

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 254 experiments for

(no ligands specified)

2 metals : Ti⁺⁺⁺, Ti⁺⁺⁺⁺

(no references specified)

(no experimental details specified)

e- HL Electron (442)

Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti ⁺⁺⁺	oth	oth/un	25°C	0.00	U				19630Ra (962)	1
								K(Ti+e=Ti ⁺⁺)=-34, -2000 mV		
Ti ⁺⁺⁺	EMF	oth/un	0°C	var	U				1924FHa (963)	2
								K(Ti+e=Ti(II))=-6.8(-370 mV)		

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti ⁺⁺⁺	sp	oth/un	25°C	1.61M	U	I		K1=1.61	1975FBa (5788)	3
Ti ⁺⁺⁺	dis	NaClO ₄	25°C	4.0M	U				1975HKa (5789)	4
								K3=0.51		
								K4=0.33		
Ti ⁺⁺⁺	EMF	non-aq	25°C	100%	U				1971DTb (5790)	5
								K3=4.92		
								K4=2.92		

Medium: SeOCl₂, 0.5 M Et₄NClO₄

Ti ⁺⁺⁺	sp	oth/un	rt	var	U		B2=0.37	1971KGa (5791)	6
							K(TiCl ₂ +3H+4Cl=H ₃ TiCl ₆)=-7.8		

Medium: HCl

Ti ⁺⁺⁺	sp	KCl	25°C	var	U		K1=0.56	1971PLa (5792)	7
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Ti ⁺⁺⁺	sp	oth/un	25°C	0.0	U		K1=-1	1967GAa (5793)	8
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Ti ⁺⁺⁺	ix	NaClO ₄		3.0M	U		K1=0.55	B2=0.15	1967NKe (5794)	9
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Metal: Ti⁰⁺⁺. In LiCl var: K3=-1.03, K4=-1.1

Ti ⁺⁺⁺	sp	oth/un	?	12.0M	U			1957J0b (5795)	10
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Medium: HCl. K(Ti(III)Cl₂+Ti(IV)Cl₆=Ti₂Cl₇(?))=1.08

Ti+++ kin NaClO₄ 40°C 0.50M U K1=0.34 1954DQa (5796) 11

F- HL Fluoride CAS 7644-39-3 (201)
Fluoride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	ix	KCl	?	0.20M	U				1966NAb	(7255) 12
								K(TiO+L)=6.65		
								K(TiOL+L)=5.09		
								K(TiOL2+L)=4.58		
								K(TiOL3+L)=4.06		

Medium: HCl

OH- HL Hydroxide (57)
Hydroxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	gl	KCl	25°C	1.00M	C				1988PFa	(12272) 13
								*K1=-2.59		
								*B(2,2)=-3.03		
Ti+++	vlt	KCl	25°C	1.0M	U				1983TMb	(12273) 14
								K[Ti(OH)+H]=2.14		
Ti+++	sol	oth/un	18°C	1.00M	U			K1=12.30 B2=22.57	1981NMB	(12274) 15
								B3=32.32		

Ti+++	nmr	oth/un	var	1.0M	U	TI			1978SSd	(12275) 16
								*K1=-3.9		
								*K1=-3.7 (I=0.3)		
								*K1=-3.6 (I=0.1)		

Medium: KBr in D₂O. Method: esr. 0-60 C. K(Ti(H₂O)₆=Ti(H₂O)₅OH+H)=-3.85

Ti+++	kin	NaCl	25°C	1.0M	C				1977BMi	(12276) 17
								*K1=-2.46		

Ti+++	kin	KCl	25°C	0.50M	U	I			1973BLc	(12277) 18
								*K1=-1.4		

In 0.5 M LiClO₄, *K1=-1.4

Ti+++	kin	oth/un	25°C	0.50M	U				1973LBa	(12278) 19
								*K1=-1.9		

Medium: LiCl

Ti+++	vlt	mixed	25°C		U			K1=14.0 B2=25.2	1972LIA	(12279) 20
								B3=32.4		

Medium: ethylene glycol and HCl varied

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	nmr	non-aq	17°C	100%	U	T	HM		1987SEc (51514)	54
K(Ti(2,4-C7H11)2 + L)=2.01										
Data for the reaction of open titanocene [Ti(2,4-C7H11)2] with L at var. T.										
DH=44.4 kJ mol-1, DS=115 J K-1 mol-1. Medium: THF										

C6H15P		L						CAS 554-70-1 (166)		
Triethylphosphine; (C2H5)3P										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	nmr	non-aq	-4°C	100%	U	T	HM		1987SEc (51549)	55
K(Ti(2,4-C7H11)2 + L)=1.09										
Data for the reaction of open titanocene [Ti(2,4-C7H11)2] with L at var. T.										
DH=41.8 kJ mol-1, DS=135 J K-1 mol-1. Medium: THF										

C7H6O3		H2L	Salicylic acid		CAS 69-72-7 (14)					
2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	sp	KCl	20°C	1.00M	U		M		1973VGa (54309)	56
K(TiA2+HL=TiA2L+H)=4.68										
K(TiH2A+H2L=(TiH2A)H2L)=11.7										
H2A= 4-(2-pyridylazo)resorcinol										

C8H11P		L						CAS 672-66-2 (2290)		
Dimethyl-phenyl-phosphine; (CH3)2.P.C6H5										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	nmr	non-aq	20°C	100%	U	T	HM		1987SEc (61325)	57
K(Ti(2,4-C7H11)2 + L)=2.38										
Data for the reaction of open titanocene [Ti(2,4-C7H11)2] with L at var. T.										
DH=54.0 kJ mol-1; DS=139 J K-1 mol-1. Medium: THF										

C10H8N2		L	2,2'-Bipyridyl		CAS 366-18-7 (25)					
2,2'-Bipyridine; (C5H4N)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	sp	KCl	50°C	0.50M	U			K1=2.86 B2=4.75 B3=6.36	1970TNb (69653)	58
K3=25.28										

Ti+++	con	none	?	0.0	U				1959KMa (69654)	59
K3=25.28										

C10H16N208		H4L	EDTA					CAS 60-00-4 (120)		

1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	sp	KCl	20°C	1.20M	U				1973YPa (74225)	60
								$K(Ti+H2L)=7.50$		

Ti+++	vlt	NaNO ₃	25°C	0.10M	U	T	K1=21.3		1954PMb (74226)	61
							$K(TiO+L)=17.3$			

C12H8N₂ L Phenanthroline CAS 66-71-7 (144)

1,10-Phenanthroline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	sp	oth/un	20°C	?	U			K1=3.78 B2=8.29 B3=12.50	1969TNa (80522)	63

C20H14N20S5 H3L Solochrome 6B CAS 3564-14-5 (3507)

1-(1-Hydroxy-2-naphthylazo)-2-naphthol-4-sulfonic acid, Mordant Black3, Eriochrome blue-black B;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	sp	oth/un	?	0.10M	U			K1=3.84 B2=7.82	1972TNa (99665)	64

C20H16N40S5 H2L Eriochrome Red B CAS 14954-75-7 (3510)

4-(4,5-Dihydro-3-Me-5-oxo-1-Phe-1H-pyrazol-4-ylazo)-3-naphthol-1-sulfonic acid;

C23H18O9S H4L Eriochrome cyan CAS 3564-18-9 (433)

4'-Hydroxy-3,3'-dimethyl-2''-sulfofuchsone-5,5'-dicarboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti+++	sp	oth/un	?	?	U			K1=5.92 B2=11.60	1973TPb (102637)	66
By polarography:								$K1=6.89, B2=10.39$		

e- HL Electron (442)

Electron;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	con	oth/un	?	dil	U				1961KPd	(6381) 77
								$K(TiOL+L)=1.61$		

Ti++++	dis	oth/un	?	?	U	M			1955DEa	(6382) 78
								$Kd(TiOH+H+4L=TiL4(CH_3COCH_2CH(CH_3)_2))=-3.63$		

F-		HL		Fluoride			CAS	7644-39-3	(201)	
Fluoride;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	ISE	NaClO4	25°C	3.00M	U				1983CPa	(7256) 79
								$K(Ti(OH)_2+HF=Ti(OH)_2F)=2.28$		
								$K(Ti(OH)_2+4HF=TiF_4)=8.34$		

Ti++++	kin	NaClO4	25°C	0.50M	U				1977TTa	(7257) 80
								$K(TiO+HL=TiOL+H)>2.30$		

Ti++++	ix	oth/un	?	?	U				1972PAb	(7258) 81
								$K_6=3.79$		

Ti++++	kin	oth/un	25°C	dil	U	T			1972RTa	(7259) 82
								$K(TiF_4(OH)+HF+F=TiF_6)=5.85$		
At 0 C: K=6.48										
Ti++++	EMF	oth/un	?	0.50M	U				1967PMa	(7260) 83
								$K_6=3.81$		
Medium: HCl										
Ti++++	EMF	NaClO4	25°C	3.0M	U		$K_1=>5.38$	$K_2=4.35$	1960CCa	(7261) 84
							$K_3=3.96$			
							$K_4=3.72$			
Metal: TiO ?. Method: quinhydrone electrode in HClO4										
Ti++++	sp	KNO3	?	0.10M	U				1952KLa	(7262) 85
								$K(TiO(?) + L)=6.44$		

OH-		HL		Hydroxide			(57)			
Hydroxide;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	nmr	NaClO4	25°C	4.0M	C				1987CMD	(12284) 86
								$K(3TiO=(TiO)_3OH+H)=0.21$		
								$K(3TiO=Ti_3O_4+2H)=-0.42$		
								$K(4TiO=(TiO)_4)=0.36$		

Method: ¹⁷⁰nmr and light scattering data.

Ti++++	gl	KNO ₃	37°C	0.32M	U	1985TMb (12285)	87
						*K(TiA ₂ (H ₂ O) ₂)=-3.5	
						*K(TiA ₂ (H ₂ O)OH)=-4.35	
A=cyclopentadiene							
Ti++++	EMF	NaCl	25°C	2.0M	U H	1981EKb (12286)	88
Spectroscopy also used. K(TiO+2H ₂ O=TiO(OH) ₂ (sat.)+2H)=-4.7							
K(TiO(OH) ₂ =TiO(OH) ₂ (solid))=-27.3							
Ti++++	sp	oth/un	25°C	0.60M	U I M	1981TMA (12287)	89
						K(TiL+SO ₄)=1.28	
Ti++++	gl	NaCl	25°C	2.0M	U	1979EIa (12288)	90
						K(8TiO=(TiO)8(OH)12+12H)=-1.68	
Ti++++	sp	KNO ₃	25°C	0.10M	U I	K1=14.15 B2=27.88 1971NAe (12289)	91
						K3=13.39	
						K4=13.06	
K1=14.29, K2=13.89, K3=13.58, K4=13.33(I=0.3); K1=14.40, K2=14.02, K3=13.75, K4=13.45(I=0.5); K1=14.70, K2=14.32, K3=14.05, K4=13.74(I=1)							
Ti++++	sol	NaClO ₄	20°C	3.50M	U	1970GTc (12290)	92
						K _{so} (Ti(OH) ₄)=-58.3	
Ti++++	sol	KCl	20°C	3.50M	U	1970GTc (12291)	93
						K(Ti(OH) ₄ (s)=Ti(OH) ₂)=-30.4	
Ti++++	gl	KCl	?	0.10M	U	1970MMk (12292)	94
						*K1=-2.53	
						*K2=-4.58	
						*K3=-8.7	
						*K4=-14.3	
*K5=-21.5, *K6=-30.0. *Kn: Ti ₂ O ₅ + nH ₂ O=Ti ₂ O ₅ (OH) _n + nH							
Ti++++	dis	NaClO ₄	25°C	1.00M	U	1969LSd (12293)	95
						B4=53.3	
Ti++++	gl	oth/un	25°C	3.00M	U	1968PGc (12294)	96
						*B(2,2)=-3.3	
						*K1=-2.25	
Medium: KBr							
Ti++++	sol	NaClO ₄	18°C	0.0	U I	K1=18.0 B2=35.20 1964NLa (12295)	97
						K3=12.5	
						K4=11.0	
I=0 corr. In 0.1 M NaClO ₄ : K(TiO ₂ (s)+2H ₂ O=Ti(OH) ₄)=-5.5							
Ti++++	dis	NaClO ₄	25°C	0.10M	U	1963LCb (12296)	98
						*K2=-1.8	

Ti++++	sp	oth/un	?	var	U		1966BVc (12710) 111
						K(TiOHL+H)=-2	
						K(TiO+HL)=13.85	
Ti++++	kin	oth/un	18°C	0.30M	U		1966LIA (12711) 112
						K(TiO+HL)=12.29	
Ti++++	sp	oth/un	?	var	U		1964JPa (12712) 113
						K(TiOC2O4+H2L)=6.15	
Ti++++	sp	oth/un	?	1.50M	U	M	1963PJa (12713) 114
							K(TiOSO4+H2L=TiLSO4+H2O)=4.2
Ti++++	sp	oth/un	?	?	U		1961VIb (12714) 115
							K(TiO+H2L)=4.31
Ti++++	sp	oth/un	?	?	U	M	1961VIb (12715) 116
							K(TiOH2L+2H+A=TiH2LA)=24.82
H4A=EDTA							
Ti++++	sp	NaClO4	25°C	3.0M	U	M	1960CLa (12716) 117
							K(TiO+H2L)=3.51
							K(TiOF+H2L)=4.22
Ti++++	dis	NaClO4	25°C	1.0M	U		1960GAa (12717) 118
							K(TiO+H2L)=3.9
							K(TiO+2H2L)=6.3
Ti++++	sp	oth/un	rt	var	U	M	1957MOb (12718) 119
							K(TiOA+H2L=TiAL+H2O)=6.37
							K(TiOH2L+A=TiAL+H2O)=20.43
H4A=EDTA							
Ti++++	sp	oth/un	rt	var	U		1954GAa (12719) 120
							K(TiO+H2L)=4.27
Ti++++	sp	oth/un	20°C	var	U		1951BVA (12720) 121
							K(TiO+H2L)=4.05
Medium: HCl							
Ti++++	sp	oth/un	20°C	30%	U		1948STA (12721) 122
							K(TiO+H2L)=5.05
Medium:H2SO4.							
Ti++++	sp	oth/un	20°C	var	U		1937RUa (12722) 123
							K(Ti(OH)6(?) +H2L)=3.95
*****							*****
PO4---		H3L	Phosphate		CAS 7664-38-2 (176)		
Phosphate;							

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	vlt	oth/un	25°C	? U					1982KNb (13345)	124
								pKa(Ti(H ₂ P _O 4)6)=18.29 (?)		
								pKa(Ti(H ₂ P _O 4)3)=10.37 (?)		
Ti++++	EMF	NaCl	25°C	2.0M	U H				1981EKb (13346)	125
								K(TiO+HL)=4.48		
								Ks(TiOHL)= -114.8		
Ti++++	sol	oth/un	20°C	0.50M	U I				1970GSh (13347)	126
								K(TiO+HL)=10.15		
K(TiOHL(H ₂ O) _x (s)+2H=TiO+H ₃ L)=-5.19 or -6.3 depending on form										
At I=0 corr: Ks=-14.2 or -15.3										

SCN-		HL			Thiocyanate		CAS	463-56-9 (106)		
Thiocyanate;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	NaClO ₄	25°C	0.10M	U I	T	K1=0.45		1980MTa (15275)	127
Ti++++	sp	NaClO ₄	25°C	0.50M	U	T			1977TTa (15276)	128
							K(TiO+L)=0.52			
In LiClO ₄ . By kinetic methods, K(TiO+L)=0.49										
Ti++++	vlt	oth/un	25°C	0.40M	C				1977VPb (15277)	129
							K(TiOH+SCN)=1.65			
Method: cyclic chronopotentiometry. Medium: 0.40 M SCN-.										
Ti++++	sp	non-aq	?	100%	U I		K1=2.31	B2=4.33	1973SMd (15278)	130
							B3=6.25			
							B4=8.15			
							B5=10.05			
							B6=11.94			
Medium: acetone. In DMF: K1=3.20, B2=6.25, B3=9.15, B4=11.91, B5=14.61, B6=17.26										
Ti++++	dis	non-aq	?	100%	U				1955DEa (15279)	131
							Kd=1.92			
Kd: K(TiOH+H+4L=TiL ₄ (Me-i-Bu-ketone)+H ₂ O)										
Ti++++	sp	NaClO ₄	?	1.0M	U				1953DEa (15280)	132
							K(TiOH+L)=1.7			

SO ₄ --		H ₂ L			Sulfate		CAS	7664-93-9 (15)		
Sulfate;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

Ti++++ dis oth/un 20°C 8.0M U 1969BMg (16591) 133
K(TiO+L)=2.23
K(TiO+2L)=4.11
B3=11.42

Medium: H₂SO₄

Ti++++ sp NaClO₄ 25°C 4.0M U I 1969VWa (16592) 134
 $K(TiO+L)=2.26$
 $K(TiO+2L)=3.80$

Medium: HClO₄; K₁=2.15(I=3), 2.47(I=5) 2.52(I=0)

Ti++++ sp oth/un 26°C 1.30M U T 1966GSg (16593) 135
 B(TiCr complex)=1.5
 B(TiFe complex)=1.4 complexes not defined

Ti++++ ix oth/un 18°C var U K1=2.40 1962NAe (16594) 136

Ti++++ ix oth/un ? var U 1960BHc (16595) 137
 $K(Ti(OH)_2 + HL) = -0.19$
 $K(Ti(OH)_3 + HL) = 1.05$

Medium: H₂SO₄

CH40 L Methyl alcohol CAS 67-56-1 (597)

Methanol; CH₃.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ EMF alc/w 20°C 100% U 1971GSA (17908) 138
 $K(Ti+2L=Ti(H-1L)2+2H) > 1$
 $K(Te(H-1L)2+H-1L)=12.82$
 $K(3TiL'3+2L'=Ti3L'11)=27.47$
 $K(Ti3L'11+L'=Ti3L'12)=9.72$

Medium: MeOH, 1 M Me4NCl. $K(2Ti3L'12+3L'=3Ti2L'9)=13.84$. L' = H-1L (i.e. CH₃O)

Ti++++ EMF alc/w 20°C 100% U 1964GUa (17909) 139
 $K(Ti_2(H-1L)^{8+H-1L}) = 11.3$

Method: H electrode, Medium: MeOH, 1.0 M Me4NCl

C₂H₂O₄ H₂L Oxalic acid CAS 144-62-7 (24)

Ethanedioic acid: $(\text{COOH})_2$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ vlt NaNO₃ 25°C 0.20M U K1=6.26 B2=10.78 1980MTb (19097) 140
 $K(Ti+2H_L)=5.69$

Ti++++ vlt NaClO₄ 25°C 1.00M U K1=5.18 1979TGa (19098) 141

Ti++++ gl NaCl04 25°C 0.50M U 1977VWa (19099) 142
 $K(Ti(OH)_2 + L) = 7.90$

$$K(Ti(OH)_{2+2L})=13.2$$

Ti++++ ix oth/un ? ? U K1=6.51 B2=11.97 1967MNa (19100) 143
 Metal ion: TiO++

Ti++++ dis NaClO4 20°C 0.10M U 1963STc (19101) 144
 $K(TiO+2L)=10.7$

Ti++++ gl oth/un 25°C 0.03M U 1960GSa (19102) 145
 $K(TiO+L)=9.7$
 $K(TiOL+L)=5.11$

By HgC204: $K(TiOL+L)=4.4$

Ti++++ sp NaClO4 ? 0.02M U 1959BDa (19103) 146
 $K(TiO+L)=6.6$
 $K(TiOL+L)=3.3$

Medium: HClO4

Ti++++ sp oth/un ? ? U K1=6.60 B2=9.90 1959BDd (19104) 147

Ti++++ con oth/un ? ? U K1=2.67 1959BSb (19105) 148

Ti++++ oth oth/un ? ? U K1=1.35 1956KPa (19106) 149

C2H8N2 L Ethylenediamine CAS 107-15-7 (23)
 1,2-Diaminoethane; H2N.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ oth non-aq ? 100% U I M 1962BBa (23236) 150
 $K(TiA4+L)=2.8$

Method: freezing point. Medium: benzene. HA=isopropyl alcohol

In cyclohexane: $K(TiA4+L)=3.7$, $K(TiA4L+TiA4)=-5.3$

C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)
 L-2-Hydroxypropanoic acid; CH3.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ sp NaCl ? 0.30M U 1971ZPa (25555) 151
 $K(Ti(OH)_{3+2L})=8.61$ at pH 4

C4H6O5 H2L Malic acid CAS 617-48-1 (393)
 2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ sp oth/un 20°C 0.30M U M 1970ZHa (30740) 152
 $K(Ti(OH)_{3+L})=6.74$

Ti++++ dis oth/un 25°C 0.10M U 1968Gpc (30741) 153
 $K(TiO+L)=6.81$

C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)
 L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	ix	NaClO4	?	2.00M	U				1973ZGc (31374) 154	
								$K(TiO+HL)=2.50$		
								$K(TiOHL+HL)=2.12$		

Ti++++ dis NaClO4 20°C 0.10M U 1963STc (31375) 155
 $K(TiO+2L)=9.7$

C5H5N L Pyridine CAS 110-86-1 (31)
 Pyridine, Azine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	oth none	25°C	0.0	U	HM				1958ERb (36685) 156	
DG(TiF4(s)+2L(g)=TiF4L2(s))=-24.2 kJ mol-1, DH=-46, DS=-75. Data also for									TiCl4, TiBr4 and TiI4	

C5H8O2 HL Acetylacetone CAS 123-54-6 (164)
 Pentane-2,4-dione; CH3.CO.CH2.CO.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	EMF alc/w	20°C	100%	U	M				1971GSa (38104) 157	
								$K(TiA2L+A)=11.25$		
								$K(2TiA3L+3A=Ti2A9+2L)=8.80$		
								$K(TiAL2+A)=11.90$		
								$K(TiAL2+TiA2L2=Ti2A3L4)=1.95$		

Medium: MeOH. HA=CH3OH
 $K(TiA2L2+A=TiA3L+L)=1.35$, $K(2TiA2L2+5A=Ti2A9+4L)=11.50$

C6H6N2O2 HL Cupferron CAS 135-20-6 (637)
 N-Nitrosophenylhydroxylamine; C6H5.N(OH).NO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	dis NaClO4	10°C	3.00M	U				$K1=10.39$ $B2=20.46$	1969AIc (43426) 158	
								$K3=9.75$		
								$K4=9.43$		

C6H6O HL Phenol CAS 108-95-2 (457)
 Hydroxybenzene, phenol; C6H5.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ sp oth/un 25°C 0.02M U M 1981VMa (43546) 159
K(TiO₂+2HL=TiOAL₂+HA+H)=-1.66

H2A=oxalic acid

C6H6O₂ H2L Catechol CAS 120-80-9 (534)
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	oth/un	?	70%	U	I			1973SPd (43845)	160
								K(TiO+HL)=1.85		
Medium: H ₂ SO ₄ . In 95% H ₂ SO ₄ , K(TiO+HL)=2.56, K(TiO+2HL)=4.75										
Ti++++	EMF	R4N.X	20°C	1.00M	U	M			1971GSa (43846)	161
								K(TiA ₂ +H2L+2A=TiA ₂ L+2HA)=27.35		
								K(2TiA ₂ L+A)=13.80		
Medium: MeOH, 1.0 M Me4NCl. HA=MeOH										
K(Ti ₂ A ₅ L ₂ +4H2L+3A=2TiL ₃ +8HA)=34.11										

Ti++++ sp oth/un ? 1.20M U I M B2=50.1 1970PLd (43847) 162
Medium: 1.2 M H2L; Medium: 0.5 M HCl; B(TiAL₂)=59.4;
Medium: unknown; K(Ti(OH)₂+A+L)=29.3, H2A=oxalic acid

Ti++++ sp alc/w 20°C 100% U I 1966SCe (43848) 163
K(TiO+HL)=6.1
Medium: MeOH. K=4.2(0%), 4.7(25%), 5.1(50%), 5.5(75%). In EtOH/H₂O:
K=4.56(25%), 4.98(50%), 5.17(75%), 6.15(100%)

Ti++++ sp mixed 20°C 100% U I 1966SCe (43849) 164
K(TiO+HL)=6.39
Medium: propanol. K=4.36(25%), 4.64(50%), 5.07(75%). In 2-propanol:
K=4.28(25%), 4.73(50%), 4.9(75%), 6.75(100%)

Ti++++ sp NaClO₄ 20°C 0.10M U 1963SOa (43850) 165
K(TiO+2H2L=TiO(HL)₂+2H)=-1.9
K(TiO(HL)₂+H2L=TiL₃+2H)=-4.7
K(TiO+2H+3L)=61.6

Ti++++ sp oth/un 22°C 0.50M U 1961SKa (43851) 166
K(TiO+L)=22.5
K(TiOL+L)=15.9

Medium: acetate buffer. At I=0.05 M: K(TiO+L)=18.8, K(TiOL+L)=17.7

C6H6O₃ H3L Pyrogallol CAS 87-66-1 (696)
1,2,3-Trihydroxybenzene; C₆H₃(OH)₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	oth/un	23°C	96%	U				1981BMe (43985)	167

$$K(Ti+H_3L)=2.13$$

Medium: 96% H₂S04

Ti++++	sp	oth/un	?	95%	U	1973SPd (43986) 168
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$$K(TiO+H_3L)=1.49$$

Medium: H₂S04

Ti++++	sp	alc/w	20°C	100%	U I	1966SCe (43987) 169
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$$K(TiO+H_2L)=6.2$$

Medium: 100% MeOH. K=4.5(0%), 5.1(25%), 5.4(50%), 5.8(75%). 18-20 C
In EtOH: K=4.8(25%), 5.7(50%), 6.0(75%), 6.4(100%)

Ti++++	sp	mixed	20°C	100%	U I	1966SCe (43988) 170
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$$K(TiO+H_2L)=6.6$$

Medium: 100% propanol. K=4.6(25%), 4.8(50%), 5.6(75%). 18-22 C
In 2-propanol: K=4.7(25%), 4.7(50%), 5.6(75%), 6.8(100%)

C6H6O8S2 H4L Tiron CAS 149-45-1 (104)
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)₂.C6H₂(SO₃H)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++	sp	NaClO ₄	20°C	0.10M	U	1963SOa (44502) 171
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$$K(TiO+2H_2L=TiO(HL)2+2H)=-0.3$$

$$K(TiO(HL)2+H2L=TiL3+2H)=-2.9$$

$$K(TiO+2H+3L=TiL3)=57.6$$

C6H7N L Aniline CAS 62-53-3 (583)
Aminobenzene, aniline; C₆H₅.NH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++	dis	oth/un	?	?	U	M	1972BAC (44882) 172
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$$K(TiA3+2HL)=0.27$$

H2A=pyrocatechol. pH 3-4

C6H8O6 H2L Ascorbic acid CAS 50-81-7 (285)
Ascorbic acid (Vitamin C);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++	sp	NaClO ₄	20°C	0.10M	U	1963SOb (45662) 173
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$$K(TiO+2HL)=24.8$$

$$K(TiO+H2L)=3.1$$

$$K(TiO+2H2L)=6.25$$

$$K(TiO+2H+3HL)=39.3$$

C6H8O7 H3L Citric acid CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH₂.CH(OH)(COOH).CH₂COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	vlt	NaClO4	25°C	0.50M	U				1978TGa (46280)	174
								$K(Ti+H3L)=2.98$		
Ti++++	ix	NaClO4	?	2.00M	U				1973ZGc (46281)	175
								$K(TiO+H2L)=2.91$		
								$K(TiOH2L+H2L)=2.49$		
Ti++++	sp	NaCl	?	0.30M	U				1971ZHa (46282)	176
								$K(Ti(OH)2+L+HL)=16.28$		

C6H9N06		H3L	NTA					CAS 139-13-9	(191)	
Nitrilotriethanoic acid; N(CH2.COOH)3										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	dis	NaClO4	20°C	0.10M	U				1963STc (47053)	177
								$K(TiO+L)=12.3$		

C6H1207		HL	Gluconic acid					CAS 526-95-4	(904)	
D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH2(CHOH)4.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	oth/un	25°C	dil	U	I			1969MCA (49766)	178
								$K_{eff}(TiO+L)=4.66$	pH 3	
								$K_{eff}(TiO+L)=4.43$	pH 6	

C7H6O3		H2L	Salicylic acid					CAS 69-72-7	(14)	
2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	cal	NaClO4	25°C	0.50M	U				1981VVb (54310)	179
								$DH(TiO+L)=-51.4$	kJ mol-1	
for 0.1 M NaClO4 $DH=-46.4$ kJ/mol (25 C); -47.7 kJ/mol (15 C); -46.9 kJ/mol (30 C)										
Ti++++	sp	oth/un	20°C	0.10M	U	I			1962BVb (54311)	180
								$K(TiO+L)=15.66$		
								$K(TiO+2L)=24.36$		
In 2-3 M KNO3: $K(TiO+2L)=24.63$										
Ti++++	sp	oth/un	35°C	?	U			$K1=6.09$	1959DGd (54312)	181

C7H6O4		H3L	Resorcylic acid					CAS 89-86-1	(876)	
2,4-Dihydroxybenzoic acid, b-Resorcylic acid; C6H3(OH)2.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

 Ti++++ sp oth/un ? 0.10M U 1972AKc (54543) 182
 $K(Ti(OH)_3+2H_2L)=6.48$

 Ti++++ sp oth/un ? ? U 1970AKb (54544) 183
 $K(TiO+2H_2L)=6.01$

 C7H604 H3L Protocatechuic CAS 99-50-3 (875)
 3,4-Dihydroxybenzoic acid; C6H3(OH)2.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ti++++ sp NaClO4 20°C 0.10M U 1963SOa (54703) 184
 $K(TiO+2H_3L=TiO(H_2L)_2+2H)=-1.35$
 $K(TiO(H_2L)_2+H_3L=TiL_3+5H)=-3.9$
 $K(TiO+2H+3L)=58.6$

 C7H605 H4L CAS 610-02-6 (3725)
 2,3,4-Trihydroxybenzoic acid; (HO)3.C6H2.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ti++++ sp mixed ? 70% U I 1972SPb (54722) 185
 $K(TiO+H_4L)=2.82$
 Medium: 70% H2SO4. In 95% H2SO4, K=2.44

 Ti++++ sp oth/un 18°C 0.10M U 1971AKe (54723) 186
 $K(Ti(OH)_3+H_2L)=3.54$
 $K(Ti(OH)_3+2HL)=7.61$
 $K(Ti(OH)_3+H_2L)$: dil HCl; $K(Ti(OH)_3+2HL)$: pH=2.8-7.5

 C7H605 H4L Gallic acid CAS 149-91-7 (446)
 3,4,5-Trihydroxybenzoic acid; C6H2(OH)3.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ti++++ sp mixed ? 95% U 1972SPb (54766) 187
 $K(TiO+H_4L)=2.35$
 Medium: 95% H2SO4

 C7H606S H3L CAS 585-42-2 (6136)
 2-Hydroxy-4-sulphobenzoic acid, 4-sulfosalicylic acid; HO.C6H3(COOH)(HSO3)

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ti++++ vlt NaClO4 21°C 0.10M U I 1977UBa (54805) 188
 $K(TiO+HL=Ti(OH)L)=4.1$
 In 0.6 M NaClO4: $K(Ti(OH)L_2+HL=TiL_3+H_2O)=1.7$

 C7H606S H3L CAS 5965-83-3 (399)

5-Sulfosalicylic acid, 2-Hydroxy-5-sulfobenzoic; H03S.C6H3(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	KCl	20°C	1.0M	U	M			1973VGa (55056)	189
								K(TiA2+HL=TiA2L+H)=3.44		
								K(TiH2A+H2L=(TiH2A)H2L)=11.40		

H2A=4-(2-pyridylazo)-resorcinol

Ti++++	sp	oth/un	20°C	0.10M	U				1963SOa (55057)	190
								K(TiO+2H+3L)=42.2		
								K(TiO+2HL)=5.4		
								K(TiO+HL)=3.1		

C7H7N02 H2L Salicylaldoxime CAS 94-67-7 (1486)

2-Hydroxybenzaldehyde oxime; HO.C6H4.CH:N.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	gl	KCl	25°C	0.10M	U	I		K1=16.30 B2=31.15	1968MDe (55314)	191
								K1=18.5(I=0),18.29(I=0.01),17.74(I=0.025),17.35(I=0.05),16.86(I=0.075);		
								K2=17.2(I=0),16.88(I=0.01),16.62(I=0.025),16.07(I=0.05),15.66(I=0.075)		

C7H8O2 HL CAS 150-19-6 (4412)

5-Methoxyphenol; HO.C6H4.OCH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	mixed	?	90%	U				1973SPd (56097)	192

K(TiO+HL)=1.92

Medium: 90% H2SO4

C8H5O2F3S HL TTA CAS 326-91-0 (165)

4,4,4-Trifluoro-1-(2-thienyl)butane-1,3-dione; F3C.CO.CH2.CO.C4H3S

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	dis	NaClO4	25°C	1.0M	U			K1=7.87 B2=15.52	1969LSe (58686)	193
								K3=7.45		
								K4=7.23		

C9H6O4 H2L Esculetin CAS 305-01-1 (3853)

6,7-Dihydroxycoumarin;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	alc/w	20°C	20%	U				1964JSb (63954)	194

K(?)=8.8

Medium: 20% EtOH, 0.4 M NaClO4

C9H7NO	HL	Oxine	CAS 148-24-3 (504)	
8-Hydroxyquinoline (8-quinolinol);				
<hr/>				
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference ExptNo
Ti++++	dis	NaClO4	20°C 1.0M U	K1=13.22 B2=25.94 1967SLa (64359) 195 K3=12.26 K4=11.0
<hr/>				
C9H7N3O2S	H2L	TAR	CAS 2246-46-0 (707)	
4-(2'-Thiazolylazo)-resorcinol; C3H2NS.N:N.C6H3(OH)2				
<hr/>				
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference ExptNo
Ti++++	sp	alc/w	25°C 50% U	1967NPb (64729) 196 K(TiO+HL)=13
Medium: 50% MeOH, 0.1 M NaClO4				
<hr/>				
C9H10O2	L	4-Tolyl-acetate	CAS 140-39-6 (3857)	
Ethanoic acid 4-methylphenyl ester; CH3.CO.O.C6H4.CH3				
<hr/>				
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference ExptNo
Ti++++	sp	non-aq	60°C 100% U T H	1966GSd (65382) 197 K(TiCl4+L)=1.93
Medium: 1,2-dichloroethane. K=2.60(25 C). DH=-37.6 kJ mol-1, DS=-71 J K-1m-1				
<hr/>				
C10H6O8Br2S2	H4L		CAS 58425-38-0 (2003)	
2,7-Dibromo-1,8-dihydroxy-naphthalene-3,6-disulfonic acid;				
<hr/>				
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference ExptNo
Ti++++	sp	NaClO4	20°C 0.10M U	1975MDa (68535) 198 B(Ti(OH)2(HL)2)=10.98
<hr/>				
C10H6O8Cl2S2	H4L		CAS 6155-33-5 (4761)	
2,7-Dichlorochromotropic acid;				
<hr/>				
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference ExptNo
Ti++++	sp	NaClO4	20°C 0.50M U	19700Mb (68537) 199 K(TiO+2HL)=7.38
<hr/>				
C10H8O2	H2L		CAS 569-42-6 (4699)	
1,8-Dihydroxynaphthalene;				
<hr/>				
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference ExptNo
Ti++++	sp	oth/un	25°C 0.10M U	1968BNc (69754) 200 K(Ti(OH)2+2H2L=Ti(OH)2(HL)2+2H)=15.34

$$K(Ti(OH)_3 + 3H_2L \rightleftharpoons Ti(OH)(HL)_3 + H) = 28.0$$

 Ti++++ sp non-aq ? 100% U I 1966SCa (69755) 201
 $K(TiOH + HL \rightleftharpoons TiOHL + H) = 7.0$: acetone
 $K = 6.38$: in dimethylformamide
 $K = 6.08$: in dioxan
 $K = 6.33$: in ethanol
 K=6.10 in methanol, K=6.52 in propanol, K=6.85 in 2-propanol

C10H804 H2L 4-Me-Esculetin CAS 529-84-0 (3890)
 4-Methyl-6,7-dihydroxycoumarin

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ti++++ sp alc/w 20°C 20% U 1964JSb (69791) 202
 $B_3 = 10.7$

Medium: 20% EtOH, 0.4 M NaClO4

 C10H804 H2L 4-Me-daphnetin CAS 2107-77-9 (6317)
 7,8-Dihydroxy-4-methylcoumarin;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ti++++ sp alc/w ? 50% U 1976SSe (69792) 203
 $K(TiO + HL) = 8.37$
 $K(TiOHL + HL) = 7.33$
 $K(TiO(HL)_2 + HL) = 5.70$

 C10H805S H3L DHNSA (877)
 2,3-Dihydroxynaphthalene-6-sulfonic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ti++++ gl oth/un 20°C 0.10M U 1963SOb (69865) 204
 $K(TiO + 2L) = 38.1$
 $K(TiO + 3L) = 54.7$
 $K(TiO + 2H + 3L) = 56.5$
 $K(TiO + 2H_2L \rightleftharpoons TiO(HL) + 2H) = -0.69$

 C10H808S2 H4L Chromotropic ac CAS 148-25-4 (1875)
 1,8-Dihydroxynaphthalene-3,6-disulfonic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ti++++ sp NaClO4 25°C 0.10M U 1975BUb (69972) 205
 $K_{eff} = 7.72$ at pH 5.32
 $B_{2eff} = 11.85$ at pH 5.32
 $B(2,2)_{eff} = 16.11$ at pH 5.32

 Ti++++ sp oth/un 25°C 0.10M U 1968BNc (69973) 206

$$K(Ti(OH)2 + 2H2L \rightleftharpoons Ti(OH)2(HL)2 + 2H) = 14.36$$

$$K(Ti(OH)3 + 3H2L \rightleftharpoons Ti(OH)(HL)3 + H) = 25.92$$

Ti++++ sp mixed 20°C 100% U I 1966CSb (69974) 207

$$K(TiO+HL) = 7.34$$

$$K(TiO+2HL) = 12.17$$

Medium: DMF/H₂O, TiO 0.005 M: K1=4.80(0% DMF), 5.60(25%), 6.26(50%-74%);

With 0.0025 TiO:K1=5.05(0%), 5.57(25%), 6.12(50%), 6.57(74%), 7.38(100%)

Ti++++ sp NaClO₄ 20°C 0.10M U 1963SOa (69975) 208

$$K(TiO+2L) = 40.5$$

$$K(TiO+3L) = 56.4$$

$$K(TiO+2H+3L \rightleftharpoons TiL3) = 60.5$$

$$K(TiOL2+2H \rightleftharpoons TiO(HL)2) = 4.4$$

Ti++++ sp oth/un 20°C 0.10M U B2=6.18 1959SOc (69976) 209
B3=10.59

Ti++++ sp oth/un 20°C 0.10M U 1957BPC (69977) 210
 $K(Ti+H2L) = 3.99$

C10H12O2 HL CAS 1946-74-3 (202)

3-Isopropyltropolone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp alc/w 25°C 50% U B2=21.17 1961HSa (71609) 211
B3=28.95

Medium: 50% EtOH, 0.01 M

C10H16N2O8 H4L EDTA CAS 60-00-4 (120)

1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ vlt KN03 25°C 0.20M U K1=22.61 1986ZFa (74228) 212

Ti++++ gl KN03 25°C 0.10M C T H K1=18.47 1985HWc (74229) 213

Data for 5-35 C. Metal is TiO++. Method: Hg and glass electrodes,
competition with Hg++. DH(K1)=-31.3 kJ mol⁻¹, DS(K1)=250 J K⁻¹ mol⁻¹.

Ti++++ sp NaClO₄ 20°C 1.0M U 1971KNa (74230) 214

$$K(TiO+L) = 18.15$$

$$K(TiO+HL) = 12.08$$

Ti++++ dis NaClO₄ 20°C 0.10M U T 1963STc (74231) 215

$$K(TiO+L) = 17.5$$

C11H9N3O2 H2L PAR CAS 1141-59-9 (636)

4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	oth/un	20°C	1.00M	U				1974LKd (77587)	216
K(Ti(OH)2+HL+A=Ti(OH)2HLA)=2.72(HA=ethanoic acid); 21.12(HA=chloroethanoic acid; 20.08(HA=dichloroethanoic acid); 19.74(HA=trichloroethanoic acid)										
Ti++++	sp	oth/un	20°C	1.00M	U				1974LKd (77588)	217
B(TiL2A2)=47.43(HA=ethanoic acid); 46.68(HA=chloroethanoic acid); 45.81(HA=dichloroethanoic acid); 45.45(HA=trichloroethanoic acid)										
Ti++++	sp	oth/un	?	?	U	M			1967SHa (77589)	218
HA=ethanoic acid								K(TiOA+HL)=13.25		
C13H11N02		HL					CAS	304-88-1	(181)	
N-Phenylbenzohydroxamic acid; C6H5.CO.N(C6H5).OH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	dis	oth/un	20°C	1.0M	U		K1=11.77	B2=23.40	1970LSd (85181)	219
							K3=11.49			
							K4=11.35			
C13H14N03P		H2L					CAS	19316-85-7	(1466)	
2-Hydroxyphenyl-N-phenylaminomethylphosphinic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	gl	NaClO4	20°C	0.10M	U		K1=7.20		1985SIb (85566)	220
C13H14N305P		H2L					CAS	80767-75-5	(1467)	
2-Hydroxy-4-nitrophenyl-N-(2-pyridylmethyl)aminemethylphosphinic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	gl	NaClO4	20°C	0.10M	U		K1=7.20		1985SIb (85644)	221
C13H14N305P		H2L					CAS	80767-76-6	(1468)	
2-Hydroxy-4-nitrophenyl-N-(3-pyridylmethyl)aminemethylphosphinic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	gl	NaClO4	20°C	0.10M	U		K1=8.10		1985SIb (85657)	222
C13H15N203P		H2L					CAS	80767-72-2	(1460)	
2-Hydroxyphenyl-(N-2-pyridylmethylamino)methylphosphinic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

Ti++++ gl NaClO₄ 20°C 0.10M U K1=11.80 1985SIa (85784) 223

C13H15N203P H2L CAS 80767-73-3 (1461)
 2-Hydroxyphenyl-(N-3-pyridylmethylamino)methylphosphinic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ gl NaClO₄ 20°C 0.10M U K1=11.70 1985SIa (85797) 224

C13H15N203P H2L CAS 80767-74-4 (1462)
 2-Hydroxyphenyl-(N-4-pyridylmethylamino)methylphosphinic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ gl NaClO₄ 20°C 0.10M U K1=11.75 1985SIa (85810) 225

C13H15N204P H3L CAS 80767-78-8 (1463)
 2-Hydroxyphenyl-(N-2-pyridylmethylamino)methylphosphonic acid;
 C₆H₄(OH)CH(PO₃H₂).NH.CH₂.C₅H₄N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ gl NaClO₄ 20°C 0.10M U K1=16.80 1985SIa (85823) 226

C13H15N204P H3L CAS 85946-85-6 (1464)
 2-Hydroxyphenyl-(N-3-pyridylmethylamino)methylphosphonic acid;
 C₆H₄(OH)CH(PO₃H₂).NH.CH₂.C₅H₄N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ gl NaClO₄ 20°C 0.10M U K1=16.90 1985SIa (85836) 227

C13H15N204P H3L CAS 85946-86-7 (1465)
 2-Hydroxyphenyl-(N-4-pyridylmethylamino)methylphosphonic acid;
 C₆H₄(OH)CH(PO₃H₂).NH.CH₂.C₅H₄N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ gl NaClO₄ 20°C 0.10M U K1=17.15 1985SIa (85849) 228

C13H20N04P H3L (1471)
 2-Hydroxyphenyl-N-(cyclohexylamino)methylphosphonic acid;
 C₆H₄(OH)CH(PO₃H₂).NH.C₆H₁₁

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ gl NaClO₄ 20°C 0.10M U K1=15.05 1985SIb (86094) 229

C14H12O3 HL Benzilic acid CAS 76-93-7 (710)
 Diphenylglycolic acid, (benzilic acid); (C₆H₅)₂C(OH).COOH

competition with Hg++. DH(K1)=-104.8 kJ mol-1, DS(K1)=89.7 J K-1 mol-1.

Ti++++ sp oth/un 20°C dil U K1=23.38 1970Kaf (89413) 238
K(Ti+HL)=14.51

C14H24N2010 EGTA CAS 67-42-5 (349)
Ethleneglycol-0,O'-bis(2-aminoethyl ether)-N,N,N',N'-tetraethanoic acid; H4L

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ gl KN03 25°C 0.10M C T H K1=20.08 1985HWc (89949) 239

Data for 5-35 C. Metal is Ti0++. Method: Hg and glass electrodes,
competition with Hg++. DH(K1)=-80.3 kJ mol-1, DS(K1)=115 J K-1 mol-1.

C15H1007 H5L Quercetin CAS 117-39-5 (5101)
3,5,7-Trihydroxy-2-(3',4'-dihydroxyphenyl)-1-benzopyran-4-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp non-aq 25°C 100% U 1969DSc (91025) 240
K(?)=4.30

Medium: BuOH

C15H13N02 HL CAS 7369-44-0 (4066)

N-3-Diphenylpropenohydroxamic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ dis oth/un 20°C 1.0M U K1=13.3 B2=26.40 1970LSd (91645) 241
K3=12.9
K4=12.7

C16H11N03 HL HPBI CAS 41836-94-6 (7740)

3-Phenyl-4-benzoyl-5-isoxazolone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ dis non-aq 30°C 100% U 1999SPa (92687) 242
Kd=1.91

Kd: Ti0+2HL(org)=TiOL2(org)+2H.

Method: Solvent extraction, H2O/xylene.

C16H14O5 H3L CAS 966-64-3 (5143)

2,3,7-Trihydroxy-9-propylfluorone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp alc/w ? 4% U 1967NBa (93591) 243

K(TiOH+2H2L)=32.72

K(Ti(OH)2+2H2L)=25.70

Medium: 4% EtOH, 0.1 M

C17H14N202 L CAS 4551-69-3 (698)

4-Benzoyl-3-methyl-1-phenyl-2-pyrazolin-5-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	dis	oth/un	20°C	1.0M	U			K1=8.11 B2=16.06 K3=7.76 K4=7.58	1969LSb (95902)	244

C17H17N03 HL CAS 58434-59-6 (1213)

2'-Hydroxy-4-methoxy-5'-methylbenzylidene acetophenone oxime

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	oth/un	30°C	8.00M	U	M			1980GKa (96192)	245
								K(TiO(SCN)+L)=2.29 K(TiO(SCN)L+L)=1.80		

C18H26N206P2 H4L CAS 53431-86-0 (5266)

Ethylenebis(imino(2-hydroxyphenyl)methylene(methyl)phosphinic acid);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	EMF	oth/un	?	?	U				1970DMc (97675)	246
								K(TiO+2H2L)=8.46 B(TiO+2L)>15		

C19H12O6 H4L Salicylfluorone (5269)

9-(2-Hydroxyphenyl)-2,3,7-trihydroxy-6-fluorone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	alc/w	?	4%	U				1967NBa (98996)	247
								K(TiOH+2H2L)=26.19 K(Ti(OH)2+2H2L)=26.19		

Medium: 4% EtOH, 0.1 M

C19H15N08 H4L Alizarin Comp. CAS 3952-78-1 (671)

(3,4-Dihydroxy-2-anthraquinonyl-methyl)iminodiethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	oth/un	RT	dil	C				1982EDa (99141)	248
								K1eff=4.2 B2eff=8.6		

Medium: borax buffer, pH 10.

C21H19N308S H4L MeNaphtholOrange (4151)

N-(1'-Hydroxy-4'-(4'''-sulfophenylazo)-2'-naphthylmethyl)-iminodiethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
Ti++++	sp	oth/un	?	?	U					1969TKc (102575)	250
									$K(Ti(OH)_2 + HL) = 5.64$		

C23H24N4O2		L	Trichachnine	CAS	1251-85-0	(2606)					
4,4'-Diantipyrylmethane,											
4,4'-phenylmethylen-bis-(1,2-dihydro-1,5-dimethyl-2-phenylpyrazol-3-one											

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	oth/un	20°C	1.0M	U				1962BTc (102679)	251
										B3=7.89

C27H30O16	H4L	Rutin		CAS	153-18-4	(4169)				
3,3',4',5,7-Pentahydroxyflavone-3-beta-rutinoside;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	alc/w	RT	50%	C				2000KMa (104509)	252
Medium: 50% EtOH/H2O. K(TiO(ox)2+2HL=TiO(Ox)2L2+2H)=10.80 at pH 6.50.										

C31H32N2O13S	H6L	Xylenol orange	CAS	63721-85-5	(432)					
5,5'-Bis-N,N-bis(carboxymethyl)aminomethyl-4'-hydroxy-3,3'-dimethylfuchsone-2"-sulfonic acid:										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	NaNO ₃	20?°C	0.20M	U				1963BGa (105500)	253
							B(Ti ₂ L ₂)=57.8			
Ti++++	sp	NaClO ₄	25°C	0.50M	U	I			19630Ta (105501)	254
							K(TiO+H ₆ L=TiOH ₅ L+H)=3.46			
Ti	0.05 M HClO ₄ :						K(TiO+H ₆ L+H ₂ O ₂ -TiH ₆ LH ₂ O ₂)=37.68			

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EXPLANATORY NOTES

DATA Flags are :-

- T Data at other TEMPERATURES
- I Data with various BACKGROUNDS
- H Data for THERMOCHEMICAL quantities
- M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC

END