(OH)_2H-2L)=1.40$   
 $K'(B(OH)_4+2L=B(H-2L)_2)=2.10$   
Medium:borax.  $K=1.79(0\text{ C}), 1.63(13\text{ C}), 1.57(25\text{ C})$ ;  $DH=-18.0\text{ kJ mol}^{-1}$ ,  $DS=-29.3\text{ J K}^{-1}\text{ mol}^{-1}$ ;  $K'=2.60(0\text{ C}), 2.45(13\text{ C}), 2.21(25\text{ C})$ ;  $DH=-23.0$ ,  $DS=-33$

\*\*\*\*\*  
C4H10O2 L CAS 5341-95-7 (3575)  
meso-Butane-2,3-diol;  $CH_3.CH(OH).CH(OH).CH_3$

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) gl oth/un 35°C .025M U T H 1967CBd (34669) 70  
 $K(B(OH)_4+L=B(OH)_2H-2L)=0.36$   
 $K'(B(OH)_4+2L=B(H-2L)_2)=0.43$   
Medium:borax.  $K=0.71(0\text{ C}), 0.51(13\text{ C}), 0.43(25\text{ C})$ ;  $DH=-14.2\text{ kJ mol}^{-1}$ ,  $DS=-42\text{ J K}^{-1}\text{ mol}^{-1}$ ;  $K'=1.11(0\text{ C}), 0.88(13\text{ C}), 0.66(25\text{ C})$ ;  $DH=-30.1$ ,  $DS=-88$

-----  
B(III) gl oth/un 25°C 0.10M U 1957RLa (34670) 71  
 $K(B(OH)_4+L=B(OH)_2H-2L)=0.54$

$$K(B(OH)_4+2L=B(H-2L)_2)=0.69$$

DL- or meso- not stated

\*\*\*\*\*

C4H10O3 L CAS 623-39-2 (3577)

3-Methoxypropan-1,2-diol; CH<sub>2</sub>(OH).CH(OH).CH<sub>2</sub>.OCH<sub>3</sub>

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
B(III) gl oth/un 25°C 0.10M U 1957RLa (34706) 72

$$K(B(OH)_4+L=B(OH)_2H-2L)=1.28$$

$$K(B(OH)_4+2L=B(H-2L)_2)=1.13$$

\*\*\*\*\*

C4H10O4 L Erythritol CAS 149-32-6 (2706)

1,2,3,4-Tetrahydroxybutane; HO.CH<sub>2</sub>.CH(OH).CH(OH).CH<sub>2</sub>.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
B(III) gl KCl 25°C 0.10M M 1987VHa (34710) 73

$$K(B(OH)_4+L)=1.85$$

$$K(B(OH)_4+2L)=2.91$$

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B(III) gl KCl 25°C 0.10M M K1=1.99 1986LHa (34711) 74

\*\*\*\*\*

C5H10O2 L CAS 5057-98-7 (3605)

cis-Cyclopentane-1,2-diol; C5H8(OH)<sub>2</sub>

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
B(III) gl oth/un 35°C .025M U T H 1967CBd (40221) 75

$$K(B(OH)_4+L=B(OH)_2H-2L)=1.32$$

$$K'(B(OH)_4+2L=B(H-2L)_2)=2.01$$

Medium:borax. K=1.65(0 C),1.49(13 C),1.42(25 C); DH=-14.6 kJ mol<sup>-1</sup>, DS=-21

J K-1 mol<sup>-1</sup>; K'=2.56(0 C),2.36(13 C),2.15(25 C); DH=-25.5,DS=-46

\*\*\*\*\*

C5H10O4 L Deoxy-Ribose CAS 533-67-5 (7470)

2-Deoxy-D-ribose, 2-Deoxy-D-erythro-pentose;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
B(III) gl KCl 25°C 0.10M U 1979HUa (40326) 76

$$K(H_2BO_3+L)=3.85$$

\*\*\*\*\*

C5H10O5 L D-Arabinose CAS 10323-20-3 (3606)

D-Arabinose;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
B(III) gl none 25°C 0.0 M K1=2.19 B2= 3.02 1979EMb (40333) 77

Metal is borate.

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B(III) gl KCl 25°C 0.10M U 1959ATa (40334) 78  
K(B(OH)4+2L=B(H-2L)2)=3.28

\*\*\*\*\*  
C5H10O5 L D-Xylose CAS 58-86-6 (3607)  
D-Xylose;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) gl KCl 25°C 0.10M U 1959ATa (40361) 79  
K(B(OH)4+2L=B(H-2L)2)=4.01

\*\*\*\*\*  
C5H10O5 L L-Arabinose CAS 5328-37-0 (1616)  
L-Arabinose

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) gl KCl 25°C 0.10M M 1987VHa (40367) 80  
K(B(OH)4+L)=2.14  
K(B(OH)4+2L)=2.99

-----  
B(III) gl KCl 25°C 0.10M U 1959ATa (40368) 81  
K(B(OH)4+2L=B(H-2L)2)=3.55

-----  
B(III) gl oth/un 25°C 0.10M U 1957RLa (40369) 82  
K(BO(OH)2+H2L=BOL)=2.11  
K(BO(OH)2+2H2L=BL2)=2.83

\*\*\*\*\*  
C5H12O2 L CAS 5396-58-7 (3611)  
2-Methylbutane-2,3-diol; CH3.C(OH)(CH3).CH(OH).CH3

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) gl oth/un 35°C .025M U T H 1967CBd (41642) 83  
K(B(OH)4+L=B(OH)2H-2L)=1.11  
K'(B(OH)4+2L=B(H-2L)2)=2.09

Medium:borax. K=1.59(0 C),1.38(13 C),1.26(25 C); DH=-20.5 kJ mol-1, DS=-46  
J K-1 mol-1; K'=2.76(0 C),2.53(13 C),2.32(25 C); DH=-33.4, DS=-67  
\*\*\*\*\*

C5H12O2 L CAS 625-69-4 (7147)  
Pentane-2,4-diol; CH3CH(OH)CH2CH(OH)CH3

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) EMF KCl 25°C 0.10M C I K1=0.50 1995BVa (41645) 84  
K(BH+L)=-0.06

In CHCl3: K(BH+L)=-0.097; CH2Cl2: K(BH+L)=-0.081. In C6H6: K(BH+L)=-0.131;  
In CCl4: K(BH+L)=-0.086. In BuOBu: K(BH+L)=-0.097. In DIBK:K(BH+L)=-0.08  
\*\*\*\*\*

C5H12O4 H2L Pentaerythritol CAS 115-77-5 (3028)  
Pentaerythritol; C(CH2.OH)4



$$K(B(OH)_3+H_2L=B(OH)_2+H)=-3.46$$

-----  
 B(III) vlt oth/un 25°C 0.10M U 1972HKd (42857) 92  
 $K(B(OH)_4+H_2L=B(OH)_2L)=3.56$

\*\*\*\*\*

C6H5NO4 H2L 4-Nitrocatechol CAS 3316-09-4 (890)  
 1,2-Dihydroxy-4-nitrobenzene; O2N.C6H3(OH)2

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
 B(III) gl KNO3 25°C 0.10M C 1977PBc (42914) 93

$$K(B(OH)_3+H_2L=B(OH)_2L+H)=-3.02$$

$K(PhB(OH)_2+H_2L=PhB(OH)L+H)=-3.82$ . PhB(OH)2 is phenylboronic acid.

-----  
 B(III) sp KCl 25°C 0.10M U 1972HKa (42915) 94

$$K(B(OH)_3+H_2L=B(OH)_2L+H)=-3.76$$

-----  
 B(III) vlt oth/un 25°C 0.10M U 1972HKd (42916) 95

$$K(B(OH)_4+H_2L=B(OH)_2L)=3.96$$

-----  
 B(III) gl KNO3 20°C 0.10M U 1968BHb (42917) 96

$$K(H_3BO_3+H_2L=BL(OH)_2+H)=-4.0$$

\*\*\*\*\*

C6H6O2 H2L Catechol CAS 120-80-9 (534)  
 1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
 B(III) nmr oth/un 27°C var C 1979YKb (43727) 97

$$K(B(OH)_4+H_2L=B(OH)_2L+2H_2O)=3.9$$

$$K(B(OH)_4+2H_2L=BL_2+4H_2O)=4.4$$

Method: 11B nmr. pH=6.5.

-----  
 B(III) gl KNO3 25°C 0.10M C 1977PBc (43728) 98

$$K(B(OH)_3+H_2L=B(OH)_2L+H)=-4.96$$

$K(PhB(OH)_2+H_2L=PhB(OH)L+H)=-4.33$ . PhB(OH)2 is phenylboronic acid.

-----  
 B(III) gl KCl 45°C 0.10M U T H 1968APc (43729) 99

$$K(B(OH)_4+H_2L=B(OH)_2L)=3.748$$

$$K'(B(OH)_4+2HL=BL_2)=3.996$$

$K=4.361(0\text{ C}), 3.972(25\text{ C}), 3.945(30\text{ C}), 3.843(35), 3.773(40)$ ;  $DH=-23.4\text{ kJ mol}^{-1}$ ,  
 $DS=-1.7\text{ J K}^{-1}\text{ mol}^{-1}$ .  $K'=4.637(0\text{ C}), 4.263(25\text{ C}), 4.077(35\text{ C})$ ;  $DH=-24.2$ ,  $DS=0$

-----  
 B(III) gl KNO3 20°C 0.10M U 1968HBa (43730) 100

$$K(H_3BO_3+H_2L=B(OH)_2L+H)=-5.17$$

-----  
 B(III) gl oth/un 35°C .025M U T H 1967CBd (43731) 101

$$K(B(OH)_4+H_2L=B(OH)_2L)=3.62$$

Medium: 0.025 borax.  $K=3.86(0\text{C}), 3.76(13\text{C}), 3.70(25\text{C})$ .  $DH=-11.3\text{ kJ mol}^{-1}$ ,  
 $DS=33\text{ J K}^{-1}\text{ mol}^{-1}$

-----  
B(III) gl KCl 23°C 0.10M U 1959AKa (43732) 102  
K(B(OH)4+H2L)=4.36-0.0145T  
K(B(OH)4+2H2L)=4.61-0.0140T

T=0-45 C

-----  
B(III) gl oth/un 25°C 0.10M U 1957RLa (43733) 103  
K(B(OH)2+H2L=BOL)=3.89

\*\*\*\*\*  
C6H6O3 H3L Pyrogallol CAS 87-66-1 (696)  
1,2,3-Trihydroxybenzene; C6H3(OH)3

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
B(III) EMF KCl 25°C 0.10M U 1971AKc (43951) 104  
K(HBO2+HL=H+HBO2L)=-5.05  
K(HBO2+2HL=H+H2BO2L2)=-4.40

-----  
B(III) gl KNO3 20°C 0.10M U 1968HBa (43952) 105  
K(H3BO3+H3L=B(OH)2HL+H)=-4.98

\*\*\*\*\*  
C6H6O5S H3L CAS 7134-09-0 (3687)  
3,4-Dihydroxybenzenesulfonic acid; (HO)2.C6H3.SO3H

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
B(III) gl KNO3 20°C 0.10M U 1968HBa (44281) 106  
K(H3BO3+H2L=B(OH)2L+H)=-4.60

\*\*\*\*\*  
C6H6O8S2 H4L Tiron CAS 149-45-1 (104)  
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)2.C6H2(SO3H)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
B(III) gl KNO3 25°C 0.10M U 1968HBa (44406) 107  
K(H3BO3+H2L=B(OH)2L+H)=-3.72

-----  
B(III) gl KCl 25°C var U I 1960NAa (44407) 108  
K(B(OH)3+H2L=B(OH)2L+H)=-4.34+3.05SQRTI/(1+1.3SQRTI)-0.16I  
At I=0: K(B(OH)4+H2L)=4.90

-----  
B(III) gl KCl 25°C 1.0M U 1960NAf (44408) 109  
K(H2L+B(OH)3=BL(OH)2+H)=3.20

\*\*\*\*\*  
C6H10O7 HL Glucuronic acid CAS 6556-12-3 (599)  
D-Glucuronic acid;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
B(III) gl KCl 25°C 0.10M M K1=1.71 1986LHa (48418) 110

\*\*\*\*\*  
 C6H1008                    H2L    Saccharic acid    CAS 87-73-0    (1191)  
 D-2,3,4,5-Tetrahydroxy-1,6-hexanedioic acid, Glucaric acid; HOOC.(CHOH)4.COOH

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 Metal            Mtd Medium Temp Conc Cal Flags Lg K values                    Reference ExptNo  
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B(III)            gl    KCl        25°C 0.10M M                    K1=2.16    B2=3.58    1986LHa (48468) 111

\*\*\*\*\*  
 C6H1202                    L                                    CAS 1792-81-0    (3657)  
 cis-1,2-Cyclohexanediol; C6H10(OH)2

-----  
 Metal            Mtd Medium Temp Conc Cal Flags Lg K values                    Reference ExptNo  
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B(III)            gl    oth/un 35°C .025M U T H                                    1967CBd (49430) 112

$$K(B(OH)4+L=B(OH)2H-2L)=0.0$$

$$K'(B(OH)4+2L=B(H-2L)2)=-0.5$$

Medium:borax. K=0.3(0 C),0.0(13-25 C); DH=-16.7 kJ mol<sup>-1</sup>, DS=-42 J K<sup>-1</sup> mol<sup>-1</sup>  
 K'=0.3(0 C), 0.3(13 C), -0.2(25 C), DH=-42, DS=-168

\*\*\*\*\*  
 C6H1205                    L    L-Rhamnose                    CAS 634-74-2    (3659)  
 6-Deoxy-L-mannose;

-----  
 Metal            Mtd Medium Temp Conc Cal Flags Lg K values                    Reference ExptNo  
 -----

B(III)            gl    KCl        25°C 0.10M U                                    1959ATa (49506) 113

$$K(B(OH)4+2L=B(H-2L)2)=2.61$$

\*\*\*\*\*  
 C6H1206                    L    D-Fructose                    CAS 57-48-7    (1561)  
 D-Fructose

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 Metal            Mtd Medium Temp Conc Cal Flags Lg K values                    Reference ExptNo  
 -----

B(III)            gl    KNO3      25°C 0.1M C T H                    K1=5.38                    1989JJa (49538) 114  
 At 15 C: K1=5.49, 35 C: K1=5.26, 45 C: K1=5.13. DH(K1)=-20.9 kJ mol<sup>-1</sup>

B(III)            cal    NaNO3    25°C 0.10M U    H                                    1988ARa (49539) 115  
 DH(B+L=BL)=-3 kJ mol<sup>-1</sup>; DS=59. DH(BL+L=BL2)=-33; DS=-84.

B(III)            gl    KCl        25°C 0.10M M                                    1987VHa (49540) 116

$$K(B(OH)4+L)=2.82$$

$$K(B(OH)4+2L)=4.97$$

-----  
 B(III)            gl    none      25°C 0.0 M                    K1=3.16    B2= 5.07    1979EMb (49541) 117

Metal is borate.

-----  
 B(III)            gl    oth/un 25°C 0.03M U T M                                    1970COa (49542) 118

$$K'(B(OH)4+L=B(OH)2(H-2)L)=3.48$$

$$K''(B(OH)4+2L=B(H-2L)2)=5.09$$

Medium: 0.027 borax. At 0 C: K'=3.70, K''=5.36. 13 C: K'=3.58, K''=5.33.  
 35 C: K'=3.21, K''=4.93

-----  
 B(III) gl KCl 45°C 0.10M U T H 1968APd (49543) 119  
 $K(B(OH)_4+L=B(OH)_2H-2L)=2.976$   
 $K'(B(OH)_4+2L=B(H-2L)_2)=4.643$   
 $K=4.142(0\text{ C}), 3.642(15\text{ C}), 3.416(25\text{ C}), 3.178(35\text{ C}); DH=-39.3\text{ kJ mol}^{-1}, DS=-66.4$   
 $K'=5.109(0\text{ C}), 5.062(15\text{ C}), 4.917(25\text{ C}), 4.772(35\text{ C}); DH=-24.6, DS=11.7$   
 -----

B(III) EMF KCl 25°C var U I 1967NEa (49544) 120  
 $K(B(OH)_4+2L)=4.723+0.470SQRTI$   
 -----

B(III) gl KCl 25°C 0.10M U 1958ANa (49545) 121  
 $K(B(OH)_4+2L=B(H-2L)_2)=5.04$   
 -----

B(III) gl oth/un 25°C ? U 1957RLa (49546) 122  
 $K(BO(OH)_2+2H_2L=BL_2)=4.98$   
 -----

\*\*\*\*\*  
 C6H12O6 L D-Galactose CAS 59-23-4 (1559)  
 D-Galactose  
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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B(III)	cal	NaNO3	25°C	0.10M	U	H			1988ARa (49558)	123
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$DH(B+L=BL)=-24.7\text{ kJ mol}^{-1}; DS=-42. DH(BL+L=BL_2)=48.5; DS=167.$   
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B(III)	gl	KCl	25°C	0.10M	M				1987VHa (49559)	124
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$K(B(OH)_4+L)=1.99$   
 $K(B(OH)_4+2L)=2.56$   
 -----

B(III)	gl	KCl	25°C	0.10M	M			$K_1=1.97$ $B_2=2.52$	1986LHa (49560)	125
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B(III)	gl	none	25°C	0.0	M			$K_1=2.09$ $B_2=2.62$	1979EMb (49561)	126
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Metal is borate.  
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B(III)	gl	oth/un	25°C	0.03M	U	T	M		1970COa (49562)	127
--------	----	--------	------	-------	---	---	---	--	-----------------	-----

$K'(B(OH)_4+L=B(OH)_2(H-2)L)=2.24$   
 $K''(B(OH)_4+2L=B(H-2L)_2)=2.63$   
 Medium: 0.027 borax. At 0 C:  $K'=2.50, K''=2.92.$  13 C:  $K'=2.38, K''=2.72.$   
 35 C:  $K'=2.19, K''=2.55$   
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B(III)	gl	KCl	25°C	0.10M	U				1958ANa (49563)	128
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$K(B(OH)_4+2L=B(H-2L)_2)=2.39$   
 -----

B(III)	gl	oth/un	25°C	0.10M	U				1957RLa (49564)	129
--------	----	--------	------	-------	---	--	--	--	-----------------	-----

$K(BO(OH)_2+H_2L=BOL)=2.10$   
 $K(BO(OH)_2+2H_2L=BL_2)=2.47$   
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\*\*\*\*\*  
 C6H12O6 L D-Glucose CAS 492-62-6 (1560)  
 D-Glucose  
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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B(III) gl KNO<sub>3</sub> 25°C 0.1M C T H K<sub>1</sub>=2.82 1989JJa (49576) 130  
 At 15 C: K<sub>1</sub>=2.86, 35 C: K=2.77, 45 C: K=2.71. DH(K<sub>1</sub>)=-8.6 kJ mol<sup>-1</sup>

-----

B(III) cal NaNO<sub>3</sub> 25°C 0.10M U H 1988ARa (49577) 131  
 DH(B+L=BL)=-17 kJ mol<sup>-1</sup>; DS=-12.5. DH(BL+L=BL<sub>2</sub>)=15; DS=58.

-----

B(III) gl KCl 25°C 0.10M M 1987VHa (49578) 132  
 K(B(OH)<sub>4</sub>+L)=1.80  
 K(B(OH)<sub>4</sub>+2L)=3.05

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B(III) gl KCl 25°C 0.10M M K<sub>1</sub>=2.07 B<sub>2</sub>=2.80 1986LHa (49579) 133

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B(III) gl NaClO<sub>4</sub> 25°C 0.02M M 1981PAa (49580) 134  
 K(B(OH)<sub>4</sub>+L)=1.62  
 Medium: 0.015 M NaClO<sub>4</sub>, pH 8.0-9.2.

-----

B(III) gl KNO<sub>3</sub> 20°C 0.10M M 1980MBc (49581) 135  
 K(B(OH)<sub>3</sub>+2H<sub>2</sub>L=BL<sub>2</sub>+H)=-6.33  
 For L=D-sorbitol, K=-3.78; L=D-dulcitol, K=-4.03; L=D-adonitol, K=-5.48.

-----

B(III) gl none 25°C 0.0 M K<sub>1</sub>=2.11 B<sub>2</sub>= 2.87 1979EMb (49582) 136  
 Metal is borate.

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B(III) gl oth/un 25°C 0.03M U T M 1970COa (49583) 137  
 K'(B(OH)<sub>4</sub>+L=B(OH)<sub>2</sub>(H-2)L)=2.18  
 K''(B(OH)<sub>4</sub>+2L=B(H-2L)<sub>2</sub>)=2.88  
 Medium: 0.027 borax. At 0 C: K'=2.41, K''=3.09. 13 C: K'=2.31, K''=3.03.  
 35 C: K'=2.04, K''=2.79

-----

B(III) gl KCl 45°C 0.10M U T H 1968APd (49584) 138  
 K(B(OH)<sub>4</sub>+L=B(OH)<sub>2</sub>H<sub>2</sub>L)=1.978  
 K'(B(OH)<sub>4</sub>+2L=B(H-2L)<sub>2</sub>)=2.407  
 K=2.305(0 C), 2.071(15 C), 2.022(25 C), 1.985(35 C); DH=-5.4 kJ mol<sup>-1</sup>, DS=20.1;  
 K'=2.894(0 C), 2.750(15 C), 2.633(25 C), 2.560(35 C); DH=-19.2, DS=-14.2

-----

B(III) gl oth/un 35°C .025M U T H 1967CBd (49585) 139  
 K(B(OH)<sub>4</sub>+L=B(OH)<sub>2</sub>H-2L)=2.10  
 K'(B(OH)<sub>4</sub>+2L=B(H-2L)<sub>2</sub>)=2.95  
 Medium: borax. K=2.33(0 C), 2.24(13 C), 2.13(25 C); DH=-14.6 kJ mol<sup>-1</sup>  
 DS=-8.4 J K<sup>-1</sup> mol<sup>-1</sup>; K'=2.95(0 C), 2.95(13 C), 2.94(25 C), DH=-0.6, DS=54.3

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B(III) EMF KCl 25°C var U I 1967NEa (49586) 140  
 K(B(OH)<sub>4</sub>+2L)=2.376+1.073SQRTI

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B(III) gl KCl 25°C 0.10M U 1958ANa (49587) 141  
 K(B(OH)<sub>4</sub>+2L=B(H-2L)<sub>2</sub>)=2.86

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B(III) gl oth/un 25°C 0.10M U 1957RLa (49588) 142  
 K(BO(OH)<sub>2</sub>+H<sub>2</sub>L=BOL)=1.90





DH(B(OH)<sub>4</sub>+L)=-19.7 kJ mol<sup>-1</sup>, DS=-7.9 J K<sup>-1</sup> mol<sup>-1</sup>. DH(B(OH)<sub>4</sub>L+L)=-23.4, DS=-50.

-----  
B(III) cal oth/un 25°C dil C H 1977EFa (51072) 159

K(B(OH)<sub>4</sub>+L=B(OH)<sub>2</sub>L+2H<sub>2</sub>O)=3.03

K(B(OH)<sub>2</sub>L+L=BL<sub>2</sub>+2H<sub>2</sub>O)=2.05

Self medium, 0.027 M borate, 0.01-0.09 M mannitol. DH(B(OH)<sub>4</sub>+L)=-18.6 kJ mol<sup>-1</sup>, DS(B(OH)<sub>4</sub>+L)=-4.6 J K<sup>-1</sup> mol<sup>-1</sup>; DH(B(OH)<sub>2</sub>L+L)=-19.9, DS=-27.

-----  
B(III) oth KCl 25°C 0.10M U 1973KAa (51073) 160

K(B(OH)<sub>4</sub>+L=B(OH)<sub>2</sub>H-2L)=2.98

K(B(OH)<sub>3</sub>+2L=B(H-2L)<sub>2</sub>+H)=-3.91

K(2B(OH)<sub>3</sub>+L=B<sub>2</sub>(OH)<sub>2</sub>H-2L)=2.0

Method: potentiostatic titration. K(B(OH)<sub>2</sub>+2L=B(H-1L)(H-2L))=-0.21

K(2B(OH)<sub>4</sub>+L=B<sub>2</sub>(OH)<sub>4</sub>(H-2L))=4.41

-----  
B(III) gl NaClO<sub>4</sub> 25°C 3.00M U K<sub>1</sub>=-0.14 1973PAb (51074) 161

K<sub>1</sub> also measured by polarimetry. K(B(OH)<sub>3</sub>+nL+H<sub>2</sub>O=B(OH)<sub>4</sub>L<sub>n</sub>+H)=-6.00 (n=1), (n=2)=-4.10; K(2B(OH)<sub>3</sub>+nL+H<sub>2</sub>O=(H-2)(B(OH)<sub>3</sub>)<sub>2</sub>L<sub>n</sub>+2H)(n=1)=-13.61, (n=2)=-10.76

-----  
B(III) EMF KCl 25°C 3.00M U 1972AAa (51075) 162

K(H<sub>3</sub>BO<sub>3</sub>+L)=-0.22

K(H<sub>3</sub>BO<sub>3</sub>L=H<sub>2</sub>BO<sub>3</sub>L+H)=-6.04

K(H<sub>3</sub>BO<sub>3</sub>L+L=H<sub>2</sub>BO<sub>3</sub>L<sub>2</sub>+H)=-4.07

-----  
B(III) gl KCl 45°C 0.10M U T H 1968APd (51076) 163

K(B(OH)<sub>4</sub>+L=B(OH)<sub>2</sub>H-2L)=3.398

K'(B(OH)<sub>4</sub>+2L=B(H-2L)<sub>2</sub>)=4.551

K=4.21(0C), 4.00(25C), 3.62(40C); DH=3.3(0C), -16.7(15C), -32.2(25C), -49.7(35C), -69.8(45C)kJmol<sup>-1</sup>. K'=5.408(0C), 4.888(25C), 4.610(35C); DH=-31.8(25C)

-----  
B(III) gl oth/un 35°C .025M U T H 1967CBd (51077) 164

K(B(OH)<sub>4</sub>+L=B(OH)<sub>2</sub>H-2L)=2.90

K'(B(OH)<sub>4</sub>+2L=B(H-2L)<sub>2</sub>)=5.05

Medium:borax. K=3.62(0 C), 3.36(13 C), 3.04(25 C); DH=-33.9 kJ mol<sup>-1</sup>, DS=-54.3 J K<sup>-1</sup> mol<sup>-1</sup>; K=5.43(0 C), 5.31(13 C), 5.14(25 C); DH=-18.8, DS=37.6

-----  
B(III) gl KCl 25°C var U 1967NEb (51078) 165

K(B(OH)<sub>4</sub>+2L)=4.225+0.554SQRTI

-----  
B(III) gl KCl 25°C 2.0M U I 1955ANa (51079) 166

K(H<sub>3</sub>BO<sub>3</sub>+L=B(OH)<sub>2</sub>H-2L+H)=-5.13

K'(H<sub>3</sub>BO<sub>3</sub>+2L=B(H-2L)<sub>2</sub>+H)=-4.29

K(H<sub>3</sub>BO<sub>3</sub>+L)=-5.22(I=0), -5.10(I=0.1), -5.02(I=0.4)

K'(H<sub>3</sub>BO<sub>3</sub>+2L)=-4.36(I=0), -4.18(I=0.1), -4.15(I=0.4)

-----  
B(III) gl KCl 25°C 0.10M U 1949RCa (51080) 167

K(H<sub>3</sub>BO<sub>3</sub>+2L=B(H<sub>2</sub>L)<sub>2</sub>+H)=-4.00

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C6H14O6 L Glucitol CAS 50-70-4 (2878)





3,4-Dihydroxybenzoic acid; C6H3(OH)2.COOH

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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
B(III)     gl  KCl    25°C 0.10M U                                1968AVa (54662) 185
                                                K(H3BO3+H2L=B(OH)2L+H)=5.01
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C7H6O5          H4L    Gallic acid      CAS 149-91-7 (446)
3,4,5-Trihydroxybenzoic acid; C6H2(OH)3.COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
B(III)     EMF KCl    25°C 0.10M U                                1969AVc (54750) 186
                                                K(HBO2+H3L=HBO2(H2L)+H)=-8.87
-----
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```
C7H6O6S          H3L          CAS 585-42-2 (6136)
2-Hydroxy-4-sulphobenzoic acid, 4-sulfosalicylic acid; HO.C6H3(COOH)(HSO3)
-----
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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
B(III)     gl  KNO3   20°C 0.30M U                                1978MBb (54803) 187
                                                K(H3BO3+HL=B(OH)2L+H2O)=0.98
-----
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```
C7H7NO3          H2L          CAS 89-73-6 (204)
2-Hydroxybenzohydroxamic acid (salicylhydroxamic acid); HO.C6H4.CO.NHOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
B(III)     gl  KCl    25°C 0.10M U                                1970BMe (55587) 188
                                                K(H3BO3+HL=B(OH)2L)=4.5
-----
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```
C7H7NO3          H2L          (1112)
4-Aminosalicylic acid; H2N.C6H3(OH).COOH
-----
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
B(III)     gl  KNO3   20°C 0.10M U                                1978MBb (55636) 189
                                                K(H3BO3+HL=B(OH)2L+H2O)=1.29
-----
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B(III)     gl  KNO3   30°C 0.10M M                                1978MBb (55637) 190
                                                K(H3BO3+HL=B(OH)2L+H2O)=1.31
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```
C7H7NS          L    Thiobenzamide  CAS 2227-79-4 (1660)
Thiobenzamide; C6H5.CS.NH2
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
B(III)     sp  non-aq 25°C 100% U                                1977SWa (55703) 191
                                                K(BF3+L)=1.08
-----
```

Medium: Et2O





Methyl  $\alpha$ -D-mannopyranoside;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) nmr NaCl 25°C 0.10M C 2003MYb (57889) 198  
K(B(OH)<sub>4</sub>+L)=1.28  
K'(B(OH)<sub>4</sub>+2L)=0.98  
K''(B(OH)<sub>4</sub>+2L)=1.69  
K'''(B(OH)<sub>4</sub>+L)=0.64

Method: 11B nmr. K: ( $\alpha$ , $\beta$  complex), K': ( $\alpha$ , $\beta$ )( $\alpha$ , $\gamma$ ) complex, K'': ( $\alpha$ , $\beta$ )( $\alpha$ , $\beta$ ) complex, K''': ( $\alpha$ , $\gamma$ ) complex.

-----  
B(III) gl oth/un 25°C var U 1965LAa (57890) 199  
K(H<sub>3</sub>B<sub>3</sub>O<sub>3</sub>+L=B(OH)<sub>2</sub>(H<sub>2</sub>L)+H)=-8  
L)H<sub>3</sub>B<sub>3</sub>O<sub>3</sub>+2L=B(H<sub>2</sub>L)<sub>2</sub>+H)=-7.2

-----  
B(III) oth oth/un 30°C .105M U 1964MGa (57891) 200  
K(B(OH)<sub>4</sub>+L)=1.7  
K(B(OH)<sub>4</sub>+2L)=2.78

Method: refractive index and optical rotation.

\*\*\*\*\*

C8H6O4 H2L Phthalic acid CAS 88-99-3 (113)  
Benzene-1,2-dicarboxylic acid; C<sub>6</sub>H<sub>4</sub>(COOH)<sub>2</sub>

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) gl none 25°C 0.0 M 1991MIa (58945) 201  
B(H<sub>3</sub>B<sub>3</sub>O<sub>3</sub>+L=B(OH)<sub>3</sub>L)=-0.07

\*\*\*\*\*

C8H10O2 L CAS 7138-28-5 (3199)  
Phenylethane-1,2-diol; C<sub>6</sub>H<sub>5</sub>.CH(OH).CH<sub>2</sub>.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) gl oth/un 25°C 0.10M U 1957RLa (60835) 202  
K(B(OH)<sub>2</sub>+H<sub>2</sub>L)=0.87  
K(B(OH)<sub>2</sub>+2H<sub>2</sub>L)=0.85

\*\*\*\*\*

C8H11NO2 H2L Dopamine CAS 579-59-9 (251)  
2-(3',4'-Dihydroxyphenyl)ethylamine; (HO)<sub>2</sub>.C<sub>6</sub>H<sub>3</sub>.CH<sub>2</sub>.CH<sub>2</sub>.NH<sub>2</sub>

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) gl KCl 25°C 0.10M M K<sub>1</sub>=4.86 1974AWa (61076) 203

\*\*\*\*\*

C8H11NO3 H2L Noradrenaline CAS 138-65-8 (253)  
Norepinephrine, 3,4-Dihydroxyphenylethanolamine; (HO)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>.CH(CH<sub>2</sub>.NH<sub>2</sub>).OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) EMF KCl 25°C 0.10M U 1971AVa (61161) 204  
K(HBO2+H2L=H2BO2L+H)=-4.74

Constants quoted for L isomer. Kor DL-isomer, K=-4.89

\*\*\*\*\*

C8H16O6 L CAS 7468-45-3 (3808)  
Methyl-4-O-methyl-a-D-mannopyranoside;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) oth oth/un 30°C .105M U 1964MGa (62736) 205  
K(B(OH)4+L=B(OH)2H-2L)=1.5  
K(B(OH)4+2L=B(H-2L)2)=3.5

Method: refractive index, optical rotation.

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C8H16O6 L CAS 99745-67-2 (3809)  
Methyl-4-O-methyl-b-D-mannopyranoside;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) oth oth/un 30°C .105M U 1964MGa (62737) 206  
K(B(OH)4+L=B(OH)2H-2L)=0.3  
K(B(OH)4+2L=B(H-2L)2)=2.60

Method: refractive index, optical rotation.

\*\*\*\*\*

C9H11NO4 H3L DOPA CAS 59-92-7 (5)  
2-Amino-3-(3,4-dihydroxyphenyl)propanoic acid;H2NCH(CH2C6H3(OH)2)COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) nmr oth/un 27°C var C 1979YKb (66394) 207  
K(B(OH)4+H2L=B(OH)2L+2H2O)=4.3  
K(B(OH)4+2H2L=BL2+4H2O)=5.0

Method: 11B nmr. pH=6.5.

\*\*\*\*\*

C9H13NO3 H2L (-)Adrenaline CAS 51-43-4 (252)  
4-(1-Hydroxy-2-(methylamino)ethyl)-1,2-dihydroxybenzene,  
Epinephrine;CH3NHCH(OH)C6H3(OH)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

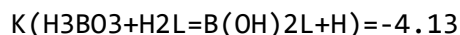
B(III) gl KCl 25°C 0.10M U 1966ATd (66858) 208  
K(H3BO3+H2L=B(OH)2L+H)=-4.67  
K(H3BO3+2H2L=BL2+H)=-3.70

\*\*\*\*\*

C10H8O2 H2L CAS 92-44-4 (1658)  
2,3-Dihydroxynaphthalene;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) gl KNO3 20°C 0.10M U 1968HBa (69764) 209



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C10H8O5S H3L DHNSA (877)  
2,3-Dihydroxynaphthalene-6-sulfonic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) EMF KCl 25°C 0.10M U 1971SBd (69837) 210  
K(HBO2+H2L=H2LBO2+H)=-3.76  
-----

B(III) gl KNO3 25°C 0.10M U 1968HBa (69838) 211  
K(H3BO3+H2L=B(OH)2L+H)=-3.98

\*\*\*\*\*

C10H8O8S2 H4L Chromotropic ac CAS 148-25-4 (1875)  
1,8-Dihydroxynaphthalene-3,6-disulfonic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) nmr oth/un 24°C 0.10M U 2000SMb (69927) 212  
K(B(OH)3+H2L=B(OH)2L+H)=-1.57  
K(B(OH)3+2H2L=BL2+H)=2.35

Method: 11B nmr.

-----  
B(III) gl KNO3 30°C 0.10M M 1978MBb (69928) 213  
K(H3BO3+HL=B(OH)2L+H2O)=-0.07  
-----

B(III) gl KNO3 20°C 0.10M U 1967BHb (69929) 214  
K(H3BO3+H2L=BL(OH)2+H)=-1.55  
K(H3BO3+2H2L=BL2+H)=-2.4

\*\*\*\*\*

C10H10O2 HL Benzoylacetone CAS 93-91-4 (197)  
1-Phenylbutane-1,3-dione; C6H5.CO.CH2.CO.CH3

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) oth oth/un low ? U 1971GMb (70706) 215  
K(H3BO3+HL=H3BO3.HL)=5.15  
K(H3BO3+2HL=H3BO3(HL)2)=8.38

Medium: glassy ether-conc. H2SO4 at -196 C. Method: phosphorescence

-----  
B(III) oth oth/un low ? U 1969MGd (70707) 216  
K(H3BO3+HL=H3BO3.HL)=5.15  
KH3BO3+2HL=H3BO3(HL)2)=7.50

Medium: glassy ether-conc. H2SO4 at -196 C. Method: phosphorescence

\*\*\*\*\*

C10H14N2O L CAS 67402-02-2 (6298)  
N-Trimethylammonio benzamidate;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) cal non-aq 25°C 100% U H 1978GMe (72070) 217  
Medium: CH2Cl2. DH(BF3L)=-116.0 kJ mol<sup>-1</sup>. Data also for related ligands

\*\*\*\*\*

C10H17NO L CAS 31039-88-0 (5637)

3-Dimethylamino-5,5-dimethylcyclohex-2-enone;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) cal non-aq 25°C 100% U H 1983AGa (74958) 218

DH(BF3+L=BF3L)=-132.4 kJ mol<sup>-1</sup> in dichloromethane.

Data also for B(III) complexes of 15 other dimethylcyclohex-2-enones.

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C11H8O3 H2L CAS 86-48-6 (1129)

1-Hydroxy-2-naphthoic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) gl KNO3 30°C 0.10M M 1978MBb (77006) 219

K(H3BO3+HL=B(OH)2L+H2O)=1.31

\*\*\*\*\*

C11H8O3 H2L CAS 2083-08-1 (1131)

2-Hydroxy-1-naphthoic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) gl KNO3 30°C 0.10M M 1978MBb (77059) 220

K(H3BO3+HL=B(OH)2L+H2O)=1.83

\*\*\*\*\*

C11H17NO3 H2L Isoproterenol CAS 949-36-0 (2671)

N-Isopropyl-DL-noradrenaline; (HO)2C6H3.CH(OH)CH2.NCH(CH3)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) gl KCl 25°C 0.10M M K1=4.81 1976AWa (79160) 221

\*\*\*\*\*

C12H16O6 L CAS 1464-44-4 (3960)

Phenyl beta-D-glucopyranoside;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) gl oth/un 25°C var U 1965LAa (81692) 222

K(H3BO3+L=H2BO3L+H)=-8

\*\*\*\*\*

C12H22O11 L Turanose CAS 547-25-1 (2701)

3-O-D-Glucopyranosyl-D-fructose;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

B(III) gl KCl 25°C 0.10M M K1=1.91 B2=2.47 1986LHa (82865) 223

\*\*\*\*\*

C12H22O11 L alpha-Lactose CAS 5989-81-1 (2486)  
4-D-Beta-D-Galactopyranosyl-alpha-D-glucose;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	gl	KCl	25°C	0.10M	M			K(B(OH)4+L)=1.43 K(B(OH)4+2L)=2.17	1987VHa (82871)	224
B(III)	gl	KCl	25°C	0.10M	M			K1=1.51	1986LHa (82872)	225
B(III)	gl	none	25°C	0.0	M			K1=1.36 B2= 2.05	1979EMb (82873)	226

Metal is borate.

C12H22O11 L Maltose CAS 6363-53-7 (2705)  
4-O-alpha-D-Glucopyranosyl-D-glucose, Maltobiose;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	gl	KCl	25°C	0.10M	M			K(B(OH)4+L)=1.41 K(B(OH)4+2L)=1.89	1987VHa (82878)	227
B(III)	gl	KCl	25°C	0.10M	M			K1=1.36	1986LHa (82879)	228

C12H22O11 L CAS 4618-18-2 (8502)  
4-O-beta-D-Galactopyranosyl-D-fructose;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	gl	KCl	25°C	0.10M	M			K(B(OH)4+L)=2.91 K(B(OH)4+2L)=5.14	1987VHa (82883)	229

C12H22O11 L Cellobiose CAS 528-50-7 (2697)  
4-O-beta-D-Glucopyranosyl-D-glucose;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	gl	KCl	25°C	0.10M	M			K1=1.25	1986LHa (82885)	230

C12H22O11 L Melibiose CAS 66009-10-7 (2699)  
6-O-D-Galactopyranose-D-glucose;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	gl	KCl	25°C	0.10M	M			K1=1.82 B2=2.44	1986LHa (82889)	231

C12H22O11 L Gentiobiose CAS 554-91-6 (2698)  
6-O-D-Glucopyranosyl-D-glucose, Amygdalose;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) gl KCl 25°C 0.10M M K1=1.14 1986LHa (82892) 232  
\*\*\*\*\*  
C12H22O11 L Trehalose CAS 6138-23-4 (2700)  
D-Glucopyranosyl-D-glucopyranoside;  
-----

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) gl oth/un 25°C dil M 2004LHa (82898) 233  
K(B(OH)4+L)=1.26  
Self medium, 0.001-0.01 M sodium borate, pH 9.2. Reaction is an  
esterification, with loss of 2H2O.  
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B(III) gl KCl 25°C 0.10M M K1=1.04 1986LHa (82899) 234  
\*\*\*\*\*  
C12H22O11 L Sucrose CAS 57-50-1 (2523)  
beta-D-Fructofuranosyl-alpha-D-glucopyranoside; Saccharose;  
-----

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) gl oth/un 25°C dil M 2004LHa (82907) 235  
K(B(OH)4+L)=0.98  
Self medium, 0.001-0.01 M sodium borate, pH 9.5. Reaction is an  
esterification, with loss of 2H2O.  
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-----  
B(III) gl KCl 25°C 0.10M M 1987VHa (82908) 236  
K(B(OH)4+L)=0.86  
K(B(OH)4+2L)=0.70  
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B(III) gl KCl 25°C 0.10M M K1=0.75 1986LHa (82909) 237  
\*\*\*\*\*  
C12H24O11 L Maltitol CAS 585-88-6 (2709)  
4-O-alpha-D-Glucopyranosyl-D-glucitol;  
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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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B(III) gl KCl 25°C 0.10M M K1=3.71 1988HLA (83682) 238  
\*\*\*\*\*  
C12H24O11 L Lactitol CAS 535-94-4 (2710)  
4-O-beta-D-Galactopyranosyl-D-glucitol;  
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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
B(III) gl KCl 25°C 0.10M M K1=3.37 1988HLA (83685) 239  
\*\*\*\*\*  
C14H8O7S H3L DASA CAS 83-61-4 (950)  
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;  
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	gl	KNO3	20°C	0.10M	U				1968BHb (86716)	240
									K(H3BO3+H3L=B(OH)2HL+H)=-3.4	
*****										
C14H10N2O4		H2L							Diaminochrysazi CAS 29706-46-5	(4039)
4,5-Diamino-1,8-dihydroxyanthraquinone;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	sp	oth/un	25°C	?	U				1964BRa (86900)	241
									K(B(OH)3+H2L=B(OH)2HL)=3.54	
*****										
C18H32O16		L							Raffinose CAS 17629-30-0	(5611)
Galactopyranosyl-[1-6]-glucopyranosyl-fructofuranoside;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	gl	KCl	25°C	0.10M	M				1987VHa (98277)	242
									K(B(OH)4+L)=1.35	
									K(B(OH)4+2L)=1.67	
*****										
C18H32O16		L							Melezitose CAS 10030-67-8	(3834)
Glucopyranosyl-[1-3]-fructofuranosyl-[2-1]-glucopyranoside;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	gl	KCl	25°C	0.10M	M				1987VHa (98279)	243
									K(B(OH)4+L)=1.05	
									K(B(OH)4+2L)=1.14	
*****										
C22H20O13		H5L							Carminic acid CAS 1260-17-9	(714)
Carminic acid;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	sp	oth/un	25°C	?	U				1970BRa (101700)	244
									K(B(OH)3+H5L)=4.29	
Medium: conc H2SO4										
*****										
C23H26N2O		L							Malechite Green CAS 510-13-4	(3517)
1-(Bis-(4-dimethylaminophenyl)methylene)-2-oxobenzene; C6H5.C(OH)(C6H4.N(CH3)2)2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	sp	KCl	20°C	0.50M	U				1962CIa (102703)	245
									K(B(OH)4+L=B(OH)2H-2L)=4.38	
									K(B(OH)4+HL=B(OH)2H-1L)=3.60	
*****										
C28H15N04		L							CAS 82-22-4	(3522)

1,1'-Iminodianthraquinone; (1,1'-dianthrimide)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	sp	oth/un	70°C	94%	U				1959LSa (104652)	246

K(HBO2+HL=BOL)=5.15

Medium: 93.8% H2SO4

\*\*\*\*\*

Polymer (4200)  
 Polyvinyl alcohol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
B(III)	gl	oth/un	25°C	0.10M	U				1957RLa (108381)	247

K'(B(OH)4+L=B(OH)2H-2L)=0.26  
 K'(B(OH)4+2L=B(H-2L)2)=0.64

See reference for definitions

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#### EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES  
 I Data with various BACKGROUNDS  
 H Data for THERMOCHEMICAL quantities  
 M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC

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