

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 128 experiments for

(no ligands specified)

2 metals : Sb(V), Sb+++

(no references specified)

(no experimental details specified)

e- HL Electron (442)
Electron;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb(V) oth none 25°C 0.0 U 1952LAb (922) 1
K=22.7(692 mV?)
K: 0.5Sb2O5(s)+2H+2e=0.5Sb2O3(s)+H2O. From thermodynamic data

Sb(V) EMF oth/un 25°C 6.0M U I 1949BSa (923) 2
K=27.66(818 mV)
Medium: HCl. K: Sb+2e=Sb(III). K=26.51(4.5 M;784 mV), 25.22(3.5 M;746 mV)

Sb(V) EMF oth/un 20°C 10.0M U I 1923GSa (924) 3
K=-20.3(-589 mV)
Medium: KOH. K: Sb+2e=Sb(III). K=-19.3(7.5 M;-561 mV), -17.7(5 M;-516 mV),
-14.7(3 M;-428 mV)

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb(V) oth oth/un 25°C 0.0 C K1=0.34 1975AAc (5672) 4
Method: use of Zr-PO4 as competitive cation-exchanger for 124Sb.
Medium: 0.01-4.0 M HCl.

Sb(V) dis oth/un 25°C var U 1972CMd (5673) 5
K((C6H5)3Sb+L)=2
K((C6H5)3SbL+L)=0.5

Sb(V) EMF non-aq 25°C 100% U 1971DTb (5674) 6
K6=5.45
Medium: SeOCl2, 0.5 M Et4NClO4

Sb(V) dis oth/un 0.0 U 1965DIa (5675) 7
K(SbL3(OH)3+H+L)=-3.07
K(SbL4(OH)2+H+L)=-3.46
K(SbL5(OH)+H+L)=-4.28

Sb(V) dis oth/un ? 0.0 U 1963IDa (5676) 8

K((SbOL)3+SbOL)=-3.22
K((SbOL)4+SbOL)=-3.48
K((SbOL)5+SbOL)=-4.06

Sb(V) ISE non-aq ? 100% U 1959BGf (5677) 9
K(SbL3OPL3=SbL6+POL2)=-5.4

Medium: POCl3

Sb(V) sp oth/un 25°C 9.0M U 1956NRa (5678) 10
Medium: LiCl. K(SbL5OH+H+Cl=SbL6+H2O)=-4.34 (or -4.65 ?)

Sb(V) sp oth/un 26°C var U 1954NEa (5679) 11
Medium:HCl. K(SbL4(OH)2+H+L=SbL5OH+H2O)=-3.43, K(SbL5OH+H+Cl=SbL6+H2O)=-4.65

ClO4- HL Perchlorate CAS 7001-90-3 (287)
Perchlorate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb(V) cal oth/un 25°C dil U H 1972CJa (6375) 12
Kso(Ph4SbL(s)=Ph4Sb+L)=-7.46

DH(Kso)=48.1 kJ mol⁻¹, DS(Kso)=21 J K⁻¹ mol⁻¹

F- HL Fluoride CAS 7644-39-3 (201)
Fluoride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb(V) dis oth/un 25°C var U 1972CMd (7140) 13
K((C6H5)3Sb+F)=4
K((C6H5)3SbF+F)=3

Sb(V) con non-aq 25°C 100% U 1965TBa (7141) 14
K(HSbF5X+HX=SbF5X+H2X)=-2.43
K(2HSbF5X=Sb2F10X+H2X)=-2.15
K(Sb2F4X7+2HX=2SbF2X4+H2X)=-2

Medium: liquid HS03F. HX=HS03F

Sb(V) sol non-aq 0°C 100% U 1961CKa (7142) 15
K(KSbF6(s)=K+SbF6)=-1.23
K(TlSbF6(s)=Tl+SbF6)=-3.56

Medium: liquid HF, I=0 corr.

Sb(V) oth non-aq ? 100% U 1961HQa (7143) 16
K(2HF+SbF5=H2F+SbF6)=-1.14
K(H2F+SbF6)=2.23

Method; ir. Medium: liquid HF

H2O L Water CAS 7732-18-5 (6115)
Water

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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Sb(V)      cal non-aq 25°C 100% U   H                      19670La (7611) 17
Medium: C2H4Cl2. DH(SbCl5+H2O)=-101.6 kJ mol-1 in C2H4Cl2(l)
*****
I-         HL      Iodide                CAS 10034-85-2 (20)
Iodide;
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Sb(V)      dis oth/un 25°C var   U                      1972CMd (8371) 18
                                K((C6H5)3Sb+L)=1.5
                                K((C6H5)3SbL+L)=-0.4
*****
OH-        HL      Hydroxide             (57)
Hydroxide;
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Sb(V)      dis oth/un 25°C      U                      1972CMe (12088) 19
                                K(R3SbF2+OH=R3SbOHF+F)=7.5
                                K(R3SbCl2+OH=R3SbOHCl+Cl)=9.5
                                K(R3SbI2+OH=R3SbOHI+I)=10
R=C6H5
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Sb(V)      dis oth/un 25°C      U                      1972CMe (12089) 20
                                K(R3SbOHF+OH=R3Sb(OH)2+F)=5
                                K(R3SbOHCl+OH=R3Sb(OH)2+Cl)=7.5
                                K(R3SbOHI+OH=R3Sb(OH)2+I)=8.5
R=C6H5
*****
C4H6O6      H2L    L-Tartaric acid CAS 87-69-4 (92)
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Sb(V)      oth oth/un ?      ? U   M                      1972MFb (31345) 21
K(2Sb(OH)3(H-1L)=Sb2(OH)4(H-2L)2)=1.26
*****
C13H9OClS      L                      CAS 6028-95-2 (5005)
1-(2-Thienyl)-3-(4'-chlorophenyl)propen-3-one;
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Sb(V)      sp non-aq ?      100% U   M                      1966TLA (84666) 22
                                K(SbCl5+L)=3.93
Medium: benzene
*****
C13H10OS      L                      CAS 3988-77-0 (4979)
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1-(2'-Thienyl)-3-phenylprop-1-en-3-one; C4H3S.CH:CH.CO.C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Sb(V)	sp	non-aq	?	100%	U	M			1966TLa (84969)	23

K(SbCl5+L)=3.18

Medium: 50% dioxan, 0.1 M NaClO4

C14H12OS L CAS 6028-90-4 (5051)
3-(4'-Methylphenyl)-1-(2'-thienyl)prop-1-en-3-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Sb(V)	sp	non-aq	?	100%	U	M			1966TLa (87322)	24

K(SbCl5+L)=3.32

Medium: benzene

C14H12O2S L CAS 6028-93-9 (5054)
3-(4'-Methoxyphenyl)-1-(2'-thienyl)prop-1-en-3-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Sb(V)	sp	non-aq	?	100%	U	M			1966TLa (87335)	25

K(SbCl5+L)=4.02

Medium: benzene

C15H14OS L (5112)
3-(4'-Ethylphenyl)-1-(2'-thienyl)prop-1-en-3-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Sb(V)	sp	non-aq	?	100%	U	M			1966TLa (91767)	26

K(SbCl5+L)=3.91

Medium: benzene.

C19H14OS L CAS 40766-17-4 (5273)
3-Biphenyl-1-(2-thienyl)prop-1-en-3-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Sb(V)	sp	non-aq	?	100%	U	M			1966TLa (99072)	27

K(SbCl5+L)=3.06

Medium: benzene.

e- HL Electron (442)
Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Sb+++	EMF	none	25°C	0.00	U	T			1973Vsa (925)	28

K=30.51(150.4mV)
K=Sb4O6(s) + 12H+ + 12e=4Sb(s) + 6H2O. K=32.43(154.5mV,15 C), 28.81(146.8mV, 35 C), 26.03(139.1mV,50 C)

Sb+++ EMF none 25°C 0.00 U 1973V Sa (926) 29
K=-32.40(-638.9mV)

K=SbO2- + 2H2O + 3e=Sb(s) + 4OH-

Sb+++ EMF none 25°C 0.00 U T 1972V Sa (927) 30
K=10.35(204.0mV)

K=SbO+ + 2H+ + 3e=Sb(s) + H2O. K=11.05(210.5mV,15 C), 9.83(200.3mV,35 C), 9.30(198.8mV,50 C)

Sb+++ EMF none 25°C 0.0 U 1924S Ca (928) 31
K=7.71(152 mV)

K: 0.5Sb2O3(s)+3H+3e=Sb(s)+1.5H2O. K(SbO+2H+3e=Sb(s)+H2O)=10.76(212 mV)

Sb+++ EMF oth/un 20°C 10.0M U 1923G Sa (929) 32
K=-34.8(-675 mV)

Medium: KOH. K: Sb(OH)4+3e=Sb(s)+4OH ?

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt oth/un 22°C 4.0M C I K1=0.96 B2= 1.52 1975BZa (5680) 33
B3=1.45
B4=1.04

Method: polarography. Medium: NaCl/HCl/HClO4 (total acid = 0.5 M).
For total acid = 4.0 M: B2=3.39, B3=4.09.

Sb+++ vlt NaClO4 20°C 4.70M U K1=1.05 B2=1.90 1975KBb (5681) 34
B3=2.20
B4=1.95
B5=1.10

Sb+++ vlt NaClO4 30°C 2.0M U K1=2.30 B2=4.1 1970BWB (5682) 35
B3=5.8 to 6.0
B4=6.8 to 7.2

Sb+++ EMF oth/un 99°C 100% U 1969BBa (5683) 36
K(2SbL3=SbL2+SbL4)=-7.8

Medium: SbCl3. Method: current-voltage studies

Sb+++ sp non-aq 99°C 100% U 1969BBc (5684) 37
K(SbL3+H2O=SbL0+2HL)=-7.5

Medium: pyridine. Method: also emf and nmr

Sb+++ dis oth/un 290°C 100% U TI 1969JSb (5685) 38

Medium:K(FeL4). K1=1.6(K(AlL4,289 C); 0.10(K(TlCl4,315 C).Gas chromatography

Sb+++ sol oth/un 25°C 4.0M U 1965HEa (5686) 39
Ks((Me4N)3(SbL4)2L)=-4.74
K4=1.0
K5.K6=-0.77

Medium:H2SO4

Sb+++ dis oth/un 15°C 0.50M U 1964IDa (5687) 40
K4=1.4
K5K6=-1.0
Kd(H+SbL4=HSbL4(in org))=0.6

Medium:0.5 H,6.3 Li(NO3). org=C6H13OH or C8H17OH

Sb+++ vlt NaNO3 25°C 4.0M U K1=2.26 B2=3.49 1959PDa (5688) 41
B3=4.18
B4=4.72
B5=4.72
B6=4.11

F- HL Fluoride CAS 7644-39-3 (201)
Fluoride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt NaClO4 30°C 2.0M U K1=3.00 B2=5.70 1970B0c (7144) 42
B3=8.30
B4=10.95

Sb+++ sol KNO3 20°C 0.10M U 1959KGc (7145) 43
*Ks=-0.37
K(Sb(OH)2+F)=5.5

Medium: HNO3, *Ks: 0.5Sb2O3(s)+0.5H2O+HF=Sb(OH)2F

I- HL Iodide CAS 10034-85-2 (20)
Iodide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ con non-aq 140°C 100% U 1967BNb (8372) 44
K(SbI2+I)=6.27

Medium: liquid I2

OH- HL Hydroxide (57)
Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sol none 80°C 0.0 C T 2003ZSa (12090) 45

Kso(valentinite)=-3.69

Kso(senarmonite)=-3.96

Kso: $0.5\text{Sb}_2\text{O}_3(\text{s})+1.5\text{H}_2\text{O}=\text{Sb}(\text{OH})_3$. Data for 15-450 C and 1 to 1000 bar.

Sb+++	sp	NaClO4	25°C	1.00M	U	K1=14.60	B2=28.62	1977ANa (12091)	46
						B3=41.57			
Sb+++	sol	NaClO4	25°C	5.00M	U			1974ABb (12092)	47
								*Ks=-10.86(orthorombic Sb406)	
								*Ks=-11.71(cubic Sb406)	
								*Ks: $\text{Sb}_4\text{O}_6(\text{s})+8\text{H}=2\text{Sb}_2(\text{OH})_2+2\text{H}_2\text{O}$	
Sb+++	sol	NaClO4	25°C	5.00M	U			1974ABb (12093)	48
								*Ks=-12.2(orthorombic Sb406)	
								*Ks=-12.7(cubic Sb406)	
								*Ks: $\text{Sb}_4\text{O}_6(\text{s})+4\text{H}+2\text{H}_2\text{O}=4\text{Sb}(\text{OH})_2$	
Sb+++	sol	NaClO4	25°C	5.00M	U			1974ABb (12094)	49
						*Ks=-13.07			
								*Ks: $\text{Sb}_4\text{O}_5(\text{OH})\text{ClO}_4(\text{H}_2\text{O})_{1/2}(\text{s})+3\text{H}+3/2\text{H}_2\text{O}=4\text{Sb}(\text{OH})_2+\text{ClO}_4$	
Sb+++	sol	NaNO3	25°C	5.00M	U			1974ABb (12095)	50
						*Ks=-13.46			
								*Ks: $\text{Sb}_4\text{O}_4(\text{OH})_2(\text{NO}_3)_2(\text{s})+2\text{H}+2\text{H}_2\text{O}=4\text{Sb}(\text{OH})_2+2\text{NO}_3$	
Sb+++	sol	NaNO3	25°C	5.00M	U			1974ABb (12096)	51
						*Ks=-16.89			
								*Ks: $\text{Sb}_4\text{O}_4(\text{OH})_2(\text{NO}_3)_2(\text{s})+6\text{H}=4\text{SbOH}+2\text{NO}_3+2\text{H}_2\text{O}$	
Sb+++	dis	NaClO4	25°C	3.00M	U			1974SMc (12097)	52
						*K3=-1.24			
Sb+++	sol	none	25°C	0.00	U	TIH		1973VSb (12098)	53
							Ks=-2.35		
								Ks($1/4\text{Sb}_4\text{O}_6(\text{s})+3/2\text{H}_2\text{O}+\text{OH}=\text{Sb}(\text{OH})_4$)=-2.44(15 C), -2.21(35 C), -2.08(50 C);	
								(DH(Ks)=18.8 kJ mol ⁻¹ . In 2 M NaClO4: Ks=-2.22	
Sb+++	sp	NaClO4	23°C	?	U			1968MGa (12099)	54
								*K(SbO+2H2O=Sb(OH)3+H)=-1.42	
Sb+++	vlt	none	12°C	0.0	U			1958K0c (12100)	55
								Kso(Sb(OH)3)=-41.5	
Sb+++	sol	none	25°C	0.0	U			1952GGb (12101)	56
								*Ks2=-3.11	
								Ks3=-4.70	
								Ks4=-2.06	
								Ks2: $0.5\text{Sb}_2\text{O}_3(\text{s})+1.5\text{H}_2\text{O}=\text{Sb}(\text{OH})_2+\text{OH}$; Ks3: $0.5\text{Sb}_2\text{O}_3(\text{s})+1.5\text{H}_2\text{O}=\text{Sb}(\text{OH})_3$;	
								Ks4: $0.5\text{Sb}_2\text{O}_3(\text{s})+\text{OH}+1.5\text{H}_2\text{O}=\text{Sb}(\text{OH})_4$	

Sb+++ vlt oth/un ? var U 1925BAa (12102) 57
Kso=-41.4

Sb+++ sol NaCl04 25°C var U 1924SCa (12103) 58
*K(PbOH+H2O=Pb(OH)2+H)=-3.1

S-- H2L Sulfide CAS 7783-06-4 (705)
Sulfide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ oth none 25°C dil C T 1989SRf (14461) 59
K(Sb2S4+H=HSb2S4)=11.50
Ks(Sb2S3+HS=Sb2S4+H)=-14.00
Ks(Sb2S3+HS=HSb2S4)=-2.50

Critical evaluation of literature data for the solubility of Sb2S3 in sulfide media. Data for 25-300 C.

Sb+++ sol oth/un 25°C var M T M 1988KRd (14462) 60
K(Sb2S3(s)+H2S=H2Sb2S4)=-5.15
K(Sb2S3(s)+H2S=HSb2S4+H)=-10.1
K(Sb2S3(s)+H2S=Sb2S4+2H)=-19.6

Also K(Sb2S3(s)+2H2O=Sb2S2(OH)2+H2S)=-7.44(200 C). Constants at I=0,25-350 C

Sb+++ sol oth/un 25°C var U 1966ADa (14463) 61
Ks(2Sb2S3(s)+SH+OH=Sb4S7)=0.7

Sb+++ oth none 25°C 0.0 U 1964PCa (14464) 62
From thermodynamic data. K(0.5Sb2L3(s)+H2O+H=SbO+1.5H2L(g))=-13.9
K(0.5Sb2L3(s)+3H2O=Sb(OH)3+1.5H2L(g))=-14.7

Sb+++ sol oth/un 30°C var U 1962DGc (14465) 63
K(Sb2L3(s)+L=Sb2L4)=2.08

Sb+++ sol oth/un 20°C var U 1956BLa (14466) 64
K(0.5Sb2L3(s)+0.5L=SbL2)=0.45. K(Sb2L3(s)+2OH=SbL2+SbL(OH)2)=-1.10

Sb+++ sol none 25°C 0.0 U 1953AKa (14467) 65
K(0.5Sb2L3(s)+1.5L=SbL3)=0.89
K(Sb2L3(s)+HL=HSb2L4?)=-2.33
Kso(Sb2L3)=-92.77

I=0 corr. K(0.5Sb2L3(s)+3H+4Cl=SbCl4+1.5H2L)=-12.24
K(0.5Sb2L3(s)+3OH=0.5SbL3+0.5SbO3+1.5H2O)=4.015. Kso from thermodynamic data

S04-- H2L Sulfate CAS 7664-93-9 (15)
Sulfate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp oth/un ? var U 1970Dwa (16530) 66

$$B(\text{SbO}+\text{L})=0.3$$

Medium: 1-4 M H₂SO₄, $K(\text{SbO}+2\text{H}_2\text{L}=\text{SbL}_2+\text{H}_2\text{O}+2\text{H})=-1.0$ (10-18 M H₂SO₄)

CH2O2 HL Formic acid CAS 64-18-6 (37)

Methanoic acid; H.CO₂H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt NaClO₄ 20°C 0.70M U K1=4.60 B2=9.53 1975W0a (17645) 67

CH4N2S L Thiourea CAS 62-56-6 (51)

Thiocarbamide, Thiourea; (H₂N)₂CS

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ cal oth/un 25°C 3.0M U IH 1984VRb (17856) 68

$$K(\text{SbO}+2\text{L})=2.80$$

in 3 M HClO₄; also for 4 M HClO₄ $K=3.48$ $\text{DH}=-103.43$ kJ/mol

for 5 M HClO₄ $K=4.38$ $\text{DH}=-108.37$ kJ/mol

Sb+++ sp NaClO₄ 25°C 3.00M U I 1979VSb (17857) 69

$$K(\text{SbO}+2\text{L})=2.72$$

CH4O L Methyl alcohol CAS 67-56-1 (597)

Methanol; CH₃.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ EMF alc/w 20°C 100% U I 1971GSa (17898) 70

$$K(\text{Sb}+\text{H}-1\text{L})=11.85$$

$$K(\text{SbH}-1\text{L}+\text{H}-1\text{L})=10.26$$

$$K(\text{SbH}-2\text{L}_2+\text{H}-1\text{L})=9.07$$

$$K(\text{Sb}+2\text{L}=\text{SbH}-2\text{L}_2+2\text{H}) > 1$$

Medium: MeOH, 1 M LiCl. With 1 M Li tosylate: $K(\text{Sb}(\text{H}-1\text{L})_2+\text{H}-1\text{L})=12.29$;

$K(\text{Sb}(\text{H}-1\text{L})_2+\text{Sb}(\text{H}-1\text{L})_3=\text{Sb}_2(\text{H}-1\text{L})_5)=2.36$

C2H4O2 HL Acetic acid CAS 64-19-7 (36)

Ethanoic acid; CH₃.CO₂H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt NaClO₄ 20°C 0.70M U K1=7.00 B2=12.64 1975W0a (20158) 71

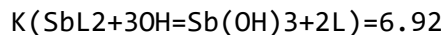
C2H4O2S H2L Thioglycolic CAS 68-11-1 (596)

Mercaptoethanoic acid; HS.CH₂.CO₂H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ gl NaClO₄ 20°C 0.10M U M 1970AMa (20363) 72

$$K(\text{SbL}_2=\text{SbL}_2(\text{OH})+\text{H})=7.58$$



C2H5NO2 HL Glycine CAS 56-40-6 (85)
2-Aminoethanoic acid; $\text{H}_2\text{N}\cdot\text{CH}_2\cdot\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt KCl 30°C 0.10M C K1=10.60 1982MNa (21700) 73
Method: polarography. By potentiometry, $K(\text{H}+\text{L})=9.53$

C2H6OS HL CAS 60-24-2 (841)
2-Mercaptoethanol; $\text{HS}\cdot\text{CH}_2\cdot\text{CH}_2\cdot\text{OH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ gl NaClO4 20°C 0.10M U 1970AMa (22078) 74
 $K(\text{SbH}-2\text{L}_2+\text{H})=17.98$

C3H4O4 H2L Malonic acid CAS 141-82-2 (79)
Propanedioic acid; $\text{CH}_2(\text{COOH})_2$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt NaClO4 20°C 0.70M U K1=10.18 B2=26.52 1975W0a (24543) 75

C3H6O2 HL Propionic acid CAS 79-09-4 (35)
Propanoic acid; $\text{CH}_3\cdot\text{CH}_2\cdot\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt NaClO4 20°C 0.70M U K1=6.68 B2=11.20 1975W0a (25047) 76

C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)
L-2-Hydroxypropanoic acid; $\text{CH}_3\cdot\text{CH}(\text{OH})\cdot\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt NaClO4 20°C 0.70M U K1=7.84 B2=12.00 1975W0a (25530) 77

C4H4O4 H2L Maleic acid CAS 110-16-7 (111)
cis-Butenedioic acid; $\text{HOOC}\cdot\text{CH}:\text{CH}\cdot\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt NaClO4 20°C 0.70M U K1=7.78 B2=14.95 1975W0a (29128) 78

C4H6O4 H2L Succinic acid CAS 110-15-6 (112)
1,4-Butanedioic acid; $\text{HOOC}\cdot\text{CH}_2\cdot\text{CH}_2\cdot\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt NaClO4 20°C 0.70M U K1=8.70 B2=17.60 1975W0a (30033) 79

C4H6O4S H3L Thiomalic acid CAS 70-49-5 (109)
2-Mercaptosuccinic acid, 2-Sulfanyl-1,4-butanedioic acid; HOOC.CH(SH).CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ gl NaClO4 20°C 0.10M U 1970AMa (30361) 80

K(SbHL2+H)=2.4
K(SbL2+H)=3.46
K(SbL2+3OH=Sb(OH)3+2L)=5.90

C4H6O4S2 H4L CAS 2418-14-6 (4264)

2,3-Dimercaptobutanedioic acid; HOOC.CH(SH).CH(SH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ gl NaClO4 20°C 0.10M U 1970AMa (30396) 81

K(SbL=SbLOH+H)=-4.90
K(SbH3L2+H)=2.57
K(SbH2L2+H)=3.60
K(SbHL2+H)=4.61

K(SbL2+H)=6.82; K(2SbLOH+OH=Sb(OH)3+SbL2)=-10.7

C4H6O4S2 H4L CAS 304-55-2 (3002)

meso-2,3-Dimercaptobutanedioic acid (meso-dithiotartaric acid)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ gl NaClO4 20°C 0.10M U 1970AMa (30434) 82

K(Sb2L2=SbLOH+2H)=13.17
K(Sb2L2+6OH=2Sb(OH)3+2L)=9.2

C4H6O5 H2L Malic acid CAS 617-48-1 (393)

2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt NaClO4 20°C 0.70M U K1=8.54 B2=17.18 1975W0a (30715) 83

C4H6O6 H2L DL-Tartaric acid CAS 133-37-9 (94)

DL-Tartaric acid,DL-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt NaClO4 20°C 0.70M U K1=8.08 B2=14.81 1975W0a (31029) 84

C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)

L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Sb+++      gl  NaClO4 20°C 0.10M U    M                      1970AMa (31346) 85
K(Sb2(H-2L)2+20H=Sb2(H-2L)OH)=10.16, K(Sb2(H-2L)2+20H=2Sb(OH)3+2L)=6.07
*****
C5H7N04S2          H3L                      CAS 36061-59-3 (1953)
Bis(carboxymethyl)dithiocarbamic acid; (HOOC.CH2)2.N.CSSH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Sb+++      EMF KNO3  22°C 1.00M U          K1=8.97  B2=17.61 1970TPb (37560) 86
                               B3=25.99
*****
C5H8O4          H2L  Glutaric acid  CAS 110-94-1 (420)
Pentanedioic acid; HOOC.CH2.CH2.CH2.COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Sb+++      vlt NaClO4 20°C 0.70M U          K1=9.48  B2=18.78 1975W0a (38350) 87
*****
C5H11NS2        HL                      CAS 147-84-2 (2126)
Diethyldithiocarbamic acid; (CH3.CH2)2N.CSSH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Sb+++      sp non-aq  ? 100% U    M                      1968SRg (41369) 88
                               K(SbAL+2HL=SbL3+H2A)=2.47
Medium: CCl4. H2A=dithizone
*****
C6H5ClO3        HL  Chlorokojic ac  CAS 7559-81-1 (8317)
2-Chloromethyl-5-hydroxy-4H-pyran-4-one;
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Sb+++      vlt KCl   30°C 0.10M C          B2=18.3          1985KNa (42337) 89
Method: polarography
*****
C6H6O2          H2L  Catechol        CAS 120-80-9 (534)
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Sb+++      gl  NaClO4 20°C 0.10M U          1970AMa (43818) 90
                               K(SbL2+H=SbLOH+H2L)=2.37
                               K(SbL2+OH=Sb(OH)3+2HL)=-5.44
*****
C6H6O3          HL  Maltol          CAS 118-71-8 (2442)
3-Hydroxy-2-methyl-4H-pyran-4-one;
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Sb+++	vlt	KCl	30°C	0.10M	C		B2=21.3	1985KNa (44100)	91
Method: polarography									

C6H6O4		HL		Kojic acid			CAS 501-30-4	(1800)	
5-Hydroxy-2-(hydroxymethyl)-4H-pyran-4-one;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Sb+++	vlt	KCl	30°C	0.10M	C		B2=19.7	1985KNa (44241)	92
Method: polarography									

C6H6O8S2		H4L		Tiron			CAS 149-45-1	(104)	
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)2.C6H2(SO3H)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Sb+++	gl	KN03	25°C	0.10M	U	I	K2=14.5 K(SbL2+H)=2.00 K(SbL+H2L=SbL2+2H)=-5.73	19710Bb (44486)	93
Medium: 0.1 M KCl: K(SbL+H2L=SbL2+2H)=-6.01									

Sb+++	gl	NaClO4	20°C	0.10M	U		K(SbL2+H=SbLOH+H2L)=1.23 K(SbL2+OH=Sb(OH)3+2HL)=-3.95	1970AMa (44487)	94

C6H9NO6		H3L		NTA			CAS 139-13-9	(191)	
Nitrilotriethanoic acid; N(CH2.COOH)3									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Sb+++	vlt	NaCl	25°C	4.0M	C	H	K1=11.66	1984GSd (47006)	95
Method: polarography. Medium pH 2.2. DH(K1)=-31.9 kJ mol-1; DS(K1)=117 J K-1 mol-1.									

C7H6O4		H3L					CAS 303-38-8	(1398)	
2,3-Dihydroxybenzoic acid; C6H3(OH)2.COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Sb+++	gl	NaClO4	20°C	0.10M	U		K(SbL2+H=SbLOH+H2L)=2.8 K(SbL+2OH=Sb(OH)2+HL)=-4.17	1970AMa (54471)	96

C8H8O4		L					(601)		
4,5-Dimethoxy-1,2-benzoquinone;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo

Sb+++ nmr non-aq 34°C 100% U M 1981KKc (60113) 97
 K(SbCl3+L)=0.68

Medium: nitromethane

C9H7N03S2 H2L CAS 58447-10-2 (4675)
 8-Mercaptoquinoline-5-sulfonic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp oth/un ? ? U K1=13.7 B2=26.1 1968ABa (64429) 98

C10H9N03S2 HL (7206)
 6-Methyl-5-sulfo-8-mercaptoquinoline;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp oth/un 20°C 0.10M U K1=14.25 B2=25.90 1985DAb (70179) 99

C10H16N2O8 H4L EDTA CAS 60-00-4 (120)
 1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ vlt NaNO3 25°C 4.00M U H K1=19.48 1982WEa (74131) 100
 DH(K1)=-25.140 kJ mol-1; DS=88 J K-1 mol-1

 Sb+++ gl KNO3 25°C 0.10M U 19710Bb (74132) 101

K(SbL+H)=1.02
 K(SbL+OH)=8.24
 K(SbLOH+H2O=SbL(OH)2+H)=7.46

 Sb+++ gl NaClO4 20°C 0.10M U T 1970AMa (74133) 102

K(SbL+2OH=Sb(OH)3+HL)=12.46

 Sb+++ sp NaClO4 25°C 1.0M U 1965BIb (74134) 103

K(SbO+L+2H)=24.8
 K(SbL+OH)=-8.7

C10H18N2O7 H3L HEDTA CAS 150-39-0 (392)
 N-(Hydroxyethyl)diaminoethane-N,N',N'-triethanoic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ gl KNO3 25°C 0.10M U 19710Bb (75488) 104

K(SbL+H2O=SbLOH+H)=-3.05

 Sb+++ gl NaClO4 20°C 0.10M U 1970AMa (75489) 105

K(SbL+OH=Sb(OH)3+H2L)=4.58
 K(SbHL=SbL+H)=-3.10

Sb+++ sp NaClO4 25°C 1.0M U 1966BIb (75490) 106
 K(SbO+L+2H)=20.2
 K(SbH-1L+H)=-3.2
 K(SbH-1L+OH)=-8.1

 C11H8O3 L CAS 18916-57-9 (581)
 4-Methoxy-1,2-naphthoquinone;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp non-aq 34°C 100% U HM 1981KKb (77140) 107
 K(SbCl3+L)=0.63

Medium: nitromethane

 C12H10N2 L CAS 103-33-3 (4893)
 Azobenzene; C6H5.N:N.C6H5

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp non-aq ? 100% U B2=4.37 1969KNa (80654) 108
 Medium: dichloroethane

 C12H10N2O HL Solvent Yellow7 CAS 1689-82-3 (1106)
 4-Hydroxyazobenzene; C6H5.N:N.C6H4.OH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp non-aq ? 100% U K(Sb+2HL)=1.66 1969KNa (80689) 109

Medium: dichloroethane

 C12H11N3 L CAS 64-09-3 (4897)
 4-Aminoazobenzene; C6H5.N:N.C6H4.NH2

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp non-aq ? 100% U K1=1.68 1969KNa (80861) 110
 Medium: dichloroethane

 C12H27O4P L CAS 126-73-8 (2432)
 Tri-n-butyl phosphate; (C4H9O)3PO

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp oth/un ? ? U M 1973RGa (84122) 111
 K(SbBr3+L)=2.63
 K(SbBr3+2L)=3.24

 C13H11N L CAS 538-51-2 (4969)
 Benzylideneaniline; C6H5.N:CH.C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp non-aq ? 100% U K1=1.77 1969KNa (85008) 112
Medium: dichloroethane

C13H11N3O4S2 HL Tenoxicam CAS 59804-37-4 (8393)
4-Hydroxy-2-methyl-N-2'-pyridinyl-2H-thien[2,2-e]-1,2-thiazine-3-carboxamide-1,1-di
oxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ gl mixed 25°C 50% C K1=3.5 2002MNa (85290) 113
Medium: 50% v/v CH3CN/H2O, 0.05 M NaNO3.

C13H12N2 L CAS 949-87-1 (4971)
4-Methylazobenzene; CH3.C6H4.N:N.C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp non-aq ? 100% U B2=4.07 1969KNa (85322) 114
Medium: dichloroethane

C14H14N2 L CAS 584-90-7 (5028)
2,2'-Dimethylazobenzene; CH3.C6H4.N:N.C6H4.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp non-aq ? 100% U B2=4.82 1969KNa (87642) 115
Medium: dichloroethane

C14H14N2 L CAS 561-60-6 (5029)
4,4'-Dimethylazobenzene; CH3.C6H4.N:N.C6H4.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp non-aq ? 100% U B2=1.30 1969KNa (87644) 116
Medium: dichloroethane

C14H14N2O L CAS 7466-38-8 (5066)
4-Ethoxyazobenzene; CH3.CH2.O.C6H4.N:N.C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp non-aq ? 100% U B2=1.66 1969KNa (87653) 117
Medium: dichloroethane

C14H14N4OBr2 HL CAS 35601-32-2 (5092)
5-(3,5-Dibromo-2-pyridylazo)-2-ethylamino-4-hydroxy-1-methylbenzene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Sb+++	dis	oth/un	?	?	U		K1=5.12	1967GUa (87689)	118
By spectrophotometry : K1=5.49									

C14H15N3		L					(5034)		
4-Dimethylaminoazobenzene; C6H5.N:N.C6H4.N(CH3)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Sb+++	sp	non-aq	?	100%	U		K1=2.70	1969KNa (87751)	119
Medium: dichloroethane									

C14H15N4OBr		HL					CAS 14337-50-9 (5095)		
5-(5-Bromo-2-pyridylazo)-2-ethylamino-4-hydroxy-1-methylbenzene;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Sb+++	dis	oth/un	?	?	U			1967GUa (87769)	120
K(?)=4.96									

C14H16N4O		HL		PAAC			CAS 13059-69-3 (5067)		
5-Ethylamino-4-methyl-2-(2'-pyridylazo)phenol;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Sb+++	dis	oth/un	?	?	U			1967GPa (88020)	121
K(?)=5.53									

C14H22N2O8		H4L		CDTA			CAS 482-54-2 (200)		
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Sb+++	vlt	NaNO3	25°C	4.00M	U	H	K1=24.78	1982WEa (88768)	122
DH(K1)=71.2 kJ mol-1; DS=234.6 J K-1 mol-1									
Sb+++	gl	NaClO4	20°C	0.10M	U			1970AMa (88769)	123
K(SbL+2OH=Sb(OH)3+HL)=11.24									

C14H23N3O10		H5L		DTPA			CAS 67-43-6 (238)		
Diethylenetriamine-pentaethanoic acid; HOOC.CH2.N(CH2.CH2.N(CH2.COOH)2)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Sb+++	gl	KNO3	25°C	0.10M	U			19710Bb (89376)	124
K(SbL+H)=3.31									
Sb+++	gl	NaClO4	20°C	0.10M	U			1970AMa (89377)	125
K(SbL+2OH=Sb(OH)3+HL)=9.82									

K(SbL+H)=3.57

C15H16N2 L CAS 889-37-2 (5104)
(4-Dimethylamino)benzalaniline; (CH3)2N.C6H4.CH:N.C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp non-aq ? 100% U K1=4.32 1969KNa (91919) 126
Medium: dichloroethane

C15H16N2 L CAS 58758-12-6 (5103)
Benzal-(4-dimethylamino)aniline; C6H5.CH:N.C6H4.N(CH3)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp non-aq ? 100% U K1=3.82 B2=1.25 1969KNa (91920) 127
Medium: dichloroethane

C17H20N4O2 H2L CAS 39965-80-5 (5221)
1,3-Dihydroxy-4-(2-N-methylanabasiny1-alpha-azo)benzene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Sb+++ sp mixed ? 60% U 1972TDa (96306) 128
K(SbCl4+H3L)=5.48

Medium: 60% v/v acetone, 0.1 M KCl

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EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES
I Data with various BACKGROUNDS
H Data for THERMOCHEMICAL quantities
M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC

END