

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 19 experiments for
(no ligands specified)

Metal : Cl

(no references specified)

(no experimental details specified)

e- HL Electron (442)
Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Cl	oth	none	25°C	0.0	U			K(1/2Cl ₂ (g)+e=Cl)=22.21(1.31V)	1972COa (401)	1
Method:Estimated data										
Cl	oth	non-aq	25°C	100%	U	I		K(0.5Cl ₂ +e=Cl)=24.27(1.436V)	1972COa (402)	2
Method:Estimated data. MeOH. K=23.18(1.37V,EtOH), 22.43(1.327V,BuOH), 22.08(1.306V,PentOH), 21.70(1.284V,acetone), 15.10(MeCN), 21.70(HCOOH)										
Cl	oth	non-aq	400°C	100%	U	T		K(1/2Cl ₂ (g)+e=Cl)=7.63(1.019V)	1971BJc (403)	3
Method:Estimated data. Medium: fused (Li,K)Cl. K=6.87(0.986V, 450 C), 5.63(0.920V,550 C)										
Cl	oth	none	25°C	0.00	U			K=41.6(1.23V)	1970JSa (404)	4
K=ClO ₄ ⁻ + 2H ⁺ + 2e=ClO ₃ ⁻ + H ₂ O. Method:combination of thermodynamic data										
Cl	EMF	none	25°C	0.00	U	T		K=22.960(1358.27mV)	1969CLb (405)	5
K=1/2Cl ₂ (g) + e=Cl ⁻ . K=21.552(1339.10mV,40 C), 19.841(1311.54mV,60 C), 17.393(1218.72mV,80 C)										
Cl	EMF	none	25°C	0.00	U	T		K=22.964(1358.52mV)	1968CLa (406)	6
K=1/2Cl ₂ (g) + e=Cl ⁻ . K=22.481(1352.24mV,30 C), 21.553(1339.19mV,40 C), 20.677(1325.74mV,50 C), 19.840(1311.44mV,60 C), 18.289(1281.53mV, 80 C)										
Cl	EMF	NaClO ₄	25°C	3.0M	U	I		K(0.5Cl ₂ (g)+e=Cl)=22.671	1967KRb (407)	7
I=2.0: K=22.957, 1358 mV; I=1.0: K=23.158, 1369.9 mV										
Cl	oth	none	25°C	0.0	U			K=40.2(1190 mV)	1952LAb (408)	8
K: Cl(VII)O ₄ +2H+2e=Cl(V)O ₃ +H ₂ O. From thermodynamic data. K(Cl(V)O ₂ ++H ₂ O+2e=Cl(III)O ₂ +2OH)=11.1(330 mV). Cl(III)O ₂ +H ₂ O+2e=Cl(I)O+H ₂ O)=55.6(1640 mV)										

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Cl sp oth/un 25°C 1.00M U 1994Wka (4602) 9
K(Cl2+Cl)=-0.74
K(BrCl+Cl)=0.78
K(Br2+Cl)=0.11

Medium: HCl

Cl oth non-aq 0°C 100% U I 1973Gma (4603) 10
K(Cl2+Cl)=2.03

Medium: CH3CN, I M LiClO4(I=0.0095). K=1.72(I=0.0388); K=1.58(I=0.0498);
1.36(I=0.0755); 1.15(I=0.1268); 0.85(I=0.3645).Method:chem. anal. and p(Cl2)

Cl EMF non-aq 127°C 100% U 1971BTa (4604) 11
K(Cl2+Cl)=2.5

Medium: dimethylsulfone, 1 M LiClO4

Cl oth NaClO4 25°C 4.0M U T 1971SHc (4605) 12
K(Cl2+Cl)=-0.62

Medium: HClO4; K1=-0.47(-4 C), -0.49(0 C). Method:chemical analysis

Cl sp non-aq 30°C 100% U 1970DBa (4606) 13
K(Cl2+Cl)=4.11

Medium: sulfolane, 0 corr. Emf also used

Cl sol oth/un 90°C var U T 1968HIa (4607) 14
Kp0=-1.88
Kp1=-2.86

With Cl2. Medium:HCl var. Kp0=-1.31(30 C), -1.46(40 C), -1.57(50 C), 1.75(70 C)
Kp1=-2.18(30 C), -2.29(40 C), -2.44(50 C), -2.66(70 C)

Cl sol oth/un 25°C var U 1931SIa (4608) 15
K(Cl2(g)=Cl2(aq))=-1.23
K(Cl2(g)+Cl=Cl3)=-2.0
K(Cl2+Cl)=-0.75

ClO2- HL Chlorite CAS 13898-47-0 (6143)
Chlorite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Cl sp oth/un 25°C dil U 1968HRc (6008) 16
K(ClO2+L)=0.20

CrO4-- H2L Chromate CAS 7738-94-5 (2382)
Chromate;

```

-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Cl         kin NaClO4 26°C 2.00M U T                        1971RKa (6480) 17
                                     K'=1.51
38 C: K'=1.37. K': Cl+2H+HCrO4=HCrO3Cl+H2O. DH=-20.8 kJ mol-1
*****
H2O                L      Water                        CAS 7732-18-5 (6115)
Water
-----

```

```

-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Cl         sol non-aq 25°C 100% U          K1=0.95 B2=1.3 1967CKa (7587) 18
Medium:MeCN. Ligand Cl-. K(L+ClO4)=-0.15; K(L+IO4)=0.0
*****
SO2                L      Sulfur dioxide      (6336)
Sulfur dioxide;
-----

```

```

-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Cl         sp non-aq 25°C 100% U T H      K1=2.56        1971WNb (15353) 19
Medium: MeCN. DH(K1)=-9.7 kJ mol-1, DS=16.3 J K-1 mol-1
-----

```

REFERENCES

- 1994WKa T Wang, M Kelley, J Cooper et al; Inorg.Chem., 33, 5872 (1994)
 1973GMA M Giordano, V Macagno, L Sereno; Anal.Chem., 45, 205 (1973)
 1972COa E Constantinescu; Rev.Roumaine Chim., 17, 1819 (1972)
 1971BJc L Boxall, K Johnson; J.Electroanal.Chem., 30, 25 (1971)
 1971BTa B Bry, B Tremillon; J.Electroanal.Chem., 30, 457 (1971)
 1971RKa P Rao, P Kamannarayana; Z.Phys.Chem., 248, 267 (1971)
 1971SHc I Shimonis; Zh.Neorg.Khim., 16, 2902(E:1543) (1971)
 1971WNb E Woodhouse, T Norris; Inorg.Chem., 10, 614 (1971)
 1970DBa M Deneux, R Benoit; Can.J.Chem., 48, 674 (1970)
 1970Jsa G Johnson, P Smith, E Appelman et al; Inorg.Chem., 9, 119 (1970)
 1969CLb A Cerquetti, P Longhi, T Mussini, G Natta; J.Electroanal.Chem., 20, 411
 (1969)
 1968CLa A Cerquetti, P Longhi, T Mussini; J.Chem.Eng.Data, 13, 458 (1968)
 1968HIa F Hine, S Inuta; Bull.Chem.Soc.Jpn., 41, 71 (1968)
 1968HRC C Hong, W Rapson; Can.J.Chem., 46, 2053 (1968)
 1967CKa M Chantooni, un, I Kolthoff; J.Am.Chem.Soc., 89, 1582 (1967)
 1967KRb W Kraft; Monatsh.Chem., 98, 1978 (1967)
 1952LAb W Latimer; "Oxidation Potentials", Prentice Hall, NY (1952)
 1931SIa M Sherrill, E Iazard; J.Am.Chem.Soc., 53, 1667 (1931)

EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES

I Data with various BACKGROUNDS
H Data for THERMOCHEMICAL quantities

END