

## SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 25 experiments for

(no ligands specified)

2 metals : Se(IV), Se(not IV)

(no references specified)

(no experimental details specified)

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e-                    HL     Electron                    (442)  
 Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Se(IV)	EMF	none	25°C	0.0	U				19660Va	(931) 1
								K=50.3, 744 mV		
K: H <sub>2</sub> SeO <sub>3</sub> + 4H + 4e = Se(s) + 3H <sub>2</sub> O										
Se(IV)	oth	none	25°C	0.0	U				1952LAb	(932) 2
								K=38.9(1150 mV)		
K: Se(VI)O <sub>4</sub> +4H+2e=H <sub>2</sub> SeO <sub>3</sub> +H <sub>2</sub> O. From thermodynamic data. K(H <sub>2</sub> SeO <sub>3</sub> +4H+4e=Se(s)+3H <sub>2</sub> O)=50.04(740 mV) plus others										

Br-                    HL     Bromide                    CAS 10035-10-6 (19)  
 Bromide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Se(IV)	sp	non-aq	25°C	100%	U	T	H		1975WSa	(2308) 3
								K(SeO <sub>2</sub> +Br)=1.11		
								K(SeOCl <sub>2</sub> +Br)=1.49		

Medium: DMSO. DH(SeO<sub>2</sub>L)=-11.1; DS=-59. DH(SeOCl<sub>2</sub>L)=21.3 kJ mol<sup>-1</sup>; DS=96 J K<sup>-1</sup> mol<sup>-1</sup>

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CN-                    HL     Cyanide                    CAS 74-90-8 (230)  
 Cyanide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Se(IV)	sol	oth/un	0°C	dil	U	T			1961HAb	(2761) 4
								K(Se(s)+HL=SeCN+H)=-3.11		

Medium: not specified; at 0.3 C. Se(all)=Se(0). At 10.8 C, K=-3.33

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Cl-                    HL     Chloride                    CAS 7647-01-0 (50)  
 Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Se(IV)	sp	non-aq	25°C	100%	U	T	H		1975WSa	(5699) 5
								K(SeO <sub>2</sub> +Cl)=1.50		

$K(\text{SeOCl}_2 + \text{Cl}) = 1.08$   
Medium: DMSO.  $DH(\text{SeO}_2\text{L}) = -4.1$ ;  $DS = -42$ .  $DH(\text{SeOCl}_2\text{L}) = 24.3 \text{ kJ mol}^{-1}$ ;  $DS = 100 \text{ J K}^{-1} \text{ mol}^{-1}$

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Se(IV) EMF non-aq 25°C 100% U 1969DTa (5700) 6  
 $K(\text{SeO}_2\text{L}_2 = \text{SeO}_2\text{L} + \text{L}) = -9.7$

Medium:  $\text{SeOCl}_2$ , 0.5 M  $(\text{C}_2\text{H}_5)_4\text{NClO}_4$

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I- HL Iodide CAS 10034-85-2 (20)  
Iodide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

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Se(IV) sp non-aq 25°C 100% U T H  $K_1 = 0.95$  1975WSa (8373) 7  
Medium: DMSO.  $DH(\text{SeO}_2\text{L}) = 12.2 \text{ kJ mol}^{-1}$ .  $DS = -56 \text{ J K}^{-1} \text{ mol}^{-1}$

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NO3- HL Nitrate CAS 7697-37-2 (288)  
Nitrate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

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Se(IV) ix oth/un 25°C >3 U  $K_1 = 2.34$   $B_2 = 3.47$  19850Ka (9917) 8  
 $K_3 = 0.69$   
 $K_4 = 0.45$   
 $K_5 = 0.28$   
 $K_6 = 0.13$

\*\*\*\*\*  
OH- HL Hydroxide (57)  
Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

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Se(IV) oth NaClO4 20°C 2.50M U  $B_2 = 29.0$  19840Ka (12121) 9  
 $K_3 = 14.15$   
 $K_4 = 12.30$

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CH40 L Methyl alcohol CAS 67-56-1 (597)  
Methanol; CH3.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

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Se(IV) EMF alc/w 20°C 100% U 1971GSa (17899) 10  
 $K(\text{Se} + 4\text{L} = \text{Se}(\text{H}-1\text{L})_4 + 4\text{H}) > 1$

Medium: MeOH, 1 M Me4NCl

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C5H11NS2 HL CAS 147-84-2 (2126)  
Diethyldithiocarbamic acid;  $(\text{CH}_3\text{CH}_2)_2\text{N.CSSH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Se(IV) sp non-aq ? 100% U M 1968SRg (41370) 11  
 $K(Se(HA)4+4HL=SeL4+4H2A)=6.5$

Medium: CCl<sub>4</sub>. H<sub>2</sub>A=dithizone

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C6H7NS HL CAS 137-07-5 (3098)  
2-Aminothiophenol (o-aminothiophenol); H2N.C6H4.SH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Se(IV) sp oth/un 25°C 2.00M U 1971BSi (45089) 12  
I=2.0-2.4.  $K(H2SeO3+4HL=SeL2+L2+3H2O)=16.2$  (L2=PhSSPh, SeL2=PhSSeSPh)

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C7H5NS2 HL CAS 149-30-3 (3752)  
2-Mercaptobenzo-1,3-thiazole;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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Se(IV) sp oth/un 25°C 1.80M U 1971BSi (53085) 13  
Range of ionic strength 1.7-2.0.  $K(H2SeO3+4RSH=RSSeSR+RSSR+3H2O)=13.3$   
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C7H6O2S H2L Thiosalicylic CAS 147-93-3 (236)  
2-Mercaptobenzoic acid; HS.C6H4.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Se(IV) sp oth/un 25°C 2.0M U 1971BSi (53916) 14  
Range of ionic strength 2.0-2.8.  $K(H2SeO3+4RSH=RSSeSR+RSSR+3H2O)=16.7$

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C14H9N04 H2L Alizarin Maroon CAS 3963-78-8 (1052)  
3-Amino-1,2-dihydroxyanthraquinone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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Se(IV) gl NaClO<sub>4</sub> 20°C 0.10M U M K1=6.9 B2=12.90 1982ISa (86813) 15  
Se(IV), selenite

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C28H15N04 L CAS 82-22-4 (3522)  
1,1'-Iminodianthraquinone; (1,1'-dianthrimide)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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Se(IV) sp mixed ? 96% U 1966DLa (104654) 16  
 $K(2HSeO2+HL)=8.75$

Medium: 96% H<sub>2</sub>S<sub>04</sub>

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Se(IV) sp oth/un 70°C 96% U 1959LSa (104655) 17  
 $K(H2SeO3+HL=HSeO2L(?))=5.04$

Medium: 96.25% H<sub>2</sub>S<sub>04</sub>

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C28H15N04 L (4171)

1,2'-Iminodianthraquinone (1,2'-dianthrimide)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Se(IV)	sp	mixed	?	96%	U				1966MLa (104658)	18

$$K(HSeO_2+HL)=4.40$$

Medium: 96.0% H<sub>2</sub>S04

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C28H15N04 L CAS 30999-75-8 (4172)

2,2'-Iminodianthraquinone (2,2'-dianthrimide)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Se(IV)	sp	mixed	?	96%	U				1966DLa (104659)	19

$$K(HSeO_2+HL)=3.89$$

$$K(2HSeO_2+2HL)=12.50$$

Medium: 95.5% H<sub>2</sub>S04

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e- HL Electron (442)  
Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Se(not IV)	oth	none	25°C	0.0	U				1952LAb (933)	20
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$$K=-12.5(-370 \text{ mV})$$

K: Se(s)+2H+2e=H<sub>2</sub>Se(g). From thermodynamic data. K(Se(s)+2e=Se(II))=-31.2 (-920 mV)

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CN- HL Cyanide CAS 74-90-8 (230)  
Cyanide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Se(not IV)	sol	oth/un	0°C	dil	U	T			1961HAb (2762)	21
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$$K_s(Se(s)+HL=Se(\theta)CN+H)=-3.11$$

At 10.8 C, K=-3.33

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CO L Carbon monoxide CAS 630-08-0 (551)  
Carbon monoxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Se(not IV)	nmr	non-aq	120°C	100%	U	T	H		1994KRa (2823)	22
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Metal:Co(0). Method:NMR. Medium:Carbon monoxide. T. 120-225 C

K:Co2L8=2CoL4. DH=79.5 kJ mol-1; DS=121

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SO<sub>3</sub>-- H2L Sulfite CAS 7782-99-2 (801)  
Sulfite;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Se(not IV) oth oth/un 20°C var U T 1947G0a (15477) 23  
K(L+Se(s)=SeS03)=0.64

By chemical analysis. Medium: Na2L. K=-0.09(97.5 C)

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SeCN- HL Selenocyanate CAS 73102-11-2 (440)  
Selenocyanate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Se(not IV)	sp	non-aq	2°C	100%	U				1972CPb (16994)	24
K((SeCN)2+L)=4.3										
Medium: acetonitrile, 0.1 M LiClO4. K1=4.6 from current-voltage studies										
C2H6OS		HL					CAS	60-24-2 (841)		
2-Mercaptoethanol; HS.CH2.CH2.OH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Se(not IV)	vlt	NaCl	0°C	0.20M	U	M			1993BSa (22079)	25
Kox((4Fe-4S)+L)=4.55										
Kred((4Fe-4S)+L)=1.01										

Metal:Fe(oxidation state unknown).

(4Fe-4S)++ and (4Fe-4S)+ are in Ferrodoxin III.

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## EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES  
H Data for THERMOCHEMICAL quantities  
M Data for TERNARY Complexes

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END