

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 348 experiments for

(no ligands specified)

5 metals : Mo(0), Mo(III), Mo(IV), Mo(V), Mo(VI)

(no references specified)

(no experimental details specified)

\*\*\*\*\*

C3H9O3P L CAS 121-45-9 (1786)  
 Trimethylphosphite; (CH3O)3.P

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Mo(0) cal non-aq 25°C 100% U HM 1991ZGa (28002) 1  
 Medium: THF. DH(Mo(CO)3A2+L)=-100.4 kJ mol-1, A=P(C6H11)3

\*\*\*\*\*

C18H33P L CAS 2622-14-2 (169)  
 Tri-(cyclohexyl)phosphine; (C6H11)3P

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Mo(0) cal non-aq 25°C 100% U HM 1991ZGa (98314) 2  
 K(Mo(CO)3py2+L)=-4.73

Medium: THF. DH=-70.7 kJ mol-1

\*\*\*\*\*

e- HL Electron (442)  
 Electron;

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Mo(III) EMF oth/un 25°C 4.0M U I 1963MAb (680) 3  
 K=14.64, 866 mV

K: Mo(Cn)6--- + e = Mo(CN)6---- . K=13.79(I=0.5 M; 816 mV). Medium: H2SO4

\*\*\*\*\*

CO L Carbon monoxide CAS 630-08-0 (551)  
 Carbon monoxide;

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Mo(III) EMF non-aq 22°C 100% U 1992PMa (2815) 4  
 K([MoL3A]2=2MoL3A)=-16.15

Metal:Mo+. Medium: MeCN, 0.1 M Bu4NPF6. A=C5H5. Dimer-monomer equilibrium

\*\*\*\*\*

Cl- HL Chloride CAS 7647-01-0 (50)  
 Chloride;

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Mo(III) EMF oth/un 25°C 3.00M U 1975ZSa (5236) 5

K(Mo(OH)+4Cl=Mo(OH)Cl4)=2.60

Medium: ethanoic acid

-----  
Mo(III) kin oth/un 25°C 1.0M U K1=1.03 1974SSd (5237) 6  
Medium: lithium p-toluenesulfonate

\*\*\*\*\*

FClBrI HL (541)  
Halides, comparative (for book data under ligand 80)

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(III) kin alc/w 22°C 100% U M 1960SHa (7409) 7

Metal:Mo++. Medium: EtOH. K(Mo6Cl8Cl6+6Br=Mo6Cl8Br6+6Cl)=-0.1

\*\*\*\*\*

OH- HL Hydroxide (57)  
Hydroxide;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(III) dis NaCl 20°C 1.00M U K1=12.0 B2=23.4 1978MMb (11754) 8  
B3=34.7

\*\*\*\*\*

S-- H2L Sulfide CAS 7783-06-4 (705)  
Sulfide;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(III) oth none 25°C 0 U 1988LIa (14420) 9

Kso(Mo2S3)=-107.8

\*Kso(Mo2S3)=-55.8

Derived from thermodynamic data and K(H+S=HS)=17.3.

\*\*\*\*\*

SCN- HL Thiocyanate CAS 463-56-9 (106)  
Thiocyanate;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(III) kin oth/un 25°C 2.00M U 1997NCa (15153) 10

K(Mo4S4(H2O)12+L)=3.11

K(Mo7S8(H2O)18+L)=2.94

Medium: Li-p-toluenesulfonate.

-----  
Mo(III) kin oth/un 25°C 2.00M U 1993HLa (15154) 11

K(Mo4S4+L)=3.11

Medium: Li toluene-p-sulfonic acid. For Mo(IV), K=3.72; for mixed Mo(III)/  
Mo(IV) (Mo4S4++++), K=3.48.

-----  
Mo(III) kin oth/un 25°C 1.0M U K1=5.0 1974SSd (15155) 12

medium:lithium p-toluenesulfonate

-----

Mo(III) sp oth/un ? 1.0M U K1=0.6 1972Kta (15156) 13  
Medium: p-toluenesulfonic acid

\*\*\*\*\*  
C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)  
Ethanedioic acid; (COOH)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(III) kin oth/un ? ? U K1=3.38 1956YAc (18966) 14  
\*\*\*\*\*  
C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)  
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(III) vlt oth/un 25°C 0.20M U K(?)=3.17 1962ZRa (31304) 15

Medium: 0.2 Na2SO4, 0.1 H2SO4, 0.04 KNO3  
\*\*\*\*\*  
C5H9N L t-Butylnitrile CAS 7188-38-7 (913)  
t-Butylcyanide;(CH3)3C.CN

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(III) con non-aq 40°C 100% U M 1992LIa (38452) 16  
K(MoL7+Cl)=3.4  
K(MoL7+Br)=3.18  
K(MoL7+I)=2.6  
K(MoL6Cl+Cl)=3.5

Medium: MeCN, 0.0063 M Bu4NClO4. K(MoL6Br+Br)=3.18, K(MoL6I+I)=3. Mo++  
\*\*\*\*\*  
C6H8O7 H3L Citric acid CAS 77-92-9 (95)  
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(III) vlt oth/un 25°C 0.20M U K(?)=3.47 1962ZRa (46178) 17

Medium: 0.2 Na2SO4,0.1 H2SO4,0.04 KNO3  
\*\*\*\*\*  
C8H19P L (6822)  
Di(t-Butyl)phosphine; ((CH3)3C)2PH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(III) nmr none ? 0.0 U T HM 1992BCc (63201) 18  
Method:NMR. Medium:toluene. DH(1,2-Mo2L2(NMe2)4 (anti-gauche isomerization))  
=-1.3 kJ mol-1, DS=-6.3 J K-1 mol-1. Data also for other phosphides

\*\*\*\*\*  
C10H10O2 HL Benzoylacetone CAS 93-91-4 (197)

1-Phenylbutane-1,3-dione; C6H5.CO.CH2.CO.CH3

```
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Mo(III)    dis NaCl   20°C  1.0M U           K1=5.71  B2=11.68  1978MMi (70752)  19
                                     B3=18.64
```

```
*****
C12H24O6          L      18-Crown-6          CAS 17455-13-9 (577)
1,4,7,10,13,16-Hexaoxacyclooctadecane;
```

```
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Mo(III)    cal alc/w  25°C  100% U  H    K1=2.44          1977ILb (83472)  20
Medium: Methanol. DH=20.0 kJ mol-1.
```

```
*****
C17H14N2O2        L              CAS 4551-69-3 (698)
4-Benzoyl-3-methyl-1-phenyl-2-pyrazolin-5-one;
```

```
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Mo(III)    dis NaCl   20°C  1.0M U           K1=6.55  B2=12.99  1978MMi (95893)  21
                                     B3=20.15
```

```
*****
C18H15P          L              CAS 603-35-0 (621)
Triphenylphosphine; (C6H5)3P
```

```
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Mo(III)    kin non-aq 35°C  100% U  M           Kout(Mo(CO)5(NHC5H10)+L)=2.78
1975EDa (97143)  22
```

```
Medium: hexane
*****
e-          HL      Electron          (442)
Electron;
```

```
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Mo(IV)     EMF oth/un  ?  4.50M U           1958CHb (681)  23
                                     K(Mo+e=Mo(III))=1.7(100 mV)
```

```
Medium: H2SO4
*****
CN-          HL      Cyanide          CAS 74-90-8 (230)
Cyanide;
```

```
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Mo(IV)     nmr KNO3   25°C  0.10M C           1994RLa (2741)  24
                                     *K(MoO(CN)4(H2O))=-9.88
```

Method: N.M.R.

Mo(IV) con oth/un 25°C dil U M 1974FIb (2742) 25  
 K(K+Mo(CN)8)=1.8  
 K(Me4N+Mo(CN)8)=2.5  
 K(Et4N+Mo(CN)8)=2.3

Mo(IV) gl none 25°C 0.0 U T H 1973BKa (2743) 26  
 K(MoOOH(CN)4+H)=8.81  
 K=8.86(30 C). K=8.90(35 C). K=8.97(40 C). K=9.04(45 C). K=9.13(50 C).  
 DH=23.4 kJ mol<sup>-1</sup>

Mo(IV) sp NaClO4 25°C var U 1973MHa (2744) 27  
 K(Fe+Mo(CN)8)=2.6

Mo(IV) sp NaClO4 25°C var U M 1971JSb (2745) 28  
 K(Fe+Mo(CN)8)=2.6

Mo(IV) sp oth/un 25°C var U M 1969KBc (2746) 29  
 K(UO2+Mo(CN)4(OH)3(H2O))=3.71

Mo(IV) sp oth/un 25°C var U M 1968DBb (2747) 30  
 K(VO+MoL4(OH)3H2O)=4.86

Mo(IV) gl oth/un 25°C 0.0 U 1968PNb (2748) 31  
 K(H+MoO2L4)=12.62  
 K(H+MoOOHL4)=9.98

Mo(IV) con oth/un 25°C dil U M 1958SEa (2749) 32  
 Ks(KAg2Y(s))=-13.96  
 Ks(Ag3Y(s))=-13.83  
 Ks(Mn3Y2(s))=-12.35  
 Ks(Fe3Y2(s))=-16.28

Y=MoSOHL4(H2O)2---. Ks(Co3Y2)=-13.92; Ks(Ni3Y2)=-18.23; Ks(Cu3Y2)=-18.46;  
 Ks(Zn3Y2)=-13.62; Ks(Cd3L2)=-18.32; Ks(Hg3Y2)=-18.73; Ks(Pb3Y2)=-18.52

\*\*\*\*\*  
 Cl- HL Chloride CAS 7647-01-0 (50)  
 Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(IV) kin oth/un 25°C 2.00M U 1993HLa (5238) 33  
 K(Mo4S4+L)=0.30

Medium: Li toluene-p-sulfonic acid. For mixed Mo(III)/Mo(IV) (Mo4S4++++)  
 \*\*\*\*\*

ClO4- HL Perchlorate CAS 7001-90-3 (287)  
 Perchlorate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(IV) kin oth/un ? 2.50M U K1=0.91 1952HSc (6326) 34  
 Medium: H2SO4

\*\*\*\*\*

NO3- HL Nitrate CAS 7697-37-2 (288)  
Nitrate;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(IV) sp NaClO4 20°C 2.0M U I K1=0.14 B2=-0.17 1967VDa (9773) 35  
Metal:MoO2++. Medium: HClO4. K1=0.10(I=1), 0.14(0.5); B2=-0.26(I=1),  
-0.08(I=0.5), corrected for assumed mononuclear hydrolysis

-----  
Mo(IV) ix NaClO4 20°C 1.0M U K1=0.15 B2=-0.15 1967VDa (9774) 36  
Metal:MoO2++. Medium: HClO4

-----  
Mo(IV) dis oth/un 20°C var U 1967VDb (9775) 37  
Kd(MoO2+2L+2TBP(org))=-0.95

Metal:MoO2++. Medium: HL var. Org=kerosene

\*\*\*\*\*

OH- HL Hydroxide (57)  
Hydroxide;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(IV) gl KCl 25°C 1.2M C 1998ARa (11755) 38  
\*K(MoO(H2O)(CN)4)=-9.88  
\*K(MoO(OH)(CN)4)=-14

Medium: KCl/KN03

-----  
Mo(IV) kin NaClO4 25°C 2.00M U 1993LMb (11756) 39  
\*K(Mo3Se4(H2O)9)=-0.49  
\*K(Mo3O3Se(H2O)9)=-0.36

Medium: LiClO4.

-----  
Mo(IV) sp NaClO4 25°C 2.00M U 1992RSb (11757) 40  
\*K(Mo3S4(H2O)9)=-0.74

Medium: 2.0 M LiClO4.

-----  
Mo(IV) EMF oth/un 16°C var U 1959LMa (11758) 41  
K(Mo(CN)4OH+OH)=8.10  
K(Mo(CN)4(OH)2+OH)=5.47  
K(Mo(CN)4(OH)3+OH)=1.55

\*\*\*\*\*

O2 L Oxygen CAS 7782-44-7 (83)  
Dioxygen, also oxide; O-- , and superoxide, O2-

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(IV) kin oth/un 25°C 2.0M U K1=2.56 1986HNa (12630) 42  
K(MoO2+Mo=Mo2O2)=2.73

Medium: Li-p-toluenesulphonate

\*\*\*\*\*

O2-- H2L Peroxide CAS 7772-84-1 (2813)  
Peroxide; -0.0-

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(IV) sp oth/un RT 0.04M M 1989CHb (12671) 43  
Medium: 0.04 M phosphate buffer.  $K(\text{MoO}_4+\text{H}_2\text{O}=\text{MoO}_2(\text{O}_2)\text{OH}+\text{OH})=-6.04$   
 $K(\text{MoO}_2(\text{O}_2)\text{OH}+\text{H}_2\text{O}=\text{MoO}(\text{O}_2)_2\text{OH}+\text{H}_2\text{O})=5.43$ . Also tris and tetra peroxy cpds.

\*\*\*\*\*  
P04--- H3L Phosphate CAS 7664-38-2 (176)  
Phosphate;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(IV) vlt NaClO4 25°C 1.00M U 1962ZRa (13249) 44  
 $K(\text{Mo(IV)}+\text{H3L})=1.16$

\*\*\*\*\*  
ReO4- HL Perrhenate (2581)  
Rhenate(VII), Perrhenate;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(IV) dis none 20°C 0.0 U 1977PRa (14105) 45  
 $K(\text{MoO}_2+\text{L})=1.43$

\*\*\*\*\*  
S-- H2L Sulfide CAS 7783-06-4 (705)  
Sulfide;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(IV) oth none 25°C 0 U 1988LIa (14421) 46  
 $K_{\text{so}}(\text{MoS}_2)=-75.6$   
 $*K_{\text{so}}(\text{MoS}_2)=-40.9$

Derived from thermodynamic data and  $K(\text{H}+\text{S}=\text{HS})=17.3$ .

\*\*\*\*\*  
SCN- HL Thiocyanate CAS 463-56-9 (106)  
Thiocyanate;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(IV) kin NaClO4 25°C 2.00M U 1993LMb (15157) 47  
 $K(\text{Mo}_3\text{Se}_4+\text{NCS})=3.38$

$K(\text{Mo}_3\text{OSe}_3+\text{NCS})=3.23$   
 $K(\text{Mo}_3\text{O}_2\text{Se}_2+\text{NCS})=3.66$   
 $K(\text{Mo}_3\text{O}_3\text{Se}+\text{NCS})=3.18$

$K(\text{Mo}_3\text{O}_4+\text{NCS})=2.99$ . Medium: 2.0 M HClO4.

-----  
Mo(IV) kin NaClO4 25°C 2.00M U 1993Vsa (15158) 48

$K(\text{Mo}_3\text{S}_4(\text{H}_2\text{O})_9+\text{L})=3.36$   
 $K(\text{Mo}_2\text{WS}_4(\text{H}_2\text{O})_9+\text{L})=3.48$

$$K(\text{MoW}_2\text{S}_4(\text{H}_2\text{O})_9+\text{L})=3.68$$

Medium: 2.0 M HClO<sub>4</sub>. For mixed Mo/W species data refer to L binding to Mo. Metals are Mo(IV) and W(IV).

-----  
 Mo(IV) kin oth/un 25°C 2.0M U T K1=2.54 19760Sa (15159) 49  
 Medium: LiClO<sub>4</sub>/HClO<sub>4</sub>, metal: MoO<sup>++</sup>. K1=2.89 (10 C); 2.73 (15 C); 2.61 (20 C)  
 \*\*\*\*\*  
 C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)  
 Ethanedioic acid; (COOH)<sub>2</sub>

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Mo(IV) kin oth/un 25°C 1.00M U 1984KRa (18967) 50  
 $K(\text{Mo}+\text{HL}=\text{MoL}+\text{H})=3.07$   
 \*\*\*\*\*  
 C2H4 L Ethylene CAS 74-85-1 (478)  
 Ethene; H<sub>2</sub>C:CH<sub>2</sub>

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Mo(IV) nmr non-aq 24°C 100% U M 1992HMa (19427) 51  
 $K(\text{MoOA}_3\text{Cl}_2+\text{L}=\text{MoOLB}_2\text{Cl}_2+\text{B})=-1.0$   
 Method:NMR. Medium:C6D6. A:PMePh<sub>2</sub>. When A=PMe<sub>3</sub>, K=-3.00  
 \*\*\*\*\*  
 C2H4O2S H2L Thioglycolic CAS 68-11-1 (596)  
 Mercaptoethanoic acid; HS.CH<sub>2</sub>.COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Mo(IV) sp oth/un 25°C ? U 1976LAg (20347) 52  
 $K(\text{MoO}(\text{OH})+\text{H}_2\text{L}=\text{MoO}_2\text{H}_2\text{L}+\text{H})=0.20$   
 $K(\text{MoO}(\text{OH})+\text{HL}=\text{MoO}_2\text{H}_2\text{L})=3.80$   
 \*\*\*\*\*  
 C2H6OS L DMSO CAS 67-68-5 (329)  
 Dimethylsulfoxide; (CH<sub>3</sub>)<sub>2</sub>.SO

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Mo(IV) kin non-aq 20°C 100% U I 1993BBc (22112) 53  
 $K(\text{MoOA}+\text{L})=1.22$   
 Medium: CH<sub>2</sub>Cl<sub>2</sub>. In DMF: K=1.90. A: S-methyl-3-(2-hydroxyphenyl)methylene-dithiocarbamate.  
 \*\*\*\*\*  
 C3H6O2S H2L Thiolactic acid CAS 79-42-5 (366)  
 2-Mercaptopropanoic acid; CH<sub>3</sub>.CH(SH).COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Mo(IV) sp oth/un 25°C ? U 1976LAg (25159) 54  
 $K(\text{MoO}(\text{OH})+\text{H}_2\text{L}=\text{MoO}_2\text{H}_2\text{L}+\text{H})=0.08$



\*\*\*\*\*

C3H7NO2S H2L Cysteine CAS 52-90-4 (96)  
2-Amino-3-mercaptopropanoic acid; H2N.CH(CH2.SH)COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(IV) sp oth/un RT 0.5M U 1977LAb (26807) 55  
K(MO(OH)+HL=M(OH)2L+H)=2.04

Medium: Na-toluenesulfonic acid

\*\*\*\*\*

C4H6O4S H3L Thiomalic acid CAS 70-49-5 (109)  
2-Mercaptosuccinic acid, 2-Sulfanyl-1,4-butanedioic acid; HOOC.CH(SH).CH2.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(IV) sp oth/un 25°C ? U 1976LAg (30347) 56  
K(MoO(OH)+H2L=MoO2H2L+H)=0.04

\*\*\*\*\*

C4H6O4S2 H4L CAS 2418-14-6 (4264)  
2,3-Dimercaptobutanedioic acid; HOOC.CH(SH).CH(SH).COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(IV) gl KNO3 25°C 0.10M C 1991HKb (30394) 57  
K(Mo3S7L3+H)=11.7  
K(Mo3S7HL3+H)=7.64  
K(Mo3S7H2L3+H)=6.77  
K(Mo3S7H3L3+H)=4.89

K(Mo3S7H4L3+H)=3.98, K(Mo3S7H5L3+H)=3.3

\*\*\*\*\*

C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)  
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(IV) vlt NaClO4 25°C 1.0M U 1962ZRa (31305) 58  
K(?)=2.06

\*\*\*\*\*

C5H6 HL Cyclopentadiene CAS 542-92-7 (4288)  
Cyclopentadiene; cyclo(-CH:CH.CH2.CH:CH-)

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(IV) nmr none 37°C 0.0 U 1991KKc (37080) 59  
\*K(MoL2)=-5.5  
\*K(MoH-1L2)=-8.5

\*\*\*\*\*

C5H9NO4 H2L MIDA CAS 4408-64-4 (190)  
N-Methyliminodiethanoic acid; CH3.N(CH2.COOH)2

-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	nmr	oth/un	35°C	? U				1966KUa (39266)	60
							K(MoO4+HL+H)=8.5 K(MoO3+L)=10.4 K(Mo7O24+7HL=7MoO3L+H)=8.9 K(MoO3L+H)=2.8		

K(2MoO3L+2H)=7

\*\*\*\*\*

C6H8O7 H3L Citric acid CAS 77-92-9 (95)  
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCC(=O)C(O)C(=O)O

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	vlt	NaClO4	25°C	1.0M U				1962ZRa (46179)	61
							K(?)=2.42		

Medium: HClO4

\*\*\*\*\*

C6H9N3O2 HL Histidine CAS 71-00-1 (1)  
2-Amino-3-(4'-imidazolyl)propanoic acid; Nc1ccc(C(=O)O)nn1

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	gl	KNO3	25°C	0.15M C				1981JJa (47588)	62
							B(MoO4+2H+A=MoO3A+H2O)=16.76		

\*\*\*\*\*

C7H14N4S L (6856)  
2,8-Dimethylnona-2,7-diene-3,4,6,7-tetraza-5-thione; CN(C)C1=CN(C)C=C1S

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	sp	alc/w	25°C	70% U			K1=3.49 B2=6.72	1991LGa (57845)	63

Complexes probably MoO2L and MoO2L2

\*\*\*\*\*

C9H18N4S L (6855)  
3,9-Dimethylundeca-3,8-diene-4,5,7,8-tetraza-6-thione; CN(C)C1=CN(C)C=C1S

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	sp	alc/w	25°C	70% U			K1=4.14 B2=8.02	1991LGa (67960)	64

Complexes probably MoO2L and MoO2L2

\*\*\*\*\*

C15H10O8 H6L Myricetin CAS 529-44-2 (4055)  
3,3',4',5,5',7-Hexahydroxyflavone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	sp	oth/un	20°C	? U				1965GKa (91026)	65
							K(MoO4+H6L=MoO3H4L)=4.62(?)		

\*\*\*\*\*

e- HL Electron (442)  
Electron;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(V) EMF none 30°C 0.00 U 1971ESb (682) 66  
K=2.96(89mV)

K: MoO2+ + 4H+ + 2e=Mo+++ + 2H2O

-----  
Mo(V) EMF oth/un 18°C 2.0M U 1952LAb (683) 67  
K=-8.7(green Mo(III), -250 mV)  
K=3.8(red Mo(III), 110 mV)

Medium:HCl. K: Mo(V)+2e=Mo(III)

-----  
Mo(V) EMF oth/un 20°C 8.75M U I 1941HGa (684) 68  
K(Mo+2e=Mo(III))=7.9(230 mV)

Medium:H2SO4. At I=4.7 M: K=3.3(95 mV), 2.25 M:1.0(30 mV); I=0.45:-0.3(10mV)

-----  
Mo(V) EMF oth/un 25°C 0.25M U I 1936KTa (685) 69  
K(Mo(CN)8+e)=13.46(796 mV)

Medium: KCl, KBr, KNO3. At I=0: K=12.28(726.0 mV)

-----  
Mo(V) EMF oth/un 0°C var U 1924COa (686) 70  
K(Mo(CN)8+e)=15.5(839 mV)

\*\*\*\*\*

Br- HL Bromide CAS 10035-10-6 (19)  
Bromide;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(V) oth oth/un 20°C var U T H 1972JRa (2134) 71  
K=13.4

K: Mo4(OH)4O4L12+4H2O=Mo4(OH)4O8L4+8H+8L)=13.7(1 C), 13.1(40 C).

DH(K)=-24.3 kJ mol-1, DS=173 J K-1 mol-1. Method: magnetic susceptibility

\*\*\*\*\*

Cl- HL Chloride CAS 7647-01-0 (50)  
Chloride;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(V) sp NaClO4 25°C 5.0M C 1984HSa (5239) 72  
K(Mo2O4Cl4+2H+4Cl=Mo2O3Cl8+H2O)=0.30

K(Mo2O3Cl8+6H+2Cl=2H2MoOC15)=-7.82

-----  
Mo(V) oth KCl 20°C var U T H 1972JRa (5240) 73  
K=4.01

Medium: HCl. K: 2MoOC15+4H2O=(MoOC12(H2O))2+4H+6Cl. K=3.82(30 C), 3.65(40 C)  
DH=-31.4 kJ mol-1, DS=39.7 J K-1 mol-1. Method: magnetic susceptibility

-----

Mo(V) sp KCl 26°C 4.0M U TIH 1971YTa (5241) 74  
K=3.07

Medium: HCl.K:  $2\text{MoOCl}_5 + \text{H}_2\text{O} = (\text{MoOCl}_4)_2\text{O} + 2\text{H} + 2\text{Cl}$ . DH(K)=-59.8 kJ mol<sup>-1</sup>. K=3.35 (59 C), 2.24(78 C), 1.86(93 C). I=6 M: DH(K)=-37.6. K=2.50(24 C), 1.27(92 C)

Mo(V) oth KCl 40°C var U T 1967JRa (5242) 75  
K(2MoOCl<sub>5</sub>+4H<sub>2</sub>O=X+4H+6Cl)=7.31

Method:magnetic susceptibility. Medium:HCl var. K=8.03(20 C), 7.64(30 C)  
X=Mo<sub>2</sub>O<sub>2</sub>(OH)<sub>4</sub>Cl<sub>4</sub>

Mo(V) sp KCl ? var U 1959BGi (5243) 76

K(MoO+3Cl)=-2.3

K(MoO<sub>2</sub>+2H+3Cl=MoOCl<sub>3</sub>+H<sub>2</sub>O)=-3.2

\*\*\*\*\*

FClBrI HL (541)

Halides, comparative (for book data under ligand 80)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) nmr oth/un 20°C 8.0M U 1966MDb (7410) 77

K: MoOCl<sub>5</sub>+nL=MoOCl<sub>5-n</sub>Ln+nCl. L=Br:K=-0.5(n=1), -0.7(n=2), -1.7(n=3), -3.2(n=4) or -4.3(n=5)

\*\*\*\*\*

OH- HL Hydroxide (57)

Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) gl NaCl 30°C 0.50M C 1991HYa (11759) 78

K(2MoO+3OH)=-0.89

K(3MoO+4OH)=1.00

K(4MoO+5OH)=2.98

Mo(V) sp oth/un 18°C 0.10M U 1984NGa (11760) 79

K(MoO<sub>2</sub>+OH=MoO<sub>2</sub>OH)=10.60

In 0.1 M HClO<sub>4</sub>/NaClO<sub>4</sub>; For 1.0 M HClO<sub>4</sub>/NaClO<sub>4</sub> K=10.98

Mo(V) sol oth/un 450°C 0.00 U 1980KKc (11761) 80

K(4MoO<sub>2</sub>+2H<sub>2</sub>O+O<sub>2</sub>=4MoO<sub>2</sub>(OH))=2.2

p(O<sub>2</sub>)=500 atm

Mo(V) sp KCl ? var U 1959BGi (11762) 81

K(MoO<sub>2</sub>+2H=MoO+H<sub>2</sub>O)=ca. -1

\*\*\*\*\*

SCN- HL Thiocyanate CAS 463-56-9 (106)

Thiocyanate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) sp mixed 20°C ? C 1986CZa (15160) 82

B(CuH-2L)=-7.88  
B(CuH-3L)=-15.12

Medium: DMSO/acetone

-----  
Mo(V) kin NaClO4 25°C 1.00M U M 1976CSa (15161) 83  
K(Mo2O4(C2O4)2+L)=0.74

By spectrophotometry: K=0.63

-----  
Mo(V) kin NaClO4 25°C 2.00M U T 1975STa (15162) 84  
K(Mo2O4+L=Mo2O4L)=2.38

Medium: LiClO4

-----  
Mo(V) sp non-aq ? 100% U K1=2.88 1970BRb (15163) 85  
Medium: (EtO)2PSSEt + EtOH(4:1)

-----  
Mo(V) nmr NaClO4 23°C 2.0M U M 1968MDf (15164) 86  
K(MoOL4+A=MoOL3A+L)=-1.64  
K(MoOL4+2A=MoOL2A2+2L)=-3.24  
K(MoOL4+3A=MoOLA3+3L)=-6.19

Medium: HClO4. A=(NH2)2CS

-----  
Mo(V) sp non-aq ? 100% U I K1=5.0 B2=9.40 1965ULa (15165) 87  
K3=4.0  
K4=3.4

Medium: Me2CO, Mo as MoCl5. In MeOH: K1=3.85

-----  
Mo(V) sp oth/un ? 3.25M U I 1959NAb (15166) 88  
K6?=1.35

Medium: H2SO4. In 3.1 M (NH4)2SO4 K3\*K4\*K5?=2.25

-----  
Mo(V) sp mixed ? 60% U K1=3.2 B2=6.2 1958PEb (15167) 89  
K3=ca.2  
K4=-1.6

Medium: 60% w/w acetone/H2O

-----  
Mo(V) sp mixed 20°C 60% U K1=3.2 B2=6.2 1958PEb (15168) 90  
K3=1.85

Medium: 60% w/w acetone/H2O, 1 M HCl. Also by electrical migration

\*\*\*\*\*  
S04-- H2L Sulfate CAS 7664-93-9 (15)  
Sulfate;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(V) sp oth/un ? var U 1974RWa (16355) 91  
K(2MoO(HL)5+4H2O=MoO2(OH)2(H2O)2(HL)6+4HL+2H)=-9.0

\*\*\*\*\*  
CH4O L Methyl alcohol CAS 67-56-1 (597)  
Methanol; CH3.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(V)	EMF	alc/w	20°C	100%	U				1971GSa (17886)	92
$K(\text{Mo}+2\text{L}=\text{Mo}(\text{L}')_2+2\text{H}) > 1$ $K(\text{Mo}(\text{L}')_2+2\text{L}'=\text{Mo}(\text{L}')_4)=24.35$										
Medium: MeOH, 1 M Me4NCl. L'=H-1L										
*****										
C3H7NO2S		H2L		Cysteine				CAS 52-90-4	(96)	
2-Amino-3-mercaptopropanoic acid; H2N.CH(CH2.SH)COOH										
-----										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(V)	gl	NaClO4	25°C	0.50M	C				1998MSa (26808)	93
$K(\text{Mo}2\text{O}2\text{S}2+2\text{L})=36.24$ $K(\text{Mo}2\text{O}2\text{S}2+\text{H}+2\text{L})=38.63$ $K(\text{Mo}2\text{O}2\text{S}2+2\text{H}+2\text{L})=40.63$ $*K(\text{Mo}2\text{O}2\text{S}2\text{H}2\text{L}2)=-2.00$										
*K(Mo2O2S2HL2)=-2.39										
*****										
C5H5N		L		Pyridine				CAS 110-86-1	(31)	
Pyridine, Azine;										
-----										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(V)	sp	NaClO4	25°C	1.00M	U	M			1976CSa (36659)	94
$K(\text{Mo}2\text{O}4(\text{C}2\text{O}4)_2+\text{L})=1.60$										
By kinetics: K=1.83										
*****										
C5H8O2		HL		Acetylacetone				CAS 123-54-6	(164)	
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3										
-----										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(V)	dis	oth/un	?	?	U				1968ABb (38030)	95
$K(\text{MoO}(\text{OH})+2\text{L})=20.36$										
*****										
C6H2O4Br2		H2L		Bromanilic acid				CAS 4379-59-6	(1279)	
3,6-Dibromo-2,5-dihydroxy-1,4-benzoquinone;										
-----										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(V)	sp	NaClO4	25°C	2.00M	U				1986VPa (42037)	96
$K(\text{MoO}2+\text{H}2\text{L}=\text{MoO}2\text{L}+2\text{H})=3.58$										
*****										
C10H8O8S2		H4L		Chromotropic ac				CAS 148-25-4	(1875)	
1,8-Dihydroxynaphthalene-3,6-disulfonic acid;										
-----										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(V)	sp	oth/un	?	0.10M	U				1968TKd (69960)	97

K(MoO+2H2L)=8.46

Metal: MoO+++

\*\*\*\*\*

C10H16N2O8 H4L EDTA CAS 60-00-4 (120)

1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(V) EMF NaClO4 ? 0.10M U 1970HPa (73970) 98  
K(Mo2O4+H2L)=11.24  
K(Mo2O4+L) > 27.4

-----  
Mo(V) sp none ? 0.0 U K1=6.36 1958SAa (73971) 99  
-----

C13H11NO2 HL CAS 304-88-1 (181)

N-Phenylbenzohydroxamic acid; C6H5.CO.N(C6H5).OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(V) dis NaClO4 20°C ? U K1=14.0 B2=28.12 1967DBa (85166) 100  
K(MoO(OH)+L)=11.83  
K(MoO(OH)+2L)=23.31

\*\*\*\*\*

C13H15NO7 H3L CAS 98531-21-6 (8057)

2-Hydroxybenzylamine-N,N,O-triethanoic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(V) gl NaCl 30°C 0.50M C 1991HYa (85760) 101  
K(MoO+H+L)=14.85  
K(MoO+2H+2L)=28.51

\*\*\*\*\*

e- HL Electron (442)  
Electron;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) oth none 25°C 0.00 U 19740Ha (687) 102  
K=-92.30(-0.910V)

K: MoO4-- + 4H2O + 6e=Mo(s) + 8OH-; method:combination of thermodynamic data

-----  
Mo(VI) sp oth/un 25°C 8.6MM U 1964ANb (688) 103  
K=0.4

Medium 8.6M HBr. K:Mo(VI)=(MoOBr4-)<sub>2</sub> + Br3-

-----  
Mo(VI) oth none 25°C 0.00 U 1956GHa (689) 104  
K=-93.3(-0.92V)

K: MoO4-- + 4H2O + 6e=Mo(s) + 8OH-; method:combination of thermodynamic data

-----  
Mo(VI) EMF oth/un 30°C 0.0 U 1953EEa (690) 105

K(MoO2+2H+e)=8.0(482.6 mV)

-----  
Mo(VI) EMF oth/un 18°C 2.0M U 1952LAb (691) 106  
K(Mo+e=Mo(V))=9.2(530 mV)  
-----

Mo(VI) oth none 25°C 0.0 U 1952LAb (692) 107  
K=-106.1(-1050 mV)  
K: MoO4+4H2O+6e=Mo(s)+8OH from thermodynamic data  
-----

Mo(VI) EMF oth/un 20°C 5.0M U I 1941HGa (693) 108  
K(Mo+e=Mo(V))=9.1(530 mV)  
Medium:HCl. In H2SO4: 9.25 M: K=9.1(530 mV), 4.85 M: K=81.(470 mV),  
2.35 M: K=7.4(430 mV), 0.5 M: K=7.0(405 mV)  
-----

\*\*\*\*\*  
Cl- HL Chloride CAS 7647-01-0 (50)  
Chloride;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(VI) sp NaCl 25°C 4.0M C 1983HHa (5244) 109  
Medium: HCl. K(Mo(OH)5(H2O)+H+2Cl=Mo(OH)4Cl2+2H2O)=-0.54  
K(Mo(OH)4Cl2+H+Cl=Mo(OH)3Cl3+H2O)=-1.44  
-----

Mo(VI) sp KCl 25°C var U 1966RCa (5245) 110  
K=-0.89  
K'=-1.42  
K(Mo2Cl2+Cl)=-2.64  
Medium: HCl. K: Mo(OH)5H2O+Cl=MoCl(OH)5+H2O). K': Mo(OH)5Cl+H+L=Mo2Cl2+3H2O  
(Mo(OH)5H2O=H3MoO4(H2O)2). HMoO4: K(H)1=4.21, K2=4.00, K3=0.93  
-----

Mo(VI) ix none 25°C 0.0 U K1=-0.3 B2=-0.8 1964PCa (5246) 111  
K3=-1.89  
-----

Mo(VI) sp NaClO4 ? 5.30M U 1959CSa (5247) 112  
K(HMo2O6+3H+4Cl=2MoO2Cl2+2H2O)=-4.1  
-----

\*\*\*\*\*  
F- HL Fluoride CAS 7644-39-3 (201)  
Fluoride;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(VI) ix oth/un ? ? U 1973PMa (7028) 113  
K(MoO2F2+F)=2.91  
K(MoO2F3+F)=3.83  
-----

Mo(VI) sp oth/un ? var U 1967KKb (7029) 114  
K(H2MoO4+F=MoO3F+H2O)=4.48  
K(H2MoO4+4F=MoO2F4(+2H))=10.58  
-----

Mo(VI) con non-aq -5°C 100% U 1960NVa (7030) 115  
-----





Medium: liquid HF, m units

\*\*\*\*\*

NH<sub>2</sub>SO<sub>3</sub>- H<sub>2</sub>L Sulfamate CAS 5329-14-6 (452)  
Sulfamate;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp oth/un 25°C ? U 1958SAc (8800) 116  
 $K(2\text{H} + 2\text{L} + \text{MoO}_4 = \text{MoO}_3\text{L}_2 + \text{H}_2\text{O}) = -7 ?$

\*\*\*\*\*

NH<sub>3</sub> L Hydroxylamine; CAS 5470-11-1 (1808)  
Hydroxylamine; NH<sub>2</sub>.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl oth/un 20°C dil U 1968JDa (9268) 117  
 $K(\text{HMo}_{11036}\text{L} + \text{H}) = 3.84$

\*\*\*\*\*

NO<sub>3</sub>- HL Nitrate CAS 7697-37-2 (288)  
Nitrate;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) vlt NaClO<sub>4</sub> 25°C 0.50M U 1983ZZa (9776) 118  
 $K(\text{MoO}_2 + \text{L}) = 0.32$

-----  
Mo(VI) dis non-aq 25°C 100% U I 1970CMb (9777) 119  
 $K(\text{Mo}(\text{OH})_6 + \text{HL}) = 2.26$

Medium: TBP, 0.5 M KNO<sub>3</sub>. In 1 M KNO<sub>3</sub>, K=2.22

\*\*\*\*\*

OH- HL Hydroxide (57)  
Hydroxide;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp NaClO<sub>4</sub> 25°C 0.0 U H 2000CHa (11763) 120  
 $K(\text{MoO}_3 + \text{H} = \text{MoO}_2(\text{OH})) = 0.95$   
 $K(\text{MoO}_2(\text{OH}) + \text{H} = \text{MoO}_2) = -1.18$   
 $K(2\text{MoO}_2(\text{OH}) = \text{Mo}_2\text{O}_5 + \text{H}_2\text{O}) = 1.99$

Medium: 0.4-8.1 M HClO<sub>4</sub>. DH( $K(\text{MoO}_2(\text{OH}) + \text{H}) = -21.4 \text{ kJ mol}^{-1}$ ).  
DS = -90 J K<sup>-1</sup> mol<sup>-1</sup> in 6.3 M HClO<sub>4</sub>. DH( $\text{Mo}_2\text{O}_5$ ) = -30.5, DS = -63.

-----  
Mo(VI) sp NaClO<sub>4</sub> 25°C 1.0M C 1988CDc (11764) 121

B(2,1)=7.11  
B(11,7)=62.9  
B(12,8)=72.0

Method: distribution between HClO<sub>4</sub>/NaClO<sub>4</sub> solution and tri-n-butyl phosphate. B(p,q):  $\text{pH} + q\text{MoO}_4 = \text{Hp}(\text{MoO}_4)_q$

-----

Mo(VI) sp NaClO4 25°C 4.0M C 1983HUa (11765) 122  
K(Mo(OH)6+H=Mo(OH)5(H2O))=1.35

Mo(VI) gl NaCl 25°C 0.10M C TIH K1=3.47 B2=7.21 1976CRa (11766) 123  
At I=0 by extrapolation: K1=3.55, B2=7.20

Mo(VI) ix NaCl 25°C 0.70M U 1976SKb (11767) 124  
K(MoO2+OH)=12.04  
B(MoO2+2OH)=23.60  
B(MoO2+3OH)=31.78  
B(2MoO2+3OH)=36.78

Mo(VI) sp KNO3 25°C 0.10M U I K1=13.81 B2=27.06 1971NSd (11768) 125  
K3=12.48

Mo(VI)=MoO2++. K1=14.06, K2=13.52, K3=12.75(I=0.3). K1=14.17, K2=13.62,  
K3=12.79(I=0.5). K1=14.68, K2=14.15, K3=13.37(I=1)

\*\*\*\*\*  
O2-- H2L Peroxide CAS 7772-84-1 (2813)  
Peroxide; -0.0-

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) nmr oth/un 25°C 0.30M C T M 2002TAa (12672) 126  
B(1,1,2,0)=11.61  
B(2,1,2,0)=13.77  
B(2,1,2,1)=14.50  
B(2,2,4,0)=23.77

Method: O17 nmr. Medium: 0.30 M Na2SO4. Also data for 5 C.

B(p,q,r,s): pH+qMoO4+rH2O2+sSO4=Hp(MoO4)q(H2O2)r(SO4)s.

Mo(VI) nmr NaCl 5°C 0.60M C T 2002TAa (12673) 127  
B(1,1,2,0)=11.61  
B(2,1,2,0)=13.86  
B(2,1,2,1)=13.87  
B(2,2,4,0)=24.08

Method: O17 nmr. B(3,2,4,0)=26.23, B(2,2,6,0)=23.9.

B(p,q,r,s): pH+qMoO4+rH2O2+sCl=Hp(MoO4)q(H2O2)r(Cl)s.

Mo(VI) gl oth/un 25°C 0.30M C M 2002THa (12674) 128  
B(1,1,1,0)=8.53  
B(2,1,1,0)=11.22  
B(1,1,2,0)=11.61  
B(2,1,2,0)=13.77

Medium: 0.30 M Na2SO4. B(2,1,2,1)=14.50, B(2,2,4,0)=23.77, B(8,7,1,0)=  
56.71. B(p,q,r,s): pH+qMoO4+rH2O2+sSO4=Hp(MoO4)q(H2O2)r(SO4)s.

Mo(VI) gl oth/un 25°C 0.30M C M 2002THa (12675) 129  
B(9,7,1,0)=62.00  
B(10,7,1,0)=65.74  
B(11,7,1,0)=68.23

Medium: 0.30 M Na2SO4. B(p,q,r,s): pH+qMoO4+rH2O2+sSO4=Hp(MoO4)q(H2O2)r(SO4)s.

-----  
 Mo(VI) kin none 25°C 0.0 C 1990CSb (12676) 130  
 K(MoO4+H2L=MoO2L(OH)+OH)=-6.04  
 K(MoO2L(OH)+H2L=MoOL2(OH)+H2O)=5.43  
 -----

Mo(VI) sp NaClO4 25°C 1.00M U T K1=6.90 1987LSa (12677) 131  
 -----

Mo(VI) kin oth/un 25°C 0.05M U 1969AYa (12678) 132  
 K(H2MoO4+H2L=H2MoO5+H2O)=4.64  
 K'(Mo2O7+H2L=H2Mo2O9)=3.4

By spectrophotometry, K'=3.5, K(Mo2O7+2H2L=H2Mo2O1+2H)=3.3  
 -----

Mo(VI) oth oth/un 25°C var U 1965MOb (12679) 133  
 K(MoL4+H2L=HMoL4+HL)=-4.7  
 K(MoL4+H2O=HMoL4+OH)=ca.-9  
 -----

Mo(VI) gl oth/un ? var U 1958CSb (12680) 134  
 K(MoO4+2H2L=HMoO2L2+OH+H2O)=-3.6  
 -----

Mo(VI) gl oth/un ? var U 1958CSb (12681) 135  
 K(H+HMoO2L2)=2.5  
 K(H+MoO2L2)=9.15  
 -----

Mo(VI) sp oth/un ? 10.0M U 1955CSa (12682) 136  
 Medium: HClO4,H2SO4. K(HMo2O2L2HSO4+2H2L+H2O=H2Mo2O3L4+HSO4+5H)=8.50

\*\*\*\*\*

PO4--- H3L Phosphate CAS 7664-38-2 (176)  
 Phosphate;

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Mo(VI) gl NaCl 25°C 0.60M C M 2000SAa (13250) 137  
 B(22,11,1,1)=141.5  
 B(21,10,2,1)=146.9  
 B(22,10,2,1)=147.2  
 B(20,9,3,1)=147.1

B(p,q,r,s): pH+qMoO4+rHV04+sHPO4. Additional methods: 31P and 51V nmr.  
 B(21,9,3,1)=150.2, B(22,9,3,1)=151.7, B(23,9,3,1)=152.2.  
 -----

Mo(VI) sp NaClO4 25°C 3.00M C 1980LPe (13251) 138  
 K(19H+11MoO4+HPO4)=125.96  
 -----

Mo(VI) gl NaClO4 25°C 3.00M C 1975PEb (13252) 139  
 B(8,5,2)=61.97  
 B(9,5,2)=67.12  
 B(10,5,2)=70.69  
 B(14,9,1)=98.41

B(15,9,1)=102.83; B(16,9,1)=105.85; B(17,9,1)=106.85;

B(p,q,r):  $pH + qMoO_4 + rHPO_4 = Hp(MoO_4)q(HP_4)r$

-----  
Mo(VI) sp oth/un 25°C var U M 1969SAb (13253) 140  
K((MoO2)12(H3L)H-27)=-14.9  
-----

Mo(VI) vlt oth/un 25°C var U 1961YBb (13254) 141  
K(H2MoO4+H3L=MoO2HL+2H2O)=3.16  
K(MoO2HL+H3L=MoO2(H2L)2)=0.19  
-----

Mo(VI) kin oth/un 22°C 0.48M U 1956YAc (13255) 142  
K(H2MoO4+H3L=MoO2L+H+2H2O)=1.02; K(H2MoO4+2H3L=MoO2L2+4H+2H2O)=2.64  
-----

\*\*\*\*\*  
S-- H2L Sulfide CAS 7783-06-4 (705)  
Sulfide;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) nmr oth/un 25°C ? U 1988BHa (14422) 143  
K(MoO4+HS+H=MoO3S+H2O)=-10.80  
K(MoO3S+HS+H=MoO2S2+aq)=-10.41  
K(MoO2S2+HS+H=MoO3S+aq)=-10.05  
K(MoO3S+HS+H=MoS4+H2O)=-9.49  
-----

Mo(VI) sol oth/un 60°C dil U T 1968SJB (14423) 144  
Kso(Tl2MoO4S)=-11.36  
K=-12.42(25 C), -11.87(40 C), -11.69(50 C)  
-----

\*\*\*\*\*  
SCN- HL Thiocyanate CAS 463-56-9 (106)  
Thiocyanate;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) nmr oth/un ? var U M 1969MDb (15169) 145  
K(MoOL4+A=MoOL3A+L)=-1.5  
K(MoOL4+2A=MoOL2A2+2L)=-3.1  
K(MoOL4+3A=MoOLA3+3L)=-5.1  
K(MoOL4+4A=MoOA4+4L)=-7.6  
-----

A=Br-. Other ternary complexes also reported. Method: esr  
\*\*\*\*\*

S04-- H2L Sulfate CAS 7664-93-9 (15)  
Sulfate;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp NaClO4 25°C 4.0M C 1983HUa (16356) 146  
K(Mo(OH)5+2HL=Mo(OH)4L2)=0.52  
-----

K:  $Mo(OH)_5(H_2O) + 2HSO_4 = Mo(OH)_4(SO_4)_2 + H$

-----  
Mo(VI) sp NaClO4 ? 8.0M U 1959CSa (16357) 147  
K(HMo2O6+3H+2HL=2MoO3LH3)=-4.74  
-----

\*\*\*\*\*  
 SiO3-- H2L Silicate CAS 7699-41-4 (747)  
 Silicate; SiO2(OH)2--

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) sp NaCl04 25°C 1.0M U 1982KCb (17215) 148

Keff=32.2  
 Measured at pH 1.2 in HNO3. Keff=(12Mo+Si(OH)4). Keff=31.9 in HClO4;  
 Keff=31.6 in H2SO4

\*\*\*\*\*  
 V04--- H3L CAS 15457-75-7 (1586)  
 Vanadate; VO2(OH)3-- or polymers

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) gl NaCl 25°C 0.60M C 1991HPa (17383) 149

B(15,1,9)=134.56  
 B(16,1,9)=137.33  
 B(16,2,8)=133.0  
 B(13,4,5)=105.22

B(p,q,r): pH+q(MoO4--)+r(HVO4--). B(14,4,5)=107.58, B(8,4,2)=60.37,  
 B(9,4,2)=64.11, B(9,5,1)=59.14, B(15,9,1)=95.0, B(11,7,1)=74.63.

-----  
 Mo(VI) gl NaCl 25°C 0.60M C 1989HPa (17384) 150

B(15,1,9)=134.56  
 B(16,1,9)=137.33  
 B(16,2,8)=132.97

From combined emf/nmr study. Also pK(16,1,9)=2.77.  
 B(p,q,r): pH+q[MoO4]+r[HVO4]=Hp[MoO4]q[HVO4]r.

\*\*\*\*\*  
 W04-- H2L Tungstate CAS 13783-36-3 (445)  
 Tungstate;

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) dis oth/un 25°C U T H 1974PRa (17441) 151

K'=5.77

K: MoO4 + HCrO4. 60 C; K'=4.65. DH=-32 kJ mol-1

\*\*\*\*\*  
 CH2O2 HL Formic acid CAS 64-18-6 (37)  
 Methanoic acid; H.COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) ix oth/un ? 0.05M U K1=1.02 B2=2.20 1970SHa (17625) 152

B3=2.78  
 B4=4.93

Medium: 0.01-0.05 HL. Metal ion: MoO2++. pH 2.5

\*\*\*\*\*

CH4O L Methyl alcohol CAS 67-56-1 (597)  
Methanol; CH3.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) nmr oth/un -70°C ? U M 1971BPg (17887) 153  
K(MoO2F2(H2O)2+L)=0.08  
K(MoO2F2(H2O)L+L)=-0.60

\*\*\*\*\*  
CH5AsO3 H2L Me-Arsonic acid CAS 124-58-3 (585)  
Methylarsonic acid; CH3.AsO3H2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) nmr NaCl 22°C 1.00M U 1997KYa (17966) 154  
B(11,6,2)=83.4  
B(12,6,2)=88.3  
B(10,5,2)=75.2

B(p,q,r): pH + qMoO4 + rCH3AsO3 = Hp(MoO4)q(CH3AsO3)r

\*\*\*\*\*  
CH5O3P H2L CAS 13590-71-1 (1752)  
Methylphosphonic acid; CH3.PO3H2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl NaCl 25°C 1.0M C 1998KYa (18130) 155  
B(10,5,2)=69.51  
B(11,5,2)=71.07  
B(11,7,1)=72.69  
B(12,7,1)=76.23

Additional method: nmr. B(p,q,r): pH+qMoO4+rL=Hp(MoO4)qLr.

B(12,6,1)=70.31.

\*\*\*\*\*  
C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)  
Ethanedioic acid; (COOH)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl NaCl 25°C 1.0M C H 1986CHa (18968) 156  
B(MoH2L)=13.619  
B(Mo2H5L2)=31.20  
B(Mo2H6L2)=34.08

Mo=MoO4--. DH(MoH2L)=-59.5, DH(Mo2H5L2)=-123.0, DH(Mo2H6L2)=-117.0 kJ mol<sup>-1</sup>

-----  
Mo(VI) gl KNO3 25°C 0.15M C 1984JJa (18969) 157  
K(MoO4+2H+L=MoO3L+H2O)=13.816

-----  
Mo(VI) sp NaClO4 30°C 1.00M U 1981BCb (18970) 158  
K(MoO4+2L+2H)=15.52  
K(2MoO2(OH)2L2+2H)=16.5

$K(\text{Mo}_2\text{O}_5(\text{OH})_2\text{L}_2+\text{H})=14.6$

-----  
Mo(VI) gl KNO3 21°C 0.22M C 1978MBc (18971) 159  
 $K(\text{MoO}_4+2\text{H}+\text{L}=\text{MoO}_3\text{L}+\text{H}_2\text{O})=13.98$

Medium pH 5-7

-----  
Mo(VI) oth oth/un ? ? U K1=1.57 1969SHd (18972) 160  
Metal ion is MoO2++

-----  
Mo(VI) dis NaClO4 20°C 0.10M U 1963STc (18973) 161  
 $K(\text{H}_2\text{MoO}_4+2\text{HL})=7.37$

-----  
Mo(VI) vlt oth/un 25°C 0.11M U I 1961YBa (18974) 162  
 $K(\text{H}_2\text{MoO}_4+\text{H}_2\text{L})=3.91$

$K=3.80(I=0.179)$ ,  $3.49(I=0.345)$

\*\*\*\*\*

C2H4N2S2 L Rubeanic acid CAS 79-40-3 (2782)

Dithiooxamide; H2N.CS.CS.NH2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(VI) sp mixed 90°C 80% U K1=23.91 1975WHb (19453) 163  
Medium: 80% 2-propanol/H2O

\*\*\*\*\*

C2H4O3 HL Glycolic acid CAS 79-14-1 (33)

2-Hydroxyethanoic acid; HO.CH2.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(VI) nmr oth/un -70°C 0.10M U M 1971BPg (20584) 164

$K(\text{MoO}_2\text{F}_2(\text{H}_2\text{O})_2+\text{L}=\text{MoO}_2\text{F}_2(\text{H}_2\text{O})\text{L}+\text{H}_2\text{O})=-0.07$ ,

$K(\text{MoO}_2\text{F}_2(\text{H}_2\text{O})\text{L}+\text{L}=\text{MoO}_2\text{F}_2\text{L}_2+\text{H}_2\text{O})=-0.70$ .

\*\*\*\*\*

C2H5NO2 HL Acetohydroxamic CAS 546-88-3 (2766)

Acetohydroxamic acid, N-Hydroxyacetamide; CH3.CO.NHOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(VI) gl KCl 25°C 0.20M C 1998FMa (21814) 165  
 $K(\text{MoO}_4+\text{L}+2\text{H}=\text{MoO}_3\text{L}+\text{H}_2\text{O})=17.16$

$K(\text{MoO}_4+2\text{L}+4\text{H}=\text{MoO}_2\text{L}_2+2\text{H}_2\text{O})=32.46$

\*\*\*\*\*

C2H6N2O2 HL CAS 5549-80-4 (833)

2-Amino-N-hydroxyacetamide, Glycine hydroxamic acid; H2N.CH2.CO.NH.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(VI) gl KCl 25°C 0.20M C 1999FCa (21994) 166

$K(\text{MoO}_4+6\text{H}+2\text{L}=\text{MoO}_2\text{H}_2\text{L}_2)=47.06$

$K(\text{MoO}_4+3\text{H}+\text{L}=\text{MoO}_3\text{HL})=24.43$

$$K(\text{MoO}_4+2\text{H}+\text{L}=\text{MoO}_3\text{L})=19.15$$

\*\*\*\*\*  
 C2H7O2As HL Cacodylic acid CAS 75-60-5 (586)  
 Dimethylarsinic acid; (CH3)2.AsO2H

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) gl NaCl 25°C 0.60M C 1997KSc (22537) 167  
 $K(4\text{MoO}_4+\text{L}+7\text{H})=46.20$

Additional method: nmr

\*\*\*\*\*  
 C3H4O4 H2L Malonic acid CAS 141-82-2 (79)  
 Propanedioic acid; CH2(COOH)2

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) sp NaClO4 23°C 1.0M C 1983BCd (24501) 168  
 $K(2\text{MoO}_4+2\text{H}+2\text{L}=\text{Mo}_2\text{O}_5(\text{OH})_2\text{L}_2+\text{H}_2\text{O})=6.96$   
 $K(\text{Mo}_2\text{O}_5(\text{OH})_2\text{L}_2+\text{H}=\text{Mo}_2\text{O}_5(\text{OH})\text{L}_2+\text{H}_2\text{O})=8.64$

\*\*\*\*\*  
 C3H6O2S H2L CAS 107-96-0 (437)  
 3-Mercaptopropanoic acid; HS.CH2.CH2.COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) sp oth/un 25°C ? U 1963SCe (25219) 169  
 $K(\text{MoO}_4+3\text{HL}=\text{MoO}_3\text{L}_3+3\text{OH})=23(?)$

Medium: acetate buffer

\*\*\*\*\*  
 C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)  
 L-2-Hydroxypropanoic acid; CH3.CH(OH).COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) gl NaCl 25°C 1.00M C H 1993CKb (25484) 170  
 $B(1,2,1)=7.46$   
 $B(1,2,2)=15.71$   
 $B(1,2,3)=16.78$   
 $B(1,1,2)=11.76$

$B(p,q,r): p\text{MoO}_4 + q\text{HL} + r\text{H} = (\text{MoO}_4)_p\text{L}_q\text{H}_r + r.$   $B(1,1,3)=12.66, B(2,2,2)=16.07,$   
 $B(2,2,3)=21.70, B(2,2,4)=24.97, B(2,1,3)=18.44;$  other  $B(p,q,r);$  also DH

-----  
 Mo(VI) sp NaClO4 25°C 1.0M C 1983BCc (25485) 171  
 $K(\text{MoO}_4+2\text{HL}+2\text{H}=\text{MoO}_2\text{L}_2)=-3.5$

$K(2\text{MoO}_2\text{L}_2+2\text{H}+3\text{H}_2\text{O}=\text{Mo}_2\text{O}_5\text{L}_2(\text{H}_2\text{O})_2+2\text{H}_2\text{L})=-8.6$

$K(\text{Mo}_2\text{O}_5\text{L}_2(\text{H}_2\text{O})_2+\text{H}=\text{Mo}_2\text{O}_3(\text{OH})_3\text{L}_2+\text{H}_2\text{O})=-7.6$

\*\*\*\*\*  
 C3H7NO2 HL B-Alanine CAS 107-95-9 (575)  
 3-Aminopropanoic acid; H2N.CH2.CH2.COOH



Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C				1998FMa (26467)	172
K(MoO4+2L+4H=MoO2L2+2H2O)=33.26										
*****										
C3H7NO2S		H2L		Cysteine				CAS 52-90-4 (96)		
2-Amino-3-mercaptopropanoic acid; H2N.CH(CH2.SH)COOH										
Mo(VI)	sp	NaClO4	25°C	0.10M	C	I			2003GDa (26809)	173
K(MoO4+L+2H=MoO3L+H2O)=21.4										
Data for 0.1-1.0 M NaClO4. K=21.1 (I=0.30), 20.9 (I=0.50), 20.7 (I=0.7), 21.2 (I=1.0).										
Mo(VI)	sp	NaCl	18°C	1.00M	U				1990CJa (26810)	174
K(MoO4+L+2H=MoO3L+H2O)=18.8										
Mo(VI)	sp	oth/un	25°C		?	U			1963SCe (26811)	175
K(MoO4+3HL=MoOL3+3OH)=18(?)										
Medium: acetate buffer. K(?) Mo(V)=6.0										
*****										
C3H8N2O2		HL		Ala-hydroxamic				CAS 16707-85-0 (1582)		
2-Amino-N-hydroxypropanamide, Alanine hydroxamic acid; CH3.CH(NH2).CO.NH.OH										
Mo(VI)	gl	KCl	25°C	0.20M	C				1999FCa (27580)	176
K(MoO4+6H+2L=MoO2H2L2)=45.90										
K(MoO4+3H+L=MoO3HL)=24.00										
K(MoO4+2H+L=MoO3L)=18.65										
*****										
C3H8N2O2		HL						(6039)		
Sarcosinehydroxamic acid; CH3.NH.CH2.CO.NH.OH										
Mo(VI)	gl	KCl	25°C	0.20M	C				1999FCa (27586)	177
K(MoO4+6H+2L=MoO2H2L2)=47.04										
K(MoO4+3H+L=MoO3HL)=24.52										
K(MoO4+2H+L=MoO3L)=19.48										
*****										
C3H8N2O2		HL						(6666)		
beta-Alaninehydroxamic acid; NH2.CH2.CH2.CO.NHOH										
Mo(VI)	gl	KCl	25°C	0.20M	C				1998FMa (27608)	178
K(MoO4+L+3H=MoO3HL+H2O)=25.81										
K(MoO4+2L+6H=MoO2H2L2+2H2O)=49.76										

\*\*\*\*\*  
 C3H8O L n-Propanol CAS 71-23-8 (1914)  
 1-Propanol; CH3.CH2.CH2.OH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) nmr mixed -80°C ? U M 1971BPg (27644) 179  
 Medium: aq.propanol. -80 - +30 C  
 $K(\text{MoO}_2\text{F}_2(\text{H}_2\text{O})_2+\text{L}=\text{MoO}_2\text{F}_2\text{L}(\text{H}_2\text{O})+\text{H}_2\text{O})=-0.09$ ;  $K(\text{MoO}_2\text{F}_2\text{L}(\text{H}_2\text{O})+\text{L}=\text{MoO}_2\text{F}_2\text{L}_2+\text{H}_2\text{O})=-0.62$   
 \*\*\*\*\*

C3H8O L isoPropanol CAS 67-63-0 (2024)  
 2-Propanol; CH3.CH(OH).CH3

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) nmr mixed -80°C ? U M 1971BPg (27646) 180  
 Medium: aq.propan-2-ol. -80 - +30 C  
 $K(\text{MoO}_2\text{F}_2(\text{H}_2\text{O})_2+\text{L}=\text{MoO}_2\text{F}_2\text{L}(\text{H}_2\text{O})+\text{H}_2\text{O})=-0.60$ ;  $K(\text{MoO}_2\text{F}_2\text{L}(\text{H}_2\text{O})+\text{L}=\text{MoO}_2\text{F}_2\text{L}_2+\text{H}_2\text{O})=-1.15$   
 \*\*\*\*\*

C4H6O4 H2L Succinic acid CAS 110-15-6 (112)  
 1,4-Butanedioic acid; HOOC.CH2.CH2.COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) ix oth/un 22°C 0.10M U K1=1.20 1973SDa (29999) 181  
 Metal ion: MoO2++. pH 2.5  
 \*\*\*\*\*

C4H6O5 H2L Malic acid CAS 617-48-1 (393)  
 2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) gl NaCl 25°C 1.0M C H 1997CRb (30675) 182  
 $B(1,1,1)=7.47$   
 $B(1,1,2)=13.23$   
 $B(1,1,3)=15.87$   
 $B(1,2,2)=15.48$   
 $B(p,q,r): p\text{MoO}_4+q\text{L}+r\text{H}=(\text{MoO}_4)_p(\text{L})_q(\text{H})_r$ .  $B(1,2,3)=20.13$ ,  $B(1,2,4)=24.14$ ,  
 $B(4,2,8)=52.92$ ,  $B(4,2,9)=54.35$ ,  $B(4,2,10)=55.27$ , etc. DH by calorimetry.

-----  
 Mo(VI) gl KNO3 20°C 0.20M U 1986BHd (30676) 183  
 $K(\text{MoO}_2(\text{OH})_4+\text{L}+2\text{H}=\text{MoO}_2(\text{OH})\text{H}-1\text{L}+3\text{H}_2\text{O})=13.7$

-----  
 Mo(VI) oth oth/un RT ? U 1981BCd (30677) 184  
 $K(\text{MoO}_2\text{L}_2+2\text{H}+3\text{H}_2\text{O}=\text{Mo}_2\text{O}_5\text{L}_2\text{OH})=22.2$   
 $K(\text{MoO}_4+2\text{L}+2\text{H}=\text{MoO}_2(\text{OH})_2\text{L}_2)=13.9$ ,  $K(\text{MoO}_2\text{L}_2(\text{OH})_2+\text{H})=8.20$   
 \*\*\*\*\*

C4H6O6 H2L D-Tartaric acid CAS 147-71-7 (93)  
 D-Tartaric acid, D-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KNO3	20°C	0.20M	M				1982HHb (30977)	185

$K(\text{MoO}_2(\text{OH})_2 + 2\text{L} + 2\text{H} = \text{MoO}_2(\text{H}-1\text{L})_2 + 4\text{H}_2\text{O}) = 16.64$   
 $K(2\text{MoO}_2(\text{OH})_2 + 2\text{L} + 4\text{H} = (\text{MoO}_2)_2(\text{H}-2\text{L})_2 + 8\text{H}_2\text{O}) = 30.90$

\*\*\*\*\*  
 C4H6O6                                  H2L      L-Tartaric acid    CAS 87-69-4    (92)  
 L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid;  $\text{HOOC}.\text{CH}(\text{OH}).\text{CH}(\text{OH}).\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Mo(VI)	gl	NaCl	25°C	1.00M	C	H			1990CHc (31306)	186
								B(1,2,2)=16.33		
								B(1,3,2)=19.99		
								B(1,4,2)=22.92		
								B(2,4,1)=24.81		

$B(2,5,1)=26.16$ ,  $B(4,7,4)=56.22$ ,  $B(4,8,4)=61.53$ ,  $B(4,9,4)=63.98$ ,  $B(4,6,2)=43.4$ ,  $B(4,7,2)=48.2$ .  $B(p,q,r): p\text{MoO}_4 + q\text{H} + r\text{L}$

Mo(VI)	sp	NaCl	18°C	1.00M	U				1989CPa (31307)	187
								$K(\text{MoO}_4 + 2\text{L} + 2\text{H} = \text{MoO}_2\text{H} - 2\text{L}_2) = 16.2$		

Data obtained from circular dichroism measurements

Mo(VI)	oth	oth/un	?	?	M				1969PFa (31308)	188
								$K(\text{HMO}_4 + \text{HL}) = 2.36$		

Method: polarimetry

Mo(VI)	dis	NaClO4	20°C	0.10M	U				1963STc (31309)	189
								$K(\text{H}_2\text{MoO}_4 + 2\text{L}) = 7.66$ ?		

\*\*\*\*\*  
 C4H7NO4                                  H2L      Aspartic acid    CAS 56-84-8    (21)  
 Aminobutanedioic acid;  $\text{H}_2\text{N}.\text{CH}(\text{CH}_2.\text{COOH}).\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Mo(VI)	sp	NaClO4	25°C	0.10M	C	I			2001GZa (31891)	190
								$K(\text{MoO}_4 + 2\text{H} + \text{L} = \text{MoO}_3\text{L}) = 18.7$		

Data for 0.1-0.8 M NaClO4.

Mo(VI)	sp	NaClO4	25°C	0.15M	C				1995GZb (31892)	191
								$K(\text{MoO}_4 + 2\text{H} + \text{L} = \text{MoO}_3\text{L}) = 1.26$		

Mo(VI)	gl	NaCl	25°C	1.00M	C	H			1993CHa (31893)	192
--------	----	------	------	-------	---	---	--	--	-----------------	-----

$B(1,1,1)=6.54$   
 $B(1,1,2)=9.84$   
 $B(1,2,1)=6.57$   
 $B(1,2,2)=11.47$

$B(p,q,r): p\text{MoO}_4 + q\text{HL} + r\text{H} = (\text{MoO}_4)_p(\text{HL})_q\text{H}_r$ .  $B(2,1,4)=21.20$ ,  $B(2,1,5)=23.50$ ,  
 $B(4,4,9)=50.86$ ,  $B(4,4,10)=53.48$ ,  $B(2,4,8)=37.74$ . Also DH by calorimetry.

Mo(VI) sp oth/un 25°C 1.0M C 1982CPa (31894) 193  
K(MoO4+2H+L=MoO3L+H2O)=15.74  
Medium not defined. pH 6.0.

Mo(VI) gl oth/un 25°C 0.16M M 1977RGa (31895) 194  
B(MoO4+2H+L=MoO3L)=16.79

Mo(VI) gl NaClO4 25°C 0.10M U K1=9.29 B2=17.00 1972SSe (31896) 195  
K3=3.88

Metal ion: MoO2++

\*\*\*\*\*

C4H7NO4 H2L IDA CAS 142-73-4 (118)  
Iminodiethanoic acid; HN(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl NaClO4 25°C 3.0M U 1979ZLa (32304) 196  
B(MoO4+L+2H=MoO3L)=18.48

Mo(VI) gl oth/un 25°C 0.15M U 1966KRa (32305) 197  
K(MoO4+L+2H=MoO3L)=18.3

\*\*\*\*\*

C4H8N2O3 HL Asparagine CAS 70-47-3 (17)  
2-Aminobutanedioic acid 4-amide; H2N.CH(CH2.CO.NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl NaClO4 25°C 0.10M U K1=8.06 B2=15.29 1973TSe (32712) 198  
K3=3.45

\*\*\*\*\*

C4H9NO2 HL 2-Aminobutyric CAS 2835-81-6 (571)  
2-Aminobutanoic acid; CH3.CH2.CH(NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KNO3 25°C 0.10M U TIH K1=8.16 B2=15.61 1980SSf (33919) 199  
K3=3.62

\*\*\*\*\*

C4H10O L Isobutanol CAS 78-83-1 (4256)  
2-Methylpropan-1-ol; CH3.CH(CH3).CH2.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) nmr oth/un -70°C ? U M 1971BPg (34650) 200  
K(MoO2F2(H2O)2+L=MoO2F2(H2O)L+H2O)=-0.22  
K(MoO2F2(H2O)L+L=MoO2F2L2+H2O)=-0.72

\*\*\*\*\*

C4H10O L Butan-2-ol CAS 15892-23-6 (3572)  
sec-Butyl alcohol; C2H5.CH(OH)CH3

-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	nmr	oth/un	-70°C	?	U	M			1971BPg (34655)	201
K(MoO2F2(H2O)2+L=MoO2F2(H2O)L+H2O)=-0.24										
K(MoO2F2(H2O)L+L=MoO2F2L2+H2O)=-0.70										
*****										
C4H10O4		L		Erythritol				CAS 149-32-6	(2706)	
1,2,3,4-Tetrahydroxybutane; HO.CH2.CH(OH).CH(OH).CH2.OH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.10M	U				1990CVb (34712)	202
B(2,2,1)=15.20										
B(2,3,1)=19.50										
B(p,q,r): pMoO4+qH+rL=MoO4pHqLr. With (2R,2R)-butantetrol, D-threitol,										
B(2,2,1)=14.60, B(2,3,1)=18.20										
*****										
C4H11N3O2		HL						CAS 471915-94-3	(8550)	
2,4-Diamino-N-hydroxybutanamide;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C				2002ECa (35179)	203
K(MoO4+L+4H=MoO3H2L+H2O)=32.7										
K(MoO4+L+3H=MoO3HL+H2O)=27.8										
K(MoO4+2L+8H=MoO2H4L2+2H2O)=63.65.										
*****										
C4H14N2O6P2		H2L		EDDPO				CAS 1733-49-9	(2435)	
1,2-Diaminoethane-N,N'-bis(methylenephosphonic) acid; (H2O3P.CH2.NH.CH2)2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	NaCl	25°C	0.60M	C	M			1987YAa (35889)	204
B(10,5,2)=68.07										
B(11,5,2)=69.40										
B(11,7,1)=71.96										
B(12,7,1)=75.70										
B(12,6,1)=69.04. B(p,q,r)=pH+q(MoO4)+r(C6H5PO3)=Hp(MoO4)q(C6H5PO3)r										
*****										
C5H8O2		HL		Acetylacetone				CAS 123-54-6	(164)	
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	dis	oth/un	25°C	?	U			K1=10.57 B2=20.49	1968ABb (38031)	205
Metal ion: MoO2++										
*****										
C5H9NO4		H2L		Glutamic acid				CAS 56-86-0	(22)	
2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH										

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp NaClO4 25°C 0.10M C I 2000GZa (39100) 206

K(MoO4+2H+L=MoO3L+H2O)=17.54

Data for 0.1-1.0 M NaClO4. K=16.94 (I=0.4 M), 16.93 (I=0.5 M), 16.84 (I=0.7 M), 16.76 (I=1.0 M).

Mo(VI) gl KNO3 25°C 0.16M M 1977RGa (39101) 207

B(MoO4+2H+L=MoO3L)=16.79

Mo(VI) gl NaClO4 25°C 0.10M U K1=9.24 B2=16.84 1972SSe (39102) 208

K3=3.50

\*\*\*\*\*

C5H9NO4 H2L MIDA CAS 4408-64-4 (190)

N-Methyliminodiethanoic acid; CH3.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) nmr oth/un 35°C 1.50M U I 1966KRa (39267) 209

K(MoO4+L+HL=MoO3L)=18.2

At 25 C, using glass electrode, I=0.15 M, K=18.73

\*\*\*\*\*

C5H10N2O3 HL Glutamine CAS 56-85-9 (18)

2-Aminopentanedioic acid 5-amide; H2N.CH(CH2.CH2.CO.NH2)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl NaClO4 25°C 0.10M U K1=7.90 B2=14.83 1973TSe (39826) 210

K3=3.35

\*\*\*\*\*

C5H10O5 L CAS 1114-34-7 (6113)

D-Lyxose

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KCl 25°C 0.10M C 1989VCa (40339) 211

B((MoO4)2H2L)=14.98

B((MoO4)2H3L)=18.68

K((MoO4)2H2L+H)=3.70

M=MoO4

\*\*\*\*\*

C5H11NO2S HL CAS 93964-73-9 (3633)

Cysteine ethyl ester; H2N.CH(CH2.SH).CO.OCH2.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp oth/un 25°C ? U 1963SCe (41146) 212

K(MoO4+3HL=MoOL3+3OH)=23(?)

Medium: acetate buffer

\*\*\*\*\*

C5H12O L n-Pentanol CAS 71-41-0 (4298)  
1-Pentanol; CH3(CH2)4.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) nmr oth/un -70°C ? U M 1971BPg (41640) 213  
K(MoO2F2(H2O)2+L=MoO2F2(H2O)L+H2O)=-0.30  
K(MoO2F2(H2O)L+L=MoO2F2L2+H2O)=-0.72

\*\*\*\*\*  
C5H12O L Isopentanol CAS 34713-94-5 (4299)  
Isopentanol; CH3.CH2.CH(CH3).CH2.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) nmr oth/un -70°C 0.10M U M 1971BPg (41641) 214  
K(MoO2F2(H2O)2+L=MoO2F2(H2O)L+H2O)=-0.24  
K(MoO2F2(H2O)L+L=MoO2F2L2+H2O)=-0.70

\*\*\*\*\*  
C5H12O5 L Arabitol CAS 488-82-4 (5403)  
Arabitol; HO.CH2.HOCH.HCOH.HCOH.CH2.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl KCl 25°C 0.10M U 1990CVb (41676) 215  
B(2,2,1)=16.35  
B(2,3,1)=20.45

B(p,q,r): pMoO4+qH+rL=MoO4pHqLr

\*\*\*\*\*  
C5H12O5 L Ribitol CAS 488-81-3 (3009)  
Ribitol, Adonitol; HO.CH2.HCOH.HCOH.HCOH.CH2.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl KCl 25°C 0.10M U 1990CVb (41680) 216  
B(2,2,1)=15.55  
B(2,3,1)=19.45

B(p,q,r): pMoO4+qH+rL=MoO4pHqLr

\*\*\*\*\*  
C5H12O5 L Xylitol CAS 87-99-0 (2139)  
Xylitol; HO.CH2.HCOH.HOCH.HCOH.CH2.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl KCl 25°C 0.10M U 1990CVb (41688) 217  
B(2,2,1)=16.25  
B(2,3,1)=19.65

B(p,q,r): pMoO4+qH+rL=MoO4pHqLr

\*\*\*\*\*  
C6H2O4Br2 H2L Bromanilic acid CAS 4379-59-6 (1279)  
3,6-Dibromo-2,5-dihydroxy-1,4-benzoquinone;

```

-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Mo(VI)    sp  NaClO4 25°C 2.00M U                                1986VPa (42038) 218
                                                K(HMoO3+H2L=MoO3HL+2H)=3.85
-----

```

```

Mo(VI)    sp  oth/un 25°C 1.00M U                                1980VPa (42039) 219
                                                K(MoO4+L+2H=MoO3L)=13.35
-----

```

```

Mo(VI)    sp  oth/un 25°C 0.20M U                            1979PVa (42040) 220
                                                K(MoO4+L+2H)=13.60
-----

```

```

*****
C6H2O4Cl2      H2L      Chloranilic acid  CAS 87-88-7 (1281)
3,6-Dichloro-2,5-dihydroxy-1,4-benzoquinone;
-----

```

```

Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Mo(VI)    sp  oth/un 30°C  ?  U                                K1=6.05          1981BMd (42055) 221
-----

```

```

Mo(VI)    sp  oth/un 25°C 1.00M U                            1980VPa (42056) 222
                                                K(MoO4+L+2H=MoO3L)=13.28
-----

```

```

Mo(VI)    sp  oth/un 25°C 0.20M U                            1979PVa (42057) 223
                                                K(MoO4+L+2H)=13.90
-----

```

```

Mo(VI)    sp  NaClO4 25°C 0.37M U                            1964LSd (42058) 224
                                                K(Mo3O11+3HL)=2.30
-----

```

```

*****
C6H4O4      H2L                                CAS 615-94-1 (1280)
2,5-Dihydroxy-1,4-benzoquinone;
-----

```

```

Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Mo(VI)    sp  oth/un 25°C 1.00M U                                1980VPa (42308) 225
                                                K(MoO4+L+2H=MoO3L)=15.30
-----

```

```

*****
C6H5Li      L                                CAS 591-51-5 (2352)
Phenyl lithium;
-----

```

```

Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Mo(VI)    sp  non-aq 25°C 100% U I M                            1981PKa (42341) 226
                                                K(MoA+L)=3.4
Medium: THF. A=Mo(CO)4(Ph2P(OCH2CH2)3.OPPh2. In benzene: K > 5
-----

```

```

*****
C6H5NO4      H2L      4-Nitrocatechol  CAS 3316-09-4 (890)
1,2-Dihydroxy-4-nitrobenzene; O2N.C6H3(OH)2
-----

```

```

Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----

```



Mo(VI) sp oth/un 25°C .575M U 1980NKa (42936) 227  
 $K(H_2MoO_4L+H_2L=MoO_2L_2+2H_2O)=3.3$   
 $K(MoO_4+H_2L)=2.21$   
Medium: 0.1 M NH<sub>4</sub>OH, 0.08 M Na<sub>2</sub>S<sub>2</sub>O<sub>5</sub>. pH 8  
Medium: 0.1 M NH<sub>4</sub>OH, 0.08 M Na<sub>2</sub>S<sub>2</sub>O<sub>5</sub>, pH 8  
\*\*\*\*\*  
C6H5O2Cl H2L 4-Cl-Catechol CAS 2138-22-9 (1656)  
1,2-Dihydroxy-4-chlorobenzene; Cl.C6H3(OH)<sub>2</sub>

---

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

---

Mo(VI) sp oth/un 25°C .575M U 1980NKa (43084) 228  
 $K(H_2MoO_4L+H_2L=MoO_2L_2+2H_2O)=3.4$   
 $K(MoO_4+H_2L)=2.07$   
Medium: 0.1 M NH<sub>4</sub>OH, 0.08 M Na<sub>2</sub>S<sub>2</sub>O<sub>5</sub>. pH 8

---

Mo(VI) sp KCl 25°C 0.10M U 1962HAb (43085) 229  
 $K(MoO_4+2H_2L)=5.85$   
\*\*\*\*\*  
C6H6N2O2 HL CAS 5657-61-4 (1430)  
Nicotinyhydroxamic acid; C<sub>5</sub>H<sub>4</sub>N.CO.NH.OH

---

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

---

Mo(VI) sp NaClO<sub>4</sub> 25°C 0.10M U 1964RMa (43437) 230  
 $K(?)=6.3$   
 $K(?)=6.7$   
\*\*\*\*\*  
C6H6O2 H2L Catechol CAS 120-80-9 (534)  
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

---

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

---

Mo(VI) sp oth/un 25°C .575M U 1980NKa (43790) 231  
 $K(H_2MoO_4L+H_2L=MoO_2L_2+2H_2O)=3.3$   
 $K(MoO_4+H_2L)=1.49$   
Medium: 0.1 M NH<sub>4</sub>OH, 0.08 M Na<sub>2</sub>S<sub>2</sub>O<sub>5</sub>. pH 8

---

Mo(VI) sp oth/un 20°C 0.10M U 1971SBd (43791) 232  
 $K(MoO_4+2H_2L=MoO_2L_2+2H_2O)=5.09$   
By electrophoresis, phosphate buffer, K=5.21

---

Mo(VI) sp KNO<sub>3</sub> 20°C 0.10M U 1969HBa (43792) 233  
 $K(MoO_4+2H_2L=MoO_2L_2)=5.00$

---

Mo(VI) sp oth/un 20°C 0.10M U 1964PCa (43793) 234  
 $K(MoO_4+2H_2L=MoO_2L_2)=5.27$   
Medium: 0.1 M NaHSO<sub>3</sub>

---

Mo(VI) sp oth/un 26°C 0.10M U 1960HAa (43794) 235

K(MoO4+2H2L=MoO2L2)=4.61

Medium: 0.1 M NaHSO3

-----  
Mo(VI) sp oth/un 20°C ? U 1959HAa (43795) 236  
K(MoO4+2H2L=MoO2L2)=5.27

\*\*\*\*\*

C6H6O3 H3L Pyrogallol CAS 87-66-1 (696)  
1,2,3-Trihydroxybenzene; C6H3(OH)3

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(VI) sp oth/un 25°C .575M U 1980NKa (43967) 237  
K(H2MoO4L+H2L=MoO2L2+2H2O)=3.2  
K(MoO4+H2L)=1.97

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

-----  
Mo(VI) sp oth/un 20°C 0.10M U 1971SBd (43968) 238  
K(MO4+2H3L=MO2(HL)2+2H2O)=5.43

By electrophoresis, phosphate buffer, K=5.57

-----  
Mo(VI) sp oth/un 20°C ? U 1959HAa (43969) 239  
K(MoO4+2H3L=MoO2(HL)2)=5.48

-----  
Mo(VI) sp oth/un 20°C ? U 1958PIa (43970) 240  
K(MoO4+2H3L=MoO2(HL)2)=5.68 ?

\*\*\*\*\*

C6H6O5S H3L CAS 7134-09-0 (3687)  
3,4-Dihydroxybenzenesulfonic acid; (HO)2.C6H3.SO3H

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(VI) sp oth/un 20°C 0.10M U 1971SBd (44284) 241  
K(MoO4+2H2L=MoO2L2+2H2O)=5.28

\*\*\*\*\*

C6H6O8S2 H4L Tiron CAS 149-45-1 (104)  
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)2.C6H2(SO3H)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(VI) gl KCl 25°C 0.20M C M 2002FCa (44474) 242  
K(MoO4+2H+L=MoO3L+H2O)=23.8

K(MoO4+4H+2L=MoO2L2+2H2O)=46.96, K(2MoO4+6H+2L=Mo2O6H2L2+2H2O)=61.6,  
K(MoO4+4H+A+L=MoO2AL+2H2O)=41.5. A is acetohydroxamic acid.

-----  
Mo(VI) sp oth/un 20°C 0.10M U 1971SBd (44475) 243  
K(MoO4+2H2L=MoO2L2+2H2O)=6.59

\*\*\*\*\*

C6H7O3As H2L Phenylarsonic CAS 98-05-5 (3690)  
Benzeneearsonic acid, phenylarsonic acid; C6H5AsO3H2

-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) gl NaCl 25°C 1MM C 1987YTa (45177) 244  
 Values given for B(pH+qMoO2+C) where C=C6H5AsO3H- and H2AsO4- .  
 \*\*\*\*\*  
 C6H8O6 H2L Ascorbic acid CAS 50-81-7 (285)  
 Ascorbic acid (Vitamin C);  
 -----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) sp oth/un ? ? U 1966SAb (45648) 245  
 K(?)=4.6  
 \*\*\*\*\*  
 C6H8O7 H3L Citric acid CAS 77-92-9 (95)  
 2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH  
 -----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) gl NaCl 25°C 1.00M U T H K1=5.12 B2=9.17 1995CRa (46180) 246  
 B3=11.94  
 DH(B1)=-1.3; DH(B2)=-5.8; DH(B3)=-10.1 kJ mol-1. TDS(B1)=28; TDS(B2)=47;  
 TDS(B3)=58 kJ mol-1. Data also at 275 K  
 -----

Mo(VI) gl NaCl 25°C 1.00M U T H 1995CRa (46181) 247  
 B(111)=8.35  
 B(121)=15.00  
 B(131)=19.62  
 B(141)=21.12  
 B(242)= 31.02; B(252)= 35.86; B(262)= 40.08; B(142)= 25.34; B(152)= 29.54;  
 B(162)= 33.34; etc. B(pqr): pMoO4 + qH + rL = (MoO4)pHqLr  
 -----

Mo(VI) gl NaCl 25°C 1.00M C H 1991CKa (46182) 248  
 B(1,1,1)=8.25  
 B(1,2,1)=15.08  
 B(1,3,1)=19.66  
 B(p,q,r): p(WO4--)+qH+rL=(WO4-- )pHqLr. By calorimetry, DH(1,1,1)=-49,  
 DH(1,2,1)=-60, DH(1,3,1)=-67 kJ mol-1.  
 -----

Mo(VI) gl KNO3 20°C 0.20M U 1986BHd (46183) 249  
 K(MoO2(OH)4+L+2H=MoO2H-1L)=16  
 K(MoO2(OH)4+L+3H=MoO2H-1L+4H2O)=20.4  
 K(MoO2(OH)H-1L+H=MoO2H-1L+H2O)=4.8  
 -----

Mo(VI) gl NaCl 25°C 1.00M C H 1986CVa (46184) 250  
 B(1,1,1)=8.25  
 B(1,1,2)=15.08  
 B(1,1,3)=19.66  
 B(2,1,4)=27.27  
 B(2,1,5)=31.86. B(p,q,r): pMoO4+qL+rH=(MoO4)pLqHr. DH(1,1,1)=-49.4 kJ mol-1;  
 DH(1,1,2)=-60.2; DH(1,1,3)=-67.4; DH(2,1,4)=-124  
 -----

\*\*\*\*\*

C6H9NO6 H3L NTA CAS 139-13-9 (191)  
Nitrilotriethanoic acid; N(CH2.COOH)3

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl NaClO4 25°C 0.10M C I 2003MZa (46922) 251  
K(MoO4+2H+L=MoO3L+H2O)=18.72

Also data for I=0.5, 0.7 and 1.0 M NaClO4. For I=1.0 M, K=17.97.

-----  
Mo(VI) gl NaCl 25°C 1.00M C H 1994CHb (46923) 252  
B(1,1,2)=17.78  
B(1,1,3)=21.02  
B(1,1,4)=22.57  
B(2,2,7)=45.16

B(p,q,r): pMoO4+qL+rH=(MoO4)pLqHr. Also B(2,2,8)=47.95, B(2,1,5)=30.74,  
B(2,1,6)=33.09. DH(1,1,2)=-69, DH(1,1,3)=-71.2, DH(2,2,7)=-123, DH(2,2,8)=-132.

-----  
Mo(VI) sp NaClO4 25°C 0.5M C 1976CLa (46924) 253  
K(MoO4+2H+L=MoO3L+H2O)=17.90

Method: stopped flow spectrophotometry

-----  
Mo(VI) nmr oth/un 28°C 1.30M U 1967MEa (46925) 254  
K(MoO4+WO3L=MoO3L+WO4)=0.15

-----  
Mo(VI) gl oth/un 25°C 0.15M U 1966KR a (46926) 255  
K(MoO4+L+2H=MoO3L)=18.94

-----  
Mo(VI) nmr oth/un 35°C 2.00M U 1966KR a (46927) 256  
K(MoO4+L+2H=MoO3L)=18.90

\*\*\*\*\*

C6H10N4O2 HL CAS 25486-00-4 (2554)  
2-Amino-3-(4'-imidazolyl)propanehydroxamic acid, Histidine-hydroxamic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl KCl 25°C 0.20M C 1999FCa (47907) 257  
K(MoO4+6H+2L=MoO2H2L2)=48.5  
K(MoO4+3H+L=MoO3HL)=24.77  
K(MoO4+2H+L=MoO3L)=18.44  
K(MoO4+8H+2L=MoO2H4L2)=55.1

K(MoO4+7H+2L=MoO2H3L2)=52.88

\*\*\*\*\*

C6H10O8 H2L Mucic acid CAS 526-99-8 (3650)  
2,3,4,5-Tetrahydroxyhexanedioic acid, Galactaric acid; HOOC.(CHOH)4.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) kin oth/un 20°C ? U 1971FPb (48438) 258  
K(?)=7.57

\*\*\*\*\*  
 C6H10O8 H2L Saccharic acid CAS 87-73-0 (1191)  
 D-2,3,4,5-Tetrahydroxy-1,6-hexanedioic acid, Glucaric acid; HOOC.(CHOH)4.COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) kin oth/un 25°C 0.10M U 1971FPb (48484) 259  
 K(?)=7.64

\*\*\*\*\*  
 C6H12N2O4 H2L EDDA CAS 5657-17-0 (119)  
 1,2-Diaminoethane-N,N'-diethanoic acid; HOOC.CH2.NH.CH2.CH2.NH.CH2.COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) gl NaClO4 25°C 3.0M U 1979ZLa (49254) 260  
 B(MoO4+L+2H=MoO3L)=19.69

\*\*\*\*\*  
 C6H12O5 L L-Rhamnose CAS 634-74-2 (3659)  
 6-Deoxy-L-mannose;

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) gl KCl 25°C 0.10M C 1989VCa (49508) 261  
 B((MoO4)2H2L)=13.89  
 B((MoO4)2H3L)=17.59  
 K((MoO4)2H2L+H)=3.70

\*\*\*\*\*  
 C6H12O6 L D-Mannose CAS 3458-28-4 (1562)  
 D-Mannose

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) gl KCl 25°C 0.10M C 1989VCa (49607) 262  
 B((MoO4)2H2L)=14.50  
 B((MoO4)2H3L)=18.10  
 K((MoO4)2H2L+H)=3.60

M=MoO4

\*\*\*\*\*  
 C6H12O7 HL Gluconic acid CAS 526-95-4 (904)  
 D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH2(CHOH)4.COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) kin oth/un ? ? U 1972FPb (49737) 263  
 K(2MoO3+L=(HMoO3)2(H-2L))=7.12

\*\*\*\*\*  
 C6H13NO2 HL CAS 4312-93-0 (4386)  
 Hexanohydroxamic acid; CH3.CH2.CH2.CH2.CH2.CO.NH.OH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(VI) sp oth/un ? ? U 1971PMd (50228) 264  
K(MoO2+L)=15.02  
K(MoO2+2L)=18.04

\*\*\*\*\*  
C6H14O6 L D-Dulcitol CAS 608-66-2 (3663)  
D-Galactitol;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl KCl 25°C 0.10M U 1990CVb (51062) 265  
B(2,2,1)=17.30  
B(2,3,1)=20.90

B(p,q,r): pMoO4+qH+rL=MoO4pHqLr  
\*\*\*\*\*  
C6H14O6 L D-Mannitol CAS 69-65-8 (3664)  
D-Mannitol;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl KCl 25°C 0.10M U 1990CVb (51085) 266  
B(2,2,1)=16.70  
B(2,3,1)=20.80

B(p,q,r): pMoO4+qH+rL=MoO4pHqLr  
-----

Mo(VI) gl KNO3 21°C 0.10M C 1978MBc (51086) 267  
Medium pH 3-5. K(2MoO4+2H+L=Mo2O5(H-4L)+3H2O)=16.89  
K(Mo2O5(H-4L)+H=HM2O5(H-4L))=3.82

Mo(VI) kin oth/un ? ? U 1972FPa (51087) 268  
K(2H2MoO4+L=(H2MoO4)2L)=7.12

\*\*\*\*\*  
C6H14O6 L Glucitol CAS 50-70-4 (2878)  
D-Sorbitol;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl KCl 25°C 0.10M U 1990CVb (51105) 269  
B(2,2,1)=16.60  
B(2,3,1)=20.50

B(p,q,r): pMoO4+qH+rL=MoO4pHqLr  
-----

Mo(VI) gl KNO3 21°C 0.10M C 1978MBc (51106) 270  
Medium pH 3-5. K(2MoO4+2H+L=Mo2O5(H-4L)+3H2O)=16.90  
K(Mo2O5(H-4L)+H=HM2O5(H-4L))=3.88

Mo(VI) kin oth/un ? ? U 1972FPa (51107) 271  
K(2H2MoO4+L=(H2MoO4)2L)=6.64

\*\*\*\*\*  
C6H15N3O2 HL CAS 52760-35-7 (6670)

Lysine hydroxamic acid; H2N.(CH2)4.CH(NH2)CO.NHOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(VI) gl KCl 25°C 0.20M C 2002ECa (51429) 272  
K(MoO4+L+4H=MoO3H2L+H2O)=34.16  
K(MoO4+L+3H=MoO3HL+H2O)=29.15  
K(MoO4+2L+8H=MoO2H4L2+2H2O)=66.65.

\*\*\*\*\*  
C7H6O3 H2L CAS 139-85-5 (881)  
3,4-Dihydroxybenzaldehyde, protocatechuic aldehyde; C6H3(OH)2.CHO

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(VI) sp oth/un 25°C .575M U 1980NKa (54356) 273  
K(H2MoO4L+H2L=MoO2L2+2H2O)=3.4  
K(MoO4+H2L)=2.15

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

-----  
Mo(VI) sp oth/un 20°C ? U 1959HAa (54357) 274  
K(MoO4+H2L=MoO2L2)=7.75

\*\*\*\*\*  
C7H6O4 H3L CAS 409-79-9 (1115)  
2,5-Dihydroxybenzoic acid; C6H3(OH)2.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(VI) sp oth/un 25°C 0.10M U K1=2.58 1976DVa (54588) 275

\*\*\*\*\*  
C7H6O4 H3L Protocatechuic CAS 99-50-3 (875)  
3,4-Dihydroxybenzoic acid; C6H3(OH)2.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(VI) sp oth/un 25°C .575M U 1980NKa (54683) 276  
K(H2MoO4L+H2L=MoO2L2+2H2O)=3.1  
K(MoO4+H2L)=1.98

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

-----  
Mo(VI) sp oth/un 20°C ? U 1959HAa (54684) 277  
K(MoO4+2H3L=MoO2H2L2)=6.68

\*\*\*\*\*  
C7H6O5 H4L CAS 610-02-6 (3725)  
2,3,4-Trihydroxybenzoic acid; (HO)3.C6H2.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(VI) sp oth/un 20°C 0.10M U 1971SBd (54721) 278  
K(MoO4+2H3L=MoO2(HL)2+2H2O)=5.24

\*\*\*\*\*

C7H6O5 H4L Gallic acid CAS 149-91-7 (446)  
3,4,5-Trihydroxybenzoic acid; C6H2(OH)3.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp oth/un 25°C .575M U 1980NKa (54754) 279  
K(H2MoO4L+H2L=MoO2L2+2H2O)=3.2  
K(MoO4+H2L)=2.32

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

-----  
Mo(VI) sp oth/un 20°C 0.10M U 1971SBd (54755) 280  
K(MoO4+2H3L=MoO2(HL)2+2H2O)=5.38

-----  
Mo(VI) sp oth/un 20°C ? U 1959HAa (54756) 281  
K(MoO4+2H3L=MoO2H2L2)=6.83

\*\*\*\*\*  
C7H7NO2 HL CAS 495-18-1 (184)  
Benzohydroxamic acid; C6H5.CO.NH.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl none 25°C 0.0 U T H K1=7.49 B2=14.40 1980ABa (55509) 282  
At 35 C: K1=7.38, K2=6.79; DH(K1)=-4.6, DH(K2)=-5.0

\*\*\*\*\*  
C7H8O2 H2L CAS 488-17-5 (1657)  
1,2-Dihydroxy-3-methylbenzene; CH3.C6H3(OH)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp oth/un 25°C .575M U 1980NKa (56056) 283  
K(H2MoO4L+H2L=MoO2L2+2H2O)=3.3  
K(MoO4+H2L)=1.52

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

\*\*\*\*\*  
C7H8O2 H2L Methylcatechol CAS 452-86-8 (525)  
1,2-Dihydroxy-4-methylbenzene; CH3.C6H3(OH)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp oth/un 25°C .575M U 1980NKa (56071) 284  
K(H2MoO4L+H2L=MoO2L2+2H2O)=3.1  
K(MoO4+H2L)=1.28

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

-----  
Mo(VI) sp oth/un 20°C .014M U 1962HAb (56072) 285  
K(MoO4+2H2L)=4.74

\*\*\*\*\*  
C8H7O3Cl H2L CAS 99-40-1 (3818)  
4-Chloro-2,3-dihydroxyacetophenone, 3-acetyl-6-chlorocatechol;



Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	KCl	25°C	0.10M	U				1963HAb (59247)	286
									K(MoO4+2H2L)=7.03	
*****										
C8H8O3		HL		Mandelic Acid				CAS 611-72-3	(80)	
2-Phenyl-2-hydroxyethanoic acid; C6H5.CH(OH).COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	vlt	NaClO4	25°C	0.50M	U				1984ZZa (59853)	287
									K(MoO2+HL)=2.42	
*****										
C8H8O3		H2L						CAS 2848-25-1	(3799)	
3,4-Dihydroxyacetophenone, (4-acetylcatechol)										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	oth/un	20°C	?	U				1961HAa (59893)	288
									K(MoO4+2H2L=MoO2L2)=6.74	
*****										
C8H11NO2		H2L		Dopamine				CAS 579-59-9	(251)	
2-(3',4'-Dihydroxyphenyl)ethylamine; (HO)2.C6H3.CH2.CH2.NH2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	KCl	25°C	0.10M	U				1963HAc (61082)	289
									K(MoO4+2H2L=MoO2L2)=5.57	
*****										
C8H11NO3		H2L		Noradrenaline				CAS 138-65-8	(253)	
Norepinephrine, 3,4-Dihydroxyphenylethanolamine; (HO)2C6H3.CH(CH2.NH2).OH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	KCl	25°C	0.10M	U				1962HAb (61166)	290
									K(MoO4+2H2L)=5.82(?)	
*****										
C9H5NOBr2		HL						CAS 521-74-4	(3279)	
5,7-Dibromo-8-hydroxyquinoline;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	dis	oth/un	20°C	var	U			K1=14.22 B2=28.32	1967DBa (63522)	291
Metal: MoO2++ , Medium: var (HCl,HClO4)										
*****										
C9H6O4		H2L		Esculetin				CAS 305-01-1	(3853)	
6,7-Dihydroxycoumarin;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Mo(VI) sp alc/w ? 50% U 1963JSa (63953) 292  
K(MoO4+2H2L)=3.65(?)

Medium: 50% EtOH

\*\*\*\*\*

C9H7NO HL Oxine CAS 148-24-3 (504)  
8-Hydroxyquinoline (8-quinolinol);

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp NaNO3 25°C 0.20M U 1968KDa (64319) 293  
K(H+HL+MoO4)=10.36

By kinetics, K=10.29

-----  
Mo(VI) dis oth/un ? ? U K1=17.04 B2=33.02 1967BDa (64320) 294  
K(MoO(OH)+L)=16.44  
K(MoO(OH)+2L)=30.02

Metal: MoO2++

-----  
Mo(VI) dis oth/un ? ? U K1=12.7 1967BDa (64321) 295

Metal: MoO++++

\*\*\*\*\*

C9H7NO3S2 H2L CAS 58447-10-2 (4675)  
8-Mercaptoquinoline-5-sulfonic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp oth/un ? ? U B2=22.8 1968ABa (64427) 296

Metal: MoO2++

\*\*\*\*\*

C9H7NO4S H2L Sulfoxine CAS 84-88-8 (448)  
8-Hydroxyquinoline-5-sulfonic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl KNO3 16°C 0.10M U 1969GTa (64565) 297  
K(MoO4+L+2H=MoO3L)=19.53

\*\*\*\*\*

C9H11NO2 HL (4650)  
5-Methyl-2-hydroxyacetophenone oxime; (CH3)(HO).C6H3.C(:N.OH).CH3

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp alc/w 30°C ? U 1970GMF (66027) 298  
K(MoO4+2H2L=MoO2L2)=8.69

Medium: EtOH

\*\*\*\*\*

C9H12N2O4 H3L (6664)  
3,4-Dihydroxyphenylalanine hydroxamic acid, DOPA hydroxamic acid;  
H2N.CH(CH2.C6H3(OH)2CO.NHOH

-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C	M			2002FCa (66621)	299
								K(MoO4+8H+2L)=65.29		
								K(MoO4+6H+2L)=56.81		
								K(2MoO4+8H+2L)=70.65		
								K(MoO4+3H+L)=28.81		
								K(MoO4+2H+L)=20.91.		

\*\*\*\*\*

C9H13NO3		H2L		Corbadrine				CAS 50731-42-5	(3880)	
1-(3',4'-Dihydroxyphenyl)-2-aminopropanol;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	KCl	25°C	0.10M	U				1962HAb (66818)	300
								K(MoO4+2H2L=MoO2L2)=5.92		
								*****		
C9H13NO3		H2L		(-)Adrenaline				CAS 51-43-4	(252)	
4-(1-Hydroxy-2-(methylamino)ethyl)-1,2-dihydroxybenzene, Epinephrine;CH3NHCH(OH)C6H3(OH)2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	KCl	25°C	0.10M	U				1962HAb (66865)	301
								K(MoO4+2H3L=MoO2(HL)2)=5.76(?)		
								*****		
C10H8O2		H2L						CAS 92-44-4	(1658)	
2,3-Dihydroxynaphthalene;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	oth/un	25°C	.575M	U				1980NKa (69774)	302
								K(H2MoO4L+H2L=MoO2L2+2H2O)=3.5		
								K(MoO4+H2L)=2.43		
Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	oth/un	20°C	0.10M	U				1973PAC (69775)	303
								K(MoO4+2H2L=MoO2L2+2H2O)=6.31		
								Medium: phosphate buffer. By electrophoresis: K(MoO4+2H2L=MoO2L2+2H2O)=6.15		
								*****		
C10H8O4		H2L		4-Me-Esculetin				CAS 529-84-0	(3890)	
4-Methyl-6,7-dihydroxycoumarin										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	alc/w	?	50%	U				1963JSa (69790)	304
								K(MoO4+2H2L=MoO2L2)=7.55		
								Medium: 50% EtOH		
								*****		
C10H8O5S		H3L		DHNSA				(877)		



$K(2\text{MoO}_3+\text{L})=19.5$ ,  $K((\text{MoO}_3)_2\text{L}+\text{H}_2\text{L}=2\text{MoO}_3\text{HL})=0.26$

\*\*\*\*\*

C10H18N4O6 H2L (4504)  
Hexanoic acid bis(3-hydroxycarbamoyl-methyl)amide; HONHCOCH2NHC(O)(CH2)4CONHCH2CONHOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(VI) gl KCl 25°C 0.20M C 1998FMa (75569) 312

$K(\text{MoO}_4+\text{L}+4\text{H}=\text{MoO}_2\text{L}+2\text{H}_2\text{O})=30.45$

$K(\text{MoO}_4+\text{L}+3\text{H}=\text{MoO}_3\text{HL}+\text{H}_2\text{O})=25.52$

For the propylamide analogue K values are 30.88; 26.76

\*\*\*\*\*

C10H20N4O4 H2L CAS 475984-27-1 (6717)  
Piperazine-1,4-bis(N-methylacetohydroxamic acid); C4H8N2(CH2.CO.N(OH)CH3)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(VI) gl KCl 25°C 0.20M C 2002FCb (75896) 313

$K(\text{MoO}_4+3\text{H}+\text{L}=\text{MoO}_3\text{HL}+\text{H}_2\text{O})=27.4$

$K(2\text{MoO}_4+9\text{H}+2\text{L}=\text{Mo}_2\text{O}_4\text{HL}_2+4\text{H}_2\text{O})=74.7$ ,  $K(2\text{MoO}_4+10\text{H}+2\text{L}=\text{Mo}_2\text{O}_4\text{H}_2\text{L}_2+4\text{H}_2\text{O})=77.5$ ,

$K(2\text{MoO}_4+11\text{H}+2\text{L}=\text{Mo}_2\text{O}_4\text{H}_3\text{L}_2+4\text{H}_2\text{O})=80.5$ . An alternative model given also.

\*\*\*\*\*

C10H25N5 L 15-Ane-N5 CAS 295-64-7 (99)  
1,4,7,10,13-Pentaazacyclopentadecane; cyclo(-(HN.CH2.CH2)5-)

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(VI) vlt NaClO4 25°C 0.20M C 1999SSe (76737) 314

$K(\text{MoO}_4+\text{H}_3\text{L})=2.11$

Method: differentail pulse polarography.

Also data for selenate, selenite and pyrophosphate as guest ions.

\*\*\*\*\*

C11H17N03 H2L Isoprenaline CAS 586-06-1 (3950)  
3,4-Dihydroxy-1-(1'-hydroxy-2'-(propylamino)ethyl)benzene;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(VI) sp KCl 25°C 0.10M U 1963HAc (79158) 315

$K(\text{MoO}_4+2\text{H}_2\text{L}=\text{MoO}_2\text{L}_2)=5.87$

\*\*\*\*\*

C11H21N3O5 H2L CAS 499238-77-6 (8837)  
N-Hydroxy-N'-[4-(hydroxymethylamino)-4-oxobutyl]-N-methylpentanediamide;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Mo(VI) gl KCl 25°C 0.20M C 2002FBb (79795) 316

$K(\text{MoO}_4+\text{L}+4\text{H}=\text{MoO}_2\text{L}+2\text{H}_2\text{O})=31.27$

$K(\text{MoO}_4+\text{L}+3\text{H}=\text{MoO}_3\text{HL}+\text{H}_2\text{O})=26.62$

\*\*\*\*\*

C12H11N3O4S H2L (4003)





$$K(\text{MoO}_4+2\text{H}_2\text{L}=\text{MoO}_2\text{L}_2)=8.20(?)$$

Medium: 50% EtOH

\*\*\*\*\*

C15H1004                      H2L    (4052)

7,8-Dihydroxy-3-phenylcoumarin    (3-phenyldaphnetin)

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

Mo(VI)            sp   alc/w   32°C   50%   U    1966JKb (90988) 330

$$K(\text{MoO}_4+\text{H}_2\text{L}=\text{MoO}_2\text{L}_2)=4.5(?)$$

Medium: 50% EtOH, 0.2 M KCl

\*\*\*\*\*

C15H1005                      H3L            Galangin                      CAS 548-83-4    (4053)

3,5,7-Trihydroxyflavone    (3,5,7-Trihydroxy-2-phenylchromone)

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

Mo(VI)            sp   oth/un   28°C   ?   U    1964KSc (90993) 331

$$K(\text{MoO}_4+\text{H}_3\text{L}=\text{MoO}_3\text{HL})=4.58(?)$$

\*\*\*\*\*

C15H11N3O4S                      H2L    (5130)

7-Phenylazo-8-hydroxyquinoline-5-sulfonic acid;

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

Mo(VI)            gl   KNO3    16°C   0.10M   U    1969GTa (91337) 332

$$B((\text{MoO}_4)\text{H}_2\text{L})=18.12$$

\*\*\*\*\*

C15H11N3O7S2                      H3L    CAS 17852-90-3    (5131)

7-(4-Sulfophenylazo)-8-hydroxyquinoline-5-sulfonic acid;

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

Mo(VI)            gl   KNO3    16°C   0.10M   U    1969GTa (91350) 333

$$B((\text{MoO}_4)\text{H}_2\text{L})=17.98$$

\*\*\*\*\*

C16H14O5                      H3L    CAS 966-64-3    (5143)

2,3,7-Trihydroxy-9-propylfluorone;

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

Mo(VI)            sp   KNO3    25°C   0.50M   U    1972ASb (93590) 334

$$K(\text{MoO}_2\text{OH}+\text{H}_2\text{L})=17.27$$

$$K(\text{MoO}_2+2\text{H}_2\text{L})=26.50$$

\*\*\*\*\*

C16H35O4P                      HL    CAS 298-07-7    (1625)

Di-(2-ethylhexyl)-phosphoric acid; (C<sub>2</sub>H<sub>5</sub>C<sub>6</sub>H<sub>12</sub>O)<sub>2</sub>P(O)OH

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----



Mo(VI) dis non-aq RT 100% C I 1992SGa (95510) 335

K'=4.28

Method: solvent extraction into CCl4. K':

-----  
Mo(VI) dis non-aq RT 100% C I 1992SGa (95511) 336

K'=4.28

By solvent extraction into CCl4. K': H2MoO4+H2L2=MoO2L2(HL)2(org)+H2O

Also data for C6H6 (K'=4.02), C2H4Cl2 (3.90), CHCl3 (3.55), MIBK (3.48).

\*\*\*\*\*

C18H30N4O12 H6L TTHA CAS 869-52-3 (694)

Triethylenetetraaminehexaethanoic acid;((HOOC.CH2)2N.CH2.CH2.N(CH2.COOH).CH2)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(VI) gl KNO3 25°C 0.10M U 1971LUa (98070) 337

K(2MoO4+H6L=(MoO3)2H2L)=13.81

K(2MoO4+H5L=(MoO3)2HL)=11.78

K(2MoO4+H4L=(MoO3)2L)=8.42

K(MoO4+H6L=MoO3H4L)=7.45

K(MoO4+H5L=MoO3H3L)=6.88, K(MoO4+H4L=MoO3H2L)=5.64, K(MoO4+H3L=MoO3HL)=3.16,  
K(MoO4+H2L=MoO3L)=3.14

\*\*\*\*\*

C19H13N3O7S2 H3L SNAZOXS CAS 117-87-3 (995)

8-Hydroxy-7-(4'-sulfo-1'-naphthylazo)-quinoline-5-sulfonic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(VI) sp NaClO4 RT 0.10M C K1=9.20 1987APb (99047) 338

-----  
Mo(VI) gl KNO3 16°C 0.10M U 1969GTa (99048) 339

K(MoO4+L+2H)=17.82

\*\*\*\*\*

C20H11N06S2 H2L CAS 66451-75-0 (8985)

6-Hydroxy-5-oxo-5H-dibenzo[a,j]phenoxazine-11-sulfonic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(VI) sp NaClO4 25°C 0.10M C 1977SLb (99529) 340

B((MoO2)2L)=10.8

B((MoO2)2L) is the effective constant at pH 2.3.

\*\*\*\*\*

C20H11N06S2 H2L CAS 55968-31-5 (8984)

6-Hydroxy-5-oxo-5H-dibenzo[a,j]phenoxazine-9-sulfonic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Mo(VI) sp NaClO4 25°C 0.10M C 1977SLb (99531) 341

K1eff=3.62

Medium pH 2.2.

\*\*\*\*\*

C20H11N09S2 H3L CAS 65501-73-7 (8982)  
6-Hydroxy-5-dibenzo[a,j]phenoxazone-8,11-disulfonic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp NaClO4 25°C 0.10M C 1977SLb (99535) 342  
B((MoO2)2L)=10.0

B((MoO2)2L) is the effective constant at pH 2.05.

\*\*\*\*\*

C20H11N09S2 H3L CAS 66451-74-9 (8983)  
6-Hydroxy-5-oxo-5H-dibenzo[a,j]phenoxazine-9,11-disulfonic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp NaClO4 25°C 0.10M C 1977SLb (99539) 343  
B((MoO2)2L)=9.9

B((MoO2)2L) is the effective constant at pH 2.05.

\*\*\*\*\*

C20H13N3O7S H3L EriochromeBla A CAS 16279-54-2 (5299)  
3-Hydroxy-4-(2-hydroxy-1-alpha-naphthylazo)-7-nitronaphthalene-1-sulfonic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp NaNO3 25°C 2.0M U 1971AAc (99584) 344  
K(MoO2+HL)=9.80

\*\*\*\*\*

C20H22N2O8 H5L Azotochelins CAS 23369-85-9 (6112)  
N,N'-Bis(2,3-dihydroxybenzoyl)lysine;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl oth/un 25°C 0.10M C 1998DHa (99918) 345  
K1=ca. 35  
K1(eff)=4

Medium: 0.10 M HEPES, pH 6.6.

\*\*\*\*\*

C22H20O13 H5L Carminic acid CAS 1260-17-9 (714)  
Carminic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) sp oth/un 22°C ? U 1966KWb (101705) 346  
K(MoO4+H5L=MoO3H3L)=3.8(?)

\*\*\*\*\*

C22H24N2O8 L Deoxycycline CAS 564-25-0 (2204)  
Deoxycycline, 6-Deoxy-5-hydroxytetracycline;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Mo(VI) gl none 20°C 0.0 C 1991JMa (101766) 347  
K(MoO4+H3L=MoO3HL)=7.99

$$K(\text{MoO}_4 + 2\text{H}_2\text{L} = \text{MoO}_3(\text{H}_2\text{L})_2) = 9.21$$

\*\*\*\*\*

C22H24N2O8 H2L Tetracycline CAS 60-54-8 (2201)

Tetracycline;

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----  
 Mo(VI) gl none 20°C dil C 1989VJa (101823) 348  
 K(MoO3+HL)=7.80

#### REFERENCES

- 2003GDa F Gharib, L Dogaheh; J. Eng. Chem. Data, 48, 999 (2003)  
 2003MZa K Majlesi, K Zare, F Teimouri; J. Chem. Eng. Data, 48, 680 (2003)  
 2002ECa E Enyedy, H Csoka, I Lazar, E Farkas; J. Chem. Soc., Dalton Trans., 2632 (2002)  
 2002FBb E Farkas, P Buglyo, E Enyedy, V Gerlei; Inorg. Chim. Acta, 339, 215 (2002)  
 2002FCa E Farkas, H Csoka; J. Inorg. Biochem., 89, 219 (2002)  
 2002FCb E Farkas, H Csoka, S Gama, M Santos; Talanta, 57, 935 (2002)  
 2002TAA F Taube, I Andersson, I Toth, A Bodor; J. Chem. Soc., Dalton Trans., 4451  
 (2002)  
 2002THa F Taube, M Hashimoto, I Andersson; J. Chem. Soc., Dalton Trans., 1002 (2002)  
 2001GZa G Gharib, K Zare, A Taghvamanesh; J. Chem. Eng. Data, 46, 1140 (2001)  
 2000CHA J Cruywagen, J Heyns; Polyhedron, 19, 907 (2000)  
 2000GZa F Gharib, K Zare, K Majlesi; J. Chem. Eng. Data, 45, 833 (2000)  
 2000SAa A Selling, I Andersson, L Pettersson; Eur. J. Inorg. Chem., 1509 (2000)  
 1999FCa E Farkas, H Csoka, G Bell, D Brown; J. Chem. Soc., Dalton Trans., 2789 (1999)  
 1999SSE A Salimi, M Shamsipur; J. Inclusion Phenom., 34, 455 (1999)  
 1998ARA A Abou-Hamdan, A Roodt, A Merbach; Inorg. Chem., 37, 1278 (1998)  
 1998DHa A Duhme, R Hider, M Naldrett; J. Biol. Inorg. Chem., 3, 520 (1998)  
 1998FMA E Farkas, K Megyeri, L Somsak, L Kovacs; J. Inorg. Biochem., 70, 41 (1998)  
 1998KYa A Kobayashi, A Yagasaki; Polyhedron, 17, 3315 (1998)  
 1998MSa A Mederos, D Sells, J Sanchiz, A Sykes; J. Chem. Soc., Dalton Trans., 2723  
 (1998)  
 1997CRb J Cruywagen, E Rohwer, R van de Water; Polyhedron, 16, 243 (1997)  
 1997KSc A Kobayashi, M Sugihashi, A Yagasaki; Polyhedron, 16, 2761 (1997)  
 1997KYa A Kobayashi, A Yagasaki; Inorg. Chem., 36, 126 (1997)  
 1997NCa M Sokolov, N Coichev, H Moya, A Sykes et al; J. Chem. Soc., Dalton Trans., 1863  
 (1997)  
 1995CRA J Cruywagen, E Rohwer, G Wessels; Polyhedron, 14, 3481 (1995)  
 1995GZb F Gharib, K Zare, S Khorrani, A Behju; J. Chem. Eng. Data, 40, 1214 (1995)  
 1994CHb J Cruywagen, J Heyns, E Rohwer; J. Chem. Soc., Dalton Trans., 45 (1994)  
 1994RLa A Roodt, J Leipoldt, L Helm et al; Inorg. Chem., 33, 140 (1994)  
 1993BBc S Battacharjee, R Battacharyya; J. Chem. Soc., Dalton Trans., 1151 (1993)  
 1993CHa J Cruywagen, J Heyns, E Rohwer; J. Chem. Soc., Dalton Trans., 1713 (1993)  
 1993CKb J Cruywagen, L Kruger, E Rohwer; J. Chem. Soc., Dalton Trans., 105 (1993)  
 1993HLa M Hong, Y Li, J Lu, M Nasreldin et al; J. Chem. Soc., Dalton Trans., 2613  
 (1993)  
 1993LMB G Lamprecht, M Martinez, M Nasreldin; J. Chem. Soc., Dalton Trans., 747 (1993)  
 1993VSA J Varey, A Sykes; J. Chem. Soc., Dalton Trans., 3293 (1993)  
 1992BCC W Buhro, M Chisholm et al; J. Am. Chem. Soc., 114, 557 (1992)

1992HMa K Hall, J Mayer; *J. Am. Chem. Soc.*, 114, 10402 (1992)  
1992LIa A Lindmark; *Inorg. Chem.*, 31, 3507 (1992)  
1992PMa J Pugh, T Meyer; *J. Am. Chem. Soc.*, 114, 3784 (1992)  
1992RSb C Routledge, A Sykes; *J. Chem. Soc., Dalton Trans.*, 325 (1992)  
1992SGa S-X Sun, Z-L Gao, Y Liu, J-L Shen; *Acta Chimica Sinica*, 50, 877 (1992)  
1991CKa J Cruywagen, L Kruger, E Rohwer; *J. Chem. Soc., Dalton Trans.*, 1727 (1991)  
1991HKb K Hegetschweiler, T Keller et al; *Inorg. Chem.*, 30, 4342 (1991)  
1991HPa O Howarth, L Pettersson et al; *J. Chem. Soc., Dalton Trans.*, 1799 (1991)  
1991HYa Z Hualin, W Yongwei; *Transition Met. Chem.*, 16, 592 (1991)  
1991JMa M Jelkic-Stankov, D Malesev; *Polyhedron*, 10, 455 (1991)  
1991KKc L Kuo, M Kanatzidis et al; *J. Am. Chem. Soc.*, 113, 9027 (1991)  
1991LGa S Lahiri, M Ghosh; *Indian J. Chem.*, 30A, 989 (1991)  
1991ZGa K Zhang, A Gonzalez et al; *J. Am. Chem. Soc.*, 113, 9170 (1991)  
1990CHc J Cruywagen, J Heyns et al; *J. Chem. Soc., Dalton Trans.*, 1951 (1990)  
1990CJa A Cavaleiro, J de Jesus, V Gil et al; *Inorg. Chim. Acta*, 172, 25 (1990)  
1990CSb L Csanyi; *Transition Met. Chem.*, 15, 371 (1990)  
1990CVb S Chapelle, J Verchere, J Sauvage; *Polyhedron*, 9, 1225 (1990)  
1989CHb L Csanyi, I Horvath, Z Galbacs; *Transition Met. Chem.*, 14, 90 (1989)  
1989CPa A Cavaleiro, J Pedrosa-de-Jesus; *Inorg. Chim. Acta*, 166, 205 (1989)  
1989HPa O Howarth, L Pettersson et al; *J. Chem. Soc., Dalton Trans.*, 1915 (1989)  
1989VCa J Verchere, S Chapelle; *Polyhedron*, 8, 333 (1989)  
1989VJa D Veselinovic, M Jelkic-Stankov; *Mikrochim. Acta*, 329 (1989)  
1988BHa J Brule, Y Hayden, K Callahan, J Edwards; *Gazz. Chim. Ital.*, 118, 93 (1988)  
1988CDc J Cruywagen, A Draaijer, T Rypstra; *S. Afr. J. Chem.*, 41, 89 (1988)  
1988LIa S Licht; *J. Electrochem. Soc.*, 135, 2971 (1988)  
1987APb M Alvarez Jimenez, M Palacios Corvillo; *An. Quim.*, 83, 203 (1987)  
1987LSa J Lydon, L Schwane, R Thompson; *Inorg. Chem.*, 26, 2606 (1987)  
1987YAa A Yagasaki, I Andersson, L Pettersson; *Inorg. Chem.*, 26, 3926 (1987)  
1987YTa A Yagasaki, H Takahama, Y Sasaki; *Bull. Chem. Soc. Jpn.*, 60, 3925 (1987)  
1986BHd M Bartusek, J Havel, D Matula; *Coll. Czech. Chem. Comm.*, 51, 2702 (1986)  
1986CHa J Cruywagen, J Heyns et al; *J. Chem. Soc., Dalton Trans.*, 1857 (1986)  
1986CVa J Cruywagen, R van de Water; *Polyhedron*, 5, 521 (1986)  
1986CZa Chen Lianshan, Zhao G L, He, Z L, Zhao H G; *Acta Chimica Sinica*, 520 (1986)  
1986HNa E Hills, P Norman, T Ramasami et al; *J. Chem. Soc., Dalton Trans.*, 157 (1986)  
1986VPa J Verchere, J Poirier; *Polyhedron*, 5, 527 (1986)  
1984HSa S Himeno, A Saito, M Hasegawa; *Inorg. Chim. Acta*, 88, 93 (1984)  
1984JJa E Johansen, O Jons; *Talanta*, 31, 743 (1984)  
1984KRa H Kelly, D Richens, A Sykes; *J. Chem. Soc., Dalton Trans.*, 1229 (1984)  
1984NGa B Nabivanets, D Gorina; *Zh. Neorg. Khim.*, 29, 1738 (1984)  
1984ZZa P Zaitsev, S Zdanov, E Dergacheva; *Zh. Neorg. Khim.*, 29, 108(60) (1984)  
1983BCc A Beltran-Porter, A Cervilla; *Transition Met. Chem.*, 8, 324 (1983)  
1983BCd A Beltran-Porter, A Cervilla, F Caturla; *Transition Met. Chem.*, 8, 222 (1983)  
1983HHa S Himeno, M Hasegawa; *Inorg. Chim. Acta*, 73, 255 (1983)  
1983HUa S Himeno, Y Ueda, M Hasegawa; *Inorg. Chim. Acta*, 70, 53 (1983)  
1983KMa T Kato, T Murayama; *Bull. Chem. Soc. Jpn.*, 56, 2129 (1983)  
1983ZZa P Zaitsev, S Zhdanov; *Zh. Neorg. Khim.*, 28, 1471(830) (1983)  
1982CPa A Cavaleiro, J Pedrosa de Jesus, V Gil; *Transition Met. Chem.*, 7, 75 (1982)  
1982HHb P Havlova, J Havel, M Bartusek; *Coll. Czech. Chem. Comm.* 47, 1570 (1982)  
1982KCb C Kircher, S Crouch; *Anal. Chem. (USA)*, 54, 1219 (1982)  
1981BCb A Beltran, F Caturla, A Cervilla et al; *J. Inorg. Nucl. Chem.*, 43, 3277 (1981)

1981BCd A Beltran, A Cervilla, J Beltran; J.Inorg.Nucl.Chem.,43,1337 (1981)  
 1981BMD G Bianchi, C Marone; J.Inorg.Nucl.Chem.,43,2985 (1981)  
 1981JJa E Johansen, O Jons; Acta Chem.Scand.,A35,233 (1981)  
 1981PKa J Powell, A Kuksis, C May et al; J.Am.Chem.Soc.,103,5941 (1981)  
 1980ABa S Abbasi; Pol.J.Chem.,54,1287 (1980)  
 1980KKc A KuoIrin, I Khodakovsky et al; Geokhim.,12,1825 (1980)  
 1980LPe L Lyhamn, L Pettersson; Chemica Scripta,16,52 (1980)  
 1980NKA S Natansohn, J Krugler et al; J.Phys.Chem.,84,2972 (1980)  
 1980SSF J Srivastava, M Srivastava; Talanta,27,763 (1980)  
 1980VPa J Verchere, J Poirier; J.Inorg.Nucl.Chem.,42,1514 (1980)  
 1979PVa J Poirier, J Verchere; Talanta,26,341 (1979)  
 1979ZLa K Zare, P Lagrange et al; J.Chem.Soc.,Dalton Trans.,1372 (1979)  
 1978MBC M Mikesova, M Bartusek; Coll.Czech.Chem.Comm.,43,1867 (1978)  
 1978MMb L Mitkina, N Melchakova et al; Zh.Neorg.Khim.,23,1258(693) (1978)  
 1978MMi L Mit'kina, N Melchakova, V Peshkova; Koord.Khim.,4,1684 (1978)  
 1977ILb R Izatt, J Lamb et al; J.Am.Chem.Soc.,99,8344 (1977)  
 1977LAB M Lamache-Duhameaux; J.Inorg.Nucl.Chem.,39,2081 (1977)  
 1977PRa A Palant, V Resnichenco, A Stepanov; Zh.Neorg.Khim.,22,538(297) (1977)  
 1977RGa D Rabenstein, M Greenburg et al; Inorg.Chem.,16,1241 (1977)  
 1977SLb J Simek, J Lasovsky, E.Ruzicka, N Song; Coll.Czech.Chem.Comm.,42,2365  
 (1977)  
 1976CLa J Collin, P Lagrange; Bull.Soc.Chim.,France,1304 (1976)  
 1976CRA J Cruywagen, E Rohwer; S.Afr.J.Chem.,29,30 (1976)  
 1976CSa G Cayley, A Sykes; Inorg.Chem.,15,2882 (1976)  
 1976DVA A Djukanovic, K Velasevic et al; Bull.Soc.Chim.Beograd,41,41 (1976)  
 1976LAG M Lamache-Duhameaux; J.Inorg.Nucl.Chem.,38,1979 (1976)  
 1976OSa J Ojo, Y Sasaki, R Taylor et al; Inorg.Chem.,15,1006 (1976)  
 1976SKb T Shpak, I Kolosov, M Senyavin; Zh.Neorg.Khim.,21,3309(1823) (1976)  
 1975EDa J Ewen, D Darensbourg; J.Am.Chem.Soc.,97,6874 (1975)  
 1975PEb L Pettersson; Chemica Scripta,7,145 (1975)  
 1975STa Y Sasaki, R Taylor, A Sykes; J.Chem.Soc.,Dalton Trans.396 (1975)  
 1975WHb D Williams, I Holcomb, D Boltz; Anal.Chem.(USA),47,2025 (1975)  
 1975ZSa T Zhilina, E Strizhev et al; Vestnik Leningr.Univ.,141 (1975)  
 1974FIb F Ferranti, A Indelli; J.Solution Chem.,3,619 (1974)  
 1974OHa P O'Hare; J.Chem.Thermodyn.,6,425 (1974)  
 1974PRa A Palant, V Reznichenko et al; Zh.Neorg.Khim.,19,2415,(E:1319) (1974)  
 1974RWA W Rudolf, A Wolniak; Z.Anorg.Allg.Chem.,408,214 (1974)  
 1974SSd Y Sasaki, A Sykes; J.Less Common Metals,36,125 (1974)  
 1973BKa M Beg, Kabir-ud-Din et al; Australian J.Chem.,26,671 (1973)  
 1973MHa G McKnight, G Haight; Inorg.Chem.,12,1934 (1973)  
 1973PAC N Poluektov, L Alakaeva, M Tischenko; Zh.Neorg.Khim.,18,1,81 (1973)  
 1973PMA N Parpiev, I Maslennikov, K Abdullaeva; Uzbeksk.Khim.Zh.,6,3 (1973)  
 1973SDa D Shishkov, H Doichinova; Dokl.Bolg.Akad.Nauk,26,927 (1973)  
 1973TSe R Tewari, M Srivastava; Talanta,20,133;360 (1973)  
 1972ASb V Antaonovich, E Shelikhina et al; Zh.Anal.Khim.,27,1,100 (1972)  
 1972FPa A Fedorov, A Pavlinova; Zh.Anal.Khim.,27,2409 (1972)  
 1972FPb A Fedorov, A Pavlinova; Zh.Anal.Khim.,27,708 (1972)  
 1972JRa B Jezowska-Trzebiatowska, M Rudolf; Trans.Roy.Inst.Tech.(Stockholm),258  
 (1972)  
 1972KTA K Kustin, D Toppen; Inorg.Chem.,11,2851 (1972)

1972SSe M Singh, M Srivastava; *J. Inorg. Nucl. Chem.*, 34, 567; 2067; 2081 (1972)  
1971AAC I Alimarin, N Arslanova, F Sudakov; *Zh. Anal. Khim.*, 26, 12, 2383 (1971)  
1971BPg Y Buslaev, S Petrosyants; *Zh. Neorg. Khim.*, 16, 5, 1330 (1971)  
1971ESb A El-Aggan, S Sidarous et al; *Egypt. J. Chem.*, 14, 315 (1971)  
1971FPb A Fedorov, A Pavlinova; *Zh. Vsesouz. Khim. Obsch.*, 5, 587 (1971)  
1971GSa R Gut, E Schmid, J Serrallach; *Helv. Chim. Acta*, 54, 593; 609 (1971)  
1971JSb D Joshi, K Sharma; *Z. Phys. Chem.*, 246, 281 (1971)  
1971LUa W Lund; *Anal. Chim. Acta*, 53, 295 (1971)  
1971NSd V Nazarenko, E Shelikhina; *Zh. Neorg. Khim.*, 16, 166(E:88) (1971)  
1971PMd V Popova, F Milkis, A Markman, K Agzamov; *Zh. Anal. Khim.*, 26, 3, 537 (1971)  
1971SBd R Soni, M Bartusek; *J. Inorg. Nucl. Chem.*, 33, 2557 (1971)  
1971YTa Y Yoshino, I Taminaga, S Uchida; *Bull. Chem. Soc. Jpn.*, 44, 1435 (1971)  
1970BGB M Bartusek, B Grebenova, L Sommer; *Publ. Fac. Sci. Univ. Brno*, E38, 381; 397  
(1970)  
1970BRb A Busev, T Rodionova; *Anal. Lett.*, 3, 325 (1970)  
1970CMB J Cruywagen, H McKay; *J. Inorg. Nucl. Chem.*, 32, 255 (1970)  
1970GMF B Gupta, W Malik; *J. Indian Chem. Soc.*, 47, 771 (1970)  
1970HPa D Hrushkova, J Podlahova, J Podlaha; *Collec. Czech. Chem. Commun.*, 35, 2738  
(1970)  
1970RBb S Rani, S Banerji; *J. Indian Chem. Soc.*, 47, 704 (1970)  
1970SHA D Shishkov; *God. Vissh. Khimikotekhnol. I. Sof.*, 15, 415 (1970)  
1969AYa I Alekseeva, K Yatsimirskii; *Zh. Neorg. Khim.*, 14, 432(E:221) (1969)  
1969GTa S Goyal, J Tandon; *Talanta*, 16, 106 (1969)  
1969HBA L Havelkova, M Bartusek; *Collec. Czech. Chem. Commun.*, 34, 2722; 2919 (1969)  
1969KBC Kabir-ud-Din, M Beg; *J. Indian Chem. Soc.*, 46, 503 (1969)  
1969MDB I Marov, Y Dubrov, A Ermakov, G Martynova; *Zh. Neorg. Khim.*, 14, 438(E:224)  
(1969)  
1969PFa A Pavlinova, A Fedorov, V Yakovlev; *Zh. Anal. Khim.*, 24, 4, 561 (1969)  
1969SAb F Sudakov, N Arslanova; *Vestnik Moskov Univ.*, 24, 3, 91 (1969)  
1969SHd D Shishkov; *Dokl. Bolg. Akad. Nauk*, 22, 763 (1969)  
1968ABa Y Atoks, Y Bankovskii; *Izv. Akad. Nauk Latv. SSR, Khim.*, 1, 122 (1968)  
1968ABb O Afanasyev, A Bantysh, D Knyazev; *Zh. Neorg. Khim.*, 13, 2, 352 (1968)  
1968DBb Kabir-ud-Din, M Beg; *J. Indian Chem. Soc.*, 45, 455 (1968)  
1968JDa W Jakob, M Dyrek; *Rocz. Chem.*, 42, 1393 (1968)  
1968KDa P Knowles, H Diebler; *Trans. Faraday Soc.*, 64, 977 (1968)  
1968MDF I Marov, Y Dubrov, A Ermakov, G Martynova; *Zh. Neorg. Khim.*, 13, 3247 (1968)  
1968NPa M Naarova, J Podlahova, J Podlaha; *Collec. Czech. Chem. Commun.*, 33, 1991  
(1968)  
1968PNb J van de Poel, H Neumann; *Inorg. Chem.*, 7, 2086 (1968)  
1968SJB R Saxena, M Jain; *Indian J. Chem.*, 6, 752 (1968)  
1968TKd I Tserkovnitskaya, N Kustova; *Zh. Anal. Khim.*, 23, 1, 72 (1968)  
1967BDa A Bantysh, E Dobizha, D Knyazev; *Zh. Neorg. Khim.*, 12, 8, 2165 (1967)  
1967DBa E Dobizha, A Bantysh, D Knyazev; *Zh. Neorg. Khim.*, 12, 10, 2740 (1967)  
1967JRa B Jezowska-Trzebiatowska, M Rudolf; *Rocz. Chem.*, 41, 453; 1879 (1967)  
1967KKb Y Karyakin, E Kryachko; *Zh. Neorg. Khim.*, 12, 2567 (1967)  
1967MEa A Merbach; *Helv. Chim. Acta*, 50, 1431 (1967)  
1967VDA S Vorobev, I Davydov, I Shilin; *Zh. Neorg. Khim.*, 12, 2142 (1967)  
1967VDb S Vorobev, I Davydov, I Shilin; *Zh. Neorg. Khim.*, 6, 2665 (1967)  
1966JKb B Jain, R Kumar; *Curr. Sci.*, 35, 557 (1966)  
1966KRa R Kula, D Rabenstein; *Anal. Chem.*, 38, 1934 (1966)

1966KUa R Kula; Anal.Chem.,38,1382 (1966)  
1966KUb R Kula; Anal.Chem.,38,1581 (1966)  
1966KWb G Kirkbright,T West,C Woodward; Talanta,13,1637 (1966)  
1966MDb I Marov,Y Dubrov,V Belyaeva et al; Zh.Neorg.Khim.,11,2443 (1966)  
1966RCa E Rohwer,J Cruywagen; S.Afr.J.Chem.,19,11 (1966)  
1966SAb K Stolyarov,I Amantova; Vestnik Leningr.Univ.,4,141;155;10,133 (1966)  
1965GKa B Gupta,M Katyal,R Singh; J.Indian Chem.Soc.,42,811 (1965)  
1965MOb C Monk; J.Chem.Soc.,2456 (1965)  
1965ULa N Ulko; Ukr.Khim.Zh.,31,887 (1965)  
1964ANb J Allen,H Neumann; Inorg.Chem.,3,1612 (1964)  
1964KSc M Katyal,R Singh; Indian J.Chem.,2,454 (1964)  
1964LSd W Lee,N Shastri,E Amis; Talanta,11,685 (1964)  
1964PCa Personal Communication etc; Chem.Soc.Spec.Publ.,no.17 (1964)  
1964RMa R Rowland,C Melon; Anal.Chem.,36,1997 (1964)  
1963HAb J Halmekoski; Suomen Kem.,B36,19 (1963)  
1963HAc J Halmekoski; Suomen Kem.,B36,29;40;46;55 (1963)  
1963JSa B Jain,H Singh; Indian J.Chem.,1,369 (1963)  
1963MAB W Malik,S Ali; Indian J.Chem.,1,374 (1963)  
1963SCe J Spence,H Chang; Inorg.Chem.,25,319 (1963)  
1963SDF R Seth,A Dey; Z.Anorg.Chem.,321,278 (1963)  
1963STc J Stary; Anal.Chim.Acta,28,132 (1963)  
1962HAb J Halmekoski; Suomen Kem.,B35,41;108;171;209;238;241 (1962)  
1962ZRa E Zahnow,R Robinson; J.Electroanal.Chem.,3,263 (1962)  
1961HAa J Halmekoski; Suomen Kem.,B34,169 (1961)  
1961YBa K Yatsimirskii,L Budarin; Zh.Neorg.Khim.,6,944 (1961)  
1961YBb K Yatsimirskii,L Budarin; Collec.Czech.Chem.Comm.,26,215 (1961)  
1960HAa G Haight; Anal.Chem.,32,642 (1960)  
1960NVa N Nikolaev,S Vlasov,Y Buslaev et al; Izv.Sib.Otd.Akad.Nauk SSR,47 (1960)  
1960SHa J Sheldon; J.Chem.Soc.,3106 (1960)  
1959BGi A Babko,T Getman; Zh.Neorg.Khim.,4,585 (1959)  
1959CSa F Chauveau,P Souchay,R Schaal; Bull.Soc.Chim.Fr.,1190 (1959)  
1959DBb A Dey,S Banerji; Proc.Symp.Chem.of Coord.Comp.,Agra,198 (1959)  
1959HAa J Halmekoski; Ann.Acad.Sci.Fennicae,96 (1959)  
1959LMA V Litvinchuk,K Mikhalevich; Ukr.Khim.Zh.,25,563 (1959)  
1959NAb B Nabivanets; Zh.Neorg.Khim.,4,1797 (1959)  
1958CHb G Charlot; Oxid-Reduction Potentials(IUPAC),London (1958)  
1958CSb L Csanyi; Acta Chim.Acad.Sci.Hung.,14,69;79;264 (1958)  
1958DSa U Durgapal,N Sogani; J.Indian Chem.Soc.,35,542,842 (1958)  
1958PEb D Perrin; J.Am.Chem.Soc.,80,3540 (1958)  
1958PIa E Pisko; Chem.Zvesti,12,95 (1958)  
1958SAa I Sajo; Acta Chim.Acad.Sci.Hung.,16,115 (1958)  
1958SAC P Sakellaridis; Chimika Chronika,23,263 (1958)  
1958SEa A Sergeeva; Nauk Zapiski L'vov Inst.,50,22 (1958)  
1956GHa R Graham,L Heppler; J.Am.Chem.Soc.,78,4846 (1956)  
1956YAc K Yatsimirskii,I Alekseeva; Zh.Neorg.Khim.,1,952 (1956)  
1955CSa F Chauveau,P Souchay,G Tridot; Bull.Soc.Chim.Fr.,1519 (1955)  
1953EEa H El-Shamy,A El-Aggan; J.Am.Chem.Soc.,75,1187 (1953)  
1952HSc G Haight,W Sager; J.Am.Chem.Soc.,74,6056 (1952)  
1952LAb W Latimer; "Oxidation Potentials",Prentice Hall,NY (1952)  
1941HGa R Holtje,R Geyer; Z.Anorg.Chem.,246,258 (1941)

1936Kta I Kolthoff,W Tomsicek; J.Phys.Chem.,40,247 (1936)  
1924COa O Collenberg; Z.Phys.Chem.,109,353 (1924)

EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES  
I Data with various BACKGROUNDS  
H Data for THERMOCHEMICAL quantities  
M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC

---

END