

SC-Database

Software version = 5.81 Data version = 4.62
Experiment list contains 1625 experiments for
(no ligands specified)
3 metals : U++, U++, UO₂++
(no references specified)
(no experimental details specified)

e- HL Electron (442)
Electron;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U++ oth none 25°C 0.0 U 1952Lab (992) 1
K(U+3e=U(s))=-91(-1800 mV)

From thermodynamic data

Br- HL Bromide CAS 10035-10-6 (19)
Bromide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U++ sp oth/un 25°C 0.0 U K1=-3.95 1965SMd (2371) 2

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U++ sp oth/un 25°C 0.0 U K1=-2.89 1965SMd (5905) 3
U++ sp NaCl 25°C var U K1=-2.85 1962SMa (5906) 4
Medium:LiCl var

P04--- H3L Phosphate CAS 7664-38-2 (176)
Phosphate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U++ oth none ? 0.0 U 1969MOc (13352) 5
K(U+H2L)=2.40
K(U+2H2L)=3.78
K(U+3H2L)=5.65

Methods: solubility, ion exchange, distribution, EMF

C2H4O3 HL Glycolic acid CAS 79-14-1 (33)
2-Hydroxyethanoic acid; HO.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ oth none ? 0.00 U K1=3.55 B2=6.10 1969MOc (20642) 6
Data extrapolated from literature

C4H4N2 L Pyrimidine CAS 289-95-2 (4247)
1,3-Diazine, pyrimidine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ nmr non-aq 25°C 100% C H 2004MBa (28779) 7
K(UA3+L)=0.8
K'(UB3+L)=3.56

1H nmr in d-toluene. DH(K)=-39 kJ mol-1, DS=-118 J K-1 mol-1; DH(K')=-70, DS=-168. A: t-butyl-cyclopentadiene; B: trimethylsilyl-cyclopentadiene.

C4H8O2 HL Isobutyric acid CAS 79-31-2 (573)
2-Methylpropanoic acid; CH3.CH(CH3).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ oth none ? 0.00 M K1=3.55 B2=6.02 1969MOc (33253) 8
B3=7.20

Data from survey of literature data

C6H7N L beta-Picoline CAS 108-99-6 (324)
3-Methylpyridine; C5H4N.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ nmr non-aq 25°C 100% C H 2004MBa (44708) 9
K(UA3+L)=1.72
K'(UB3+L)=4.7

1H nmr in d-toluene. DH(K)=-46 kJ mol-1, DS=-123 J K-1 mol-1; DH(K')=-83, DS=-193. A: t-butyl-cyclopentadiene; B: trimethylsilyl-cyclopentadiene.

C6H9NO6 H3L NTA CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH2.COOH)3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ oth none ? 0.00 M K1=12.4 1969MOc (47072) 10
Constant obtained from survey of literature data

C7H9N L 3,5-Lutidine (323)
3,5-Dimethylpyridine; C5H3N.(CH3)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ nmr non-aq 25°C 100% C H 2004MBa (56290) 11
K(UA3+L)=1.74

					K'(UB3+L)=4.9
1H nmr in d-toluene. DH(K)=-44 kJ mol-1, DS=-116 J K-1 mol-1; DH(K')=-85, DS=-194. A: t-butyl-cyclopentadiene; B: trimethylsilyl-cyclopentadiene.					

C10H8N2	L	2,2'-Bipyridyl	CAS 366-18-7	(25)	
2,2'-Bipyridine; (C5H4N)2					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo
U+++	nmr	non-aq	21°C	100%	U HM K(UI3+L)=1.04 K(UI3L+L)=2.11 2001RNa (69658) 12
Medium: pyridine. At -40 C K(UI3L2+L)=0.20. DH(UI3+L)=-21 kJ mol-1, DS=-52 J K-1 mol-1; DH(UI3L+L)=-9, DS=8; DH(UI3L2+L)=-12, DS=-47.					

C14H23N3O10	H5L	DTPA	CAS 67-43-6	(238)	
Diethylenetriamine-pentaethanoic acid; HOOC.CH2.N(CH2.CH2.N(CH2.COOH)2)2					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo
U+++	oth	oth/un	?	0.0	U K1=25.1 1969MOc (89424) 13
From survey of literature data					

C18H18N4	L		CAS 16858-01-8	(1528)	
Tris(2-pyridylmethyl)amine; (C5H4NCH2)3N					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo
U+++	nmr	non-aq	RT	100%	U K2=1.1 2000WMa (97274) 14
Method: 1H nmr. Medium: C5D5N					

e-	HL	Electron			(442)
Electron;					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo
U+++	EMF	NaClO4	25°C	1.0M	C TI 1990CVc (993) 15
Method: cyclic voltammetry. K(U +e=U(III))=-10.63 (-630 mV).					
Data for 5-55 C. Data extrap to 1.0 M (HClO4) using SIT.					

U+++	cal	NaClO4	25°C	0.50M	U H 1958FOa (994) 16
Medium: HClO4. DH(U+e=U(III))=99.0 kJ mol-1. DH(U(III)+3e=U(s))=512					

U+++	EMF	KCl	25°C	1.0M	U 1958HMa (995) 17
K(U+e=U(III))=-10.70(-633 mV)					

U+++	oth	none	25°C	0.0	U 1952LAb (996) 18
K(U+e=U(III))=-10.3(-610 mV)					
From thermodynamic data					

U++++ vlt NaClO₄ 25°C 1.0M U I 1949KHa (997) 19
 K(U=e=U(III))=-10.67(-631 mV)
 Medium: HClO₄. In 1 M HCl: K=-10.82(-640 mV)

U++++ EMF none 25°C 0.0 U 1949KNa (998) 20
 K=9.3(0.55 V)
 Metal ion: U(V). K(UO₂+4H+e=U(IV)+2H₂O)

Br- HL Bromide CAS 10035-10-6 (19)
 Bromide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U++++	nmr	NaClO ₄	?	2.0M	U			K1=0.30	1964PCa	(2372) 21

Method: NMR; medium: HClO₄.

U++++ EMF NaClO₄ 20°C 1.60M U K1=0.18 1954ALa (2373) 22

CO₃-- H2L Carbonate CAS 465-79-6 (268)
 Carbonate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U++++	EMF	NaClO ₄	25°C	0.0	C	TI			1990CVc	(3406) 23

B3=6.6
 Method: cyclic voltammetry.

U++++ sp NaClO₄ 25°C 0.00 U I 1989BGa (3407) 24
 K5=-1.12
 Value extrapolated to infinite dilution

U++++ cal KCl 25°C 3.0M C 1984GSe (3408) 25
 DH(U+5CO₃)=-20 kJ mol⁻¹, DS(U+5CO₃)=672

Cl- HL Chloride CAS 7647-01-0 (50)
 Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U++++	EMF	none	25°C	0.0	U	T	H	K1=3	1980LTb	(5907) 26

100 C: K1=3; 200 C: K1=5. Evaluated data

U++++ ix NaClO₄ 25°C 0.20M C K1=0.28 1978SGg (5908) 27
 Method: polarography. Medium: 0.20 M HClO₄.

U++++ EMF KCl 70°C 0.56M U T K1=1.35 1977NNb (5909) 28
 Temps from 70 to 150 Degrees. At 150 C: K1=2.45. Range of I: 0.56 to 4.0.

U++++ ix NaClO₄ 25°C 2.06M U I K1=-0.5 1974BUa (5910) 29
 Medium: HClO₄. K1=-0.1(I=3), 0 approx(I=3.93)

U+++ ix NaClO4 48°C 4.0M U T K1=-0.3 B2=-0.96 1974Buc (5911) 30
Medium: HClO4. K1=0.23(16 C); K1=0.02, B2=-1.1(25 C); K1=0, B2=-0.3(31 C);
K1=-0.2, B2=-0.7(41 C); K1=-0.4, B2=-0.8(56 C)

U+++ nmr NaClO4 ? 2.0M U K1=0.78 1964PCa (5912) 31
Method: NMR, medium: HClO4

U+++ EMF oth/un 25°C 0.0 U I K1=0.8 1961SOe (5913) 32

U+++ dis NaClO4 10°C 3.0M U T K1=0.52 1955DWa (5914) 33
Medium: 2 M NaClO4, M HClO4. At 25 C: K1=0.26 (or K1=0.08, K2=-0.002)
40C: K1=0.18 (or K1=-0.04, K2=-0.06)

U+++ EMF NaClO4 20°C 1.60M U K1=0.30 1954ALa (5915) 34

U+++ sp NaClO4 25°C 0.50M U I K1=-0.20 1950KNa (5916) 35
At I=0 corr.: K1=0.85

ClO2- HL Chlorite CAS 13898-47-0 (6143)

Chlorite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ kin NaClO4 25°C 2.0M U 1972BGb (6011) 36
K(U+HL)=0.6

Also at 10, 20, 55 C

ClO4- HL Perchlorate CAS 7001-90-3 (287)

Perchlorate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ nmr NaClO4 ? 2.0M U K1=-0.92 1964PCa (6390) 37

U+++ nmr NaClO4 20°C 2.0M U K1=-0.85 1963VRb (6391) 38

Medium:HClO4

F- HL Fluoride CAS 7644-39-3 (201)

Fluoride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ cal NaClO4 25°C 4.0M U H 1990AHa (7281) 39
DH(U+HF=UF+H)=17.52 kJ mol-1; DH(UF+HF=UF2+H)=9.8; DH(UF2+HF=UF3+H)=8

U+++ ISE NaClO4 23°C 1.0M C K1=8.48 B2=14.66 1990SCa (7282) 40
B3=19.51
B4=23.92

Medium: 1.0 M HClO4/NaClO4. Method: F ion selective electrode.

U++++ EMF none 25°C 0.0 U T H K1=9 B2=14 1980LTb (7283) 41
B3=19
B4=24
B5=25
B6=28

100 C: K1=9, B2=16, B3=21, B4=25, B5=27, B6=29; 200 C: K1=11, B2=18, B3=23,
B4=27, B5=29, B6=32. Evaluated data

U++++ ISE KC1 25°C 1.00M U K1=7.34 B2=13.12 1974Kia (7284) 42
B3=17.46
B4=21.8

U++++ EMF NaClO4 25°C 4.0M U 1969GVa (7285) 43
K(U+HF=UF+H)=5.37
B(U+2HF=UF2+2H)=8.29
B(U+3HF=UF3+3H)=9.4

U++++ ISE NaClO4 20°C 4.0M U 1969NOb (7286) 44
K(U+HF=UF+H)=5.54
K(UF+HF=UF2+H)=3.18
K(UF2+HF=UF3+H)=2.0

Medium: HClO4

U++++ nmr NaClO4 20°C var U K1=8.78 B2=14.48 1966VRa (7287) 45
By nuclear magnetic resonance; K2/K1(Al+++)= -0.55. Spectrophotometry also

U++++ nmr oth/un ? 0.0 U K1=7.15 B2=12.40 1964PCa (7288) 46
B3=17.30

Method: nmr

U++++ sol oth/un ? var U 1963LNa (7289) 47
Kso(UF4(H2O)2.5)=-21.24

U++++ nmr NaClO4 ? 2.0M U K1=7.15 B2=12.41 1963VRa (7290) 48

U++++ nmr NaClO4 20°C 2.0M U K1=7.15 B2=12.4 1963VRb (7291) 49
B3=17.7

U++++ oth oth/un 25°C dil U 1961NLa (7292) 50
K(U(OH)4+HF=U(OH)3F)=16.37
K(U(OH)3F+HF=U(OH)2F2)=10.38
K(U(OH)2F2+HF=U(OH)F3)=6.21
K(U(OH)F3+HF=UF4)=2.48

U++++ sol NaClO4 25°C 0.12M U 1960SBb (7293) 51
Ks(UF4(s)=UF2+2F)=-12.46
Ks(UF4(s)=UF3+F)=-3.96
K3=4.23
K4=4.27

Medium: HClO₄. K₅=1.59, K₆=2.30

FClBrI HL (541)

Halides, comparative (for book data under ligand 80)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U+++	nmr	oth/un	20°C	2.0M	U				1963VRb (7437)	52
								K1=0.8(Cl) K1=0.3(Br) K1=0.2(I)		

I- HL Iodide CAS 10034-85-2 (20)

Iodide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U+++	nmr	NaClO ₄	?	2.0M	U			K1=0.18	1964PCa (8412)	53

Method: NMR. Medium: HClO₄

NH₃ L Ammonia CAS 7664-41-7 (414)

Ammonia

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U+++	gl	R4N.X	25°C	5.00M	U			K1=4.2	1985MMa (9220)	54

NO₃- HL Nitrate CAS 7697-37-2 (288)

Nitrate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U+++	sp	KNO ₃		var	U			K1=-3.8	1973LEa (9970)	55

U+++	sp	KNO ₃	?	var	U				1973RAa (9971)	56
B(U(H ₂ O) ₈ +6L=UL ₆ +8H ₂ O)=6.23										

U+++	dis	NaClO ₄	20°C	8.0M	U			K1=-0.08 B2=0.58 B3=-0.43 B4=-0.30	1970LKa (9972)	57
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U+++	dis	NaClO ₄	0°C	3.80M	U			K1=0.1	1969RPb (9973)	58
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U+++	sp	oth/un	27°C	0.0	U			K1=1.55	1966SNe (9974)	59
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U+++	sp	non-aq	20°C	100%	U				1965WMa (9975)	60
Kd(UL ₄ (TBP)+H ₂ O)=-1.46										

Medium: TBP. Kd: UL₄(TBP)+H₂O=U(OH)L₃(TBP)+HL(TBP)

U+++	sp	NaClO ₄	20°C	1.0M	U	I		K1=0.04 B2=-0.3	1964MWb (9976)	61
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Medium: LiClO₄. K₁=0.06(I=2), 0.20(I=3), 0.18(I=4);
B₂=0.0(I=2), 0.3(I=3), 0.8(I=4); B₃ < (I=1 to 4)

U+++ dis KN03 25°C 1.75M U I 1963SKb (9977) 62
Kd(U+4L+2TBP(kerosene))=0.65
Kd=1.13(I=2.75)

U+++ sp NaClO₄ 26°C 3.50M U I K₁=0.36 B₂=0.47 1962EKa (9978) 63
B₃=0.42
B₄=0.18

Medium: HClO₄. At I=2 M: K₁=0.2, B₂=0.17, B₃=-0.02, B₄=-0.46 plus others

OH- HL Hydroxide (57)
Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ sol NaClO₄ 25°C 0.01M C 1998CPa (12347) 64
K(UO₂(s)+2H₂O=U(OH)₄)=-7.3
*K_{so}(UO₂)=-2.3

Medium: 0.008 M NaClO₄. *K_{so}: UO₂(s)+4H=U+2H₂O.

U+++ sol oth/un 21°C 0.05M U 1990RFa (12348) 65
*K(1,2) < -4.0
*K(1,3) < -8.0
*K(1,4) < -12.0
*K(1,5) < -26
*K₁=-0.50. K_{so}=-52.0. *K_s(UO₂.xH₂O(am)+3H=UOH)=3.5

U+++ EMF NaClO₄ 25°C 3.00M C 1986BFa (12349) 66
*K_{so}(UO₂(s)+4H=U+2H₂O)=-1.2
I=0 M corr. *K_{so}=-1.6

U+++ EMF none 25°C 0.0 U T H 1980LTb (12350) 67
*K₁=-1
*B₂=-2
*B₃=-5
*B₄=-9. *B₅=-13

100 C: *K₁=1, *B₂=0, *B₃=-2, *B₄=-5, *B₅=-10; 200 C: *K₁=2, *B₂=2, *B₃=1,
*B₄=-3, *B₅=-8. Evaluated data

U+++ sp oth/un 25°C 0.15M U T K₁=13.57 1978NNa (12351) 68
Temps from 25 to 150 Degrees C. At 150, K₁=13.40. Range of I: 0.12 to 2.0

U+++ sp NaClO₄ ? U 1972GKd (12352) 69
*K₁=-1.29

Medium: HClO₄

U+++ nmr oth/un ? U B₂=26.2 1969VSa (12353) 70

U++++	sp	none	25°C	0.0	M		1967STe (12354)	71
						*K1=-1.11		
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U++++	sp	NaClO ₄	25°C	1.00M	U		1964MWb (12355)	72
						*K1=-1.57		
Medium: 1 M LiClO ₄								
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U++++	oth	oth/un	700°C	100%	U		1964NTa (12356)	73
						Ks=-0.88		
Medium: molten Na0.5K0.5Cl. Ks: UO ₂ (s)+UCl ₄ =(UOCl ₂) ₂ , by analysis, m units								
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U++++	nmr	oth/un	?	var	U	K1=12.5	1963VRc (12357)	74
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U++++	EMF	oth/un	25°C	var	C I		1962RAb (12358)	75
						*K1=3.05		
						*K2=-1.95		
Medium: UC ₁₄ ; method: H electrode. In 1.5% EtOH *K2=-3.13								
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U++++	sol	oth/un	?	?	U		1960SGa (12359)	76
						K _{so} (U(OH) ₄)=-51.96		
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U++++	gl	NaClO ₄	25°C	2.0M	U I		1959SHa (12360)	77
						*K1=-1.68		
In D ₂ O *K1=-1.74								
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U++++	sol	none	25°C	0.0	U		1957GLb (12361)	78
						K _s (U(OH) ₄ (s)+OH)=-3.77		
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U++++	oth	none	25°C	0.0	U		1956DPa (12362)	79
						*K _{so} (UO ₂ (s))=3.80		
						*K(UO ₂ (s)+3H=U(OH))=2.60		
-----	-----	-----	-----	-----	-----	-----	-----	-----
U++++	oth	none	25°C	0.0	U		1956DPa (12363)	80
						*K _{so} (U(OH) ₄ (s))+4H)=3.80		
						*K _s (U(OH) ₄ (s)+3H)=8.78		
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U++++	gl	NaClO ₄	25°C	3.0M	U		1956HIa (12364)	81
						*K1=-2.0		
						*B(n+1,3n)=-1.2-3.4n		
*B(n+1,3n): K((n+1)M+3nH ₂ O=M(n+1)(OH) ₃ n+3nH)								
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U++++	sp	NaClO ₄	24°C	0.19M	U T		1955BEa (12365)	82
						*K1=-1.12		
Medium: HClO ₄ . *K1=-1.38(15.2 C)								
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U++++	sp	none	25°C	0.0	U H		1955BEa (12366)	83
DH(*K1(U+H ₂ O=UOH+H))=44.8 kJ mol ⁻¹ , DS=138 J K ⁻¹ mol ⁻¹								
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U++++	sp	NaClO ₄	25°C	0.50M	U T		1955KNa (12367)	84
						*K1=-1.47		

*K1=-1.90(10 C), -1.00(43 C)

U++++ sp none 25°C 0.0 U T H 1955KNa (12368) 85

*K1=-0.68

DH(*K1)=49.0 kJ mol-1, DS=151 J K-1 mol-1(25 C); *K1=-1.12(10 C), -0.18(43 C)

U++++ sp NaClO4 25°C 0.50M U I 1950KNa (12369) 86

*K1=-1.50

*K1=-1.63(I=2), -1.56(I=1))

U++++ oth oth/un 20°C var U 1934LAa (12370) 87
*B2(U(H2O)6)=-2.30

method:magnetic susceptibility

P04--- H3L Phosphate CAS 7664-38-2 (176)

Phosphate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U++++ EMF none 25°C 0.0 U T H 1980LTb (13353) 88

K(U+HP04)=12

K(U+2HP04)=22

K(U+3HP04)=31

K(U+4HP04)=39

100 C: values: 14, 24, 32, 39; 200 C: values: 17, 27, 37, 43.

Evaluated data

U++++ sol KCl 20°C var U 1967MEb (13354) 89

K(U(HL)2(s)+4H=U+2H3L)=-9.96

Kso(U(HL)2)=-26.80

K(U+HL)=12.0

K(U+2HL)=22.0

Medium:HCl var. K(U+3HL)=30.6, K(U+4HL)=38.6 plus other sol. products

U++++ oth oth/un 25°C ? U 1960DMa (13355) 90

Ks(U(HL)2=U+2HL)=-27.5

U++++ sol oth/un 20°C ? U 1960MAd (13356) 91

Ks(U(HL)2=U+2HL)=-27.74

P207--- H4L Pyrophosphate CAS 2466-09-3 (198)

Diphosphate; from (HO)2PO.O.PO(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U++++ sol NaClO4 25°C 0.10M U K1=19.07 1967MSc (13666) 92

Kso(UL(H2O)20)=-23.87

SCN- HL Thiocyanate CAS 463-56-9 (106)

Thiocyanate;

U+++ dis NaClO₄ 25°C 2.0M U 1950BLb (16625) 105

*K1=2.53

*K2=-0.13

Medium: HClO₄

W04-- H2L Tungstate CAS 13783-36-3 (445)

Tungstate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ oth oth/un 16°C 0.10M U 1971BRc (17447) 106
K'=4.15

K': 3U + 4HW6021(5-) = 3UW8028(4-) + 4H. Method: paper electrophoresis

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)

Ethanedioic acid; (COOH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ sol oth/un 25°C 0.50M U K1=9.01 1970MKe (19116) 107

C2H4O2 HL Acetic acid CAS 64-19-7 (36)

Ethanoic acid; CH₃.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ oth oth/un ? 0.50M U K1=2.34 B2=4.30 1969MOc (20208) 108
B3=6.73
B4=8.97
B5=11.2
B6=13.8

Metal ion: U0++. B7=15.9, B8=18.9

Data from survey of literature data

C2H5NO2 HL Glycine CAS 56-40-6 (85)

2-Aminoethanoic acid; H₂N.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ vlt oth/un 22°C ? U K1=10.3 B2=11.3 1976NFa (21738) 109
B4=17.9

C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)

L-2-Hydroxypropanoic acid; CH₃.CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ gl NaClO₄ 25°C 1.00M C K1=4.4 B2=8.3 1984LLa (25559) 110
B3=11.8
B4=15.1

B5=17.5

B6=19.0

C3H7NO2 HL Alanine CAS 56-41-7 (86)
2-Aminopropanoic acid; H2N.CH(CH3).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
U++++	vlt	oth/un	22°C	?	U			K1=10.3 B4=18.8	B2=13.4	1976NFa (26282)	111

U++++ vlt NaClO4 25°C 0.1M U K1=9.00 1975FNa (26283) 112

C4H6O4 H2L Succinic acid CAS 110-15-6 (112)
1,4-Butanedioic acid; HOOC.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++++ sol oth/un 25°C 0.50M U K1=9.78 1970MKe (30057) 113

C5H8O2 HL Acetylacetone CAS 123-54-6 (164)
Pentane-2,4-dione; CH₃.CO.CH₂.CO.CH₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

C5H8O4 H2L Glutaric acid CAS 110-94-1 (420)
Pantanedioic acid; HOOC.CH2.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++++ sol oth/un 25°C 0.50M U K1=8.81 1970MKe (38364) 115

C5H11NO₂ HL DL-Valine CAS 516-06-3 (186)
DL-2-Amino-3-methylbutanoic acid; H₂N.CH(CH(CH₃)₂).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U++++ vlt oth/un 22°C ? U K1=9.8 B2=13.3 1976NFa (40897) 116
B4=19.6

C6H5NO₂ HL Nicotinic acid CAS 59-67-6 (419)
3-Pyridine-carboxylic acid; C5H4N.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U++++ sp alc/w 25°C 100% U M 1972RKb (42689) 117

$$K(UCl_2+HL)=1.95$$

$$K(UCl+2L)=3.84$$

Medium: EtOH

C6H6O4 HL Kojic acid CAS 501-30-4 (1800)

5-Hydroxy-2-(hydroxymethyl)-4H-pyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U+++ vlt NaNO₃ 20°C 1.0M U K1=4.9? 1967HAa (44248) 118

C6H8O7 H3L Citric acid CAS 77-92-9 (95)

2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH₂.CH(OH)(COOH).CH₂COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U+++ gl oth/un 25°C 0.50M U K1=11.53 B2=19.46 1966NUa (46293) 119

U+++ sp NaClO₄ 20°C 0.10M U 1960ASa (46294) 120

$$K(U(OH)_2+L)=13.5$$

C6H10O4 H2L Adipic acid CAS 124-04-9 (401)

1,6-Hexanedioic acid; HOOC.(CH₂)₄.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U+++ sol oth/un 25°C 0.50M U K1=9.28 1970MKe (48093) 121

C6H11N05 H2L HIMDA CAS 93-62-9 (192)

N-(2-Hydroxyethyl)iminodiethanoic acid; HO.CH₂.CH₂.N(CH₂.COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U+++ gl KCl 25°C 0.10M U 1968CMb (48805) 122

$$K(U(HL)_2OH+H)=3.67$$

C6H13N02 HL Leucine CAS 61-90-5 (47)

2-Amino-4-methylpentanoic acid; H₂N.CH(CH₂.CH(CH₃)₂).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U+++ vlt oth/un 22°C ? U K1=7.0 B2=10.5 1976NFa (50114) 123

$$B3=15.6$$

C7H7N02 HL CAS 495-18-1 (184)

Benzohydroxamic acid; C₆H₅.CO.NH.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U+++ gl NaClO₄ 25°C 0.01M U K1=9.89 B2=18.00 1966BBf (55518) 124

B3=26.32

B4=32.95

Medium: HClO4

C8H5O2F3S HL TTA CAS 326-91-0 (165)

4,4,4-Trifluoro-1-(2-thienyl)butane-1,3-dione; F3C.CO.CH2.CO.C4H3S

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ sp oth/un 25°C 0.10M U K1=7.2 1964PCa (58691) 125

C9H7NO HL Oxine CAS 148-24-3 (504)

8-Hydroxyquinoline (8-quinolinol);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ sp non-aq 25°C 100% U HM K1=-0.22 B2=2.23 1976EWb (64366) 126

DH(K1)=-11.7 kJ mol-1 and DS(K1)=-43.5 J mol-1 K-1. DH(K2)=-25.9,

DS(K2)=39.3 in 1,2-dichloroethane.

C9H11N02 HL Phenylalanine CAS 63-91-2 (2)

2-Amino-3-phenylpropanoic acid; H2N.CH(CH2.C6H5)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ vlt oth/un 22°C ? U K1=10.1 B2=13.3 1976NFa (65982) 127

B3=18.7

C9H16O4 H2L Azelaic acid CAS 123-99-9 (3255)

Nonanedioic acid; HOOC.(CH2)7.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ dis oth/un 25°C 0.50M U K1=9.08 1970MKe (67799) 128

C10H16N2O8 H4L EDTA CAS 60-00-4 (120)

1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ sp oth/un 25°C ? U M 1980PPa (74264) 129

K(UL+Benzoylacetone)=3.22

K(UL+Thenoylacetone)=2.80

K(UL+Dibenzoylmethane)=2.50

U+++ oth oth/un ? 0.0 U K1=20.3 1969MIb (74265) 130

From survey of literature data

U+++ gl KCl 20°C 0.10M U T 1968CMb (74266) 131

K(ULOH+H)=4.72

$$K((ULOH)2+2H)=6.53$$

$$K(2ULOH=(ULOH)2)=2.9$$

U+++ gl KCl 25°C 0.10M U M K1=25.8 1967CMd (74267) 132
 K(UL+A)=5.61
 K(UL+B)=16.22
 K(UL+C)=11.08
 K(UL+D)=14.2

H4A=dihydroxybenzene-3,5-disulfonic acid, H4C=sulfosalicylic acid,
 H4B=dihydroxynaphthalene-3,6-disulphonic acid, H2D=catechol

U+++ gl KCl 25°C 0.10M U M 1967CMd (74268) 133
 K(UL+A)=8.2
 K(UL+B)=4.2
 K(UL+C)=9.72
 K(ULC(OH)+H)=7.14

H2A=iminodiacetic acid, H2B=phthalic acid, H2C=8-hydroxyquinoline-5-sulfonic

U+++ gl oth/un 25°C 0.0 U I 1963EKc (74269) 134
 K(UL+OH)=8.95
 K(2ULOH=(ULOH)2)=2.78
 K(ULOH+H)=4.94
 K((ULOH)2+2H)=7.01

K(UL+OH)=9.00(I=0.01), 9.07(I=0.1), 9.08(I=0.25), 9.17(I=0.5), 9.13(I=1);
 K(2UL(OH)=U2L2(OH)4)=2.84(I=0.01), 2.75(I=0.1), 2.79(I=0.25), 2.48(0.5), 2.86(1)

U+++ gl oth/un 25°C 1.0M U I 1963EKc (74270) 135
 K(ULOH+OH)=6.87
 K(UL(OH)2+2H)=12.83
 K(2UL2+H2L)=3.3
 K(2UL2+L)=12.93

Data at I=0.01 to 1.0. K(U2L3+2OH=12.11(I=0.1), K(HU2L3+H)=2.8(I=0.1)

U+++ sp oth/un 25°C 0.10M U K1=25.83 1962KEa (74271) 136

U+++ sol oth/un 25°C ? U K1=25.6 1959KSa (74272) 137

U+++ sp NaClO4 ? 0.10M U I 1959SMA (74273) 138
 K(UF+L)=17.50

C12H8N4O4S2 H2L CAS 3385-61-8 (2586)

7-(2-Thiazolylazo)-8-hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U+++ sp diox/w 25°C 50% U K1=8.55 B2=15.21 1977RIA (80557) 139

C14H22N2O8 H4L CDTA CAS 482-54-2 (200)

trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U+++	sp	oth/un	RT	6.0M	U				1997RRc (101654)	147
K1eff=6.12 B2eff=12.04										
Medium: 6 M HCl										

C26H28O4		H2L		B(CH2AcAcCH2)2B		(2253)				
3,5,16,18-Tetraoxo[7.7]metacyclophane ; Cyclo-(-C6H4.(CH2)2.CO.CH2.CO.(CH2)2-)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U+++	gl	diox/w	24°C	50%	U		K1=11.2		1979ACa (104023)	148

C34H46N4O14		H2L					CAS 226947-33-7	(8530)		
N,N'-Bis[(benzo-15-crown-5)-oylmethyl]diaminoglyoxime;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U+++	gl	mixed	25°C	60%	U		K1=11.48		1999ADd (106078)	149
B(UO2HL)=15.48										
B(UO2H2L2)=30.32										
B(UO2H-1L)=5.77										
B(UO2H-2L)=-2.62										
Medium: 60% v/v acetone/H2O, 0.20 M KNO3.										

C76H52O46		H9L		Gallotannin		CAS 1401-55-4	(2795)			
Tannic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U+++	oth	NaCl	25°C	0.01M	U				1980LVa (107865)	150
K1eff=6.93										
K2eff=5.04										
Method: dialysis at pH 6										

Polymer				Fulvic acid		(1523)				
Fulvic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U+++	oth	NaCl	25°C	0.01M	U				1980LVa (108183)	151
K1eff=6.64										
K2eff=4.94										
Method: dialysis at pH 6										

Polymer				Humic acid		(1524)				
Humic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

U++++ oth NaCl 25°C 0.01M U 1980LVa (108244) 152
K1eff=6.98
K2eff=4.51

Method: dialysis at pH 6

e- HL Electron (442)
Electron;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ EMF KN03 25°C 0.0 U TIH 1990BGb (999) 153
K(UO2+2H+H2(g)=U++++)=9.1
E(2e + UO2++=U++++)=0.269 V

UO2++ EMF NaClO4 25°C 1.0M C TI 1990CVc (1000) 154
K(UO2++ +e=UO2+)=1.013 (60 mV)

Method: cyclic voltammetry. Data for 5-55 C. Data extrap to 1.0 M (HC1O4) using SIT.

UO2++ EMF NaClO4 25°C 3.00M C 1986BFa (1001) 155
E(2e + UO2++=UO2(s))=0.305 V
E(2e+4H+UO2++=U++++)=0.268 V

UO2++ EMF none 25°C 0.0 U T H 1980LTb (1002) 156
K(4UO2+2H2O=4U(V)O2+4H+O2)=-72
K'=-64.7
K"=-248

K': 2UO2+4H=2U(IV)+2H2O+O2. K": 4UO2+4H=4U(III)+2H2O+3O2. At 100 C, values are: -53, -54.6, -193. At 200 C: -37, -46.0, -148. Evaluated data

UO2++ sp none 25°C 0.00 U H 1974BFc (1003) 157
K=-4.85

K: UO2++ + U++++ + 2H2O=2UO2+ + 4H+; DH=77.8 kJ mol-1. Data for 25-90 C

UO2++ sp oth/un 450°C 100% U T H 1974LLb (1004) 158
K=1.15

Medium:(Li,K)Cl eutectic; K: UO2++ + U++++ + 2H2O(g) + 4Cl=2UO2+ + 4HCl(g); DH=107.9 kJ mol-1; K=1.95(550 C), 2.44(600 C), 2.85(650 C)

UO2++ vlt none 25°C 0.00 U 1970BCc (1005) 159
K(UO2 + e=U(V)O2)=2.76(0.163V)

UO2++ kin oth/un 25°C 1.00M U H 1970NEc (1006) 160
K(UO2 + Np++++=UO2+ +NpIV)=-1.5

Medium: HC1O4; DH=-36.4 kJ mol-1

UO2++ oth oth/un 25°C 0.10M U 1970STa (1007) 161
K=-7.81

Medium: HCl; K: UO2 + U++++ + 2H2O=2U(V)O2 + 4H+

U02++ sp non-aq 650°C 100% U 1964WAa (1008) 162
K=-5.92

Medium fused KCl. K: $UO_2^{++} + 2Cl^- = 2UO_2^+ + Cl_2(g)$

U02++ EMF oth/un 25°C 1.0M U IH 1961SOe (1009) 163
K=11.12(328.8 mV)

Medium: HClO4. K: $UO_2 + 4H + 2e = U(IV) + 2H_2O$. DS(K)=-205 J K-1 mol-1. Data also for 1 M HCl (DS=-195) and 1 M H2SO4 (DS=-137) and data for various I values

U02++ cal NaClO4 25°C 0.50M U H 1958FOa (1010) 164
Medium: HClO. DH($UO_2 + 4H + 2e = U(IV) + 2H_2O$)=-137.8 kJ mol-1

U02++ EMF oth/un 25°C 1.0M U 1958HMa (1011) 165
K=18.26(540 mV)

Medium: HCl. K: $UO_2 + 4H + 2e = U(IV) + 2H_2O$

U02++ EMF none 25°C 0.0 U H 1957GUa (1012) 166
K=13.76(407 mV)
K: $UO_2 + 4H + 2e = U(IV) + 2H_2O$. DH(K)=-142 kJ mol-1

U02++ EMF none 25°C 0.0 U I 1952LAb (1013) 167
K=11.29(334 mV)
K($UO_2 + e = UO_2(V)$)=0.88(52 mV)
K($UO_2 + 2e = UO_2(s)$)=15.1(447 mV)
K: $UO_2 + 4H + 2e = U(IV) + 2H_2O$. Other values from thermodynamic data at I=0

U02++ sp none 25°C 0.0 U 1951NKa (1014) 168
K=-6.23
K: $UO_2 + U(IV) + 2H_2O = 2UO_2(V) + 4H$. Polarography also used

U02++ vlt oth/un 25°C 1.0M U 1949KHa (1015) 169
K($UO_2 + e$)=1.07(63 mV)

U02++ EMF none 25°C 0.0 U 1949KNa (1016) 170
K=10.5(0.31 V)
K: $UO_2 + 4H + 2e = U(IV) + 2H_2O$

U02++ vlt NaClO4 25°C 0.50M U I 1949KOa (1017) 171
K($UO_2 + e$)=1.05(62 mV)
K: $UO_2 + e = UO_2(V)$. Same value in 0.1 M KCl

U02++ EMF oth/un 18°C 0.05M U 1910TIa (1018) 172
K=14.0(404 mV)
Medium: 0.05 to 0.5 M H2SO4. K: $UO_2 + 4H + 2e = U(IV) + 2H_2O$

U02++ EMF oth/un 18°C var U 1908LMa (1019) 173
K=14.5(419 mV)
Medium: H2SO4. K: $UO_2 + 4H + 2e = U(IV) + 2H_2O$

AsO4--- H3L Arsenate CAS 7778-39-4 (1557)
Arsenate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sol	oth/un	20°C	var	U	M			1956CSc (1164)	174
								Ks(UO2HL(s)=UO2+HL)=-10.50		
								Ks(UO2LiL(s)=UO2+Li+L)=-18.82		
								Ks(UO2NaL(s)=UO2+Na+L)=-21.87		
								Ks(UO2LL(s)=UO2+K+L)=-22.60		

Ks(UO2(NH4)L(s)=UO2+NH4+L)=-23.77

Br- HL Bromide CAS 10035-10-6 (19)
Bromide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	NaClO4	20°C	0.05M	C			K1=1.79	1989RAb (2374)	175
Medium:	0.05 M	NaClO4/HClO4.								

UO2++	sp	none	25°C	0.0	U			K1=-0.20	1957DMa (2375)	176
UO2++	EMF	NaClO4	20°C	1.0M	U			K1=-0.30	1951AHa (2376)	177
Method:	quinhydrone electrode.									

BrO3- HL Bromate (6017)
Bromate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	dis	NaClO4	25°C	0.10M	U	H		K1=0.20	1988KCb (2438)	178
DH=0.1	kJ mol-1,	DS=4	J K-1 mol-1							

CN- HL Cyanide CAS 74-90-8 (230)
Cyanide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	oth/un	30°C	dil	U				1971JSb (2770)	179
								K(UO2+Mo(CN)8)=2.97		

UO2++	sp	oth/un	25°C	var	U				1969KBC (2771)	180
								K(UO2+Mo(CN)4(OH)3H2O)=3.71		

CO3-- H2L Carbonate CAS 465-79-6 (268)
Carbonate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sol	NaClO4	24°C	0.10M	U				1993MKb (3409)	181

$K_{so}(UO_2CO_3) = -13.89$. By pH titration under CO₂.

U02++ sp NaClO₄ 25°C 0.50M C 1991BCd (3410) 182
K(UO₂(CO₃)₂+CO₃)=6.35

Method: thermal lensing spectrophotometry.

U02++ gl NaClO₄ 25°C 0.50M C H 1991GLa (3411) 183
B(1,2,1)=-8.71
B(1,4,2)=-19.57
B(3,12,6)=-49.68
B(2,5,1)=-19.40

B(p,q,r); pUO₂+qH₂O+rCO₂(g)=(UO₂)p(OH)q(CO₂)r+qH

B(1,6,3)=-29.45; B(11,24,6)=-72.48

U02++ EMF NaClO₄ 25°C 0.0 C TI 1990CVc (3412) 184
B3=21.3

Method: cyclic voltammetry.

U02++ cal oth/un 25°C U 1988USA (3413) 185
DH(UO₂+3L)=-42.1 kJ mol⁻¹

Ionic strength is variable within 0.27-1.08

U02++ sp NaClO₄ 25°C 3.0M C 1986GRb (3414) 186
K(3(UO₂)(CO₃)₃=(UO₂)₃(CO₃)₆+3(CO₃))=-11.3

U02++ cal oth/un 25°C 1.6M C H 1985SFa (3415) 187
Medium: 1.6 M (Na₂CO₃ + Na₂SO₄). DH(B2)=-39.6 kJ mol⁻¹,
DH(B3)=-57.5 kJ mol⁻¹.

U02++ gl NaClO₄ 25°C 3.00M C I K1=8.3 1984GFa (3416) 188
B(UO₂L₂)=16.20
B(UO₂L₃)=22.61
B((UO₂)₃L₆)=56.2

K_{so}(UO₂L) = -13.94; Data also at 0.5 M NaClO₄, and calc for 0.0 M

U02++ cal KCl 25°C 3.0M C H 1984GSe (3417) 189
DH(UO₂+3L)=-35.9 kJ mol⁻¹, DS=312 J K⁻¹ mol⁻¹; DH((UO₂)₃L₆+3L)=-46.2, DS=67.7
DH(UO₂+2L=0.33(UO₂)₃L₆)=-20.5, DS=290; DH(UO₂L₂=0.33(UO₂)₃L₆)=-35.1, DS=-69

U02++ gl NaClO₄ 25°C 3.0M C 1983FGb (3418) 190
B3=13.3

K(2UO₂(CO₃)₃ + 4HC₃ = UO₂(CO₃)₃ + U(CO₃)₅ + 2CO₃)=4.98

U02++ gl NaClO₄ 25°C 0.10M U M B2=16.15 1982MAC (3419) 191
B3=21.81

U02++ gl NaClO₄ 25°C 3.0M C 1981CFb (3420) 192
K(3UO₂(CO₃)₃ + 3CO₂(g) + 3H₂O = (UO₂)₃(CO₃)₆ + 6 HC₃)=-6.45

U02++ sp NaClO₄ 25°C 3.0M C 1981FGc (3421) 193

$$K(3UO_2(CO_3)_3 + 6H = (UO_2)_3(CO_3)_6 + 3CO_2 + 3H_2O) = 1.62$$

U02++ EMF none 25°C 0.0 U T H K1=10.1 B2=17.1 1980LTb (3422) 194
B3=21.4

100 C: K1=10.6, B2=18, B3=21.3; 200 C: K1=13, B2=19, B3=24.0. Evaluated data

U02++ gl NaClO4 25°C 3.00M C M 1979CFa (3423) 195
K(UO2+2OH+CO2)=-8.99
K(3UO2+5OH+CO2)=-16.40
K(11UO2+25OH+6CO2)=-76.5
K(13UO2+30OH+7CO2)=-91.8

U02++ sp NaNO3 20°C 0.10M U B2=16.22 1977JSa (3424) 196
K3=5.48

U02++ dis oth/un 20°C 0.10M U 1975CSa (3425) 197
B(UO2+3L)=21.54

U02++ vlt oth/un 25°C var U B2=4.0 1973AGa (3426) 198
B3=7.7
K1=2.2-2.8

Medium: Na2CO3

U02++ EMF NaClO4 25°C 3.00M U 1972CIa (3427) 199
K(UO2+CO2(g)+H2O=UO2L+2H)=-9.00. K(3UO2+CO2(g)+4H2O=(UO2)3(OH)3+5H)=-16.6

U02++ sol none 25°C 0.0 U T 1972SNb (3428) 200
+Kps0=3.9
Ks1=-4.39
+Ksp2=-15.66

+Kps0=3.65(50 C); Ks1=-4.37(50 C), -4.35(100 C), -4.34(150 C), -4.21(200 C);
+Ksp2(UO2CO3(s)+CO2(g)+H2O=UO2(CO3)2+2H)=-16.00(50 C)

U02++ EMF NaClO4 25°C 0.10M U B2=16.16 1969TSb (3429) 201
B3=21.57

U02++ EMF NaClO4 25°C 0.10M U 1968BIa (3430) 202
K3=4.7

U02++ ix NaNO3 ? 0.50M U 1962PNa (3431) 203
K3=7.0
B3=ca.23

U02++ gl oth/un 20°C var U 1962PNb (3432) 204
K3=5.5

U02++ sol R4N.X rt 0.20M U B2=15.57 1960BKa (3433) 205
B3=20.70
Kso(UO2CO3(s))=-11.73
Ks=1.22

Also by glass electrode. Medium: NH₄N₃. K_s: UO₂(OH)₂(s)+H₂L=UO₂L(s)+2H₂O

UO₂++ sol R4N.X 25°C 1.0M U 1959KSc (3434) 206
B3=22.8

Also by glass electrode. Medium: NH₄Cl

UO₂++ sp oth/un 26°C ca.2 U 1956BCb (3435) 207
K3=ca.3.5

By solubility K_s(Na₄UO₂L₃(s)=4Na+UO₂L₃)=-2.8 to -2.0

UO₂++ sol none 25?°C 0.0 U 1955MBd (3436) 208
K_s=4

I=0 corr. K(UO₂L₂+2HC₀3=UO₂L₃+CO₂(g))=1.81. From thermodynamic quantities
B₂=14.6, B₃=18.3. K_s: UO₂(OH)₂H₂O(s)+CO₂(g)=UO₂CO₃(s)+2H₂O

UO₂++ ix oth/un ? var U 1955PAb (3437) 209
K3=7.0

UO₂++ sol none 25?°C 0.0 U 1954BUa (3438) 210
K3=3.78

C₆N₆Fe---- H4L (2191)
Hexacyanoferrate (II); Fe(II)(CN)₆----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ sol oth/un 25°C var U 1956TGb (3613) 211
K_{so}((OU₂)₂L)=-13.15

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ con non-aq ? 100% U M 1992RSa (5917) 212
K(UO₂A+L)=2.60

Medium: 1% DMSO+99% MeCN. A=N,N'-disalicylidene-1,2-benzenediamine (+others)

UO₂++ sp NaClO₄ 20°C 0.05M C K1=1.98 1989RAb (5918) 213
Medium: 0.05 M NaClO₄/HClO₄.

UO₂++ sp NaClO₄ 25°C 1.0M C TI K1=-0.29 B2=-2.00 1981ASc (5919) 214

Medium: 1.0-4.9 M HCl/HClO₄. Data for 25-40 C. At 25 C and I=1.0 M:

DH(K1)=9.62 kJ mol⁻¹, DS(K1)=26 J K⁻¹ mol⁻¹; DH(B2)=20.1, DS(B2)=29.

UO₂++ EMF none 25°C 0.0 U T H K1=2 1980LTb (5920) 215
100 C: K1=2; 200 C: K1=4. Evaluated data

UO₂++ EMF KCl 70°C 0.03M U T K1=1.87 1977NNa (5921) 216
Temps from 70 to 150 Degrees. At 150 C: K1=3.25

ClO₃- HL Chlorate CAS 7790-93-4 (971)
Chlorate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ dis NaClO₄ 25°C 0.10M U H K1=0.08 1988KCb (6065) 234
DH=-3.9 kJ mol⁻¹, DS=-11 J K⁻¹ mol⁻¹

ClO₄- HL Perchlorate CAS 7001-90-3 (287)
Perchlorate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ EMF non-aq 25°C 100% U K2=5.80 1967AJa (6392) 235
Medium:MeCOOH

UO₂++ oth non-aq 20°C 100% U K(R4NL+R4NUO2L3)=1.53 1964VSa (6393) 236
Method:infrared spectra. Medium:C6H6. R=C10H21

F- HL Fluoride CAS 7644-39-3 (201)
Fluoride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ gl NaClO₄ 25°C 1.0M C M K1=4.24 B2= 7.21 1999ASa (7294) 237
K3=2.07
K(UO₂+L+ac)=6.66
K(UO₂+2L+ac)=9.63
K(UO₂+3L+ac)=11.70

Additional method: nmr. K(UO₂+2F+2ac)=10.15. Ac: ethanoate.

UO₂++ ISE NaClO₄ 25°C 3.00M C K1=4.86 B2=8.62 1993FSa (7295) 238
B3=11.71
B4=13.78

Specific ion interaction parameters are also given

UO₂++ sp NaClO₄ 20°C 0.05M C K1=3.15 1989RAb (7296) 239
Medium: 0.05 M NaClO₄/HClO₄.

UO₂++ ISE non-aq 185°C 100% M K1=4.89 B2=9.255 1988JHa (7297) 240
B3=12.91
B4=16.26
B5=18.28

Medium: molten KSCN. K1=mol⁻¹ kg, B2=mol⁻² kg² etc.

UO₂++ ISE NaClO₄ 21°C 1.0M C K1=4.56 B2= 7.99 1985SCe (7298) 241
B3=10.34

UO₂++ EMF none 25°C 0.0 U T H K1=5.1 B2=9.0 1980LTb (7299) 242

		B3=11.3	
		B4=12.6	
100 C: K1=5.3, B2=9.1, B3=11.4, B4=12.9; 200 C: K1=6.0, B2=9.8, B3=11.6, B4=14.2. Evaluated data			

U02++	dis NaClO ₄ 25°C 2.00M U	K1=1.56	1976PRA (7300) 243

U02++	ISE NaClO ₄ 25°C 1.0M U	K1=4.54 B3=10.41 B4=11.9	1971AKA (7301) 244

Method: quinhydrone and fluoride-ISE			

U02++	cal NaClO ₄ 25°C 1.0M U H		1971AKA (7302) 245
DH(K1)=1.7 kJ mol ⁻¹ , DH(K2)=0, DH(K3)=0.3, DH(K4)=-2.1; DS(K1)=92.5 J K ⁻¹ mol ⁻¹ , DS(K2)=67.4, DS(K3)=47.7, DS(K4)=21.3			

U02++	EMF NaClO ₄ 25°C 4.0M U		1969GVA (7303) 246
		K(U02+HF=U02F+H)=1.5	

U02++	nmr oth/un 20°C 0.50M U	K1=4.65	1969VSA (7304) 247
Method: nmr			

U02++	ix NaClO ₄ 25°C 2.10M U I		1968KKD (7305) 248
		*K1=1.36	
Medium: HClO ₄ . Method:cation exchange. *K1=1.52(I=1.04 to 0.51) At I=0.2: *K1=1.57, *B2=1.64, *B3=1.68			

U02++	dis NaClO ₄ 25°C 2.0M U H		1967AHA (7306) 249
DH(K1)=-8.36 kJ mol ⁻¹ , DS=54.3 J K ⁻¹ mol ⁻¹			

U02++	EMF non-aq 760°C 100% U		1967KSa (7307) 250
		B4=-3.93-4380/T	
Medium:molten (Na/K)Cl, 690-830 C, x units			

U02++	sp NaClO ₄ 25°C 0.65M U		1961CPc (7308) 251
		K(U02+HF=U02F+H)=1.18	

U02++	sp none 20°C 0.0 U	K1=4.77	1961KUA (7309) 252

U02++	con non-aq -5°C 100% U		1960NVa (7310) 253
		K(UO2F ₂ +4HF=UF ₆ +2H ₂ O)=-3.95	
Medium: liquid HF			

U02++	EMF NaClO ₄ 20°C 1.0M U	K1=4.54 K3=2.57 K4=1.34	1956ALa (7311) 254

U02++	EMF NaClO ₄ 20°C 1.00M U	K1=4.59 K3=2.56 K4=1.36	1954ALb (7312) 255

U02++ oth none 25°C 0.0 U K2=4.4 1954BBb (7313) 256

UO₂++ dis NaClO₄ 25°C 2.0M U IH 1954DPa (7314) 257
 *K1=1.42

*K1=1.74(10 C), 1.32(40 C). At 25 C: *K1=1.43(C=1), 1.38(C=0.5), 1.71(0.05)
 DH(*K1)=-23 kJ mol⁻¹. DS=-50

By centrifuge, in U02E2, $K=0.42$ ($0^\circ C$)

UO₂++ sp oth/un 0°C var U K1=5.5 1951BLA (7317) 260
B4-c2 8

UO2++ ix oth/up 25°C var II K1=4.32 1951BLa (7318) 261

UO₂++ ix KC1 25°C var U 1950Mkb (7319) 262
 K(UO₂+HE=UO₂E+H)=1.18

K(002111-002111)-1-18

IO₃- **HL** **Iodate** **CAS 7782-68-5 (1257)**

Todate:

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ dis NaClO4 25°C 0.10M U H K1=1.58 1988KCb (8571) 263
 PH 0.8 kJ mol⁻¹ DS 63.7 K 1 mol⁻¹

DH=9.8 kJ mol⁻¹, DS=63 J K⁻¹ mol⁻¹

$$\begin{aligned} K(U02L2(s)=U02L2) &= -4.28 \\ K(U02L2(s)+L=U02L3) &= -3.34 \\ K_{so}(U02L2) &= -0.71 \end{aligned}$$

Medium:NH4Cl. At 60 C: B2=2.74, K3=0.69, Kso=-6.65, K(UO2L2(s)=UO2L2)=-3.91, K(UO2L2(s)+L=UO2L3)=-3.22

IO4- HL Periodate CAS 13444-71-8 (6063)

Periodate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂⁺⁺ oth NaNO₃ 25°C 0.50M C 1975HUa (8617) 265

Keff=2.079

1971HUa (8)

Medium: KI04. K=1.2 to 1.7

NH3 L Ammonia CAS 7664-41-7 (414)
Ammonia

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl R4N.X 25°C 5.00M U K1=2.0 1985MMa (9221) 267

NH3O L Hydroxylamine; CAS 5470-11-1 (1808)
Hydroxylamine; NH2.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ sol KCl 22°C 2.0M U 1961KOb (9274) 268
Ks(UO2L'2(H2O)3(s)=UO2L'2+3H2O)=-3.39, K(UO2+L=UO2L'+L)=-1.15,
K(UO2L'2+L=HO2L'3+H)=-7.38. L'=NH2O-

NO2- HL Nitrite CAS 7782-77-6 (635)
Nitrite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ con non-aq ? 100% U M 1992RSa (9409) 269
K(UO2A+L)=2.49

Medium: 1% DMSO+99% MeCN. A=N,N'-disalicylidene-1,2-benzenediamine (+others)

NO3- HL Nitrate CAS 7697-37-2 (288)
Nitrate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ sp oth/un 90°C 6.25M U K1=0.15 1980BHb (9979) 270

Raman spectroscopy. Ion pair constant

UO2++ sp KN03 20°C var U B2=-1.7 1970KKc (9980) 271
K(UO2+2L+HL)=-1.7

UO2++ dis NaClO4 20°C 8.0M U K1=0.47 B2=-1.5 1970LKa (9981) 272

UO2++ EMF NaClO4 1.06M U I K1=-0.72 19670Mb (9982) 273
K1=-0.70(I=0.82), -0.43(I=0.54)

UO2++ con alc/w 25°C 100% U K2=3.67 1966EJa (9983) 274
Medium:MeOH. K2=3.45(MeOH+2H2O), 3.21(MeOH+6H2O/UO2)
In EtOH: K2=4.47, 4.25(EtOH+2H2O), 4.16(EtOH+6H2O/UO2)

UO2++ con non-aq 25°C 100% U I K2=5.70 1966JEa (9984) 275
Medium acetone. In MeCOEt: K2=6.02; in i-BuCOMe: K2=6.39

U02++ con non-aq 25°C 100% U I K2=6.36 1966JEa (9985) 276
 Medium: acetone. K2=6.52(MeCOEt), 6.93(i-BuCOMe, 6.64(heptan-2-one))

U02++ con alc/w 25°C 100% U I K2=3.67 1964JEa (9986) 277
 Medium: MeOH anhydrous. K2=4.47(in EtOH), 5.59(PrOH). With U02L2(H2O)6:
 K2=3.21(in MeOH), 4.16(in EtOH)

U02++ dis oth/un 25°C 0.0 U 1962AAa (9987) 278
 Kd(U02+2L+2T(TBP))=1.68
 Medium: 0 corr. Product: U02L2T2(TBP), M units

U02++ ix NaClO4 32°C 1.0M U K1=-1.4 B2=-1.4 1961BTa (9988) 279
 K3=0.9

U02++ con alc/w 25°C 100% U I K2=3.4 1961JCa (9989) 280
 Medium: MeOH. K2=3.3 to 3.9 in EtOH, 6.1 to 6.7 in MeCOEt and other solvents

U02++ sp non-aq ? 100% U 1961RYb (9990) 281
 K4=0.67
 Medium: MeNO2

U02++ sp non-aq ? 100% U M 1960MLa (9991) 282
 Medium: BuOH. K(U02L2+TBP=U02L2(TBP))=0.8

U02++ dis oth/un 25°C 0.0 U H 1960NAC (9992) 283
 Kd=1.43; DH(Kd)=-18.0 kJ mol-1, DS=-32 J K-1 mol-1. TBP and CC14 or kerosene

U02++ dis oth/un ? var U M 1960SSc (9993) 284
 Medium: HL. K(BHL(org)+U02+2L=(BH)2U02L3(org))=0.31(org=CC14), 0.46(org=o-xylene); B=(C8H17)3N

U02++ dis oth/un 25°C 1.0M U H 1959SIa (9994) 285
 Medium: HL. Kd(U02+2L+2TBP(org)=U02L2(TBP)2(org))=1.35, org=n-dodecane. DH(Kd)=26 kJ mol-1. Also Kd for 20 compounds R3PO4 and R3PO3 in place of TBP.

U02++ dis oth/un 19°C 0.0 U M 1959SSa (9995) 286
 Kd(U02+2L+2T(kerosene)=U02L2T2(kerosene))=3.40. T=(isopentyloxy)2(CH3)PO

U02++ dis NaClO4 25°C 0.72M U M K1=-0.2 1959VNa (9996) 287
 Medium: HClO4. Kd(U02+2L+2T(CC14)=U02L2T2(CC14))=1 T=(BuO)3PO;
 K=2 T=(BuO)2BuPO; K=6 T=Bu3PO

U02++ dis oth/un 25°C 0.0 U M 1959VSa (9997) 288
 Kd(U02+2L+hH2O=U02L2(H2O)h(org))=-1.0(Et20,h=4), -2.37(Pr20,h=4),
 -3.22(Bu20,h=2.5), -4.06(Isopetyl ether,h=2.2). Also Kd for 9 other esters

U02++ dis oth/un 25?°C 0.0 U M 1958C0a (9998) 289
 Kd(U02+2L+2TBP(org)=U02L2(TBP)2(org))=1.71. org=amsco 125-90W

U02++ dis oth/un 25°C var U M 1958I0a (9999) 290

Medium:HL. Kd(UO₂+2L+2TBP(kerosene)=UO₂L₂(TBP)2(kerosene)] = 1.08

U02++ con alc/w 25°C 100% U I K2=3.15 1958JEb (10000) 291
K3(?)=1.39

Medium: EtOH, I=0 corr.. In acetone: K2=3.96, K3(?)=2.46

U02++ dis oth/un 25°C 0.0 U M 1957ROa (10001) 292
Kd(UO₂+2L=UO₂L₂(org)) = -3.20(org=Bu₂O), -2.52(Pr₂O), -1.82(i-C₅H₁₁OCOCH₃),
-0.73(BuCOMe), 0.87(cyclohexanone), -0.94(Et₂O), 1.80(20% (BuO)₃PO, 80% kero)

U02++ sp non-aq ? 100% U 1957VLa (10002) 293
K3=3.6

Medium: Me₂CO

U02++ dis NaClO₄ 25°C 2.0M U K1=-0.62 1954DPa (10003) 294
K1=-0.52(10 °C), -0.77(40 °C)

U02++ EMF NaClO₄ 20°C 1.0M U K1=-0.3 1951AHa (10004) 295

U02++ sp NaClO₄ 25°C 7.0M U I K1=-0.57 1949BMa (10005) 296
K1=-0.68(I=5.38)

N3- HL Azide CAS 7782-79-8 (441)
Azide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ 25°C 2.00M U K1=2.14 B2=3.92 1983CNa (10266) 297
B3=5.69
B4=5.85
B5=6.61
B6=7.78

U02++ sp oth/un rt ? U K1=2.64 1962SAb (10267) 298

U02++ sp NaClO₄ 35?°C 0.30M U K1=3.50 1961NPb (10268) 299

U02++ sp oth/un 25°C var U K1=2.31 1961SAd (10269) 300

OH- HL Hydroxide (57)
Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ cal NaCl 25°C 0.0 C IH 2004CDB (12371) 301
DH(p,q): pUO₂+qH₂O=(UO₂)p(OH)q+qH. From data for 0.1, 0.51 and 1.02 m NaCl
DH(1,1)=40.7 kJ m⁻¹, DH(2,2)=47.8, DH(3,4)=98.9, DH(3,5)=119.5, DH(3,7)=177

U02++ cal NaClO₄ 25°C 1.0M C IH 2004CDB (12372) 302
*B(1,1)=-5.2

*B(2,2)=-5.95
 *B(3,5)=-16.2

*B(p,q) and DH(p,q) for the reaction: pUO₂+qH₂O=(UO₂)_p(OH)_q+qH.
 DH(1,1)=37 kJ mol⁻¹, DH(2,2)=44.2, DH(3,5)=106.6. Medium: 1.0 m NaClO₄.

U02++ gl oth/un 25°C 0.0 C I 2004GMb (12373) 303
 *B(2,2)=-5.76
 *B(3,4)=-11.82,
 *B(3,5)=-15.89
 *B(3,7)=-29.26

Calculated from data for 0.5-3.0 m LiCl. For I=1.0 m LiCl, *B(2,2)=-6.32, *B(3,4)=-12.53, *B(3,5)=-17.45, *B(3,7)=-29.99. Data for 0.05-0.20 M CaCl₂

U02++ gl NaNO₃ 25°C 0.0 C I 2004GMb (12374) 304
 *K1=-5.19

Calculated from data for 0.10-1.00 M NaNO₃ and NaCl.

U02++ gl NaClO₄ 25°C 0.10M C I 2002BRa (12375) 305
 *K1=-5.01
 *B(2,2)=-5.98
 *B(3,4)=-12.39
 *B(3,5)=-16.36

0.1 M KCl: *K1=-5.17, *B(2,2)=-5.86, *B(3,4)=-12.00, *B(3,5)=-16.09. At I=0, *K1=-5.1, *B(2,2)=-5.5, *B(3,4)=-11.7, *B(3,5)=-15.4, *B(4,7)=-22.1

U02++ gl NaCl 25°C 0.10M C I 2002DGa (12376) 306
 *K1=-5.45
 *B(2,2)=-5.98,
 *B(3,4)=-12.22
 *B(3,5)=-16.55

Data for 0.10-4.5 M NaCl and 0.10-1.0 M NaNO₃. *B(3,7)=-29.68. At I=0.0 M:
 *K1=-5.19, *B(2,2)=-5.76, *B(3,4)=-11.82, *B(3,5)=-15.89, *B(3,7)=-29.26

U02++ gl oth/un 25°C 0.10M C I 2000CBa (12377) 307
 *K1=-5.91
 *B(2,2)=-6.65
 *B(3,4)=-13.31
 *B(3,5)=-18.27

Medium: 0.10 M Na₂SO₄. *B(4,7)=-24.80, *B(5,8)=-26.64. *B(m,n): mUO₂+nH₂O=(UO₂)_m(OH)_n+nH. At I=1.0, *B(2,2)=-7.25, *B(3,4)=-13.53, *B(4,7)=-26.82.

U02++ sol NaClO₄ 25°C 5.0M M 1998DGa (12378) 308
 K_s(UO₃.2H₂O+2H=UO₂+3H₂O)=5.14

Medium: 0.5 m NaClO₄. Using Pitzer theory, at I=0, K_s=4.7. In 3 m NaCl, K_s(Na_{0.33}UO₃.16.2H₂O+2.33H=UO₂+3.16H₂O+0.33Na)=7.95. At I=0, K_s=13.

U02++ gl R4N.X 25°C 0.10M C 1995PNa (12379) 309
 *B(2,2)=-5.51
 *B(3,5)=-15.3
 *B(3,7)=-27.77

*B(3,8)=-37.65
*B(3,10)=-62.14. *B(m,n)=mUO₂+nH₂O=(UO₂)m(OH)n+nH

UO₂++ gl NaClO₄ 25°C 3.00M C 1993FSa (12380) 310
*B(2,2)=-5.98
*B(3,5)=-16.23

UO₂++ sol NaClO₄ 24°C 0.10M U 1993MKb (12381) 311
K(UO₃.2H₂O(s)=UO₂+2OH)=-22.19. By pH titration.

UO₂++ sp NaClO₄ 24°C 0.10M C 1993MKc (12382) 312
*B(2,2)=-5.97.

Method: laser-induced fluorescence spectroscopy.

UO₂++ sol NaClO₄ 25°C 0.50M C 1992SBa (12383) 313
*B(3,7)=-33.32
*B(1,3)=-20.18
*K_s(UO₂(OH)₂+2H=UO₂+2H₂O)=6.23

Solubility constants for crystalline Schoepite. Also at I=0

UO₂++ gl NaClO₄ 25°C 0.50M C H 1991GLa (12384) 314
*B(2,2)=-6.07
*B(3,5)=-16.40

UO₂++ sol none 100°C dil C T 1988PPd (12385) 315
K_{s4}=-9.47

Data for 100-300 C at 50 MPa H₂. K_{s4}: UO₂+2H₂O=U(OH)₄.

UO₂++ sp NaNO₃ 20°C 0.5M U 1983DBC (12386) 316
K(UO₂(OH)+H)=3.34

UO₂++ con none 23°C 0.0 C 1983SGe (12387) 317
*K₁=-5.2

UO₂++ gl NaNO₃ 25°C 0.50M C I 1982MSh (12388) 318
*B(2,2)=-6.01
*B(3,4)=-12.24

Data for 0.50-3.0 M NaNO₃. At I=1.0 M, *B(2,2)=-6.07, *B(3,4)=-12.31.

At I=3.0 M, *B(2,2)=-6.13, *B(3,5)=-16.65.

UO₂++ sol oth/un 25°C var C T H 1981TCc (12389) 319
K(UO₂+2H₂O+OH=U(OH)₅)=-5.75

Data for 25-300 C. Solubility of UO₂ at pOH=1.5 and 2.5

DH(K)=-0.6 kJ mol⁻¹.

UO₂++ EMF none 25°C 0.0 U T H 1980LTb (12390) 320
*K₁=-5.8. *B₂=-12
*B(2,2)=-5.6
*B(3,5)=-15.6
*B(3,7)=-31

100 C: *K1=-4.2, *B2=-10, *B(2,2)-4.4, *B(3,5)=-12.4, *B(3,7)=-23; 200 C:
-2.9, -8, -3.8, -10.7, -16. Evaluated data

U02++ gl NaNO₃ 25°C 0.20M M 1980PDc (12391) 321
*K(UO₂)=-4.80

*K: UO₂+H₂O=UO₂(OH)+H

U02++ gl NaClO₄ 25°C 3.0M C 1979CFa (12392) 322
*B(2,2)=-6.0
*B(3,5)=-16.6

U02++ gl NaClO₄ 25°C 0.50M C 1979LPC (12393) 323
*B(1,2)=-3.81
*B(2,2)=-6.03
*B(3,4)=-13.17
*B(3,5)=-16.78

*B(q,p)=K(qUO₂(2+)+pH₂O = (UO₂)q(OH)p(2q-p)++pH+)

U02++ gl KCl 25°C 3.00M U 1979MIb (12394) 324
*B(2,2)=-6.30
*B(2,3)=-11.2
*B(4,6)=-17.85

U02++ gl R4N.X 25°C 0.10M U 1979SAC (12395) 325
*B(2,2)=-5.63
*B(3,5)=-15.87

Medium: 0.10 M Et₄NClO₄.

U02++ gl KN₃ 25°C 0.10M U 1979SDa (12396) 326
*K1=-5.50, *B(2,2)=-5.89
*B(3,4)=-12.31
*B(3,5)=-16.46
*B(4,7)=-22.76

U02++ con oth/un 25°C 0.00 U 1977VBA (12397) 327
*K(UO₂=UO₂(OH)+H)=-4.20

U02++ gl KN₃ 25°C 1.00M U 1974CGb (12398) 328
*B(2,2)=-6.02
*B(3,4)=-12.48
*B(3,5)=-16.22

U02++ sp NaClO₄ 22°C 0.50M U 1974MAb (12399) 329
*B(2,2)=-6.0
*K'=-7.0

Medium: 0.5-2 M. K': (UO₂)₂(OH)₂ + UO₂ + 2H₂O = (UO₂)₃(OH)₄ + 2H

U02++ gl NaClO₄ 25°C 3.00M U 1972MAa (12400) 330
*B(2,2)=-6.61
*B(3,4)=-14.28

					*B(3,5)=-18.16
Medium:	80% w/w D2O/H2O, 3 M NaClO4				
U02++	gl	NaClO4	25°C	3.00M	U
					1972MAa (12401) 331
					*B(2,2)=-6.80
					*B(3,4)=-14.00
					*B(3,5)=-18.63
Medium:	D2O, 3 M NaClO4				
U02++	gl	NaClO4	25°C	3.0M	U
					1972MAa (12402) 332
					*B(2,2)=-6.17
					*B(3,4)=-12.92
					*B(3,5)=-17.04
U02++	kin	NaNO3	3°C	0.50M	U
					1970FWa (12403) 333
					K((UO2)2(OH)2+H)=2.9
					K((UO2)2(OH)+H)=1.6
U02++	sol	oth/un	25°C		U
					1969TSa (12404) 334
					Kso((NH4+)2U7022(?))=-14.3
U02++	gl	KN03	25°C	0.50M	U
					1969VOa (12405) 335
					*K1=ca.-5.7
					*B(2,2)=-5.95
					*B(3,5)=-16.36
U02++	cal	NaClO4	25°C	3.00M	U H
					1968ASb (12406) 336
					*B(2,2)=-6.02
					*B(5,3)=-16.54
DH(*2,2))=39.7 kJ mol-1, DS=18.0 J K-1 mol-1. DH(*3,5))=97.9; DS=26.1					
U02++	gl	NaClO4	25°C	0.20M	U
					19680Ca (12407) 337
					*B(2,2)=-5.92
					*B(5,3)=-16.16
U02++	gl	oth/un	25°C	5.0M	U I
					1968SFb (12408) 338
					*K1=-5.53
					*B(2,2)=-6.52
					*B(3,5)=-17.76
Medium: MgNO3. At I=3.0: *K1=-5.38, *B(2,2)=-6.34, *B(3,5)=-17.37					
U02++	sol	none	20°C	0.0	M T
					1967GKc (12409) 339
					Kso(UO2(OH)2)=-21.12
Kso=-20.87(25 C), -20.63(30 C), -20.35(40 C), -20.06(50 C)					
U02++	sol	NaClO4	25°C	1.00M	U
					1966BTb (12410) 340
					Ks=-23.92
Ks: K(Na0.14UO2(OH)2.14(s)=0.14Na + 2.14UO2(OH)2)					
U02++	sp	NaClO4	25°C	2.00M	U H
					1965NBa (12411) 341

K=1.2

Medium: 2 M (H,Li)ClO4. K: UO2++ + UO2+ = U2O4+++. DH=-7.9 kJ mol-1, DS=-4

UO2++ gl NaClO4 25°C 0.10M U I 1964BSf (12412) 342
*B(2,2)=-6.09
*B(2,1)=-2.5

By spectrophotometry: *B(2,2)=-6.28, *B(1,2)=-1.9. In 30% EtOH, 0.1 M NaClO4
*B(2,2)=-6.2; -4.8(50 %)

UO2++ cal oth/un 25°C 6.00M U H 1964C0c (12413) 343
DH(UO3(s)+2H=UO2+H2O)=-77.8 kJ mol-1(alpha-UO3), -74.8(Beta-UO3), -71.2
(gamma-UO3), -78.0(epsilon-UO3) plus others. Medium: 6 M HNO3

UO2++ gl NaCl 25°C 3.0M U I 1963DHa (12414) 344
*B(2,2)=-6.64
*B(3,5)=-18.07
*B(3,4)=-12.54
*B(4,6)=-20.0

*B(4,7)=-24.9. Also quinhydrone electrode. In 1 M KNO3: *B(2,1)=-4.2,
*B(2,2)=-5.96, *B(3,5)=-16.21, *B(4,3)=-12.8

UO2++ gl NaClO4 25°C 3.0M U I 1963HRa (12415) 345
*B(2,2)=-6.04
*B(3,5)=-16.53
*B(3,4) ca.-13.6
*B(4,6) < -19.2

*K1=<-5.9 (range with up to 0.1 M UO2++); *B(m,n): K(mM+nH2O=Mm(OH)n+nH)
Also in 3 M Mg(ClO4)2 *B(2,1)=-3.81, *B(2,2)=-6.25, *B(3,5)=-17.18 etc.

UO2++ gl oth/un 27°C var U 1963PSb (12416) 346
*K1=-4.59

UO2++ gl NaClO4 25°C 1.0M U 1963RJa (12417) 347
*B(2,2)=-5.94
*B(3,5)=-16.41

UO2++ gl KNO3 25°C 0.50M U T H 1962BMb (12418) 348
*K1=-5.7
*B(2,2)=-5.92
*B(3,5)=-16.22

*B(m,n): K(mM+nH2O=Mm(OH)n+nH); DH(*K1)=46 kJ mol-1, DS=-46; DH(*B(2,2))=42.7
DS=30; DH(3,5)=105.0, DS=42 J K-1 mol-1. 94 C: *K1=-4.19, *B(2,2)=-4.51

UO2++ gl oth/un ? 1.0M U 1962NPa (12419) 349
*B(2,2)=-6.1

UO2++ ix NaNO3 ? 0.50M U 1962NPa (12420) 350
*B(3,5)=-16

UO2++ sol oth/un 25°C dil U 1962PPa (12421) 351

Kso=-19.82

U02++ sol NaClO4 20°C 1.0M U 1962RJa (12422) 352
*B(2,2)=-5.96
*B(3,5)=-16.74

U02++ gl NaCl 25°C 1.0M U 1962RJa (12423) 353
*B(2,2)=-6.17
*B(3,4)=-12.33
*B(3,5)=-17.00

U02++ cal NaClO4 25°C 3.0M U H 1962SCe (12424) 354
DH(*B(2,2))=39.5 kJ mol-1, DH(*B(3,4))=75, DH(*B(3,5))=105, DH(*B(4,6))=100,
DS(*B(2,2))=17, DS(*B(3,5))=33. *B(m,n): mM+nH2O=Mm(OH)n+nH

U02++ gl oth/un 25°C 1.50M U 1961PEa (12425) 355
*B(2,2)=-8.17
*B(3,4)=-16.20
*B(4,6)=-24.51
*B(5,8)=-32.14

Medium: Na2SO4

U02++ sol R4N.X rt 0.20M U 1960BKa (12426) 356
Kso(U02(OH)2)=-21.74

Medium: NH4NO3

U02++ sol none 20°C 0.0 U 1960BRb (12427) 357
Kso(U02(OH)2)=-17.22
K(UO2(OH)2(s)=UO2OH+OH)=-11.89
K(UO2)2(s)=UO2(OH)2)=-5.89

U02++ gl KN03 25°C 0.10M U 1960GRa (12428) 358
*K1=-6.10
*B(2,2)=-5.84
*B(2,2)=-5.83
*B(4,6)=-17.6

U02++ gl NaClO4 25°C 3.0M U 1960HIa (12429) 359
*B(2,2)=-6.03
*B(3,4)=-13.20
*B(3,5)=-16.55
*B(4,6)=-19.42

U02++ dis oth/un ? var U 196000a (12430) 360
Kso(U02(OH)2)=-23.74

U02++ dis NaClO4 20°C 0.10M U K1=9.2 B2=17.2 1960STc (12431) 361
B3=25.5

U02++ gl NaClO4 25°C 3.0M U I 1959HSa (12432) 362

						*B(2,1)=-3.68 *B(2,2)=-6.31 *B(3,4)=-12.6 ?
*B(m,n): K(mM+nH2O=Mm(OH)n+nH). Method: also quinhydrone electrode						
U02++	sol	none	20°C	0.0	U	1958BRb (12433) 363 Kso(U02(OH)2)=-17.22(?) Ks(U02(OH)2=U02(OH)+OH)=-10.89
U02++	gl	NaCl	25°C	0.16M	U	1958LDa (12434) 364 *B(n+1,2n)=0.30-6.42n
*B(n+1,2n): K((n+1)M+2nH2O=M(n+1)(OH)2n+2nH)						
U02++	gl	oth/un	25°C	0.35M	U	1957HWa (12435) 365 *K1=-5.40 *B(2,2)=-5.82
Medium: 0.347 M Ba(ClO4)2; *B(2,2): K(2U02+2H2O=(U02)2(OH)2+2H)						
U02++	gl	oth/un	25°C	0.34M	U T H	1957HWa (12436) 366 *K1=-5.82 *B(2,2)=-6.15
Medium: 0.347 M Ba(ClO4)2; *B(2,2): K(2U02+2H2O=(U02)2(OH)2+2H); DH(*K1)=87.0 DH(*B(2,2)=28.0, DS(*K1)=180, DS(*B(2,2))=-25; *K1=-5.10(40 C), *B(2,2)=-5.92						
U02++	oth	none	25°C	0.0	U	1956DPa (12437) 367 *Kso(U03(s)+2H)=14.74 *Kso(U02(OH)2(s)+2H)=4.97 *Kso(U02(OH)2H2O(s)+2H)=5.60
*Kso(U03(s)+2H=U02+H2O); *Kso(U02(OH)2(s)+2H=U02+2H2O); *Kso(U02(OH)2H2O(s)+2H=U02+3H2O); method: combination of thermodynamic data						
U02++	gl	none	25°C	0.0	U	19560Ba (12438) 368 *B(2,2)=-5.06 *B(4,2)=-1.26
*B(m,n): K(mM+nH2O=Mm(OH)n+nH)						
U02++	sol	none	25°C	0.0	U	1955GLc (12439) 369 *Kso=6.04 *Ks(U02(OH)2(s)+H)=1.90 Ks(U02(OH)2(s)+OH)=-3.60 Ks(U02(OH)2(s)+2OH)=-3.77
*K1=-4.14						
U02++	sp	oth/un	?	var	U	1955KTa (12440) 370 *K1=-4.19
U02++	dis	NaClO4	25°C	0.10M	U	1955RYa (12441) 371 *K1=-4.2 *K2=-5.20

U02++ gl NaClO₄ 20°C 1.0M U 1954AHa (12442) 372
 *B(2,2)=-6.05
 *B(n+1,2n)=0.30-6.35n
 *B(m,n)(mM+nH₂O=Mm(OH)n+nH). Method: also quinhydrone electrode

U02++ gl oth/un 15°C 0.06M U I 1954FAa (12443) 373
 *B(2,2)=-5.72
 Medium: Ba(NO₃)₂. In 0.6 M Ba(NO₃)₂ *B(2,2)=-5.97

U02++ EMF NaClO₄ 20°C 1.0M U 1949AHa (12444) 374
 *K1=-4.70

U02++ gl NaClO₄ ? 0.15M U I 1949SUa (12445) 375
 *B(2,2)=-5.99
 *B(3,4)=-13.29
 *K(U308)=-3.55
 *B(m,n): K(mM+nH₂O=Mm(OH)n+nH); *K(U307(OH)n)=-3.55(n=0), -6.5(n=1), -7.4(n=2)
 -11.0(n=3), -11.4(n=4). Method: freezing point and spectrophotometry

U02++ sp oth/un ? var U 1947GUb (12446) 376
 *K1=-4.50
 *B(2,2)=-4.95

U02++ EMF none 25°C 0.0 U 1947HKa (12447) 377
 *K1(U02+H₂O=UO₂OH+H)=-4.09
 Method: quinhydrone electrode

U02++ gl oth/un 20°C var U 1947MLa (12448) 378
 *B(2,2)=-5.87

U02++ EMF none 25°C 0.0 U 1942HEa (12449) 379
 *K1=-4.3
 Method: quinhydrone electrode

O2-- H₂L Peroxide CAS 7772-84-1 (2813)
 Peroxide; -0.0-

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	sp	NaClO ₄	RT	0.70M	C			K1=31.95 K _{1eff} =5.30 (pH 5.0)	1992DBa (12723)	380
Medium: 0.7 M LiClO ₄ , pH 5.0. Additional method: DPP.										
U02++	sp	oth/un	0°C	?	U	M		B((UO ₂) ₂ L ₂ (SO ₄) ₂)=71.8 B((UO ₂) ₂ L(SO ₄) ₃)=40.3	1972GSF (12724)	381
U02++	sp	oth/un	?	var	U	M		K(UO ₂ +3H ₂ L=6H+UO ₂ L ₃)=-38.4 B ₃ =71.7	1968G0a (12725)	382

Equilibrium constants for mixed UO₂-L-citrate complexes

UO₂++ sp oth/un ? var U M 1968GPe (12726) 383

$$K = -3.77$$

$$K' = -1.15$$

$$K'' = -16$$

$$K((UO_2)_2YL(OH)+H) = 8$$

H4Y=EDTA. K: (UO₂)₂Y+H₂L=2H+(UO₂)₂YL; K': 2UO₂Y+H₂L=2H+(UO₂)₂Y₂L;
K'': (UO₂)₂Y₂L+H₂L=2H+(UO₂)₂YL₂+Y

UO₂++ sp oth/un ? var U M 1968GSg (12727) 384

$$B_3 = 72.95$$

Equilibrium constants given for reactions involving (UO₂)₂LF₅,
(UO₂)₂L₂F₅ and (UO₂)₂L₃F₂

UO₂++ sol oth/un ? var U K₁=32.04 B₂=60.15 1968M0c (12728) 385
K_s(UO₂L(H₂O)₄(s)+2H)=-2.0

Other solubilities also given

UO₂++ sp oth/un ? 1.0M U 1965MAb (12729) 386
K(UO₂L₃+H)=11.06
K=-18.4

Medium: LiCl. K: 2HUO₂L₃+4H₂O=(UO₂)₂L₃+3H₂L+4OH

UO₂++ sp KCl ? 1.0M U M 1965SMa (12730) 387
K=-11.1

K: UO₂(CO₃)₃+H₂L+2OH=UO₂L(CO₃)₂+CO₃+2H₂O

UO₂++ sol none 25°C 0.0 U 1964PCa (12731) 388
K(UO₂L(s)+H=UO₂LH)=-1.44
K(UO₂L(s)+2H=UOL+H₂O)=0.18
K(UO₂L(s)+OH=UO₂LOH)=-1.96
K(UO₂L(s)+2OH=UO₃L+H₂O)=-0.05

UO₂++ gl oth/un ? var U 1960GPa (12732) 389
K(UO₂L₃+H)=ca.12.5

UO₂++ sol oth/un 20°C ? U 1960MAa (12733) 390
K(UO₂L(H₂O)₄(s)+2H=UO₂+H₂L+4H₂O)=-2.86

UO₂++ sol oth/un 78°C var U T 1959GJa (12734) 391
K(UO₂L(s)=UO₂L)=-4.0
K(UO₂L(s)+2H=UO₂+H₂L)=-1.44
K(UO₂L(s)+2H=UO₂+H₂L)=-1.44(78-114 °C)

UO₂++ sp KN03 ? 0.40M U I M 1959K0b (12735) 392
K(UO₂(CO₃)₃+H₂L=UO₂(CO₃)₂HL+HC03)=2.0. At I=0 corr. K=2.2

UO₂++ sp oth/un 0°C var U M 1959KPb (12736) 393
K(UO₂(CO₃)₂L+H)=10.6

U02++	sp	oth/un	?	var	U	1958GPa (12737) 394 K(2U02+2H2L+H2O=H2U205L2)=-2.7 K(HU205L2+H)=ca.7 K(U205L2+H)=ca.10
U02++	sp	oth/un	0°C	var	U	1958GPa (12738) 395 K(U02L3+H)=12.3
U02++	gl	none	25°C	0.0	U	1958GTa (12739) 396 Ks(U02L(s)+H=U02LH)=-1.44? Ks(U02L(s)+OH=U02LOH)=-1.96?
I=0 corr. Complexes may be polymers						
U02++	sp	oth/un	?	var	U	1957MAc (12740) 397 K1eff=4.71
Medium: 0.05-1.5 M Na2CO3						*****
P04---		H3L	Phosphate		CAS 7664-38-2 (176)	
Phosphate;						
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values Reference ExptNo
U02++	con	non-aq	?	100%	U M	1992RSa (13357) 398 K(U02A+H2L)=4.15
Medium: 1% DMSO+99% MeCN. A=N,N'-disalicylidene-1,2-benzenediamine (+others)						
U02++	sol	NaClO4	25°C	0.50M	C	K1=11.29 1992SBa (13358) 399 K(U02+HL)=11.29 Kso((U02)3L2)=-48.48
Also extrapolated values to I=0						
U02++	dis	NaClO4	25°C	1.00M	U TIH	1991MAb (13359) 400 K(U02+H2L)=2.94 K(U02+2H2L)=4.93
DH(U02+H2L)=8.0 kJ mol-1, DS=84 J K-1 mol-1; DH(U02+2H2L)=16.3; DS=151 at 1.0 M NaClO4, 25 C						
U02++	sp	NaClO4	20°C	0.05M	C	K1=3.26 1989RAb (13360) 401
Medium: 0.05 M NaClO4/HC1O4.						
U02++	dis	oth/un	25°C	0.20M	C	1987EBa (13361) 402 K(U02+H3L=(U02)H2L+H)=1.70 K(U02+2H3L=U02(H2L)2+2H)=1.40 K(U02+2H3L=U02(H2L)H3L+H)=1.78
Medium: HC1O4/H3P04. Distribution of 230U and 233U into benzene/HDEHP. K(U02+3H3L=U02(H2L)(H3L)2+H)=3.57; K(U02+3H3L=U02(H2L)3+3H)=2.04						
U02++	sol	none	25°C	0.0	U	1983MPa (13362) 403 K(U02+H3L=U02H2L+H)=1.50

$$\begin{aligned} K(UO_2 + H_3L) &= 1.30 \\ K(UO_2 + 2H_3L = UO_2H_4L_2 + 2H) &= 1.30 \\ K(UO_2 + 3H_3L = UO_2H_7L_3 + 2H) &= 2.30 \end{aligned}$$

U02++ EMF none 25°C 0.0 U T H 1980LTb (13363) 404

$$\begin{aligned} K(UO_2 + HP_4 + H) &= 10.2 \\ K(UO_2 + 2HP_4 + 2H) &= 19.9 \\ K(UO_2 + 3HP_4 + 3H) &= 28.8 \\ K(UO_2 + HP_4) &< 8 \end{aligned}$$

$K(UO_2 + 2HP_4) < 19$. At 100 C: values are 11, 19, 27, <9, <19; At 200 C: values: 12, 20, 28, ,10, <22. Evaluated data

U02++ oth none ? 0.0 U 1969M0c (13364) 405
 $K(UO_2 + HL) = 8.43$

U02++ sol KN03 25°C 0.50M U I 1967MSH (13365) 406
 $K(UO_2 + HL) = 7.18$
 $K(UO_2 + 2HL) = 17.30$

At I=0 corr: $K(UO_2 + HL) = 8.43$, $K(UO_2 + 2HL) = 18.57$

Also many solubility products

U02++ sol NaNO3 20°C 0.32M U 1965VPa (13366) 407
 $K(UO_2HL(s) = UO_2 + HL) = -12.17$
 $K_{so}((UO_2)_3L_2) = -49.7$
 $K_{so}(NaUO_2L) = -24.21$
 $K_{so}(KUO_2L) = -25.50$
 $K_{so}(RbUO_2L) = -25.72$, $K_{so}(CsUO_2L) = -25.41$, $K_{so}((NH_4)UO_2L) = -26.23$

U02++ sol oth/un 25°C dil U 1964MZA (13367) 408
 $K_{so}(LiUO_2L) = -25.6$
 $K_{so}(NaUO_2L) = -28.2$
 $K_{so}(KUO_2L) = -23.1$
 $K_{so}(RbUO_2L) = -27.0$

U02++ sol oth/un 20°C dil U 1961CAa (13368) 409
 $K_{so}((UO_2)_3L_2) = -49.1$

U02++ sol oth/un 25°C var U 1961KAb (13369) 410
 $K_{so}((UO_2)_3L_2) = -46.68$

U02++ sol oth/un 25°C dil U M 1961KZa (13370) 411
 $K_{so}(NH_4(UO_2)L_3(H_2O)_3) = -25.44$

U02++ sp NaClO4 25°C 1.00M U 1958BAa (13371) 412
 $K(UO_2 + H_3L) = 0.76$
 $B((UO_2)(H_3L)H-1) = 0.72$
 $B((UO_2)(H_3L)2H-2) = 0.41$
 $B((UO_2)(H_3L)2H-1) = 1.33$

U02++ sp NaClO4 25°C 1.07M U 1957THb (13372) 413

$$\begin{aligned}B((UO_2)H-1(H_3L)) &= 1.19 \\B((UO_2)H-2(H_3L)_2) &= 1.34 \\B((UO_2)H-2(H_3L)_3) &= 1.01\end{aligned}$$

Also by distribution

UO ₂ ++	sol	oth/un	20°C	var	U	M	1956CSd (13373)	414
							K _{so} (UO ₂ (NH ₄)L) = -26.36	
							K _{so} (UO ₂ KL) = -23.11	
							K _s (UO ₂ HL = UO ₂ + HL) = -10.67	

UO ₂ ++	gl	none	20°C	0.0	U		1955MAa (13374)	415
							K(UO ₂ + H ₂ L) = 3.0	
							K(UO ₂ H ₂ L + H ₂ L) = 2.5	
							K(UO ₂ (H ₂ L) ₂ + H ₂ L) = 1.9	
							K(UO ₂ (H ₂ L) ₂ HL + H) = 0.5	

Also by anion exchange. K(UO₂H₂L(HL)₂+H)=2.3, K(UO₂(HL)₃+H)=1.4, K(UO₂(H₃L)(H₂L)+H)=1.5 and others

UO ₂ ++	sol	NaClO ₄	25°C	1.00M	U		1953SBa (13375)	416
							K(UO ₂ +2H ₃ L = UO ₂ (H ₂ L) ₂ +2H) = 1.18	
							K(UO ₂ HL(s) + H ₃ L = UO ₂ (H ₂ L) ₂) = -1.7	

Medium: HClO₄. Plus many other equilibria and solubility data

P207---- H4L Pyrophosphate CAS 2466-09-3 (198)
Diphosphate; from (HO)₂PO.O.PO(OH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	NaClO ₄	?	0.10M	U			1973DVb (13667)	417	
								K(5UO ₂ +H ₂ L) = -1.46		

SCN- HL Thiocyanate CAS 463-56-9 (106)
Thiocyanate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	con	non-aq	?	100%	U	M		1992RSa (15314)	418	
								K(UO ₂ A+L) = 1.70		

Medium: 1% DMSO+99% MeCN. A=N,N'-disalicylidene-1,2-benzenediamine (+others)

UO₂++ cal NaClO₄ 25°C 1.0M U H 1971AKb (15315) 419
DH(K1) = -3.2 kJ mol⁻¹, DS = 3.5 J K⁻¹ mol⁻¹, DH(K2) = -5.7, DS = -19.8,
DH(K3) = 2.9, DS = 18

UO₂++ sp NaClO₄ 20°C 1.0M U K1=0.74 B2=0.97 1970SWa (15316) 420
K3=-0.22 to -0.15

UO₂++ sp NaClO₄ 20°C 1.0M U K1=0.73 B2=0.96 1968SWa (15317) 421
B3=0.8

U02++ sp NaNO₃ 23°C 4.0M U I K1=0.71 B2=0.72 1964VMa (15318) 422
At I=2.5 M: K1=0.72, B2=0.70; at 0 corr: K1=1.5, B2=1.9

U02++ ix NaClO₄ 32°C 1.0M U K1=-1.3 B2=1.05 1961BTa (15319) 423
B3=1.08

U02++ sp alc/w 25°C 20% U T H K1=1.00 1957BD_b (15320) 424
Medium: 20% w/w MeOH/H₂O; DH(K1)=4.81 kJ mol⁻¹, DS=35.6 J K⁻¹ mol⁻¹(25 °C).
K1=0.99(15 °C), 1.05(35 °C), 1.07(45 °C)

U02++ sp none 25°C 0.0 U T K1=0.93 1957DMa (15321) 425

U02++ sp NaClO₄ 20°C 1.0M U T K1=0.76 B2=0.74 1949AHa (15322) 426
K3=0.44

S03-- H2L Sulfite CAS 7782-99-2 (801)
Sulfite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl oth/un 23°C var U I K1=5.85 1967Z0c (15481) 427
By spec., 0.1 NH₄ClO₄: K1=6.01

U02++ sol NaCl 23°C 1.0M U K1=5.26? B2=9.17 1967Z0c (15482) 428

U02++ sol oth/un ? 2.0M U B3(K3?)=1.01 1967Z0d (15483) 429

U02++ sol oth/un 25°C var U B2=7.10 1959KK_b (15484) 430
Kso(U02L)=-8.59

S04-- H2L Sulfate CAS 7664-93-9 (15)
Sulfate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaNO₃ 25°C 0.0 C I K1=3.32 B2= 4.26 2004GMB (16626) 431
B(1,1,1)=-2.30
B(2,2,1)=-2.64
B(3,4,1)=-8.45
B(3,5,1)=-13.58
Calc. from data for 0.25-1.0 M NaNO₃/Na₂SO₄, 0.5-1.5 M Na₂SO₄ and 0.25-1.0 M NaCl/Na₂SO₄. B(p,q,r): pU02+qH₂O+rSO₄=(U02)p(OH)q(SO₄)r+qH.

U02++ gl oth/un 25°C 0.10M C I M K1=1.92 B2= 2.90 2000CBa (16627) 432
B(2,2,2)=-2.17
B(3,4,3)=-6.60
B(4,7,4)=-15.85
B(5,8,4)=-17.69

In 0.10 M Na₂SO₄. Data for I=1.03 and 1.57 m. B(p,q,r): pU02+qH₂O+rSO₄=

(U02)p(OH)q(SO₄)r+qH. I=0: K₁=3.15, B₂=4.14, B(2,2,2)=-0.64, B(3,4,3)=-5.9

U02++ sp oth/un 20°C 1.0M C I K₁=3.14 B₂= 4.20 1992BTa (16628) 433
Method: Raman spectroscopy. Calculated from data for 0.1-0.4 M Na₂SO₄.

U02++ con non-aq ? 100% U M 1992RSa (16629) 434
K(UO₂A+HL)=1.70

Medium: 1% DMSO+99% MeCN. A=N,N'-disalicylidene-1,2-benzenediamine (+others)

U02++ cal none 25°C 0.0 U TIH K₁=3.185 1990THa (16630) 435
Data for T=10-55 C: K₁=3.009 (10 C), 3.365 (40 C), 3.548 (55 C). DH(K₁)=
20.2 kJ mol⁻¹.

U02++ oth oth/un 25°C dil C I 1990VGa (16631) 436
K((UO₂)₂(OH)₂+SO₄)=3.01

Analysis of literature data. Value is for 0.025 M (UO₂)SO₄ solution.
In 1.5 M Na₂SO₄, K((UO₂)₂(OH)₂+SO₄)=3.95.

U02++ sp NaClO₄ 20°C 0.05M C K₁=2.26 1989RAb (16632) 437
Medium: 0.05 M NaClO₄/HClO₄.

U02++ EMF none 25°C 0.0 U T H K₁=2.9 1980LTb (16633) 438
60 C: K₁=3.4; 100 C: K₁=4.4; 150 C: K₁=6; 200 C: K₁=7. Evaluated data

U02++ ix NaClO₄ 25°C 0.20M C K₁=0.59 1978SGg (16634) 439
Method: polarography. Medium: 0.20 M HClO₄.

U02++ dis NaClO₄ 25°C 2.00M U K₁=0.88 B₂=1.23 1976Pra (16635) 440

U02++ con diox/w 0°C 82% U I K₁=6.03 1974EJa (16636) 441
K(triple ion)=3.45

Medium: w/w dioxan/H₂O. In 20% dioxan: K₁=2.58; 45%: 3.48; 70%: 5.90

U02++ cal NaClO₄ 25°C 1.0M U H K₁=1.81 B₂=2.76 1971AKb (16637) 442
DH(K₁)=18.2 kJ mol⁻¹, DH(K₂)=16.9; DS(K₁)=96 J K⁻¹ mol⁻¹, DS(K₂)=75

U02++ cal none 25°C 0.0 U H 1971BLc (16638) 443
DH(K₁)=20.8 kJ mol⁻¹, DS(K₁)=121.6 J K⁻¹ mol⁻¹

U02++ EMF KCl 25°C var U T K₁=2.93 1971NOb (16639) 444
K₁=3.2(50 C), 3.68(70 C), 4.13(90 C), 4.37(100 C), 4.99(125 C), 5.63(150 C)

U02++ oth oth/un 50°C 0.0 U T H K₁=3.43 B₂=4.60 1967WAa (16640) 445
Method: membrane equil Na-UO₂. K₁=3.14(25 C), 3.26(35 C); B₂=4.21(25 C),
4.36(35 C). DH(K₁)=21.3, DH(B₂)=29.3 kJ mol⁻¹

U02++ sol R4N.X 17°C 8.0M U M 1963KGa (16641) 446
K_s((NH₄)₂UO₂L₂(s)+L)=-0.97

Medium: NH₄NO₃. K_s: (NH₄)₂UO₂L₂(s)+L=2NH₄+UO₂L₃

U02++ gl oth/un 25°C var U K1=3.85 1963PSb (16642) 447
 K(UO₂(OH)+L)=3.32

U02++ ix NaClO₄ 32°C 1.0M U K1=1.63 B2=3.78 1961BTa (16643) 448

U02++ sp alc/w 25°C 20% U K1=3.88 B2=5.48 1961MMc (16644) 449
 Medium: 20% MeOH

U02++ sol oth/un 25°C 0.0 U T H 1960LSa (16645) 450
 25 to 250 C: K1, K2 and DH(K1), DH(K2) as functions of T. K1=2.72+(0.02939(
 t-25))+.000323(T-25)²log(e) etc. DH(K1)=20 kJ mol⁻¹(25 C) to 263(250 C)

U02++ sp NaClO₄ 25°C 1.0M U K1=1.81 B2=2.29 1960MAb (16646) 451

U02++ vlt oth/un 25°C var U K1=-0.9 1959EKa (16647) 452
 Metal ion possibly UO₂₊

U02++ dis oth/un 25°C 1.0M U I K1=1.53 B2=2.31 1958ALa (16648) 453
 B3 < 2.1
 At I=0 corr. K1=2.76, K2=0.78. By quinhydrone elec. K1=2.03, K2=0.85,
 B3=-0.38. By spec. K1=2.98, K2=0.90

U02++ dis oth/un 25°C 1.0M U I K1=1.53 B2=2.31 1958ALa (16649) 454
 B3<2.1
 At I=0 corr. K1=2.76, K2=0.78

U02++ sp oth/un 25°C 0.0 U K1=2.96 B2=4 1957DMa (16650) 455

U02++ sol oth/un 250°C 0.0 U T M 1957GLa (16651) 456
 B((UO₂)L₂Ba)=9.3
 At 39 C: B((UO₂)L₂Ag₂)=6.18

U02++ dis NaClO₄ 25°C 2.0M U T H K1=1.88 B2=2.85 1954DPa (16652) 457
 20 C: K1=1.80, K2=0.96; 40 C: K1=1.98, K2=0.93. DH(K1)=9.6 kJ mol⁻¹, DS=67;
 DH(K2)=-4, DS=8

U02++ sp NaClO₄ 25°C 4.50M U K1=1.83 1953WDa (16653) 458

U02++ EMF NaClO₄ 20°C 1.0M U K1=1.70 B2=2.54 1951AHa (16654) 459
 K3=0.86
 B((UO₂)LA)=3.78
 B((UO₂)L₂A)=4.60
 Method: quinhydrone electrode. HA=etahnoic acid. By spec. K1=1.75, K2=0.90

U02++ sp NaClO₄ 25°C 4.50M U 1949BMa (16655) 460
 *K1=0.70

****S203-- H2L Thiosulfate CAS 73686-28-7 (177)
 Thiosulfate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	EMF	NaClO4	20°C	1.00M	U			K1=1.83 B2=3.13 B3=3.52	1972MPa (17655)	469

UO2++	oth	NaClO4	31°C	0.10M	U	M			1972SSb (17656)	470
								K(UO2+A+L)=5.06 K(UO2+B+L)=5.13 K(UO2+C+L)=5.18 K(UO2+D+L)=4.40		

HA=benzoic acid, HB=phenylethanoic acid, HC=phenylpropanoic acid,
HD=phenoxyethanoic acid, K(UO2+HE+L)=4.2, where H2E=4-hydroxybenzoic acid

UO2++	gl	NaClO4	31°C	0.10M	U		K1=2.61		1968RSa (17657)	471
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UO2++	sp	NaClO4	20°C	1.0M	U		K1=1.89	B2=2.97	1967MNd (17658)	472
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UO2++	vlt	NaNO3	?	1.0M	U		K1=2.4	B2=3.0	1962H0a (17659)	473
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CH606P2 H4L Medronic acid CAS 1984-15-2 (2384)

Methanediphosphonic acid; CH₂(PO₃H₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO2++	dis	NaClO4	25°C	0.1M	C				1993NAa (18296)	474
								K(UO2+H+H2L)=7.82 K(UO2+2H+H2L)=14.40 K(UO2+2H2L)=11.67		

CH7N06P2 H4L (6919)

Aminomethylenebis(phosphonic acid); NH₂.CH(PO₃H₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO2++	gl	R4N.X	25°C	0.10M	C		K1=25.9	B2=30.5	1994BRa (18315)	475
							K(UO2(OH)L+H)=10.4			
							K(UO2(OH)2L+H)=11.3			
							K(UO2L+H)=6.7			
							K(UO2HL+H)=6.0			

Medium: Me₄NNO₃. K(UO2L₂+H)=11.0, K(UO2HL₂+H)=10.3,

K(UO2H₂L₂+H)=6.6, K(UO2H₃L₂+H)=5.0

C2H02C13 HL Trichloroacetic CAS 76-03-9 (1205)

Trichloroethanoic acid; Cl₃C.CO₂H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO2++	dis	NaClO4	25°C	0.10M	U	H	K1=0.73		1988KCb (18337)	476
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DH=0.5 kJ mol⁻¹, DS=16 J K⁻¹ mol⁻¹

C2H2N2S3 H2L Bismuthiol I CAS 1072-71-5 (6261)
2,5-Dimercapto-1,3,4-thiadiazole;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	NaClO4	25°C	0.15M	U	I		K1=10.10	1977ZIa (18371)	477

C2H2O2C12 HL CAS 79-43-6 (1282)
Dichloroethanoic acid; Cl2CH.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	dis	NaClO4	25°C	0.10M	U	H		K1=1.14	1988KCb (18403)	478

DH=5.0 kJ mol-1, DS=39 J K-1 mol-1

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)
Ethanedioic acid; (COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	NaClO4	25°C	3.0M	C	I		K1=6.31 B2=11.21 B3=13.8 B(2,3)=18.5 B(2,5)=28.5	2002HSa (19117)	479

By application of SIT, at I=0, K1=7.41, B2=11.80, B3=13.96.

UO2++	gl	NaClO4	25°C	2.00M	C	I		K1=6.20 B2=11.21 B3=14.9	2000FIa (19118)	480
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At I=3.0 M, K1=6.39, B2=11.52, B3=15.2. By extrapolation (SIT) to I=0.0 M
K1=7.38, B2=11.72, B3=13.6.

UO2++	ISE	NaClO4	25°C	1.0M	C			K1=6.03 B2=10.87 B3=14.0	2000VCa (19119)	481
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Method: Hg,Hg2 oxalate electrode

UO2++	sp	NaClO4	20°C	0.05M	C			K1=3.22	1989RAb (19120)	482
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Medium: 0.05 M NaClO4/HClO4.

UO2++	gl	KN03	25°C	0.10M	U	M		K1=4.48 B2=8.43 B(UO2AL)=7.24 K(UO2A+L)=2.11 K(UO2L+A)=2.76	1985VSb (19121)	483
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H2A=phthalic acid

UO2++	oth	NaClO4	40°C	0.10M	C	M		B2=6.47 B(UO2L(nta))=8.98	1984SIa (19122)	484
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Method: Paper electrophoresis, pH 10.0.

UO2++	dis	NaClO4	25°C	4.0M	U			K1=6.28	1983CBa (19123)	485
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Medium: 4 M HClO4/NaClO4

 U02++ sp KN03 25°C 0.50M C K1=9.36 1976BVa (19124) 486
 K(UO2+2HL)=6.00

Additional method: polarography.

 U02++ sp NaClO4 20°C 0.10M U I K1=6.36 B2=10.59 1969HAa (19125) 487
 At I=1, K1=5.99, B2=10.64, B3=11.0

 U02++ sp NaClO4 20°C 1.0M U K1=4.63 B2=8.68 1967MNd (19126) 488
 K3=3.31

 U02++ gl KN03 25°C 1.00M U B2=9.1 1967RMc (19127) 489

 U02++ oth KCl 25°C 0.10M U K1=6.7 B2=11.8 1967SMe (19128) 490
 Method: electromigration

 U02++ sol R4N.X 23°C 1.00M U I M 1967Z0b (19129) 491
 K(UO2L+S03)=4.38
 K(UO2LS03+S03)=3.35 (spect.)

Medium : NH4Cl. I=2.5 M, K(UO2L+S03)=4.54, K(UO2LS03+S03)=3.72.

 U02++ dis NaClO4 20°C 0.10M U B2=11.08 1960STa (19130) 492

 U02++ sol NaClO4 20°C 1.0M U I K1=6.72 B2=11.92 1959MZa (19131) 493
 Kso=-8.66

Medium: HClO4. In 1 M HNO3: K1=6.85, B2=12.10, Kso=-8.52

 U02++ ISE oth/un 25°C 0.0 U K1=4.44 B2=10.44 1959PTa (19132) 494

 U02++ ISE oth/un 25°C 0.07M U I 1959TVa (19133) 495
 K((UO2)2L3+2L)=4.42

I=0.02: K((UO2)2L3+L)=1.32

C2H3O2Cl HL Chloroacetic CAS 79-11-8 (34)
 Chloroethanoic acid; ClCH₂.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ dis NaClO4 25°C 0.10M U H K1=1.49 1988KCb (19389) 496
 DH=14.1 kJ mol⁻¹, DS=76 J K⁻¹ mol⁻¹

 U02++ gl NaClO4 25°C 1.00M C H K1=1.436 B2=2.24 1974PBa (19390) 497
 B3=2.57

DH(K1)=1.93, DH(B2)=1.91 and DH(B3)=1.98 kJ mol⁻¹, obtained via calorimetry.

 U02++ vlt NaClO4 ? 1.0M U K1=1.6 B2=2.3 1962HOa (19391) 498

 U02++ EMF NaClO4 20°C 1.0M U K1=1.44 B2=2.24 1949AHa (19392) 499
 K3=0.51

By spectrophotometry: K1=1.38, K2=0.80, K3=0.37

C2H4N4S HL CAS 16691-43-3 (9032)
3-Amino-5-mercaptop-1,2,4-triazole;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KNO₃ 25°C 0.10M C K1=5.95 2003AHa (19500) 500

C2H4O2 HL Acetic acid CAS 64-19-7 (36)
Ethanoic acid; CH₃.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ 25°C 1.0M C T H K1=2.58 B2= 4.37 2002JRa (20209) 501
B3=6.86

Data for 35-70 °C. By calorimetry, DH(K1)=10.6 kJ mol⁻¹, DS(K1)=86 J K⁻¹ mol⁻¹; DH(B2)=20, DS(B2)=152; DH(B3)=17.5, DS(B3)=192.

U02++ gl NaClO₄ 25°C 1.0M C M K1=2.12 B2= 3.77 1999ASa (20210) 502
K3=1.29
K(UO₂+L+F)=6.66
K(UO₂+L+2F)=9.63
K(UO₂+L+3F)=11.70

Additional method: nmr. K(UO₂+2L+2F)=10.15.

U02++ dis NaCl 25°C 0.30M C I K1=2.60 1999MBb (20211) 503

Method: Solvent extraction into n-heptane, 0.05 M di-(2-ethylhexyl)-phosphoric acid. Data for 0.3-5.0 M NaCl. At I=0.0, K1=3.01.

U02++ dis NaClO₄ 25°C 0.10M U H K1=4.56 1988KCb (20212) 504
DH=21.8 kJ mol⁻¹, DS=123 J K⁻¹ mol⁻¹

U02++ vlt KCl 30°C 0.50M C K1=2.83 B2= 5.17 1982CKb (20213) 505
Method: polarography.

U02++ gl NaClO₄ 25°C 1.00M C H K1=2.457 B2=4.38 1974PBa (20214) 506
B3=6.518

DH(K1)=2.83, DH(B2)=1.45 and DH(B3)=-0.29 kJ mol⁻¹, obtained via calorimetry

U02++ EMF oth/un 25°C 0.10M U T K1=3.00 1972NPa (20215) 507
25-150 °C

K1(50 °C)=3.27, K1(90 °C)=3.63, K1(100 °C)=3.71, K1(125 °C)=3.88, K1(150 °C)=3.94

U02++ oth oth/un ? ? U K1=2.40 B2=4.43 1967MBa (20216) 508
B3=5.61

Method: paper electrophoresis

U02++ sp NaClO₄ 20°C 1.0M U K1=2.40 B2=4.43 1967MNd (20217) 509
K3=1.95

U02++ ix oth/un ? 0.50M U K1=2.52 B2=4.4 1966PKa (20218) 510
B3=6.2

U02++ gl NaClO4 30°C 1.0M U K1=1.48? B2=4.82 1964BSe (20219) 511
B3=6.00
B4=7.54

U02++ gl KN03 25°C 0.20M U K1=2.70 1963FKa (20220) 512

U02++ dis NaClO4 20°C 0.10M U K1=2.61 B2=4.9 1960STa (20221) 513
B3=6.3

U02++ EMF NaClO4 20°C 1.0M U K1=2.38 B2=4.36 1951AHa (20222) 514
K3=1.98

C2H402S H2L Thioglycolic CAS 68-11-1 (596)
Mercaptoethanoic acid; HS.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	NaClO4	25°C	1.0M	C	H		K1=1.89 B2= 3.21 B3=4.51	1978DRa (20379)	515

By calorimetry: DH(K1)=8.62 kJ mol-1, DS=65.3 J K-1 mol-1; DH(B2)=10.6,
DS=61.1; DH(B3)=0.0, DS=25.

U02++ gl oth/un 25°C .065M U TIH K1=7.45 B2=14.03 1975GSa (20380) 516
At 35 C: K1=7.56, K2=6.41; 45 C: 7.40, 6.23. At 35 C, I=0.15: 7.70, 6.45.
At 35 C, I=0.25: K1=7.90, K2=6.57. DH(K1)=-46.4 kJ mol-1

U02++ sp NaClO4 30°C 0.10M U 1969RRa (20381) 517
K(U02+HL)=2.40
K(U02+2HL)=5.75

U02++ gl KCl 30°C 0.10M U 1962CTb (20382) 518
K(U02+HL)=2.88
K(U02HL+HL)=2.40

C2H403 HL Glycolic acid CAS 79-14-1 (33)
2-Hydroxyethanoic acid; HO.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++	gl	NaClO4	25°C	1.0M	C	M			2000SGa (20643)	519
								B(U02H-1L)=-1.26		
								B(U02H-1L2)=0.19		
								B(U02H-2L2)=-4.17		
								B(U02LF2)=10.36		

B(U02LF3)=11.89, B(U02H-1LF3)=5.1, B((U02)2H-2L2F4)=11.09,
B(U02H-2L2F)=-2.40

U02++ gl NaClO₄ 31°C 0.10M U M K1=2.93 B2=5.15 1977SSb (20644) 520
B(UO₂L(Ala))=12.01
K(ML₂+M(Ala)₂=2ML(Ala))=3.54

U02++ cal NaClO₄ 25°C 1.00M C H T K1=2.35 B2=3.97 1976BBf (20645) 521
B3=5.17

DH(K1)=5.4 kJ mol⁻¹, DS=63.1 J K⁻¹ mol⁻¹; DH(K2)=7.5, DS=63.5; DH(K3)=-0.8, DS=20.0

U02++ gl NaClO₄ 31°C 0.10M U M 1976SSa (20646) 522
B((UO₂)L(glycollate))=6.62

U02++ gl NaClO₄ 20°C 1.00M C T K1=2.38 B2=3.95 1974MTa (20647) 523
B3=5.18

U02++ gl KCl 30°C 0.10M U K1=2.97 B2=5.37 1962CTb (20648) 524

U02++ dis NaClO₄ 20°C 1.0M U K1=2.71 B2=4.08 1962SBb (20649) 525
B3=5.5

C₂H₅N₀2 HL Glycine CAS 56-40-6 (85)
2-Aminoethanoic acid; H₂N.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ nmr NaClO₄ 25°C 1.0M C M B2=13.0 2000SGa (21739) 526
B(UO₂LF₃)=13.80

Method: 19F nmr.

U02++ gl NaClO₄ 25°C 1.00M C K1=9.43 B2=17.55 1994LSa (21740) 527

U02++ gl NaClO₄ 25°C 1.00M C H K1=1.16 B2=2.20 1983BRa (21741) 528
DH(K1)=3.9, DH(K2)=0.9 kJ mol⁻¹

U02++ oth NaClO₄ 35°C 0.10M C K1=7.88 1983PYa (21742) 529
B3=18.93

Method: paper electrophoresis.

U02++ vlt KCl 30°C 0.50M C K1=1.58 1982CKb (21743) 530
Method: polarography.

U02++ vlt NaClO₄ 30°C 0.10M U T 1979RRa (21744) 531
K(UO₂+2HL)=2.14

U02++ gl NaClO₄ 31°C 0.10M U M K1=7.53 B2=14.68 1977SSb (21745) 532
B(UO₂L(Malonate))=12.06
B(UO₂L(Diglycolate))=11.71
B(UO₂L(Maleate))=12.67
B(UO₂L(Glycolate))=11.11

B((UO₂)L(Thiodiglycolate))=10.45

 U02++ EMF oth/un 25°C 0.50M U K1=7.15 1973SKb (21746) 533
 By spectrophotometry: K1=7.34

 U02++ gl KCl 30°C 0.10M U T K1=7.53 B2=14.68 1962CTb (21747) 534

C2H5N02 HL Acetohydroxamic CAS 546-88-3 (2766)
 Acetohydroxamic acid, N-Hydroxyacetamide; CH₃.CO.NHOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl NaCl 25°C 0.15M U I K1=7.63 B2=14.25 1995SKb (21816) 535
 Also data for 42% MeOH/H₂O, 52% EtOH/H₂O, 59% isopropanol/H₂O and
 61% dioxane/H₂O.

 U02++ gl KN03 25°C 0.10M C K1=8.22 B2=15.30 1989KUb (21817) 536

C2H5O5P H3L CAS 4408-78-0 (4225)
 Phosphonoethanoic acid; HOOC.CH₂.PO₃H₂

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ dis NaClO₄ 25°C 0.1M C 1993NAa (21896) 537
 K(UO₂+H+HL)=7.57
 K(UO₂+HL)=6.06
 K(UO₂+2H+2HL)=14.17
 K(UO₂+2HL)=10.80

C2H6N20 L Glycinamide CAS 598-41-4 (60)
 2-Aminoethanoic acid amide; H₂N.CH₂.CO.NH₂

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl oth/un 25°C 0.15M U K1=5.15 1957LDa (21955) 538

C2H6N20 HL Acetamidoxime CAS 22059-22-9 (818)
 Acetamidoxime; CH₃.C(:N.OH).NH₂

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl KN03 25°C 0.10M C 1986HKa (21959) 539
 B(UO₂H-1L)=-0.97
 B(UO₂H-2L2)=-4.4

C2H6N202 HL CAS 5549-80-4 (833)
 2-Amino-N-hydroxyacetamide, Glycine hydroxamic acid; H₂N.CH₂.CO.NH.OH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl KN03 25°C 0.10M C K1=10.45 B2=18.95 1989KUb (21995) 540

C2H6OS HL CAS 60-24-2 (841)
2-Mercaptoethanol; HS.CH2.CH2.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	NaClO4	10°C	0.10M	U	T		K1=8.05 B2=15.13	1977SKe (22084)	541

At 20 C: K1=7.73, K2=6.80; 30 C: 8.088, 7.03

C2H6OS L DMSO CAS 67-68-5 (329)
Dimethylsulfoxide; (CH3)2.SO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	non-aq	25°C	100%	U	M			1976DBa (22129)	542

$$K((UO2A2)_2 + 2L = 2UO2A2L) = 0.06$$

HA=tropolone. Medium: benzene

C2H6O6P2 H4L (5706)
Ethene-1,1-diphosphonic acid; H2C:C(PO3H2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	dis	NaClO4	25°C	0.1M	C				1993NAa (22177)	543

$$K(UO2+H+H2L) = 7.64$$

$$K(UO2+2H+2H2L) = 13.82$$

$$K(UO2+2H2L) = 11.27$$

C2H7N03S HL Taurine CAS 107-35-7 (2214)
2-Aminoethane sulfonic acid; H2N.CH2.CH2.SO3H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	alc/w	25°C	50%	C			K1=7.68 B2=14.60	1978MCa (22440)	544

C2H8N2 L Ethylenediamine CAS 107-15-7 (23)
1,2-Diaminoethane; H2N.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO2++	ISE	non-aq	25°C	100%	C	H	K1=3.88 B2=5.88	1990CDa (23243)	545
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Medium: DMSO, 0.1 M Et4NC1O4. DH(K1)=-41.4, DH(B2)=-76.7 kJ mol-1,

DS(K1)=-64, DS(B2)=-145 J K-1 mol-1

UO2++	gl	KN03	25°C	0.10M	U	M	K1=9.02 B(UO2AL)=12.97 K(UO2A+L)=7.84 K(UO2L+A)=3.95	1985VSb (23244)	546
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H2A=phthalic acid

C2H8O6P2 H4L CAS 6145-31-9 (2579)
1,2-Ethylenediphosphonic acid; H2O3P.CH2.CH2.PO3H2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	dis	NaClO4	25°C	0.1M	C				1993NAa (23260)	547
								K(UO2+H2L)=5.34		
								K(UO2+2H2L)=8.31		

C2H8O7P2 H4L HEDPA CAS 2809-21-4 (436)
1-Hydroxyethane-1,1-diphosphonic acid; CH3.C(OH)(PO3H2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	dis	NaClO4	25°C	0.1M	C				1993NAa (23402)	548
								K(UO2+H+H2L)=7.99		
								K(UO2+2H+2H2L)=14.53		
								K(UO2+2H2L)=11.76		

C3H4O2 HL Acrylic acid CAS 79-10-7 (2044)
Propenoic acid; CH2:CH.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	NaClO4	25°C	0.10M	U			K1=2.77	1988GAc (23998)	549

Additional method: polarography.

C3H4O3 HL Pyruvic acid CAS 127-17-3 (1152)
2-Oxopropanoic acid; CH3.CO.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	NaClO4	25°C	0.11M	U	TIH		K1=1.79 B2= 3.57	1984GMc (24076)	550
Data for 30-50 C. Data for 0.03-0.11 M NaClO4. At I=0.0 M, K1=2.39										
K2=2.11; DH(K1)=30.6 kJ mol-1, DS(K1)=172 J K-1 mol-1.										
UO2++	gl	NaClO4	31°C	0.10M	C	M			1975BSa (24077)	551
								B((UO2)LA)=8.32		
								K(UO2L2+UO2A=UO2LA+UO2L)=2.54		
								K(UO2L+A)=6.17		
								K(UO2A+L)=3.13		

H2A=maleic acid

UO2++	gl	NaClO4	31°C	0.10M	C	M			1975BSa (24078)	552
								B((UO2)LA)=5.88		
								K(UO2L2+UO2A=UO2LA+UO2L)=2.24		
								K(UO2L+A)=3.73		
								K(UO2A+L)=2.83		

H2A=fumaric acid

U02++ gl NaClO₄ 31°C 0.10M C M 1975BSa (24079) 553
B((UO₂)LA)=6.94
K(UO₂L₂+UO₂A=UO₂LA+UO₂L)=1.88
K(UO₂L+A)=4.79
K(UO₂A+L)=2.46

H₂A=succinic acid

U02++ gl NaClO₄ 31°C 0.10M C M 1975BSa (24080) 554
B((UO₂)LA)=7.28
K(UO₂L₂+UO₂A=UO₂LA+UO₂L)=2.61
K(UO₂L+A)=5.13
K(UO₂A+L)=3.20

H₂A=adipic acid

U02++ gl NaClO₄ 31°C 0.10M C M 1975BSa (24081) 555
B((UO₂)LA)=6.69
K(UO₂L₂+UO₂A=UO₂LA+UO₂L)=2.39
K(UO₂L+A)=4.54
K(UO₂A+L)=2.98

H₃A=thiomalic acid

U02++ sp NaClO₄ 30°C 0.10M U K1=2.71 B2=5.33 1969RRa (24082) 556

U02++ gl NaClO₄ 31°C 0.10M U K1=2.15 B2=2.74 1968RSa (24083) 557

C3H4O₄ H₂L Malonic acid CAS 141-82-2 (79)

Propanedioic acid; CH₂(COOH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ 25°C 0.10M M M K1=5.56 1987NCa (24575) 558
K(UO₂(nta)+L)=4.29

U02++ gl NaClO₄ 30°C 0.10M M I K1=5.56 B2= 9.36 1985ARc (24576) 559
Also data for 20-80% dioxane/H₂O. For 40% dioxane/H₂O, K1=7.41, K2=5.50.

U02++ gl KN0₃ 25°C 0.20M U T 1985KMc (24577) 560
K(UO₂A+L)=4.66

H₂A=iminodiacetic acid; 5 C:K=4.92; 45 C: K=4.43, DH=-23.0 kJ mol⁻¹,
DS=12 J K⁻¹ mol⁻¹

U02++ gl NaClO₄ 30°C 0.10M C I K1=5.56 B2= 9.36 1978SJb (24578) 561
Data for 20-80% v/v dioxane/H₂O, 0.03-0.11 M NaClO₄.

In 40% dioxane/H₂O, K1=7.41, K2=5.50.

U02++ gl NaClO₄ 25°C 1.00M U K1=5.42 B2=9.48 1977BNa (24579) 562

U02++ gl NaClO₄ 31°C 0.10M U M K1=5.20 B2=9.21 1977SSb (24580) 563
B(UO₂L(Ala))=13.19
K(ML₂+M(Ala))=1.77

U02++ EMF NaClO₄ 31°C 0.10M U 1974BSa (24581) 564
B((U02)L(succinate))=9.23
B((U02)L(glutarate))=8.59
B((U02)L(adipate))=8.21
B((U02)L(thiomalate))=8.9
B(U02+L+diglycollate)=9.60.

U02++ oth oth/un ? ? U 1971GPa (24582) 565
K((U02)2O₂+2L)=4.48

From survey of literature data

U02++ gl KN₃O 25°C 0.50M U K1=5.66 B2=9.66 1969V0b (24583) 566

U02++ gl NaClO₄ 31°C 0.10M U K1=5.28 B2=9.29 1968RSa (24584) 567

U02++ gl NaClO₄ 30°C 0.20M U K1=4.88 B2=8.63 1967AMa (24585) 568

U02++ gl KN₃O 25°C 1.00M U K1=5.66 B2=9.66 1967RMc (24586) 569

C₃H₅O₂C₁ HL CAS 107-94-8 (1436)

3-Chloropropanoic acid; Cl.CH₂.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ 25°C 1.00M C H K1=2.056 B2=3.580 1974PBa (24736) 570
B3=5.18

DH(K1)=2.70, DH(B2)=2.30 and DH(B3)=0.00 kJ mol⁻¹, obtained via calorimetry.

U02++ EMF NaClO₄ 20°C 1.00M U K1=2.05 B2=3.55 1972MPa (24737) 571
B3=4.98

C₃H₆N₆ L Melamine CAS 108-78-1 (889)

2,4,6-Triamino-1,3,5-triazine, sym-Triaminotriazine; C₃N₃(NH₂)₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ 30°C 0.10M U K1=3.50 B2=6.66 1981JKa (24839) 572

C₃H₆O₅ HL CAS 1892-31-5 (3550)

Thiopropanoic acid; CH₃.CH₂.CO.SH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ ? 0.00 U K1=3.7 B2=11.10 1968MNa (24861) 573

C₃H₆O₂ HL Propionic acid CAS 79-09-4 (35)

Propanoic acid; CH₃.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ vlt KCl 30°C 0.50M C K1=2.85 B2= 5.20 1982CKb (25064) 574
B3=7.20

Method: polarography.

U02++ gl NaClO4 31°C 0.10M C M 1975BSa (25065) 575
 $B((U_02)LA)=10.70$
 $K(U_02L_2+U_02A=U_02LA+U_02L)=3.66$
 $K(U_02L+A)=5.42$
 $K(U_02A+L)=7.67$

H2A=malonic acid

U02++ gl NaClO4 31°C 0.10M C M 1975BSa (25066) 576
 $B((U_02)LA)=7.65$
 $K(U_02L_2+U_02A=U_02LA+U_02L)=0.14$
 $K(U_02L+A)=3.17$
 $K(U_02A+L)=4.62$

H2A=succinic acid

U02++ gl NaClO4 31°C 0.10M C M 1975BSa (25067) 577
 $B((U_02)LA)=7.69$
 $K(U_02L_2+U_02A=U_02LA+U_02L)=1.97$
 $K(U_02L+A)=3.99$
 $K(U_02A+L)=4.66$

H2A=glutaric acid

U02++ gl NaClO4 31°C 0.10M C M 1975BSa (25068) 578
 $B((U_02)LA)=7.21$
 $K(U_02L_2+U_02A=U_02LA+U_02L)=0.10$
 $K(U_02L+A)=3.13$
 $K(U_02A+L)=4.18$

H2A=adipic acid

U02++ gl NaClO4 31°C 0.10M C M 1975BSa (25069) 579
 $B((U_02)LA)=7.32$
 $K(U_02L_2+U_02A_2=U_02LA+U_02L)=0.58$
 $K(U_02L+A)=3.61$
 $K(U_02A+L)=4.29$

H3A=thiomalic acid

U02++ gl NaClO4 31°C 0.10M C M 1975BSa (25070) 580
 $B((U_02)LA)=10.15$
 $K(U_02L_2+U_02A=U_02LA+U_02L)=4.28$
 $K(U_02L+A)=5.25$
 $K(U_02A+L)=7.12$

H2A=diglycollic acid

U02++ gl NaClO4 31°C 0.10M U K1=3.03 1968RSa (25071) 581

U02++ sp NaClO4 20°C 1.0M U K1=2.53 B2=4.68 1967MNd (25072) 582

K3=1.81

K4=1.76

U02++ vlt NaNO3 ? 1.0M U 1962H0a (25073) 583
K(U(V)O2L+U02=U02L+U(V)O2)=4.7
K'=5.2

K': U(V)O2L2)+U02=U02L2+U(V)O2

C3H6O2S HL CAS 2444-37-3 (1074)
(Methylthio)ethanoic acid; CH3.S.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ vlt KN03 25°C 0.45M C K1=1.75 1985CEa (25094) 584
Method: differential pulse polarography, using anodically generated Hg++
as indicator ion.

C3H6O2S H2L Thiolactic acid CAS 79-42-5 (366)
2-Mercaptopropanoic acid; CH3.CH(SH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 20°C 0.10M U T K1=8.36 B2=15.73 1974SSa (25177) 585
At 30 C: K1=8.72, B2=16.13; 40 C: K1=9.20, B2=16.62

C3H6O2S H2L CAS 107-96-0 (437)
3-Mercaptopropanoic acid; HS.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 30°C 0.10M U K1=3.25 1973RSa (25232) 586

C3H6O3 HL CAS 81598-26-7 (2521)
3-Hydroxypropanoic acid; HO.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 30°C 0.12M U K1=2.74 B2=4.94 1962CMb (25282) 587
K3=2

U02++ gl KCl 30°C 0.10M U K1=3.25 B2=6.13 1962CTb (25283) 588

C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)
L-2-Hydroxypropanoic acid; CH3.CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ dis NaCl 25°C 0.30M C I K1=2.60 1999MBb (25560) 589

Method: Solvent extraction into n-heptane, 0.05 M di-(2-ethylhexyl)-
phosphoric acid. Data for 0.3-5.0 m NaCl. At I=0.0, K1=3.16.

U02++	gl	NaClO4	25°C	1.00M	C	K1=2.68 B3=5.64	B2=4.47	1984LLa (25561)	590
U02++	gl	NaClO4	20°C	1.00M	C	T K1=2.77 B3=5.78	B2=4.52	1974MTa (25562)	591
U02++	sp	NaClO4	20°C	1.0M	U	K1=2.43 K3=1.86	B2=4.49	1967MNd (25563)	592
U02++	gl	NaClO4	25°C	1.0M	U	K1=2.77 K3=1.33	B2=4.45	1967TGa (25564)	593
U02++	gl	KNO3	25°C	0.10M	U	K1=2.48 K(UO2(OH)L+H)=4.0 K(2UO2L=(UO2(OH)L)2+2H)=-5.13	1967VAA (25565)	594	
U02++	gl	NaClO4	31°C	0.12M	U	K1=3.36 K3=2	B2=5.56	1962CMb (25566)	595
U02++	dis	NaClO4	20°C	1.0M	U	K1=2.81 B3=5.46	B2=4.56	1962SBb (25567)	596

C3H7NO2 HL Alanine CAS 56-41-7 (86)
2-Aminopropanoic acid: H₂N.CH(CH₃).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptN_o

$$K(UO_2(dtapa)+L) = 5.84; \quad K(UO_2(hedta)+L) = 6.94.$$

hedta is N-(2-hydroxyethyl)-1,2-diaminoethane-N,N',N'-triethanoic acid

UO2++ vlt KC1 30°C 0.50M C K1=1.54 1982CKb (26285) 598

Method: polarography.

UO2++ gl KNO₃ 25°C 0.10M U T K1=7.33 B2=14.97 1982NMa (26286) 599

UO₂++ vlt NaClO₄ 30°C 0.10M U R 1979RRa (26288) 601
 $K(UO_2+2HL)=2.15$

UO₂++ EMF oth/un 25°C 0.50M U K1=7.0 1973SKb (26289) 602
By spectrophotometry, K1=7.0

U02++ EMF oth/un ? ? U K1=9.00 1970FMB (26290) 603

C3H7N02 HL B-Alanine CAS 107-95-9 (575)
3-Aminopropanoic acid; H2N.CH2.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	KNO ₃	25°C	0.20M	U	M		K1=7.70 B2=14.79 K(UO2(ida)+L)=7.35 K(UO2(nta)+L)=7.21 K(UO2(edta)+L)=6.73 K(UO2(cdtta)+L)=6.35	1992SSf (26485)	604

K(UO2(dtpa)+L)=5.80; K(UO2(hedta)+L)=6.85.

hedta is N-(2-hydroxyethyl)-1,2-diaminoethane-N,N',N'-triethanoic acid

U02++	gl	NaClO ₄	25°C	1.0M	U	H	T	K1=1.93 B2=3.44 B3=4.82	1987BRa (26486)	605
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DH₁ = 6.5, DH(B₂) = 12.0, DH(B₃) = 11.3, DS₁ = 59, DS(B₂) = 106, DS(B₃) = 130

U02++	oth	NaNO ₃	35°C	0.10M	U	M			1985VSA (26487)	606
								K(UO2(NTA)+L)=5.12		

By electrophoresis.

U02++	gl	NaClO ₄	30°C	0.10M	U		T		1980RRa (26488)	607
								K(UO2+HL)=2.44		

U02++	vlt	NaClO ₄	30°C	0.10M	U		T		1979RRa (26489)	608
								K(UO2+2HL)=3.49		

U02++	gl	NaClO ₄	30°C	0.10M	U			K1=9.20	1973RSa (26490)	609
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U02++	EMF	oth/un	25°C	0.50M	U			K1=7.86	1973SKb (26491)	610
								By spectrophotometry, K1=7.93		

U02++	EMF	oth/un	? ?	U				K1=9.90	1970FMB (26492)	611
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U02++	gl	KCl	30°C	0.10M	U			K1=7.78 B2=15.31	1962CTb (26493)	612

C3H7N02 HL DL-Alanine CAS 302-72-7 (189)
DL-2-Aminopropanoic acid; H2N.CH(CH₃).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++	EMF	NaClO ₄	31°C	0.10M	U			K1=8.55	1977RRa (26543)	613

C3H7N02S H2L Cysteine CAS 52-90-4 (96)
2-Amino-3-mercaptopropanoic acid; H2N.CH(CH₂.SH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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C3H11N06P2 H4L (6772)
(Dimethylamino)-N-methylenediphosphonic acid; (CH₃)₂N.CH(P(03H₂)₂)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	R4N.X	25°C	0.10M	C			K1=24.8 K(UO ₂ (OH)L+H)=10.7 K(UO ₂ (OH) ₂ L+H)=12.0 K(UO ₂ L+H)=9.8 K(UO ₂ HL+H)=5.3	1994BRa (28416)	627

Medium: Me4NN03. $K(UO_2H_2L_2+H) = 7.4$, $K(UO_2H_3L_2+H) = 5.0$

C4H2O4 H2L Squaric acid CAS 2892-51-5 (439)
3,4-Dihydroxy-3-cyclobutene-1,2-dione:

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgO^{++} sp. NaClO₄, 25°C, 0, 50M, II K1-3 08 1969Twa (28670) 628

C4H4N2O2S H2L Thiobarbituric CAS 504-17-6 (4279)
4,6-Dihydroxy-2-mercapto-5-pyrimidine-2-thiobarbituric acid:

Metal	Mtd	Medium	Temp	Conc	CaL	Flags	Lg	K values	Reference	ExptNo
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UO₂++ sp alc/w 25°C 100% U I 1968PPb (28898) 629
K(?)=4.6

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U02++ sp alc/w rt 100% U I B2=9.2 1968PPb (28921) 630

Solvent: EtOH. In MeOH, $K(?)=3.7$

C4H4N6 L 8-Azaadenine CAS 1123-54-2 (1884)
8-Aza-6-aminopurine:

Metal	Mtd	Medium	Temp	Conc	Ca _l	Flags	Lg	K values	Reference	ExptNo
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UO₂++ g1 KNO₃ 35°C 0 10M H₂O M K1-4 20 1982BK₃ (28956) 631

00244 g¹ KN03 35 °C 0.10M H₂O KI-4.20 1982RKA (28558) 031
 $K(UO_2(EDTA)+L)=3.18$
 $K(UO_2(EDTA)L+H)=5.91$

C4H4N6O L-8-Azaguanine CAS 134-58-7 (114)
2-Amino-6-hydroxy-8-azapurine:

U02++	gl alc/w	25°C	50%	U	K1=9.69	1978MCb (28964)	632			

C4H404	H2L	Maleic acid		CAS 110-16-7	(111)					
cis-Butenedioic acid; HOOC.CH:CH.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo		

U02++	sp	oth/un	25°C	?	U		K1=4.05	B2=6.15	1985GGa (29145)	633

U02++	gl	KNO3	25°C	0.20M	U	T			1985KMc (29146)	634
K(UO2A+L)=5.12										
H2A=iminodiacetic acid; 5 C:K=5.38; 45 C: K=4.88, DH=-18.3 kJ mol-1,										
DS=34 J K-1 mol-1										

U02++	gl	KNO3	25°C	0.10M	U	M	K1=4.80		1985VSb (29147)	635
B(UO2LA)=9.05										
K(UO2A+L)=3.92										
K(UO2L+A)=4.25										
H2A=phthalic acid										

U02++	gl	NaClO4	31°C	0.10M	U	M	K1=5.20		1977SSb (29148)	636
B((UO2L(Ala)))=13.30										

U02++	gl	NaClO4	31°C	0.10M	U	M			1976SSa (29149)	637
B((UO2)L(glycollate))=8.31										

U02++	gl	NaClO4	31°C	0.10M	U	M			1976SSa (29150)	638
B((UO2)L(malonate))=8.13										

U02++	EMF	NaClO4	31°C	0.10M	U				1974BSa (29151)	639
B((UO2)L(succinate))=8.42										
B((UO2)L(glutarate))=8.27										
B((UO2)L(adipate))=8.01										
B((UO2)L(thiomalate))=8.7										

U02++	gl	NaClO4	31°C	0.10M	U		K1=5.15		1968RSa (29152)	640

U02++	gl	KNO3	25°C	1.00M	U		K1=4.46		1967RMc (29153)	641

U02++	gl	KNO3	25°C	1.0M	U		K1=4.45		1964PCa (29154)	642

C4H404	H2L	Fumaric acid		CAS 110-17-8	(289)					
trans-Butenedioic acid; HOOC.CH:CH.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values		Reference ExptNo	

U02++	gl	KNO3	25°C	0.20M	U	T			1985KMc (29225)	643
K(UO2A+L)=3.82										
H2A=iminodiacetic acid; 5 C:K=3.94; 45 C: K=3.72, DH=-8.4 kJ mol-1,										
DS=46 J K-1 mol-1										

U02++ gl KN03 25°C 0.10M U M K1=3.47 1985VSb (29226) 644
B(UO2AL)=6.15
K(UO2A+L)=1.02
K(UO2L+A)=2.68

H2A=phthalic acid

U02++ EMF NaCl04 31°C 0.10M U M 1974BSa (29227) 645
B((UO2)L(succinate))=6.73
B((UO2)L(adipate))=6.56
B((UO2)L(thiomalate))=6.9

U02++ gl NaCl04 31°C 0.10M U M K1=3.05 1968RSa (29228) 646

C4H5N3O HL Cytosine CAS 71-30-7 (1096)
2-Oxy-6-aminopyrimidine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KN03 35°C 0.10M U M K1=10.42 1982RKa (29420) 647
K(UO2+HL)=3.70

C4H5O4Cl1 H2L CAS 16045-92-4 (2232)

Chlorosuccinic acid; HOOC.CH(Cl).CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaCl04 30°C 0.10M M I K1=3.57 B2= 6.52 1985ARC (29438) 648

Also data for 20-80% dioxane/H2O. For 40% dioxane/H2O, K1=6.42, K2=4.26.

C4H6O2 HL Crotonic acid CAS 107-93-7 (2990)

But-2-enoic acid; CH3.CH:CH.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaCl04 25°C 0.10M U M K1=2.98 1983GAa (29726) 649

U02++ gl NaCl04 31°C 0.10M U M K1=2.74 B2=5.27 1968RSa (29727) 650

C4H6O2Br2 HL CAS 41459-42-1 (6308)

3-Bromo-2-(bromomethyl)-propanoic acid; BrCH2.CH(CH2Br).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaCl04 31°C 0.10M U M K1=3.49 1976RRb (29735) 651

C4H6O4 H2L Succinic acid CAS 110-15-6 (112)

1,4-Butanedioic acid; HOOC.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 30°C 0.50M U K1=3.54 B2= 6.06 1990PNa (30058) 652

U02++ gl NaClO4 25°C 0.10M M M K1=4.48 1987NCa (30059) 653
K(UO2(nta)+L)=3.23

U02++ gl NaClO4 30°C 0.10M M I K1=4.48 B2= 7.78 1985ARc (30060) 654
Also data for 20-80% dioxane/H2O. For 40% dioxane/H2O, K1=6.89, K2=4.85.

U02++ gl KN03 25°C 0.20M U T HM 1985KMc (30061) 655
K(UO2A+L)=3.80

H2A=iminodiethanoic acid; 5 C:K=3.92; 45 C: K=3.72. DH=-9.6 kJ mol-1,
DS=42 J K-1 mol-1

U02++ vlt KCl 20°C 0.60M U T H K1=2.00 B2=2.30 1985SKb (30062) 656
B3=3.14

30 C, K1=1.78, B2=2.70. 40 C, K1=1.65, B2=2.53. DH(K1)=-30.5 kJ mol-1,
DH(K2)=20.0

U02++ gl KN03 25°C 0.10M U M K1=4.38 1985VSb (30063) 657
B(UO2AL)=8.51
K(UO2A+L)=3.38
K(UO2L+A)=4.13

H2A=phthalic acid

U02++ gl NaClO4 25°C 1.0M U H K1=3.85 1981BCg (30064) 658
B(UO2HL)=7.41
B(UO2HL2)=11.28

By calorimetry: DH(K1)=21.7 kJ mol-1, DS(K1)=146 J K-1 mol-1; DH(UO2HL)=
8.28, DS(UO2HL)=170; DH(UO2HL2)=13.6, DS(UO2HL2)=262.

U02++ gl NaClO4 30°C 0.10M U K1=4.48 1973KJa (30065) 659

U02++ gl NaClO4 31°C 0.10M U M 1971RBc (30066) 660
K(UO2+L+HA)=7.67
K(UO2+L+B)=7.87
K(UO2+L+C)=8.00
K(UO2+L+D)=8.66

H2A=4-hydroxybenzoic acid; HB=benzoic acid; HC=phenylacetic acid
HD=phenoxyacetic acid

U02++ gl NaClO4 31°C 0.10M U M K1=4.48 1970RSb (30067) 661
K(UO2+L+A)=7.23
K(UO2+L+B)=5.87
K(UO2+L+C)=6.87

H2A=adipic acid, H2B=thiomalic acid, H2C=itaconic acid

U02++ gl KN03 25°C 0.50M U K1=3.87 1969V0b (30068) 662
K(UO2+HL)=2.13

U02++ gl NaClO₄ 31°C 0.10M U K1=4.48 1968RSa (30069) 663

 U02++ sp NaClO₄ 20°C 1.0M U 1967MNd (30070) 664
 K(UO₂+HL)=2.53

 U02++ gl KN0₃ 25°C 1.00M U K1=3.68 1967RMc (30071) 665
 separation of solid phase.

 U02++ gl KN0₃ 25°C 0.20M U 1963FKa (30072) 666
 K(UO₂+HL)=2.62

C4H6O4 H2L Me-Malonic Acid CAS 516-15-2 (816)
 Methylpropanedioic acid; HOOC.CH(CH₃).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	30°C	40%	M	I		K1=9.95 B2=16.68	1985ARC (30141)	667
Medium: 0.10 M NaClO ₄ in 40% dioxane/H ₂ O. Also data for 20, 60 and 80% dioxane/H ₂ O.										
U02++	gl	KN0 ₃	25°C	0.50M	U			K1=5.56 B2=9.53	1969VOb (30142)	668
K(UO ₂ L=UO ₂ LOH+H)=-5.55										

C4H6O4S H2L Thiodiacetic CAS 123-93-3 (140)
 2,2'-Thiodiglycolic acid, Thiodiethanoic acid; HOOC.CH₂.S.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	vlt	KN0 ₃	25°C	0.20M	C			K1=3.11	1985CEa (30235)	669
Method: differential pulse polarography, using anodically generated Hg ⁺⁺ as indicator ion.										
U02++	gl	NaClO ₄	30°C	0.10M	U	T		K1=3.43	1981SJb (30236)	670
At 20 C: K1=3.37; 40 C: 3.48										
U02++	gl	NaClO ₄	30°C	0.10M	U	M		K1=3.43	1981SJc (30237)	671
B((UO ₂)L(malonate))=8.44										
B((UO ₂)L(succinate))=6.71										
B((UO ₂)L(itaconate))=7.55										
B((UO ₂)L(glutarate))=6.44										
B(M+L+adipic acid)=6.25.										
U02++	gl	NaClO ₄	25°C	1.00M	U	H		K1=2.97	1980BTa (30238)	672
B((UO ₂)HL)=5.43										
B((UO ₂)HL2)=8.39										
DH(K1)=14.8, DH(MHL)=17.8 and DH(MHL2)=25.7 kJ mol ⁻¹ .										
Alternative method: Calorimetry.										
U02++	gl	NaClO ₄	30°C	0.10M	U	M		K1=3.43	1978SJa (30239)	673
B((UO ₂)L(Asp))=7.96										

$$B((UO_2)L(Glu))=7.55$$

 U02++ gl NaClO4 31°C 0.10M U M K1=2.52 B2=4.49 1977SSb (30240) 674
 $B(UO_2L(Ala))=11.49$
 $K(ML_2+M(Ala)_2)=2ML(Ala))=3.19$

 U02++ gl NaClO4 20°C 1.00M U K1=3.16 1973CBc (30241) 675
 $K(UO_2+HL+L)=4.38$

 C4H604S H3L Thiomalic acid CAS 70-49-5 (109)
 2-Mercaptosuccinic acid, 2-Sulfanyl-1,4-butanedioic acid; HOOC.CH(SH).CH2.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl NaClO4 30°C 0.10M M I K1=3.70 B2= 6.35 1985ARC (30370) 676
 Also data for 20-80% dioxane/H2O. For 40% dioxane/H2O, K1=6.05, K2=4.38.

 U02++ gl NaClO4 30°C 0.10M U T K1=3.70 1981SJb (30371) 677
 Data also for 20 and 40 C

 U02++ gl NaClO4 30°C 0.10M U M K1=3.70 1978SJa (30372) 678
 $B((UO_2)L(Asp))=8.94$
 $B((UO_2)L(Glu))=8.81$

 U02++ gl NaClO4 31°C 0.10M U M 1971RSa (30373) 679
 $K(UO_2+HL+A)=7.38$
 $K(UO_2+HL+B)=8.18$
 $K(UO_2+HL+C+A)=9.32$
 $K(UO_2+HL+A+B)=9.29$

H2A=adipic acid; HB=itaconic acid; H2C=succinic acid

 U02++ gl NaClO4 31°C 0.10M U 1968RSa (30374) 680
 $K(UO_2+HL)=3.71$

 U02++ gl NaClO4 45°C 0.10M U T 1968RSf (30375) 681
 $K(UO_2+HL)=3.91$

At 31 C: K=3.82

 U02++ sp oth/un 5°C ? U 1963MNb (30376) 682
 $K(UO_2+HL)=3.0$

 U02++ gl KCl 30°C 0.10M U 1962CTb (30377) 683
 $K(UO_2+HL)=3.56$
 $K(UO_2HL+HL)=3.42$

 C4H604S2 H4L CAS 2418-14-6 (4264)
 2,3-Dimercaptobutanedioic acid; HOOC.CH(SH).CH(SH).COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 40% M I K1=6.09 B2=10.11 1985ARc (30398) 684
 Medium: 0.10 M NaClO₄ in 40% dioxane/H₂O. Also data for 20, 60 and 80% dioxane/H₂O.

C4H6O5 H2L Malic acid CAS 617-48-1 (393)
 2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH₂.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	30°C	40%	M	I		K1=8.26 B2=14.36	1985ARc (30746)	685
Medium: 0.10 M NaClO ₄ in 40% dioxane/H ₂ O. Also data for 20, 60 and 80% dioxane/H ₂ O.										

U02++	gl	KN03	25°C	0.20M	U	T	HM		1985KMc (30747)	686
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$$K(U02A+L)=5.04$$

H2A=iminodiethanoic acid; 5 C:K=5.22; 45 C: K=4.85. DH=-15.0 kJ mol⁻¹, DS=46 J K⁻¹ mol⁻¹

U02++	sp	NaNO ₃	25°C	0.50M	U			K1=18.35	1978KPC (30748)	687
								B((U02)HL)=19.8		
								B((U02)2L2)=38.03		
								B((U02)3L5)=79.50		
								B((U02)3H-2L5)=57.70		

Malic acid defined as H₃L with protonation constants K1=15.46, K2=4.49, K3=3.14

U02++	dis	oth/un	25°C	?	U				1972MKc (30749)	688
								K(2U02L=(U02L)2)=7.0		

U02++	gl	oth/un	25°C	?	U				1972MKc (30750)	689
K(U02+H2L=U02H-1L+3H)=-7.40										

U02++	dis	oth/un	25°C	?	U				1970AKa (30751)	690
								K(2U02L=(U02L)2)) > 7		
Keff(InL2+0.5(U02L)2=InU02L2+L)=1.49 pH 4										

U02++	gl	KN03	25°C	1.0M	U				1964PCa (30752)	691
								K(U02+H2L=U02H-1L+3H)=-5.55		
								K(2U02H-1L=(U02)2H-2L2)=3.35		

U02++	gl	KN03	25°C	1.0M	U				1964RMB (30753)	692
								K(U02+L=U02H-1L+H)=1.66		
								K(U02+H2L=U02H-1L+3H)=-5.55		
								K((U02)2(H-1L)2(OH)+H)=6.1		
K(2U02+2H2L=(U02)2(H-1L)2+6H)=-7.75, K(2(U02)3(H-1L)3(OH)2=3(U02)2(H-1L)2)=19.35										

U02++	gl	KCl	30°C	0.10M	U			K1=5.50 B2=9.13	1962CTb (30754)	693

C4H6O5 H2L Diglycolic acid CAS 110-99-6 (243)

Di(carboxy)methyl ether, 2,2'-Oxydiethanoic acid; HOOC.CH2.O.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	NaClO4	25°C	1.00M	U	H		K1=5.11 B2=7.54 B((U02)HL2)=10.03	1980BTa (30942)	694
DH(K1)=29.17, DH(B2)=23.5 and DH(MHL2)=16.86 kJ mol-1.										
Alternative method: Calorimetry.										
U02++	gl	NaClO4	31°C	0.10M	U	M		K1=4.90 B2=7.74 B(U02L(Ala))=12.78 K(ML2+M(Ala)2=2ML(Ala))=2.51	1977SSb (30943)	695

U02++	gl	NaClO4	20°C	1.00M	U			K1=5.11	1973CBc (30944)	696
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U02++	gl	NaClO4	31°C	0.10M	U			K1=4.90 B2=7.74	1968RSa (30945)	697
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C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)
 L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++	sp	NaClO4	20°C	0.05M	C			K1=3.30	1989RAb (31381)	698
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Medium: 0.05 M NaClO4/HClO4.

U02++	oth	NaClO4	40°C	0.10M	C			K1=4.91	1982SYb (31382)	699
								*K(U02H2L)=-4.6 K(2U02HL=(U02)2L2+2H)=-9.4		

Method: paper electrophoresis. Medium: 0.10 M HClO4.

U02++	oth	oth/un	40°C	0.10M	U	M			1981YSa (31383)	700
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Method: paper electrophoresis

U02++	dis	oth/un	25°C	?	U				1972MKc (31384)	701
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$$\begin{aligned} K(2U02L=(U02)2L2) &= 5.7 \\ K(U02+H2L=U02H-1L+3H) &= -6.85 \end{aligned}$$

U02++	dis	oth/un	25°C	?	U				1970AKa (31385)	702
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see comment

$$K(2U02L=(U02)2L2) = 6.7$$

$K'(InL2+0.5(U02L)2=InU02L2+L)=1.49$, conditional constant

U02++	gl	KNO3	25°C	1.0M	U				1964PCa (31386)	703
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$$\begin{aligned} K(U02+H2L=U02H-1L+3H) &= -5.62 \\ K(2U02H-1L=(U02)2H-2L2) &= 3.24 \end{aligned}$$

U02++	gl	KNO3	25°C	1.0M	U				1964RMb (31387)	704
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$$\begin{aligned} K(U02+L=U02H-1L+H) &= 0.75 \\ K(U02+H2L=U02H-1L+3H) &= -5.62 \end{aligned}$$

$K((UO_2)_2(H-1L)_2(OH)+H)=5.26$
 $K((UO_2)_3(H-1L)_3(OH)_2+4H)=17.91$

K(2UO₂+2H₂L=(UO₂)₂(H-1L)₂+6H)=-8.00

UO₂++ dis NaClO₄ 20°C 0.10M U B2=9.73 1963STc (31388) 705

C4H₇N₀2 HL Acetoacetamide CAS 2044-64-6 (1407)
3-Oxobutanamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	NaClO ₄	25°C	0.5M	C			K1=9.48	1998HCb (31448)	706

C4H₇N₀3 HL CAS 543-24-8 (3586)
N-Acetylglycine; CH₃.CO.NH.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	NaClO ₄	30°C	0.10M	U			K1=3.30	1973RSa (31510)	707

C4H₇N₀4 H₂L Aspartic acid CAS 56-84-8 (21)
Aminobutanedioic acid; H₂N.CH(CH₂.COOH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	KNO ₃	25°C	0.10M	C			K1=8.45	2003AHa (31958)	708

UO₂++ gl KNO₃ 25°C 0.10M M M K1=8.89 1996AEa (31959) 709
Data for ternary complexes with dipicolinic acid.

UO ₂ ++	gl	NaNO ₃	25°C	1.10M	U	M		K1=9.11	1995ADc (31960)	710
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K(UO₂+Hegta+HL)=8.30
K(UO₂(Hegta)L+H)=5.94
K(UO₂(egta)L+H)=8.64

UO ₂ ++	gl	NaClO ₄	30°C	0.10M	M	M			1995JSa (31961)	711
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K(UO₂+2HL)=3.77
K(UO₂+HL)=2.65
B(UO₂AL)=11.30
B(UO₂CL)=10.05

K(UO₂+2HL) by polarography. B(UO₂DL)=9.60, B(UO₂EL)=6.60. H₂A is oxalic acid, H₂C is malonic acid, H₂D is succinic acid, H₂E is lactic acid

UO₂++ gl NaClO₄ 25°C 1.00M U H K1=2.41 B2=4.14 1989BRc (31962) 712
DH(K1)=8.9, DH(B2)=10.5 kJ mol⁻¹; DS(K1)=76, DS(B2)=114 J mol⁻¹ K⁻¹

UO₂++ gl NaClO₄ 30°C 0.10M M I K1=8.71 B2=16.11 1985ARc (31963) 713
Also data for 20-60% dioxane/H₂O. For 40% dioxane/H₂O, K1=10.70, K2=8.44.

UO ₂ ++	dis	NaCl	25°C	0.1M	U			K1=3.32	1984SCa (31964)	714
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UO₂++ gl NaClO₄ 30°C 0.10M U M K1=8.71 1978SJa (31965) 715

UO2++ EMF oth/un 25°C 0.50M U K1=8.40 1973SKb (31966) 716

UO2++ sp oth/un 25°C 0.50M U K1=8.62 1973SKb (31967) 717

U02++ g1 NaClO₄ 30°C 0.10M U T K1=8.34 1971TMc (31968) 718
 K1(40 °C)=8.93; K1(50 °C)=10.40

UO₂++ g1 KNO₃ 25°C 0.20M U 1963FKa (31969) 719
K(UO₂+HL)=2.61

UO2++ gl KCl 30°C 0.10M U K1=8.00 1962CTb (31970) 720

C4H7NO4 H2I TDA CAS 142-73-4 (118)

Iminodiethanoic acid: $\text{HN}(\text{CH}_2\text{COOH})_2$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ gl NaClO₄ 25°C 0.20M U K1=13.00 1986SLb (32384) 721

UO₂++ g1 NaClO₄ 25°C 3.0M C K1=9.63 1984BLb (32385) 722

BIG (SCL) 12111 1978

g₁ KNO₃ 25 °C 0.1M 0 KI 0.75 DE 17.20 192ENBG (3230)

B₁(UO₂)HL=11.19

$B((U02)HL2)=19.81$
 $B((U02)H2L2)=22.50$

$\Delta H(K1) = -2.2 \text{ kJ mol}^{-1}$, $\Delta H((UO_2)HL) = -30.9$, $\Delta H((UO_2)HL2) = -50.0$, $\Delta H((UO_2)H_2L2) = -52.0$. Alternative method: Calorimetry.

UO2++ gl NaClO4 20°C 1.00M U K1=8.66 1973CBc (32388) 725

UO₂++ vlt NaClO₄ 30°C 0.15M U 1967LCa (32389) 726

R(882+212)=3.92

K1=8.93(I=0.1 M) *****

C4H8N2O3 HL Asparagine CAS 70-47-3 (17)

2-Aminobutanedioic acid 4-amide; H₂N.CH(CH₂.CO.NH₂).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl KNO3 25°C 0.10M M M K1=8.12 1996AEa (32736) 728

Data for ternary complexes with dipicolinic acid.

UO₂++ gl NaClO₄ 30°C 0.10M M 1995JSa (32737) 729
 $K(UO_2+2HL) = 2.47$
 $K(UO_2+HL) = 2.00$

K(UO₂+2HL) by polarography.

UO2++ EMF NaClO4 31°C 0.10M U K1=7.23 1977RRa (32738) 730

UO2++ gl NaClO4 25°C 0.10M U K1=6.79 B2=12.95 1973TSe (32739) 731

UO2++ EMF oth/un ? ? U K1=6.85 1970FMb (32740) 732

C4H8N2O3 HL Gly-Gly CAS 556-50-3 (54)

Glycyl-glycine; H₂N.CH₂.CO.NH.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ gl NaClO₄ 25°C 1.0M U H 1992BRc (33056) 733

K(UO₂+HL)=1.61

$$K(UO_2+2HL) = 2.99$$

$$K(UO_2+3HL) = 3.90$$

DH(UO₂+HL)=7.0 kJ mol⁻¹, DS(UO₂+HL)=54 J K⁻¹ mol⁻¹; DH(UO₂+2HL)=11.5, DS(UO₂+2HL)=96; DH(UO₂+3HL)=14.7, DS(UO₂+3HL)=124

UO₂++ oth NaClO₄ 35°C 0.10M C M K1=5.40 B2= 9.00 1986SYa (33057) 734
 K(UO₂(nta)₂⁺)=5.15

Method: paper electrophoresis. Medium pH 8.5.

UO2++ gl KC1 25°C 0.10M U K1=3.76 B2=10.15 1982ZZa (33058) 735

UO2++ vlt NaClO4 30°C 0.10M C K1=0.0 B2= 2.45 1980SBe (33059) 736

Method: polarography.

S4H202 III. Technotriol acid - CAS 70-31-2 (572)

3-Methylbutyrynic acid: CH₃CH(CH₃)COOH

2-Methylpropanoic acid; CH₃.CH(CH₃).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ vlt KC1 30°C 0.50M C K1=2.95 B2= 5.52 1982CKb (33254) 738

B3=.61

HOCl + NaClO_4 , 20°C, 0–12M, H₂O, K1, 3, 12, P3, 5, 83, 1080GBPs (33355), 738

HC2111-111-NaCl04-3086C-0-1CM-HI B2-E-66 1070BPA-(2225E)-712

UCC - N-0104-00000-0-10M-N - W1-S-54 - DS-1-01 - 1077EKM (PDS577)

K3=1.66

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U02++ EMF NaClO₄ 20°C 1.00M U K1=2.48 B2=4.87 1972MPa (33258) 742
B3=7.20

U02++ gl oth/un ? 0.10M U K1=3.40 B2=5.83 1969RRa (33259) 743
pH=1.5-3.5

C4H8O₂ HL CAS 107-92-6 (1118)
n-Butanoic acid; CH₃.CH₂.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp NaClO₄ 20°C 0.10M U K1=2.96 B2=5.62 1975KMb (33353) 744
K3=2.43
K4=2.23

U02++ EMF NaClO₄ 20°C 1.00M U K1=2.58 B2=4.71 1972MPa (33354) 745
B3=6.91

U02++ gl NaClO₄ 31°C 0.10M U K1=2.91 B2=4.53 1969RRa (33355) 746
pH=1.5-3.5

C4H8O₃ HL CAS 594-61-6 (81)
2-Hydroxy-2-methylpropanoic acid; (CH₃)₂C(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ 25°C 1.0M C M 2000SGa (33532) 747
B(UO2H-1L)=<-2
B(UO2H-1L2)=0.59
B(UO2H-2L2)=-5.23
B(UO2LF2)=10.96

B(UO2LF3)=12.96.

U02++ gl NaClO₄ 20°C 1.00M C T K1=3.18 B2=5.13 1974MTa (33533) 748
B3=6.67

U02++ EMF NaClO₄ 25°C 1.0M U K1=3.02 B2=4.85 1967TGa (33534) 749
K3=1.54

Method: quinhydrone electrode

C4H8O₃ HL CAS 965-70-8 (423)
2-Hydroxybutanoic acid; CH₃.CH₂.CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ 31°C 0.10M U K1=3.29 B2=4.99 1962CMb (33586) 750

U02++ dis NaClO₄ 20°C 1.0M U K1=3.58 B2=5.3 1962SBB (33587) 751
B3=7.01

C4H8O3 HL CAS 300-85-6 (30)
3-Hydroxybutanoic acid; CH₃.CH(OH).CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	NaClO ₄	20°C	1.00M	C			K1=2.38 B2=4.35 B3=6.25	1974MTa (33631)	752

U02++	gl	NaClO ₄	31°C	0.10M	U			K1=2.70 B2=4.10	1962CMB (33632)	753
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C4H8O3 HL CAS 591-81-1 (39)
4-Hydroxybutanoic acid; HO.CH₂.CH₂.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	NaClO ₄	20°C	1.00M	C			K1=2.34 B2=4.49 B3=6.28	1974MTa (33660)	754

C4H9N02 HL Aminoisobutyric CAS 144-90-1 (188)
2-Amino-2-methylpropanoic acid; H₂N.C(CH₃)₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	NaClO ₄	30°C	0.10M	U				1980RRa (33842)	755

U02++	EMF	NaClO ₄	31°C	0.10M	U			K1=7.72	1977RRa (33843)	756
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C4H9N02 HL 2-Aminobutyric CAS 2835-81-6 (571)
2-Aminobutanoic acid; CH₃.CH₂.CH(NH₂).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	NaClO ₄	30°C	0.10M	U				1980RRa (33925)	757

U02++	gl	KN03	25°C	0.10M	U	TIH		K1=6.48 B2=15.06	1980SSF (33926)	758
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U02++	vlt	NaClO ₄	30°C	0.10M	U				1979RRa (33927)	759
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$$K(U02+2HL)=2.12$$

C4H9N02 HL 4-Aminobutyric CAS 56-12-2 (574)
4-Aminobutanoic acid; H₂N.CH₂.CH₂.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	NaClO ₄	25°C	1.0M	U	H		K1=2.25 B2=4.02 B3=6.08	1987BRa (33984)	760

DH₁ = 10.6, DH(B2) = 15.5, DH(B3) = 13.0, DS₁ = 79, DS(B2) = 129, DS(B3)=160

U02++ gl NaClO₄ 30°C 0.10M U 1980RRa (33985) 761
K(UO₂+HL)=2.34
K(UO₂HL+HL)=2.15

U02++ vlt NaClO₄ 30°C 0.10M U 1979RRa (33986) 762
K(UO₂+2HL)=4.44

U02++ gl NaClO₄ 31°C 0.10M U K1=9.13 1976RRb (33987) 763

C4H₉N₀3 HL Threonine CAS 72-19-5 (48)
2-Amino-3-hydroxybutanoic acid; H₂N.CH(CH(OH).CH₃)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ vlt NaClO₄ 25°C 0.10M C K1=0.87 1986SPb (34331) 764

Method: polarography.

U02++ gl KN₃ 25°C 0.10M C K1=6.65 B2=12.08 1983NMB (34332) 765

U02++ gl NaClO₄ 30°C 0.10M U K1=7.30 B2=14.20 1973RSa (34333) 766

U02++ gl KC₁ 25°C 0.05M U TIH K1=6.35 B2=12.50 1973SCe (34334) 767

Data for 0.15 and 0.25 M KCl and 45 C. At I=0, B2=12.68. DH(K1)=-21
kJ mol⁻¹, DS(K1)=52 J K⁻¹ mol⁻¹; DH(K2)=-21, DS(K2)=48.

U02++ EMF oth/un 25°C 0.50M U K1=6.00 1973SKb (34335) 768

U02++ sp oth/un 25°C 0.50M U K1=5.95 1973SKb (34336) 769

C4H₁₀S₂ H₂L CAS 2150-02-9 (2896)
2,2'-Dimercaptoethyl ether; HS.CH₂CH₂.O.CH₂CH₂.SH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 25°C 40% U K1=12.60 1975SSe (34663) 770

At 35 C: K1=12.55

C4H₁₁N L Diethylamine CAS 109-89-7 (1331)
Diethylamine, 3-azapentane; (C₂H₅)₂NH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp non-aq 25°C 100% U K1=7.03 K2=<1 1989LMB (34821) 771

Medium: propylene carbonate, 0.1 M Et₄NClO₄

C4H₁₁NS HL CAS 108-02-1 (1792)
1-Mercapto-2-(N,N-dimethyl)aminoethane; HS.CH₂.CH₂.N(CH₃)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KN03 20°C 0.25M U I K1=8.00 B2=15.20 1973MSd (35140) 772
0.25 KN03, 25% MeOH: K1=9.88, K2=8.62; 25% EtOH: K1=10.19, K2=8.92

C5H2O2F6 HL HFA CAS 1522-22-1 (195)
1,1,1,5,5,5-Hexafluoropentane-2,4-dione; F3C.CO.CH2.CO.CF3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ dis oth/un 25°C 0.10M U B2=3.24 1970GRa (35929) 773

C5H4NOC1 L CAS 1121-76-2 (328)
4-Chloropyridine-N-oxide; C5H4N(O)Cl

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ sp non-aq 25°C 100% U M 1976DBa (36029) 774
 $K((U02A2)2+2L=2U02A2L2)=0.61$

HA=tropolone. Medium: benzene

C5H4N202 HL CAS 98-97-5 (1879)
Pyrazine-2-carboxylic acid; cyclo(-CH:CH.N:C(COOH).CH:N-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl NaClO4 25°C 0.50M C K1=2.45 B2=4.4 1989NMa (36066) 775

C5H4O2S HL 2-Thenoic acid CAS 527-72-0 (2312)
Thiophene-2-carboxylic acid; C4H3S.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl NaClO4 30°C 0.20M U T H K1=2.20 1976SKc (36266) 776
At 40 C:K1=2.13; 50 C:2.11

C5H5N L Pyridine CAS 110-86-1 (31)
Pyridine, Azine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ sp non-aq 25°C 100% U M 1976DBa (36689) 777
 $K((U02A2)2+2L=2U02A2L)=-1.66$

HA=tropolone. Medium: benzene

C5H5NOS (4389)
2-Mercaptopyridine N-oxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ sp NaClO4 25°C 0.10M C 1975EMa (36722) 778
B3=12.72

At pH 4.5, B3eff=12.41

C5H5N02 HL CAS 16867-04-2 (2316)

2,3-Dihydroxypyridine, 3-Hydroxypyridin-2(1H)-one; C5H3N(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	25°C	50%	U			K1=9.35 B2=17.62	1970GDa (36798)	779
Medium: 50% dioxan, 0.1 M NaClO4										
UO2++	gl	NaClO4	25°C	0.10M	U			K1=8.14 B2=14.96	1970GDa (36799)	780

C5H5N02 CAS 1121-47-7 (6252)

2-Furancarboxaldehyde oxime, 2-Furfuraldoxime; C4H3O.CH:NOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	20°C	60%	U	I		K1=11.23	1979GBd (36804)	781
UO2++	sp	diox/w	21°C	40%	U	I		K1=8.74	1978GMd (36805)	782

C5H5N5 L Adenine CAS 73-24-5 (237)

6-Aminopurine; H2N.C5H3N4

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KN03	35°C	0.10M	U	M		K1=8.38	1982RKa (36984)	783
								K(UO2(EDTA)+L)=2.88		
								K(UO2(EDTA)L+H)=6.65		

C5H6N6 HL Diaminopurine CAS 1904-98-9 (4290)

2,6-Diaminopurine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KN03	35°C	0.10M	U	M		K1=9.60	1982RKa (37340)	784
								K(UO2(EDTA)+L)=3.01		
								K(UO2(EDTA)L+H)=6.57		

C5H6OS HL Furfurylmercaptan; C4H3O.CH2.SH CAS 98-02-2 (4309)

Furfurylmercaptan; C4H3O.CH2.SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	alc/w	25°C	50%	U	T		K1=7.75 B2=15.09	1973SSF (37346)	785
Medium: 50% EtOH, 0.1 M NaClO4										
K1(15 C)=7.77, K1(35 C)=7.73, K2(15 C)=7.37, K3(35 C)=7.33										

C5H6O4 H2L Itaconic acid CAS 97-65-4 (398)

Methylenesuccinic acid; HOOC.CH2.C(:CH2).COOH

U02++ dis NaClO₄ 25°C 0.10M U K₁=6.8 B₂=13.10 1955RYb (38116) 797
B((UO₂)(OH)L) < 5.6
K(UO₂+L+HL)=8.7
K(UO₂+2L+HL)=14.8
K(UO₂+HL) < 0

Medium: H₂O-CHCl₃

U02++ gl diox/w 30°C 50% U K₁=9.32 B₂=16.92 1954BRC (38117) 798

C5H8O₄ H₂L CAS 595-46-0 (1144)

Dimethylmalonic acid; HOOC.C(CH₃)₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 40% M I K₁=9.40 B₂=16.00 1985ARC (38218) 799

Medium: 0.10 M NaClO₄ in 40% dioxane/H₂O. Also data for 20, 60 and 80% dioxane/H₂O.

U02++ gl KN₃ 25°C 0.50M U K₁=5.55 B₂=9.38 1969VOb (38219) 800

C5H8O₄ H₂L Glutaric acid CAS 110-94-1 (420)

Pentanedioic acid; HOOC.CH₂.CH₂.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ 25°C 0.10M M M K₁=4.07 1987NCa (38365) 801
K(UO₂(nta)+L)=2.85

U02++ gl NaClO₄ 31°C 0.10M U M K₁=3.70 B₂=6.39 1977SSb (38366) 802
B(UO₂L(Ala))=11.38
K(ML₂+M(Ala)₂-2ML(Ala))=1.10

U02++ gl NaClO₄ 20°C 1.00M U 1973CBc (38367) 803
K(UO₂+HL)=1.89
K(UO₂+2HL)=3.58
K(UO₂+L+HL)=4.01

U02++ gl KN₃ 25°C 0.50M U K₁=3.53 1969VOb (38368) 804
K(UO₂+HL)=2.30

C5H₉N₂ HL Proline CAS 147-85-3 (44)

Pyrrolidine-2-carboxylic acid; C₄H₈N.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ vlt KCl 30°C 0.10M U K₁=1.43 B₂=2.11 1978DKa (38650) 805

U02++ EMF oth/un 25°C 0.50M U K₁=7.54 1973SKb (38651) 806

U02++ sp oth/un 25°C 0.50M U K₁=7.72 1973SKb (38652) 807

U02++	EMF	oth/un	?	?	U	K1=10.45	1970FMb (38653) 808
U02++	gl	KCl	20°C	0.10M	U	K1=7.75	1970GVa (38654) 809

C5H9N03 HL Hydroxyproline CAS 51-35-4 (416)							
4-Hydroxy-2-pyrrolidinecarboxylic acid; C4H7N(OH)(COOH)							
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo
U02++	gl	KCl	25°C	0.05M	U TIH	K1=7.02 B2=13.84	1973SCe (38758) 810
Data for 0.15 and 0.25 M KCl and 45 C. At I=0, B2=13.90. DH(K1)=-29 kJ mol-1, DS(K1)=37 J K-1 mol-1; DH(K2)=-29, DS(K2)=33.							
U02++	EMF	oth/un	25°C	0.50M	U	K1=6.52	1973SKb (38759) 811
By spectrophotometry, K1=6.48							

C5H9N04 H2L Glutamic acid CAS 56-86-0 (22)							
2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH							
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo
U02++	sp	NaClO4	25°C	0.1M	C		2004GZa (39133) 812
K(U02+H+L)=7.78							
For 0.3 mol/L NaClO4 K(U02+H+L)=7.25; K(U02+2H+L)=10.51							
For 0.7 mol/L NaClO4 K(U02+H+L)=7.6.95; K(U02+2H+L)=9.79							
U02++	gl	NaNO3	25°C	1.10M	U M	K1=8.53 K(U02+Hegta+HL)=7.95 K(U02(Hegta)L+H)=5.99 K(U02(egta)L+H)=8.72	1995ADc (39134) 813
U02++	gl	NaClO4	30°C	0.10M	M M	K(U02+2HL)=5.19 K(U02+HL)=2.90 B(UO2AL)=11.45 B(UO2CL)=9.89	1995JSa (39135) 814
K(U02+2HL) by polarography. B(UO2DL)=8.80, B(UO2EL)=6.00. H2A is oxalic acid, H2C is malonic acid, H2D is succinic acid, H2E is lactic acid							
U02++	gl	NaClO4	25°C	0.10M	C	K1=8.25 B(UO2HL)=12.40	1982PMa (39136) 815
U02++	vlt	NaClO4	25°C	0.10M	C	K1=3.10	1980SKd (39137) 816
Method: polarography.							
U02++	gl	NaClO4	30°C	0.10M	U M	K1=8.43	1978SJa (39138) 817
U02++	gl	KNO3	25°C	0.10M	U	K1=8.25 B2=14.75	1976GPd (39139) 818

U02++ gl KN03 25°C 0.20M U 1963FKa (39140) 819
 K(UO2+HL)=2.66
 ****=
 C5H9N04 H2L MIDA CAS 4408-64-4 (190)
 N-Methyliminodiethanoic acid; CH₃.N(CH₂.COOH)2

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl KN03 25°C 0.10M U K1=9.70 1970FSa (39289) 820
 K(UO2(OH)L+H=UO2L)=5.92
 K(2UO2(OH)L=(UO2)2(OH)2L2)=3.41
 K(2UO2L+2H2O=(UO2)2(OH)2L2+2H)=-8.43
 ****=
 C5H9N3 L Histamine CAS 51-45-6 (103)
 4(5)-(2'-Aminoethyl)imidazole; C3H3N2.CH₂.CH₂.NH2

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ sp NaNO3 25°C 0.10M U K1=7.5 1993GAa (39548) 821
 ****=
 C5H9N3O4S H2L CAS 16907-58-7 (2106)
 Thiosemicarbazone-diethanoic acid; H₂N.CS.NH.N(CH₂.COOH)2

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ sp NaClO4 25°C 0.05M U K1=5.54 1987CDa (39575) 822
 ****=
 C5H10N2O3 HL Glutamine CAS 56-85-9 (18)
 2-Aminopentanedioic acid 5-amide; H₂N.CH(CH₂.CH₂.CO.NH₂).COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl NaClO4 30°C 0.10M M 1995JSa (39840) 823
 K(UO2+2HL)=2.70
 K(UO2+HL)=1.90
 K(UO2+2HL) by polarography.

 U02++ gl NaClO4 25°C 0.10M U K1=6.63 B2=12.85 1973TSe (39841) 824
 ****=
 C5H10O2 HL IsoValeric acid CAS 503-74-2 (1311)
 3-Methyl-butanoic acid, Isovaleric acid; (CH₃)₂CH.CH₂.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ sp NaClO4 20°C 0.10M U K1=2.90 B2=4.81 1975KMb (40185) 825
 ****=
 C5H10O2 HL n-Valeric acid CAS 109-52-4 (3027)
 Pentanoic acid; CH₃(CH₂)₃.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp NaClO4 20°C 0.10M U K1=2.91 B2=5.43 1975KMb (40204) 826

C5H11N02 HL Valine CAS 72-18-4 (43)
2-Amino-3-methylbutanoic acid; H2N.CH(CH(CH3)2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ oth NaClO4 35°C 0.10M U M K1=8.02 B2=14.53 1984SYa (40765) 827
B(UO2(NTA)+L)=5.18

Method: paper electrophoresis

U02++ vlt KCl 30°C 0.50M C K1=1.55 1982CKb (40766) 828
Method: polarography.

U02++ gl KN03 25°C 0.10M U K1=7.10 B2=14.72 1982NMa (40767) 829

U02++ gl NaClO4 30°C 0.10M U T 1980RRa (40768) 830
K(UO2+HL)=2.01

U02++ EMF oth/un ? ? U K1=8.60 1970FMb (40769) 831

C5H11N02 HL DL-Valine CAS 516-06-3 (186)
DL-2-Amino-3-methylbutanoic acid; H2N.CH(CH(CH3)2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KN03 25°C 0.10M C K1=7.10 B2=14.72 1983Nmb (40898) 832

U02++ EMF NaClO4 31°C 0.10M U K1=7.97 1977RRa (40899) 833

C5H11N02S HL Methionine CAS 63-68-3 (42)
2-Amino-4-(methylthio)butanoic acid; H2N.CH(CH2.CH2.S.CH3).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KN03 25°C 0.10M U K1=6.41 B2=13.38 1982NMa (41129) 834

U02++ gl NaClO4 30°C 0.10M U K1=7.65 B2=13.95 1973RSa (41130) 835

U02++ gl KCl 25°C 0.10M U T K1=6.52 B2=11.88 1971SSc (41131) 836
K1(35 C)=6.35, K1(45 C)=6.14, B2(35 C)=11.55, B2(45 C)=11.24

C5H11NS2 HL CAS 147-84-2 (2126)
Diethyldithiocarbamic acid; (CH3.CH2)2N.CSSH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp NaClO4 25°C 1.0M U 1956ZIa (41374) 837
B4=17.2 to 17.8

C6H4N204 H2L CAS 89-01-0 (5801)
Pyrazine-2,3-dicarboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 25°C 0.50M C K1=3.10 B2=5.40 1989NMa (42207) 838

C6H4O4 H2L CAS 615-94-1 (1280)
2,5-Dihydroxy-1,4-benzoquinone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KCl 30°C 25% M TIH K1=7.37 B2=14.07 1991GDe (42310) 839

Medium: 35% Dioxan/H2O, 0.1 M NaClO4. Other solvents and backgrounf concs.

C6H4O5 H2L Comenic acid CAS 499-78-5 (2544)
3-Hydroxypyran-4-one-6-carboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ oth oth/un 20°C 0.10M U B2=6.86 1972DVa (42321) 840

C6H4O6 H4L CAS 5678-48-2 (871)
Tetrahydroxy-1,4-benzoquinone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ EMF NaClO4 30°C 0.10M U K1=7.00 B2=10.00 1981HIa (42328) 841

C6H5N02 HL Picolinic acid CAS 98-98-6 (391)
2-Pyridine-carboxylic acid; C5H4N.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KN03 25°C 0.20M U T HM 1986KMc (42612) 842

K(U02(ida)+L)=4.52

K(U02(edda)+L)=3.96

K(U02(nta)+L)=4.42

Data for 5, 45 C. DH(U02(ida)L)=-14 kJ mol-1, DS(U02(ida)L)=38 J K-1 mol-1
DH(U02(edda)L)=-16, DS(U02(edda)L)=25; DH(U02(nta)L)=-20, DS(U02(nta)L)=17

U02++ gl NaClO4 25°C 0.10M U K1=4.51 1970ERa (42613) 843

K(U02HL=U02L+H)=-1

C6H5N02 HL Nicotinic acid CAS 59-67-6 (419)
3-Pyridine-carboxylic acid; C5H4N.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ g1 KN03 25°C 0.10M U K1=9.56 B2=18.56 1988ZMa (42690) 844
K3=8.10

C6H5N03 HL 4-Nitrophenol CAS 100-02-7 (454)
4-Nitrohydroxybenzene; HO.C6H4.N02

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ g1 KN03 20°C 0.10M U K1=4.40 1967BAb (42818) 845

C6H5N03 HHL CAS 824-40-8 (878)
Pyridine-2-carboxylic acid N-oxide (Picolinic acid N-oxide); C5H4N(0)COO

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ EMF oth/un 25°C 0.10M U K1=3.62 1970ROa (42842) 846

C6H5N04 H2L 4-Nitrocatechol CAS 3316-09-4 (890)
1,2-Dihydroxy-4-nitrobenzene; O2N.C6H3(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ g1 KN03 20°C 0.10M U K1=12.9 B2=22.70 1967BAb (42945) 847
K(U02L+H)=2.7
K(U02L2+H)=4.97

C6H5O4C1 HL Chlorokojic aci (3086)
3-Chloro-5-hydroxy-2-hydroxymethyl-4-pyrone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ g1 diox/w 25°C 75% U K1=9.93 1960KFc (43138) 848

C6H6N20 HL CAS 873-69-8 (1258)
Pyridine-2-aldoxime; C5H4N.CH:NOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp diox/w 21°C 40% U I K1=7.12 1978GMd (43302) 849

C6H6N202 HL Cupferron CAS 135-20-6 (637)
N-Nitrosophenylhydroxylamine; C6H5.N(OH).NO

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp oth/un 26°C 0.0 U B2=11.0 1961KKa (43427) 850
Ks(NH4+U02L2+L)=-9.2

Ks by solubility

C6H6N202 HL CAS 5657-61-4 (1430)

Nicotinylhydroxamic acid; C5H4N.CO.NH.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	NaClO4	30°C	0.10M	U			K1=7.50 B2=14.65	1969DSb (43438)	851

C6H6N203 HL CAS 99-57-0 (469)
2-Amino-4-nitrophenol; H2N.C6H3(OH)(NO2)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	30°C	50%	U			K1=7.59 B2=14.72	1966VMa (43447)	852

Medium: 50% dioxan, 0.1 M NaClO4

C6H6O HL Phenol CAS 108-95-2 (457)
Hydroxybenzene, phenol; C6H5.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KN03	20°C	0.10M	U			K1=5.8	1965BSd (43547)	853

C6H6O2 H2L Catechol CAS 120-80-9 (534)
1,2-Dihydroxybenzene, pyrocatechol; H0.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KN03	25°C	0.20M	U	M		K1=13.85 B2=23.43 K(UO2(IMDA)+L)=12.40 K(UO2(NTA)+L)=11.86 K(UO2(HEDTA)+L)=11.36 K(UO2(EDTA)+L)=10.83 K(UO2(CDTA)+L)=10.76, K(UO2(DTPA)+L)=9.49	1990SSc (43855)	854

UO2++	gl	NaClO4	25°C	0.20M	U	M	K1=14.90	1986SLb (43856)	855
							K(UO2(ida)+L)=13.80		
							K(UO2(nta)+L)=13.40		
							K(UO2(edta)+L)=10.42		

UO2++	gl	KN03	25°C	0.10M	U	M	K1=13.23	1985VSb (43857)	856
							B(UO2AL)=17.28		
							K(UO2A+L)=12.15		
							K(UO2L+A)=4.05		

H2A=phthalic acid

UO2++	gl	KN03	20°C	0.10M	U	I	K1=15.9	1965BSd (43858)	857
							K(UO2+HL)=6.2		
							K(UO2L+HL)=4.9		
							K(UO2HL2+HL)=3.7		

By spectrophotometry, 0.1 M NaClO4: K1=15.9, K(UO2+HL)=6.3, K(UO2L+HL)=4.9

U02++ sp oth/un ? 0.0 U 1963SGb (43859) 858
 $K(UO_2L+H_2L=UO_2L_2+2H)=-10.5$
 $K(UO_2L+H)=3.76$

C6H6O2 H2L Resorcinol CAS 108-46-3 (3645)
1,3-Dihydroxybenzene; HO.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	NaClO4	25°C	0.20M	U	M		K1=9.66 $K(UO_2(ida)+L)=8.96$ $K(UO_2(nta)+L)=8.64$ $K(UO_2(edta)+L)=6.71$	1986SLb (43890)	859
U02++	gl	KNO3	20°C	0.10M	U			K1=16.9	1966BRC (43891)	860
U02++	sp	oth/un	?	?	U			$K(UO_2+HL)=6.0$	1966GSb (43892)	861
U02++	gl	KNO3	20°C	0.10M	U			$K(UO_2+HL)=5.9$	1965BSd (43893)	862

C6H6O2 H2L Hydroquinone CAS 123-31-9 (3646)
1,4-Dihydroxybenzene; HO.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	oth/un	25°C	?	U			K1=10.32	1967RBa (43898)	863
C6H6O3	H3L	Pyrogallol						CAS 87-66-1 (696)		
1,2,3-Trihydroxybenzene; C6H3(OH)3										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	sp	oth/un	25°C	?	U			K1=27.19	1989WZa (43990)	864
U02++	gl	NaClO4	25°C	0.20M	U	M		K1=13.81 $K(UO_2(ida)+L)=13.55$ $K(UO_2(nta)+L)=12.85$ $K(UO_2(edta)+L)=10.31$	1986SLb (43991)	865

U02++ gl KNO3 25°C 0.10M U 1965BAb (43992) 866
 $K(2UO_2+H_3L=(UO_2)_2L+3H)=-6.84$
 $K(UO_2L+H_3L=UO_2H_2L_2+H)=-4.69$

C6H6O3 H3L Phloroglucinol CAS 6099-90-7 (2525)
1,3,5-Trihydroxybenzene; C6H3(OH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl NaClO₄ 25°C 0.20M U M K1=8.34 1986SLb (44027) 867
 K(UO₂(ida)+L)=7.46
 K(UO₂(nta)+L)=7.08
 K(UO₂(edta)+L)=5.28

U02++ sp none 25°C 0.0 C 1983EEa (44028) 868
 K(UO₂+H₂L)=6.67

Medium pH 3.5. Extrapolated from data for I=0.15-0.25 M. K(H₂L+H)=8.45.

C6H603 HL Maltol CAS 118-71-8 (2442)

3-Hydroxy-2-methyl-4H-pyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl KN03 25°C 0.10M U K1=8.18 B2=14.76 1969CBb (44102) 869
 K3=3.51

U02++ sp NaClO₄ 25°C 0.10M U K1=8.3 B2=15.00 1968CHc (44103) 870
 K3=3.26

C6H604 HL Kojic acid CAS 501-30-4 (1800)

5-Hydroxy-2-(hydroxymethyl)-4H-pyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl KN03 25°C 0.20M U T HM 1986KMc (44249) 871
 K(UO₂(ida)+L)=6.98
 K(UO₂(edda)+L)=6.50
 K(UO₂(nta)+L)=7.03

Data for 5, 45 C. DH(UO₂(ida)L)=-17 kJ mol⁻¹, DS(UO₂(ida)L)=75 J K⁻¹ mol⁻¹
 DH(UO₂(edda)L)=-15, DS(UO₂(edda)L)=71; DH(UO₂(nta)L)=-18, DS(UO₂(nta)L)=71

U02++ vlt NaNO₃ 20°C 1.0M U M K1=7.27 1967HAa (44250) 872
 K(UO₂L+A)=1.57

HA=formic acid

U02++ gl KN03 20°C 0.10M U 1966BRC (44251) 873
 K(UO₂+HL=UO₂L+H)=-0.5
 K(UO₂L+HL=UO₂L₂+H)=-2.10
 K(UO₂L₂+HL=UO₂L₃+H)=-4.33

U02++ sp NaClO₄ 20°C 0.10M U K1=7.05 B2=12.66 1966SKb (44252) 874
 K3=3.53

U02++ gl KN03 20°C 0.10M U I K1=7.2 B2=12.65 1965BSd (44253) 875
 K3=3.4

By spectrophotometry, 0.1 M NaClO₄: K1=7.05, K2=5.4, K3=3.5

U02++ gl diox/w 30°C 75v% U K2=10.23 1960KFc (44254) 876

U02++ gl diox/w 30°C 50% U K1=10.1 B2=17.5 1954BFa (44255) 877

C6H6O5S H3L CAS 7134-09-0 (3687)
3,4-Dihydroxybenzenesulfonic acid; (HO)2.C6H3.SO3H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	KNO ₃	20°C	0.10M	U				1965BSb (44287)	878
								K((UO ₂) ₂ L ₂ O ₄ H ₃)=9.0		
U02++	gl	KNO ₃	20°C	0.10M	U				1965BSD (44288)	879
								K(UO ₂ +HL)=6.4		

C6H6O8S ₂		H4L	Tiron				CAS	149-45-1 (104)		
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)2.C6H ₂ (SO ₃ H) ₂										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	KNO ₃	25°C	0.20M	U	M	K1=14.29	B2=26.58	1990SSc (44505)	880
							K(UO ₂ (IMDA)+L)=12.78			
							K(UO ₂ (NTA)+L)=12.41			
							K(UO ₂ (HEDTA)+L)=12.10			
							K(UO ₂ (EDTA)+L)=11.14			
K(UO ₂ (CDTA)+L)=10.99, K(UO ₂ (DTPA)+L)=10.03										

U02++ sp NaClO₄ 20°C 0.10M U 1965BSb (44506) 881
K((UO₂)₂L₂O₄H₃)=8.9

U02++ gl KNO₃ 20°C 0.10M U 1965BSD (44507) 882
K(UO₂+HL)=6.3

U02++ sp NaClO₄ 20°C 0.10M U 1965SSc (44508) 883
K(UO₂+HL)=6.5

U02++ gl KNO₃ 25°C 0.10M U K1=15.90 1958GRd (44509) 884

C6H7NO HL 2-Aminophenol CAS 95-55-6 (2868)
2-Amino-1-hydroxybenzene; HO.C6H₄.NH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	KNO ₃	25°C	0.10M	U	M	K1=11.40	B2=21.23	1985VSb (44940)	885
							B(UO ₂ A)=14.98			
							K(UO ₂ A+L)=9.85			
							K(UO ₂ L+A)=3.58			

H2A=phthalic acid

C6H7NO L CAS 1003-67-4 (331)
4-Methylpyridine-N-oxide; C5H₄N(O)CH₃

$$K(UO_2LH=UO_2LH(OH)_2+2H) = -10.92$$

$$K(UO_2(LH)_2=UO_2(LH)_2OH+H) = -5.08$$

$$K(UO_2+LH+2OH) = 19.4$$

C6H8O6S H3L CAS 99-68-3 (3692)
(Carboxymethylthio)butanedioic acid; HOOC.CH(S.CH₂.COOH).CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	NaClO ₄	30°C	0.10M	M	I		K ₁ =4.65 B ₂ = 8.06	1985ARc (45717)	893

Also data for 20-80% dioxane/H₂O. For 40% dioxane/H₂O, K₁=8.50, K₂=4.25.

UO ₂ ++	gl	NaClO ₄	30°C	0.10M	U	I		K ₁ =4.65 B ₂ =8.06	1983ASa (45718)	894
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UO ₂ ++	gl	KNO ₃	25°C	0.05M	M			K ₁ =4.55	1975DPb (45719)	895
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C6H8O7 H3L Citric acid CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH₂.CH(OH)(COOH).CH₂COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	NaClO ₄	20°C	0.05M	C			K ₁ =3.93	1989RAb (46295)	896

Medium: 0.05 M NaClO₄/HClO₄.

UO ₂ ++	sp	oth/un	25°C	>1.0	U			K ₁ =7.17 B((UO ₂ L) ₂)=17.00 B(UO ₂ HL)=9.68 B(UO ₂ H ₂ L)=11.43	1980VKa (46296)	897
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Medium: 1 M (H, Na, (UO₂)_{0.5})₃L

UO ₂ ++	ix	oth/un	25°C	0.10M	U			K(UO ₂ +H ₃ L)=2.79 K(UO ₂ +H ₂ L)=4.25 K(UO ₂ +HL)=7.25	197500a (46297)	898
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I=0.1(Na citrate)

UO ₂ ++	dis	oth/un	25°C	?	U			K(UO ₂ +H ₃ L=UO ₂ L+3H)=-6.30 K(2UO ₂ +2H ₃ L=(UO ₂)L ₂ +6H)=-6.59	1972MKc (46298)	899
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UO ₂ ++	dis	oth/un	25°C	pH 4	U	M		K(2UO ₂ L=(UO ₂ L) ₂) > 6	1970AKa (46299)	900
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Keff(InL₂+0.5(UO₂L)₂=InUO₂L₂+L)=2.86

UO ₂ ++	gl	KNO ₃	25°C	1.0M	U	I		K ₁ =6.9 K(2UO ₂ +2L)=17.70 K'(2UO ₂ L=(UO ₂) ₂ L ₂)=4.0	1965RMa (46300)	901
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At I=0.1 M: K₁=7.4, K=18.87, K'=4.1

K(3(UO₂)₂L₂=(UO₂)₂L₂[(OH)₅(UO₂)₂L₂]₂+10H)=-47.9

U02++ gl KN03 25°C 0.10M U I K1=7.40 B2=18.87 1965RMa (46301) 902
K(2U02L=(U02L)2)=4.07

I=1.0 M: K1=6.87, B2=17.70, K=3.96

U02++ dis NaClO4 20°C 1.0M U B2=11.2 1962SBb (46302) 903

U02++ gl KN03 25°C .136M U I 1960FNa (46303) 904
K(2U02+2HL=(U02HLOH)2+2H)=7.68

At I=0.05 M K=9.04

U02++ gl oth/un 25°C 0.15M U K1=8.5 1959LLa (46304) 905

C6H9N06 H3L NTA CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH2.COOH)3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KN03 5°C 0.10M U TI K1=9.94 1987AKb (47073) 906

Data for 25, 45 C, I=0.05-0.2 M KN03. Also data for 10-40% MeOH/H2O and EtOH/H2O, 0.20 M KN03, 25 C.

U02++ gl NaClO4 25°C 0.20M U K1=13.44 1986SLb (47074) 907

U02++ oth NaClO4 35°C 0.10M C K1=9.85 1986SYa (47075) 908
Method: paper electrophoresis. Medium pH 8.5.

U02++ oth NaClO4 35°C 0.10M C M K1=9.85 1985SGc (47076) 909
K(U02L+his)=5.31

Method: paper electrophoresis. Medium pH 8.5.

U02++ gl NaClO4 25°C 3.0M C 1984BLb (47077) 910
B((U02)HL)=12.19

U02++ oth NaClO4 35°C 0.10M U K1=9.85 1984SYa (47078) 911
Method: paper electrophoresis

U02++ dis oth/un 20°C 0.10M U K1=7.88 1968MTa (47079) 912
Method: paper electrophoresis

U02++ dis NaClO4 20°C 0.10M U T K1=9.56 1963STc (47080) 913

C6H9N302 HL Histidine CAS 71-00-1 (1)
2-Amino-3-(4'-imidazolyl)propanoic acid; H2N.CH(CH2.C3H3N2)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KN03 35°C 0.10M U 1997RVa (47622) 914
K(U02+HL)=4.56

U02++ oth NaClO4 35°C 0.10M C K1=8.70 B2=14.05 1985SGc (47623) 915

Method: paper electrophoresis. Medium pH 8.5.

U02++ gl oth/un 25°C 0.20M U K1=7.71 1957LDa (47624) 916

C6H10N2O5 H2L ADA CAS 26239-55-4 (2747)
N-(2-Acetamido)iminodiethanoic acid; H2N.CO.CH2.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KN03 25°C 0.10M C M K1=7.05 2003AHa (47856) 917
K((UO2)L+A)=3.99

HA is 3-amino-5-mercaptop-1,2,4-triazole.

U02++ gl KN03 25°C 0.10M M M K1=6.89 1996AEa (47857) 918
Data for ternary complexes with dipicolinic acid

C6H1003 HL CAS 141-97-9 (3068)

Ethyl acetoacetate; CH3.CO.CH2.CO2.C2H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 75% U K1=12.48 1973AAa (48019) 919

C6H1004 H2L Adipic acid CAS 124-04-9 (401)
1,6-Hexanedioic acid; HOOC.(CH2)4.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 25°C 0.10M M M K1=4.08 1987NCa (48094) 920
K(UO2(nta)+L)=2.98

U02++ gl KN03 25°C 0.20M U T 1985KMc (48095) 921
K(UO2A+L)=4.10

H2A=iminodiacetic acid; 5 C:K=4.28; 45 C: K=3.86, DH=-18.8 kJ mol-1,
DS=17 J K-1 mol-1

U02++ oth oth/un 40°C 0.10M U K1=11.8 1981SSe (48096) 922

Method: Paper electrophoresis.

U02++ gl NaClO4 30°C 0.10M U K1=4.08 1973KJa (48097) 923

U02++ gl KN03 25°C 0.50M U K1=3.54 1969VOb (48098) 924
K(UO2+HL)=2.38

U02++ gl NaClO4 31°C 0.10M U K1=4.08 1968RSa (48099) 925

C6H1004S H2L CAS 111-17-1 (139)
3,3'-Thiodipropanoic acid; HOOC.CH2.CH2.S.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 30°C 0.10M U T K1=4.04 1981SJb (48195) 926
At 20 C: K1=3.99; 40 C: 4.10

U02++ gl NaClO4 30°C 0.10M U M K1=4.04 1981SJc (48196) 927
B((U02)L(malonate))=9.33
B((U02)L(succinate))=8.29
B((U02)L(itaconate))=8.53
B((U02)L(glutarate))=7.53

B(M+L+adipic acid)=7.43.

U02++ gl NaClO4 30°C 0.10M U M K1=4.04 1978SJa (48197) 928
B((U02)L(Asp))=9.76
B((U02)L(Glu))=9.49

U02++ gl KN03 25°C 0.05M M K1=3.90 1975DPb (48198) 929

U02++ vlt alc/w 30°C 30% U I K1=0.74 B2=0.08 1972RGc (48199) 930
B3=1.53

Medium: 0-30% MeOH, 0.1 M KC1. 0%: K1=0.52, B2=0.08, B3=0.93

C6H1004S2 H2L CAS 7244-02-2 (438)
1,2-Bis(carboxymethylthio)ethane; HOOC.CH2.S.CH2.CH2.S.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 25°C 0.50M U K1=3.06 B2=4.85 1980NAa (48251) 931

C6H1006 H2L CAS 23243-68-7 (242)
1,2-Bis(carboxymethoxy)ethane; HOOC.CH2.O.CH2.CH2.O.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 25°C 1.00M U H K1=3.06 B2=5.22 1986BSb (48359) 932
B((U02)HL)=5.51
B((U02)HL2)=8.34

U02++ gl NaClO4 25°C 1.0M U H K1=3.08 1984TAb (48360) 933
By calorimetry: DH(K1)=26.6 kJ mol-1, DS(K1)=148.4 J K-1 mol-1.

C6H1007 HL Galacturonic CAS 685-73-4 (290)
D-Galacturonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 25°C 1.00M U B2=6.19 1990DGb (48396) 934
B((U02)H-2L2)=-2.03
B((U02)H-3L3)=-4.724

C6H11N02 HL CAS 2044-64-6 (4374)

N,N-Dimethylacetamide; CH₃.CO.CH₂.CO.N(CH₃)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	20°C	50%	U			K1=12.10 B2=22.14	1969KSd	(48542) 935
Medium:	50% dioxan,	0.025 M	NaClO ₄							

C₆H₁₁N₀5 H₂L HIMDA CAS 93-62-9 (192)
N-(2-Hydroxyethyl)iminodiethanoic acid; HO.CH₂.CH₂.N(CH₂.COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	KNO ₃	25°C	0.10M	U			K1=8.34	1970FSa	(48806) 936
								K(UO ₂ (OH)L+H=UO ₂ L)=5.86		
								K(2UO ₂ (OH)L=(UO ₂) ₂ (OH) ₂ L ₂)=3.40		
								K(2UO ₂ L+2H ₂ O=(UO ₂) ₂ (OH) ₂ L ₂ +2H)=--8.32		

UO ₂ ++	gl	KNO ₃	25°C	0.10M	U	I		K1=8.32	1964RMc	(48807) 937
								K(H+UO ₂ OH _L)=5.92		
								K(2UO ₂ OH _L =(UO ₂ OH _L) ₂)=3.50		

In 1 M KNO₃ K1=7.99, K(UO₂OH_L+H)=5.87, K(2UO₂OH_L=(UO₂OH_L)₂)=3.65

C₆H₁₂N₂O₄ H₂L EDDA CAS 5657-17-0 (119)
1,2-Diaminoethane-N,N'-diethanoic acid; HOOC.CH₂.NH.CH₂.CH₂.NH.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	NaClO ₄	25°C	1.00M	U			K1=11.5	1986BSb	(49277) 938
UO ₂ ++	gl	NaClO ₄	25°C	3.0M	C			K1=16.02	1984BLb	(49278) 939
UO ₂ ++	gl	KNO ₃	25°C	0.10M	U			K1=11.41	1970FSa	(49279) 940
								K(UO ₂ (OH)L+H=UO ₂ L)=5.96		

C₆H₁₂N₂O₄ H₂L CAS 4726-83-4 (5911)

N,N-Dihydroxyhexanediamide; HN(OH).CO.(CH₂)₄.CO.NH(OH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	NaNO ₃	25°C	0.10M	C			K1=13.27	1989EHa	(49337) 941
								B((UO ₂)L)=17.50		

C₆H₁₂N₄ L Methenamine CAS 100-97-0 (619)

Hexamethylenetetramine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	vlt	oth/un	25°C	1.0M	C T H			K1=2.50 B2= 3.80	1980PSc	(49386) 942
								B3=-1.14		
								B4=1.25		

Method: polarography. Medium: 1.0 M potassium acetate. Also data at 30 and 35 C. At 30 C, DH(K1)=-15.1 kJ mol-1, DS(K1)=-50 J K-1 mol-1.

C6H13N02 HL Isoleucine CAS 73-32-5 (424)
2-Amino-3-methylpentanoic acid; CH₃.CH₂.CH(CH₃).CH(NH₂).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KN03 25°C 0.10M U K1=7.02 B2=14.66 1982NMa (49918) 943

C6H13N02 HL Leucine CAS 61-90-5 (47)
2-Amino-4-methylpentanoic acid; H₂N.CH(CH₂.CH(CH₃)₂).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ vlt KCl 30°C 0.50M C K1=1.50 1982CKb (50115) 944

Method: polarography.

U02++ gl KN03 25°C 0.10M U K1=7.13 B2=14.36 1982NMa (50116) 945

U02++ gl KCl 25°C 0.10M U K1=5.60 B2=13.20 1982ZZa (50117) 946

U02++ gl KCl 25°C 0.10M U T T K1=6.83 B2=12.49 1971SSc (50118) 947
K1(35 C)=6.58, K1(45 C)=6.17, B2(35 C)=11.83, B2(45 C)=11.23

U02++ EMF oth/un 25°C 0.10M U K1=8.60 1970FMb (50119) 948

C6H13N02 HL Norleucine CAS 616-06-8 (602)
2-Aminohexanoic acid (2-Aminocaproic acid) CH₃.(CH₂)₃.CH(NH₂).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ 20°C 0.10M U T H K1=7.81 B2=14.73 1983SDc (50197) 949
Data for 30 and 40 C. DH(B2)=-43.4 kJ mol-1, DS(B2)=133 J K-1 mol-1.

C6H13N04 HL Bicine CAS 150-25-4 (2124)
N,N-Bis(2-hydroxyethyl)glycine; (HO.CH₂.CH₂)₂N.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ 30°C 0.10M U T H K1=6.20 B2=11.38 1980SGh (50416) 950
Also data at 20 and 40 C. DH(B2)=-66.9 kJ mol-1, DS(B2)=-4.8J K-1 mol-1.

C6H13N05 HL Tricine CAS 5704-04-1 (1239)
N-(Tris(hydroxymethyl)methyl)glycine; (HO.CH₂)₃C.NH.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KN03 25°C 0.10M C M K1=6.79 2003AHA (50511) 951
K((U02)L+A)=3.65

HA is 3-amino-5-mercaptop-1,2,4-triazole.

U02++ gl KNO₃ 25°C 0.10M M I K1=6.84 B2=13.11 1997EAa (50512) 952
Also values in 40% w/w ethanol, DMF, dioxane, acetonitrile.

C6H13N06 HL CAS 84518-56-9 (4387)

2-Amino-2-deoxy-D-gluconic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ 25°C 1.00M C M K1=7.01 B2=13.36 1991DGa (50537) 953
B(U02AL)=11.43
B(U02H-1AL)=7.40

HA=D-galacturonic acid.

C6H14N202 HL Lysine CAS 56-87-1 (41)
2,6-Diaminohexanoic acid; H₂N.(CH₂)₄.CH(NH₂)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO₄ 20°C 0.10M U T K1=7.90 B2=14.90 1986SHa (50839) 954
Data for 20-40 °C.

C6H16N202 L CAS 929-59-4 (915)
3,6-Dioxaoctane-1,8-diamine; H₂N.CH₂.CH₂.O.CH₂.CH₂.O.CH₂.CH₂.NH₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp non-aq 25°C 100% U K1=3.81 1989LMb (51702) 955
Medium: propylene carbonate, 0.1 M Et₄NClO₄

C6H17N3 L CAS 56-18-8 (968)
1,5,9-Triazanonane, 4-azaheptane-1,7-diamine; H₂N.CH₂.CH₂.CH₂.NH.CH₂.CH₂.CH₂.NH₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp non-aq 25°C 100% C H K1=0.75 1995CBa (51903) 956
Medium: DMSO, 0.1 M NEt₄ClO₄. DH=-23.4 kJ mol⁻¹, DS=-64 J K⁻¹ mol⁻¹.
Method: FTIR and calorimetry.

C6H18N206P2 H4L (1363)
N,N'-Dimethyldiaminoethane-N,N'-dimethylphosphonic acid;
CH₃N(CH₂PO₃H₂).CH₂.CH₂.N(CH₂PO₃H₂)CH₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl oth/un 25°C 0.10M U K1=14.9 1976MDa (51955) 957
K(U02+HL)=9.9

C6H18N30P L HMPA CAS 680-31-9 (603)

Hexamethylphosphoramide, Tris-(dimethylamino)phosphine oxide;((CH₃)₂N)₃PO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	non-aq	20°C	100%	U				1983KBc (51988)	958
								K(UO ₂ C ₁₂ +L)=5.06		

Medium: acetone

C₆H₁₈N₄ L Trien-tetramine CAS 112-24-3 (11)
1,4,7,10-Tetraazadecane; H₂N.CH₂.CH₂.NH.CH₂.CH₂.NH.CH₂.CH₂.NH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO ₂ ++	EMF	non-aq	25°C	100%	C	H	K1=7.63		1995CBa (52132)	959
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Medium: DMSO, 0.1 M NEt₄ClO₄. DH=-70.0 kJ mol⁻¹, DS=-89 J K⁻¹ mol⁻¹.

Method: Ag electrode and calorimetry.

C₇H₄N₂O₇ H₂L CAS 609-99-4 (400)
3,5-Dinitrosalicylic acid; (O₂N)₂.C₆H₂(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO ₂ ++	gl	NaClO ₄	30°C	0.10M	U	M	K1=6.39	B2=11.13	1973KJa (52504)	960
							K(UO ₂ +L+A)=8.80			
							K(UO ₂ +L+B)=9.88			
							K(UO ₂ +L+C)=8.20			
							K(UO ₂ +L+D)=13.20			

H₂A=succinic acid, H₂B=phthalic acid, H₂C=adipic acid, H₃D=5-sulfosalicylic acid

UO ₂ ++	gl	oth/un	35°C	dil	U		K1=7.0	B2=12.50	1970DDc (52505)	961
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UO ₂ ++	gl	KNO ₃	20°C	0.10M	U		K1=7.55	B2=13.05	1967BAa (52506)	962
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C₇H₄O₃Br₂ H₂L CAS 3147-55-5 (1116)
3,5-Dibromosalicylic acid; C₆H₂(OH)(Br)₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO ₂ ++	gl	NaClO ₄	31°C	0.10M	U		K1=9.80	B2=17.34	1973JKa (52544)	963
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C₇H₄O₃C₁₂ H₂L CAS 320-72-9 (1117)
3,5-Dichlorosalicylic acid; C₆H₂(OH)(Cl)₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO ₂ ++	gl	NaClO ₄	31°C	0.10M	U		K1=9.52	B2=16.73	1973JKa (52556)	964
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C₇H₄O₇ H₃L Meconic acid CAS 497-59-6 (3723)
3-Hydroxy-4-pyrone-2,6-dicarboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	alc/w	20°C	30%	U	I		K1=12.4 B2=21.40 K(UO2+H3L=UO2H2L+H)=1.0	1966SKb (52566)	965
Medium: 30% EtOH, 0.1 M NaClO4. 0%: K1=11.8, K2=8.9, K=0.6										
UO2++	sp	NaClO4	20°C	0.10M	U			K1=12.5 B2=21.0	1965BSd (52567)	966

C7H5N04		H2L	Quinolinic acid	CAS	89-00-9	(567)				
2,3-Pyridinedicarboxylic acid; C5H3N.(COOH)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	NaClO4	25°C	0.50M	C			K1=4.72	1989NMa (52631)	967
UO2++	gl	NaClO4	30°C	0.10M	U	M		K1=4.65 B2=8.30 K(UO2+HL+malonate)=10.56 K(UO2+HL+succinate)=8.78 K(UO2+HL+itaconate)=9.19 K(UO2+HL+glutarate)=7.88	1979SJc (52632)	968
K(UO2+HL+adipate)=8.72										
UO2++	gl	NaClO4	30°C	0.10M	U	M		K1=4.65 B2=8.30 B((UO2)L(Asp))=8.48	1978SJa (52633)	969

C7H5N04		H2L	Dipicolinic aci	CAS	449-83-2	(418)				
2,6-Pyridinedicarboxylic acid; C5H3N.(COOH)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KN03	25°C	0.10M	M	M		K1=5.70	1996AEa (52816)	970
Data for ternary complexes with aspartic acid, serine, asparagine and N-(2-acetamido)iminodiacetic acid										

C7H5N04		H2L	Dinicotinic	CAS	499-81-0	(2857)				
3,5-Pyridinedicarboxylic acid; C5H3N.(COOH)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	NaClO4	30°C	0.10M	U	M		K1=5.44 B2=10.68 B((UO2)L(Asp))=9.21	1978SJa (52846)	971

C7H5N05		H2L	Nitrosalicylic	CAS	85-38-1	(1416)				
2-Hydroxy-3-nitrobenzoic acid; HO.C6H3(NO2).COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	NaClO4	31°C	0.10M	U			K1=8.42 B2=15.04	1973JKa (52978)	972

U02++ gl KN03 28°C 0.10M U K1=8.57 1966RSa (52979) 973

 C7H5N05 H2L Nitrosalicylic CAS 96-97-9 (148)
 2-Hydroxy-5-nitrobenzoic acid; HO.C6H3(NO2).COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl NaClO4 31°C 0.10M U K1=8.65 B2=15.21 1973JKa (53056) 974

 C7H5O3As HL CAS 50722-40-2 (8008)
 2-Arsenosobenzoic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl alc/w 35°C 20% U K1=3.70 1973SPf (53279) 975
 Medium: 20% EtOH/H2O, 0.1 M KNO3.

 C7H5O3Br H2L CAS 3883-95-2 (1111)
 3-Bromosalicylic acid; Br.C6H3(OH).COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl NaClO4 31°C 0.10M U K1=9.72 B2=17.34 1973JKa (53291) 976

 C7H5O3C1 H2L CAS 321-14-2 (1113)
 5-Chlorosalicylic acid; Cl.C6H3(OH).COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl NaClO4 31°C 0.10M U K1=10.81 B2=19.69 1973JKa (53348) 977

 U02++ sp NaClO4 22°C 0.10M U K1=12.11 B2=16.68 1970HSb (53349) 978

 C7H5O3I H2L CAS 16870-28-3 (4435)
 2-Hydroxy-4-iodobenzoic acid (4-iodosalicylic acid); HO.C6H3(I).COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl NaClO4 31°C 0.10M U K1=10.71 B2=19.51 1973JKa (53354) 979

 C7H6N03Br H2L CAS 87353-69-3 (207)
 4-Bromosalicylhydroxamic acid; Br.C6H3(OH).CO.NH.OH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl diox/w 30°C 50% U K1=6.505 1977DJb (53397) 980

 C7H6N03Br H2L CAS 5798-94-7 (206)
 5-Bromosalicylhydroxamic acid; Br.C6H3(OH).CO.NH.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
U02++	gl	diox/w	30°C	50%	U		K1=7.05		1977DJb (53408)	981
<hr/>										
C7H6N03Cl		H2L					(205)			
3-Chlorosalicylhydroxamic acid; Cl.C6H3(OH).CO.NH.OH										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
U02++	gl	diox/w	30°C	50%	U		K1=6.63		1977DJb (53418)	982
<hr/>										
C7H6N03Cl		HL					(6263)			
4-Chlorosalicylhydroxamic acid; Cl.C6H3(OH).CO.NH.OH										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
U02++	gl	diox/w	30°C	50%	U		K1=5.62		1977DJb (53421)	983
<hr/>										
C7H6N03Cl		HL					CAS 37551-43-2	(6262)		
5-Chlorosalicylhydroxamic acid; Cl.C6H3(OH).CO.NH.OH										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
U02++	gl	diox/w	30°C	50%	U		K1=7.11		1977DJb (53424)	984
<hr/>										
C7H6N205		H2L					CAS 831-51-6	(208)		
5-Nitrosalicylhydroxamic acid; O2N.C6H3(OH).CO.NH.OH										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
U02++	gl	diox/w	30°C	50%	U		K1=5.62		1977DJb (53524)	985
<hr/>										
C7H6N2S		HL					CAS 583-39-1	(2043)		
2-Mercaptobenzimidazole;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
U02++	gl	alc/w	25°C	50%	U		K1=8.60	B2=16.40	1978ZIa (53533)	986
<hr/>										
C7H6O2		HL	Salicylaldehyde				CAS 90-02-8	(193)		
2-Hydroxybenzaldehyde, Salicylaldehyde; HO.C6H4.CHO										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
U02++	gl	alc/w	?	50%	U		K1=12.83		1957HSa (53633)	987
<hr/>										
C7H6O2		HL	Tropolone				CAS 533-75-5	(3129)		
2-Hydroxycyclohepta-2,4,6-trien-1-one;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

U02++ gl NaClO4 20°C 1.00M U K1=8.18 B2=15.07 1973MBb (53698) 988

C7H6O2 HL Benzoic Acid CAS 65-85-0 (462)
Benzene carboxylic acid; C6H5.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 31°C 0.10M U M 1971RBc (53860) 989
K(UO2+L+A)=6.08
K(UO2+L+B)=4.36
K(UO2+L+HC)=5.08

HA=phenylacetic acid, HB=phenoxyacetic acid, H2C=4-hydroxybenzoic acid

U02++ gl NaClO4 31°C 0.10M U K1=2.57 1968RSa (53861) 990

U02++ gl NaClO4 31°C 0.10M U K1=2.59 1968RSg (53862) 991

C7H6O2S H2L Thiosalicylic CAS 147-93-3 (236)
2-Mercaptobenzoic acid; HS.C6H4.CO0H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 25°C 40% U M K1=4.60 B2=8.32 1986S1b (53922) 992
Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4

U02++ gl diox/w 30°C 50% U K1=4.60 B2=8.50 1973RSa (53923) 993
Medium: 50% dioxan, 0.1 M NaClO4

C7H6O3 H2L CAS 95-01-2 (4407)
2,4-Dihydroxybenzaldehyde; (OH)2.C6H3.CH0

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 50% U 1969VMa (53942) 994
K(UO2+HL)=6.60
K(UO2HL+HL)=5.20

Medium: 50% dioxan, 0.1 M NaClO4

C7H6O3 H2L CAS 1194-98-5 (4408)
2,5-Dihydroxybenzaldehyde; (OH)2.C6H3.CH0

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 50% U 1969VMa (53949) 995
K(UO2+HL)=8.20
K(UO2HL+HL)=6.85

Medium: 50% dioxan, 0.1 M NaClO4

C7H6O3 H2L Salicylic acid CAS 69-72-7 (14)

2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2 ⁺⁺	gl	alc/w	24°C	20%	C	M	K1=2.95 K(UO2(ada)+L)=3.14		1996MIA (54319)	996
Medium: 20% w/w EtOH/H ₂ O, 0.10 M KNO ₃ .										
ada: N-(acetamido)-iminodiethanoic acid.										
UO2 ⁺⁺	sp	KNO ₃	25°C	0.10M	U				1996SMb (54320)	997
K _{eff} =2.72										
Method: synchronous fluorescence spectroscopy. pH 3.5.										
UO2 ⁺⁺	sp	none	25°C	0	M	T	K1=13.12 K(UO2+HL=UO2L+H)=1.43 K(UO2+H2L=UO2L+2H)=-3.55		1989YAa (54321)	998

UO₂++ gl NaClO₄ 25°C 0.10M U I K1=12.04 1987G Ma (54)
 $B((UO_2)HL) = 14.68$
 $T=0.1: K1=11.97, B((UO_2)HL) = 15.56; T=0.7: K1=12.00, B((UO_2)HL) = 15.41$

H₂A=phthalic acid

C7H6O3 H2L CAS 99-96-7 (1371)

4-Hydroxybenzoic acid; H0.C6H4.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	NaClO ₄	31°C	0.10M	U	M			1971RBc (54436)	1006
								$K(UO_2+HL+A)=5.19$		
								$K(UO_2+HI+B)=4.68$		

HA=phenylacetic acid, HB=phenoxyacetic acid

C7H6O4	H3L	Resorcylic acid	CAS 89-86-1	(876)					
2,4-Dihydroxybenzoic acid, <i>b</i> -Resorcylic acid; C6H3(OH)2.COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference ExptNo

UO2++	gl	NaClO4	25°C	0.20M	U	M	K1=8.10 K(UO2(ida)+L)=7.41 K(UO2(nta)+L)=6.97 K(UO2(edta)+L)=4.85	1986SLb (54545)1007	

UO2++	gl	NaClO4	31°C	0.10M	U				1973JKa (54546)1008
								K(UO2+HL)=14.35 K(UO2HL+HL)=11.79	

UO2++	sp	NaClO4	30°C	0.10M	U	IH			1971SMa (54547)1009
								K(?)=4.40	
Range of ionic strength 0.02-0.20. DH=6.7 kJ mol-1									
								K(?) (I=0.02)=4.49, K(?) (I=0.20)=4.38	pH=4.5

UO2++	gl	diox/w	30°C	50%	U				1971VMa (54548)1010
								K(UO2+HL)=14.73 K(UO2HL+HL)=10.55	
Medium: 50% dioxan, 0.1 M NaClO4									

UO2++	gl	NaClO4	25°C	0.20M	U				19680Ca (54549)1011
								K(UO2+H2L)=2.10 K(UO2+H2L=UO2HL+H)=-0.66	
K(2UO2+2H2L=(UO2)2HL2+3H)=-4.17									

UO2++	gl	KNO3	28°C	0.10M	U		K1=11.98		1966RSa (54550)1012

UO2++	sp	oth/un	25°C	?	U				1965DDb (54551)1013
								K(UO2+H2L=UO2HL+H)=3.70	

C7H6O4	H3L						CAS 409-79-9	(1115)	
2,5-Dihydroxybenzoic acid; C6H3(OH)2.COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference ExptNo

UO2++	gl	NaClO4	31°C	0.10M	U				1973JKa (54589)1014
								K(UO2+HL)=13.16 K(UO2HL+HL)=11.41	

UO2++	gl	diox/w	30°C	50%	U				1971VMa (54590)1015
								K(UO2+HL)=12.75	
Medium: 50% dioxan, 0.1 M NaClO4									

UO2++	gl	NaClO4	25°C	0.20M	U				19680Ca (54591)1016
								K(UO2+H2L)=1.51	

$$K(UO_2 + H_2L \rightleftharpoons UO_2HL + H) = -1.00$$

C7H604 H3L Protocatechuic CAS 99-50-3 (875)
3,4-Dihydroxybenzoic acid; C₆H₃(OH)₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	NaClO ₄	25°C	0.20M	U	M	K1=15.47		1986SLb (54706)	1017
							K(UO ₂ (ida)+L)=15.04			
							K(UO ₂ (nta)+L)=14.50			
							K(UO ₂ (edta)+L)=11.63			

C7H604 H3L CAS 99-10-5 (4409)
3,5-Dihydroxybenzoic acid; C₆H₃(OH)₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	NaClO ₄	25°C	0.20M	U				19680Ca (54715)	1018
							K(UO ₂ +H ₂ L)=2.13			
							K(UO ₂ +H ₂ L=UO ₂ HL+H)=-2.02			

C7H605 H4L CAS 83-30-7 (4410)
2,4,6-Trihydroxybenzoic acid; (OH)₃.C₆H₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	NaClO ₄	?	0.10M	U				1969HKb (54726)	1019
							K(UO ₂ +H ₄ L=UO ₂ H ₂ L+2H)=-3.32			
							K(UO ₂ +H ₃ L=UO ₂ H ₂ L+H)=-0.92			

$$K(UO_2H_2L+H_3L=UO_2(H_2L)_2+H)=-1.08$$

C7H605 H4L Gallic acid CAS 149-91-7 (446)
3,4,5-Trihydroxybenzoic acid; C₆H₂(OH)₃.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	oth/un	25°C	?	U		K1=25.60		1989WZa (54768)	1020
UO ₂ ++	sp	NaClO ₄	?	0.10M	U	M			1969HSa (54769)	1021
							K(UO ₂ +H ₃ L)=2.3			
							K(UO ₂ +2HL)=27.5			

$$K(UO_2(OH)_2UO_2HL+H_3L=(UO_2)_2(OH)_2HLH_2L+H)=-4.1; K(UO_2(CO_3)_3+2H_3L=UO_2(HL)_2+3HC_3O_3+H)=-5.61. \text{ Data for other complexes also given}$$

C7H606S H3L CAS 5965-83-3 (399)
5-Sulfosalicylic acid, 2-Hydroxy-5-sulfobenzoic; H₃SO₃.C₆H₃(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	kin	NaClO ₄	25°C	1.0M	C				1985GBa (55062)	1022

					K(UO ₂ (OH)+HL)=3.21 K(UO ₂ (OH)+OH)=8.58
UO ₂ ++	ix	oth/un	25°C	0.10M	U K1=11.0 B2=19.20 1979CPa (55063)1023 K(UO ₂ L+H)=2.1 K(UO ₂ L+2H)=5.85
UO ₂ ++	gl	NaClO ₄	31°C	0.10M	U K1=11.20 B2=19.61 1973JKa (55064)1024
UO ₂ ++	sp	NaClO ₄	?	0.10M	U K1=11.27 B2=19.41 1968HSa (55065)1025 K(UO ₂ +HL)=2.07
UO ₂ ++	gl	NaClO ₄	30°C	0.20M	U K1=10.85 B2=19.38 1967AMa (55066)1026
UO ₂ ++	gl	KNO ₃	20°C	0.10M	U K1=11.25 B2=18.75 1967BAa (55067)1027
UO ₂ ++	gl	KNO ₃	28°C	0.10M	U K1=10.70 1966RSa (55068)1028
UO ₂ ++	vlt	NaNO ₃	20°C	1.0M	U K1=5.1 1964HAa (55069)1029
Metal ion:	UO ₂ +				
UO ₂ ++	vlt	NaNO ₃	20°C	1.0M	U K1=11.7 B2=17.6 1964HAa (55070)1030
UO ₂ ++	gl	KNO ₃	25°C	0.10M	U I K1=10.62 1964RMc (55071)1031
At I=1.0 M	K1=10.44				
UO ₂ ++	gl	NaClO ₄	25°C	0.10M	U K1=11.14 B2=19.20 1960BSb (55072)1032
UO ₂ ++	sp	oth/un	25°C	.015M	U 1949FAa (55073)1033 K(UO ₂ +HL)=3.89

C7H6O ₉ S ₂		H3L			CAS 56507-30-3 (2659)
3,5-Disulfosalicylic acid;	(HO ₃ S) ₂ .C ₆ H ₂ (OH).COOH				
Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo
UO ₂ ++	gl	NaClO ₄	25°C	0.50M	U T K1=10.77 B2=18.45 1979LPe (55104)1034 B((UO ₂) ₂ L)=13.07 B((UO ₂)(OH)L)=4.21 B((UO ₂)(OH)L ₂)=11.37

C7H ₇ N ₀ 2		HL	Anthranilic		CAS 118-92-3 (1589)
2-Aminobenzoic acid,	Anthranilic acid;	H ₂ N.C ₆ H ₄ .COOH			
Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo
UO ₂ ++	gl	alc/w	24°C	20%	C M K1=2.95 1996MIA (55268)1035 K(UO ₂ (ada)+L)=3.78

Medium: 20% w/w EtOH/H₂O, 0.10 M KNO₃.

ada: N-(acetamido)-iminodiethanoic acid.

U02++ gl diox/w 30°C 50% U K1=5.15 B2=9.05 1973RSa (55269)1036
Medium: 50% dioxan, 0.1 M NaClO4

C7H7N02 HL Salicylamide CAS 65-45-2 (3155)
2-Hydroxybenzamide; HO.C6H4.CO.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp oth/un ? ? U K1=6.40 B2=11.37 1953CSb (55332)1037

C7H7N02 HL CAS 495-18-1 (184)
Benzohydroxamic acid; C6H5.CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaNO3 25°C 0.10M M K1=7.42 B2=14.19 1996KSc (55519)1038

U02++ gl KNO3 25°C 0.10M C K1=7.49 B2=14.17 1989Kub (55520)1039

U02++ gl diox/w 37°C 30% C M K1=6.93 1983MAd (55521)1040
B(UO2(bpy)L)=8.23

U02++ gl NaClO4 30°C 0.10M U K1=9.03 B2=17.94 1969DSb (55522)1041

U02++ sp NaClO4 20°C 1.0M U K1=7.72 1966MRa (55523)1042

U02++ gl NaClO4 25°C 0.10M U K1=8.72 B2=16.77 1965BGa (55524)1043
Medium: HClO4

C7H7N03 H2L CAS 89-73-6 (204)
2-Hydroxybenzohydroxamic acid (salicylhydroxamic acid); HO.C6H4.CO.NHOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaNO3 25°C 0.10M M K1=11.93 B2=17.78 1996KSc (55613)1044

U02++ gl diox/w 30°C 50% U K1=7.31 1977DJb (55614)1045

U02++ gl NaClO4 30°C 0.10M U K1=7.71 B2=14.51 1969DSb (55615)1046

U02++ EMF mixed 30°C 50% U K1=6.70 B2=12.16 1969GMc (55616)1047

Medium: 50% acetone/H2O, 0.5 M NaClO4

C7H7N03 H2L (1112)
4-Aminosalicylic acid; H2N.C6H3(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp NaClO4 22°C 0.10M U K1=13.0 B2=22.6 1970HSb (55642)1048

U02++ gl KNO₃ 28°C 0.10M U K1=14.41 1966RSa (55643)1049

C7H7N06S H2L CAS 35379-88-5 (4464)
3-Nitro-p-cresol-5-sulfonic acid; (CH₃)(HO).C₆H₂(NO₂).SO₃H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ dis NaCl 25°C 1.0M U K1=5.90 1971BEa (55700)1050

C7H8N202 HL CAS 5623-04-1 (1917)
2-Amino-benzohydroxamic acid; H₂N.C₆H₃.CO.NH.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl KNO₃ 25°C 0.10M C K1=7.82 B2=14.71 1989Kub (55855)1051

C7H8O2 H2L Methylcatechol CAS 452-86-8 (525)
1,2-Dihydroxy-4-methylbenzene; CH₃.C₆H₃(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl KNO₃ 25°C 0.20M U M K1=13.94 B2=23.39 1990SSc (56081)1052
K(UO₂(IMDA)+L)=12.54
K(UO₂(NTA)+L)=11.96
K(UO₂(HEDTA)+L)=11.49
K(UO₂(EDTA)+L)=10.94
K(UO₂(CDTA)+L)=10.89, K(UO₂(DTPA)+L)=9.60

C7H10N203S HL CAS 71691-06-0 (1247)
2-(N-Pyrrolideneimino)ethane sulfonic acid; C₄H₄N.CH:N.CH₂.CH₂.SO₃H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl NaClO₄ 25°C 0.10M U T K1=10.50 B2=18.35 1979GSa (56693)1053

C7H1003 H2L (793)
Heptane-2,4,6-trione; CH₃.CO.CH₂.CO.CH₂.CO.CH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ sp alc/w 25°C 70 % U B((UO₂)HL)=8.48 1991HKe (56718)1054
Medium: 70% v/v MeOH/H₂O, 0.5 M NaClO₄

C7H11N302 L CAS 7389-87-9 (3162)
Histidine methyl ester

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl oth/un 25°C 0.20M U K1=5.76 1957LDa (57006)1055

 C7H12N202 HL (6181)
 2-(N-2-Pyrrolidimino)propanoic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl NaClO4 25°C 0.10M U TIH B2=22.72 1988GRb (57074)1056
 35 C:B2=22.82, 45 C:22.92. DH(B2)=18.1 kJ mol-1, DS=495.8 J K-1 mol-1

 C7H12O2 HL CAS 7424-54-6 (4421)
 Heptane-3,5-dione; CH3.CH2.CO.CH2.CO.CH2.CH3

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ sp NaClO4 25°C 0.5M C K1=7.70 1998BLa (57247)1057

 C7H12O4 HL CAS 96740-23-7 (2249)
 1,5-Dimethoxy-pent-2,4-dione, CH3.O.CH2.CO.CH2.CO.CH2.O.CH3

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl diox/w 24°C 50% U K1=8.7 1979ACa (57295)1058

 C7H12O4 H2L Pimelic acid CAS 111-16-0 (985)
 1,7-Heptanedioic acid; HOOC.(CH2)5.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl KNO3 25°C 0.50M U K1=3.68 1969V0b (57312)1059
 K(U02+HL)=2.45

 C7H12O4 H2L CAS 510-20-3 (482)
 Diethylpropanedioic acid (Diethylmalonic acid); HOOC.C(C2H5)2.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl KNO3 25°C 0.50M U K1=6.36 B2=11.04 1969V0b (57374)1060

 C7H13N04S HL (6310)
 Acetylacetone-2-aminoethane sulfonic acid schiff base;
 CH3.CO.CH2.C(CH3):N.CH2.CH2.HS03

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 U02++ gl diox/w 25°C 50% U T H K1=15.40 19760Ma (57536)1061

 C7H21N2010P3 H6L (7004)
 N-(2-Hydroxyethyl)-1,2-diaminoethane-N,N'N'-trimethylenephosphonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	g1	KCl	25°C	0.10M	U			K1=10.04 K(UO ₂ +HL)=7.65	1974KRd	(58374)1062

C8H4O4Cl2 H2L CAS 16110-99-9 (1173)
3,6-Dichloro-phthalic acid; C12.C6H2(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
UO ₂ ++	gl	NaClO ₄	30°C	0.10M	U			K1=3.99		1976PJa (58447)	1067
UO ₂ ++	ix	KNO ₃	25°C	1.0M	U			K1=3.6		1973NKb (58448)	1068
UO ₂ ++	gl	NaClO ₄	31°C	0.10M	U			K1=4.02		1967SPe (58449)	1069

C8H5O2F3S HL TTA CAS 326-91-0 (165)
4,4,4-Trifluoro-1-(2-thienyl)butane-1,3-dione; F3C.CO.CH2.CO.C4H3S

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	30°C	75%	U			K1=7.48 B2=14.45	1977AHb	(58692)1070

UO₂++ g1 diox/w 30°C 75% U K1=8.7 B2=16.62 1965RGa (58693)1071

C8H5O4Br H2L CAS 116-69-8 (1169)
3-Bromo-phthalic acid; Br.C6H3(COOH)2

Metal	Melt Medium	Temp	Conc	Cat	Flags	Lg R	values	Reference	ExptNo
UO ₂ ++	gl NaClO ₄	30 °C	0.10M	U		K1=4.30		1976PJa (58722)	1072

C8H5O4Cl H2L CAS 27563-65-1 (1168)
3-Chloro-phthalic acid; Cl.C6H3(COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 30°C 0.10M U K1=4.20 1976PJa (58729)1073

C8H5O4I H2L CAS 6737-34-3 (1170)
3-Iodo-phthalic acid; I.C6H3(COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 30°C 0.10M U K1=4.07 1976PJa (58730)1074

C8H6N2O HL CAS 17056-99-4 (3220)
5-Hydroxyquinoxaline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=8.40 B2=15.91 1954IRa (58748)1075
Medium: 50% dioxan, I=0.3 M NaClO4

C8H6N2O HL (6290)
8-Hydroxycinnoline, (2-Hydroxybenzo)pyrimidine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=8.68 B2=15.84 1954IRa (58769)1076
Medium: 50% dioxan, 0.3 M NaClO4

C8H6N2O HL 8-Quinazolinol CAS 7757-02-2 (3221)
8-Hydroxyquinazoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=8.99 B2=16.69 1954IRa (58779)1077
Medium: 50% dioxan, 0.3 M NaClO4

C8H6O4 H2L Phthalic acid CAS 88-99-3 (113)
Benzene-1,2-dicarboxylic acid; C6H4(COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 24°C 20% C M K1=4.96 1996MIa (59021)1078
K(UO2(ada)+L)=5.19

Medium: 20% w/w EtOH/H2O, 0.10 M KNO3.
ada: N-(acetamido)-iminodiethanoic acid.

U02++ gl NaClO4 25°C 0.50M C K1=4.97 1989NMa (59022)1079

U02++ gl NaClO₄ 25°C 0.10M U I K1=4.742 B2=7.73 1987GMa (59023)1080
I=0.4: K1=4.46, B2=7.38; I=0.7: K1=4.43, B2=6.97

U02++ gl KN03 25°C 0.20M U T 1985KMc (59024)1081
K(U02A+L)=3.90

H2A=iminodiacetic acid; 5 C:K=4.08; 45 C: K=3.68, DH=-16.3 kJ mol-1,
DS=21 J K-1 mol-1

U02++ gl NaClO₄ 30°C 0.10M U K1=4.88 1976PJa (59025)1082

U02++ gl NaClO₄ 30°C 0.10M U M K1=4.78 1973KJa (59026)1083
K(U02+L+A)=7.32
K(U02+L+B)=7.63

H2A=adipic acid, H2B=succinic acid

U02++ ix KN03 25°C 1.0M U K1=4.1 1973NKb (59027)1084

U02++ gl KN03 25°C 1.00M U K1=4.38 1967RMc (59028)1085

U02++ gl NaClO₄ 31°C 0.10M U K1=4.81 1967SPe (59029)1086

U02++ gl KN03 25°C 1.0M U K1=4.38 1964RAa (59030)1087

C8H7NO2C12 HL CAS 13538-26-6 (6286)

3,5-Dichloro-2-hydroxyacetophenone oxime; C12(HO)C₆H₂.C(CH₃):NOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 27°C 75% U I K1=8.48 B2=15.98 1976LGa (59119)1088

Data in 75% EtOH. Data also in 75% acetone and 75% dioxan

C8H8NO2C1 HL CAS 2153-11-9 (4570)

N-Chloroacetyl-N-phenylhydroxylamine; Cl.CH₂.CO.N(C₆H₅).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ EMF mixed 30°C 50% U K1=7.20 B2=13.00 1970GSf (59285)1089

Medium: 50% acetone/H₂O, 0.5 M NaClO₄

C8H8O2 HL Phenylacetic CAS 103-82-2 (1361)

Phenylethanoic acid; C₆H₅.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ EMF NaClO₄ 25°C 1.00M C H K1=2.22 1992BCb (59568)1090

DH(K1)=10.80 kJ mol-1, DS=79 J K-1 mol-1

U02++ gl NaClO₄ 31°C 0.10M U M 1971RBC (59569)1091

K(U02+L+A)=3.72

HA=phenoxyacetic acid

U02++	gl	NaClO4	31°C	0.10M	U	K1=3.21	1968RSa (59570)1092		
U02++	gl	NaClO4	31°C	0.10M	U	K1=3.25	1968RSF (59571)1093		

C8H8O2	HL		CAS 583-80-2 (3191)						
beta-Methyltropolone;									
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo		
U02++	sp	alc/w	?	50%	U	K1=9.62 B2=16.60	1965DSb (59605)1094		
Medium: 50% EtOH, 0.5 M KNO3. By glass electrode: K2=6.93									

C8H8O2S	HL		3-Thenoylacetone CAS 21808-13-9 (2736)						
3-Thenoylacetone, 1-(3'-Thienyl)butane-1,3-dione; C4H3S.CO.CH2.CO.CH3									
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo		
U02++	gl	diox/w	30°C	75%	U	K1=12.3 B2=22.62	1965RGa (59647)1095		

C8H8O3	H2L		o-Cresotic acid CAS 83-40-9 (2338)						
2-Hydroxy-3-methylbenzoic acid; CH3.C6H3(OH).COOH									
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo		
U02++	gl	diox/w	30°C	50%	U	K1=12.18	1971VMa (59703)1096		
Medium: 50% dioxan, 0.1 M NaClO4									

C8H8O3	HL		Mandelic Acid CAS 611-72-3 (80)						
2-Phenyl-2-hydroxyethanoic acid; C6H5.CH(OH).COOH									
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo		
U02++	gl	KNO3	25°C	0.20M	U T	1985KMc (59883)1097			
K(UO2A+L)=3.54									
H2A=iminodiacetic acid; 5 C:K=3.66; 45 C: K=3.46, DH=-10.0 kJ mol-1,									
DS=33 J K-1 mol-1									
U02++	gl	NaClO4	20°C	1.00M	C	T K1=2.57 B2=4.10 B3=5.32	1974MTa (59884)1098		

U02++	gl	KNO3	25°C	0.10M	U	K1=2.47 K(UO2(OH)L+H)=3.9 K((UO2(OH)L)2+2H)=4.94	1967VAa (59885)1099		

C8H8O3	HL		m-Anisic acid CAS 586-38-9 (2804)						
3-Methoxybenzoic acid; CH3O.C6H4.COOH									
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo		

U02++ dis NaClO₄ 25°C 0.10M U 1956H0a (59922)1100
 $B((U02)LOH)=11.9$
 $K(U02+HL+2H)=-5.85$

C8H8O₃ HL CAS 121-33-5 (4476)
4-Hydroxy-3-methoxybenzaldehyde; CH₃O.C₆H₃(OH).CHO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	con	oth/un	?	?	U			B2=6.04	1972LLa (59934)	1101
U02++	sp	oth/un	?	?	U			B2=5.51	1972LLa (59935)	1102

C8H8O₃ H₂L m-Cresotic acid CAS 50-85-1 (1244)
4-Methylsalicylic acid; CH₃.C₆H₃(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	30°C	50%	U			K1=13.46	1971VMa (60000)	1103

Medium: 50% dioxan, 0.1 M NaClO₄

C8H8O₃ HL Phenoxyacetic CAS 122-59-8 (1153)
Phenoxyethanoic acid; C₆H₅.O.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	NaClO ₄	31°C	0.10M	C	M			1975BSa (60042)	1104

$B((U02)LA)=7.64$
 $K(U02L2+U02A2=2U02LA)=-0.03$
 $K(U02L+A)=5.20$
 $K(U02A+L)=2.45$

H₂A=maleic acid

U02++	gl	NaClO ₄	31°C	0.10M	C	M			1975BSa (60043)	1105
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$B((U02)LA)=5.08$
 $K(U02L2+U02A2=2U02LA)=-0.42$
 $K(U02L+A)=2.64$
 $K(U02A+L)=2.03$

H₂A=fumaric acid

U02++	gl	NaClO ₄	31°C	0.10M	C	M			1975BSa (60044)	1106
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$B((U02)LA)=5.79$
 $K(U02L2+U02A2=2U02LA)=-0.73$
 $K(U02L+A)=3.35$
 $K(U02A+L)=1.71$

H₂A=adipic acid

U02++	gl	NaClO ₄	31°C	0.10M	C	M			1975BSa (60045)	1107
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$B((U02)LA)=6.84$
 $K(U02L2+U02A2=2U02LA)=0.69$

$$K(UO2L+A)=4.40$$

$$K(UO2A+L)=3.13$$

H3A=thiomalic acid

U02++	gl	NaClO4	31°C	0.10M	C	M	1975BSa (60046)1108
							$B((UO2)LA)=5.12$
							$K(UO2L+UO2A2=UO2LA+UO2A)=2.09$
							$K(UO2L+A)=2.68$
							$K(UO2A+L)=2.96$

HL=pyruvic acid

U02++	gl	NaClO4	31°C	0.10M	U	K2=2.59	1968RSa (60047)1109
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U02++	gl	NaClO4	31°C	0.10M	U	K1=2.41	1968RSf (60048)1110
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C8H8O4 HL CAS 520-45-6 (4478)

3-Acetyl-2-hydroxy-6-methylpyran-4-one, Dehydroethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	35°C	50%	U			K1=6.52 B2=11.62	1971MAa (60100)1111	

Medium: 50% dioxan, 0.1 M NaClO4

C8H8O9 H4L (6951)

Tetrahydrofuran-2,3,4,5-tetracarboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++	gl	NaClO4	25°C	0.10M	C	H			2000MNa (60139)1112	
								$B(UO2HL)=11.18$		
								$B(UO2H2L)=14.27$		

$$B(UO2HL2)=16.04$$

By calorimetry: DH(UO2+HL)=22.5 kJ mol-1, DS=166. DH(UO2+H2L)=18.2, DS=121

C8H9NOS HL CAS 4822-44-0 (3240)

N-(Mercaptoacetyl)aniline (thioglycolanilide); C6H5.NH.CO.CH2.SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++	gl	diox/w	30°C	75%	U			K1=10.14 B2=19.10	1961MAe (60164)1113	
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C8H9N02 HL CAS 17194-82-0 (1382)

2-Hydroxyacetophenone oxime; HO.C6H4.C(CH3):NOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++	gl	diox/w	30°C	50%	U			K1=9.33 B2=17.26	1982UVa (60218)1114	
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C8H9N02 HL (2591)

N-Phenyl-N-acetohydroxamic acid; CH3.CO.N(OH)C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

U02++	EMF	mixed	30°C	50%	U		K1=8.56	B2=15.36	1971GSc	(60286)1115
Medium: 50% acetone/H ₂ O, 0.5 M NaClO ₄										

C8H9N02		HL					CAS	5330-97-2	(6248)	
Phenylacetohydroxamic acid; C ₆ H ₅ .CH ₂ .CO.NH.OH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

U02++	sp	NaClO ₄	30°C	0.10M	U		K1=8.44	B2=15.78	1980RSb	(60357)1116

C8H9N02S		HL					CAS	104-18-7	(4575)	
(4-Aminophenylthio)ethanoic acid; H ₂ N.C ₆ H ₄ .S.CH ₂ .COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

U02++	gl	KNO ₃	25°C	0.05M	M		K1=3.90		1975DPb	(60377)1117

C8H9N02S		HL					CAS	6310-11-8	(4576)	
3-Mercaptoacetamidophenol; HS.CH ₂ .CO.NH.C ₆ H ₄ .OH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

U02++	gl	oth/un	17°C	?	U		K1=6.57	B2=12.28	1973KPD	(60386)1118

C8H9N03		HL	Pyridoxal				CAS	65-22-5	(110)	
3-Hydroxy-5-(hydroxymethyl)-2-methyl-4-pyridinecarboxaldehyde;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

U02++	gl	NaClO ₄	25°C	0.10M	C		K1=7.14	B2=13.34	1978MCa	(60429)1119

C8H9N03		H2L					CAS	26071-07-8	(209)	
5-Methylsalicylhydroxamic acid; CH ₃ .C ₆ H ₃ (OH).CO.NH.OH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

U02++	gl	diox/w	30°C	50%	U		K1=7.22		1977DJb	(60439)1120

C8H9N03		HL					CAS	24618-17-5	(4526)	
N-Methylsalicylohydroxamic acid; HO.C ₆ H ₄ .CO.N(CH ₃)OH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

U02++	EMF	mixed	30°C	50%	U		K1=5.92	B2=10.40	1969GMc	(60452)1121
Medium: 50% acetone, 0.5 M NaClO ₄										

C8H9N04		H2L					(4520)			

Dehydroethanoic acid oxime;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	35°C	50%	U				1971MAa (60506)1122	

$$K(UO_2+HL)=6.62$$
$$K(UO_2+2HL)=11.63$$

Medium: 50% dioxan, 0.1 M NaClO4

C8H9N3OS H2L CAS 5351-90-6 (2103)

Salicylideneethiosemicarbazone; HO.C6H4.CH:N.NH.CS.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	NaClO4	25°C	0.05M	U			K1=18.46	1987CDa (60559)1123	

C8H9N307 H2L Uramildiacetic CAS 13055-06-5 (185)
5-Amino-2,4,6-trioxo-1,3-perhydrodiazimino-N,N-diethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	EMF	oth/un	25°C	0.10M	U			K1=9.52	1967DSa (60661)1124	

C8H11N03 HL Vitamin B6 CAS 65-23-6 (254)
5-Hydroxy-6-methyl-3,4-pyridinedimethanol, Pyridoxine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	KCl	30°C	0.50M	U			K1=11.49 B2=19.96 K3=3.76	19710Sb (61125)1125	

C8H12N203 H2L Barbital CAS 57-44-3 (2744)
5,5-Diethylbarbituric acid, Veronal, Barbitone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	alc/w	20°C	50%	C	TIH		K2=5.48 K3=3.65	1987EAa (61445)1126	

$$\Delta H(K2)=-29.05 \text{ kJ mol}^{-1}$$

C8H1205 H2L CAS 103435-40-1 (4481)
1-Hydroxy-1,2-cyclohexanedicarboxylic acid; HO.C6H9(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	oth	oth/un	?	?	U				1972MKc (61733)1127	

$$K(UO_2+H_2L=UO_2(H-1)L+3H)=-8.63$$

C8H1404 H2L Suberic acid CAS 505-48-6 (517)
Octanedioic acid; HOOC.(CH2)6.COOH

C9H5N02Br2 HL CAS 16846-41-1 (4666)
5,7-Dibromo-8-hydroxyquinoline N-oxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 35°C 75% U K1=11.59 B2=20.69 1970GMh (63584)1141
Medium: 75% v/v dioxan, 0.2 M NaClO4

C9H5N02Cl2 HL CAS 21168-33-2 (4665)
5,7-Dichloro-8-hydroxyquinoline N-oxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 35°C 75% U K1=11.46 B2=20.46 1970GMh (63594)1142
Medium: 75% dioxan, 0.1 M NaClO4

C9H5N04 HL CAS 22308-86-7 (4607)
3-Nitroso-4-hydroxycoumarin (oximidobenzotetronic acid);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 21°C 50% U K1=3.48 B2=6.38 1970MGd (63616)1143
Medium: 50% dioxan, 0.3 M NaClO4

C9H5N305 HL CAS 1084-32-8 (4608)
5,7-Dinitro-8-hydroxyquinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 35°C 75% U K1=6.43 B2=11.56 1970GMh (63629)1144
Medium: 75% dioxan, 0.2 M NaClO4

C9H5N306 HL CAS 21168-36-3 (4609)
5,7-Dinitro-8-hydroxyquinoline-N-oxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 35°C 75% U K1=6.03 B2=11.00 1970GMh (63637)1145
Medium: 75% v/v dioxan, 0.2 M NaClO4

C9H6N04BrS H2L CAS 3062-37-1 (3889)
7-Bromo-8-hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp oth/un 30°C 0.10M U K1=9.04 1970ABd (63706)1146

C9H6N04ClS H2L CAS 3244-71-1 (4687)
5-Chloro-8-hydroxyquinoline-7-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	oth/un	25°C	0.02M	U	T	H		1970BBb (63712)1147	
K(UO2+2HL=UO2L2+2H)=8.40										
30 C:	K=8.47,	35 C:	K=8.50,	40 C:	K=8.52,	45 C:	K=8.90			
DH=-17.14 kJ mol-1, DS=218.6 J K-1 mol-1										

C9H6N04IS		H2L						CAS 3075-21-6 (4689)		
5-Iodo-8-hydroxyquinoline-7-sulfonic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	oth	oth/un	25°C	0.0	U			B2=8.85	1972BBb (63718)1148	

C9H6N04IS		H2L	Ferron					CAS 547-91-1 (275)		
7-Iodo-8-hydroxyquinoline-5-sulfonic acid; (HO)(HO3S)C9H4NI										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	NaClO4	45°C	1.0M	C			K1=1.16	1994DAb (63833)1149	
Data for 30-30 C.	DH(K1)=36.6	kJ mol-1,	DS(K1)=106.3	J K-1	mol-1.					
By kinetics at 45 C,	K1=1.15.									
UO2++	gl	KN03	25°C	0.10M	C			K1=9.30 B2=14.10	1985ZHa (63834)1150	
UO2++	sp	oth/un	25°C	0.10M	U			B2=13.32	1972HKc (63835)1151	
pH=5.08										

C9H6O3		HL						CAS 939-19-5 (8274)		
3-Hydroxycoumarin;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	KN03	RT	0.10M	C			K1=6.48 B2=10.62	1983SSe (63937)1152	

C9H7NO		HL	Oxine					CAS 148-24-3 (504)		
8-Hydroxyquinoline (8-quinolinol);										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	vlt	diox/w	20°C	30%	U	T	H	K1=2.02 B2=2.30 B3=4.00 K(UO2L+succinate)=3.72	1985SKb (64367)1153	
30 C,	K1=1.60,	B2=2.89.	40 C,	K1=1.30,	B2=2.86.	DH(K1)=-63.1	kJ mol-1			
DH(K2)=48.9.	Medium:	30% EtOH/H2O,	0.6 M	KCl						
UO2++	gl	KN03	25°C	0.10M	U	T	M	K1=9.66 B(UO2AL)=17.33 K(UO2A+L)=12.20	1985VSb (64368)1154	

$$K(UO2L+A)=7.67$$

H2A=phthalic acid

U02++ gl diox/w 25°C 50% U K1=11.42 B2=21.09 1971CAd (64369)1155
Medium: 50% dioxan, 0.1 M NaClO4

U02++ sol oth/un 25°C ? U 1958KKa (64370)1156
 $K_s(UO2HL3)=-28.72$

U02++ gl diox/w 20°C 50% U K1=11.25 B2=20.89 1954IRa (64371)1157

C9H7N02 HL CAS 1127-45-3 (4614)
8-Hydroxyquinoline-N-oxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	25°C	50%	U			K1=10.45 B2=18.00	1970GMb (64413)1158	
Medium:	50% dioxan,	0.3 M	NaClO4					*****		

C9H7N04S H2L Sulfoxine CAS 84-88-8 (448)
8-Hydroxyquinoline-5-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	KN03	25°C	0.10M	U			K1=8.52 B2=15.68	1959RGa (64584)1159	
								$K(UO2L2OH+H)=6.68$		
								$K((UO2L2OH)2+2H=2UO2L2)=11.7$		

C9H7N04S H2L CAS 3062-35-9 (4676)
8-Hydroxyquinoline-7-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	sp	oth/un	25°C	0.02M	U	T	H		1970BBb (64596)1160	
								$K(UO2+2HL=UO2L2+2H)=8.90$		

30 C, K=8.96, 35 C, K=9.04, 40 C, K=9.06, 45 C, K=9.14

DH=-22.15 kJ mol-1, DS=246.6 J K-1 mol-1

C9H7N302S H2L TAR CAS 2246-46-0 (707)
4-(2'-Thiazolylazo)-resorcinol; C3H2NS.N:N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	alc/w	25°C	50%	U				1967NPb (64734)1161	
								$K(UO2+HL)=10.7$		
								$K(UO2(HL)+HL)=9.7$		

Medium: 50% MeOH, 0.1 M NaClO4

U02++	sp	NaClO4	20°C	0.10M	U			K1=11.35	1967SIC (64735)1162	
								$K(UO2L+H)=4.5$		

$$K(UO_2 + HL) = 9.8$$

C9H8N20 HL CAS 17056-96-1 (3258)
8-Hydroxy-4-methylcinnoline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	20°C	50%	U			K1=9.00 B2=16.30	1954IRa	(64791)1163

Medium: 50% dioxan, 0.3 M NaClO₄

C9H8N204S2 HL CAS 219931-32-5 (8394)
3-Phenylsulfonamidorhodanine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	alc/w	30°C	20%	C T H			K1=8.43 B2=15.82	1998EGA	(64834)1164

Medium: 20% v/v EtOH/H₂O, 0.10 M KCl. Also data for 35 and 45 C.

DH and DS values reported

C9H8N50Cl L (2723)
4-(4'-Chlorophenylazo)-3-amino-pyrazolin-5-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	alc/w	20°C	100%	U	H		K1=6.65 B2=10.36	1983EAa	(64863)1165

C9H8N603 L CAS 76043-30-6 (2724)
4-(4'-Nitrophenylazo)-3-amino-pyrazolin-5-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	alc/w	20°C	100%	U	H		K1=7.0 B2=11.0	1983EAa	(64864)1166

C9H8O2 HL CAS 140-10-3 (3245)
trans-Cinnamic acid; C₆H₅.CH:CH.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	NaClO ₄	25°C	0.10M	U			K1=3.08 B2=5.86	1983GAa	(64873)1167

C9H8O4 H₂L CAS 97652-17-0 (3855)
3-Carboxy-4-methyltropolone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	NaClO ₄	?	0.20M	U			K1=9.60	1967GDb	(64956)1168

By glass electrode: K₁=9.72, K₂=6.78

UO ₂ ++	sp	KNO ₃	?	0.50M	U			K1=9.22 B2=15.97	1965DSb	(64957)1169
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By glass electrode: K₂=6.80

C9H9N3O2S2 HL Sulfathiazole CAS 72-14-0 (8357)
4-Amino-N-2-thiazolyl-benzenesulfonamide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 25°C 50% C K1=6.40 B2=12.24 1999GAa (65136)1170
Medium: 50% EtOH/H2O, 0.10 M NaNO3.

C9H9N50 L CAS 13197-14-9 (2720)
4-Phenylazo-3-amino-pyrazolin-5-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp alc/w 20°C 100% U H K1=5.86 B2=8.41 1983EAa (65155)1171

C9H10N2O2 HL CAS 52829-64-8 (4627)
2-Acetoacetamidopyridine; C5H4N.NH.CO.CH2.CO.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KNO3 25°C 0.10M U K1=7.26 B2=13.89 1967HAb (65230)1172

C9H10O2 HL CAS 1450-72-2 (4596)
2-Hydroxy-5-methylacetophenone; HO(CH3).C6H3.CO.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp oth/un 30°C ? U 1970GMe (65337)1173
K(Uo2+2HL)=7.15

C9H10O2 HL Benzylacetic CAS 501-52-0 (1362)
3-Phenylpropanoic acid; C6H5.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ oth NaClO4 31°C 0.10M U M 1972SSb (65377)1174
K(UO2+benzoate+L)=6.05
K(UO2+phenylacetate+L)=6.16
K(UO2+hydroxybenzoate+L)=5.42

C9H10O8 H4L CAS 3724-52-5 (1264)
cis-1,2,3,4-Cyclopentanetetracarboxylic acid; C5H6.(COOH)4

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 30°C 0.19M U K1=6.48 B2=11.63 1985MSb (65652)1175

C9H11NOS HL CAS 34282-30-9 (3287)
N-(Mercaptoacetyl)-4-methylanilide; CH3.C6H4.NH.CO.CH2.SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
UO2++	sp	non-aq	?	100%	U			K1=2.84 B2=8.32	1991SKc (66488)	1186
Medium: EtOH										
UO2++	sp	alc/w	?	100%	U			K1=2.84 B2=8.32	1991SKd (66489)	1187
Medium: EtOH										

C9H13N3O5		L	Cytidine				CAS	65-46-3 (2152)		
Cytidine, Cytosine-1-beta-D-ribofuranoside;										
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KNO3	35°C	0.10M	U			K1=3.5	1982RKa (67084)	1188

C9H15N02		HL					CAS	15871-65-5 (4655)		
N-Acetoacetyl piperidine; C5H10N-CO.CH2.CO.CH3										
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	20°C	50%	U			K1=12.98 B2=24.26	1969KSe (67381)	1189
Medium: 50% dioxan, 0.025 M NaClO4										

C9H16O4		H2L	Azelaic acid				CAS	123-99-9 (3255)		
Nonanedioic acid; HOOC.(CH2)7.COOH										
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KNO3	25°C	0.10M	U	TI	M	K1=5.88 K(UO2(nta)+L)=4.02	1987AKb (67800)	1190
Data for 5, 25, 45 C, I=0.05-0.2 M KNO3. Also data for 10-40% MeOH/H2O and EtOH/H2O, 0.20 M KNO3, 25 C.										

C9H18N204		H2L					CAS	18992-11-5 (5913)		
N,N-Dihydroxy nonanediamide; HN(OH).CO.(CH2)7.CO.NH(OH)										
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	NaNO3	25°C	0.10M	C			K1=13.07 B((UO2)HL)=17.48	1989EHa (67942)	1191

C9H28N3O15P5		10L	DTPPH				CAS	15827-60-8 (2921)		
Diethylenetriamine-N,N,N',N'',N''-penta(methylphosphonic acid); H2O3PCH2.N(CH2CH2.N(CH2PO3H2)2)2 H										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KNO3	20°C	0.10M	U			K1=16.87 K(UO2+HL)=13.85 K(UO2+H2L)=11.59	1979ZKb (68415)	1192

C10H7N08S2 H3L Nitroso-R acid CAS 525-05-3 (1811)
 1-Nitroso-2-hydroxynaphthalene-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	KNO ₃	25°C	0.20M	U	T	HM		1986KMc	(69033)1200
K(UO ₂ (ida)+L)=4.86 K(UO ₂ (edda)+L)=5.68 K(UO ₂ (nta)+L)=4.99										
Data for 5, 45 C. DH(UO ₂ (ida)L)=-15 kJ mol ⁻¹ , DS(UO ₂ (ida)L)=42 J K ⁻¹ mol ⁻¹ DH(UO ₂ (edda)L)=-15, DS(UO ₂ (edda)L)=56; DH(UO ₂ (nta)L)=-7.5, DS(UO ₂ (nta)L)=71										
U02++	gl	KCl	25°C	0.10M	U			K1=5.87	1974LSa	(69034)1201
U02++	oth	oth/un	30°C	0.0	U			K1=6.90	B2=12.10	1973GBa (69035)1202
U02++	gl	NaClO ₄	25°C	0.10M	U			K1=5.44	B2=9.85	1966BDa (69036)1203
								K3=2.68		

C10H8N2 L 2,2'-Bipyridyl CAS 366-18-7 (25)
 2,2'-Bipyridine; (C₅H₄N)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	KNO ₃	25°C	0.10M	U	M		K1=3.77	B2=6.92	1985VSb (69659)1204
B(UO ₂ A ₁ L)=10.44 K(UO ₂ A ₁ +L)=5.31 K(UO ₂ L+A ₁)=6.67										

H2A=phthalic acid

U02++	gl	diox/w	37°C	30%	C	M	K1=3.93	1983MAd	(69660)1205
B(UO ₂ (bha)L)=8.23									

bha: benzohydroxamic acid

U02++	gl	NaClO ₄	30°C	0.10M	M	M	K1=3.58	1982RSb	(69661)1206
K(UO ₂ L+OH)=9.60 K(UO ₂ (OH)L+A)=10.58 K(UO ₂ +OH+A+L)=24.03 K(UO ₂ (OH)L+B)=3.96									

B(UO₂(OH)BL)=17.14. H2A=N,N'-1,2-ethanediylbis(2-mercaptoproactamide),
 H2B=N,N'-1,2-ethanediylbis(3-mercaptopropanamide).

C10H8N202S2 L (7069)

3-Benzamidophodanine; C₆H₅.CO.NH.C₃H₂NS₂:0

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	alc/w	25°C	20%	U	T	H	K1=10.43	B2=16.83	1994BSd (69696)1207

K3=4.78

Medium: 20% v/v EtOH/H₂O, 0.1 M KCl. Also at 35 C, 45 C.

DH(K1)=-27 kJ mol-1, DH(K2)=-15, DH(K3)=-13

C10H802 H2L CAS 92-44-4 (1658)

2,3-Dihydroxynaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KNO3	25°C	0.20M	U	M	K1=13.75 K(UO2(IMDA)+L)=12.72 K(UO2(NTA)+L)=12.05 K(UO2(HEDTA)+L)=11.60 K(UO2(EDTA)+L)=11.03	1990SSc (69782)	1208	

K(UO2(CDTA)+L)=10.95, K(UO2(DTPA)+L)=9.83

UO2++	gl	KNO3	20°C	0.10M	U		K1=15.0 K(UO2L+H)=3.9 K(UO2HL+H)=6.5	1967BAd (69783)	1209
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C10H805S H3L DHNSA (877)

2,3-Dihydroxynaphthalene-6-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	NaClO4	20°C	0.10M	U	I	K1=15.6 K(UO2+HL)=6.2 K(UO2L+HL)=4.2	1965BSd (69867)	1210	

By glass electrode, 0.1 M KNO3: K1=15.5, K2=10.65, K(UO2+HL)=5.6,
K(UO2L+HL)=4.2

C10H807S2 H3L CAS 1330-52-5 (3904)

2-Hydroxynaphthalene-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	NaClO4	25°C	0.10M	U		K1=7.42 B2=13.12	1968Bdc (69879)	1211	

C10H808S2 H4L Chromotropic ac CAS 148-25-4 (1875)

1,8-Dihydroxynaphthalene-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KNO3	25°C	0.20M	U	M	K1=16.39 K(UO2(IMDA)+L)=14.71 K(UO2(NTA)+L)=14.64 K(UO2(HEDTA)+L)=14.20 K(UO2(EDTA)+L)=14.15	1990SSc (69978)	1212	

K(UO2(CDTA)+L)=13.90, K(UO2(DTPA)+L)=13.27

UO2++	gl	NaClO4	25°C	0.10M	U		K1=13.58 B2=22.12	1968BDe (69979)	1213
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U02++ gl NaClO₄ 30°C 0.20M U K1=16.60 B2=28.00 1967AMa (69980)1214

U02++ gl KN₃O 20°C 0.10M U K1=16.1 1965BSd (69981)1215
K(UO₂+HL)=3.9

U02++ sp NaClO₄ 20°C 0.10M U K1=16.6 B2=28.10 1965BSd (69982)1216
K(UO₂+HL)=4.0
K(UO₂L+HL)=1.5

U02++ gl oth/un 25°C 0.11M U 1957JAc (69983)1217
K(UO₂+H₂L=UO₂L+2H)=-4.62

C10H9NO HL 8-OH-Quinaldine CAS 826-81-3 (998)
2-Methyl-8-hydroxyquinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U I K1=9.4 B2=17.4 1954IRa (70056)1218
Medium: 50% dioxan, 0.3 M NaClO₄

C10H9NO HL CAS 5541-67-3 (999)
5-Methyl-8-hydroxyquinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=11.25 B2=20.77 1954IRa (70068)1219
Medium: 50% dioxan, 0.3 M NaClO₄

C10H9NO HL CAS 5541-68-4 (1000)
7-Methyl-8-hydroxyquinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=11.28 B2=21.06 1954IRa (70076)1220
Medium: 50% dioxan, 0.3 M NaClO₄

C10H9NO HL CAS 20984-33-2 (3321)
8-Hydroxy-6-methylquinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=10.89 B2=20.15 1954IRa (70101)1221
Medium: 50% dioxan, 0.3 M NaClO₄

C10H9N02 HL CAS 83010-87-2 (4717)
8-Hydroxy-2-methoxyquinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 50% U K1=10.28 1971CAd (70122)1222

Medium: 50% dioxan, 0.1 M NaClO4

C10H9N02C12

HL

(3333)

N-2,5-Dichlorophenylacetooacetamide (Acetoacet-2,5-dichloroanilide)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	20°C	50%	U			K1=8.66 B2=15.68	1969KSe	(70147)1223

Medium: 50% dioxan, 0.025 M NaClO4

C10H9N07S2

H3L

CAS 25149-18-2 (3927)

7-Amino-1-hydroxynaphthalene-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	NaClO4	25°C	0.10M	U			K1=6.19 B2=11.06	1968BDC	(70208)1224

C10H9N302S

HL

CAS 3012-52-0 (217)

2-(2'-Thiazolylazo)-4-methoxyphenol; CH₃₀.C₆H₃(OH).N:N.C₃H₂N₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	sp	alc/w	20°C	30%	U			K1=8.8	1968SSd	(70405)1225

Medium: 30% EtOH, 0.1 M

C10H9N302S

HL

CAS 15574-54-6 (3925)

2-(2'-Thiazolylazo)-5-methoxyphenol; CH₃₀.C₆H₃(OH).N:N.C₃H₂N₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	sp	alc/w	20°C	30%	U			K1=8.1	1968SSd	(70408)1226

Medium: 30% EtOH, 0.1 M

C10H10N02Cl

HL

CAS 91573-19-2 (4783)

1-Acetoacetamido-3-chlorobenzene; CH₃.CO.CH₂.CO.NH.C₆H₄.Cl

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	20°C	50%	U			K1=9.69 B2=17.54	1969KSe	(70470)1227

Medium: 50% dioxan, 0.025 M NaClO4

C10H10N02Cl

HL

CAS 3027-00-7 (4784)

1-Acetoacetamido-4-chlorobenzene; CH₃.CO.CH₂.CO.NH.C₆H₄.Cl

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	20°C	50%	U			K1=9.78 B2=17.80	1969KSe	(70478)1228

Medium: 50% dioxan, 0.025 M NaClO4

C10H10N02Cl

HL

CAS 6144-11-0 (247)

Acetoacet-2-chloroacetanilide; CH₃.CO.CH₂.CO.NH.C₆H₄.Cl

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	20°C	50%	U			K1=8.97 B2=16.25	1969KSe	(70493)1229
Medium:	50% dioxan,	0.025 M	NaClO ₄							

C₁₀H₁₀N₂O HL CAS 37920-81-3 (3323)
8-Hydroxy-2,4-dimethylquinazoline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	20°C	50%	U			K1=8.77 B2=16.10	1954IRa	(70540)1230
Medium:	50% dioxan,	0.3 M	NaClO ₄							

C₁₀H₁₀N₂O₄ HL CAS 7418-44-2 (4726)
1-Acetoacetamido-3-nitrobenzene; CH₃.CO.CH₂.CO.NH.C₆H₄.NO₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	20°C	50%	U			K1=8.99 B2=16.50	1969KSe	(70571)1231
Medium:	50% dioxan,	0.025 M	NaClO ₄							

C₁₀H₁₀N₂O₄ HL CAS 91573-21-6 (4727)
1-Acetoacetamido-4-nitrobenzene; CH₃.CO.CH₂.CO.NH.C₆H₄.NO₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	20°C	50%	U			K1=9.39 B2=17.05	1969KSe	(70579)1232
Medium:	50% dioxan,	0.025 M	NaClO ₄							

C₁₀H₁₀N₄O₂S HL CAS 68-35-9 (1885)
4-Amino-N-(2-pyrimidinyl)benzenesulfonamide; C₄H₃N₂NHSO₂C₆H₄NH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	alc/w	25°C	50%	C			K1=5.88 B2=11.48	1993EEa	(70620)1233
								K(UO ₂ (nta)+L)=10.71		

Medium: 50% v/v EtOH/H₂O, 0.10 M NaClO₄.

C₁₀H₁₀O₂ HL CAS 93-91-4 (197)
1-Phenylbutane-1,3-dione; C₆H₅.CO.CH₂.CO.CH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	30°C	75%	U			K1=10.67 B2=20.89	1977AHb	(70779)1234
UO ₂ ++	dis	NaClO ₄	20°C	0.10M	U			K1=7.2	1960STb	(70780)1235
								K(UO ₂ +L+OH)=15.9		
								K(UO ₂ +L+2OH)=24.1		

U02++ gl diox/w 30°C 75% U K1=12.15 B2=23.27 1955H0a (70781)1236

C10H1003 HL CAS 16636-62-7 (3298)
2-Hydroxybenzoylacetone; HO.C6H4.CO.CH2.CO.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl diox/w 30°C 75% U K1=10.97 B2=21.20 1955H0a (70801)1237

C10H1006 H2L CAS 5411-14-3 (2394)
1,2-Phenylenedioxodiethanoic acid; C6H4(O.CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ dis NaClO4 25°C 0.10M C H K1=3.01 1990RCa (70863)1238
B((UO2)HL)=5.22
K(UO2+HL)=1.75

DH(K1)=16.8, DH(MHL)=10.4 kJ mol-1. DS(K1)=114, DS(MHL)=68 J K-1 mol-1

C10H11N02 L CAS 102-01-2 (250)
Acetoacetanilide; CH3.CO.CH2.CO.NH.C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl diox/w 20°C 50% U K1=9.94 B2=18.02 1969KSe (70915)1239
Medium: 50% dioxan, 0.025 M NaClO4

C10H11N50 L (2721)
4-(4'-Methylphenylazo)-3-amino-pyrazolin-5-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ sp alc/w 20°C 100% U H K1=5.94 B2=8.8 1983EAa (71087)1240

C10H11N502 L (2722)
4-(4'-Methoxyphenylazo)-3-amino-pyrazolin-5-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ sp alc/w 20°C 100% U H K1=6.60 B2=9.81 1983EAa (71098)1241

C10H12N203S HL CAS 93100-65-3 (6199)
2-(2-Pyrrolideneamino)benzene sulfonic acid; C4H7N:N.C6H4.HSO3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl NaClO4 25°C 0.10M U T H K1=17.65 1987RDb (71211)1242
35 C:K=18.32, 45 C:18.78. DH=102.52 kJ/mol-1, DS=670 J K-1 mol-1

C10H12N204		HL	(6004)							
N-Benzylloxycarbonylglycyl hydroxamic acid; C6H5.CH2.O.CO.NH.CH2.CO.NHOH										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
UO2++	gl	KNO3	25°C	0.10M	U		K1=7.6	B2=14.2	1987CSb	(71306)1243
<hr/>										
C10H12O2		HL					CAS	1946-74-3	(202)	
3-Isopropyltropolone;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
UO2++	sp	alc/w	?	50%	U		K1=9.62	B2=16.54	1965DSb	(71610)1244
Medium: 50% EtOH, 0.5 M KNO3. By glass electrode: K2=6.93										
<hr/>										
UO2++	dis	NaClO4	25°C	0.10M	U		K1=9.5	B2=18.00	1962DYa	(71611)1245
<hr/>										
C10H13NOS		HL					CAS	99075-17-9	(3339)	
2-Mercapto-N-phenylbutyramide (2-Mercaptobutyranilide)										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
UO2++	gl	diox/w	30°C	75%	U		K1=10.78	B2=20.46	1961MAe	(71703)1246
<hr/>										
C10H13NOS		HL					CAS	34282-28-5	(3338)	
N-(Mercaptoacetyl)-2,6-dimethylaniline; (CH3)2.C6H3.NH.CO.CH2.SH										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
UO2++	gl	diox/w	30°C	75%	U		K1=10.30	B2=19.47	1961MAe	(71709)1247
<hr/>										
C10H13N03S		HL					(3340)			
N-(Mercaptoacetyl)-2,5-dimethoxyaniline; HS.CH2.CO.NH.C6H3(OC(CH3)2)										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
UO2++	gl	diox/w	30°C	75%	U		K1=10.21	B2=19.31	1961MAe	(71753)1248
<hr/>										
C10H13N504		L	Adenosine				CAS	58-61-7	(2154)	
Adenosine, Adenine-9-beta-D-ribofuranoside;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
UO2++	gl	KNO3	35°C	0.10M	U	M	K1=2.9		1982RKA	(71954)1249
							K(UO2(EDTA)+L)=1.08			
<hr/>										
C10H13N505		HL	Guanosine				CAS	118-00-3	(1402)	
2-Aminopurin-6-one-9-riboside;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

U02++	gl	KNO ₃	35°C	0.10M	U	M	1997RVa (72021)1250
							K(UO ₂ +HL)=3.10 K(UO ₂ +HL+HA)=10.31 K(UO ₂ +HL+HC)=12.44
H2A is histidine, H2C is cysteine.							
U02++	gl	KNO ₃	35°C	0.10M	U	M	K1=3.1 1982RKA (72022)1251 K(UO ₂ (EDTA)+L)=2.88 K(UO ₂ (EDTA)L+H)=6.38
U02++	gl	NaNO ₃	20°C	1.0M	U		1965FBa (72023)1252 K(UO ₂ +HL)=0.7
C10H16N2O8		H4L	EDDS				CAS 52759-67-8 (1100)
1,2-Diaminoethane-N,N'-di-1,4-butanedioic acid; (CH ₂ .NH.CH(COOH)CH ₂ .COOH) ₂							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values Reference ExptNo
U02++	gl	KNO ₃	30°C	0.10M	U		K1=10.6 1971TSF (73195)1253
C10H16N2O8		H4L	EDTA				CAS 60-00-4 (120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values Reference ExptNo
U02++	gl	NaClO ₄	25°C	0.20M	U		K1=15.37 1986SLb (74274)1254
U02++	gl	NaClO ₄	25°C	3.0M	C		K1=15.65 1984BLb (74275)1255 B((UO ₂)HL)=18.59 B((UO ₂)2L)=20.24
U02++	gl	oth/un	25°C	0.10M	U	H	K1=7.40 1983LGa (74276)1256
U02++	EMF	KNO ₃	25°C	0.10M	C		19820La (74277)1257 K(2UO ₂ +OH+L)=26.2 K(2UO ₂ +2OH+L)=34.4 K(2UO ₂ +HL)=11.4 K(2UO ₂ +L)=17.8
U02++	gl	KNO ₃	25°C	0.10M	U		1970FSa (74278)1258 K(UO ₂ +HL)=7.40 K(UO ₂ (OH)HL+H)=5.62 K(2UO ₂ (OH)HL)=3.27 K(2UO ₂ +L)=17.87
K(2UO ₂ HL+2H ₂ O=(UO ₂) ₂ (OH) ₂ H ₂ L ₂ +2H)=-7.97							
U02++	gl	KNO ₃	25°C	0.10M	U	I	1968FSa (74279)1259 K(UO ₂ +HL)=7.40 K(2UO ₂ +L)=17.87

I=1.0, K(UO₂+HL)=7.35, K(2UO₂+L)=17.77

UO₂++ gl KN03 25°C 1.0M U 1968FSa (74280)1260
K(UO₂(OH)L+H)=6.30
K(UO₂(OH)HL+H)=5.62
K(2UO₂(OH)L+2H)=15.87
K((UO₂)₂(OH)₂H₂L₂+2H)=7.97

polymeric species are also formed

UO₂++ sp NaClO₄ 25°C 0.15M U M 1964BKb (74281)1261
K(UO₂+HL)=7.8
K(2UO₂+L)=17.8
K(UO₂(OH)LH+H)=5.6
K((UO₂OH)₂L+2H)=11.1

K(2UO₂LH+Ca=(UO₂)₂L+2H+CaL)=-8.2

UO₂++ sp R4N.X 24°C 0.10M U K1=10.4 1960KKa (74282)1262
K(2UO₂+L)=15.2

Medium: NH₄Cl

UO₂++ dis NaClO₄ ? 0.10M U T 1960STa (74283)1263
K(UO₂+HL)=7.32

UO₂++ sol oth/un 25°C ? U 1959KSa (74284)1264
K(UO₂+HL)=4.13
Ks(UO₂H₂L)=-5.64

C10H17N3O6S H3L Glutathione CAS 70-18-8 (333)

Glutamyl-cysteinyl-glycine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO₂++ gl NaClO₄ 25°C 1.0M U H 1992BRc (75148)1265
K(UO₂+H₂L)=2.24

DH(UO₂+H₂L)=12.6 kJ mol⁻¹, DS(UO₂+H₂L)=85 J K⁻¹ mol⁻¹

C10H18N2O7 H3L HEDTA CAS 150-39-0 (392)

N-(Hydroxyethyl)diaminoethane-N,N',N'-triethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO₂++ vlt NaClO₄ 30°C 0.50M U B2=9.2 1969LLa (75529)1266
K(UO₂+2HL)=6.4
K(UO₂+2H₂L)=5.57

UO₂++ sp NaClO₄ 25°C 0.20M U 1967BRa (75530)1267
K(UO₂+HL)=6.33
K(2UO₂+L)=16.70
K(UO₂(H₂O)HL=UO₂(OH)HL+H)=5.33

K((UO₂)₂(H₂O)2L=(UO₂)₂(OH)2L+2H)=9.93

C10H19NO2 HL (4752)
N,N-Dipropylacetooacetamide; CH₃.CO.CH₂.CO.N(CH₂.CH₂.CH₃)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	20°C	50%	U		K1=12.31	B2=23.30	1969KSe	(75628)1268
Medium: 50% dioxan, 0.025 M NaClO ₄										

C10H20N2O4 H2L CAS 5578-84-7 (5914)
N,N-Dihydroxydecanediamide: HN(OH).CO.(CH₂)₈.CO.NH(OH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	g1	NaNO ₃	25°C	0.10M	C			K1=13.28 B((UO ₂)H ₄)=16.92	1989EHa (75804)	1269

C10H20O2 HL Capric acid CAS 334-48-5 (2542)
Decanoic acid; CH₃(CH₂)₈COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	dis	non-aq	25°C	100%	U				1973NHa (75907)1270	
								K(UO ₂) ₂ (H ₂ O) ₂ · (H ₂ O) ₂ -	0.17	

Medium: benzene

C10H20O5 L 15-Crown-5 CAS 33100-27-5 (576)
1,4,7,10,13-Pentaoxacyclopentadecane; cyclo(-(O.CH₂.CH₂)₅-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	R4N.X	25°C	0.10M	U		K1=0.7	B2=3.3	1985BFa	(76145)1271
Measured in competition with Na+ K1=0.5 B2=3.8 in competition with Pb++										

C10H22N2O3 L Cryptand 2,1 CAS 31249-95-3 (835)
4,7,13-Trioxa-1,10-diazacyclopentadecane (Trioxa(2,1)cryptand);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ⁺⁺	sp	non-aq	25°C	100%	U		K1=4.96	B2=8.56	1989LMb	(76344)1272
Medium: propylene carbonate, 0.1 M Et ₄ NClO ₄										

C10H22O5 L Tetraglyme CAS 143-24-8 (121)
2,5,8,11,14-Pentaoxapentadecane; (CH₃.O.CH₂.CH₂.O.CH₂.CH₂.O)20

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ⁺⁺	sp	non-aq	25°C	100%	U	I		K1=2.99	1989LMb	(76478)1273
Medium: 0.1 M Et ₄ NClO ₄ in propylene carbonate										

C11H802 HL CAS 708-06-5 (1889)
2-Hydroxy-1-naphthaldehyde;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	alc/w	27°C	50%	U	M		K1=7.91 B((UO2)L(Gly))=14.05 B((UO2)L(Phe))=13.78 B((UO2)L(Ala))=13.27 B((UO2)L(Val))=12.89	1985RSc (76967)	1274
B(ML(Leu))=13.93, B(ML(Ser))=13.31, B(ML(Thr))=13.08, B(ML(His))=16.96										

UO2++	gl	diox/w	25°C	50%	U			K1=7.00	1974LSa (76968)	1275
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C11H803 H2L CAS 86-48-6 (1129)
1-Hydroxy-2-naphthoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	oth/un	30°C	0.10M	U	T		K1=3.42	1971PSb (77020)	1276
K1(20 C)=3.45, K1(40 C)=3.39, K1(50 C)=3.35, K1(60 C)=3.33. I=0: K1=3.88										
UO2++	sp	oth/un	30°C	?	U				1959TPa (77021)	1277
K(UO2+HL=UO2L+H)=1.62(?)										

C11H803 H2L CAS 2083-08-1 (1131)
2-Hydroxy-1-naphthoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	NaClO4	30°C	0.10M	U	T			1972PSa (77066)	1278
K(UO2+HL)=3.52										
K(20 C)=3.59, K(40 C)=3.47, K(50 C)=3.41. I=0: K(UO2+HL)=4.19										

C11H803 HL CAS 483-35-6 (3347)
2-Hydroxy-3-methyl-1,4-naphthoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	30°C	75%	U			K1=8.39 B2=15.59	1960KFc (77080)	1279

C11H803 H2L CAS 92-70-6 (1130)
2-Hydroxy-3-naphthoic acid (3-Hydroxy-2-naphthoic acid);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	oth/un	25°C	?	U				1965DEa (77134)	1280
K(UO2+HL=UO2L+H)=3.45										

C11H804	HL	CAS 7555-37-5 (4812)
3-Acetyl-4-hydroxycoumarin		
<hr/>		
Metal	Mtd Medium Temp Conc Cal Flags Lg K values	Reference ExptNo
<hr/>		
U02++	gl diox/w 35°C 50% U	K1=5.90 B2=10.51 1971MAa (77188)1281
Medium: 50% dioxan, 0.01 M NaClO4		
<hr/>		
C11H804	HL	CAS 6724-42-1 (6183)
8-Formyl-7-hydroxy-4-methyl-2H-1-benzopyran-2-one; CHO.C9H30(:O)(CH3)(OH)		
<hr/>		
Metal	Mtd Medium Temp Conc Cal Flags Lg K values	Reference ExptNo
<hr/>		
U02++	gl alc/w 35°C 70% U	K1=6.86 B2=12.40 1988Krc (77208)1282
<hr/>		
C11H806S	H3L	CAS 66695-90-7 (1996)
1-Hydroxy-4-sulfo-2-naphthoic acid;		
<hr/>		
Metal	Mtd Medium Temp Conc Cal Flags Lg K values	Reference ExptNo
<hr/>		
U02++	gl NaClO4 25°C 0.50M C	K1=11.77 B2=20.78 1988LKa (77236)1283 B((U02)H-1L2)=10.86
K1 measured by spectrophotometry		
<hr/>		
C11H806S	H3L	CAS 6407-91-6 (1994)
1-Hydroxy-7-sulfo-2-naphthoic acid;		
<hr/>		
Metal	Mtd Medium Temp Conc Cal Flags Lg K values	Reference ExptNo
<hr/>		
U02++	gl NaClO4 25°C 0.50M C	K1=13.35 B2=21.42 1988LKa (77240)1284 B((U02)H-1L)=5.92 B((U02)H-1L2)=11.69
K1 measured by spectrophotometry		
<hr/>		
C11H809S2	H4L	CAS 67097-84-1 (1995)
1-Hydroxy-4,7-disulfo-2-naphthoic acid;		
<hr/>		
Metal	Mtd Medium Temp Conc Cal Flags Lg K values	Reference ExptNo
<hr/>		
U02++	gl NaClO4 25°C 0.50M C	K1=10.94 B2=18.83 1988LKa (77289)1285 B3=22.23
<hr/>		
C11H809S2	H4L	CAS 67097-83-0 (1618)
3-Hydroxy-5,7-disulfo-2-naphthoic acid;		
<hr/>		
Metal	Mtd Medium Temp Conc Cal Flags Lg K values	Reference ExptNo
<hr/>		
U02++	gl NaClO4 25°C 0.50M C	K1=9.809 B2=17.398 1978Lkb (77296)1286
<hr/>		
C11H9N02	H2L	CAS 7470-09-9 (8481)

2-Hydroxy-1-naphthaldoxime;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ gl diox/w 25°C 75% U K1=11.30 B2=20.91 1978MCd (77319)1287
Medium: 75% v/v dioxane/H₂O, 0.10 M NaClO₄.

C11H9N02S HL CAS 29556-13-6 (1450)
N-Phenyl-2-thenoylhydroxamic acid; C₄H₃SCON(C₆H₅)OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ gl diox/w 25°C 50% M T H K1=9.88 B2=18.33 1977ABb (77352)1288

C11H9N03 HL CAS 1137-48-0 (1449)
N-Phenyl-2-furylhydroxamic acid; C₄H₃O.CO.N(C₆H₅).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ gl NaClO₄ 30°C 0.10M U K1=8.14 B2=16.05 1969DSb (77394)1289

C11H9N04 H₂L CAS 4321-82-7 (4829)
3-Acetyl-4-hydroxycoumarin oxime;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ gl diox/w 35°C 50% U 1971MAa (77431)1290
K(UO₂+HL)=5.83
K(UO₂+2HL)=10.40

Medium: 50% dioxan, 0.01 M NaClO₄

C11H9N30 HL CAS 10335-29-2 (3937)
2-(2'-Pyridylazo)phenol; C₅H₄N.N:N.C₆H₄.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ sp alc/w 20°C 50% U K1=10.7 1967ANa (77462)1291

Medium: 50% MeOH, 0.1 M NaClO₄

C11H9N302 H₂L PAR CAS 1141-59-9 (636)
4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C₅H₄N.N:N.C₆H₃(OH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ sp oth/un 20°C 0.10M U K1=11.9 1967SI_d (77595)1292
K(UO₂+HL)=12.9

UO₂++ gl diox/w 25°C 50% U I K1=16.2 B2=25.80 1962GNa (77596)1293
Medium: 50% dioxan, 0.1 M. In 0% dioxan: K1=12.5, K2=8.4

C11H9N305S		HL	(6249)						
1,2-Naphthoquinone-4-sulfonic acid 2-semicarbazone; C10H5(:O)(HSO ₃):N.NH.CO.NH2									
<hr/>									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
<hr/>									
U02++	gl	NaClO ₄	28°C	0.10M	U	T H	K1=6.92 B2=12.94	1980MGd	(77643)129
<hr/>									
C11H11N06		H3L					CAS 1147-65-5	(425)	
N-(2'-Carboxyphenyl)iminodiethanoic acid; HOOC.C ₆ H ₄ .N(CH ₂ .COOH) ₂									
<hr/>									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
<hr/>									
U02++	gl	KNO ₃	25°C	0.10M	U		K1=9.71 B2=17.99	1982NBa	(77839)129
<hr/>									
C11H11N202Br		HL					(9228)		
3-[4-Bromophenylazo]penta-2,4-dione;									
<hr/>									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
<hr/>									
U02++	gl	alc/w	25°C	0.1M	U		K1=7.29	2004GMc	(77878)1296
Medium: 0.1 mol/L KCl in 3:7 EtOH/H ₂ O mixture									
<hr/>									
C11H11N202I		HL					(9227)		
3-[4-Iodophenylazo]penta-2,4-dione;									
<hr/>									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
<hr/>									
U02++	gl	alc/w	25°C	0.1M	U		K1=7.26	2004GMc	(77901)1297
Medium: 0.1 mol/L KCl in 3:7 EtOH/H ₂ O mixture									
<hr/>									
C11H11N305		L					(7162)		
2-(2'-Thiazolylazo)-4,6-dimethylphenol; C ₉ H ₁₁ NS.N:C ₆ H ₂ (CH ₃) ₂ OH									
<hr/>									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
<hr/>									
U02++	sp	alc/w	rt	40%	U		K1=9.04 B2=17.37	1988SSh	(77903)129
Medium: 40% v/v EtOH/H ₂ O, 0.25 M NaClO ₄									
<hr/>									
C11H11N302S		HL	Sulfapyridine		CAS 144-83-2		(8356)		
4-Amino-N-2-pyridinyl-benzenesulfonamide;									
<hr/>									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
<hr/>									
U02++	gl	alc/w	25°C	50%	C	M	K1=9.59 B2=15.71 K(UO ₂ (nta)+L)=6.55	1993EEa	(77935)129
Medium: 50% v/v EtOH/H ₂ O, 0.10 M NaClO ₄ .									
<hr/>									
C11H11N304		HL					(9230)		
3-[4-Nitrophenylazo]penta-2,4-dione;									

C11H13N02 HL CAS 3026-99-1 (249)
Acetoacet-2-toluidide; CH₃.CO.CH₂.CO.NH.C₆H₄.CH₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=9.35 B2=17.03 1969KSe (78466)1314
Medium: 50% dioxan, 0.025 M NaClO₄

C11H13N02 HL CAS 20222-64-4 (4842)
N-3-Tolylacetoacetamide; CH₃.CO.CH₂.CO.NH.C₆H₄.CH₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=10.14 B2=18.37 1969KSe (78474)1315
Medium: 50% dioxan, 0.025 M NaClO₄

C11H13N03 HL CAS 101374-66-7 (4844)
1-Acetoacetamido-3-methoxybenzene; CH₃.CO.CH₂.CO.NH.C₆H₄.OCH₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=10.0 B2=18.18 1969KSe (78484)1316
Medium: 50% dioxan, 0.025 M NaClO₄

C11H13N03 HL CAS 3006-35-7 (4845)
1-Acetoacetamido-4-methoxybenzene; CH₃.CO.CH₂.CO.NH.C₆H₄.OCH₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=10.28 B2=18.70 1969KSe (78492)1317
Medium: 50% dioxan, 0.025 M NaClO₄

C11H13N03 HL CAS 91099-10-4 (246)
Acetoacet-2-aniside; CH₃.CO.CH₂.CO.NH.C₆H₄.OCH₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=9.70 B2=17.76 1969KSe (78522)1318
Medium: 50% dioxan, 0.025 M NaClO₄

C11H16N2010 H5L CEDTA CAS 62394-58-5 (1080)
1-Carboxy-1,2-diaminoethane-N,N,N',N'-tetraethanoic acid;
(HOOCCH₂)₂NCH(COOH)CH₂N(CH₂COOH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KCl 25°C 0.10M U 1987HGa (79113)1319

B((U02)H3L)=23.04
B((U02)H2L)=20.11

$B((UO_2)HL)=17.06$
 $B((UO_2)2L_2)=26.27$
 $B((UO_2)2HL)=19.99; B((UO_2)2L)=16.53; B((UO_2)4H-2L_2)=26.32; B((UO_2)4H-4L_2)=15.31$

C11H18N2O8 H4L CAS 4408-81-5 (923)
1,3-Diaminopropane-N,N,N',N'-tetraethanoic acid; ((HOOC.CH2)2N.CH2.)2.CH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	KNO ₃	25°C	0.10M	C	I			1984GMb (79474)1320	
								$B(UO_2HL)=18.80$		
								$B((UO_2)2L)=18.66$		
								$B((UO_2)2H-1L)=14.05$		
								$B((UO_2)2L_2)=30.2$		

$B((UO_2)4H-4L_2)=20.64.$ For I=1.0 M KNO₃: $B(UO_2HL)=18.12,$ $B((UO_2)2L)=17.3,$
 $B((UO_2)2H-1L)=12.44,$ $B((UO_2)2L_2)=28.44,$ $B((UO_2)4H-4L_2)=17.96.$

UO ₂ ++	gl	KNO ₃	25°C	0.10M	C	I		1984GMb (79475)1321	
								*K((UO ₂)2L)=-4.61	
								K(2UO ₂ HL=(UO ₂ (OH)H-1L)2+2H)=-7.39; K'(2(UO ₂)2L=(UO ₂ L)2(OH)4+4H)=-16.75	
								In 1.0 M KNO ₃ : K(2UO ₂ HL=(UO ₂ (OH)H-1L)2+2H)=-7.83; *K((UO ₂)2L)=-4.89;	

UO ₂ ++	gl	KNO ₃	25°C	0.10M	U			1968FSa (79476)1322	
								K(UO ₂ +HL)=8.94	

C11H20O2 HL Dipivaloylmeth. CAS 1118-71-4 (363)
2,2,6,6-Tetramethyl-3,5-heptanedione; (CH₃)₃C.CO.CH₂.CO.C(CH₃)₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	30°C	75%	U			K1=12.11 B2=23.92	1977AHb (79753)1323	

C12H8N2 L Phenanthroline CAS 66-71-7 (144)
1,10-Phenanthroline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	NaClO ₄	30°C	0.10M	M	M	K1=3.90		1982RSb (80527)1324	
								K(UO ₂ L+OH)=9.93		
								K(UO ₂ (OH)L+A)=10.15		
								K(UO ₂ +OH+A+L)=23.98		
								K(UO ₂ (OH)L+B)=3.24		

$K(UO_2+OH+B+L)=17.07.$ H₂A=N,N'-1,2-ethanediylbis(2-mercaptopropanamide),
H₂B=N,N'-1,2-ethanediylbis(3-mercaptopropanamide).

C12H9N03 HL CAS 63098-85-1 (6279)
2-(N-2'-Furfuralideneimino)benzoic acid; C₄H₃₀.CH:N.C₆H₄.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl NaClO4 25°C 0.10M U TI K1=4.32 B2=7.58 1978SKg (80582)1325

C12H10N20 HL CAS 1823-47-8 (3969)
2-Salicylideneaminopyridine; (2-OH).C6H4.CH:N.C5H4N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp alc/w 20°C 100% U H K1=5.53 1984EAb (80678)1326

Data also for related hydroxybenzilidene-aminopyridines, -aminopyrimidines, -amino-1,2,4-triazine.

U02++ gl diox/w 25°C 50% U K1=8.1 1962GNb (80679)1327

C12H10N6O4S H2L CAS 77327-19-6 (8343)
2-[4-Amino-3-(1,2,4-triazolylazo)]naphthol-4-sulphonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 30°C 0.10M U T K1=5.37 B2=10.16 1981GMi (80789)1328

Also data for 40-50 C.

C12H11N02S HL CAS 29556-14-7 (2049)
N-(4-Tolyl)-2-thenoylhydroxamic acid; C4H3SCON(OH)C6H4CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 50% M T H K1=9.94 B2=18.43 1977ABb (80836)1329

50% v/v dioxan -water; Data also for Pd(II), Cu(II), Zn, Ni, Co, Mn
also values of K at 35 C and DH values

C12H11N30S HL (6787)

2-Hydroxy-1-naphthaldehyde thiosemicarbazone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 75% U K1=10.09 B2=18.47 1992SSc (80897)1330

Medium: 75% v/v dioxan/H2O and other mixtures, 0.1 M NaClO4

C12H11N302 HL CAS 50536-09-5 (6323)

2-Hydroxy-1-naphthaldehyde-semicarbazone; HO.C10H6.CH:N.NH.CO.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 75% U K1=9.97 B2=19.33 1992SSc (80926)1331

Medium: 75% v/v dioxan/H2O and other mixtures, 0.1 M NaClO4

C12H12N20 HL CAS 70301-52-9 (1940)

2-(Hydroxyphenyliminomethyl)pyridine; C5H4N.CH2.NH.C6H4.OH

C12H14O3 HL CAS 543-05-8 (4900)
Ethyl 2-phenylacetooacetate; CH₃.CO.CH(C₆H₅).CO.O.CH₂.CH₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 75% U K1=12.90 1973AAa (81402)1340

C12H15N02 HL (248)
Acetoacet-2,4-dimethylanilide; CH₃.CO.CH₂.CO.CH₂.NH.C₆H₃(CH₃)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U T K1=9.84 B2=17.98 1969KSe (81445)1341

Medium: 50% dioxan, 0.025 M NaClO₄

C12H15N02 HL (4921)
N-3,5-Dimethylphenylacetooacetamide; CH₃.CO.CH₂.CO.NH.C₆H₃(CH₃)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=9.83 B2=17.93 1969KSe (81450)1342

Medium: 50% dioxan, 0.025 M NaClO₄

C12H15N04 HL (4922)
1-Acetoacetamido-2,4-dimethoxybenzene; CH₃.CO.CH₂.CO.NH.C₆H₃(OCH₃)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=10.05 B2=18.36 1969KSe (81470)1343

Medium: 50% dioxan, 0.025 M NaClO₄

C12H15N04 HL (4923)
1-Acetoacetamido-2,5-dimethoxybenzene; CH₃.CO.CH₂.CO.NH.C₆H₃(OCH₃)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=9.47 B2=17.28 1969KSe (81475)1344

Medium: 50% dioxan, 0.025 M NaClO₄

C12H17NOS HL CAS 34282-27-4 (3393)
N-(2,6-Diethylphenyl)mercaptoacetamide; HS.CH₂.CO.NH.C₆H₃(CH₂.CH₃)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 75% U K1=10.42 B2=19.68 1961MAe (81711)1345

C12H18N2010 H5L CAS 105147-09-9 (1081)

1-Carboxy-1,3-diaminopropane-N,N,N',N'-tetraethanoic acid;
(HOOCCH₂)₂NCH(COOH)(CH₂)₂N(CH₂COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KNO3	25°C	0.10M	U			K1=13.64 B((UO2)H3L)=25.00 B((UO2)H2L)=22.25 B((UO2)HL)=19.36 B((UO2)H-1L)=7.33	1987HGa	(81911)1346
$B((UO2)2L2)=29.62; B((UO2)2HL)=22.16; B((UO2)2L)=18.61; B((UO2)4H-2L2)=30.12$										
$B((UO2)4H-4L2)=19.21$										

C12H20N208			H4L					CAS 40623-42-5	(1101)	
1,2-Diaminoethane-N,N'-di(2-pentane-1,5-dioic acid); (CH2NHCH(COOH)CH2CH2COOH)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KNO3	30°C	0.10M	U			K1=12.55	1971TSf	(82107)1347

C12H20N208			H4L					CAS 2458-58-4	(922)	
1,4-Diaminobutane-N,N,N',N'-tetraethanoic acid; (HOOC.CH2)2N.(CH2)4.N(CH2.COOH)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KNO3	25°C	0.10M	C	I			1984GMb	(82239)1348
$B(UO2HL)=19.61$										
$B((UO2)2L)=19.06$										
$B((UO2)2H-1L)=13.83$										
$B((UO2)2L2)=31.04$										
$B((UO2)4H-4L2)=19.76.$ For I=1.0 M KNO3: $B(UO2HL)=19.21,$ $B((UO2)2L)=18.38,$										
$B((UO2)2H-1L)=13.44,$ $B((UO2)2L2)=30.33,$ $B((UO2)4H-4L2)=20.15.$										
UO2++	gl	KNO3	25°C	0.10M	C	I			1984GMb	(82240)1349
$*K((UO2)2L)=-5.26$										
$K(2UO2HL=(UO2(OH)H-1L)2+2H)=-8.18;$ $K'(2(UO2)2L=(UO2L)2(OH)4+4H)=-18.35$										
In 1.0 M KNO3: $K(2UO2HL=(UO2(OH)H-1L)2+2H)=-8.22;$ $*K((UO2)2L)=-4.95;$										

C12H2406			L	18-Crown-6				CAS 17455-13-9	(577)	
1,4,7,10,13,16-Hexaoxacyclooctadecane;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	nmr	non-aq	27°C	100%	C	I		K1=1.51	2001KZa	(83672)1350
Method: 7Li nmr; competitive binding study. Medium: nitromethane.										
In acetonitrile, K1=1.06										
UO2++	sp	non-aq	25°C	100%	U	I		K1=5.29	1989LMb	(83673)1351
Medium: 0.1 M Et4NClO4 in propylene carbonate										
In acetonitrile, K1=3.80										
UO2++	gl	R4N.X	25°C	0.10M	U			K1=2.1	B2=3.9	1985BFa (83674)1352

Measured in competition with Na+. K1=2.0, B2=3.7 in competition with Pb++

U02++ sp non-aq 25°C 100% U K1=5.29 1985BFa (83675)1353
Medium: propylene carbonate. In H2O, by potentiometry: K1=2.0, B2=3.7

U02++ ISE non-aq 25°C 100% C K1=5.31 1984FLa (83676)1354
In propylenecarbonate; electrolyte Et4NClO4

C12H26N2O4 L Cryptand 2,2 CAS 23978-55-4 (925)
4,7,13,16-Tetraoxa-1,10-diazacyclooctadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp non-aq 25°C 100% U K1=7.45 B2=12.40 1989Lmb (83913)1355
Medium: propylene carbonate, 0.1 M Et4NClO4

U02++ sp non-aq 25°C 100% U K1=7.45 B2=12.40 1985BFa (83914)1356
B((UO2)2L)=14.49

Medium: propylene carbonate, 0.01 M Et4NClO4

U02++ gl R4N.X 25°C 0.10M C K1=10.87 1983SEa (83915)1357
Medium: 0.10 M Me4NCl.

C12H27O4P L CAS 126-73-8 (2432)
Tri-n-butyl phosphate; (C4H9O)3PO

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp non-aq 20°C 100% U 1983Kbc (84123)1358
K(UO2Cl2+L)=1.99
K(UO2Cl2+2L)=3.68

Medium: acetone

C13H8N2O3C12 HL (6202)
2-Carboxy-2'-hydroxy-3',5'-dichloroazobenzene; HOOC.C6H4.N:N.C6H2(OH)C12

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 70% U I K1=14.98 B2=26.89 1987Kbc (84473)1359

C13H8O3 HL CAS 719-41-5 (3397)
1-Hydroxyxanthone (1-Hydroxy-9-xanthenone)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp alc/w 25°C 50% U K1=9.97 1968GDb (84499)1360
Medium: 50% EtOH, 0.1 M NaClO4

C13H9F02S HL CAS 43191-66-8 (6154)
1-(2'-Thienyl)-3"-fluoro-2"-hydroxyphenyl)-prop-1-one-2-ene;

$$\text{C}_4\text{H}_3\text{S} \cdot \text{CH}:\text{CH} \cdot \text{CO} \cdot \text{C}_6\text{H}_3(\text{OH})\text{F}$$

C13H10N204 HL CAS 2029-61-0 (178)
N-Phenyl-2-nitrobenzohydroxamic acid; O2N.C6H4.CO.N(C6H5).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 50% U T K1=10.20 B2=18.65 1977VKA (84901)1368
At 35 C: K1=10.00, K2=8.25

U02++ EMF mixed 30°C 50% U K1=7.12 B2=12.84 1970GSF (84902)1369
Medium: 50% v/v acetone/H2O, 0.5 M NaClO4

C13H10N204 HL CAS 17120-18-2 (220)
N-Phenyl-3-nitrobenzohydroxamic acid; O2N.C6H4.CO.N(C6H5).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 50% U T K1=10.38 B2=19.00 1977VKA (84911)1370
At 35 C: K1=10.13, K2=8.45

U02++ EMF mixed 30°C 50% U K1=7.90 B2=14.00 1970GSF (84912)1371
Medium: 50% v/v acetone/H2O, 0.5 M NaClO4

C13H10N205 H3L (1389)
2,4-Dihydroxy-5-nitrobenzophenone oxime; (HO)2(NO2)C6H2.C(:NOH)C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 50% U K1=9.70 B2=18.55 1982UVa (84919)1372

C13H10N205S H2L CAS 98789-35-6 (5012)
4-Hydroxy-3-formylazobenzene-4'-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ EMF alc/w 25°C 42% U 1972DSc (84924)1373
K(U02+HL=U02L+H)=4.83
K(U02L+HL=U02L2+H)=4.19

Medium: 42% EtOH, 0.2 M NaClO4

U02++ sp oth/un 30°C aq U B2=7.13 1972DTb (84925)1374

C13H10N206S H2L MordentYellow10 CAS 21542-82-5 (1390)
5-(4'-Sulfophenylazo)salicylic acid; HO3S.C6H4.N:N.C6H3(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl oth/un 20°C 0.10M M T H K1=8.2 1978MBe (84943)1375

Medium: 0.10 M KClO4. Data for 44 C. DH and DS values reported.

C13H11N02 H2L (1383)

2-Hydroxybenzophenone oxime; HO.C6H4.C(:NOH)C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	30°C	50%	U			K1=9.17 B2=17.10	1982UVa (85077)	1376

C13H11N02 H2L CAS 78-75-2 (6258)

3-(Salicylideneamino)phenol; HO.C6H4.CH:N.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	alc/w	25°C	50%	U			K1=11.30 B2=17.10	1977DWa (85089)	1377

C13H11N02 HL CAS 91-40-7 (1276)

N-Phenyl-anthranilic acid; C6H5.NH.C6H4.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	30°C	50%	U			K1=4.95	1973RSa (85100)	1378

Medium: 50% dioxan, 0.1 M NaClO4

C13H11N02 HL CAS 304-88-1 (181)

N-Phenylbenzohydroxamic acid; C6H5.CO.N(C6H5).OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	EMF	mixed	30°C	50%	U			K1=8.52 B2=15.04	1970GSf (85183)	1379

Medium: 50% acetone, 0.5 M NaClO4

UO2++	gl	NaClO4	30°C	0.10M	U			K1=8.77 B2=16.98	1969DSb (85184)	1380
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C13H11N03 H3L CAS 3147-44-2 (1388)

2,4-Dihydroxy-benzophenone oxime; (HO)2C6H3.C(:NOH)C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	30°C	50%	U			K1=11.04 B2=19.57	1982UVa (85195)	1381

C13H11N03 HL (4987)

N-Phenylsalicylohydroxamic acid; HO.C6H4.CO.N(C6H5)OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	EMF	mixed	30°C	50%	U			K1=5.58 B2=10.00	1969GMc (85204)	1382

Medium: 50% acetone/H2O, 0.5 M NaClO4

C13H11N203F3 HL (5563)

3-(2-Acetylphenylhydrazone)-1,1,1-trifluoropentane-2,4-dione;

CF3.CO.C(CO.CH3):N.HN.C6H4.COCH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	25°C	75%	U			K1=10.75 B2=20.60	1990ASb (85255)	1383

C13H11N3O2		HL					(4984)			
1-Isonicotinyl-2-salicylidene hydrazone; C5H4N.CO.NH.N:CH.C6H4.OH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	alc/w	36°C	60%	U				1970GPb (85269)	1384
K(?)=4.2										
Medium: 60% EtOH, 0.02 M KCl										

C13H11N3O2		H2L					CAS 62031-25-8	(1119)		
4-Hydroxy-3-oximinomethylazobenzene; (HO)(HO.N:CH)C6H3.N:N.C6H5										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	alc/w	25°C	42%	U			K1=6.16 B2=11.94	1974MSb (85281)	1385

C13H11N3O5S		H3L					(5019)			
4-Hydroxy-3-oximinomethylazobenzene-4'-sulfonic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	alc/w	25°C	50%	U			K1=5.22 B2=9.84	1973DSa (85301)	1386
Medium: 42% EtOH, 0.2 M NaClO4										

C13H12N2O		HL					CAS 69067-12-5	(4986)		
Benzanilidoxime; C6H5.C(:N.OH).NH.C6H5										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	diox/w	25°C	50%	U			K1=10.22 B2=20.05	1969MKd (85336)	1387
Medium: 50% dioxan, 0.1 N NaClO4										

C13H12N2O		HL					(2728)			
Salicylidene phenyl hydrazone; HO.C6H4.CH:N.NH.C6H5										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	alc/w	20°C	100%	U	H		K1=4.22 B2=8.76	1983EAb (85347)	1388

C13H12N4O		L	Diphenylcarbaz.				CAS 538-62-5	(1195)		
Diphenylcarbazone; C6H5.NH.NH.CO.N:N.C6H5										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	non-aq	25°C	100%	U	T	HM		1976EWb (85421)	1389

$$K((UO_2)_2(NO_3)_2(TBP)_2+L) = -0.4$$

TBP=tributylphosphate. Medium:dichloromethane. In tetrachloromethane, $K=0.48$

C13H13NO HL CAS 24403-51-8 (3410)

1,2,3,4-Tetrahydro-9-hydroxyacridine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl diox/w 20°C 50% U K1=10.10 B2=18.30 1954IRa (85492)1390
Medium: 50% dioxan, 0.3 M NaClO4

C13H14N2O3 HL (4940)

3-(2-Acetylphenylhydrazone)pentane-2,4-dione; (CH₃.CO)2C:N.NH.C₆H₄(CO.CH₃)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl diox/w 25°C 75% U K1=13.16 B2=25.92 1990ASb (85618)1391

C13H15N02 HL (4990)

2-Butoxy-8-hydroxyquinoline; CH₃.CH₂.CH₂.CH₂.O.C₉H₅N.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl diox/w 25°C 50% U K1=10.39 1971CAd (85701)1392
Medium: 50% dioxan, 0.1 M NaClO4

C13H15N02 HL (4991)

7-t-Butoxy-8-hydroxyquinoline; (CH₃)₃C.O.C₉H₅N.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl diox/w 25°C 50% U K1=13.4 B2=25.00 1971CAd (85703)1393
Medium: 50% dioxan, 0.1 M NaClO4

C13H15N3O5 HL CAS 76877-50-4 (1291)

2-(4',5'-Dimethyl-2-thiazolylazo)-4,6-dimethylphenol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ sp alc/w rt 40% U K1=8.75 B2=19.56 1988SSh (85861)1394
Room temperature. Medium: 0.25 M NaClO4 in 40% v/v EtOH/H₂O

C13H17NO HL (3412)

4-(2,6-Dimethylphenylimino)pentan-2-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl diox/w 30°C 50% U K1=11.44 B2=21.18 1961MJa (85968)1395

C13H17N3O5 HL (6006)

N-Benzylloxycarbonyl-alanylglycyl hydroxamic acid;
C6H5.CH2.O.CO.NH.CH(CH3).CO.NH.CH2.CO.NHOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl KNO₃ 25°C 0.10M U K1=6.4 B2=12.4 1987CSb (86016)1396

C13H18N2O4 L (6005)
N-Benzylloxycarbonyl-valyl hydroxamic acid; C6H₅.CH₂.O.CO.NH.CH(CH₃)₂.CO.NHOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl KNO₃ 25°C 0.10M U K1=6.7 B2=12.1 1987CSb (86034)1397

C13H20N2O10 H5L CAS 88897-18-1 (1082)
1-Carboxy-1,4-diaminobutane-N,N,N',N'-tetraethanoic acid;
(HOOCC₂)₂NCH(COOH)(CH₂)₃N(CH₂COOH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl KNO₃ 25°C 0.10M U K1=14.14 1987HGa (86134)1398
B((UO₂)H3L)=25.17
B((UO₂)H2L)=22.96
B((UO₂)HL)=19.91
B((UO₂)H-1L)=7.73

B((UO₂)2L2)=31.34; B((UO₂)2HL)=22.99; B((UO₂)2L)=19.22; B((UO₂)4H-2L2)=31.86
B((UO₂)4H-4L2)=20.86

C14H8N3O8S2F3 HL (9231)
1-(2-Thenoyl),4-trifluoro,2-[2-hydroxy-2-sulpho-5-nitrophenylazo]butadi-1,3-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl KCl 25°C 0.1M U K1=8.10 B2=14.72 2004ACa (86612)1399

C14H8O4 H2L CAS 117-10-8 (3425)
1,8-Dihydroxyanthraquinone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl diox/w 30°C 75% U K1=12.13 B2=23.16 1960KFc (86676)1400

C14H8O7S H3L DASA CAS 83-61-4 (950)
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ sp NaClO₄ 30°C 0.15M U 1963SDa (86765)1401
K(?)=4.5

U02++ sp NaClO₄ 25°C 0.15M U K1=4.22 1960SDa (86766)1402
 At 30 C: K1=4.56 (I=0.1 M)

C14H9N02 HL CAS 641-63-4 (4038)
 2-(2'-Pyridyl)indan-1,3-dione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	30°C	75%	U			K1=11.76 B2=22.37	1964CMB	(86790)1403

C14H9N04 H2L Alizarin Maroon CAS 3963-78-8 (1052)
 3-Amino-1,2-dihydroxyanthraquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	alc/w	25°C	0.10M	U			K1=6.45 B2=11.52	1986SIB	(86815)1404

Medium: 40% v/v EtOH/H₂O, 0.1 M NaClO₄. K(UO₂L+A)=4.87; K(UO₂A+L)=6.72;
 B((UO₂)LA)=11.32, H₂A=thiosalicylic acid

C14H10N02F HL CAS 87221-43-0 (6155)
 1-(2'-Pyridyl)-3-(3-fluoro-2-hydroxyphenyl)-prop-1-one-2-ene;
 C₅H₄N.CH:CH.CO.C₆H₃(OH)F

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	NaClO ₄	30°C	0.10M	U			K1=2.97	1989SHa	(86890)1405

Data also for the 2-hydroxy-3-ethyl-5-fluoro analogue for all metal.

C14H11N03 H2L CAS 7316-93-5 (5047)
 N-Salicylideneanthranilic acid; HO.C₆H₄.CH:N.C₆H₄.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	30°C	50%	U			K1=11.15	1971MGa	(86952)1406

Medium: 50% dioxan, 0.1 M NaClO₄

C14H11N03 H2L CAS 67707-86-2 (8476)
 Salicylideneaniline-3-carboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	25°C	30%	U			K1=6.95	1978CPb	(86958)1407

Medium: 30% v/v dioxane/H₂O, 0.20 M NaClO₄.

C14H11N04 HL (2727)
 Salicylidene-4-amino salicylic acid; HO.C₆H₄.CH:N.C₆H₃(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	alc/w	27°C	40%	M			K1=11.48 B2=21.72	1993MRA	(86981)1408

Medium: 40% v/v EtOH/H₂O, 0.10 M NaCl.

U02++ sp alc/w 20°C 100% U H K1=4.9 1983EAb (86982)1409

C14H11N05 H4L CAS 245062-92-4 (8423)
4-[(E)-[(2,4-Dihydroxyphenyl)methylene]amino-2-hydroxybenzoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 27°C 40% M K1=11.38 B2=17.42 1993MRa (86985)1410

Medium: 40% v/v EtOH/H₂O, 0.10 M NaCl.

C14H11N508S2 H5L CAS 1105-53-9 (5084)
1,5-Bis(2-hydroxy-5-sulfophenyl)-3-cyanoformazan;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaNO₃ 20°C 0.10M U K1=20.19 1971SEa (87021)1411

C14H12N203 H2L CAS 4870-46-6 (3432)
2-Hydroxy-5-methyl-2'-carboxy-azobenzene; HO.C₆H₃(CH₃).N:N.C₆H₄.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp none 25°C 0.0 U K1=11.63 1984MSc (87225)1412

C14H12N204 HL (179)
N-3-Tolyl-3-nitrobenzohydroxamic acid; O₂N.C₆H₄.CO.N(C₆H₄.CH₃).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 50% U T K1=10.10 B2=18.48 1977VKa (87267)1413

At 35 C: K1=9.90, K2=8.21

C14H12N204 HL CAS 85407-74-5 (180)
N-4-Tolyl-2-nitrobenzohydroxamic acid; O₂N.C₆H₄.CO.N(C₆H₄.CH₃).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 50% U T K1=10.45 B2=19.15 1977VKa (87280)1414

At 35 C: K1=10.20, K2=8.45

C14H12N204 HL (221)
N-4-Tolyl-3-nitrobenzohydroxamic acid; O₂N.C₆H₄.CO.N(C₆H₄.CH₃).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 50% U T K1=10.70 B2=19.71 1977VKa (87293)1415

At 35 C: K1=10.45, K2=8.75

C14H13N03 H2L (5064)
3-Methoxysalicylaldehyde aminophenol Schiff base

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ dis oth/un 0.30M U K1=4.34 1968ZSa (87542)1423

Medium: 0.3 M, acetate buffer

C14H13N03 H2L CAS 51931-02-1 (5063)
N-(2-Hydroxy-1-naphthalidene)-beta-alanine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ oth NaClO4 30°C 0.10M U K1=9.30 1972MSe (87552)1424

C14H13N03 HL CAS 13664-49-8 (5065)
N-Phenyl-(4-methoxybenzo)hydroxamic acid; CH30.C6H4.CO.N(C6H5)OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 30°C 0.10M U K1=8.68 B2=17.03 1969DSb (87556)1425

C14H22N2010 H5L (1083)
1-Carboxy-1,5-diaminopentane-N,N,N',N'-tetraethanoic acid;
(HOOCCH2)2NCH(COOH)(CH2)4N(CH2COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KN03 25°C 0.10M U K1=14.27 1987HGa (88900)1426

B((UO2)H3L)=25.32

B((UO2)H2L)=23.17

B((UO2)HL)=20.05

B((UO2)H-1L)=7.73

B((UO2)2L2)=31.03; B((UO2)2HL)=23.08; B((UO2)2L)=19.47; B((UO2)4H-2L2)=32.30
B((UO2)4H-4L2)=21.15

C14H22O5 H2L CAS 85785-29-1 (2250)
Di(hepta-4,6-dione)ether, (CH3.CO.CH2.CO.(CH2)3)20

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 24°C 50% U K1=11.8 1979ACa (88995)1427

C14H23N3010 H5L DTPA CAS 67-43-6 (238)
Diethylenetriamine-pentaethanoic acid; HOOC.CH2.N(CH2.CH2.N(CH2.COOH)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 20°C 1.0M U M K1=14.0 1998BMA (89427)1428

K(UO2L+H)=5.50

$K(UO_2HL+H) = 4.15$
 $K(UO_2H_2L+H) = 2.45$
 $K(UO_2+UO_2L) = 5.5$
 $K(2UO_2CrL + 2H_2O = (UO_2)_2(OH)_2(CrL)2 + 2H) = -5.60$, $K(UO_2+CrL) = 6.70$. Cr=Cr(III)

UO ₂ ++	EMF	KNO ₃	25°C	0.10M	C	19820La (89428)1429
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$K(2UO_2+HL) = 27.3$
 $K(UO_2+HL) = 8.8$
 $K(2UO_2+HL) = 8.8$
 $K(2UO_2+L) = 19.0$

$K(2UO_2+2OH+L) = 35.1$

UO ₂ ++	sp	NaClO ₄	30°C	0.10M	U	1980KJa (89429)1430
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$B((UO_2)H_3L) = 26.9$
 $B((UO_2)2H_3L) = 31.2$
 $B((UO_2)2HL) = 22.9$
 $B((UO_2)HL) = 18.8$

C14H24N208 H4L HMDTA CAS 1633-00-7 (920)
1,6-Diaminohexane-N,N,N',N'-tetraethanoic acid; ((HOOC.CH₂)₂N.CH₂.CH₂.CH₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	KNO ₃	25°C	0.10M	C	I			1984GMb (89613)1431	

$B(UO_2HL) = 20.22$
 $B((UO_2)2L) = 19.43$
 $B((UO_2)2H-1L) = 14.23$
 $B((UO_2)2L2) = 31.89$

$B((UO_2)4H-4L2) = 20.18$. For I=1.0 M KNO₃: $B(UO_2HL) = 19.3$, $B((UO_2)2L) = 18.74$,
 $B((UO_2)2H-1L) = 13.1$, $B((UO_2)2L2) = 30.98$, $B((UO_2)4H-4L2) = 19.7$.

UO ₂ ++	gl	KNO ₃	25°C	0.10M	C	I	1984GMb (89614)1432
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$*K((UO_2)2L) = -5.20$

$K(2UO_2HL = (UO_2(OH)H-1L)2 + 2H) = -8.55$; $K'(2(UO_2)2L = (UO_2L)2(OH)4 + 4H) = -16.68$
 In 1.0 M KNO₃: $K(2UO_2HL = (UO_2(OH)H-1L)2 + 2H) = -7.98$; $*K((UO_2)2L) = -5.65$;

UO ₂ ++	gl	KNO ₃	25°C	0.10M	U	1968FSa (89615)1433
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$K(UO_2+HL) = 9.96$

C14H24N2010 EGTA CAS 67-42-5 (349)
Ethyleneglycol-0,0'-bis(2-aminoethyl ether)-N,N,N',N'-tetraethanoic acid; H4L

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	NaNO ₃	25°C	1.10M	U			K1=13.44	1995ADc (89955)1434	
UO ₂ ++	gl	KNO ₃	25°C	0.10M	U			K1=11.23 B2=19.03	1982NBa (89956)1435	
UO ₂ ++	gl	KNO ₃	25°C	0.10M	U			K1=9.41	1970FSa (89957)1436	

$B((UO_2)2L) = 17.66$

					K(UO ₂ (OH)HL+H)=5.98
K(2UO ₂ (OH)HL=(UO ₂) ₂ (OH) ₂ H ₂ L ₂)=3.48, K(2UO ₂ HL+2H ₂ O=(UO ₂) ₂ (OH) ₂ H ₂ L ₂ +2H)=-8.48					

UO ₂ ++	sp	NaClO ₄	25°C	0.20M	U
					1967BRa (89958)1437
					K(UO ₂ +HL)=9.84
					B((UO ₂) ₂ L)=19.03
					K(UO ₂ (OH)HL+H)=5.61
K((UO ₂) ₂ (H ₂ O) ₂ L=(UO ₂) ₂ (OH) ₂ L+2H)=-9.93					

UO ₂ ++	EMF	NaClO ₄	25°C	0.20M	U
					1967BRa (89959)1438
					K(UO ₂ (OH)HL+H)=5.44
K((UO ₂) ₂ (H ₂ O) ₂ L=(UO ₂) ₂ (OH) ₂ L+2H)=-10.55					*****

C14H2807		L	21-Crown-7		CAS 33089-36-0 (2264)
1,4,7,10,13,16,19-Heptaoxacycloheicosane;					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo

UO ₂ ++	sp	non-aq	25°C	100%	U
					K1=3.09
					1989LMB (90544)1439
Medium: 0.1 M Et ₄ NClO ₄ in propylene carbonate					*****

C14H30N204		L			CAS 31255-13-7 (2448)
N,N'-Dimethyl-cyclo-1,10-diaza-4,7,13,16-tetraoxaoctadecane;					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo

UO ₂ ++	sp	non-aq	25°C	100%	U
					K1=6.90 B2=14.28
Medium: propylene carbonate, 0.1 M Et ₄ NClO ₄					1989LMB (90592)1440

C14H30N205		L			CAS 23978-10-1 (2955)
1,10-Diaza-4,7,13,16,19-pentaoxacycloheicosane;					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo

UO ₂ ++	sp	non-aq	25°C	100%	U
					K1=6.79 B2=12.96
Medium: propylene carbonate, 0.1 M Et ₄ NClO ₄					1989LMB (90616)1441

C15H10N3OCl		HL			CAS 16195-35-0 (27)
5-(4-Chlorophenylazo)-8-hydroxyquinoline; Cl.C ₆ H ₄ .N:N.C ₉ H ₅ N.OH					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo

UO ₂ ++	gl	alc/w	27°C	40%	U
					K1=9.53 B2=28.99
*****					1984EIA (90950)1442

C15H1003		HL			CAS 577-85-5 (3443)
3-Hydroxyflavone;					

Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values Reference ExptNo

UO ₂ ++	sp	alc/w	20°C	80%	U
					K1=8.68
					1990MRa (90977)1443

C15H11N02 HL CAS 55022-23-6 (4061)
2-(6'-Methyl-2'-pyridyl)indan-1,3-dione;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 75% U K1=12.54 B2=24.12 1964CMB (91064)1444

C15H11N30 HL PAN CAS 85-85-8 (572)
1-(2-Pyridylazo)-2-naphthol; C5H4N.N:N.C10H6.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 31°C 0.10M U M K1=7.78 B2=15.31 1977SSb (91245)1445
B(UO2L(Malonate))=13.19
B(UO2L(Diglycolate))=12.78
B(UO2L(Glutarate))=11.38
B(UO2L(Maleate))=13.30

B((UO2)L(Glycolate))=12.01, B((UO2)L(Thiodiglycolate))=11.49

C15H11N30 HL CAS 4312-09-8 (989)
5-Phenylazo-8-hydroxyquinoline; C6H5.N:N.C9H5N.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 27°C 40% U K1=9.31 B2=18.38 1984EIA (91272)1446
Data also for 4-Cl-phenyl, 4-Br-, 4-MeO-, 4-Me2N- and 4-HSO3- analogues

C15H11N304S H2L CAS 574-70-9 (6238)
5-(4-Sulfophenylazo)-8-hydroxyquinoline,
4-((8-hydroxy-5-quinolinyl)azo)-benzenesulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 27°C 40% U K1=8.46 B2=24.38 1984EIA (91331)1447

C15H12N20 HL CAS 19726-12-6 (8336)
3-(2'-Hydroxyphenyl)-5-phenylpyrazole;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 35°C 60% U K1=8.38 B2=16.20 1993ALB (91433)1448

Medium: 60% v/v MeOH/H2O, 0.1 M KNO3. For 4-Cl-phenylpyrazole deriv.

K1=8.26, K2=7.55; for 1,5-diphenylpyrazole deriv. K1=9.60, K2=9.00.

C15H12N20 HL (3449)
4-Methyl-2-phenylquinazolin-8-ol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 20°C 50% U K1=8.53 B2=16.38 1954IRa (91437)1449

Medium: 50% dioxan, 0.3 M NaClO4

C15H12N3O4As H3L CAS 81315-66-2 (6237)

5-(2-Dihydroxyasenophenylazo)-8-hydroxyquinoline; (HO)2AsO.C6H4.N:N.C9H5N.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 27°C 40% U K1=7.64 B2=18.71 1984EIa (91447)1450

C15H12OS HL (1261)

mono-Thiodibenzoylmethane; C6H5.CO.CH2.CS.C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 75% U K1=10.34 B2=19.81 1966USA (91506)1451

C15H12O2 HL Diphenylacac CAS 120-46-7 (362)

1,3-Diphenylpropane-1,3-dione, Dibenzoylmethane; C6H5.CO.CH2.CO.C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 75% U K1=11.61 B2=23.14 1977AHb (91566)1452

U02++ dis oth/un 25°C 0.10M U B2=21.74 1970GRa (91567)1453

C15H12O3 H2L CAS 1469-94-9 (3445)

2-Hydroxydibenzoylmethane; HO.C6H4.CO.CH2.CO.C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 75% U K1=11.40 B2=22.43 1955HOa (91610)1454

C15H14N2O3 HL (6201)

2-Carboxy-2'-hydroxy-3',5'-dimethylazobenzene; HOOC.C6H4.N:N.C6H2(OH)(CH3)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 70% U I K1=15.68 B2=28.94 1987Kbc (91716)1455

C15H14N2O5S HL (9232)

3-(5-Sulphonylnaphthylazo)penta-2,4-dione;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KCl 25°C 0.1M U H K1=7.56 2004ACb (91737)1456

for 35 C K1=7.37; for 45 C K1=7.19

C15H14O3 HL CAS 84-79-7 (3446)

2-Hydroxy-3-(3-methylbut-2-enyl)-1,4-naphthoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
UO2++	gl	mixed	25°C	75%	U		K1=9.78	B2=18.75	1972Mcb	(92703)1463
Medium: 75% acetone, 0.1 M KNO ₃										

C16H11N20Cl		HL					CAS	24390-65-6	(5170)	
1-(2-Chlorophenylazo)-2-hydroxynaphthalene;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	mixed	25°C	75%	U		K1=9.29	B2=17.31	1972Mcb	(92718)1464
Medium: 75% acetone, 0.1 M KNO ₃										

C16H11N20Cl		HL					CAS	10149-93-6	(5171)	
1-(4-Chlorophenylazo)-2-hydroxynaphthalene;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	mixed	25°C	75%	U		K1=9.66	B2=18.50	1972Mcb	(92733)1465
Medium: 75% acetone, 0.1 M KNO ₃										

C16H11N20I		HL					CAS	25023-35-2	(5173)	
1-(4-Iodophenylazo)-2-hydroxynaphthalene;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	mixed	25°C	75%	U		K1=9.92	B2=19.03	1972Mcb	(92748)1466
Medium: 75% acetone, 0.1 M KNO ₃										

C16H11N208ClS2		H4L	Solochrome FN				CAS	25747-11-9	(8527)	
6-[(5-Chloro-2-hydroxy-3-sulfophenyl)azo]-5-hydroxy-1-naphthalenesulfonic acid;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	oth/un	20°C	0.10M	M T H		K1=11.2		1978MBe	(92780)1467
Medium: 0.10 M KCLO ₄ . Data for 44 C. DH and DS values reported.										

C16H11N209ClS2		H4L	Plasmocorinth				CAS	1058-92-0	(5203)	
3-(5-Chloro-2-hydroxyphenylazo)chromotropic acid (Eriochrome Blue SE)										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	oth/un	25°C	dil	U		B2=11.13		1968SMa	(92786)1468

C16H11N303		HL					CAS	6410-09-9	(5151)	
1-(2-Nitrophenylazo)-2-hydroxynaphthalene;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	mixed	25°C	75%	U		K1=5.37	B2=9.16	1972Mcb	(92802)1469

Medium: 75% acetone, 0.1 M KNO₃

C16H11N303 HL CAS 6410-46-1 (5152)
1-(4-Nitrophenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ gl mixed 25°C 75% U K1=6.23 B2=11.75 1972MCb (92817)1470

Medium: 75% acetone, 0.1 M KNO₃

C16H11N304 HL (2910)
1,3-Diphenyl-5-hydroxyimino-hexahydropyrimidine-2,4,6-trione;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ gl diox/w 30°C 75% C K1=6.18 B2=12.03 1978MGb (92837)1471

C16H12N20 HL CAS 842-07-9 (5156)
1-Phenylazo-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ gl mixed 25°C 75% U K1=10.64 B2=20.48 1972MCb (92923)1472

Medium: 75% acetone, 0.1 M KNO₃

C16H12N202 H2L CAS 9486-98-2 (3462)
1-(2-Hydroxyphenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ gl mixed 25°C 75% U 1972MCb (92959)1473
K(UO₂+HL)=10.57
K(UO₂HL+HL)=9.85

Medium: 75% acetone, 0.1 M KNO₃

C16H12N202 H2L CAS 14934-27-1 (5157)
1-(4-Hydroxyphenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO₂++ gl mixed 25°C 75% U 1972MCb (92975)1474
K(UO₂+HL)=10.42
K(UO₂HL+HL)=9.66

Medium: 75% acetone, 0.1 M KNO₃

C16H12N203 HL CAS 49747-16-2 (8340)
7-Hydroxy-4-methyl-8-(phenylazo)coumarin;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 25°C 60% U K1=7.08 B2=12.25 1992IOa (92980)1475
Medium: 60% v/v EtOH/H₂O, 0.1 M NaCl. Data for a range of aryl-substituted derivatives.

C16H12N2O4S H2L CAS 13964-82-4 (3475)

1-(4-Sulfophenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl mixed 25°C 75% U K1=6.28 B2=11.58 1972McB (93006)1476
Medium: 75% acetone, 0.1 M KNO₃

C16H12N3O4ClS H2L CAS 133131-00-7 (8468)

7-Amino-8-[(4-chlorophenyl)azo]-4-hydroxy-2-naphthalenesulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaCl 25°C 0.10M U K1=8.38 B2=15.39 1997IHa (93121)1477
B3=20.52

Also data for the 4'-bromo-, 4'-fluoro-, 4'-nitro-, 4'-methoxy-, 4'-di-methylamino-, 4'-hydroxy-, 4'-carboxy-, 4'-AsO(OH)₂-, 2'-hydroxy- analogue

C16H12O4 H2L CAS 1795-39-7 (4071)

3-Benzyl-4,5-dihydroxycoumarin

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp alc/w 21°C 40% U 1966JKa (93145)1478
K(?)=5.0

Medium: 40% EtOH, 0.4 M NaClO₄

C16H13N2O10AsS2 H5L Thorin I CAS 3688-92-4 (2609)

1-((2-Arsonophenyl)azo)-2-hydroxy-3,6-naphthalyl disulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp oth/un 25°C ? U 1966SAe (93216)1479
K(?)=4.3

U02++ gl oth/un 30°C ? U K1=15 1964PCa (93217)1480

C16H14N2O HL (1318)

2-(2-Hydroxynaphthyliminomethyl)pyridine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 50% A K1=8.23 1981RUa (93415)1481

C16H14N4O2 H2L (3467)

5-Hydroxy-4-(2-hydroxyphenylazo)-3-methyl-1-phenylpyrazole;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	alc/w	25°C	100%	U		K1=7.08	B2=11.49	1991EHa (93478)	1482
Medium: EtOH. Data also for other analogues										
C16H14N4O4S		HL					(5184)			
5-Methyl-1-phenyl-4-(2-sulfophenylazo)-3-pyrazolone;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	30°C	75%	U		K1=9.71		1969SSc (93508)	1483

C16H14O3		HL					CAS 41126-22-1	(3457)		
2-Methoxydibenzoylmethane; CH3.O.C6H4.CO.CH2.CO.C6H5										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	30°C	75%	U		K1=13.30	B2=24.36	1955HOa (93552)	1484

C16H14O6		H2L					CAS 20210-97-3	(8309)		
Ethylene disalicylate;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	alc/w	25°C	61%	C		K1=18.27		1991DSb (93594)	1485
Medium: 61.1% w/w EtOH/H2O, 0.50 M LiCl. K(H+L)=9.89, K(HL+H)=9.45.										
Data for the propyl and higher analogues.										
C16H15NO		HL					CAS 18594-93-9	(3468)		
3-Phenylimino-1-phenylbutan-1-one; C6H5.CO.CH2.C(:N.C6H5).CH3										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	30°C	50%	U		K1=11.32	B2=21.74	1961MJa (93603)	1486

C16H15N3O2S		H2L					(2105)			
S-Methyl-N1,N4-bis(salicylidene)isothiosemicarbazone; HO.C6H4.CH:N.N:C(SCH3).N:CH.C6H4.OH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	NaClO4	25°C	0.05M	U				1987CDa (93636)	1487
K(UO2+L=UO2L)=10.6										

C16H16N2O2		H2L					CAS 94-93-9	(2101)		
N,N'-Bis(salicylidene)ethylenediamine;(HO(C6H4)CH:NCH2-)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

U02++ dis oth/un 20°C 0.30M U K1=24.35 1966SZA (93685)1488

Medium: acetate

C16H16N204 H2L CAS 6345-72-8 (6729)

N,N'-Ethylenebis(salicylamide), N,N'-1,2-Ethanediylbis(2-hydroxybenzamide);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp alc/w 25°C 61% C K1=16.83 1991DSb (93704)1489

Medium: 61.1% w/w EtOH/H2O, 0.50 M LiCl. K(H+L)=9.28, K(HL+H)=8.48.

Data for the N,N'-1,3-propyl and higher analogues.

C16H18N203 HL (5564)

2-(2-Acetylphenylhydrazone)-5,5-dimethyl-1,3-cyclohexanedione;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 75% U K1=11.92 B2=22.87 1990ASb (93788)1490

C16H18N205S HL Penicillin V CAS 87-08-1 (943)

Phenoxyethylpenicillanic acid, 4-Thia-1-azabicyclo[3.2.0]heptane-2-carboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KNO3 25°C 0.10M M T H K1=7.50 B2=13.50 1983SBc (93820)1491

Also data for 35 C. DH(B2)=-5.86 kJ mol-1, DS(B2)=220 J K-1 mol-1.

C16H20N2010 H6L (704)

1,2-Dihydroxy-3,6-di-(methyleneiminodiethanoic acid)-benzene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl KNO3 25°C 0.10M C K1=19.28 1988ZHa (94068)1492

K(UO2+H2L)=11.43

K(UO2+HL)=16.31

K(UO2HL+H)=6.25

K(UO2L+H)=8.90

C16H24O14 H4L CAS 61696-54-6 (6104)

1,4,7,10,13,16-Hexaoxacyclooctadeca-2,3,11,12-tetracarboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp non-aq 25°C 100% U K1=5.61 1989LMb (94504)1493

Medium: 0.1 M Et4NClO4 in propylene carbonate

C16H34N206 L CAS 69930-74-1 (1321)

N,N'-Bis(2-hydroxyethyl)-1,7,10,16-tetraoxa-4,13-diazacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	mixed	25°C	75%	U			K1=11.28 B2=21.66	1972MCb (95834)	1500
Medium: 75% acetone, 0.1 M KNO3										
C17H14N2O2		HL					CAS	13441-91-1	(5217)	
1-(4-Methoxyphenylazo)-2-hydroxynaphthalene;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	mixed	25°C	75%	U			K1=10.82 B2=20.75	1972MCb (95849)	1501
Medium: 75% acetone, 0.1 M KNO3										
C17H14N2O2		L					CAS	4551-69-3	(698)	
4-Benzoyl-3-methyl-1-phenyl-2-pyrazolin-5-one;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	dis	oth/un	25°C	0.10M	U	I		B2=12.85	1973BKc (95905)	1502
I=1.0, B2=13.07										
C17H14N2O5S		H3L	Calmagite				CAS	3147-14-6	(2875)	
1-(1-Hydroxy-4-methyl-2-phenylazo)-2-naphthol-4-sulfonic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	NaClO4	25°C	0.10M	U			K1=16.87 B2=30.77	1973MPd (95931)	1503
C17H14O3		H2L					CAS	1467-40-9	(795)	
1,5-Diphenylpentane-1,3,5-trione; C6H5.CO.CH2.CO.CH2.CO.C6H5										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	alc/w	25°C	70 %	U				1991HKe (95978)	1504
B((UO2)HL)=8.95										
Medium: 70% v/v MeOH/H2O, 0.5 M NaClO4										
C17H15N03		HL					(6321)			
Benzoylacetoneanthranilic acid; C6H5.CO.CH2.C(CH3):N.C6H4.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	30°C	50%	U			K1=11.55	1975PNa (95986)	1505
C17H16O4		HL					CAS	18362-51-1	(3485)	
Di-2-methoxybenzoylmethane; CH3.O.C6H4.CO.CH2.CO.C6H4.O.CH3										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

U02++ gl diox/w 30°C 75% U K1=13.28 B2=24.80 1955H0a (96173)1506

C17H16O6 HL (4111)

2-Hydroxy-2',4',4-trimethoxydibenzoyl; HO.C6H4.CO.CO.C6H2(OCH3)3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaClO4 ? 0.10M U K1=8.43 B2=15.45 1963DSa (96184)1507

C17H18N2O2 H2L (6774)

1,3-Bis(salicylaldimino)propane; CH2(CH2.N:CH.C6H4.OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp non-aq 25°C 100% U 1990EGa (96202)1508
K(U02+2L=U02H-2L2+2H)=5.86

Medium: ethylacetate. For analogues with -(CH2)6- K=5.74; -(CH2)2NH(CH2)2-
K=6.18; -(CH2)2NH(CH2)2NH(CH2)2- K=7.40

C18H11N02 HL CAS 83-08-9 (4126)
2-(2'-Quinolyl)indan-1,3-dione;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 75% U K1=12.95 B2=25.02 1964CMB (96843)1509

C18H13N03 H2L (5238)

N-(2-Hydroxy-1-naphthalidene)anthranilic acid Schiff base;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 50% U K1=11.62 B2=19.15 1971MGa (96894)1510
Medium: 50% dioxan, 0.1 M NaClO4

U02++ gl diox/w 30°C 50% U K1=11.62 B2=18.15 1971MSH (96895)1511
Medium: 50% dioxan, 0.1 M NaClO4

C18H13N04 H3L CAS 698-51-6 (8424)
2-Hydroxy-4-[[2-hydroxy-1-naphthalenyl)methylene]amino]benzoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 27°C 40% M K1=7.87 B2=12.65 1993MRa (96898)1512
Medium: 40% v/v EtOH/H2O, 0.10 M NaCl.

C18H13N06 H3L CAS 216243-28-6 (8614)
5,7-Dihydroxy-6-[(2-carboxyphenyl)imino]methyl]-2-methyl-4H-1-benzopyran-4-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl alc/w 25°C 70% U TIH K1=5.54 B2=10.17 1998ISd (96901)1513
Medium: 70% v/v EtOH/H₂O, 0.106 M NaCl. Data for 60-100% EtOH/H₂O,
0.15-0.03 M NaCl and 0-55 C. At 25 C, I=0 M: K1=6.84, B2=13.07. DH and DS.

C18H13N5O3S4 HL CAS 683787-43-1 (9097)
4-[(4-Oxo-3-phenyl-2-thioxo-5-thiazolidinyl)azo]-N-2-thiazolyl-benzenesulfonamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	alc/w	25°C	30%	U	T H		K1=7.90 B2=12.60	2003EEa (96906)1514	
Medium: 30% v/v EtOH/H ₂ O, 0.10 M KCl. Data for 25-45 C. DH(K1)=44 kJ mol ⁻¹										
DS=297 J K ⁻¹ mol ⁻¹ . DH(K2)=55, DS=274. Protonation constants not reported.										

C18H14N2O2 HL CAS 15017-21-7 (6859)
2-Hydroxynaphthalidene benzoyl hydrazone; C₆H₅.CO.NH.N:CH.C₁₀H₆.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	20°C	75%	U	T		K1=8.76 B2=14.99	1992MCb (96910)1515	
30 C: B1=8.65, B2=14.78; 40 C: B1=8.52, B2=14.56										

C18H14N2O3 H2L CAS 54009-54-0 (6860)
2-Hydroxynaphthalidene salicylic hydrazone; H₂O.C₆H₄.CO.NH.N:CH.C₁₀H₆.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	20°C	75%	U	T		K1=7.99 B2=13.69	1992MCb (96920)1516	
30 C: B1=7.64, B2=13.41; 40 C: B1=7.48, B2=13.11										

C18H14N2O4 H2L (3499)
2-(2-Hydroxy-1-naphthylazo)phenoxyethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	30°C	75%	U			K1=15.01	1964PCa (96930)1517	

C18H14N2O11S2 H5L (4133)
2-(2'-(Carboxymethoxy)phenylazo)chromotropic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	sp	KNO ₃	25°C	0.10M	U				1969SHb (96955)1518	
K(UO ₂ +HL)=10.10										

C18H15N3O3S HL CAS 61625-17-0 (4139)
Di-4-tolylthiovioluric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	diox/w	30°C	25%	M T H			K1=5.09 B2= 9.46	1978MGe (97016)1519	

Medium: 25% dioxane/H₂O, 0.10 M NaClO₄. Data for 40, 45 and 50 C. DH(K1)=-35.8 kJ mol⁻¹, DS(K1)=-21.2 J K⁻¹ mol⁻¹; DH(K2)=-47.3, DS(K2)=-71.3.

C18H150P L CAS 791-28-6 (32)
Triphenylphosphine oxide; (C₆H₅)₃PO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	non-aq	25°C	100%	U	M			1976DBa	(97101)1520

K((UO₂A₂)₂+2L=2UO₂A₂L)=1.21

HA=tropolone. Medium: benzene

C18H16N2O3 HL (5560)
2-(2-Acetylphenylhydrazone)-1-phenyl-but-1,3-dione;
C₆H₅.CO.C(CO.CH₃):N.NH.C₆H₄.COCH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	25°C	75%	U			K1=12.72 B2=24.37	1990ASb	(97181)1521

C18H16N4O4 H2L (3500)
2-(4,5-Dihydro-3-methyl-5-oxo-1-phenyl-1H-pyrazol-4-ylazo)phenoxyethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	30°C	75%	U			K1=11.93	1962SCc	(97214)1522

C18H18O3 HL (5233)
Ethyl-2,4-diphenyl acetoacetate; C₆H₅.CH₂.CO.CH(C₆H₅).CO.O.CH₂.CH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	30°C	75%	C			K1=13.20	1973AAa	(97299)1523

C18H20N2O4 H2L (4131)
1,2-Bis(3'-methoxysalicylideneamino)ethane; (CH₃O.C₆H₃(OH).CH:N.CH₂.)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	dis	oth/un	20°C	0.30M	U			K1=19.6	1966SZa	(97337)1524

Medium: acetate

C18H22O4 H2L B(CH₂AcAcH)₂ (2252)
1,3-Di(hexa-3,5-dione)-benzene; C₆H₄((CH₂)₂.CO.CH₂.CO.CH₃)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	24°C	50%	U			K1=11.4	1979ACa	(97563)1525

C18H28O6 H2L O(EAcAcE)20 CAS 73199-63-0 (2251)

1,11-Dioxacycloicosane-5,7,15,17-tetraone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	24°C	50%	U			K1=12.5	1979ACa (97833)1526	

C18H28O10 H2L (OEOAcAcOE)2 CAS 62950-36-1 (2254)

1,4,10,13,16,22-Hexaoxacyclotetracos-6,8,18,20-tetraone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO2++	gl	diox/w	24°C	50%	U			K1=11.0	1979ACa (97871)1527	
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C18H30N4O12 H6L TTHA CAS 869-52-3 (694)

Triethylenetetraaminehexaethanoic acid;((HOOC.CH2)2N.CH2.CH2.N(CH2.COOH).CH2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO2++	gl	NaClO4	25°C	0.50M	U				1982NAc (98102)1528	
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K(UO2+H2L)=6.15

K(UO2L+H)=6.40

K(UO2HL+H=UO2H2L)=5.69

UO2++	EMF	KNO3	25°C	0.10M	C				19820La (98103)1529	
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K(2UO2+HL)=17.4

K(UO2+H2L)=7.6

K(UO2+H3L)=5.5

K(UO2+H4L)=4.5

K(2UO2+H2L)=11.8, K(2UO2+2OH+HL)=30.4

C18H36N2O6 L Cryptand 2,2,2 CAS 23978-09-8 (514)

1,10-Diaza-4,7,13,16,21,24-hexaoxabicyclo[8.8.8]hexacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO2++	sp	non-aq	25°C	100%	U			K1=7.70 B2=13.00	1985BFa (98763)1530	
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B((UO2)2L)=13.10

Medium: propylene carbonate

C19H12O8S H4L Pyrogallol red CAS 85531-30-2 (638)

Pyrogallolsulfonephthalein;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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UO2++	sp	none	25°C	0.0	U			K2=12.64	1979PKa (99001)1531	
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K(UO2L2+4A)=22.48

A=cetyl ammonium ion

C19H12O9Br2S H6L Bromo Pyrog.Red CAS 16574-43-9 (706)

5',5"-Dibromopyrogallolsulfonephthalein;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	mixed	25°C	75%	U			K1=10.44 B2=20.10	1972MCb (99619)	1545
Medium: 75% acetone, 0.1 M KNO ₃										

C ₂₀ H ₁₄ N ₂₀ S ₅		H3L	Solochrome 6B	CAS 3564-14-5	(3507)					
1-(1-Hydroxy-2-naphthylazo)-2-naphthol-4-sulfonic acid, Mordant Black3, Eriochrome blue-black B;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	oth/un	20°C	0.10M	M T H			K1=9.7	1978MBe (99666)	1546
Medium: 0.10 M KClO ₄ . Data for 44 C. DH and DS values reported.										
UO ₂ ++	gl	NaClO ₄	25°C	0.10M	U			K1=15.50 B2=27.56	1973MPd (99667)	1547

C ₂₀ H ₁₄ N ₂₀ S ₃		H2L	Hydroxynaphthol	CAS 63451-35-4	(2835)					
Hydroxynaphthol blue, 1-(2-Hydroxy-4-sulfo-1-naphthylazo)-2-naphthol-3,										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	none	25°C	0.0	U				1978BRb (99738)	1548
K _{eff} =4.10										
Keff at pH 10										

C ₂₀ H ₁₅ N ₀ 3		H2L			(2120)					
2-(alpha-Phenyl-2-hydroxybenzylideneimino)benzoic acid; HO.C ₆ H ₄ .C(C ₆ H ₅):N.C ₆ H ₄ .COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	NaClO ₄	25°C	0.10M	U	TIH		K1=10.85 B2=19.88	1986SGb (99749)	1549
35 C: K ₁ =11.13, K ₂ =9.35; 45 C:K ₁ =11.55, K ₂ = 9.60										
DH(K ₁)=-95.3 kJ mol ⁻¹ , DS=105 J K ⁻¹ mol ⁻¹										

C ₂₀ H ₁₅ N ₃ 04		HL			(4147)					
8-Hydroxy-7-(3-nitroanilinofurfuryl)-quinoline;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sol	oth/un	25°C	?	U				1961TZA (99753)	1550
K _s (UO ₂ L ₂ HL+2H=UO ₂ +3HL)=-30.81										
Acetate buffer										

C ₂₀ H ₁₆ N ₂ 02		H2L			CAS 3946-91-6	(2733)				
N,N'-Bis(2'-hydroxybenzylidene)-1,2-diaminobenzene; (HO ₂ C ₆ H ₄ CH:N) ₂ .C ₆ H ₄										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	non-aq	25°C	100%	C				2000MRa (99776)	1551

$$K(U02L+A)=0.89$$

$$K(U02L+B)=1.23$$

Medium: CHCl₃. A: 2-cyclohexen-1-one; B: 4,4-dimethyl-2-cyclohexen-1-one.

For the 3'-phenyl-2'-hydroxybenzylidene derivative of L: K(U02L+A)=2.95.

U02++ sp alc/w 20°C 100% U K1=6.08 1984EAa (99777)1552

U02++ dis oth/un 20°C 0.30M U K1=20.9 1966Sza (99778)1553

Medium: acetate

C20H16N2O2 H2L (2730)

N,N'-Bis(salicylidene)-1,4-phenylenediamine; (HO.C6H4.CH:N)2C6H4

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp alc/w 20°C 100% U H K1=3.98 B2=7.9 1983EAb (99785)1554

C20H17NO HL (6215)

N-(2-Hydroxy-5-phenylbenzylidene)-2-methylaniline; C6H5.C6H3(OH).CH:N.C6H4.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 75% U K1=7.964 B2=15.34 1986MBd (99811)1555

C20H18N4O2 HL (5917)

Pyruvic monohydrazone-3-hydrazino-4-benzyl-6-phenylpyridazine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 75% U B2=18.52 1985RSb (99843)1556
K(U02+HL)=5.54
K(U02+2HL)=10.58
K(U02+L+HL)=15.50

C20H19N3O3S HL CAS 380496-11-7 (9099)

1,3-Di(2-ethylphenyl)-4,5,6-pyrimidinetrione-2-thioxo-5-oxime;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 75% U T H K1=5.43 B2= 9.76 2001SSd (99866)1557
Medium: 75% v/v dioxan/H₂O, 0.10 NaClO₄. Data for 30 and 35 C.
DH(B2)=-0.21 kJ mol⁻¹.

C20H19N3O3S HL CAS 380496-12-8 (9100)

1,3-Di(3-ethylphenyl)-4,5,6-pyrimidinetrione-2-thio-5-oxime;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 75% U T H K1=5.89 B2=11.03 2001SSd (99876)1558
Medium: 75% v/v dioxan/H₂O, 0.10 NaClO₄. Data for 30 and 35 C.

DH(B2)=-0.42 kJ mol-1.

C20H19N3O3S HL CAS 380496-13-9 (9101)

1,3-Di(4-ethylphenyl)-4,5,6-pyrimidinetrione-2-thio-5-oxime;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 75% U T H K1=5.50 B2=11.11 2001SSd (99885)1559

Medium: 75% v/v dioxan/H₂O, 0.10 NaClO₄. Data for 30 and 35 C.

DH(B2)=-0.13 kJ mol-1.

C20H24O6 L DiBz-18-Crown-6 CAS 14187-32-7 (604)

2,3:11,12-Dibenzo-1,4,7,10,13,16-hexaoxacyclooctadeca-2,11-diene

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp non-aq 25°C 100% U I K1=5.51 1989LMb (100256)1560

Medium: 0.1 M Et₄NClO₄ in propylene carbonate

In acetonitrile, K1=6.00

U02++ sp non-aq 25°C 100% U I K1=5.51 1985BFa (100257)1561

Medium: propylene carbonate

U02++ ISE non-aq 25°C 100% C K1=5.50 1984FLa (100258)1562

In propylenecarbonate; electrolyte Et₄NClO₄

C20H36O6 L DiCy-18-crown-6 CAS 16069-36-6 (1653)

2,3:11,12-Dicyclohexyl-1,4,7,10,13,16-hexaoxacyclooctadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ nmr non-aq 27°C 100% C I K1=2.93 2001KZa (100721)1563

Method: 7Li nmr; competitive binding study. Medium: nitromethane.

In acetonitrile, K1=2.52

U02++ ISE non-aq 25°C 100% C K1=5.63 1984FLa (100722)1564

In propylenecarbonate; electrolyte Et₄NClO₄

C21H17NO HL CAS 20964-94-7 (3512)

1-(Phenylimino)-1,3-diphenylpropan-3-one; C₆H₅.N:C(C₆H₅).CH₂.CO.C₆H₅

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 50% U K1=11.59 B2=22.08 1961MJa (101074)1565

C21H18N6O3S3 HL CAS 364325-74-6 (9094)

N-(4,6-Dimethyl-2-pyrimidinyl)-4-[(4-oxo-3-phenyl-2-thioxo-5-thiazolidinyl)azo]-benzenesulfonamid

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp oth/un 25°C 0.01M U K1=4.77 1958MDa (101514)1573

C22H17N3O3 HL CAS 53855-37-1 (4154)
8-Hydroxy-7-(3'-nitroanilinobenzyl)-quinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sol oth/un 25°C ? U 1961TZA (101571)1574
Ks(U02L2HL+2H=U02+3HL)=-30.04

Acetate buffer

C22H17N4O14ClP2S2 H8L ClPhosphonazo 3 CAS 1914-99-4 (2577)
2,7-Bis((4-chloro-2-phosphophenyl)azo)chromotropic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp KN03 25°C 0.20M U 1967BMc (101583)1575
B((U02)H12L2)=47.7

C22H17N4O14ClP2S2 H8L CAS 86253-02-3 (4159)
2-(4'-Chloro-2'-phosphonophenylazo)-7-(2''-phosphonophenylazo)chromotropic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp KN03 25°C 0.20M U 1967BMc (101586)1576
B((U02)H10L2)=103.0

C22H18N4O14As2S2 H8L Arsenazo III CAS 1668-00-4 (1148)
2,7-Bis(2'-arsonophenylazo)chromotropic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp oth/un RT 0.03M U 1997RRc (101655)1577
K1eff=7.37

Medium: HCl, pH 1.5.

U02++ sp oth/un 25°C ? C K1=5.41 B2=11.0 1987SLa (101656)1578

U02++ sp oth/un ? 7.0M U 1970KSc (101657)1579
K(U02(NO3)2+H8L)=4.53

Medium: 2-12 M HNO3

U02++ vlt KCl ? 0.60M U 1967TBa (101658)1580
K(U02+H6L)=4.22
K(U02+2H6L)=8.11

C22H18N4O14P2S2 H8L Phosphonazo III CAS 16017-11-1 (4158)
2,7-Bis(2'-phosphonophenylazo)chromotropic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp KN03 25°C 0.20M U 1967BMc (101670)1581
B((U02)H10L2)=106.8

C22H18O2 HL (364)
4-Phenylbenzoyl(phenylpropionoloyl)methane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 30°C 75% U K1=9.62 B2=18.72 1977AHb (101675)1582

C22H24N208 H2L Tetracycline CAS 60-54-8 (2201)
Tetracycline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaNO3 25°C 0.10M C K1=4.6 1992GAa (101829)1583

U02++ vlt NaNO3 25°C 0.10M C K1=4.04 1992GAb (101830)1584
Method: polaography.

C22H24N209 H2L Oxotetracycline CAS 79-57-2 (2202)
Oxytetracycline, 5-Hydroxy-tetracycline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl NaNO3 25°C 0.10M C K1=4.97 1992GAa (101888)1585

C23H16O9Cl2S H4L Chrome azurol S CAS 1667-99-8 (711)
Chromazurol S;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ sp oth/un ? 0.10M U 1970CSb (102576)1586

$$\begin{aligned}K(U02+H2L=U02(HL)+H) &= 0.45 \\K(U02+H3L=U02(HL)+2H) &= -2.6 \\K(U02+HL) &= 5.35 \\K(2U02+H2L=(U02)2L+2H) &= 1.6\end{aligned}$$

$$K(2U02+L)=18.3$$

U02++ sp KCl 30°C 0.2M U K1=4.7 1960SDa (102577)1587

C23H18N203 HL (5561)
2-(2-Acetylphenylhydrazone)-1,3-diphenyl-prop-1,3-dione;
C6H5.CO.C(CO.C6H5):N.NH.C6H4.COCH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

U02++ gl diox/w 25°C 75% U K1=12.28 B2=23.51 1990ASb (102603)1588

C23H18N203 H2L (4160)
7-(4'-Carboxyphenylaminobenzyl)-8-hydroxyquinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ sol oth/un 25°C ? U 1961TZA (102609)1589

Acetate buffer. $K_s(UO_2(HL)2H_2L+2H=2UO_2+3H_2L)=-22.98$

C23H18O3 L CAS 29549-01-7 (5321)

Ethyl alpha-(alpha-naphthyl)phenylpropiolethanoate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl diox/w 30°C 75% U K1=11.64 B2=21.39 1973AA (102617)1590

C23H27N07 HL CAS 203302-24-3 (8395)

4'-(omega-Salicylaldiminoacetyl)benzo-15-crown-5;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ gl KNO3 25°C 0.10M M K1=8.15 1998ADb (102712)1591

B(UO2H-1L)=3.36

B(UO2H-2L)=-2.17

B(UO2H-3L)=-10.35

C24H20N4014C12P2S2 H8L (4165)

2,7-Bis(4'-chloro-5'-methyl-2'-phosphonophenylazo)chromotropic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ sp KNO3 25°C 0.20M U 1967BMC (102917)1592

B((UO2)H12L2)=108.7

C24H32O8 L DiBz-24-Crown-8 CAS 14174-09-5 (580)

2,3:14,15-Dibenzo-1,4,7,10,13,16,19,22-octaoxacyclotetracosa-2,14-diene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ sp non-aq 25°C 100% U I K1=3.63 1989LMb (103182)1593

Medium: 0.1 M Et4NClO4 in propylene carbonate

In acetonitrile, K1=5.16

C24H51N L CAS 1116-76-3 (4161)

Trioctylamine; (CH3.(CH2)7)3.N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

UO2++ dis KNO3 ? 2.50M U 1960SSa (103532)1594

Medium: HNO3. $K(UO_2+2NO_3+LHN_03=UO_2HL(NO_3)_2)=0.31(\text{org}=CCl_4), 0.46(2\text{-xylene})$

C24H51O_P L CAS 78-50-2 (4162)
Trioctylphosphine oxide; (C₈H₁₇)₃P:O

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	non-aq	20°C	100%	U				1983KBC	(103544)1595

K(UO₂Cl₂+L)=2.56
K(UO₂Cl₂+2L)=5.32

Medium: acetone. Data also for other phosphonic acid esters

C26H22N4O HL (1410)
1-Phenyl-1-propanone-3-(4-benzyl-6-phenyl)-pyridazinyl hydrazone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	30°C	75%	U			K1=11.81	1983RRa	(103867)1596

C26H23N5O₂ HL (5918)
Hippuric monohydrazone-3-hydrazino-4-benzyl-6-phenylpyridazine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	gl	diox/w	30°C	75%	U			K1=14.42 B2=25.80	1985RSb	(103890)1597

C27H30O₁₆ H4L Rutin CAS 153-18-4 (4169)
3,3',4',5,7-Pentahydroxyflavone-3-beta-rutinoside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	KNO ₃	20°C	0.50M	U				1963DJa	(104510)1598

K(?)=9.35

C27H54N2O₂ L THMA CAS 170126-54-2 (7624)
N,N,N',N'-Tetrahexylmalonamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	dis	non-aq	25°C	100%	U	I		B2=0.59	1999LMA	(104642)1599

B3=1.38

Media: t-butylbenzene and 1 M NaNO₃. Also data for 2, 3, 4, 5 M NaNO₃.

Bn: UO₂(aq)+2NO₃(aq)+nL(org)=UO₂(NO₃)₂Ln(org)

C28H24N2O₂ H2L Solvent Green 3 CAS 128-80-3 (1021)
1,4-Bis(4'-methylanilino)anthraquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO ₂ ++	sp	mixed	25°C	40%	U			K1=8.45 B2=10.29	1985ISb	(104668)1600

In 40% DMF/H₂O, 0.1 M NaClO₄.

C28H30N207 L CAS 105169-83-3 (7173)
4, '5-Bis(salicylideneimino)-1,4,7,10,13-pentaoxa[13]orthocyclophan;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	KCl	25°C	1.00M	C			B2=11.60 B(UO2H-1L=UO2(OH)L)=0.80 B(UO2H-2L2=UO2(OH)2L2)=-1.35	1995ABB (104733)	1601

C28H40010 L DiBz-30-crown10 CAS 104946-67-0 (1776)
2,3:17,18-Dibenzo-1,4,7,10,13,16,19,22,25,28-decaoxacyclotriaconta-2,17-diene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	sp	non-aq	25°C	100%	U			K1=2.95	1989LMb (104921)	1602

Medium: propylene carbonate, 0.1 M Et4NClO4

C28H56N202 L CAS 252344-64-2 (7625)
N,N,N',N'-Tetrahexyl-2-methylmalonamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	dis	non-aq	25°C	100%	U	I		B2=0.53 B3=1.69	1999LMA (105028)	1603

Media: t-butylbenzene and 1 M NaNO3. Also data for 2, 3, 4, 5 M NaNO3.
Bn: UO2(aq)+2NO3(aq)+nL(org)=UO2(NO3)2Ln(org)

C29H58N202 L CAS 252344-66-4 (7626)
N,N,N',N'-Tetrahexyl-2,2-dimethylmalonamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	dis	non-aq	25°C	100%	U	I		B2=-1.10 B3=-2.14	1999LMA (105168)	1604

Media: t-butylbenzene and 1 M NaNO3. Also data for 2, 3, 4, 5 M NaNO3.
Bn: UO2(aq)+2NO3(aq)+nL(org)=UO2(NO3)2Ln(org)

C31H24N40 HL CAS 88700-85-0 (1409)
1,2-Diphenyl-1,2-ethanedione-3-(4-benzyl-6-phenyl)-pyridazinyl hydrazone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
UO2++	gl	diox/w	30°C	75%	U	I		K1=11.80 In 75% DMF: K1=8.90, B2=16.67 B2=23.09	1983RRa (105412)	1605

C31H32N2013S H6L Xylenol orange CAS 63721-85-5 (432)
5,5'-Bis-N,N-bis(carboxymethyl)aminomethyl-4'-hydroxy-3,3'-dimethylfuchsone-2"-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ gl KCl 25°C 0.10M U 1982NAd (105504)1606
 $K(UO_2+H_2L)=8.22$
 $K(UO_2+H_2L+OH)=15.48$
 $K(UO_2H_2L+OH)=7.26$
 $K(UO_2+H_2L+2OH)=21.78$
 $K(UO_2+HL+2OH)=23.3; K(UO_2+L+2OH)=24.19; K(UO_2.H_2L(OH)_2=UO_2HL(OH)_2+H)=-8.9$

U02++ sp none 25°C 0.0 U 1974BUb (105505)1607
 $B((UO_2)H_2L)=29.80$
 $B((UO_2)H_4L_2)=56.60$

U02++ sp oth/un 25?°C ? U 19630Ta (105506)1608
 $K(?)=11.46$

U02++ sp NaNO3 20?°C 0.20M U 1962BUa (105507)1609
 $B((UO_2)2L_2)=38.57$

C32H66N204 L 22DD Kryptofix CAS 79495-97-9 (6655)
1,10-Didecyl-1,10-diaza-4,7,13,16-tetraoxacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	sp	non-aq	25°C	100%	U			$K_1=3.88$ $B_2=7.74$	1989LMb (105866)1610	
Medium: propylene carbonate, 0.1 M Et4NClO4										

C37H44N2013S H6L MeThymol Blue (428)
3,3'-Bis(N,N-di(carboxymethyl)aminomethyl)thymolsulfonephthalein;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	sp	NaClO4	?	0.50M	U				1973CPb (106624)1611	

C54H62N8014S4 H2L CAS 187828-35-9 (8875)
Bis[(4,10-Diaza-4,10-ditosyl-benzo-12-crown-4)4'-yl]diaminoglyoxime;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
U02++	gl	mixed	25°C	70%	U				1996ADc (107539)1612	

Medium: 70% v/v acetone/H2O, 1.0 M NaNO3. $K(UO_2+HL=UO_2H-2L+3H)=-3.98$,
 $K(UO_2+HL=UO_2H-3L+4H)=-15.71$.

C66H408 H4L CAS 173173-83-6 (9060)
[C60]fullerene dimalonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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U02++ sp KN03 25°C 0.10M U 1998DMb (108184)1619
K1eff=5 2

Method: fluorescence quenching. Medium: pH 3.5.
Fulvic acid extracted from sewage sludge.

UO₂++ sp KNO₃ 25°C 0.10M U I 1996SMb (108185)1620

K1eff=3.93

UO_2^{++} 0.1M KNO_3 25°C 0.01M H_2O_2 1980Lya (108186)1621

K1eff=7.43
K2eff=5.56

Method: dialysis at pH 6

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

H2O2
N-ClO4-22886-0-1CM-C-T-H
22888E1-(100045)-1532

002++ Tx NaCl0.4 20°C 0.10M C + H K1a-SS 8-26 2000B3a

Aldrich humic acid. $K_{1\text{eff}}$ at pH 4.5. Also data for

Aldrich humic acid. Riem at pH 4.5. Also data for $\text{DH}(\text{K1eff}) = -33 \text{ kJ mol}^{-1}$, $\text{DS} = 63 \text{ J K}^{-1} \text{ mol}^{-1}$.

BR(RICHT)- 23 RS M01 1, DS-02 3 R 1 M01 1.

UO_2^{++} TSE NaCl 104 25°C C

Mathematics Education in Early Childhood

Method: uranyl ion selective electrode. Humic acid extracted from brown coal. Conditions: pH 5.0, NaCl [M] 11.7

U0311 - disc NaCl04-2586-2-10M_H K1_E_11_B2_S_24_10815cb_108247_1624

UO2-Li_xctb-KNO₃, 25°C, 0.01M, II, 1880 LV₂ (108248) 1625

K1eff=6.73
K2eff=4.72

Method: dialysis at pH 6

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EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES
I Data with various BACKGROUNDS
H Data for THERMOCHEMICAL quantities
M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC
R or IUP=R signifies EVALUATION RATING = Recommended by IUPAC

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