

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 889 experiments for

(no ligands specified)

5 metals : V(IV), V(V), V++, V++, V0++

(no references specified)

(no experimental details specified)

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	ISE	KNO ₃	37°C	0.32M	U	M			1985TMb (4439)	1
								K(VA ₂ Cl+Cl=VA ₂ Cl ₂)=2.57		

A=Cyclopentadiene

MoO₄-- H₂L Molybdate (443)
Molybdate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	sp	oth/un	25°C	2.0M	C				1994MFb (8709)	2
								K(VO ₄ +1/2Mo ₂ O ₇ =VOH-1MoO ₂ +H)=1.5		

Medium: HNO₃.

OH- HL Hydroxide (57)
Hydroxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	gl	KNO ₃	37°C	0.32M	U				1985TMb (10880)	3
								*K(VA ₂ (H ₂ O) ₂)=-4.73		
								*K(VA ₂ (H ₂ O)OH)=-5.15		

A=cyclopentadiene

V04--- H₃L CAS 15457-75-7 (1586)
Vanadate; VO₂(OH)₃-- or polymers

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	sp	NaCl		0.10M	U				19640Sa (17376)	4
								K=-8.4		

K: 2HV04+V409+H₂O=HV6015+3OH

V(IV)	sp	NaCl	45°C	1.0M	U	T			19640Sa (17377)	5
								K=71		

K: 0.7H₂V10028+3VO₄+7.2OH=HV10026+3.8H₂O. At rt: K(0.3H₂V10028+7VO₄+16.8H₂O=HV10024+8.2H₂O)=143. Many other complex equilibria given

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)
Ethanedioic acid; (COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	sp	NaClO4	20°C	0.1M	U				1997KVb (18773)	6
								K(VO+L)=7.11		
								K(VO+2L)=10.45		

C2H4O2 HL Acetic acid CAS 64-19-7 (36)
Ethanoic acid; CH3.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	nmr	KCl	RT	1.0M	C				1990TLa (19879)	7
								K(VO4H2+2HL=V02L2+2H2O)=1.01		

Method: 51V nmr spectroscopy.

V(IV)	nmr	KCl	RT	1.0M	C				1990TLa (19880)	8
								Keff(VO4H2+2HL=V02L2)=0.20		

Method: 51V nmr spectroscopy. pH 5.0.

C3H4O4 H2L Malonic acid CAS 141-82-2 (79)
Propanedioic acid; CH2(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	sp	NaClO4	20°C	0.1M	U				1997KVb (24381)	9
								K(VO+OH+L)=18.71		
								K(VO+2OH+L)=28.52		

V(IV)	nmr	KCl	RT	1.0M	C				1990TLa (24382)	10
								Keff(VO4H2+H2L=V02L+2H2O)=0.70		

Method: 51V nmr spectroscopy. pH 5.0.

C3H7N02S H2L Cysteine CAS 52-90-4 (96)
2-Amino-3-mercaptopropanoic acid; H2N.CH(CH2.SH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	gl	KNO3	25°C	0.15M	U T		K1=5.78	B2= 8.91	1985KRa (26750)	11
At 35 C,	K1=6.01,	K2=3.83.	At 45 C,	K1=6.27,	K2=5.06.					

C4H5O4Cl H2L CAS 16045-92-4 (2232)
Chlorosuccinic acid; HOOC.CH(Cl).CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	nmr	KCl	RT	1.0M	C				1990TLa (29434)	12

$$K_{eff}(V04H2+2HL=V02L2) = <-0.7$$

$$K_{eff}(V04H2+H2L=V02L+2H2O) = 1.22$$

Method: 51V nmr spectroscopy. pH 5.0.

C4H6O4 H2L Succinic acid CAS 110-15-6 (112)
1,4-Butanedioic acid; HOOC.CH2.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	sp	NaClO4	20°C	0.1M	U				1997KVb (29926)	13
V(IV)	nmr	KCl	RT	1.0M	C				1990TLa (29927)	14

$$K_{eff}(V04H2+2HL=V02L2) = 0.76$$

$$K_{eff}(V04H2+H2L=V02L+2H2O) = 0.86$$

Method: 51V nmr spectroscopy. pH 5.0.

C4H9N04 HL CAS 17149-11-0 (8049)
(1-Hydroxymethyl)serine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	gl	KNO3	25°C	0.10M	M			B2=12.57 B(VOH-1L2)=4.48 B(VOH-2L2)=-4.39 B(VOH-3L2)=-15.09	1995KKb (34399)	15

C5H8O2 HL Acetylacetone CAS 123-54-6 (164)
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	sp	non-aq	25°C	100%	C				1983MGa (37909)	16

$$K(VOL2+py)=1.83$$

$$K(VOL2+2-Me-py)=1.03$$

$$K(VOL2+4-Me-py)=2.11$$

$$K(VOL2+2,4-di-Me-py)=1.28$$

Medium: dichloromethane. Data for other di-methylpyridine analogues.

C5H8O4 H2L CAS 498-21-5 (2234)
Methylsuccinic acid; HOOC.CH2.CH(CH3).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	nmr	KCl	RT	1.0M	C				1990TLa (38256)	17

$$K_{eff}(V04H2+2HL=V02L2) = 0.96$$

$$K_{eff}(V04H2+H2L=V02L+2H2O) = 0.15$$

Method: 51V nmr spectroscopy. pH 5.0.

C5H8O4 H2L Glutaric acid CAS 110-94-1 (420)
Pantanedioic acid; HOOC.CH2.CH2.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	nmr	KCl	RT	1.0M	C				1990TLa (38308)	18
$K_{eff}(V_04H_2+2HL=V_02L_2)=0.95$										
$K_{eff}(V_04H_2+H_2L=V_02L+2H_2O)=0.26$										
Method: 51V nmr spectroscopy. pH 5.0.										

C5H1002		HL		n-Valeric acid		CAS	109-52-4	(3027)		
Pentanoic acid; CH ₃ (CH ₂) ₃ .COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	nmr	KCl	RT	1.0M	C				1990TLa (40200)	19
$K_{eff}(V_04H_2+2HL=V_02L_2)=0.59$										
Method: 51V nmr spectroscopy. pH 5.0.										

C6H9N3O2		HL		Histidine		CAS	71-00-1	(1)		
2-Amino-3-(4'-imidazolyl)propanoic acid; H ₂ N.CH(CH ₂ .C ₃ H ₃ N ₂).COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	gl	KNO ₃	25°C	0.15M	U T		K1=9.79		1985KRa (47528)	20
At 35 C, K1=9.28. At 45 C, K1=9.04.										

C6H1004		H ₂ L		Adipic acid		CAS	124-04-9	(401)		
1,6-Hexanedioic acid; HOOC.(CH ₂) ₄ .COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	nmr	KCl	RT	1.0M	C				1990TLa (48063)	21
$K_{eff}(V_04H_2+2HL=V_02L_2)=1.04$										
$K_{eff}(V_04H_2+H_2L=V_02L+2H_2O)=0.28$										
Method: 51V nmr spectroscopy. pH 5.0.										

C6H11N05		H ₂ L		HIMDA		CAS	93-62-9	(192)		
N-(2-Hydroxyethyl)iminodiethanoic acid; HO.CH ₂ .CH ₂ .N(CH ₂ .COOH) ₂										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	sp	NaClO ₄	20°C	0.1M	U				1997KVb (48686)	22
$K(V_0+L)=9.87$										
$K(V_0+OH+L)=18.23$										

C8H15N08		HL				CAS	5616-22-8	(6474)		
N-(2,3,4,5,6-Pentahydroxyhexanoyl)glycine, N-D-Gluconylglycine;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	gl	KCl	25°C	0.20M	C		K1=1.75		2001GJa (62229)	23

$B((V0)2H-2L2)=-1.37$
 $B((V02)2H-3L2)=-5.57$
 $B((V02)2H-4L2)=-10.17$
 $B((V02)2H-5L2)=-16.67$
 $B((V02)2H-6L2)=-25.01$, $B((V0)H-4L)=-24.80$, $K(V0+HL=VOL+H)=-1.57$.

C9H11N02 HL Phenylalanine CAS 63-91-2 (2)
2-Amino-3-phenylpropanoic acid; H2N.CH(CH2.C6H5)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	gl	KNO3	25°C	0.15M	U	T		K1=7.58	1985KRa	(65922) 24
At 35 C, K1=7.24. At 45 C, K1=6.73.										

C9H12N206 HL Uridine CAS 58-96-8 (828)
Uracil-1-beta-D-ribofuranoside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	gl	NaCl	25°C	0.60M	U				1998ECa	(66691) 25
$B(0,2,2,0)=7.66$ $B(-1,2,2,0)=-1.09$ $B(-1,1,1,0)=-7.43$ $B(0,1,1,1)=3.12$										

$B(p,q,r,s)$: $pH+qH2V04+rHL+sHA=Hp(H2V04)q(HL)r(HA)s$.
 $B(-1,1,1,1)=-6.26$. A is imidazole.

C9H17N08 HL CAS 94231-90-0 (7909)
N-(2,3,4,5,6-Pentahydroxyhexanoyl)-beta-alanine, N-D-gluconyl-beta-alanine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	gl	KCl	25°C	0.20M	C			K1=1.91	2001GJa	(67845) 26
$B((V0)2H-2L2)=-1.19$ $B((V02)2H-3L2)=-5.48$ $B((V02)2H-4L2)=-10.17$ $B((V02)2H-5L2)=-16.52$										

$B((V02)2H-6L2)=-24.43$, $B((V0)H-4L)=-23.98$, $K(V0+HL=VOL+H)=-2.23$.

C9H17N09 HL CAS 168107-24-2 (7910)
N-(2,3,4,5,6-Pentahydroxyhexanoyl)serine, N-D-gluconyl-L-serine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	gl	KCl	25°C	0.20M	C			K1=2.19	2001GJa	(67849) 27
$B((V0)2H-2L2)=-1.37$ $B((V02)2H-3L2)=-5.08$ $B((V02)2H-4L2)=-9.21$ $B((V02)2H-5L2)=-15.11$										

$B((V02)2H-6L2)=-22.84$, $B((V0)H-4L)=-23.74$, $K(V0+HL=VOL+H)=-0.91$

C15H15N03 HL (6240)
N-4-Tolyl-4'-methoxybenzohydroxamic acid; CH3O.C6H4.CO.N(C6H4.CH3).OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
V(IV)	dis	NaClO4	30°C	0.20M	C			K1=5.11	B2= 9.79	1980GKb (91862)	28
								B3=9.79			

Method: distribution from 0.2 M NaClO₄ into chloroform.

C16H13N2O11AsS2 H6L Arsenazo I CAS 520-10-5 (277)
2-(2'-Arsonophenylazo)chromotropic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	sp	oth/un	?	?	U				1967LPa (93247)	29
								$K(?)=12.7$		

C19H15N08 H4L Alizarin Comp. CAS 3952-78-1 (671)
(3,4-Dihydroxy-2-anthraquinonyl-methyl)iminodiethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(IV)	sp	oth/un	RT	dil	C				1982EDa (99128)	30
								K1eff=4.7		
								B2eff=8.6		

Medium: borax buffer, pH 10.

e- HL Electron (442)
Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	EMF	NaCl	25°C	0.60M	C			1985Pha (1020) 31 K(e+V02+2H=V0++)=17.37 E=1.028 V		

V(V) cal none 25°C 0.0 M H 1966BDb (1021) 32
 $\Delta H(\text{Fe(II)} + \text{VO}_2^+ + 2\text{H} = \text{Fe(III)} + \text{VO}^{++} + \text{H}_2\text{O}) = 82.346 \text{ kJ mol}^{-1}$

V(V) EMF none 25°C 0.0 U 1940HPa (1022) 33
 K=16.90(999.6 mV)

K: VO₂+2H+e=VO+H₂O

AsO₄--- H3L Arsenate CAS 7778-39-4 (1557)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
V(V)	¹ nmr	KCl	25°C	1.00M	II	I			1986GTa	(1165)	34

K1eff=1.11

At pH 6.47: K1eff: H₂V₀4+H₂L=H₂V₀3L+H₂O

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) sp NaClO₄ 25°C 4.00M U I 1985IKa (5937) 35
K(VO₂ + L)=-0.483

V(V) sp NaClO₄ 19°C 1.01M U 1966IVc (5938) 36
K(VO₂+L)=-0.38

F- HL Fluoride CAS 7644-39-3 (201)
Fluoride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) ISE NaNO₃ 20°C 4.00M U K1=5.34 B2=10.34 1985VRb (7320) 37
B3=15.04
V(V) ix oth/un ? ? U 1972PAb (7321) 38
K(VOF₃+F)=3.68
V(V) sp NaClO₄ 20°C 1.0M U K1=3.04 B2=5.59 1969IVa (7322) 39
B3=6.86

Metal ion: VO₂₊

V(V) sp NaClO₄ 20°C 1.0M U 1967IVa (7323) 40
K(VO₂+L)=3.67
K(VO₂+2L)=6.32

H₂PO₂- HL Hypophosphite CAS 6303-21-5 (6304)
Hypophosphite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) sp NaClO₄ 25°C 1.00M U K1=1.22 1971CHc (7652) 41
Medium:LiClO₄, metal ion: VO₂₊. K1 calc. on the basis of K(H₂L+H)=1.00

NH₂SO₃- H₂L Sulfamate CAS 5329-14-6 (452)
Sulfamate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) sp oth/un 18°C dil U 1958SAd (8803) 42
K(2VO₃+L=V₂O₅L)=5.12

NH₃O L Hydroxylamine; CAS 5470-11-1 (1808)

Hydroxylamine; NH₂.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	25°C	1.0M	C			K1=2.91 K(H ₂ V ₀ 4+H+2L)=12.95	1997PAa (9275)	43

Method: 51V nmr.

N03-	HL	Nitrate	CAS	7697-37-2	(288)
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Nitrate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sol	NaNO ₃	20°C	0.22M	U	I		K1=-0.07	1973IYb (10006)	44

Metal: V₀2+. At I=1, K1=-0.49

V(V)	sp	NaClO ₄	20°C	1.0M	U				1966IVb (10007)	45
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$$K(V_02+L)=-0.5$$

OH-	HL	Hydroxide	(57)
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Hydroxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	NaCl	25°C	1.00M	U	H			1991CHc (12450)	46

$$\begin{aligned} K(2HV_04=V_207+H_2O) &= 0.83 \\ K(HV_04+OH=V_04+H_2O) &= 0.431 \\ K(HV_04=V_04+H) &= -13.27 \end{aligned}$$

DH(2HV₀4=V₂07+H₂O)=-28 kJ mol⁻¹, DS=-80 J K⁻¹ mol⁻¹;

DH(HV₀4+OH=V₀4+H₂O)=-32, DS=97

V(V)	gl	KNO ₃	20°C	0.50M	M				1983BSb (12451)	47
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$$\begin{aligned} K(3V_03=V_309) &= 7.20 \\ K(4V_03=V_4012) &= 10.15 \end{aligned}$$

V(V)	nmr	NaClO ₄	0°C	2.0M	U				1981HHa (12452)	48
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$$\begin{aligned} *K(HV_04) &= -7.1 \\ *K(H_2V_207) &= -7.2 \\ *K(HV_207) &= -8.9 \\ K(3HV_04+H=V_3010+2H_2O) &= 21.1 \end{aligned}$$

Method: 51V and 170 nmr. Medium: 2.0 M LiClO₄. K(2HV₀4=V₂07+H₂O)=1.39

K(4HV₀4+4H=V₄012+4H₂O)=40.0

V(V)	dis	oth/un	20°C	0.10M	U			K1=12.16 B2=23.78	1978TKa (12453)	49
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V(V)	sp	NaClO ₄	25°C	1.00M	U				1977BMb (12454)	50
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$$\begin{aligned} B(10,14) &= -7.63 \\ B(10,15) &= -11.57 \\ B(10,16) &= -17.40 \\ B'(2,1) &= 11.69 \end{aligned}$$

$B'(3,3)=33.51; B'(-1,1)=-12.09$; $B(m,n): K(mV02=(V02)m(OH)n+nH)$
 $B'(m,n): K(V04=(V04)m(OH)n+nH)$

V(V) ix NaCl04 20°C 2.00M U K1=14.86 B2=29.52 1977LPa (12455) 51

Metal ion: V0+++

O2-- H2L Peroxide CAS 7772-84-1 (2813)

Peroxide; -0.0-

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) nmr NaCl 25°C 0.15M C 2004AGa (12741) 52

$B(2,1,1,1)=20.22$
 $B(2,1,1,2)=25.96$
 $B(0,1,2,1)=14.67$

$B(p,q,r,s): pH+q(H2V04)+r(H2O2)+sL=(H)p(H2V04)q(H2O2)r(L)s.$

V(V) nmr NaCl 25°C 0.15M C M 2004GGb (12742) 53

$B(2,2,1,1)=20.06$
 $B(1,2,3,1)=27.06$
 $B(2,2,3,1)=32.59$
 $B(2,2,2,1)=29.43$

51V nmr. $B(p,q,r,s): pH+q(H2V04)+rH2O2+sL=(H)p(H2V04)q(H2O2)r(L)s.$

$B(2,1,1,1)=19.56$, $B(0,1,2,1)=11.87$, $B(1,1,2,1)=17.21$.

V(V) gl NaCl 25°C 0.15M C 2001SAb (12743) 54

$B(0,1,1,1)=7.68$
 $B(1,1,1,1)=14.80$
 $B(-1,1,2,1)=6.29$
 $B(0,1,2,1)=14.48$

Additional method: 51V nmr. HA is alanyl-histidine.

$B(p,q,r,s): pH+qH2V04+rH2O2+sHA=Hp(H2V04)q(H2O2)r(HA)s$

V(V) gl NaCl 25°C 0.15M C 2000AAb (12744) 55

$B(-1,1,1)=-2.27$
 $B(-1,1,2)=3.61$
 $B(0,1,2)=11.28$
 $B(-1,1,3)=5.14$

Additional method: 51V nmr. $B(p,q,r): pH+qH2V04+rH2L=Hp(H2V04)q(H2L)r$

$B(-1,2,4)=16.48$, $B(2,2,3)=23.73$, $B(-1,2,1)=-0.34$, $B(-1,2,2)=5.33$

V(V) gl NaCl 25°C 0.15M C M 2000SAC (12745) 56

$B(0,1,1,1)=7.69$
 $B(1,1,1,1)=14.82$
 $B(-1,1,2,1)=6.24$
 $B(0,1,2,1)=14.50$

Additional method: 51V nmr.

$B(p,q,r,s): pH+qH2V04+rH2O2+sHL=(H)p(H2V04)q(H2O2)r(HL)s$

V(V) nmr KCl 22°C 1.00M U 1993TJa (12746) 57

$$K(2V_03(02)2=(V_03)2(02)4)=1.52$$

$$*K(V_03(H_2O_2))=-7.38$$

V(V) sp NaClO₄ 23°C 1.0M U 19670Wa (12747) 58
 $K(VOL+H_2L=VOL_2+2H)=2.15$?

V(V) sp oth/un ? var U 1966BVc (12748) 59
 $K(V_02HL+H)=0.4$
 $K(V_02+HL)=15.92$

V(V) kin oth/un 65°C var U T 1965BYb (12749) 60
 $K(V_03+H_2L=V_02L+H_2O)=7.18$

$K=5.88(45\text{ }^\circ\text{C}); 6.74(55\text{ }^\circ\text{C})$

V(V) sp oth/un 24°C var U 1963FLa (12750) 61
 $K(H_2V_03L+2H)=7.8$
 $K(HV_03L+H)=6.5$

V(V) sp NaClO₄ 22°C 1.0M U I 1961DEC (12751) 62
 $K(V_02+H_2L=VOL(\text{red})+H_2O)=4.53$

Medium: HClO₄. In 0.5 M H₂SO₄ K=4.50. Data also for formation of a yellow anionic complex

V(V) gl NaCl ? 1.0M U I 1960CHb (12752) 63
 $K(VOL_2+HV_02L_2=HV_03L_4)=2.30$
 $K(HV_02L_2+H=VOL_2+H_2O)=7.15$
 $K(HV_02L_2+H_2L=HVOL_3+H_2O)=1.68$

In 4 M NaClO₄, 0 °C: $K(HVOL_3+HL=VL_4+H_2O)=0.45$, $K(VOL_3+HL=VL_4+OH)=-1.60$

V(V) sp oth/un 20°C dil U 1956TSc (12753) 64
 $K(V(V)+H_2L)=4.11$

V(V) sp oth/un ? var U 1952BVa (12754) 65
 $K(V(V)+H_2L)=4.52$

V(V) sp oth/un 20°C 20% U 1948STa (12755) 66
 $K(V(V)+H_2L)=4.91$

V(V) sp oth/un 20°C var U 1937RUa (12756) 67
 $K(V(V)+H_2L)=3.63$

Po4--- H3L Phosphate CAS 7664-38-2 (176)
Phosphate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V) nmr KCl 25°C 1.00M U I 1986GTa (13376) 68
 $K_{1\text{eff}}=1.40$

At pH 6.69. $K_{1\text{eff}}$: $H_2V_04+H_2L=H_2V_03L+H_2O$

V(V) sp NaClO₄ 25°C 1.00M U 1974IKb (13377) 69
 $K(VO_2+H_2L)=1.37$
 $K(VO_2+HL)=5.20$
 $K(VO_2+2HL)=8.20$

P207---- H4L Pyrophosphate CAS 2466-09-3 (198)
Diphosphate; from $(HO)_2PO_0.O.PO(OH)_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	25°C	1.00M	U				1986GTa (13668)	70
								K1eff=1.59		

At pH 7.98: K_{1eff}: H₂V₀₄+HL=HV_{03L}+H₂O

V(V) sp NaClO₄ ? 1.00M U 1972LGb (13669) 71
 $K(V_0+HL)=6.52$
 $K(V_0+H_2L)=3.30$
 $K(V_0+2H_2L)=6.41$

P3010----- H5L CAS 10380-08-2 (1001)
Tripolyphosphate; from (HO)2PO.O.PO(OH).O.PO(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	NaClO4	25°C	1.00M	U				1984IKa (13916)	72
								$K(V_2O_5 + H_2P_3O_10) = 4.60$		
								$K(V_2O_5 + HP_3O_10) = 8.13$		
								$K(V_2O_5 + 2(HP_3O_10)) = 12.9$		

S04-- H2L **Sulfate** CAS 7664-93-9 (15)
Sulfate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
V(V)	sp	NaClO ₄	22°C	5.80M	U					1978FHa (16656)	73
									$K(V_2O_5 + SO_4) = 0.84$		
									$B(V_2O_5 + 2SO_4 + H) = 1.22$		

V(V) sol NaClO₄ 20°C 1.0M U 1973IYb (16657) 74
 $K(VO_2+L) = 0.95$

V(V) con non-aq 25°C 100% U 1966GKa (16658) 75
 $K(VO(HL)_4 + H) = -2.1$

Method: also freezing point. Medium: H₂SO₄, m units

V(V) sp NaClO₄ 19°C 1.01M U K1=0.97 1966IVc (16659) 76
 *K1=-0.14

Metal:V02+

WO₄-- H₂L Tungstate CAS 13783-36-3 (445)

Tungstate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaCl	25°C	0.60M	C			B(6,4,2)=56.67 (cis isomer) B(7,4,2)=58.90 (cis isomer) B(6,4,2)=54.75 (trans isomer) *K(trans-HW4V2019)=-2.5	1996AHa (17448)	77
B(p,q,r): pH+q(WO4)+r(H2V04)=Hp(WO4)q(H2V04)r. B(6,1,9)=65.52, B(7,1,9)=65.7										
V(V)	gl	NaCl	25°C	0.60M	C			B(4,3,3)=41.56 (fac isomer) B(5,3,3)=48.55 (fac isomer) B(4,3,3)=40.60 (mer isomer) B(5,3,3)=48.59 (mer isomer)	1996AHa (17449)	78

B(p,q,r): pH+q(WO4)+r(H2V04)=Hp(WO4)q(H2V04)r.

V(V)	gl	NaCl	25°C	1.00M	U	M			1981AIa (17450)	79
								K(V3W3019)=33.95 K(HV3W3019)=41.82		

Equations: 0.75V4012+3WO4+4H=V3W3019+2H2O

0.75V4012+3WO4+5H=HV3W3019+2H2O.

V(V)	gl	NaCl	25°C	1.00M	U	M			1981IAa (17451)	80
								K(V2W4019)=51.06		

Equation: V4012/2+4WO4+6H=V2W4019+3H2O

V(V)	sp	NaCl	25°C	0.50M	U				1960CHb (17452)	81
								K(V2W4019+H)=2.8		

CH40 L Methyl alcohol CAS 67-56-1 (597)
Methanol; CH3.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	alc/w	?	100%	U	M		K(VOA+L)=-0.770 K(VOAL+2L=VO(H-1L)3+H3A)=-4.22 K(VOBL+2L=VO(H-1L)3+H3B)=-5 K(VOB+L)=-1.44	1993CCc (17910)	82

Method:NMR. Medium: MeOH. H3A:Triethanolamine.

H3B:Tri(2-propanol)amine. Also data for L=i-PrOH and t-BuOH.

CH5NO L CAS 593-77-1 (7603)
N-Methylhydroxylamine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	25°C	1.0M	C			K1=3.53	1997PAa (18034)	83

$$K(H_2V_04+H+2L)=13.43$$

Method: ^{51}V nmr.

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)
Ethanedioic acid; (COOH_2)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	21°C	1.0M	C				1987TGc (19134)	84
								$K(H_2V_04+H+L=H_3V_04L)=8.15$		
								$K(H_3V_04L+H+L=V_02L_2+2H_2O)=9.26$		

Method: ^{51}V nmr.

V(V)	gl	NaCl	25°C	0.60M	C			1986EAb (19135)	85
								$K(2H+H_2V_04+2L)=17.11$	
								$K(2H+H_2V_04+L)=13.07$	

V(V)	gl	KNO ₃	20°C	0.50M	M			1983BSb (19136)	86
								$K(V_03+2H+2L=V_02L_2+H_2O)=17.42$	
								$K(V_03+H+L=V_02(OH)L)=8.58$	

V(V)	sp	none	20°C	0.00	U		K1=6.49	B2=9.99	1969VIa (19137)	87
Metal ion:	V ₀₂₊									

V(V)	sp	NaClO ₄	20°C	1.00M	U		K1=5.08		1966IVa (19138)	88
Metal ion:	V ₀₂₊									

V(V)	sp	NaCl	25°C	1.00M	U		K1=2.32	B2=0.04	1964NNa (19139)	89
Metal ion:	V ₀₂₊									

V(V)	dis	NaClO ₄	20°C	0.10M	U		K1=6.4	B2=9.0	1963STc (19140)	90
Metal:	V ₀₂₊ . Medium:	KClO ₄								

C2H4O2 HL Acetic acid CAS 64-19-7 (36)
Ethanoic acid; CH₃.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	NaClO ₄	20°C	1.00M	U		K1=2.28		1966IVa (20223)	91
Metal ion:	V ₀₂₊									

C2H4O3 HL Glycolic acid CAS 79-14-1 (33)
2-Hydroxyethanoic acid; HO.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaCl	25°C	1.00M	C				1975VIa (20650)	92
								$K(V_4O_12+2L+2H=V_4O_11L_2)=14.3$		
								$K(0.5V_4O_12+2L+2H=V_2O_5L_2)=13$		

C2H5NO2 HL Glycine CAS 56-40-6 (85)
2-Aminoethanoic acid; H2N.CH2.COOH

Measured at pH 8.5

C2H5NO2 HL Acetohydroxamic CAS 546-88-3 (2766)

Acetohydroxamic acid, N-Hydroxyacetamide; CH₃.CO.NHOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	25°C	0.02M	C				2002BPa (21818)	98
							K1eff=3.32			
							K2eff=2.63			
Medium: 0.02 M MOPS, pH 7.50.										

Medium: 0.02 M MOPS, pH 7.50.

V(V) gl NaCl 25°C 0.15M C 1997YPa (21819) 99
 $B(1,1,1)=7.54$
 $B(0,1,1)=2.70$
 $B(-1,1,1)=-5.66$
 $B(1,1,2)=8.95$

B(p,q,r): pH+q(H₂VO₄)+rHL = H_p(H₂VO₄)q(HL)r. B(0,1,2)=5.40,
 B(-1,1,2)=-3.8. Additional methods: spectrophotometry, NMR.

C2H6N2O L Glycinamide CAS 598-41-4 (60)

2-Aminoethanoic acid amide: H₂N.CH₂.CO.NH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	alc/w	?	100%	U	M			1993CCc (22031)	101
K(VOA+L)=-0.569 K(VOAL+2L=VO(H-1L)3+H3A)=-2.40 K(VOB+L)=-1.24 K(VOBL+2L=VO(H-1L)3+H3B)=-4.22										
Method:NMR. Medium: EtOH. H3A:triethanolamine, H3B=tri(2-propyl)amine										

C2H7NO		L					CAS	5725-96-2	(7602)	
N,N-Dimethylhydroxylamine;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	25°C	1.0M	C				1997PAa (22419)	102
K1eff=2.86 B2eff=6.91										
Method: 51V nmr. Keff for pH 6.7.										

C3H4N2		L	Imidazole				CAS	288-32-4	(90)	
1,3-Diazole, imidazole; C3H4N2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaCl	25°C	0.15M	C	M			2000AAb (23929)	103
B(0,1,1,1)=7.31 B(0,12,1)=14.84 B(-1,2,4,1)=19.27										
B(p,q,r,s): pH+qH2V04+rH2O2+sHL=Hp(H2O4)q(H2O2)r(HL)s										
Additional method: 51V nmr.										
V(V)	nmr	KCl	22°C	1.00M	U				1993TJa (23930)	104
K(V03(02)2+L)=2.61										

C3H7NO2		HL	Alanine				CAS	56-41-7	(86)	
2-Aminopropanoic acid; H2N.CH(CH3).COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaClO4	25°C	0.10M	C		K1=10.56	B2=19.55	2003GFB (26291)	105
V(V)	nmr	KCl	22°C	1.00M	U				1993TJa (26292)	106
K(V03(02)2+L)=0.04 K(V03(02)2L+L)=-0.045										

C3H7NO3		HL	Serine				CAS	56-45-1	(49)	
2-Amino-3-hydroxypropanoic acid; H2N.CH(CH2.OH)COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

V(V)	sp	NaClO4	25°C	1.00M	U	K1=9.20 K(VO2L+H)=0.90	1995KGa (27194) 107

C3H8N2O2		HL				(6666)	
beta-Alaninehydroxamic acid; NH2.CH2.CH2.CO.NHOH							
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo
V(V)	gl	NaClO4	25°C	0.10M	C	K1=14.2 B(VO2HL)=18.72 B(VO2H-1L)=6.95 B(VO2H-2L)=-3.17 B(VO2H3L2)=37.5	1999YPa (27609) 108
Additional method: 51V nmr. B(VO2H2L2)=34.96, B(VO2HL2)=27.78							

C3H12N09P3		H6L	NTPA			CAS 6419-19-8	(2920)
Nitrilotris(methylenephosphonic acid); N(CH2PO3H2)3							
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo
V(V)	sp	oth/un	20°C	1.00M	U		1982SPb (28594) 109
K(VO2+H2L)=12.6							

C4H6N2		L	N-Me-Imidazole			CAS 616-47-7	(354)
N-Methyl-1,3-diazole; C3H3N2.CH3							
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo
V(V)	nmr	KCl	22°C	1.00M	U		1993TJa (29609) 110
K(VO3(02)2+L)=2.57							

C4H6O6		H2L	L-Tartaric acid			CAS 87-69-4	(92)
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH							
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo
V(V)	gl	KNO3	20°C	0.50M	M		1983BSb (31389) 111
K(4VO3+4H+2L=V4O8(H-2L)2)=39.7 K(2VO3+H+L=HV2O5(H-2L))=11.82							
V(V)	sp	oth/un	25°C	?	U		1963GMa (31390) 112
K(2HV03+H2L)=4.9 ?							

C4H7N04		H2L	IDA			CAS 142-73-4	(118)
Iminodiethanoic acid; HN(CH2.COOH)2							
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo
V(V)	gl	NaClO4	25°C	3.0M	U	K1=11.7 B2=22.2	1979ZLa (32391) 113

C4H8N203 HL Asparagine CAS 70-47-3 (17)
2-Aminobutanedioic acid 4-amide; H2N.CH(CH2.CO.NH2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	NaClO4	25°C	0.15M	U				2001GMB (32741)	114
								$K(V_02+L)=9.47$		
								$K(V_02+2L)=14.57$		

C4H8O3 HL CAS 594-61-6 (81)
2-Hydroxy-2-methylpropanoic acid; (CH3)2C(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	NaCl	25°C	1.0M	C				2001HBa (33535)	115
								$K_{eff}(H_2V_04+L)=1.41$		
								$K_{eff}(2H_2V_04+2L)=6.66$		
								$K_{eff}(3H_2V_04+2L)=10.2$		
								$*K(H_2V_04L)=-6.2$		

Method: 51V NMR spectroscopy. Medium pH: 7.06 (HEPES buffer).

C4H9N03 HL Threonine CAS 72-19-5 (48)
2-Amino-3-hydroxybutanoic acid; H2N.CH(CH(OH).CH3)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	NaClO4	25°C	1.00M	U			$K_1=10.29$	1995KGa (34337)	116
								$K(V_02L+H)=1.17$		

C4H9N303 H2L CAS 36212-68-7 (2042)
2-Aminobutanedioic acid-4-hydrazide; HOOC.CH(NH2).CH2.CO.NH.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	NaClO4	20°C	2.00M	U				1977MSb (34424)	117
								$K(V_02+H_2L)=2.06$		

C4H1003 L CAS 111-46-6 (3579)
2,2'-Oxydiethanol; (HO.CH2.CH2)2.0 (Diethylene glycol)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	25°C	0.40M	U				1988CSb (34704)	118
								$K(H_2V_04+L)=0.08$		

C4H11N02 L Diethanolamine CAS 111-42-2 (89)
2,2'-Iminodiethanol; HN(CH2.CH2.OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V) nmr KCl RT 0.40M C H 1998CBc (34963) 119
K(H₂V₀₄+L=V₀₂H-2L+2H₂O)=2.71

Method: 51V and 1H nmr. DH(K)=-21.2 kJ mol⁻¹, DS(K)=-13.5 J K⁻¹ mol⁻¹.

V(V) gl NaCl 25°C 0.60M M 1991CEa (34964) 120
K(H₂V₀₄+L)=3.02

V(V) nmr KCl 25°C 0.40M U 1988CSb (34965) 121
K(H₂V₀₄+HL=H₂V₀₄L+H)=-6.17
K(HV₀₄+HL)=2.04
K(H₂V₀₄+L)=2.71
K(HV₀₄+L=V₀₃L+OH)=-3.06

C4H11N08P2 H5L CAS 2439-99-8 (2129)

N-Carboxymethyl-N,N-bis(methylenephosphonic acid); HOOC.CH₂.N(CH₂.PO₃H₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	20°C	1.00M	U				1982SPb (35116)	122

K(V₀₂+HL)=13.24

C5H5NO HL 2-Pyridinol CAS 142-08-5 (1890)

2-Hydroxypyridine, Pyridin-2-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	25°C	0.01M	U				1989GTa (36696)	123

K_{eff}(V+HL=VL+H)=-0.05

At pH 7.0. V=H₂V₀₄

C5H9N04 H2L Glutamic acid CAS 56-86-0 (22)

2-Aminopentanedioic acid; H₂N.CH(CH₂.CH₂.COOH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	NaClO ₄	25°C	1.00M	U			K1=11.89 B2=15.99	1997GAa (39141)	124

C5H9N04 H2L MIDA CAS 4408-64-4 (190)

N-Methyliminodethanoic acid; CH₃.N(CH₂.COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaClO ₄	25°C	1.00M	U T H			K1=10.16	1976YNa (39290)	125

K(V₀₂(H₂O)=V₀₂(OH)+H)=-6.13
K(V₀₂LOH+H₂O=HV₀₄+L+2H)=-19.7
K(V₀₂L+2H₂O=HV₀₄+L+3H)=-25.9

V(V) as V₀₂⁺. Data also at 15 and 35 C

DH=-25.0 kJ mol⁻¹ and DS=117 J mol⁻¹ K⁻¹.

C5H10N07P H4L PMIDA CAS 5994-61-6 (2433)
 N-(Phosphonomethyl)iminodiethanoic acid; H2O3P.CH2.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	KCl	25°C	0.20M	C				1999CJa (39687)	126
						B(5,1,1)=35.93		(V02H2L)		
						B(4,1,1)=34.77		(V02HL)		
						B(3,1,1)=31.03		(V02L)		

B(p,q,r): pHVO4+qL+rH=(HV04)pLqHr.

Additional methods: 1H, 13C, 31P, 17O, 51V nmr.

C5H10N203 HL Glutamine CAS 56-85-9 (18)
 2-Aminopentanedioic acid 5-amide; H2N.CH(CH2.CH2.CO.NH2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	NaClO4	25°C	0.1M	C I		K1=9.91	B2=17.01	2003GAa (39842)	127
for I=0.3 M	K1=9.71;	B2=16.69;	for I=0.5 M	K1=9.93;	B2=17.04					
for I=0.7 M	K1=9.96;	B2=17.24;	for I=1.0 M	K1=10.12;	B2=17.36					

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	NaClO4	25°C	0.15M	U				2001GMb (39843)	128
						K(V02+L)=9.54				
						K(V02+2L)=14.81				

V(V) sp NaClO4 25°C 1.00M U K1=8.70 B2=11.98 1997GAa (39844) 129

C5H10N203 HL Ala-Gly CAS 687-69-4 (55)
 Alanyl-glycine; H2N.CH(CH3).CO.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaClO4	25°C	0.10M	C		K1=9.85		2003GFb (39894)	130
						B(V02HL)=4.60				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaCl	25°C	0.60M	C				1997FEa (39895)	131
						B(0,1,1)=1.715				
						B(0,2,2)=5.76				
						B(0,1,2)=2.86				
						B(0,2,1)=4.66				

B(p,q,r): pH+qH2V04+rHL=Hp(H2V04)q(HL)r

C5H11N02 HL Valine CAS 72-18-4 (43)
 2-Amino-3-methylbutanoic acid; H2N.CH(CH(CH3)2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaCl	25°C	0.10M	U	TIH	K1=7.30	B2=10.08	2002GNa (40770)	132
Cation V02+; For 15 C	K1=7.61;	B2=9.21;	for 35 C	K1=7.80;	B2=11.01					
For 45 C	K1=8.30;	B2=11.91;	Also data for I=0.3;	0.5;	0.7 and 1.0					

C5H11NO2S HL Methionine CAS 63-68-3 (42)
2-Amino-4-(methylthio)butanoic acid; H2N.CH(CH2.CH2.S.CH3)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	alc/w	25°C	0.15M	U	I			2002GVa (41132)	133

Medium: 0.15 M NaClO₄ in 0.194 mol parts MeOH in H₂O; for 0.123 m.p. K₁=9.26
for 0.15 M NaClO₄ in H₂O K₁=8.14

V(V) gl alc/w 25°C 0.15M U I 2002GVa (41133) 134
K(V02+L)=9.41

Medium: 0.15 M NaClO₄ in 0.18 mol parts dioxane/H₂O; for 0.128 m.p. K₁=9.32
 for 0.053 mol parts dioxane/H₂O K₁=9.21

C5H12O3 L (5881)
1,1,1-Tris(hydroxymethyl)ethane; CH₃.C(CH₂.OH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
V(V)	nmr	KCl	25°C	1.0M	U					1988TGb (41646)	135

Keff(H2V04+L)=-0.29 pH 7.5
Keff(H2V04L+L)=-0.74 pH 7.5

C5H13NO2 L CAS 105-59-9 (1070)
N-Methyldiethanolamine: CH₃.N(CH₂.CH₂.OH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
V(V)	nmr	KCl	RT	0.40M	C	H				1998CBc (41743)	136

Method: 51V and 1H nmr. $\Delta\text{H}(\text{K}) = -22.2 \text{ kJ mol}^{-1}$, $\Delta\text{S}(\text{K}) = -13.9 \text{ J K}^{-1} \text{ mol}^{-1}$.

C6H2O4Cl2 H2L Chloranilic acid CAS 87-88-7 (1281)
3,6-Dichloro-2,5-dihydroxy-1,4-benzoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	30°C	?	U			K1=3.48	1981BMD	(42059) 137

C6H30Cl3 **HL** **CAS 88-06-2 (508)**

2,4,6-Trichlorophenol; HO.C₆H₂(Cl)₃

V(V) nmr mixed 25°C 42% C 1988GTa (42164) 138
 $K(H_2V_04+H_2)_-1$ 46

Medium: 42% acetone

medium. 4% acetone

C6H5NO₂ HL Picolinic acid CAS 98-98-6 (391)
2-Pyridine-carboxylic acid; C5H4N.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo	
V(V)	nmr	NaCl	25°C	0.15M	C				B(1,1,1)=9.31 B(2,1,1)=14.06 B(2,1,2)=18.92	2004AGa	(42614)	139

B(p,q,r): pH+q(H₂VO₄)+rL=(H)p(H₂VO₄)q(L)r. Isomers detected for MHL and MH₂L₂. Ternary complexes with H₂O₂ reported.

C6H5N03 H2L CAS 609-71-2 (5910)
2-Hydroxypyridine-3-carboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
V(V)	nmr	KCl	25°C	0.50M	U					1989GTa (42725)	140
									$K_{eff}(V+HL=VL+H)=-0.24$		
									$K_{eff}(VI+HI=VI\ 2+H)=1.23$		

At pH 7.0, V=H2V04

C6H5NO₃ H2L CAS 874-24-8 (4356)
3-Hydroxypyridine-2-carboxylic acid: C₅H₃N.(OH)(COOH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
V(V)	nmr	KCl	25°C	0.01M	U					1989GTa (42755)	141
										$K_{eff}(V+HL=VL+H)=2.76$	
										$K_{eff}(VL+HL=VL2+H)=2.64$	

At pH 7.0, V=H2V04

C6H5NO3 HL 4-Nitrophenol CAS 100-02-7 (454)
4-Nitrohydroxybenzene; HO-C6H4-NO2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
V(V)	nmr	mixed	25°C	42%	C					1988GTa (42819)	142
										$K(H_2V_04+HL) = 0.52$	

Medium: 42% acetone

C6H5OCl HL 2-Chlorophenol CAS 95-57-8 (3671)
2-Chlorophenol; HO-C₆H₄-Cl

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
V(V)	nmr	mixed	25°C	42%	C					1988GTa (43032)	143
									$K(H_2V_04+HL) = 0.41$		
									$K(H_3V_04I+HL) = 0.46$		

Medium: 42% acetone

C6H5OCl	HL	3-Chlorophenol	CAS 108-43-0	(3672)				
3-Chlorophenol; HO.C6H4.Cl								

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo

V(V)	nmr	mixed	25°C	42%	C			1988GTa (43042) 144
							K(H2V04+HL)=0.43	
Medium: 42% acetone								

C6H6O	HL	Phenol	CAS 108-95-2	(457)				
Hydroxybenzene, phenol; C6H5.OH								

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo

V(V)	nmr	KCl	22°C	1.0M	M	M		1993JTa (43548) 145
							K(V(peroxo)+HL=VL+H)=-5.21	
V is peroxovanadate. Method:NMR.							K(V(diperoxo)+L=VL+H)=-6.22	

V(V)	nmr	mixed	25°C	42%	C			1988GTa (43549) 146
							K(H2V04+HL)=0.25	
							K(H3V04L+HL)=0.04	
Medium: 42% acetone. Data also for complexes of other substituted phenols								

C6H6O2	H2L	Catechol	CAS 120-80-9	(534)				
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH								

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo

V(V)	kin	NaClO4	25°C	1.00M	M			1974KLa (43860) 147
							Keff(V02+H2L)=2.64	
Medium: 0.2 - 1 M HClO4								

C6H6O3	H3L	Pyrogallol	CAS 87-66-1	(696)				
1,2,3-Trihydroxybenzene; C6H3(OH)3								

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo

V(V)	sp	oth/un	23°C	85%	U			1981BMe (43993) 148
							K(V(V)+2H3L)=2.96	
Medium: 96% H2SO4. In 85%: K(V(V)+H3L)=2.6								

V(V)	kin	NaClO4	25°C	1.00M	M			1974KLa (43994) 149
							Keff(V02+H2L)=3.72	
Medium: 0.2 - 1 M HClO4								

C6H6O3	H3L		CAS 533-73-3	(1734)				
1,2,4-Trihydroxybenzene; C6H3(OH)3								

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo

V(V) kin NaClO₄ 25°C 1.00M M 1974KLa (44005) 150
 K_{eff}(V0₂+H₂L)=3.96

Medium: 0.2 - 1 M HClO₄

 C6H6O₃ HL Maltol CAS 118-71-8 (2442)

3-Hydroxy-2-methyl-4H-pyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V)	gl	NaCl	25°C	0.15M	C				1996EBa (44104) 151
								B(0,1,1)=2.66	
								B(0,1,2)=7.02	
								B(-1,1,1)=-7.37	

B(p,q,r): pH+qH₂V0₄+rHL=Hp(H₂V0₄)q(HL)r

V(V)	gl	NaCl	25°C	0.15M	U		K1=7.5	B2=13.70	1995CGc (44105) 152
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Metal ion: V0₂₊

V(V)	gl	KNO ₃	20°C	0.50M	M	M		1983BSb (44106) 153
								K(V0 ₃ +2HL=V0 ₂ L ₂ +H ₂ O)=7.31
								K(V0 ₃ +H+HL+A=V0 ₂ LA+H ₂ O)=12.67

H₂A is oxalic acid.

C6H6O₃ HL Allomaltol CAS 644-46-2 (2688)

5-Hydroxy-2-methyl-4H-pyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V)	gl	NaCl	25°C	0.16M	C		K1=9.55	B2=16.86	2002SSb (44129) 154
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C6H₆O₈S₂ H₄L Tiron CAS 149-45-1 (104)

4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)₂.C₆H₂(SO₃H)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V)	sp	oth/un	22°C	0.10M	U			1969BFa (44510) 155	
								K(V0 ₂ +H ₂ L=V0 ₁ L+H ₂ O)=1.15	

Medium: NaSO₄, pH=1

C6H₈O₇ H₃L Citric acid CAS 77-92-9 (95)

2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH₂.CH(OH)(COOH).CH₂COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V)	nmr	NaCl	25°C	0.15M	C	M		2004GGb (46305) 156
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B(2,1,1)=14.19

B(3,1,1)=18.25

B(1,2,1)=12.58

B(2,2,1)=20.06

51V nmr. B(p,q,r): pH+q(H2V04)+rL=(H)p(H2V04)q(L)r. B(3,2,1)=25.16, B(6,2,2)=40.69. Ternary complexes with H2O2 and lactic acid also reported

V(V)	nmr	NaCl	25°C	0.60M	C	1989EAa (46306) 157
					K(2(H2V04)+H+L)=12.84	
					K(2(H2V04)+2H+L)=19.68	
					K(3H+2(H2V04)+L)=24.12	

$$K(3H+H2V04+L)=18.1$$

C6H9N06	H3L	NTA	CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH2.COOH)3			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	kin	NaClO4	25°C	3.00M	U	T			1984LAa (47081) 158	

$$K(V02L+H) < 0.0$$

Data at 10 to 40C

V(V)	sp	NaClO4	25°C	3.00M	C	1978LLa (47082) 159
						K(V02+L)=13.8

V(V)	gl	NaClO4	25°C	1.00M	U	T	K1=13.8	1976YNa (47083) 160
							K(V02L+2H2O=HV04+L+3H)=-28.3	

V(V) as V02+. Data also at 15 and 35 C
DH(K1)=0 kJ mol-1 and DS(K1)=263 J mol-1 K-1.

V(V)	gl	NaClO4	25°C	3.00M	C	1975LLb (47084) 161
						K(V02+H3L=V02L+3H)=-0.07

By spectrophotometry: 270nm: K=-0.03; 300nm: K=-0.11. Kinetics also studied.

C6H9N3O2	HL	Histidine	CAS 71-00-1 (1)
2-Amino-3-(4'-imidazolyl)propanoic acid; H2N.CH(CH2.C3H3N2)COOH			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	22°C	1.00M	U				1993TJa (47625) 162	

$$K(V03(02)2+L)=2.36$$

C6H1008	H2L	Mucic acid	CAS 526-99-8 (3650)
2,3,4,5-Tetrahydroxyhexanedioic acid, Galactaric acid; HOOC.(CHOH)4.COOH			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	KCl	25°C	0.20M	C				2004DGa (48440) 163	

$$B(1,1,4)=22.51$$

$$B(1,1,2)=16.17$$

$$B(1,1,1)=10.13$$

$$B(2,1,4)=32.46$$

B(p,q,r): p(HV04)+qL+rH=(V02)p(L)q(H)r+2p(H2O). B(2,1,3)=28.05

B(2,2,6)=42.79, B(2,2,5)=39.54, B(2,2,4)=35.35, B(3,2,5)=46.5

C6H1008 H2L Saccharic acid CAS 87-73-0 (1191)
D-2,3,4,5-Tetrahydroxy-1,6-hexanedioic acid; Glucaric acid; HOOC.(CHOH)4.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	KCl	25°C	0.20M	C				2004DGa (48489)	164
								B(1,1,4)=22.52		
								B(1,1,2)=16.11		
								B(1,1,1)=10.24		
								B(2,1,4)=33.69		

B(p,q,r): p(HV04)+qL+rH=(V02)p(L)q(H)r+2p(H2O). B(2,1,3)=29.01
B(2,2,6)=43.14, B(2,2,5)=39.50, B(2,2,4)=35.20, B(3,2,5)=48.1

C6H11N05 H2L HIMDA CAS 93-62-9 (192)
N-(2-Hydroxyethyl)iminodiethanoic acid; HO.CH2.CH2.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	25°C	1.00M	U			K1=11.86	1987TKa (48808)	165

C6H12N203 HL Ala-Ala CAS 1948-31-8 (53)
Alanyl-alanine; H2N.CH(CH3).CO.NH.CH(CH3).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaClO4	25°C	0.10M	C			K1=9.93	2003GFb (49108)	166
								B(VO2HL)=4.66		

C6H12N204 H2L EDDA CAS 5657-17-0 (119)
1,2-Diaminoethane-N,N'-diethanoic acid; HOOC.CH2.NH.CH2.CH2.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	kin	NaClO4	25°C	3.00M	U	T			1984LAa (49280)	167
								K(VO2L+H)=0.72		
Also data at 15 and 35 C										

V(V) gl NaClO4 25°C 3.0M U K1=15.98 1979ZLa (49281) 168

V(V) gl NaClO4 25°C 1.00M U T H K1=14.5 1976YNa (49282) 169
K(VO2L+2H2O=HV04+L+3H)=-29.7

V(V) as V02+. Data also at 15 and 35 C
DH(K1)=-59.0 kJ mol-1 and DS=79.0 J mol-1 K-1.

C6H1202 L CAS 1792-81-0 (3657)
cis-1,2-Cyclohexanediol; C6H10(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V) nmr KCl 25°C 1.0M U 1988TGa (49431) 170
 Keff(V+L)=-0.17
 Keff(VL+L)=-0.62

At pH 7.5. V=H2V04. Data are for trans-L, for cis isomer: Keff(V+L)=-0.62

 C6H12O3 HL DiEtGlycolic CAS 3639-21-2 (421)
 2-Ethyl-2-hydroxybutanoic acid; (C2H5)2.C(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	NaCl	25°C	1.0M	U				2001HBa (49467)	171
								Keff(H2V04+L)=1.30		
								Keff(2H2V04+2L)=6.52		
								Keff(3H2V04+2L)=11.43		
								*K(H2V04L)=-6.6		

Method: 51V NMR spectroscopy. Medium pH: 7.1 (HEPES buffer).

 C6H13N02 HL Isoleucine CAS 73-32-5 (424)
 2-Amino-3-methylpentanoic acid; CH3.CH2.CH(CH3).CH(NH2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	NaClO4	25°C	0.20M	C	I		K1=8.61 B2=13.88	2000GFa (49919)	172
Data for 0.2-0.8 M NaClO4. For I=0.4 M, K1=8.41, B2=12.85; I=0.6 M, K1=7.86, B2=11.66; I=0.6 M, K1=8.05, B2=11.94.										

 C6H13N02 HL Leucine CAS 61-90-5 (47)
 2-Amino-4-methylpentanoic acid; H2N.CH(CH2.CH(CH3)2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaClO4	25°C	0.15M	C	I		K1=9.51	1999GMa (50120)	173
Other media: 0.05-0.19 mole fraction MeOH/H2O and 0.05-0.18 mole fraction dioxane/H2O, 0.15 M NaClO4.										

V(V)	sp	NaClO4	25°C	0.15M	C	I			1999GMa (50121)	174
								K(V02+HL=V02L+H)=-7.99		
								K(V02L+HL=V02L2+H)=-6.74		

Other media: 0.05-0.19 mole fraction MeOH/H2O and 0.05-0.18 mole fraction dioxane/H2O, 0.15 M NaClO4.

 C6H13N05 HL Tricine CAS 5704-04-1 (1239)
 N-(Tris(hydroxymethyl)methyl)glycine; (HO.CH2)3C.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaCl	25°C	0.60M	M				1991CEa (50513)	175
								K(H2V04+L)=3.65		
								K(H+H2V04+L)=6.69		

C6H15N02 L CAS 7087-40-5 (8757)
2-[(2-Hydroxyethyl)amino]-2-methyl-1-propanol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	RT	0.40M	C				1998CBc (51198)	176
								K(H2V04+L=V02H-2L+2H2O)=0.30		

Method: 51V and 1H nmr.

C6H15N02 L CAS 139-87-7 (3707)
N-Ethyl-2,2'-iminodiethanol; CH3.CH2.N(CH2.CH2.OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	RT	0.40M	C	H			1998CBc (51209)	177
								K(H2V04+L=V02H-2L+2H2O)=2.477		

Method: 51V and 1H nmr. DH(K)=-27.5 kJ mol-1, DS(K)=-42.0 J K-1 mol-1.

C6H15N03 Triethanolamine CAS 102-71-6 (447)
Tris-(2-hydroxyethyl)amine; L

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	RT	0.40M	C	H			1998CBc (51305)	178
								K(H2V04+L=V02H-2L+2H2O)=2.72		
V(V)	nmr	KCl	25°C	0.40M	U				1988CSb (51306)	179
								K(H2V04+HL=H2V04L+H)=-5.57		
								K(HV04+HL)=2.62		
								K(H2V04+L)=3.30		
								K(HV04+L=V03L+OH)=-3.0		

C6H15N04 L CAS 7343-51-3 (8756)
2-[(Hydroxyethyl)amino]-2-(hydroxymethyl)-1,3-propanediol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	RT	0.40M	C				1998CBc (51308)	180
								K(H2V04+L=V02H-2L+2H2O)=3.20		

Method: 51V and 1H nmr.

C6H20N2012P4 H8L EDTPA CAS 1429-50-1 (434)
Ethane-1,2-bis(iminobis(methyleneephosphonic acid)); ((H2O3PCH2)2NCH2.)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	20°C	1.00M	U				1982SPb (52369)	181
								K(V02+H4L)=10.0		

C7H5N04 H2L Dipicolinic aci CAS 449-83-2 (418)
 2,6-Pyridinedicarboxylic acid; C5H3N.(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	KCl	25°C	0.40M	C				2000CYa (52817)	182

$$K(H_2V_04+2H+L)=15.79$$

$$K(H_2V_04+3H+L)=16.3$$

V(V)	sp	NaClO4	25°C	1.00M	C T		K1=8.65	1977FHb (52818)	183
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$$K((V_02)L(OH)+H)=5.87$$

Metal: V02+. At 17 C: K((V02)LOH+H)=5.91; 35 C: 5.82

C7H5N05 H3L CAS 499-51-4 (3150)
 4-Hydroxypyridine-2,6-dicarboxylic acid; HO.C5H2N(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	KCl	25°C	0.20M	C				2003JJa (53078)	184

$$K(H_2V_04+3H+L=V_02HL+2H_2O)=24.26$$

$$K(H_2V_04+2H+L=V_02L+2H_2O)=20.56$$

$$K(H_2V_04+H+L=V_02H-1L+2H_2O)=12.7$$

V(V)	nmr	oth/un	24°C	var	C			2002YLb (53079)	185
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$$K(V_02+H_2L=V_02L+2H)=1.96$$

$$K(H_2V_04+H+HL=V_02L+2H_2O)=10.20$$

Self-medium: 0.004-0.04 M vanadate. Method: 1H and 51V nmr.

C7H6N204 HL CAS 1613-76-9 (8273)
 4-Nitro-benzohydroxamic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	25°C	0.02M	C				2002BPa (53515)	186

$$K_{eff}=4.46$$

$$K_{2eff}=1.0$$

Medium: 0.02 M MOPS, pH 7.50. By 1H nmr, K1eff=4.48, K2eff=1.13.

C7H6O3 H2L Salicylic acid CAS 69-72-7 (14)
 2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	25°C	1.00M	U				1989GTa (54329)	187

$$K_{eff}(V+HL=VL+H)=-0.64$$

$$K_{eff}(VL+HL=VL2+H)=0.23$$

At pH 7.0. V=H2V04

C7H6O3 H2L CAS 99-96-7 (1371)
 4-Hydroxybenzoic acid; HO.C6H4.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	mixed	25°C	42%	C				1988GTa (54437)	188
K(H ₂ V ₀ 4+H ₂ L)=0.34										
Medium: 42% acetone										

C7H6O5		H4L					CAS	610-02-6	(3725)	
2,3,4-Trihydroxybenzoic acid; (HO) ₃ .C ₆ H ₂ .COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	?	?	U				1969KSb (54724)	189
K(V ₀ 2+2H ₂ L)=5.05										

C7H6O5		H4L		Gallic acid			CAS	149-91-7	(446)	
3,4,5-Trihydroxybenzoic acid; C ₆ H ₂ (OH) ₃ .COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	alc/w	20°C	50%	U	I			1970CSc (54770)	190
K(V ₀ 2+H ₂ L)=3.06										
K(V ₀ 2+2H ₂ L)=5.85										
K(V ₀ 2+3H ₂ L)=8.48										
Medium: 0-100% methanol. Range of temperture 18-22C. K(V ₀ 2+H ₂ L)(0%)=2.08										
(100%)=3.22; K(V ₀ 2+2H ₂ L)(25%)=5.02, (100%)=5.89; K(V ₀ 2+3H ₂ L)(100%)=8.28										

C7H7NO2		HL					CAS	495-18-1	(184)	
Benzohydroxamic acid; C ₆ H ₅ .CO.NH.OH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	25°C	0.02M	C				2002BPa (55525)	191
K _{1eff} =3.47										
K _{2eff} =2.62										
Medium: 0.02 M MOPS, pH 7.50. By ¹ H nmr, K _{1eff} =3.45, K _{2eff} =2.61.										

C7H8O3		HL					CAS	2298-99-9	(8830)	
3-Hydroxy-2,6-dimethyl-4H-pyran-4-one;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaCl	25°C	0.16M	C		K ₁ =10.60	B ₂ =18.04	2002SSb (56104)	192

C8H5O2F3S		HL	TTA				CAS	326-91-0	(165)	
4,4,4-Trifluoro-1-(2-thienyl)butane-1,3-dione; F ₃ C.CO.CH ₂ .CO.C ₄ H ₃ S										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	dis	oth/un	25°C	0.10M	U				1972KMe (58694)	193

$$K(VO(OH)+L)=14.91$$

C8H9N02 HL CAS 5330-97-2 (6248)
 Phenylacetohydroxamic acid; C6H5.CH2.CO.NH.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	25°C	0.02M	C				2002BPa (60358)	194
								K1eff=2.64		

Medium: 0.02 M MOPS, pH 7.50.

C8H9N03 HL CAS 10507-69-4 (8584)
 4-Methoxybenzohydroxamic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	25°C	0.02M	C				2002BPa (60430)	195
								K1eff=2.99		
								K2eff=2.08		

Medium: 0.02 M MOPS, pH 7.50. By 1H nmr, K1eff=2.99, K2eff<2.

C8H12N403 HL Gly-His CAS 3486-76-8 (273)
 Glycyl-histidine; H2N.CH2.CO.NH.CH(CH2.C3H3N2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	KCl	21°C	1.0M	M				1995TJb (61595)	196
								K1eff=2.0		
								K(H2V04+HL=H2V04L+H)=-4.96		
								*K(H2V04L)=-7.0		

Additional methods: 1H, 13C and 51V nmr. K1eff at pH 7.0.

C8H12N403 HL His-Gly CAS 2578-58-7 (274)
 Histidyl-glycine; H2N.CH(CH2.C3H3N2).CO.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	KCl	21°C	1.0M	M				1995TJb (61630)	197
								K1eff=1.83		
								K(H2V04+HL=H2V04L+H)=-4.70		
								*K(H2V04L)=-6.7		

Additional methods: 1H, 13C and 51V nmr. K1eff at pH 7.0.

C8H14N203 HL Pro-Ala CAS 6422-36-2 (263)
 Proyl-alanine; C4H8N.CO.NH.CH(CH3).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaCl	25°C	0.60M	C				1997FEa (61930)	198
								B(0,1,1)=2.44		

$$\begin{aligned}B(0,2,2) &= 6.98 \\B(0,1,2) &= 4.08 \\B(0,2,1) &= 5.23\end{aligned}$$

$B(p,q,r): pH+qH_2V_04+rHL=Hp(H_2V_04)q(HL)r$

C8H19N02 L CAS 102-79-4 (3841)

N-Butyl-2,2'-iminodiethanol (butyldiethanolamine);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	RT	0.40M	C				1998CBC (63035)	199

$$K(H_2V_04+L=V_02H-2L+2H_2O)=2.64$$

Method: 51V and 1H nmr.

C9H7N3O2S H2L TAR CAS 2246-46-0 (707)

4-(2'-Thiazolylazo)-resorcinol; C3H2NS.N:N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	oth	oth/un	?	?	U			K1=13.15	1969MGB (64736)	200

$$K(V_02+HL)=10.6$$

Metal: V02+

V(V)	sp	alc/w	25°C	50%	U				1967NPB (64737)	201
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$$K(V_03+H_2L)=12.5(?)$$

Medium: 50% MeOH, 0.1 M NaClO4

C9H11N03 H2L Tyrosine CAS 60-18-4 (4)

2-Amino-3-(4-hydroxyphenyl)propanoic acid; HO.C6H4.CH2.CH(NH2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	NaClO4	25°C	0.10M	C	I			2004GNA (66243)	202

$$K(V_02+H+HL)=7.45$$

$$K(V_02+2H+2HL)=5.57$$

Data for 0.3-1.0 M NaClO4. At I=1.0 M, K(V02+H+HL)=7.82,

K(V02+2H+2HL)=6.85.

C9H11N04	H3L	DOPA	CAS 59-92-7 (5)
2-Amino-3-(3,4-dihydroxyphenyl)propanoic acid; H2NCH(CH2C6H3(OH)2)COOH			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	kin	NaClO4	25°C	1.00M	M				1974KLA (66403)	203

$$K_{eff}(V_02+H_2L)=2.31$$

Medium: 0.2 - 1 M HClO4

C9H11N030	L	CAS 4813-04-1 (4646)
Acetone isonicotinylhydrazone; C5H4N.CONHN:C(CH3)2		

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
<hr/>										
V(V)	oth	oth/un	20°C	0.10M	U			K1=3.63 B2=6.98	1973Z0a (66473)	204
Metal: V02+										

C9H12N206		HL		Uridine			CAS	58-96-8 (828)		
Uracil-1-beta-D-ribofuranoside;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	25°C	0.40M	C			K1=1.41 K(2H2V04+2L=(H2V04)2L2)=7.72	1991CHa (66714)	205
<hr/>										
V(V)	nmr	KCl	25°C	0.10M	U			Beff(2V04+2L)=7.45 pH 7.5	1988TGc (66715)	206

C9H13N03		H2L		(-)Adrenaline			CAS	51-43-4 (252)		
4-(1-Hydroxy-2-(methylamino)ethyl)-1,2-dihydroxybenzene,										
Epinephrine; CH3NHCH(OH)C6H3(OH)2										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	kin	NaClO4	25°C	1.00M	M				1974KLa (66869)	207
<hr/>										
Medium: 0.2 - 1 M HClO4										

C9H13N305		L		Cytidine			CAS	65-46-3 (2152)		
Cytidine, Cytosine-1-beta-D-ribofuranoside;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	25°C	0.40M	C			K1=1.26 K(2H2V04+2L=(H2V04)2L2)=7.18	1991CHa (67085)	208

C9H14N403		HL		Ala-His			CAS	3253-17-6 (5767)		
Alanyl-histidine;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	NaCl	25°C	0.15M	C				2001SAb (67341)	209
<hr/>										
Method: 51V nmr. B(p,q,r): pH+qH2V04+rHL=Hp(H2V04)q(HL)r										
<hr/>										
V(V)	gl	NaCl	25°C	0.15M	C				2000SAC (67342)	210
<hr/>										
Additional method: 51V nmr. B(p,q,r): pH+qH2V04+rHL=(H)p(H2V04)q(HL)r										
<hr/>										
V(V)	oth	NaCl	25°C	0.60M	C				1994EFb (67343)	211

$$B(0,1,1)=2.55$$

$$B(1,1,1)=9.44$$

$B(p,q,r)$; $pH+q(H_2V_04)+rHL=H_p(H_2V_04)q(HL)r$

Method: potentiometric titrations (glass electrode) + 51V-NMR

C9H14N4O4 HL His-Ser CAS 21438-60-8 (7466)

Histidyl-serine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V)	gl	KCl	21°C	1.0M	M				1995TJb (67348)	212
								K1eff=2.50		
								K(H ₂ V ₀₄ +HL=H ₂ V ₀₄ L+H)=-4.82		
								*K(H ₂ V ₀₄ L)=-7.5		

Additional methods: 1H, 13C and 51V nmr. K_{1eff} at pH 7.0.

C9H21N03 L CAS 122-20-3 (946)

Tri-isopropanolamine; (CH₃.CH(OH).CH₂)₃N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V)	nmr	KCl	RT	0.40M	C				1998CBc (68143)	213
								K(H ₂ V ₀₄ +L=V ₀₂ H-2L+2H ₂ O)=2.98		

Method: 51V and 1H nmr.

C9H28N3015P5 10L DTPPH CAS 15827-60-8 (2921)

Diethylenetriamine-N,N,N',N'',N''-penta(methylphosphonic acid);
H₂O₃PCH₂.N(CH₂CH₂.N(CH₂P₀3H₂)₂)₂ H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V)	sp	oth/un	20°C	1.00M	U				1982SPb (68416)	214
								K(V ₀₂ +H ₆ L)=9.3		

C10H1002 HL Benzoylacetone CAS 93-91-4 (197)

1-Phenylbutane-1,3-dione; C₆H₅.CO.CH₂.CO.CH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V)	dis	oth/un	25°C	0.10M	U				1972KMe (70782)	215
								K(V ₀ (OH)+2L)=21.62		

Metal: V₀+++

C10H12N2O4 H₂L CAS 16598-05-3 (967)

2-Pyridylmethyliminodiethanoic acid; C₅H₄N.CH₂.N(CH₂.COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V)	nmr	KCl	RT	0.40M	U				1997CKb (71279)	216
								K(H ₂ V ₀₄ +L)=3.89		

Medium: 20% D20

C10H13N5O4 L Adenosine CAS 58-61-7 (2154)

Adenosine, Adenine-9-beta-D-ribofuranoside;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) gl NaCl 25°C 0.60M C 1997ECa (71955) 217

B(1,2,2,0)=11.89

B(0,2,2,0)=7.68

Additional method: 51V nmr. B(p,q,r,s): pH+qH2V04+rHL+sA= Hp(H2V04)q(HL)r(A)s. A is imidazole.

V(V) gl NaCl 25°C 0.60M C M 1997ECa (71956) 218

B(0,1,1,1)=3.04

B(0,1,2,1)=4.84

B(0,1,1,2)=3.60

B(0,1,2,2)=5.43

B(p,q,r,s): pH+qH2V04+rHL+sA=Hp(H2V04)q(HL)r(A)s. A: imidazole.

B(0,2,2,2)=8.68, B(0,2,2,1)=8.08, B(0,2,1,2)=6.91, B(0,2,1,1)=6.30.

V(V) nmr KCl 25°C 0.40M C K1=1.41 1991CHa (71957) 219

K(2H2V04+2L=(H2V04)2L2)=7.82

C10H13N5O5 HL Guanosine CAS 118-00-3 (1402)

2-Aminopurin-6-one-9-riboside;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) nmr KCl 25°C 0.40M C K1=1.08 1991CHa (72024) 220

K(2H2V04+2L=(H2V04)2L2)=7.36

C10H14N5O7P H2L AMP-5 CAS 18422-05-4 (842)

Adenosine-5'-monophosphoric acid, 5-Adenylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) nmr KCl 25°C 0.10M U 1988TGc (72500) 221

Beff(2V04+2L)=6.53 pH 7.5

C10H16N2O2 L (7408)

N-(2-Pyridylmethyl)iminodiethanol; C5H4N.CH2.N(CH2CH2.OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) nmr KCl RT 0.40M U 1997CKb (73036) 222

K(H2V04+L)=3.59

Medium: 20% D20.

C10H16N2O8 H4L EDTA CAS 60-00-4 (120)

1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	kin	NaClO4	25°C	3.00M	U	T			1984LAa (74285)	223
								$K(VO_2H_2L+H) > 0.0$		
Data at 10 to 40C										
V(V)	sp	NaClO4	25°C	3.00M	C				1978LLa (74286)	224
								$K(VO_2+H_2L)=6.9$		
								$K(VO_2+HL)=10.6$		
								$K(VO_2+L)=15.5$		

V(V) sp KCl RT 0.10M C K1=17.38 1975IYb (74287) 225
Method: competition with PAR.

V(V) g1 NaClO4 25°C 3.0M U 1972LLb (74288) 226
 $K(V_02 + H4L \rightleftharpoons V_02H2L + 2H) = 2.22$

V(V) sp oth/un 18°C 0.02M U 1971PLb (74289) 227
 $K(V_02+HL+H02)=24.44$

V(V) sp KC1 20°C 0.10M U T 1965PSa (74290) 228
 $K(V_02+L)=15.55$
 $K(V_02+HL)=9.60$
 $K(V_02+H2L)=6.93$
 $K(V_02+H3L)=5.6$

V(V) sp NaClO₄ 25°C 0.10M U K1=18.05 1957RSa (74292) 230

C11H9N03 H2L

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V(V) sp oth/un ? ? U 1969M0b (77356) 231
K(V02+L)=3.58
K(V02L+L)=3.20

C11H9N3O2 H2L PAR CAS 1141-59-9 (636)

4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C₁₀H₁₀N₂O₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) sp KCl RT 0.10M C 1975IYb (77597) 232
K(V02+HR)=17.16
K(V02R+H)=3.95

V(V) sp oth/un ? ? U M K1=17.10 1973LPa (77598) 233
 $K(VO_2(NH_2OH)+HL)=13.52$
 $K(VO_2+NH_2OH+L)=24.85$

V(V) sp oth/un 25°C ? U 1967ADa (77599) 234
 $K(?)=4.2$

V(V) sp oth/un 15°C 0.01M U 1966BVb (77600) 235
 $K(VO_2+L)=16.49$

C11H12N20 L Antipyrine CAS 60-80-0 (2026)

2,3-Dimethyl-1-phenyl-3-pyrazolin-5-one, Phenazone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	?	?	U	M			1969KSb (78006) 236	
								$K(VO_2(H_2A)2+3HL)=1.19$		

H4A=2,3,4-trihydroxybenzoic acid

C11H14N204 H2L Gly-Tyr CAS 658-79-5 (533)
Glycyl-tyrosine; H2N.CH2.CO.NH.CH(CH2.C6H4.OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	22°C	1.0M	M	M			1993JTa (78860) 237	
								$K(V(\text{peroxo})+H_2L=VHL+H)=-5.15$		

V is peroxovanadate. Method:NMR. Coordination site is phenolate.

V(V)	nmr	KCl	22°C	1.0M	M	M			1993JTa (78861) 238	
								$K(V(\text{diperoxo})+HL=VL+H)=-5.82$		

V is diperoxovanadate. Method:NMR. Coordination site is amino.

For phenolate coordination $K=-6.08$

C11H14N204 H2L Tyr-Gly CAS 673-08-5 (532)
Tyrosyl-glycine; H2N.CH(CH2.C6H4.OH).CO.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	KCl	21°C	1.0M	M				1995TJb (78919) 239	
								$K_{\text{eff}}=1.53$		

Method: 1H, 13C and 51V nmr. K_{eff} at pH 7.0.

Also data for HL=Trp-Tyr, Trp-Trp, Trp-Phe, Tyr-Tyr, Phe-Glu.

C11H16N205 HL CAS 213412-33-0 (8754)
N-(2-Hydroxy-5-nitrobenzyl)iminodiethanol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	RT	0.40M	C				1998CBc (79097) 240	

$$K(H_2V_04+L=V_02H-2L+2H_2O)=2.56$$

Method: 51V and 1H nmr.

C11H21N305 H2L CAS 499238-77-6 (8837)
N-Hydroxy-N'-[4-(hydroxymethylamino)-4-oxobutyl]-N-methylpentanediamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	KCl	25°C	0.20M	C				2002FBb (79797) 241	
								K(HV04+L+5H=VOL+3H2O)=37.73		
								K(HV04+L+4H=V02HL+2H2O)=36.11		
								K(HV04+L+3H=V02L+2H2O)=31.69		

By spectrophotometry, $K(2HV_04+3L+14H=V_2L_3+8H_2O)=100.9$.

C12H13N304 H2L (7410)
N-(Benzimidazol-2-ylmethyl)iminodiethanoic acid; C7H5N2.CH2.N(CH2COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	RT	0.40M	U				1997CKb (81310) 242	
								$K(H_2V_04+L)=2.99$		

Medium: 20% D2O.

C12H17N302 L (7409)
N-(Benzimidazol-2-ylmethyl)iminodiethanol; C7H5N2.CH2.N(CH2CH2.OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	nmr	KCl	RT	0.40M	U				1997CKb (81726) 243	
								$K(H_2V_04+L)=1.83$		

Medium: 20% D2O

C12H23N305 H2L CAS 499238-78-7 (8836)
N-Hydroxy-N'-[5-(hydroxymethylamino)-5-oxopentyl]-N-methylpentanediamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	KCl	25°C	0.20M	C				2002FBb (82987) 244	
								$K(HV_04+L+5H=VOL+3H_2O)=36.61$		
								$K(HV_04+L+4H=V_02HL+2H_2O)=35.78$		
								$K(HV_04+L+3H=V_02L+2H_2O)=31.39$		

By spectrophotometry, $K(2HV_04+3L+14H=V_2L_3+8H_2O)=99.6$.

C12H23N305 H2L CAS 499238-79-8 (8835)
N-Hydroxy-N'-[6-(hydroxymethylamino)-6-oxohexyl]-N-methylbutanediamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	gl	KCl	25°C	0.20M	C				2002FBb (82997) 245	
								$K(HV_04+L+5H=VOL+3H_2O)=38.56$		

$$K(HV04+L+4H=V02HL+2H2O)=37.33$$

$$K(HV04+L+3H=V02L+2H2O)=33.01$$

By spectrophotometry: $K(2HV04+3L+14H=V2L3+8H2O)=103.7$

C13H10NO2Cl HL CAS 36016-24-7 (1818)

N-(4-Chlorophenyl)benzohydroxamic acid; C6H5.CO.N(C6H4Cl)OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	28°C	?	U				1965MDa (84720)	246

$$K(V03+L=V02(H2L))=7.05(?)$$

C13H11N3O2 HL (4985)

2- or 4-Hydroxybenzaldehyde isonicotinylhydrazone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	oth	oth/un	20°C	0.10M	U				1973Z0a (85272)	247

$$K(V02+L)=4.13$$

C13H11N3O6S H2L (2811)

1-(2-Carboxy-5-sulfonatophenyl)-3-hydroxy-phenyltriazene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	none	25°C	0.0	U			K1=4.7	1974CHa (85305)	248

C13H15N3O3 HL Gly-Trp CAS 2390-74-1 (3411)
Glycyltryptophan;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	KCl	21°C	1.0M	M				1995TJb (85899)	249

$$K_{\text{eff}}=2.16$$

Method: 1H, 13C and 51V nmr. K_{eff} at pH 7.0.

C13H15N3O3 HL Trp-Gly CAS 7360-09-0 (5804)
Tryptophyl-Glycine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	KCl	21°C	1.0M	M				1995TJb (85902)	250

$$K_{\text{eff}}=2.00$$

Method: 1H, 13C and 51V nmr. K_{eff} at pH 7.0.

C13H21N02 L CAS 213412-34-1 (8755)
N-2,5-(Dimethylbenzyl)iminodiethanol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V(V) nmr KCl RT 0.40M C 1998CBc (86159) 251
 $K(H_2V_04+L=V_02H-2L+2H_2O)=1.34$

Method: 51V and 1H nmr.

C14H8N4O4Cl2S H2L (6672)
7-((3,5-Dichloro-2-pyridyl)azo)-8-hydroxyquinoline-5-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	vlt	KNO ₃	25°C	0.10M	U				1993HKa (86622) 252	

$$K(V_0(OH)_2(H_2O)+L)=16.35$$

For 3,5-dibromo analogue K=16.38

C14H8O7S H3L DASA CAS 83-61-4 (950)
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	NaClO ₄	25°C	0.10M	U				1962SDa (86767) 253	

$$K(V_03+H_2L=V_02L)=8.5(?)$$

V(V)	sp	oth/un	25°C	?	U				1961BDc (86768) 254	
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$$K(V_03+H_2L=V_02L)=8.6(?)$$

V(V)	sp	oth/un	25°C	?	U			B2=8.6	1959DBb (86769) 255	
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C14H12N4O2Br₂ HL CAS 72833-87-5 (2533)
2-(2-(3,5-Dibromopyridyl)azo)-5-dimethylaminobenzoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	diox/w	25°C	40%	C			K1=8.97	1986KHa (87320) 256	

C14H13N02 HL CAS 1143-74-2 (4044)
N-2-Tolylbenzohydroxamic acid; C₆H₅.CO.N(C₆H₄.CH₃).OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	oth/un	28°C	?	U				1964MDa (87484) 257	

$$K(V_03+2HL=VOL2)=8.1(?)$$

C14H13N5OS HL (5394)
1-(2-Pyridylmethylideneamino)-3-(salicylideneamino)thiourea;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	mixed	25°C	40%	U				1985RGa (87618) 258	

$$K_{1eff}=6.09$$

Medium: 40% DMF, pH 4.5

C14H13N5O2	HL	(5393)								
1-(2-Pyridylmethylideneamino)-3-(salicylideneamino)urea;										
<hr/>										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	mixed	25°C	32%	U				1985RGa (87625)	259
			K _{1eff} =4.68							
Medium: 32% DMF, pH 4.5			<hr/>							
C14H14N40Br2	HL						CAS	35601-32-2	(5092)	
5-(3,5-Dibromo-2-pyridylazo)-2-ethylamino-4-hydroxy-1-methylbenzene;			<hr/>							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	dis	oth/un	?	?	U				1967GUa (87690)	260
			K(?)=7.52							
<hr/>			<hr/>							
C14H15N40Br	HL						CAS	14337-50-9	(5095)	
5-(5-Bromo-2-pyridylazo)-2-ethylamino-4-hydroxy-1-methylbenzene;			<hr/>							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	dis	oth/un	?	?	U				1967GUa (87771)	261
			K(?)=5.59							
<hr/>			<hr/>							
C14H16N40	HL	PAAC					CAS	13059-69-3	(5067)	
5-Ethylamino-4-methyl-2-(2'-pyridylazo)phenol;			<hr/>							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	dis	oth/un	?	?	U				1967GSb (88022)	262
			K(?)=6.73							
<hr/>			<hr/>							
C14H22N208	H4L	CDTA					CAS	482-54-2	(200)	
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;			<hr/>							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	none	25°C	0.0	U		K ₁ =16.61		1975IYa (88814)	263
			K(VO ₂ +HL)=8.91							
<hr/>			<hr/>							
V(V)	sp	KCl	RT	0.10M	C		K ₁ =16.59		1975IYb (88815)	264
Method: competition with PAR.			<hr/>							
<hr/>			<hr/>							
C14H24N2010	EGTA						CAS	67-42-5	(349)	
Ethleneglycol-0,0'-bis(2-aminoethyl ether)-N,N,N',N'-tetraethanoic acid; H4L			<hr/>							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V(V)	sp	NaClO ₄	22°C	0.10M	C		K ₁ =11.18		1980DCc (89960)	265

$$K(VO_2+HL)=7.73$$

$$*K(VO_2HL)=-5.49$$

C15H9N3O4Cl2S H2L (6673)

1-((3,5-Dichloro-2-pyridyl)azo)-2-hydroxynaphthalene-4-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) vlt KN03 25°C 0.10M U 1993HKa (90937) 266

B(VO(OH)2+L=VO(OH)2L)= 21.91. For 3,5-dibromo analogue K=21.96;

For 3,5-dichloro...-1-hydroxynaphthalene analogue K=16.12, -dibromo- K=15.97

C15H1007 H5L Morin CAS 104363-16-8 (5100)

2',3,4',5,7-Pentahydroxyflavone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) sp non-aq ? 100% U 1972CMb (91006) 267

$$K(VO_2+H4L)=(?)4.21$$

$$K(VO_2+2H4L)=(?)7.95$$

Medium: MeOH. pH=3

C15H11N3O4S H2L (5130)

7-Phenylazo-8-hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) vlt KN03 25°C 0.10M U 1993HKa (91338) 268

B(VO(OH)2(H2O)+L=VO(OH)2LH2O)= 17.07. For 4-chlorophenyl analogue K=16.86

C15H16N4OBr2 HL CAS 14337-54-3 (993)

2-(3,5-Dibromo-2-pyridylazo)-5-diethylaminophenol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) vlt oth/un 25°C ? U 1990WZa (91944) 269

$$K(VO_2+HL)=10.40$$

C15H17N4OBr HL CAS 14357-53-2 (712)

2-(5-Bromo-2-pyridylazo)-5-diethylaminophenol; BrC5H3N.N:N.C6H3(OH)N(CH3)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) vlt oth/un 25°C 0.60M U M 1989WZb (91983) 270

$$K(V+L+H2O2=VL(H2O2))=10.3$$

C16H14N4O2S HL CAS 83688-78-2 (2534)

2-(2-Benzothiazolylazo)-5-dimethylaminobenzoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) sp diox/w 25°C 40% C K1=6.62 1986KHa (93484) 271

C16H19N02 L CAS 157008-41-8 (8758)
meso-2,2'-Diphenyliminodiethanol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) nmr KCl RT 0.40M C H 1998CBc (93911) 272
 $K(H_2V_04+L=V_02H-2L+2H_2O)=2.18$

Method: 51V and 1H nmr. L is R,S-stereoisomer. DH(K)=-26.8 kJ mol⁻¹, DS(K)=-50.4 J K⁻¹ mol⁻¹. For R,R-isomer, K=1.79, DH(K)=-30.6, DS=-83.6.

C16H20N206 H2L CAS 488827-72-1 (8831)
N,N'-Bis(3-hydroxy-6-methyl-2-methylene-4-pyrone)ethylenediamine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) gl NaCl 25°C 0.16M C K1=17.10 2002SSb (94008) 273
 $K(V_0+HL)=13.29$
 $K(V_0+H_2L)=8.75$
 $K(V_0+H_3L)=5.3$
 $*K(VOL)=-8.75$

C19H19N07 H2L (7003)
3-Methoxy-5-(N,N-dicarboxymethyl)aminomethyl-4-hydroxybenzophenone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) gl KCl 20°C 0.10M U K1=14.0 1981SYa (99257) 274
 $K(V_02+HL)=7.7$

C22H14N405S H3L CAS 74261-72-6 (9033)
5-Hydroxy-6-(2-hydroxy-5-sulfophenylazo)benzo[a]phenazine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V(V) sp KCl 25°C 0.02M C K1=13.05 B2=21.39 1980NKb (101481) 275
 $K(V_02+H_2L=V_02L+2H)=-1.29$
 $K(V_02(OH)_2+H_2L=V_02L+2H_2O)=5.4$

V(V) sp KCl 25°C 1.0M C 1980NKc (101482) 276
 $K(V_02+H_2L=V_02L+2H)=-1.29$
 $K(V_02(OH)_2+H_2L=V_02L+2H_2O)=5.46$

For protonation reactions, K1=13.05, K2=8.34 (O Navratil, Radiokhimiya, 19, 626 (1977)). Data for 0.1 and 0.01 M Septonex solutions.

C25H28N2013 H6L CAS 42281-29-8 (5335)
(Carbonylbis((6-hydroxy-5-methoxy-3-phenylene)methylenenitrilo))tetraethanoic acid;

CN- HL Cyanide CAS 74-90-8 (230)
Cyanide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V++	kin	oth/un	25°C	1.00M	U	I M			1969DEa (2772)	283

$$\begin{aligned} K(V+Co(III)(CN)5(SCN)) &= 1.5 \\ K(V+Co(III)(CN)5Cl) &= 1.6 \\ K(V+Co(III)(CN)5I) &= 1.3 \\ K(V+Co(III)(CN)5H_2O) &= 1.7 \end{aligned}$$

$$I=0.2: K(V+Co(CN)5Br)=2.0$$

V++	cal	oth/un	25°C	var	U	H			1964GHc (2773)	284
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V++	cal	oth/un	25°C	?	U	H			1961GUa (2774)	285
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$$DH(B6)=-196.5 \text{ kJ mol}^{-1}$$

HP03-- H2L Phosphite CAS 13598-36-2 (6305)
Phosphite;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V++	gl	oth/un	25°C	0.0	U		K1=4.61		1966POa (7515)	286
							B(V+HL)=2.50			
							B(VHL+HL)=1.67			
							B(VL+HL)=2.01			

OH- HL Hydroxide (57)
Hydroxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V++	EMF	oth/un	?	var	U	H			1964PPa (12456)	287
							*K1=-6.85(15 C)			
							*K1=-6.49(25 C)			
							*K1=-6.10(35 C)			

Medium: VS04 in various concentrations. DH(*K1)=64.8 kJ mol⁻¹

SCN- HL Thiocyanate CAS 463-56-9 (106)
Thiocyanate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V++	kin	NaClO4	24°C	1.0M	U		K1=1.04		1968KTc (15323)	288
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Method: Temperature jump

V++	sp	NaClO4	45°C	0.84M	U T H	T	K1=1.18		1968MSc (15324)	289
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Medium: 0.84 M LiClO4, 0.05 H+. K1(11 C)=1.61, K1(25 C)=1.43

DH(K1)=-21.7 kJ mol⁻¹, DS=-46 J K⁻¹ mol⁻¹

V++ sp NaClO₄ 25°C 1.0M U K1=1.43 19680Pa (15325) 290

C4H4O4 H2L Maleic acid CAS 110-16-7 (111)
cis-Butenedioic acid; HOOC.CH:CH.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V++	sp	oth/un	23°C	var	U				1970VSa (29155)	291
								K(V+H2L)=-0.68		
								K(V+HL)=1.81		

C4H6O4S H2L Thiodiacetic acid CAS 123-93-3 (140)
2,2'-Thiodiglycolic acid, Thiodiethanoic acid; HOOC.CH₂.S.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V++	gl	NaClO ₄	25°C	0.10M	U			K1=1.73	1970PPa (30242)	292

C5H8O2 HL Acetylacetone CAS 123-54-6 (164)
Pentane-2,4-dione; CH₃.CO.CH₂.CO.CH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V++	gl	KCl	25°C	1.00M	U			K1=5.38 B2=10.19 B3=14.70	1965SCd (38118)	293

C6H5NO₂ HL Picolinic acid CAS 98-98-6 (391)
2-Pyridine-carboxylic acid; C5H₄N.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V++	sp	KCl	25°C	0.50M	U	I		K1=4.43 B2=9.00 B3=12.84	1965MBb (42615)	294

In 0.5 M K₂SO₄: B3=12.77. By polarography: B2=8.54, B3=12.46

C6H₉NO₆ H3L NTA CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH₂.COOH)₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V++	vlt	KCl	22°C	1.0M	C			K1=3.46	1988MWb (47085)	295

Method: cyclic voltammetry and differential pulse polarography.

C6H10O4S₂ H2L CAS 7244-02-2 (438)
1,2-Bis(carboxymethylthio)ethane; HOOC.CH₂.S.CH₂.CH₂.S.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V++	gl	NaClO ₄	25°C	0.10M	U			K1=1.39	1971PPb (48252)	296

C8H13N06S H3L (5675)
2-Mercapto-1-aminoethane-N,N,S-triethanoic acid; HOOC.CH2.S.CH2.CH2.N(CH2COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V++ gl NaClO4 25°C 0.10M U K1=7.21 1975POa (61833) 297
K(V+HL)=1.7

C8H1404S3 H2L (2526)
3,6,9-Trithiaundecanedioic acid; HOOC.CH2.S.C2H4.S.C2H4.S.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V++ gl NaClO4 25°C 0.10M U K1=1.2 1971PPc (62127) 298

C10H16N208 H4L EDTA CAS 60-00-4 (120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V++ vlt NaCl 25°C 1.3M C 1983KNb (74293) 299
K(VL+H)=3.40
*K(VL(H2O))=-12.6

Method: dc and ac polarography.

V++ gl KCl 20°C 0.10M U K1=12.70 1953SSa (74294) 300

C12H8N2 L Phenanthroline CAS 66-71-7 (144)
1,10-Phenanthroline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V++ sp NaCl ? 0.10M U 1972BBh (80528) 301
B3=19.68
K(V20+L)=6.60

C24H16N2 L Bathophenan CAS 1662-01-7 (2749)
4,7-Diphenyl-1,10-phenanthroline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V++ sp NaCl ? 0.10M U 1973BBh (102860) 302
B3=15.65
K(V20+L)=6.44

e- HL Electron (442)
Electron;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ EMF NaCl 25°C 1.00M U 1993FOa (1024) 303
E(e + V+++)=-0.242V

Method: mercury electrode. At I=0.0, E=-0.196V.

V+++ EMF oth/un 17°C 1.0M U I 1962AND (1025) 304
K(V+e=V(II))=-5.35(-308 mV)

Medium:H₂SO₄. In 1 M HCl: K=-4.92(-283 mV), 1 M HClO₄: K=-4.79(-276 mv)

V+++ EMF none 25°C 0.0 U T 1944JCb (1026) 305
K(V+e=V(II))=-4.31(-255 mV)

At 0 C: K=-4.89(-265 mV)

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ sp alc/w 20°C 100% U M 1987KAa (5939) 306
B(V2(1,2-diazole)6C15)=2.4

V+++ sp alc/w 20°C 100% U M 1987KAa (5940) 307
B(V2(imidazole)6C15)=3.2

V+++ sp alc/w 20°C 100% U M 1987KAa (5941) 308
B(V2(thiazole)6C15)=3.4

V+++ sp alc/w 22°C 100% U M 1983KAa (5942) 309
B(V(a-pic)2C14)=3.9
B(V(b-pic)4C12)=4.7
B(V(g-pic)4C12)=4.7

Measurements in 99.8% ethyl alcohol; a-, b-, g-pic are alpha-, beta-, gamma-picoline (2-, 3-, 4-methylazine) respectively

V+++ vlt NaCl 25°C 5.00M U K1=-0.46 1981JGa (5943) 310

F- HL Fluoride CAS 7644-39-3 (201)
Fluoride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ ISE non-aq 185°C 100% M K1=4.56 B2=8.58 1988JHa (7324) 311
B3=12.15
B4=15.53
B5=18.04

Medium: molten KSCN. K1=mol-1 kg, B2=mol-2 kg2 etc.

V+++ ISE NaClO₄ 25°C 1.00M U K1=5.00 1980HMa (7325) 312

NO₃- HL Nitrate CAS 7697-37-2 (288)

*B(2,3)=-8.7

V+++ gl none 23°C 0.0 U 1963PAc (12464) 322
*K1=-2.4
*K2=-3.85

V+++ gl NaCl 20°C 3.00M U I 1963PAc (12465) 323
*K1=-3.15
*B2=-7.3
*B(2,2)=-4.1

In 1 M NaCl: *K1=-2.85, *B2=-6.7, *B(2,2)=-3.9

V+++ gl KCl 22°C 1.0M U 1960JPa (12466) 324
*K1=-2.57
*K2=-3.70

V+++ gl NaClO4 25°C 1.0M U 1959GSa (12467) 325
*K1=-2.53

V+++ gl oth/un 25°C var U 1953MEa (12468) 326
*K1=-2.9
*K2=-3.5

V+++ gl oth/un 25°C var U 1950FDa (12469) 327
*K1=-2.7

V+++ sp NaClO4 25°C 1.0M U H 1950FGa (12470) 328
DH(*K1)=42 kJ mol-1

SCN- HL Thiocyanate CAS 463-56-9 (106)
Thiocyanate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ sp non-aq 25°C 100% U H K1=2.22 1992DSb (15326) 329
In DMSO, I=0.5M Natriflate. DH1=2.9 kJ mol-1, DS1=52.1 J K-1 mol=1

V+++ sp oth/un 25°C ? U K1=2.20 1989SDc (15327) 330
isothiocyanate complex

V+++ kin NaClO4 23°C 1.0M U K1=2.18 1968KTc (15328) 331
Method: Temperature jump

V+++ sp NaClO4 37°C 1.0M U T H T K1=1.94 1967BSa (15329) 332
K1(5 C)=2.16, K1(12 C)=2.14, K1(25 C)=2.07. DH(K1)=-14.6 kJ mol-1, DS=-9.2

V+++ sp alc/w 24°C 100% U 1963GKb (15330) 333
B6=15 to 16

V+++ sp NaClO4 25°C 2.60M U H T K1=2.0 1951FGa (15331) 334

DH(K1)=-15.1 kJ mol-1.

S04-- H2L Sulfate CAS 7664-93-9 (15)
Sulfate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ sp NaClO4 25°C 1.0M U K1=1.45 1972KMD (16660) 335

C2H3O2Cl HL Chloroacetic CAS 79-11-8 (34)
Chloroethanoic acid; ClCH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 22°C 1.0M U 1960JPa (19393) 336
K(V+2VOH+6L)=18.52

C2H5NO2 HL Glycine CAS 56-40-6 (85)
2-Aminoethanoic acid; H2N.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ sp oth/un ? ? U K1=9.08 1970PKb (21753) 337

C3H3NS L Thiazole CAS 288-47-1 (382)
Thiazole; cyclo(-S.CH:N.CH:CH-) C3H3NS

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ sp alc/w 20°C 100% U M 1987KAa (23529) 338
B(V2L6C15)=3.4

C3H4N2 L Imidazole CAS 288-32-4 (90)
1,3-Diazole, imidazole; C3H4N2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ sp alc/w 20°C 100% U M 1987KAa (23931) 339
B(V2L6C15)=3.2

C3H7NO2 HL Alanine CAS 56-41-7 (86)
2-Aminopropanoic acid; H2N.CH(CH₃).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl NaCl 20°C 0.50M C K1=8.44 B2=15.55 2003BPa (26293) 340
B(VHL)=12.08
B(VH2L2)=22.02
B(VHL2)=19.02
B(VH-2L2)=-1.48

C3H7NO3 HL Serine CAS 56-45-1 (49)
2-Amino-3-hydroxypropanoic acid; H2N.CH(CH2.OH)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 25°C 0.20M C K1=8.75 B2=15.35 1986KDa (27195) 341

C4H7NO4 H2L Aspartic acid CAS 56-84-8 (21)
Aminobutanedioic acid; H2N.CH(CH2.COONa).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl NaCl 20°C 0.50M C K1=9.83 B2=16.86 2003BPa (31971) 342
B(VHL)=12.89
B(VH2L2)=24.52
B(VHL2)=20.42
B(V2H-2L2)=14.51

B(V2H-4L2)=-1.69, B(V2H-2L4)=20.32.

V+++ gl KCl 20°C 0.20M C K1=9.48 B2=17.16 1988KDa (31972) 343

C4H8N2O3 HL Asparagine CAS 70-47-3 (17)
2-Aminobutanedioic acid 4-amide; H2N.CH(CH2.CO.NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 20°C 0.20M C K1=8.08 B2=15.33 1988KDa (32742) 344

C4H9NO3 HL Threonine CAS 72-19-5 (48)
2-Amino-3-hydroxybutanoic acid; H2N.CH(CH(OH).CH3)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 25°C 0.20M C K1=8.36 B2=14.76 1986KDa (34338) 345

C5H5N5 L Adenine CAS 73-24-5 (237)
6-Aminopurine; H2N.C5H3N4

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl NaCl 25°C 0.50M C 2001BKa (36985) 346
K(V(OH)+HL=VL)=3.15
K(2V(OH)L=V20L+H2O)=4.86
K(V20(OH)+HL=V20L+H2O)=4.13

C5H8O2 HL Acetylacetone CAS 123-54-6 (164)
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ dis NaCl04 25°C 0.10M U K1=10.19 B2=19.18 1986ISb (38119) 347
B3=26.10

C5H9N04 H2L Glutamic acid CAS 56-86-0 (22)
2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 20°C 0.20M C K1=9.35 B2=17.00 1988KDa (39142) 348

C5H10N203 HL Glutamine CAS 56-85-9 (18)
2-Aminopentanedioic acid 5-amide; H2N.CH(CH2.CH2.CO.NH2)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 20°C 0.20M C K1=8.20 B2=15.82 1988KDa (39845) 349

C6H5N02 HL Picolinic acid CAS 98-98-6 (391)
2-Pyridine-carboxylic acid; C5H4N.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 25°C 0.20M C K1=6.28 B2=12.16 2005BNa (42616) 350
B3=16.63
B(VH-1L)=3.87
B(VH-1L2)=9.06
B(VH-2L2)=1.1

V+++ sp KCl 25°C 0.50M U 1965MBb (42617) 351
B3=15.41

C6H6O3 HL Maltol CAS 118-71-8 (2442)
3-Hydroxy-2-methyl-4H-pyran-4-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 25°C 0.20M C K1=10.38 B2=18.97 2005BNa (44107) 352
B3=26.01
B(VH-1L)=6.39
B(VH-1L2)=13.1

C6H6O8S2 H4L Tiron CAS 149-45-1 (104)
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)2.C6H2(SO3H)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 25°C 0.20M C K1=18.65 B2=37.53 2005BNa (44511) 353
B3=43.10
B(VH-1L)=13.69

$$B(VH-1L2)=24.80$$

C6H7N L Picoline CAS 109-06-8 (320)
2-Methylpyridine; C5H4N.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ sp alc/w 22°C 100% U M 1983KAA (44617) 354
 $B(VL2C14)=3.9$

C6H7N L beta-Picoline CAS 108-99-6 (324)
3-Methylpyridine; C5H4N.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ sp alc/w 22°C 100% U M 1983KAA (44709) 355
 $B(VL4CL2)=4.7$

C6H7N L gamma-Picoline CAS 108-89-4 (325)
4-Methylpyridine; C5H4N.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ sp alc/w 22°C 100% U M 1983KAA (44835) 356
 $B((VL4C12)+)=4.7$

C6H9N06 H3L NTA CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH₂.COOH)₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ sp NaCl 20°C 1.0M U K1=16.9 1974PIa (47086) 357
V+++ vlt NaClO₄ 20°C 0.10M U T K1=13.41 B2=22.09 1970PHb (47087) 358

C6H9N3O2 HL Histidine CAS 71-00-1 (1)
2-Amino-3-(4'-imidazolyl)propanoic acid; H2N.CH(CH₂.C3H3N2)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 25°C 0.10M C K1=12.9 1997BKA (47626) 359
 $B(VHL)=15.3$
 $B(VH2L)=18.3$
 $B(VH2L2)=29.2$
 $B(VH4L2)=34.4$

C6H13N02 HL Leucine CAS 61-90-5 (47)
2-Amino-4-methylpentanoic acid; H2N.CH(CH₂.CH(CH₃)₂)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 25°C 0.20M C K1=8.96 B2=16.26 1986KDa (50122) 360

C7H5N05 H2L Nitrosalicylic CAS 96-97-9 (148)
2-Hydroxy-5-nitrobenzoic acid; HO.C6H3(NO2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ kin oth/un 25°C 1.00M U 1980PTa (53057) 361
K(V+HL=VL+H)=-3.05
K(V+HL)=2.93

C7H6O3 H2L Salicylic acid CAS 69-72-7 (14)
2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ kin NaClO4 27°C 1.00M U T 1979PTa (54330) 362
K(V+HL=VL+H)=2.3
K(V+HL)=3.3

C7H7N02 HL CAS 3222-47-7 (3154)
6-Methylpyridine-2-carboxylic acid; CH3.C5H3N.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 25°C 0.20M C K1=5.98 2005BNa (55433) 363
B(VH-1L)=2.67
B(VH-1L2)=6.90

C7H9N02 HL CAS 30652-11-0 (2458)
3-Hydroxy-1,2-dimethylpyridin-4(1H)-one; (OH)(CH3)(O:)C5H2N.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ gl KCl 25°C 0.20M C K1=13.36 B2=24.45 2005BNa (56443) 364
B3=33.39
B(VH-1L)=9.10
B(VH-1L2)=18.39

C8H9N3O7 H2L Uramildiacetic CAS 13055-06-5 (185)
5-Amino-2,4,6-trioxo-1,3-perhydrodiazimino-N,N-diethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V+++ sp NaCl 20°C 1.0M U K1=13.51 1974PIa (60662) 365

C10H16N2O8 H4L EDTA CAS 60-00-4 (120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V+++	gl	NaClO4	25°C	1.00M	C				1983AHa (74295)	366
								K(VL+H)=0.22		

V+++	vlt	NaCl	25°C	1.3M	C				1983KNb (74296)	367
								K(VL+H)=2.54		

$$*K(VL(H2O))=-9.27$$

Method: dc and ac polarography.

V+++	gl	NaClO4	25°C	1.0M	M	I			1977KSc (74297)	368
								*K(VL(H2O))=-9.57		
								K(2VL(H2O)=LVOVL+2H)=-15.99		
								K(2VL(OH)=LVOVL+H2O)=3.14		

At I=0.20, *K(VL(H2O))=-10.16, K(2VL(H2O)=LVOVL+2H)=-16.71, K(2VL(OH)=LVOVL+H2O)=3.62.

V+++	sp	NaCl	25°C	0.10M	U		K1=0.67		1974TPa (74298)	369
V+++	EMF	KCl	20°C	0.10M	U	T	K1=25.9		1953SSa (74299)	370

$$K(VLOH+H)=9.54$$

C10H18N2O7 H3L HEDTA CAS 150-39-0 (392)
N-(Hydroxyethyl)diaminoethane-N,N',N'-triethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V+++	vlt	NaCl	25°C	1.3M	C	I			1983KNb (75531)	371
								K(2V(OH)L=V2(OH)2L2)=4.20		

$$*K(VL(H2O))=-6.52$$

Method: dc and ac polarography. By pH titration at I=0.6 M NaCl:
*K(VL(H2O))=-6.31, K(2V(OH)L=V2(OH)2L2)=4.05.

V+++	gl	NaClO4	25°C	1.0M	M	I			1977KSc (75532)	372
								*K(VL(H2O))=-6.59		
								K(2VL(H2O)=LVOVL+2H)=-9.16		

$$K(2VL(OH)=LVOVL+H2O)=4.01$$

At I=0.20, *K(VL(H2O))=-6.39, K(2VL(H2O)=LVOVL+2H)=-9.05, K(2VL(OH)=LVOVL+H2O)=3.74.

V+++	sp	NaCl	20°C	1.0M	U		K1=17.6		1974PIa (75533)	373

C11H9N3O2 H2L PAR CAS 1141-59-9 (636)
4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V+++	sp	oth/un	?	?	U				1970BBg (77601)	374
								K(VO+L)=15.75		

C14H23N3O10	H5L	DTPA	CAS 67-43-6 (238)		
Diethylenetriamine-pentaethanoic acid; HOOC.CH2.N(CH2.CH2.N(CH2.COOH)2)2					
<hr/>					
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference	ExptNo
V+++	sp	NaCl	25°C 0.10M U	K2=5.6	1974TPa (89430) 375
V+++	sp	oth/un	20°C ? U	K1=27.89	1970KAF (89431) 376
<hr/>				<hr/>	
C15H12N2O5	H2L		CAS 1562-85-2 (5111)		
Gallocyanine;				<hr/>	
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference	ExptNo
V+++	sp	oth/un	? ? U	K1=11.64 B2=17.30	1973TPb (91443) 377
By polarography: K1=11.04, B2=16.24				<hr/>	
<hr/>				<hr/>	
C20H16N4O5S	H2L	Eriochrome Red B	CAS 14954-75-7 (3510)		
4-(4,5-Dihydro-3-Me-5-oxo-1-Phe-1H-pyrazol-4-ylazo)-3-naphthol-1-sulfonic acid;				<hr/>	
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference	ExptNo
V+++	sp	oth/un	20°C 0.20M U	1972TPb (99799) 378	
K(V+HL)=6.96, pH=4-6				<hr/>	
<hr/>				<hr/>	
C23H18O9S	H4L	Eriochrome cyan	CAS 3564-18-9 (433)		
4'-Hydroxy-3,3'-dimethyl-2''-sulfofuchsone-5,5'-dicarboxylic acid;				<hr/>	
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference	ExptNo
V+++	sp	oth/un	? ? U	K1=5.04 B2=10.21	1973TPb (102639) 379
By polarography: K1=5.96, B2=10.26				<hr/>	
<hr/>				<hr/>	
C32H18N8O12S4	H4L		CAS 33308-41-7 (5367)		
Phthalocyanine tetrasulfonic acid;				<hr/>	
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference	ExptNo
V+++	sp	oth/un	? ? U	K1=4.48 B2=8.98	1971TPa (105561) 380
<hr/>				<hr/>	
e-	HL	Electron	(442)		
Electron;				<hr/>	
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference	ExptNo
V0++	kin	oth/un	25°C 1.00M U	1968SEa (1027) 381	
K' = 2.85				<hr/>	
K': V0++ + Cu(I) + 2H = V+++ + Cu(II) + H2O. I=3.0 M, K'=3.04				<hr/>	
<hr/>				<hr/>	
V0++	EMF	none	25°C 0.0 U T	1944JCa (1028) 382	

K=5.70(337 mV)

K: V0+2H+e=V+H2O. At 0 C: K=6.66(361 mV)

BrO3- HL Bromate (6017)
Bromate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ kin NaClO4 20°C 0.10M U K1=1.95 1969FOa (2439) 383
medium: (H,Li)ClO4

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ kin NaClO4 25°C 2.0M U K1=-0.63 1972MGc (5944) 384

V0++ sp KC1 rt var U B2=-0.75 1971KGa (5945) 385
K(VOC12+2H+2Cl=H2VOC14)=-3.75

Medium: HCl

V0++ nmr KC1 27°C var U H K1=-1.9 B2=-4.90 1971ZMb (5946) 386

Medium: HCl. Method: nmr. DH(K1)=31.4 kJ mol-1, DH(K2)=10.9 kJ mol-1

V0++ EMF NaClO4 20°C 1.0M U K1=0.04 1958ANb (5947) 387

F- HL Fluoride CAS 7644-39-3 (201)
Fluoride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ ISE non-aq 185°C 100% M K1=3.94 B2=7.64 1988JHa (7326) 388
B3=11.08
B4=13.99

Medium: molten KSCN. K1=mol-1 kg, B2=mol-2 kg2 etc.

V0++ ISE NaClO4 25°C 1.0M U K1=3.38 B2=5.75 1971AKa (7327) 389
B3=7.31
B4=8.0

Method: quinhydrone and fluoride-ISE

V0++ cal NaClO4 25°C 1.0M U H 1971AKa (7328) 390
DH(K1)=7.9 kJ mol-1, DH(K2)=6.5, DH(K3)=5.9, DH(K4)=6.3;
DS(K1)=90.8 J K-1 mol-1, DS(K2)=66.9, DS(K3)=49.4, DS(K4)=35.6

V0++ EMF NaClO4 20°C 1.00M U K1=3.31 B2=5.57 1961RYa (7329) 391
K3=1.56
B3=7.13
B4=7.8

VO++ gl none 25°C 0.0 U K1=3.15 1951DUa (7330) 392

FC1BrI HL (541)
Halides, comparative (for book data under ligand 80)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ sp non-aq 100% U I 1968GMc (7438) 393
K(VO(acac)2+N3)=3.1
K(VO(acac)2+SCN)=1.96
K(VO(acac)2+Cl)=0.43
K(VO(acac)2+Br) < -0.1

Medium: MeCN. In C2H4Cl2: K=2.64(N3), 1.74(SCN), 0.38(Cl), <-0.1(Br)

HPO3-- H2L Phosphite CAS 13598-36-2 (6305)
Phosphite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ gl oth/un 20°C dil U K1=3.80 1967PSa (7516) 394
B(VO+HL)=1.80
B(VOHL+HL)=1.22

By solubility: K(VOL+H2L)=2.79

IO3- HL Iodate CAS 7782-68-5 (1257)
Iodate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ kin NaClO4 16°C 5.50M U K1=0.35 1972G0b (8573) 395

NH3O L Hydroxylamine; CAS 5470-11-1 (1808)
Hydroxylamine; NH2.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ sp NaClO4 25°C 1.0M U 1973BEa (9276) 396
K(VO2+H3NOH)=1.2

OH- HL Hydroxide (57)
Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ gl NaClO4 25°C 1.00M C 1982NFa (12471) 397
*K1=-6.07
*B(2,2)=-6.59

VO++ gl oth/un 25°C 0.10M C 1977KHa (12472) 398

$*K1=-5.05$
 $*B(2,2)=-6.72$
 $Kso(VO(OH)2)=-22.18$
 $B((VO)2(OH)5)=47.26$
Medium: 0.10 M LiClO₄. $Ks(Na(VO)2(OH)5)=Na+(VO)2(OH)5)=-3.74$,
 $Ks(2VO(OH)2(s)+H2O=(VO)2(OH)5+H)=-11.1$. *B(p,q): pVO+qH2O=(VO)p(OH)q+qH.

V0++ nmr none 25°C 0.0 U 1975FCb (12473) 399
* $B(2,2)=-6.82$
 $Kso(VO(OH)2)=-21.97$

V0++ gl none 25°C 0.00 U 1973HMb (12474) 400
* $K1=-5.67$
* $B(2,2)=-6.67$

V0++ dis oth/un 25°C 0.10M U K1=11.46 B2=22.5 1972KEb (12475) 401
Medium: HClO₄

V0++ gl KCl 25°C 3.00M U 1972MBa (12476) 402
* $K1=-6.4$
* $B2=-10.0$
* $B(2,2)=-7.45$

V0++ dis NaClO₄ 25°C 0.10M U K1=11.32 B2=22.3 1971MKc (12477) 403

V0++ kin NaClO₄ 25°C 0.30M U T H 1970LWa (12478) 404
* $B(2,2)=-6.96$
DH(*B(2,2))=53.1 kJ mol⁻¹.
*B(2,2)=-7.62(4 °C), -7.22(16 °C), -6.60(36 °C), -6.30(46 °C)

V0++ nmr NaClO₄ rt var U I 1962RIa (12479) 405
 $K(VO+H)=-0.9$
Medium: HClO₄. K=-0.5(H₂SO₄), -0.6(HNO₃)

V0++ gl oth/un 20°C var U 1960WEa (12480) 406
 $K(H18V12037+H)=10.7$?

V0++ gl NaClO₄ 25°C 3.0M U 1955RRa (12481) 407
* $K1=-6.0$
* $B(2,2)=-6.88$
 $K(2VO(OH)=(VOOH)2)=5.1$

V0++ gl oth/un 25°C var U 1953MEa (12482) 408
* $K1(VO+H2O=VO(OH)+H)=-5.36$

V0++ gl none 20°C 0.0 U 1951DUa (12483) 409
* $K1=-4.77$
 $Kso(VO(OH)2)=-22.13$
 $K(H2V205(s)=HV205+H)=-10.3$

O2-- H2L Peroxide CAS 7772-84-1 (2813)
Peroxide; -O.O-

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	nmr	KCl	25°C	1.00M	U				1991JTb (12757)	410
$K(VO_2(OH)_2 + H_2L = VO(OH)_2L) = 2.78$ $K(VO(OH)_2L + H_2L = V(OH)_2L_2) = 5.78$ $K(V(OH)_2L_2 + H_2L = VOHL_3) = -5.57$ $K(VO(OH)_2L = VO_2OH + H) = -6.2$ $K(V(OH)_2L_2 = VO(OH)L_2 + H) = -7.2$; $K(2V(OH)_2L_2 = (VOL)_2 + 2OH + H) = -4.77$.										

P04--- H3L Phosphate CAS 7664-38-2 (176)
Phosphate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C	M		$K_1 = 10.8$ $B(VOH_2L) = 20.3$ $B(VOHL) = 16.8$ $B(VOH-2L) = -3.0$ $B((VO)_2H-2L_2) = 13.35$	1998KKe (13378)	411
$B(VOLA) = 18.81$, $B(VOHLA) = 25.00$, $B(VOH_2LA) = 28.69$, HA = maltol.										

V0++	sp	NaClO4	25°C	1.0M	C				1976CKb (13379)	412
									$K(VO + H_2PO_4) = 3.20$ $K(VO(H_2PO_4) + H_2PO_4) = 1.95$ $*K(VO(H_2PO_4)) = -3.22$ $*K(VO(H_2PO_4)_2) = -3.5$	

V0++	sp	NaClO4	25°C	1.00M	U				1975IVa (13380)	413
									$K(VO + H_2L) = 1.49$ $K(VO + HL) = 5.33$ $K(VO + 2HL) = 8.25$	

V0++	kin	KCl	25°C	0.20M	U				1971KYa (13381)	414
									$K(VO + H_2L) = 2.18$	

V0++	vlt	none	25°C	0.0	U				1956ZKa (13382)	415
									$K_{so}((VO)_3L_2) = -24.1$	

P207--- H4L Pyrophosphate CAS 2466-09-3 (198)
Diphosphate; from $(HO)_2PO_0.O.PO(OH)_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			$B_2 = 17.67$ $B(VOH_2L) = 16.36$ $B(VOHL) = 15.00$ $B(VOH-1L) = 4.45$	1995BKb (13670)	416

B(VOHL2)=23.18

B((VO)3L3)=42.55

V0++ kin KCl 25°C 0.20M U 1971KYa (13671) 417
K(VO+H2L)=4.20

P3010---- H5L CAS 10380-08-2 (1001)
Tripolyphosphate; from (HO)2PO.O.PO(OH).O.PO(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=10.33 B2=14.40 B((VO)H2L)=15.17 B((VO)HL)=14.16 B((VO)H-1L)=2.72 B((VO)H-2L)=-6.06	1995BKb (13917)	418

B((VO)HL2)=20.88

V0++ gl NaClO4 25°C 0.10M U K1=9.87 1989CGb (13918) 419
B((VO)HL)=14.06

V0++ kin KCl 25°C 0.20M U 1971KYa (13919) 420
K(VO+H2L)=4.81

SCN- HL Thiocyanate CAS 463-56-9 (106)
Thiocyanate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	NaCl	30°C	3.0M	U			K1=1.38 B2=2.24 B3=3.08	1970LNa (15332)	421

V0++ ISE oth/un 40°C 0.0 U T H T K1=2.15 B2=3.47 1968SWb (15333) 422
Medium:0 corr. K1=2.48(10 C), 2.32(25 C); K2=1.4(10 C), 1.36(25 C)
DH(K1)=-17.7 kJ mol-1, DS=-11.7 J K-1 mol-1

V0++ EMF oth/un rt var U K1=1.7 1963GKd (15334) 423

V0++ sp oth/un rt var U K1=1.25 1963GKd (15335) 424

V0++ sp alc/w rt 100% U I 1963GKd (15336) 425
B2=ca.5.2

Medium: MeOH. In acetone B2=ca.6.9

V0++ sp NaClO4 25°C 2.60M U H T K1=0.92 1951FGa (15337) 426
DH(K1)=1.7 kJ mol-1.

S04-- H2L Sulfate CAS 7664-93-9 (15)
Sulfate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sol	oth/un	20°C	5.25M	U				1973GTa (16661)	427
K(V0+HL)=0.55										
Medium: Na2S04										
V0++	sp	NaClO4	20°C	1.0M	U		K1=1.74	B2=2.51	1973IVb (16662)	428
K(V0+HL)=0.23										
V0++	cal	oth/un	25°C	0.0	U	H			1971BLc (16663)	429
DH(K1)=17.2 kJ mol-1, DS=105.3 J K-1 mol-1, DG=-14.13 kJ mol-1										
V0++	kin	oth/un	25°C	var	U		K1=2.40		1966KWa (16664)	430
K1in/K1out=0.0										
K1out=2.10										
K1 by conductivity, I=0 corr.										
V0++	sol	oth/un	25°C	0.0	U		K1=2.48		1963SWc (16665)	431

CH2O2		HL		Formic acid			CAS	64-18-6	(37)	
Methanoic acid; H.CO0H										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	NaClO4	20°C	1.00M	U		K1=1.98	B2=2.77	1973IVc (17660)	432

CH305P		H3L		Phosphonoformic			CAS	4428-95-9	(5654)	
Phosphonoformic Acid; O:P(OH)2.CO0H										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		K1=9.20	B2=15.22	1996SMa (17704)	433
B((VO)HL)=11.53										
B((VO)H-1L)=2.73										
B((VO)HL2)=18.35										

CH503P		H2L					CAS	13590-71-1	(1752)	
Methylphosphonic acid; CH3.PO3H2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		K1=5.87		1996SMa (18135)	434
B((VO)HL)=9.55										
B((VO)H-2L)=-5.98										
B((VOH-1L)2)=6.72										

CH504P		H2L					CAS	2617-47-2	(1977)	
Hydroxymethylphosphonic acid; HO.CH2.PO3H2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

V0++ gl KCl 25°C 0.20M C K1=6.59 1996SMa (18152) 435
 B((VO)HL)=10.14
 B((VO)H-2L)=-6.39
 B((VO)H-1L2)=4.24
 B((VOH-1L)2)=7.69

CH606P2 H4L Medronic acid CAS 1984-15-2 (2384)
 Methanediphosphonic acid; CH₂(PO₃H₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			B2=19.75 B((VO)H2L)=21.64 B((VO)HL)=19.03 B((VO)3L3)=53.29	1996SMa (18297)	436

C2H02C13 HL Trichloroacetic acid CAS 76-03-9 (1205)
 Trichloroethanoic acid; Cl₃C.CO₂H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	dis	oth/un	25°C	0.35M	U	M		B2,3(VOL2+3(antipyrine))=6.39 B2,3(VOL2+3py)=4.78 3-picoline, B=6.05, 4-pic,6.88 quinoline, 6.02, iso-quin,9.4	1976RSa (18338)	437

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)
 Ethanedioic acid; (COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=5.77 B(VOH-1L)=0.44	2000BKa (19141)	438

V0++	gl	KN03	35°C	0.10M	U T HM			B(VO(bpy)L)=8.88	1978JKa (19142)	439
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Data for 45 C. DH and DS values reported.

V0++	gl	KN03	25°C	0.10M	U	M		B(VO(Iminodiethanoate)L)=7.66	1978JSb (19143)	440
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V0++	gl	KN03	30°C	0.10M	M			K1=4.65	1978SVa (19144)	441
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V0++	sp	NaClO ₄	21°C	1.00M	U			K1=6.45 B2=11.77	1970IVa (19145)	442
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V0++	gl	NaClO ₄	25°C	0.10M	U			K(VOL(OH)+H)=5.7	1966KFc (19146)	443
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V0++ sp NaCl 25°C 1.00M U K1=6.48 B2=9.28 1964NNa (19147) 444

V0++ sp oth/un ? 0.50M U K1=9.76 1959TTa (19148) 445

V0++ ISE oth/un 25°C ? U B2=12.3 1959Z0a (19149) 446

Medium: saturated K2C2O4

V0++ sp oth/un ? 0.05M U B2=9.80 1957TTa (19150) 447

V0++ sp oth/un ? 0.05M U K1=9.76 1956TTa (19151) 448

C2H3O2Cl HL Chloroacetic CAS 79-11-8 (34)

Chloroethanoic acid; ClCH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl NaClO₄ 25°C 1.00M C K1=1.04 B2=1.60 1981LLc (19394) 449
B3=2.15

V0++ gl NaClO₄ 25°C 1.00M U K1=1.15 B2=1.78 1969DDa (19395) 450
K3=0.36

C2H4O2 HL Acetic acid CAS 64-19-7 (36)

Ethanoic acid; CH₃.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl NaClO₄ 25°C 1.0M U H K1=1.97 B2= 3.46 1988DTa (20224) 451
By calorimetry: DH(K1)=10.42 kJ mol⁻¹, DS(K1)=73 J K⁻¹ mol⁻¹; DH(B2)=19.33
DS(B2)=131.

V0++ gl NaClO₄ 25°C 1.00M U K1=1.86 B2=2.96 1981LLa (20225) 452

V0++ gl NaClO₄ 25°C 1.00M C K1=1.86 B2=2.96 1981LLc (20226) 453

C2H4O2S H2L Thioglycolic CAS 68-11-1 (596)

Mercaptoethanoic acid; HS.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KCl 25°C 0.20M C K1=8.6 B2=17.6 1993KBb (20383) 454
B((VO)HL)=12.8
B((VO)HL2)=22.3
K(VO+H2L=VOHL+H)=-1.81

C2H4O3 HL Glycolic acid CAS 79-14-1 (33)

2-Hydroxyethanoic acid; HO.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KCl 25°C 0.20M C T K1=2.66 B2=4.39 1993MSa (20651) 455
B((VO)H-1L)=-1.35
B((VO)H-1L2)=0.89
B((VO)H-2L2)=-3.98

V0++ gl NaClO4 25°C 1.0M U H K1=2.56 B2= 4.22 1988DTa (20652) 456
B3=5.19

By calorimetry: DH(K1)=-0.59 kJ mol⁻¹, DS(K1)=47 J K⁻¹ mol⁻¹; DH(B2)=6.40, DS(B2)=102; DH(B3)=4.39, DS(B3)=114.

V0++ gl NaClO4 25°C 1.00M C T K1=2.56 B2=4.22 1981LLc (20653) 457
B3=5.19

V0++ gl KN03 35°C 0.10M U T HM 1978JKa (20654) 458
B(VO(bpy)L)=23.26

Data for 45 C. DH and DS values reported.

V0++ gl KN03 25°C 0.10M U M 1978JSb (20655) 459
B(VO(Iminodiethanoate)L)=17.96

C2H5N02 HL Glycine CAS 56-40-6 (85)
2-Aminoethanoic acid; H2N.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++ gl NaClO4 25°C 1.0M C I K1=11.24 1995GZa (21754) 460
Additional method: spectrophotometry

V0++ nmr KCl 25°C 1.00M U K1=6.81 B2=12.23 1994NSb (21755) 461
Method: H1 Nuclear magnetic relaxation of solvent (H2O).

V0++ gl NaClO4 25°C 1.0M U H 1988DTa (21756) 462
K(VO+HL)=1.06
K(VO+2HL)=2.25

By calorimetry: DH(VO+HL)=4.51 kJ mol⁻¹, DS(VO+HL)=35 J K⁻¹ mol⁻¹; DH(VO+2HL)=3.50, DS(VO+2HL)=55.

V0++ gl NaClO4 25°C 1.00M C T K1=6.51 B2=11.82 1982FNb (21757) 463
B((VO)HL)=10.81
B((VO)HL2)=16.63
B((VO)H-1L2)=4.10
B((VO)H-1L)=1.3

B((VO)H-2L)=-6.3; B((VO)2H-2L2)=5.1

V0++ gl oth/un 25°C U K1=8.24 B2=15.66 1970CBb (21758) 464

C2H5N02 HL Acetohydroxamic CAS 546-88-3 (2766)
Acetohydroxamic acid, N-Hydroxyacetamide; CH3.CO.NHOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++ gl NaCl04 25°C 0.10M C K1=8.58 B2=16.21 2003SCa (21820) 465
B(VOH-1L2)=9.23
B(VOH-2L2)=0.70

By spectrophotometry: K1=8.30, B2=15.91, B(VOH-1L2)=9.21, B(VOH-2L2)=1.28.

C2H5N03 HL N-OH Glycine CAS 4746-62-7 (1235)
N-Hydroxy-2-aminoethanoic acid; HO.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO3	25°C	0.10M	U			K1=6.4 B2=12.60 K3=5.1	1987BKa	(21831) 466

C2H5O5P H3L CAS 4408-78-0 (4225)
Phosphonoethanoic acid; HOOC.CH2.PO3H2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=9.28 B2=15.13 B((VO)HL)=12.99 B((VO)H-1L)=2.52 B((VO)H2L2)=23.76 B((VO)HL2)=19.86	1996SMa	(21897) 467

C2H8N2 L Ethylenediamine CAS 107-15-7 (23)
1,2-Diaminoethane; H2N.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	con oth/un	20°C	0.01M	U				B2=5.7	1961BSa	(23245) 468

Medium: 0.017 VOSO4

C2H8O7P2 H4L HEDPA CAS 2809-21-4 (436)
1-Hydroxyethane-1,1-diphosphonic acid; CH3.C(OH)(PO3H2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO3	25°C	0.10M	C				1998DKa	(23403) 469

B((VO)HL)=18.56
B((VO)3L3)=53.25

V0++	gl	KCl	25°C	0.20M	C				1996SMa	(23404) 470
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B((VO)HL)=19.27
B((VO)3L3)=54.73

C2H9N06P2 H4L IDPA CAS 32545-63-4 (1335)
Imino-N,N-bis(methylenephosphonic acid); HN(CH2PO3H2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++	gl	KCl	25°C	0.20M	C	K1=12.22 B(VOHL2)=26.29 B((VO)2H-2L2)=12.67 B(VOHL)=17.55	1999SBb (23460)	471

C3H4O4		H2L	Malonic acid		CAS 141-82-2	(79)		
Propanedioic acid; CH2(COOH)2								
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C	K1=5.62 B(VOHL)=7.2 B(VOH-1L)=-0.71	2000BKa (24587)	472
V0++	gl	NaClO4	25°C	1.00M	C	K1=5.594 B((VO)HL)=6.20 B((VO)H-1L)=0.52 B((VO)H-1L2)=2.56	1982NFa (24588)	473
V0++	gl	NaClO4	30°C	0.10M	U	K1=6.10 B2=10.60	1980NSd (24589)	474
V0++	gl	KNO3	35°C	0.10M	U T HM	B(VO(bpy)L)=11.34	1978JKa (24590)	475
Data for 45 C. DH and DS values reported.								
V0++	gl	KNO3	25°C	0.10M	U M	B(VO(Iminodiethanoate)L)=11.85	1978JSb (24591)	476
V0++	gl	NaClO4	25°C	3.00M	U	K(VO+H+L)=7.41	1973ITa (24592)	477
V0++	sp	NaClO4	21°C	1.00M	U	K1=5.23 B2=8.85	1970IVa (24593)	478

C3H5O2Cl		HL				CAS 107-94-8	(1436)	
3-Chloropropanoic acid; Cl.CH2.CH2.COOH								
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference	ExptNo
V0++	gl	NaClO4	25°C	1.00M	C	K1=1.54 B2=2.63 B3=3.20	1981LLc (24738)	479

C3H6O2		HL	Propionic acid		CAS 79-09-4	(35)		
Propanoic acid; CH3.CH2.COOH								
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference	ExptNo
V0++	gl	NaClO4	25°C	1.0M	C	K1=1.91 B2=3.46	1983LLb (25074)	480

C3H6O2S		H2L	Thiolactic acid		CAS 79-42-5	(366)		

2-Mercaptopropanoic acid; CH₃.CH(SH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=8.40 B2=17.55 B((VO)HL)=11.85 B((VO)HL2)=20.68	1993KBb (25178)	481

C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)

L-2-Hydroxypropanoic acid; CH₃.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		T	K1=2.79 B2=5.15 B((VO)H-1L)=-1.34 B((VO)H-1L2)=1.28 B((VO)H-2L2)=-3.80	1993MSa (25568)	482

V0++	sp	NaClO ₄	20°C	1.0M	U		K1=2.68 B2=4.83	1965JLa (25569)	483
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Also by circular dichroism

C3H7N02 HL Alanine CAS 56-41-7 (86)

2-Aminopropanoic acid; H₂N.CH(CH₃).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO ₄	25°C	1.0M	C	I		K1=11.46	1995GZa (26294)	484
Additional method: spectrophotometry										
V0++	gl	NaNO ₃	25°C	2.25M	C			K1=6.87 B2=12.4 B((VO)HL)=11.16 B((VO)H2L2)=21.4 B((VO)HL2)=17.6 B((VO)2H-2L2)=5.8	1988PBa (26295)	485
B((VO)H-1L2)=4.4, B((VO)2H-3L2)=-1.3, B((VO)H-2L)=-6.7										

V0++	gl	oth/un	25°C	dil	U		K1=8.34 B2=15.63	1970CBb (26296)	486
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V0++	EMF	oth/un	?	?	U		K1=8.70	1970FMb (26297)	487
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C3H7N02 HL B-Alanine CAS 107-95-9 (575)

3-Aminopropanoic acid; H₂N.CH₂.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	oth/un	25°C	dil	U			K1=8.34	1970CBb (26494)	488
V0++	EMF	oth/un	?	?	U			K1=9.80	1970FMb (26495)	489

C3H7N02 HL DL-Alanine CAS 302-72-7 (189)

DL-2-Aminopropanoic acid; H2N.CH(CH3).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	nmr	KCl	25°C	1.00M	U			K1=6.64 B2=12.06	1994NSb (26544)	490

Method: H1 Nuclear magnetic relaxation of solvent (H2O).

C3H7NO2S	H2L	Cysteine	CAS	52-90-4	(96)
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2-Amino-3-mercaptopropanoic acid; H2N.CH(CH2.SH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++	gl	NaNO3	25°C	2.25M	C			B2=19.6 B((VO)H2L)=19.9 B((VO)HL)=16.1 B((VO)H4L2)=39.1 B((VO)H3L2)=35.8	1990CVa (26851)	491
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B((VO)H2L2)=31.4; B((VO)HL2)=26.4; B((VO)2L2)=25.3

V0++	sp	NaNO3	25°C	2.25M	U			B2=19.2 B((VO)HL)=16.1 B((VO)H2L)=19.9 B((VO)H4L2)=39.3 B((VO)H3L2)=35.8	1989PBc (26852)	492
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B((VO)H2L2)=31.0, B((VO)HL2)=26.0, B((VO)2L2)=25.2

C3H7NO3	HL	Serine	CAS	56-45-1	(49)
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2-Amino-3-hydroxypropanoic acid; H2N.CH(CH2.OH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++	gl	NaNO3	25°C	2.25M	C			K1=6.38 B2=11.70 B((VO)HL)=10.37 B((VO)H2L2)=19.9 B((VO)HL2)=16.44 B((VO)2H-2L2)=4.99	1989PBd (27196)	493
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B((VO)H-1L2)=4.45, B((VO)2H-3L2)=-1.23, B((VO)H-2L2)=-5.0, B((VO)H-2L)=-6.0,
B((VO)H-3L)=-18.0

V0++	EMF	oth/un	?	?	U			K1=7.50	1970FMb (27197)	494
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C3H7NO3	H2L		CAS	6252-11-5	(1236)
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N-Hydroxy-N-methyl-2-aminoethanoic acid; CH3(OH)N.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++	gl	KNO3	25°C	0.10M	U			K1=6 B2=11.00 K3=4	1987BKa (27233)	495
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C3H7O7P	H4L		CAS	820-11-1	(8695)
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D-3-Phosphoglyceric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=6.99 B(VOHL)=10.34 B((VO)2H-1L2)=13.35 B((VO)2H-2L2)=9.23 B((VO)2H-3L2)=2.31	2001HJa (27335)	496

C3H8N05P		H3L	Glyphosate		CAS	1071-83-6	(1617)			
N-(Phosphonomethyl)glycine; H2O3P.CH2.NH.CH2.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=10.69 B(VOHL2)=23.35 B((VO)2H-2L2)=10.74 B(VOHL)=14.37	1999SBb (27410)	497

C3H12N09P3		H6L	NTPA		CAS	6419-19-8	(2920)			
Nitrilotris(methylenephosphonic acid); N(CH2PO3H2)3										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=17.66 B(VOHL)=24.84 B(VOH-1L)=8.33 B(VOH2L)=28.94 B(VOH3L)=31.39	1999SBb (28595)	498
B(VOH4L)=33.3. By spectrophotometry: K1=18.0.										

C4H2O4		H2L	Squaric acid		CAS	2892-51-5	(439)			
3,4-Dihydroxy-3-cyclobutene-1,2-dione;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	NaClO4	25°C	3.00M	C			K1=2.47	1974AVa (28671)	499

C4H4O4		H2L	Maleic acid		CAS	110-16-7	(111)			
cis-Butenedioic acid; HOOC.CH:CH.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO4	20°C	0.1M	U				1999MKc (29156)	500
K(V0+OH+L)=16.39										
V0++	gl	NaClO4	30°C	0.10M	U			K1=5.19	1980NSd (29157)	501
V0++	gl	KNO3	35°C	0.10M	U T HM				1978JKa (29158)	502

$$B(V_0(bpy)L) = 11.44$$

Data for 45 C. DH and DS values reported.

V0++ gl KNO₃ 30°C 0.10M M K1=4.41 1978SVa (29159) 503

C4H6O4 H2L Succinic acid CAS 110-15-6 (112)
1,4-Butanedioic acid; HOOC.CH2.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
VO++	gl	KCl	25°C	0.20M	C			K1=3.20 B(VOHL)=7.2 B(VOH-2L)=-7.25 B(VOHL2)=9.56	1998PGb (30073)	504
VO++	gl	NaClO4	30°C	0.10M	U			K1=3.65	1980NSd (30074)	505
VO++	EMF	NaClO4	30°C	0.10M	U	M		B((VO)LA)=8.70 B((VO)LB)=8.34 B((VO)LC)=11.29 B((VO)LD)=9.40	1977SJa (30075)	506

H2A=malonic, H2B=phthalic, H3C=5-sulphosalicylic, H2D=3,5-dinitrosalicylic acid

C4H6O4S H2L Thiodiacetic CAS 123-93-3 (140)
2,2'-Thiodiglycolic acid, Thiodiethanoic acid; HOOC.CH₂.S.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
VO++	gl	NaClO4	25°C	0.50M	U			K1=3.14	1973NAc	(30243) 507
C4H6O4S	H3L	Thiomalic acid		CAS 70-49-5	(109)					
2-Mercaptosuccinic acid.	2-Sulfanyl-1,4-butanedioic acid:							$\text{HOOC}-\text{CH}(\text{SH})-\text{CH}_2-\text{COOH}$		

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
VO++	g1	KCl	25°C	0.20M	C			K1=10.40 B((VO)HL)=13.51 B((VO)HL2)=22.42 B((VO)H-1L)=4.44	1993KBb (30378)	508

C4H6O4S2 H4L CAS 2418-14-6 (4264)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++ g1 KC1 25°C 0.20M C B2=20.71 1993KBb (30399) 509
 $B((V0)HL)=20.91$
 $B((V0)HL2)=30.19$
 $B((V0)H2L)=23.39$

B((VO)3L2)=38.31

B((VO)2L2)=35.60

C4H604S2 H2L CAS 505-73-7 (3585)

Dithiodiethanoic acid; HOOC.CH2.S.S.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ gl NaClO4 25°C 0.10M M TIH K1=2.55 B2= 4.08 1980BDc (30414) 510
Data for 0.1-0.3 M NaClO4 and for 35 C. At I=0.0 M, K1=2.39, K2=1.37.
DH(K1)=-14.1 kJ mol-1, DS(K1)=1.1 J K-1 mol-1; DH(K2)=-3.5, DS(K2)=17.0.

C4H605 H2L Malic acid CAS 617-48-1 (393)

2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ gl NaNO3 25°C 2.25M C K1=4.3 B2=7.8 1992HTa (30755) 511
B((VO)H2L2)=13.1
B((VO)HL)=6.9
B((VO)HL2)=11.4
B((VO)H-1L)=0.6

B((VO)H-2L)=-4.4; B((VO)H-1L2)=2.8; B((VO)H-2L2)=-4.2

C4H605 H2L Diglycolic acid CAS 110-99-6 (243)

Di(carboxy)methyl ether, 2,2'-Oxydiethanoic acid; HOOC.CH2.O.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ gl NaClO4 25°C 0.10M M TIH K1=3.10 B2= 5.16 1980BDc (30946) 512
Data for 0.1-0.3 M NaClO4 and for 35 C. At I=0.0 M, K1=3.01, K2=1.95.
DH(K1)=-15.8 kJ mol-1, DS(K1)=5.8 J K-1 mol-1; DH(K2)=-0.0, DS(K2)=38.4.

VO++ gl NaClO4 25°C 0.50M U K1=5.01 1973NAc (30947) 513

C4H606 H2L D-Tartaric acid CAS 147-71-7 (93)

D-Tartaric acid, D-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ gl KCl 25°C 0.20M C K1=4.04 1995KBa (30980) 514
B((VO)2H-1L2)=9.34
B((VO)2H-2L2)=6.37
B((VO)2H-3L2)=-0.49
B((VO)2H-4L2)=-5.61

VO++ sp NaClO4 25°C 2.00M U I 1981HTb (30981) 515
K((VO)2L2+2L)=0.8

VO++ gl KN03 25°C 0.10M C 1978PSa (30982) 516

$$\begin{aligned}
 B((VO)H-1L) &= 1.33 \\
 B((VO)2H-1L2) &= 9.78 \\
 B((VO)2H-2L2) &= 5.97 \\
 B((VO)2H-3L2) &= -0.9
 \end{aligned}$$

C4H6O6 H2L DL-Tartaric acid CAS 133-37-9 (94)
 DL-Tartaric acid,DL-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		K1=3.99 B((VO)2H-1L2)=9.01 B((VO)2H-2L2)=6.05 B((VO)2H-3L2)=-0.48 B((VO)2H-4L2)=-4.86	1995KBa (31033)	517	
V0++	gl	none	25°C	0.0	M			K((VO)2L2+H)=4.69 K(H(VO)2L2+H)=7.09	1982H0c (31034)	518
V0++	sp	NaClO4	25°C	2.00M	U	I			1981HTb (31035)	519
V0++	gl	KNO3	25°C	0.10M	C		K1=3.86 B((VO)H-1L)=1.20 B((VO)H-2L)=-4.31 B((VO)2H-1L2)=9.53 B((VO)2H-2L2)=6.11	1978PSa (31036)	520	

C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)
 L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		K1=3.80 B((VO)2H-1L)=9.31 B((VO)2H-2L2)=6.24 B((VO)2H-3L2)=-0.81 B((VO)2H-4L2)=-5.80	1995KBa (31391)	521	
V0++	gl	none	25°C	0.0	M			K((VO)2L2+H)=5.99 K(H(VO)2L2+H)=7.01	1982H0c (31392)	522
V0++	gl	NaClO4	30°C	0.10M	U		K1=6.24	B2=11.12	1980NSd (31393)	523
V0++	gl	KNO3	25°C	0.10M	C		K1=4.0 B((VO)H-1L)=1.50 B((VO)2H-1L2)=9.84 B((VO)2H-2L2)=6.21	1978PSa (31394)	524	

$$B((VO)2H-3L2)=-0.3$$

V0++ sp KN03 25°C 0.25M U 1970KPb (31395) 525
 $K(VO+HL=VOL+H)=1.10$
 $K(VOL+HL=VOL2+H)=0.30$

C4H6O6 H2L meso-Tartaric CAS 147-73-9 (91)
meso-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KCl 25°C 0.20M C K1=4.52 1995KBa (31431) 526
 $B((VO)3H-3L3)=8.56$
 $B((VO)3H-4L3)=4.55$
 $B((VO)3H-5L3)=-2.04$
 $B((VO)3H-6L3)=-10.46$

$B((VO)H-1L)=0.56$, $B((VO)H-1L2)=3.45$, $B((VO)H-2L2)=-2.48$

V0++ gl KN03 25°C 0.10M C K1=4.42 1978PSa (31432) 527
 $B((VO)2H-1L2)=7.75$
 $B((VO)2H-2L2)=4.46$
 $B((VO)2H-3L2)=-0.41$

C4H7N02 HL Acetoacetamide CAS 2044-64-6 (1407)
3-Oxobutanamide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KCl 25°C 0.20M C 2001CKa (31449) 528
 $K(VO+HL=VOL+H)=-2.30$
 $K(VO+2HL=VOL2+2H)=-6.04$

C4H7N04 H2L Aspartic acid CAS 56-84-8 (21)
Aminobutanedioic acid; H2N.CH(CH2.COOH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ nmr KCl 25°C 1.00M U K1=7.23 B2=12.64 1994NSb (31973) 529
Method: H1 Nuclear magnetic relaxation of solvent (H2O). For the DL isomer
K1=7.23, B2=12.64

V0++ gl NaNO3 25°C 2.25M C K1=9.05 B2=16.25 1990CMa (31974) 530
 $B((VO)H2L)=15.55$
 $B((VO)HL)=12.56$
 $B((VO)HL2)=21.49$
 $B((VO)H3L2)=27.77$

V0++ gl KN03 25°C 0.10M U K1=9.2 1987BKa (31975) 531

V0++ gl NaClO4 25°C 0.10M U K1=8.98 B2=15.47 1972SSe (31976) 532

K3=4.42

V0++ gl KNO₃ 25°C 0.10M U K1=8.39 B2=14.43 1972TSd (31977) 533

C4H₇N₀4 H₂L IDA CAS 142-73-4 (118)
Iminodethanoic acid; HN(CH₂.COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=8.84 B2=15.32 B(VOHL2)=20.08 B((VO)2H-2L2)=9.56	1999SBb (32392)	534

V0++	gl	KNO ₃	25°C	0.10M	C			K1=9.00 K(VO(OH)L+H)=5.8 K((VO)2(OH)2L2+2H)=9.1	1984FVa (32393)	535
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V0++	gl	KNO ₃	25°C	0.10M	U			K1=8.98	1983FSa (32394)	536
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V0++	gl	KNO ₃	25°C	0.10M	U	M		B((VO)LA)=7.66 B((VO)LB)=11.85 B((VO)LC)=18.35 B((VO)LD)=18.35	1978JSb (32395)	537
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B((VO)LE)=17.52, B((VO)LF)=17.96. H₂A=oxalic, H₂B=malonic, H₂C=succinic,
H₃D=5-sulfosalicylic, HE=mandelic, HF=glycolic acid

V0++	gl	KNO ₃	25°C	0.10M	U	T	M	K1=9.01 B2=15.63 K(VO(OH)L+H=VOL)=5.71 K((VO(OH)L)2+2H=2VOL)=8.31	1973STc (32396)	538
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30 C: K1=8.98; K2=6.51; K(VO(OH)L+H=VOL)=5.59; K((VO(OH)L)2+2H=2VOL)=8.14

V0++	gl	NaClO ₄	25°C	0.10M	U			K(VOLOH+H)=5.50	1966KFc (32397)	539
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C4H₇N₀5 H₂L (1237)
N-Hydroxyaminobutanedioic acid; HO.NH.CH(CH₂.COOH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO ₃	25°C	0.10M	U			K1=7.24 B2=12.64	1987BKa (32411)	540

C4H₇N₀5 H₂L (1234)
N-Hydroxyiminodethanoic acid; HO.N(CH₂.COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++	gl	NaClO ₄	25°C	1.00M	C	H		B2=21.9	1987AKa (32427)	541
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DH(B2)=-56.5 kJ mol⁻¹, DS=229 J K⁻¹ mol⁻¹

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO4	25°C	1.0M	C			K1=1.97 B2=3.39	1983LLb (33260)	551

C4H8O2		HL						CAS 107-92-6 (1118)		
n-Butanoic acid; CH3.CH2.CH2.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO4	25°C	1.0M	C			K1=1.94 B2=3.37	1983LLb (33356)	552

C4H8O3		HL						CAS 594-61-6 (81)		
2-Hydroxy-2-methylpropanoic acid; (CH3)2C(OH).COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		T	K1=3.10 B2=5.89 B((VO)H-1L)=-1.03 B((VO)H-1L2)=1.64 B((VO)H-2L2)=-3.03	1993MSa (33536)	553

C4H9N03		HL	Threonine					CAS 72-19-5 (48)		
2-Amino-3-hydroxybutanoic acid; H2N.CH(CH(OH).CH3)COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaNO3	25°C	2.25M	C			K1=6.41 B2=11.93 B((VO)HL)=10.30 B((VO)H2L2)=20.0 B((VO)HL2)=16.43 B((VO)2H-2L2)=4.98	1989PbD (34339)	554
B((VO)H-1L2)=4.80, B((VO)2H-3L2)=-1.35, B((VO)H-2L2)=-4.8, B((VO)H-2L)=-6.0, B((VO)H-3L)=-18.0										

V0++	vlt	NaClO4	25°C	0.10M	C			K1=0.89	1986SPb (34340)	555
Method: polarography.										

C4H11N08P2		H5L						CAS 2439-99-8 (2129)		
N-Carboxymethyl-N,N-bis(methylenephosphonic acid); HOOC.CH2.N(CH2.PO3H2)2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=16.78 B(VOHL)=21.26 B(VOH-1L)=7.87 B(VOH2L)=24.5	1999SBb (35117)	556
By spectrophotometry: K1=16.0										

C4H13N07P2		H4L						CAS 63132-40-1 (1347)		

1-Hydroxy-4-aminobutyl-1,1-diphosphonic acid; (PO₃H₂)₂C(OH).CH₂.CH₂.CH₂.NH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	g1	KNO ₃	25°C	0.10M	C			B((VO)HL)=17.41 B((VO)3L3)=50.17	1998DKa (35618)	557

C5H4N2O2 HL CAS 98-97-5 (1879)
Pyrazine-2-carboxylic acid; cyclo(-CH:CH.N:C(COOH).CH:N-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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VO++ g1 NaClO4 25°C 0.50M C K1=3.30 B2=6.25 1989NMa (36067) 558
B3=7.80

C5H4N2O4 H2L CAS 570-22-9 (7544)

Imidazole-4,5-dicarboxylic acid; C₃H₂N₂(COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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VO++ gl KCl 25°C 0.20M C K1=7.80 B2=13.55 1998SMa (36131)

$B(VOH-2L2) = -2.7$
 $B(VOHL) = 11.04$
 $B((VO)4H-4L4) = 23.15$
 $B((VO)4H-5L4) = 16.55$

$$B((V0)4H-6L4)=8.65, \quad B((V0)4H-7L4)=-0.30, \quad B((V0)4H-8L4)=-9.46.$$

C5H4N2O4 H2L CAS 85908-17-4 (7545)

Pyrazole-3,5-dicarboxylic acid: CN1C=CC(=O)C(=O)N=C1

Pyrazole 3,5-dicarboxylic acid,

Metal Mtd Medium Temp Conc Cai Flags Lg R values Reference ExpNO

V0++ g1 KCl 25°C 0.20M C K1=3.72 B2= 6.52 1998SMa (36133) 560
 $B((VO)2H-2L2)=1.72$
 $B((VO)2H-3L2)=-4.26$
 $B((VO)2H-4L2)=-10.6$

VO++ g1 KC1 25°C 0.20M C K1=3.72 B2= 6.52 1998SMa (36134) 561
 $B((VO)2H-2L2)=1.72$
 $B((VO)2H-3L2)=-4.26$
 $B((VO)2H-4L2)=-10.6$

C5H5N L Pyridine CAS 110-86-1 (31)

Pyridine, Azine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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VO++ sp oth/un 25°C ? U K1=-1.70 1956ERa (36690) 562

C5H5NOS

(4389)

2-Mercaptopyridine N-oxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C	I		K1=6.7 B2=13.00	2003KKa (36723)	563
In 60% w/w DMSO/H ₂ O,								K1=7.7, B2=14.9.		
V0++	gl	KCl	25°C	0.20M	C	M			2003KKa (36724)	564
								B(VOHLA)=17.01		
								B(VOLA)=13.80		
								B(VOLB)=12.39		
								B(VOLC)=5.17		

$B(VOH-1LC)=5.47$
 $B(VOHL(P04))=23.40$. H3A is citric acid, H2B is oxalic acid, HC is lactic acid.

including the following: (a) the name of the individual or entity that will receive the information; (b) the purpose for which the information will be used; (c) the type of information that will be provided; (d) the source of the information; (e) the date the information was last updated; and (f) the date the information will be deleted.

C5H5NO₂ HI CAS 13161-30-3 (5582)

1-Hydroxypyridin-2(1H)-one **2-Hydroxypyridine 1-oxide:**

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	g1	KCl	25°C	0.20M	C	I		K1=8.30 B2=16.01	2003KKa (36757)	565
In 60% w/w DMSO/H ₂ O,				K1=8.80,	B2=16.87.					

B(VOH-1LB)=7.18, B(VOLC)=11.88, B(VOH-1LC)=6.79, B(VOH2L(PO4))=30.44, B(VOHL(PO4))=25.62. H3A is citric, H2B is oxalic and HC is lactic acid.

C5H5O2F3 HL CAS 367-57-7 (163)

1,1,1-Trifluoropentane-2,4-dione; CF₃.CO.CH₂.CO.CH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
V0++	sp	NaClO4	25°C	1.0M	U					1980HRa (37060)	567
										$K(VO+HL=VOL+H)=-0.96$	

C5H6N2O2 HL CAS

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=6.10 B(VOH-1L2)=3.1 B((VO)2H-2L2)=4.29	1998SMa (37319)	568

C5H6O4 H2L Citraconic acid CAS 498-23-7 (3021)

Citraconic acid; CH₃.C(COOH):CH.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO ₄	30°C	0.10M	U			K1=6.33	1980NSd (37375)	569

C5H6O₄ H₂L Itaconic acid CAS 97-65-4 (398)
Methylenesuccinic acid; HOOC.CH₂.C(:CH₂).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO ₄	30°C	0.10M	U			K1=3.91	1980NSd (37459)	570
V0++	EMF	NaClO ₄	30°C	0.10M	U	M			1977SJ _a (37460)	571

H2A=malonic acid, H3B=5-sulphosalicylic acid, H2C=3,5-dinitrosalicylic acid

C5H8O₂ HL Acetylacetone CAS 123-54-6 (164)
Pentane-2,4-dione; CH₃.CO.CH₂.CO.CH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	dis	NaClO ₄	25°C	0.10M	U			K1=8.59 B2=16.10	1986ISb (38120)	572
V0++	sp	NaClO ₄	25°C	1.0M	U				1980HR _a (38121)	573
V0++	sp	none	25°C	0.0	U	M		K(V0+HL=VOL+H)=-0.26	1978JZ _a (38122)	574
V0++	sp	non-aq	20°C	100%	U	M			1976K _{Ta} (38123)	575

K(VOL2+(Me)₂SO)=0.63
K(VOL2+(Et)₂SO)=0.68
K(VOL2+(Pr)₂SO)=0.68
K(VOL2+(Bu)₂SO)=0.70

V0++	sp	non-aq	20°C	100%	U	M			1976K _{Ta} (38124)	576
V0++	sp	non-aq	20°C	100%	U	M			K(VOL2+diethylamine)=2.21 K(VOL2+diisopropylamine)=2.26 K(VOL2+diisobutylamine)=2.23 K(VOL2+dicyclohexylamine)=2.03	577

Medium: CH₂Cl₂

V0++	sp	non-aq	20°C	100%	U	M			1976K _{Ta} (38125)	577
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$K(VOL_2 + \text{morpholine}) = 3.64$
 $K(VOL_2 + \text{piperidine}) = 4.11$
 $K(VOL_2 + \text{quinoline}) = 1.18$
 $K(VOL_2 + \text{isoquinoline}) = 2.92$

Medium: CH₂Cl₂

V0++	sp	non-aq	20°C	100%	U	M	1976KTa (38126) 578
							$K(VOL_2 + \text{pyridine}) = 2.94$
							$K(VOL_2 + (2\text{-picoline})) = 1.02$
							$K(VOL_2 + (3\text{-picoline})) = 3.19$
							$K(VOL_2 + (4\text{-picoline})) = 3.28$

Medium: CH₂Cl₂

V0++	sp	non-aq	20°C	100%	U	M	1976KTa (38127) 579
							$K(VOL_2 + (2,4\text{-lutidene})) = 1.62$
							$K(VOL_2 + (3,5\text{-lutidene})) = 3.26$
							$K(VOL_2 + 2\text{-hydroxypyridine}) = 2.81$
							$K(VOL_2 + (2\text{-cyanopyridine})) = 1.22$

Medium: CH₂Cl₂

V0++	sp	non-aq	20°C	100%	U	M	1976KTa (38128) 580
							$K(VOL_2 + (3\text{-cyanopyridine})) = 2.23$
							$K(VOL_2 + (4\text{-cyanopyridine})) = 2.20$
							$K(VOL_2 + (4\text{-ethylpyridine})) = 3.26$
							$K(VOL_2 + (2\text{-aminopyridine})) = 1.89$

Medium: CH₂Cl₂

V0++	sp	non-aq	20°C	100%	U	M	1975KTa (38129) 581
							$K(VO(L)_2 + A) = 1.54$
							$K(VO(L)_2 + B) = 1.59$
							$K(VO(L)_2 + C) = 1.59$
							$K(VO(L)_2 + D) = 1.62$

Medium: CH₂Cl₂. A=N,N-dimethylacetamide, B=N,N-diethylacetamide.
C=N,N-di-n-propylacetamide, D=N,N-diisopropylacetamide

V0++	sp	non-aq	20°C	100%	U	M	1975KTa (38130) 582
							$K(VO(L)_2 + A) = 1.44$
							$K(VO(L)_2 + B) = 1.39$
							$K(VO(L)_2 + C) = 1.43$
							$K(VO(L)_2 + D) = 1.42$

Medium: CH₂Cl₂. A=N,N-di(n-butyl)acetamide, B=N,N-di(isobutyl)acetamide,
C=N-methyl-N-phenylacetamide, D=N-ethyl-N-phenylacetamide

V0++	sp	non-aq	20°C	100%	U	M	1975KTa (38131) 583
							$K(VO(L)_2 + A) = 1.42$
							$K(VO(L)_2 + B) = 2.10$
							$K(VO(L)_2 + C) = 1.42$
							$K(VO(L)_2 + D) = 1.61$

Medium: CH₂Cl₂. A=N-methyl-N-benzylacetamide, B=N,N-dicyclohexylacetamide,
C=N-acetylmorpholine, D=N-acetylpirperidine

VO++ sp non-aq 20°C 100% U M 1975KTa (38132) 584
 K(VO(L)2+A)=1.62
 K(VO(L)2+B)=0.97
 K(VO(L)2+C)=1.38
 Medium: CH₂Cl₂. A=N,N,N',N'-tetramethylurea, B=N,N-dimethylbenzamide,
 C=N,N-dimethyl-chloroacetamide

VO++ gl oth/un 25°C ? U H K1=8.68 B2=15.79 1956BTa (38133) 585
 DH(VO₂+HL=VOL+H)=-7.1 kJ mol⁻¹, DS=142; DH(VOL+HL=VOL₂+H)=-28, DS=40

C5H8O6 H2L CAS 34618-90-1 (4292)
 2-Methyltartaric acid; HOOC.C(OH)(CH₃).CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
VO++	gl	none	25°C	0.0	M				1982H0c (38408)	586
								K((VO)2L2+H)=5.98		
								K(H(VO)2L2+H)=5.19		

For the L-isomer, K((VO)2L2+H)=6.31, K(H(VO)2L2+H)=5.44

C5H8O7 H2L CAS 40120-71-6 (3022)
 2,3,4-Trihydroxypentanedioic acid, Trihydroxyglutaric acid; HOOC.(CH(OH))₃.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
VO++	sp	NaClO ₄	?	0.01M	U			K1=3.82	1962GMa (38444)	587
Medium:	HC1O4									

C5H9NO2 HL Proline CAS 147-85-3 (44)
 Pyrrolidine-2-carboxylic acid; C₄H₈N.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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VO++ nmr KCl 25°C 1.00M U K1=7.28 B2=13.12 1994NSb (38655) 588
 Method: H₁ Nuclear magnetic relaxation of solvent (H₂O). For the DL isomer
 K1=7.28, B2=13.11

VO++ EMF oth/un ? ? U K1=10.30 1970FMb (38656) 589

C5H9NO3S H2L Thiopronin CAS 1953-02-2 (2162)
 N-2-Mercaptopropanoyl-glycine; CH₃.CH(SH).CO.NH.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
VO++	gl	KCl	25°C	0.20M	C	M		K1=5.13	2002JBa (38787)	590
								B(VOH2L2)=19.4		
								B(VOHL)=10.02		
								B(VOH-1L)=1.10		
								B(VOH-1L2)=4.57		

$B(VOH-2L)=-5.80$, $B(VOH-1(bpy)L)=6.14$, $B(VOH-1AL)=14.06$, $B(VOH-1(mal)L)=6.84$, $B(VOH-1(ox)L)=3.2$. H4A is tiron.

C5H9N04 H2L Glutamic acid CAS 56-86-0 (22)
2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO4	25°C	0.20M	C	TIH		K1=11.53 K(VO2L+H)=4.81 K(VO2+H2L=VO2HL+H)=2.58	2001MMb (39143)	591

Data for 20-30 °C and for 0.06-0.31 mole fraction MeOH/H2O.

DH(K1)=-3.57 kJ mol-1, DS(K1)=208 J K-1 mol-1.

V0++	nmr	KCl	25°C	1.00M	U		K1=8.06	B2=13.51	1994NSb (39144)	592
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Method: H1 Nuclear magnetic relaxation of solvent (H2O). For the DL isomer
K1=8.01, B2=13.59

V0++	gl	NaNO3	25°C	2.25M	C				1992CAa (39145)	593
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B((VO)H2L)=14.9
B((VO)H4L2)=29.0
B((VO)HL)=12.0
B((VO)H2L2)=23.3

B((VO)HL2)=18.9; B((VO)H-1L2)=7.5; B((VO)2H-5L2)=-10.4; B((VO)H-3L)=-10.4;
B((VO)2H-6L2)=-19.0

V0++	gl	NaClO4	25°C	0.10M	U		K1=7.73	B2=14.10	1972SSe (39146)	594
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K3=3.90

C5H9N04 H2L MIDA CAS 4408-64-4 (190)
N-Methyliminodiethanoic acid; CH3.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO3	25°C	0.10M	U		K1=9.56		1983FSa (39291)	595
V0++	gl	NaClO4	25°C	0.50M	U		K1=9.44 *K(VOL)=-5.76 K(2VOL=(VO(OH)L)2+2H)=-9.05		1977NAa (39292)	596

C5H10N07P H4L PMIDA CAS 5994-61-6 (2433)
N-(Phosphonomethyl)iminodiethanoic acid; H2O3P.CH2.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		K1=15.14 B(VOH2L)=20.4 B(VOH-1L)=7.06		1999SBb (39688)	597

By spectrophotometry: K1=14.9

C5H10N2O3 HL Glutamine CAS 56-85-9 (18)
2-Aminopentanedioic acid 5-amide; H2N.CH(CH2.CH2.CO.NH2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO4	25°C	0.10M	U			K1=7.40 B2=14.40 K3=4.07	1973TSe (39846)	598

C5H10O3 HL CAS 3739-30-8 (3612)
2-Hydroxy-2-methylbutanoic acid, Methylethylglycolic acid; CH3.CH2.C(OH)(CH3)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=2.88 B2=5.55 B((VO)H-1L)=-1.15 B((VO)H-1L2)=1.19 B((VO)H-2L2)=-2.97	1993MSa (40266)	599

C5H11N L Piperidine CAS 110-89-4 (105)
Perhydropyridine; cyclo(-CH2.CH2.CH2.NH.CH2.CH2-) C5H11N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	non-aq	25°C	100%	U	M			1975WHa (40459)	600

K(VOW+L)=-0.17
K(VOX+L)=-0.28
K(VOY+L)=-0.40
K(VOZ+L)=-0.44

Medium: toluene. W=(p-Cyano)tetraphenylporphyrin complex of V0++.
X=(p-chlor-), Y=(H-) analogue, Z=(OCH3-) analogus

V0++	sp	non-aq	25°C	100%	U	T	HM		1975WHa (40460)	601
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K(VOY+L)=-0.42

Medium: toluene. VOY: (p-Methyl)tetraphenylporphyrin complex of V0++
DH(VOY+L)=-23.4 kJ mol-1 at 35 C. 35 C: K=-0.55; 45 C: -0.68

C5H11N02 HL Valine CAS 72-18-4 (43)
2-Amino-3-methylbutanoic acid; H2N.CH(CH(CH3)2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	nmr	KCl	25°C	1.00M	U			K1=6.37 B2=11.42	1994NSb (40771)	602

Method: H1 Nuclear magnetic relaxation of solvent (H2O). For the DL isomer
K1=6.37, B2=11.40

V0++	EMF	oth/un	?	?	U			K1=8.65	1970FMB (40772)	603
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C5H11N02S H2L D-Penicillamine CAS 52-67-5 (1323)
D-2-Amino-3-mercaptop-3-methylbutanoic acid; (CH3)2C(SH)CH(NH2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaNO3	25°C	2.25M	C		B2=21.0 B((VO)H2L)=20.09 B((VO)HL)=16.3 B((VO)H4L2)=39.4 B((VO)H3L2)=36.1	1990CVa (41193)	604	
							B((VO)H2L2)=32.2; B((VO)HL2)=27.5; B((VO)2L2)=26.7; B((VO)H-1L2)=7.9; B((VO)H-2L)=-5.2			
V0++	sp	NaNO3	25°C	2.25M	U		B2=19.7 B((VO)HL)=16.3 B((VO)H2L)=19.8 B((VO)H4L2)=39.5 B((VO)H3L2)=36.1	1989PBc (41194)	605	
							B((VO)H2L2)=31.7, B((VO)HL2)=26.7, B((VO)2L2)=26.0			

C5H14O7P2		H4L					(7243)			
1-Hydroxypentane-1,1-diphosphonic acid; HO.C(PO(OH)2)2.(CH2)3CH3										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO3	25°C	0.10M	C				1998DKa (41938)	606
							B((V02)HL)=18.23 B((VO)3L3)=51.52			

C6H204Br2		H2L	Bromanic acid	CAS 4379-59-6			(1279)			
3,6-Dibromo-2,5-dihydroxy-1,4-benzoquinone;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	oth/un	25°C	1.00M	U				1980VPa (42041)	607
							K(VO(OH)2+2L+2H=VOL2)=18.98			

C6H204Cl2		H2L	Chloranilic acid	CAS 87-88-7			(1281)			
3,6-Dichloro-2,5-dihydroxy-1,4-benzoquinone;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	oth/un	25°C	1.00M	U				1980VPa (42060)	608
							K(VO(OH)2+2L+2H=VOL2)=18.82			

C6H4N204		H2L					CAS 89-01-0 (5801)			
Pyrazine-2,3-dicarboxylic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO4	25°C	0.50M	C		K1=3.90 B2=6.45		1989NMa (42208)	609

C6H4O4		H2L					CAS 615-94-1 (1280)			

2,5-Dihydroxy-1,4-benzoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	oth/un	25°C	1.00M	U				1980VPa (42311)	610
K(VO(OH)2+2L+2H=VOL2)=21.27										
C6H5N02		HL	Picolinic acid	CAS 98-98-6	(391)					
2-Pyridine-carboxylic acid; C5H4N.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C	M	K1=6.66	B2=12.11	2000KGd (42618)	611
B((VO)2H-2L2)=6.15 B((VO)H-1L2)=5.13 B((VO)AL)=12.38 B((VO)BL)=13.33										
B((VO)HBL)=16.96, B((VO)H-1BL)=7.57. H2 A is oxalic acid, H3B is citric acid.										
V0++	gl	KCl	25°C	0.20M	C	M			2000KGd (42619)	612
B((VO)AL)=9.29 B((VO)H-1AL)=5.20 B((VO)BL)=15.52 B((VO)HBL2)=26.21										
B((VO)BL2)=18.90, B((VO)H2BL)=26.97, B((VO)HBL)=22.73. HA is lactic acid, H3B is phosphoric acid.										
V0++	gl	KCl	25°C	0.20M	C		K1=6.66	B2=12.11	2000KPa (42620)	613
B((VOH-1L)2)=6.15 B(VOH-1L2)=5.13										
V0++	gl	NaNO3	25°C	0.10M	C		K1=6.68	B2=11.99	1994DHa (42621)	614
V0++	gl	NaClO4	25°C	0.50M	C		K1=6.50	B2=11.90	1987NMb (42622)	615
pK values of ligand: in Annali di Chimica, 76, 277 (1986)										
V0++	gl	KN03	30°C	0.10M	U				1979VSa (42623)	616
B((V02)LA)=18.63 B((V02)LB)=16.44										
H2A=2-hydroxybenzoic acid; H3B=5-sulpho-4-hydroxybenzoic acid										
V0++	gl	KN03	30°C	0.10M	U	M			1975STD (42624)	617
B((VO)AL)=21.27 K(VOA+H2L=VOAL+2H)=-5.79 K(VOA+L)=15.34 K(VO+HA+H2L=VOAL+3H)=-5.14										
H2A=catechol										
V0++	gl	KN03	30°C	0.10M	U	M			1975STD (42625)	618

$B((VO)AL)=21.67$
 $K(VOA+H2L=VOAL+2H)=-4.30$
 $K(VOA+L)=15.74$
 $K(VO+HA+H2L=VOAL+3H)=-3.65$

H4A=tiron (4,5-dihydroxynaphthalene-3,6-disulfonic acid)

V0++	gl	KNO ₃	30°C	0.10M	U	M	1975STd (42626) 619
							$B((VO)LA)=22.14$
							$K(VOL+H2A=VOAL+2H)=-4.73$
							$K(VOL+A)=16.21$
							$K(VO+HL+H2A=VOAL+3H)=-4.08$

H4A=4,5-dihydroxynaphthalene-3,6-disulfonic acid (chromotropic acid)

V0++	gl	NaClO ₄	25°C	0.10M	U	1966KFc (42627) 620	
							$K(VOLOH+H)=5.03$
							$K(VOL2OH+H)=6.95$

By spectrophotometry: $K(VOL2OH+H)=6.98$

C6H₅N₀3 H₂L CAS 609-71-2 (5910)
2-Hydroxypyridine-3-carboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		$B2=19.09$ $B(VOHL)=17.30$ $B(VOH2L2)=33.71$ $B(VOHL2)=26.69$ $B(VOH-1L2)=10.71$	2000KPa (42726) 621	

C6H₅N₀3 H₂L CAS 874-24-8 (4356)
3-Hydroxypyridine-2-carboxylic acid; C₅H₃N.(OH)(COOH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		$B2=20.55$ $B(VOHL)=17.27$ $B(VOH2L2)=33.93$ $B(VOHL2)=28.31$ $B(VOH-1L2)=10.03$	2000KPa (42756) 622	

$B((VOL)4)=66.70$

C6H₆N₂O HL CAS 873-69-8 (1258)
Pyridine-2-aldoxime; C₅H₄N.CH:NOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
V0++	gl	KNO ₃	30°C	0.10M	U	I M		1979SVa (43303) 623	
							$K(VOA+H2L=VOAL+2H)=5.12$		
							$K(VO+H2A+H2L=VOAL+4H)=8.53$		
							$K(VOB+H2L=VOBL+2H)=5.42$		

$K(VO + H_2B + H_2L = VOBL + 4H) = 8.15$
 $K(MC + H_2L = MCL + 2H) = 5.32$ and $K(M + H_2C + H_2L = MCL + 4H) = 7.39$.
 H₂A=Salicylic acid,
 H₂B=sulfosalicylic acid and H₂C=8-Hydroxyquinoline-5-sulfonic acid

V0++ gl KN03 30°C 0.10M M M 1978SVa (43304) 624
 $K(VO(A)+L)=8.18$
 $B((VO)AL)=12.83$
 $K(VO(B)+L)=8.62$
 $B((VO)BL)=12.63$

$K(VO(C)+L)=9.24$, $B(VO(C)L)=13.65$. H₂A is oxalic acid, H₂B is phthalic acid
 H₂C is maleic acid.

V0++ gl KN03 30°C 0.10M M 1975STc (43305) 625
 $K(VO + H_2L = VO(OH)L + H) = -8.40$

C6H6O2 H2L Catechol CAS 120-80-9 (534)
 1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			$K1=16.75$ $B2=31.58$	1990JKb (43861)	626
								$B(VOH-1L)=10.21$		
								$B(VO2H-2L2)=22.92$		

V0++ gl KN03 20°C 0.10M U K1=17.7 B2=33.50 1971ZBa (43862) 627

V0++ gl oth/un 30°C 0.10M U M K1=15.28 B2=28.30 1967Lab (43863) 628
 $K(VO(OH)L+H)=5.90$
 $K(VO(phen)+L)=16.69$

C6H6O3 H3L Pyrogallol CAS 87-66-1 (696)
 1,2,3-Trihydroxybenzene; C6H3(OH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KN03	20°C	0.10M	U				1971ZBa (43995)	629
								$K(VO+HL)=15.0$		
								$K(VOHL+HL)=13.7$		

V0++ sp alc/w 25°C 50% U I 1970CSF (43996) 630
 $K(VO2+HL)=5.3$

Medium: 0-100% MeOH, 0.01 M LiNO₃. $K(0\%)=4.7$, $K(75\%)=6.3$, $K(100\%)=7.0$

V0++ sp oth/un ? 0.10M U 1970CSg (43997) 631
 $K(VO+HL)=7.02$

C6H6O3 HL Maltol CAS 118-71-8 (2442)
 3-Hydroxy-2-methyl-4H-pyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++ gl KCl 25°C 0.20M C K1=8.69 B2=16.29 2000BKa (44108) 632
B((VOH-1L)2)=9.88
B(VOH-1L2)=7.5

V0++ gl KCl 25°C 0.20M C M K1=8.69 B2=16.29 1998KKe (44109) 633
B((VOH-1L)2)=9.88
B(VOH-1L2)=7.5
B(VOLA)=12.5(A=AMP), 13.97(ADP), 13.72(ATP), 25.32(dopamine), 25.08(DOPA),
13.92(oxlate), 15.41(citrate), 17.41(diphosphate), 16.31(triphosphate).

V0++ gl NaCl 25°C 0.15M U K1=8.80 B2=7.51 1995CGc (44110) 634

C6H6O3 HL Allomaltol CAS 644-46-2 (2688)
5-Hydroxy-2-methyl-4H-pyran-4-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl NaCl 25°C 0.16M C K1=7.90 B2=14.83 2002SSb (44130) 635
*K(VOL2)=-8.80

C6H6O4 HL Kojic acid CAS 501-30-4 (1800)
5-Hydroxy-2-(hydroxymethyl)-4H-pyran-4-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KCl 25°C 0.20M C K1=7.63 B2=14.37 2000BKa (44256) 636
B((VOH-1L)2)=7.93
B(VOH-1L2)=5.9

V0++ gl NaCl 25°C 0.15M U K1=7.61 B2=14.50 1997YCa (44257) 637

C6H6O5S H3L CAS 7134-09-0 (3687)
3,4-Dihydroxybenzenesulfonic acid; (HO)2.C6H3.SO3H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KN03 20°C 0.10M U K1=16.7 B2=31.20 1971ZBa (44289) 638

C6H6O6 HL CAS 490-83-5 (2575)
Dehydroascorbic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl NaClO4 25°C 0.20M U K1=7.20 1976KKe (44293) 639

C6H6O8S2 H4L Tiron CAS 149-45-1 (104)
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)2.C6H2(SO3H)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KCl 25°C 0.20M C K1=16.47 B2=31.22 1991Bkb (44512) 640
B((VO)H2L3)= 52.5
B((VO)2H-1L4)= 51.9

V0H2L3=V(IV)L3.H2O containing "bare" V(IV).

V0++ gl KN03 20°C 0.10M U K1=16.8 B2=31.20 1971ZBa (44513) 641

V0++ gl KCl 30°C 0.10M U M K1=16.61 1967LAd (44514) 642
K(VO(phen)+L)=17.19

V0++ gl KN03 25°C 0.10M U T H K1=16.74 B2=30.94 1966MMb (44515) 643
K(VO(OH)L+H=VOL)=6.3
K(2VO(OH)L=(VO(OH)L)2)=4.3
K1=16.05(35C), DH(K1)=-10.5 kJ mol-1, DS=280 J K-1 mol-1

V0++ gl oth/un 25°C 0.10M U K1=15.88 1960GRa (44516) 644

V0++ gl KN03 25°C 0.10M U K1=17.2 1959CGa (44517) 645
K((VOOHL+H=VOL)=5.1

C6H8N2 L 2-Picolylamine CAS 29722-36-9 (502)
2-(Aminomethyl)pyridine; C5H4N.CH2NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl NaNO3 25°C 0.10M C K1=6.75 B2=12.04 1994DHa (45360) 646

C6H8O6 H2L Ascorbic acid CAS 50-81-7 (285)
Ascorbic acid (Vitamin C);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl mixed 25°C 80% U 1980KKd (45665) 647
K(VO+HL)=4.6

Medium: 80% DMF

V0++ gl NaClO4 25°C 0.20M U K1=2.18 1976KKe (45666) 648

C6H8O7 H3L Citric acid CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KCl 25°C 0.20M C K1=7.85 1995KBa (46307) 649
B((VO)HL)=10.65
B((VO)2H-1L2)=15.71
B((VO)2H-2L2)=10.73
B((VO)2H-3L2)=1.6

B((VO)H-2L)=-5.7, B((VO)H-2L2)=-2.89

V0++	sp	NaCl	20°C	1.0M	U	K1=8.83 K(2V0+L)=11.50	1967NNd (46308) 650	

C6H9N05	H2L	N-Acetyl-Asp	CAS 997-55-7	(7440)				
N-Acetyl-aminobutanedioic acid; N-acetyl-L-aspartic acid;								
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo	
V0++	gl	KCl	25°C	0.20M	C	K1=2.7 B(VOHL)=6.2 B(VOH-2L)=-7.32 B(VOHL2)=9.56	1998PGb (46354) 651	

C6H9N06	H3L	NTA	CAS 139-13-9	(191)				
Nitrilotriethanoic acid; N(CH ₂ .COOH) ₃								
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo	
V0++	gl	KCl	25°C	0.20M	C	K1=13.18 B(VOHL)=15.69 B(VOH-1L)=6.07	1999SBb (47088) 652	
By spectrophotometry: K1=12.64								
V0++	sp	NaClO ₄	20°C	0.10M	U	M	K1=13.94	1998MKb (47089) 653
V0++	EMF	NaClO ₄	25°C	0.10M	U		1985NSa (47090) 654	K((VO)H-1L+H)=7.0
V0++	gl	KNO ₃	25°C	0.10M	U		1983FSa (47091) 655	
V0++	gl	NaClO ₄	25°C	0.50M	U		1977NAa (47092) 656	K(VOL)=-7.15
V0++	sp	oth/un	25°C	1.0M	U		K1=15.34	1975TPa (47093) 657
Medium: (NH ₄) ₂ SO ₄								
V0++	gl	KNO ₃	25°C	0.10M	U T	T	K1=10.82 K(VO(OH)L+H=VOL)=7.23 K(VO(OH)2L+2H=VOL)=12.81	1973STc (47094) 658
30 C: K1=10.70, K(VO(OH)L+H=VOL)=7.17, K(VO(OH)2L+2H=VOL)=12.97								
V0++	gl	NaClO ₄	25°C	0.10M	U		1966KFc (47095) 659	K(VOLOH+H=VOL)=7.38

C6H9N3O2	HL	Histidine	CAS 71-00-1	(1)				
2-Amino-3-(4'-imidazolyl)propanoic acid; H ₂ N.CH(CH ₂ .C ₃ H ₃ N ₂).COOH								
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference ExptNo	

V0++	gl	NaNO ₃	25°C	2.25M	C	B2=15.74 B((VO)H ₂ L)=17.4 B((VO)HL)=13.6 B((VO)H ₄ L ₂)=33.5 B((VO)H ₃ L ₂)=30.2 B((VO)H ₂ L ₂)=26.3, B((VO)HL ₂)=21.85, B((VO)H-2L)=-5.0, B((VO)H-1L ₂)=6.4, B((VO)H-4L ₂)=-7.8	1994CLa (47627)	660
V0++	gl	KNO ₃	25°C	0.10M	C	T K1=9.04 B2=15.48 B(VOHL ₂)=21.42 B(VOH ₂ L ₂)=26.0 B(VOH-1L)=3.48	1976PSb (47628)	661
V0++	gl	KNO ₃	25°C	0.10M	C	K1=9.06 B2=15.49 B(VOHL ₂)=21.33 B(VO(HL) ₂)=25.9 B(VOH-1L)=3.43	1976PSb (47629)	662

Ligand: D-His

C6H10N2O5 H2L Asp-Gly CAS 3790-51-0 (6521)
Aspartyl-glycine; H2N.CH(CH₂.COOH)CO.NH.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=6.42 B(VOHL)=10.46 B(VOH ₂ L)=13.4 B(VOH-1L)=0.83	1998PGb (47760)	663
V0++	gl	KCl	25°C	0.20M	C			B(VOHL)=11.52 B(VOH ₂ L)=15.1 B(VOH-1L)=1.7 B(VOH ₂ L ₂)=22.5	1998PGb (47780)	664
								B(VOH ₃ L)=26.6, B(VOHL ₂)=17.1		

C6H10O4 H2L Adipic acid CAS 124-04-9 (401)
1,6-Hexanedioic acid; HOOC.(CH₂)₄.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	EMF	NaClO ₄	30°C	0.10M	U	M		B((VO)LA)=9.68 B((VO)LB)=11.67 B((VO)LC)=9.36	1977SJ _a (48100)	665

H2A=malonic acid, H3B=5-sulphosalicylic acid, H2C=3,5-dinitrosalicylic acid

C6H1004S2 H2L CAS 7244-02-2 (438)
1,2-Bis(carboxymethylthio)ethane; HOOC.CH2.S.CH2.CH2.S.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO4	25°C	0.50M	C	M			1981NAe (48253)	666

B(VOL)=2.68

C6H1006 H2L CAS 14812-53-4 (5460)
2,3-Dimethyl-2,3-dihydroxybutanedioic acid; HOOC(OH)(CH3)C(OH)(CH3)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	none	25°C	0.0	M				1982H0c (48362)	667

K((VO)2L2+H)=6.94
K(H(VO)2L2+H)=5.21

For the L-isomer, K((VO)2L2+H)=7.19, K(H(VO)2L2+H)=5.24

C6H1007 HL Galacturonic CAS 685-73-4 (290)
D-Galacturonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO3	25°C	0.10M	C			K1=4.57 B2= 8.75 2001GLa (48397)	668	

B(VOH-1L)=1.46
B(VOH-2L2)=-0.76
B(VOH-3L2)=-8.31
B(VOH-4L2)=-17.64

B((VO)2H-4L2)=-4.20, B((VO)2H-5L2)=-10.48.

C6H1007 HL Glucuronic acid CAS 6556-12-3 (599)
D-Glucuronic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO3	25°C	0.10M	C			K1=3.28 B2= 6.23 2001GLa (48424)	669	

B(VOH-1L)=-0.28
B(VOH-4L2)=-22.89
B((VO)2H-4L2)=-6.91
B((VO)2H-5L2)=-15.75

C6H1008 H2L Mucic acid CAS 526-99-8 (3650)
2,3,4,5-Tetrahydroxyhexanedioic acid, Galactaric acid; HOOC.(CHOH)4.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=3.56 B((VO)2H-2L2)=3.08	2004DGa (48441)	670

B((VO)2H-3L2)=-1.00
 B((VO)2H-4L2)=-7.01

C6H10O8 H2L Saccharic acid CAS 87-73-0 (1191)
 D-2,3,4,5-Tetrahydroxy-1,6-hexanedioic acid; Glucaric acid; HOOC.(CHOH)4.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C				2004DGa (48490)	671
								B(VOHL)=6.84		
								B((VO)2H-2L2)=5.87		
								B((VO)2H-3L2)=0.65		
								B((VO)2H-4L2)=-5.55		

C6H11NO2 H2L CAS 2044-64-6 (4374)
 N,N-Dimethylacetamide; CH₃.CO.CH₂.CO.N(CH₃)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C				2001CKa (48543)	672
								K(VO+HL=VOL+H)=-2.34		
								K(VO+2HL=VOL2+2H)=-6.46		

C6H11NO4 H2L (1232)
 2,2'-Iminodipropionic acid; HN(CH(CH₃)COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO ₃	25°C	0.10M	C			K1=9.4	1987AKa (48577)	673

C6H11NO4 H2L CAS 103954-11-6 (5805)
 N-(1-Carboxyethyl)-alanine; HOOC.CH(CH₃).NH.CH₂.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO ₃	25°C	0.10M	C			K1=9.54	1984FVa (48594)	674
								K(VO(OH)L+H)=6.1		
								K((VO)2(OH)2L2+2H)=9.2		

C6H11NO5 H2L CAS 50825-12-2 (5806)
 N-(1-Carboxyethyl)-N-hydroxy-alanine; HOOC.CH(CH₃).N(OH).CH₂.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO ₃	25°C	0.10M	C			K1=7.34 B2=12.85	1984FVa (48625)	675
								K(VO(OH)L+H)=5.0		
								K((VO)2(OH)2L2+2H)=-6.6		

C6H11NO5 H2L HIMDA CAS 93-62-9 (192)
 N-(2-Hydroxyethyl)iminodiethanoic acid; HO.CH₂.CH₂.N(CH₂.COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=2.61 B2=5.05 B((VO)H-1L)=-1.09 B((VO)H-1L2)=1.03 B((VO)H-2L2)=-2.85	1993MSa (49468)	685

C6H12O7		HL		Gluconic acid	CAS	526-95-4	(904)			
D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH2(CHOH)4.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	oth/un	25°C	?	M			K1=3.34 K'=5.26	1976GSc (49767)	686
Metal ion: VO(OH)2. K': VO(OH)2 + 2L = VO(OH)(H-1L)2										
C6H13N04		HL		Bicine	CAS	150-25-4	(2124)			
N,N-Bis(2-hydroxyethyl)glycine; (HO.CH2.CH2)2N.CH2.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO4	30°C	0.10M	U T H			K1=6.42 B2=11.87	1980SGh (50417)	687
Also data at 20 and 40 C. DH(B2)=-56.7 kJ mol-1, DS(B2)=38.0 J K-1 mol-1.										
C7H4N2O7		H2L						CAS 609-99-4 (400)		
3,5-Dinitrosalicylic acid; (O2N)2.C6H2(OH).COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	EMF	NaClO4	30°C	0.10M	U	M			1977SJa (52507)	688
B((VO)LA)=11.86 B((VO)LB)=14.30										
H2A=5-sulphosalicylic acid, H3B=4-hydroxysalicylic acid										
V0++	gl	KNO3	35°C	0.10M	U			K1=6.8 B2=12.40 K3=2.5	1970DDc (52508)	689

C7H5N04		H2L		Quinolinic acid	CAS	89-00-9	(567)			
2,3-Pyridinedicarboxylic acid; C5H3N.(COOH)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO4	25°C	0.50M	C			K1=5.95 B2=11.00	1987NMB (52634)	690
pK values of ligand: in Annali di Chimica, 76, 277 (1986)										
C7H5N04		H2L						CAS 499-80-9 (566)		
2,4-Pyridinedicarboxylic acid; C5H3N.(COOH)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

V0++ gl NaCl04 25°C 0.50M C K1=5.20 B2=9.85 1987Nmb (52654) 691

C7H5N04 H2L CAS 100-26-5 (2528)
2,5-Pyridinedicarboxylic acid; Isocinchomeric acid; C5H3N.(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++ gl NaCl04 25°C 0.50M C K1=4.90 B2=8.85 1987Nmb (52670) 692

C7H5N04 H2L Dipicolinic aci CAS 449-83-2 (418)
2,6-Pyridinedicarboxylic acid; C5H3N.(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++ gl KCl 25°C 0.20M C K1=6.69 B2= 9.47 2003JJa (52819) 693
B(VOH-1L)=-0.03
B(VOHL2)=12.96
B(VOH-1L2)=2.62

V0++ gl NaCl04 25°C 0.50M C K1=6.70 B2=9.53 1987Nmb (52820) 694

V0++ sp NaCl04 25°C 1.0M C K1=6.77 1982FFa (52821) 695

C7H5N05 H2L Nitrosalicylic CAS 96-97-9 (148)
2-Hydroxy-5-nitrobenzoic acid; HO.C6H3(NO2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++ sp oth/un ? ? U 1971KHb (53058) 696
K(VO+H2L=VOL+2H)=-3.37

C7H5N05 H3L CAS 499-51-4 (3150)
4-Hydroxypyridine-2,6-dicarboxylic acid; HO.C5H2N(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++ gl KCl 25°C 0.20M C K1=10.43 B2=17.96 2003JJa (53080) 697
B(VOHL)=14.52
B(VOH2L2)=27.36
B(VOHL2)=23.32

C7H6N03Br H2L CAS 87353-69-3 (207)
4-Bromosalicylhydroxamic acid; Br.C6H3(OH).CO.NH.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++ gl diox/w 30°C 50% U K1=7.88 1977DJb (53398) 698

C7H6N03Br H2L CAS 5798-94-7 (206)

5-Bromosalicylhydroxamic acid; Br.C₆H₃(OH).CO.NH.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
V0++	gl	diox/w	30°C	50%	U			K1=8.04		1977DJb (53409)	699

C7H6NO3Cl H2L (205)
 3-Chlorosalicylhydroxamic acid; Cl.C6H3(OH).CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl dioxygen 30°C 50% U K1=7.52 1977DJb (53419) 700

C₆H₆N₀3Cl HL (6263)
4-Chlorosalicylhydroxamic acid; Cl.C₆H₃(OH).CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ g1 dioxygen 30°C 50% U K1=6.02 1977DB (5342) 701

C7H6NO3Cl
5-Chloro-1,1-dimethyl-3-hydroxy-1,2-dihydro-1,4-dioxin-2-one
CAS 37551-43-2 (6262)

5-Chlorosalicylhydroxamic acid; Cl.C₆H₃(OH).CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl dioxygen 30°C 50% U KI=8.34 1977DB (53425) 702

C₇H₆N₂O₅ H₂L CAS 831-51-6 (208)
5-Nitro-2-(2-hydroxyethylamino)-2-oxo-2H-1,3-dioxole-4-carboxylic acid (22N, SGH2(OH), GS, NH, OH)

5-Nitrosalicylhydroxamic acid; O₂N.C₆H₃(OH).CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ g1 diox/w 30°C 50% U K1=6.02 1977Db (53525) 703

C/H6N2S HL CAS 583-39-1 (2043)
2-Mercaptohexanimidomalon

2-Mercaptobenzimidazole;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ g1 alc/w 25°C 50% U K1=9.05 1978Zia (53534) 704

C7H602S H2L Thiosalicylic CAS 147-93-3 (236)

2-Mercaptobenzoic acid; HS.C6H4.CO₂H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ g1 alc/w 25°C 50% U K1=10.24 B2=18.63 1971RFb (53924) 705

C7H6O3 H2L Salicylic acid CAS 69-72-7 (14)
3-Hydroxybenzoic acid

2-Hydroxybenzoic acid, Salicylic acid; HO.C₆H₄.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C	M	K1=12.97 B(VOH-1L2)=13.16 B(VOH-1L)=6.32 B(VO2H-2L2)=16.61	1990JKb (54331)	706	

B(VOLA)=27.66, where H2A=catechol

V0++	gl	NaClO4	25°C	0.10M	U	I	K1=12.68 B((VO)HL)=14.68	1987GMa (54332)	707
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I=0.4: K1=12.52, B((UO2)HL)=14.88; I=0.7: K1=12.56, B((UO2)HL)=15.25

V0++	gl	KNO3	30°C	0.10M	M	M	K1=13.18 K(VO+H2L=VOL+2H)=-3.41 K(VO+HL+A)=14.78 K(VOL+A)=1.60	1981VSd (54333)	708
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HA is hippuric acid

V0++	gl	KNO3	35°C	0.10M	U	T	HM	1978JKa (54334)	709
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B(VO(bpy)L)=27.78

Data for 45 C. DH and DS values reported.

V0++	gl	KNO3	25°C	0.10M	U	M		1978JSb (54335)	710
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B(VO(Iminodiethanoate)L)=18.35

V0++	gl	KNO3	30°C	0.10M	U	M	K1=13.18 K(VOL(OH)+H=ML)=4.63 K(VOL+Sulfoxine)=12.7	1975STb (54336)	711
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V0++	gl	KNO3	20°C	0.10M	U		K1=12.7	B2=22.40	1971ZBa (54337)	712
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V0++	gl	KNO3	35°C	0.10M	U	T	H	K1=12.89	1966MMb (54338)	713
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K1=13.38(25 C). DH(K1)=-8.8 kJ mol-1, DS=226 J K-1 mol-1

V0++	sp	oth/un	?	0.05M	U		K1=15.4	1962LZa (54339)	714
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C7H6O4 H3L CAS 303-38-8 (1398)

2,3-Dihydroxybenzoic acid; C6H3(OH)2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++	gl	KCl	25°C	0.20M	C	T	K1=9.97 B(VOH-1L)=4.02 B(VOH-1L2)=10.46 B(VO2H-2L2)=12.86 B(VOH-2L2)=2.00	1990JKb (54473)	715
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B(VOH-2L)=-2.88, B(VOH-3L)=-14.13

C7H6O4 H3L Resorcylic acid CAS 89-86-1 (876)

2,4-Dihydroxybenzoic acid, *b*-Resorcylic acid; C₆H₃(OH)₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		T	K1=8.50 B(VOH-1L2)=5.93 B(VOH-1L)=1.48 B(VO2H-2L2)=8.60	1990JKb (54552)	716
V0++	gl	KNO ₃	35°C	0.10M	U				1970DDc (54553)	717

C7H6O₄ H₃L CAS 409-79-9 (1115)

2,5-Dihydroxybenzoic acid; C₆H₃(OH)₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		T	K1=9.61 B(VOH-1L2)=7.48 B(VOH-1L)=2.49 B(VO2H-2L2)=10.58	1990JKb (54592)	718

C7H6O₄ H₃L *g*-Resorcylic ac CAS 303-07-1 (1624)

2,6-Dihydroxybenzoic acid; C₆H₃(OH)₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		T	K1=12.25 B(VOH-1L2)=12.04 B(VOH-1L)=4.96 B(VO2H-2L2)=22.92	1990JKb (54608)	719

C7H6O₄ H₃L Protocatechuic CAS 99-50-3 (875)

3,4-Dihydroxybenzoic acid; C₆H₃(OH)₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=17.13 B(VOH-1L)=11.23 B(VOHL)=21.54 B(VO2H-2L2)=24.36 B(VOH2L2)=40.62	1990JKb (54707)	720

B(VOHL2)=36.61

V0++	gl	NaClO ₄	25°C	0.10M	U	I	K1=16.63	B2=20.83	1987GMa (54708)	721
I=0.4: K1=16.34, B2=20.65; I=0.7: K1=16.26, B2=20.76										

C7H6O₅ H₄L CAS 610-02-6 (3725)

2,3,4-Trihydroxybenzoic acid; (HO)₃.C₆H₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=8.72 B((VO)HL)=11.36 B((VO)H-1L)=4.49 B((VO)H-2L)=-1.41 B((VO)4H-8L4)=4.89	1994KMa (54725)	722

B((VO)H-1L2)=10.24, B((VO)H-2L2)=4.88

C7H6O5 H4L Gallic acid CAS 149-91-7 (446)
3,4,5-Trihydroxybenzoic acid; C6H2(OH)3.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	alc/w	20°C	50%	U	I			1973CSc (54771)	723

K(VO+HL)=4.10
K(VO+2HL)=7.40
K(VO+3HL)=10.44

Medium: 25-100% MeOH, 0.01 M LiNO3

K(VO+HL)(100%)=4.20; K(VO+2HL)(100%)=7.62; K(VO+3HL)=10.87

C7H6O6S H3L CAS 5965-83-3 (399)
5-Sulfosalicylic acid, 2-Hydroxy-5-sulfobenzoic; HO3S.C6H3(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KN03	30°C	0.10M	M	M		K1=11.37 K(VO+H2L=VOL+2H)=-2.73 K(VO+HL+A)=13.54 K(VOL+A)=2.17	1981VSd (55074)	724

HA is hippuric acid

V0++	gl	KN03	35°C	0.10M	U	T	HM		1978JKa (55075)	725
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B(VO(bpy)L)=22.93

Data for 45 C. DH and DS values reported.

V0++	gl	KN03	25°C	0.10M	U	M			1978JSb (55076)	726
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B(VO(Iminodiethanoate)L)=18.35

V0++	gl	KN03	30°C	0.10M	U	M		K1=11.73 K(VOL(OH)+H)=4.34 K(VOL+Sulfoxine)=10.7	1975STb (55077)	727
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K(VOL(OH)+H)=4.34

K(VOL+Sulfoxine)=10.7

V0++	gl	KN03	20°C	0.10M	U			K1=12.0	B2=20.60	1971ZBa (55078)	728
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V0++	gl	KN03	25°C	0.10M	U	T	H	K1=11.71 K(VO(OH)L+H)=7.22 K(2VO(OH)L=(VO(OH)L)2)=5.33	1966MMb (55079)	729
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K(VO(OH)L+H)=7.22

K(2VO(OH)L=(VO(OH)L)2)=5.33

K1=11.29(35 C); DH(K1)=-7.1 kJ mol-1, DS=188 J K-1 mol-1

C7H7N02 HL Anthranilic CAS 118-92-3 (1589)
2-Aminobenzoic acid, Anthranilic acid; H2N.C6H4.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KNO₃ 30°C 0.10M M 1975STc (55270) 730
 $K(VO+H_2L=VO(OH)L+H)=-6.32$

C7H7N02 HL CAS 3222-47-7 (3154)
6-Methylpyridine-2-carboxylic acid; CH₃.C5H₃N.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KCl 25°C 0.20M C M 2000KGd (55434) 731
 $B((VO)AL)=8.21$
 $B((VO)H-1AL)=3.28$
 $B((VO)BL)=15.40$
 $B((VO)HBL)=20.94$

HA is lactic acid, H₃B is phosphoric acid.

V0++ gl KCl 25°C 0.20M C M K1=5.13 B2= 9.28 2000KGd (55435) 732
 $B((VO)HL)=7.27$
 $B((VO)2H-2L2)=3.25$
 $B((VO)H-2L)=-6.56$
 $B((VO)AL)=11.22$

$B((VO)BL)=13.05$, $B((VO)HBL)=16.19$, $B((VO)H-1BL)=7.39$.

H₂A is oxalic acid, H₃B is citric acid.

C7H7N03 H2L CAS 89-73-6 (204)
2-Hydroxybenzohydroxamic acid (salicylhydroxamic acid); HO.C6H₄.CO.NHOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl diox/w 30°C 50% U K1=8.90 1977DJb (55617) 733

C7H7O6P H3L CAS 6064-83-1 (822)
Salicyl phosphate; HO.C6H₄.CO₂.PO₃H₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KNO₃ 25°C 0.10M U T H K1=5.81 1966MMb (55777) 734
 $K(VO(OH)L+H)=5.7$
 $K(2VO(OH)L=(VO(OH)L)2)=2.3$

At 35 C: K1=5.68; DH(K1)=-2.1 kJ mol⁻¹, DS=104 J K⁻¹ mol⁻¹

C7H8O3 HL CAS 2298-99-9 (8830)
3-Hydroxy-2,6-dimethyl-4H-pyran-4-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl NaCl 25°C 0.16M C K1=9.23 B2=17.06 2002SSb (56105) 735
*K(VOL2)=-8.70

C7H10N2O3S HL CAS 71691-06-0 (1247)
2-(N-Pyrrolideneimino)ethane sulfonic acid; C4H4N.CH:N.CH2.CH2.SO3H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl NaClO4 25°C 0.10M U T K1=10.80 B2=19.10 1979GSa (56694) 736

C7H11N06 H3L MNTA (1026)
Nitrilo(2-propanoic)-diethanoic acid; HOOC.CH(CH3).N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KN03 25°C 0.10M U K1=11.77 1983FSa (56921) 737

C7H12N2O2 HL (6181)
2-(N-2-Pyrrolidinimino)propanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl NaClO4 25°C 0.10M U TIH B2=22.40 1988GRb (57075) 738
35 C:B2=24.52, 45 C:24.61. DH(B2)=19.1 kJ mol-1, DS=531.3 J K-1 mol-1

C7H12O2 HL CAS 7424-54-6 (4421)
Heptane-3,5-dione; CH3.CH2.CO.CH2.CO.CH2.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ sp NaClO4 25°C 0.5M C K1=9.39 1998BLa (57248) 739

V0++ EMF NaClO4 25°C 0.50M U K1=9.39 1990BHa (57249) 740

C7H12O6 HL Quinic acid CAS 77-95-2 (2578)
1,3,4,5-Tetrahydroxycyclohexane-1-carboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KN03 25°C 0.10M C K1=2.83 B2= 5.42 2001GLa (57413) 741
B(VOH-1L)=0.32
B(VOH-2L2)=-0.74
B(VOH-3L2)=-9.92
B(VOH-4L2)=-21.31

V0++ gl NaCl 25°C 0.15M C 2000AFb (57414) 742
K(VO+H-1L)=11.56
K(VO+2H-1L)=22.01
K(VO+H-1L=VOH-2L+H)=6.28

$$K(VO+2H-1L=VOH-3L2+H)=11.34$$

$$K(VO+2H-1L=VOH-4L2+2H)=0.54$$

C8H5O2F3S HL TTA CAS 326-91-0 (165)
4,4,4-Trifluoro-1-(2-thienyl)butane-1,3-dione; F3C.CO.CH2.CO.C4H3S

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ sp NaClO4 25°C 1.0M U 1980HRa (58695) 743
K(VO+HL=VOL+H)=-0.49

C8H6O4 H2L Phthalic acid CAS 88-99-3 (113)
Benzene-1,2-dicarboxylic acid; C6H4(COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl NaClO4 20°C 0.1M U 1999MKc (59031) 744
K(VO+OH+L)=16.14

V0++ gl NaClO4 25°C 0.50M C K1=3.73 1989NMa (59032) 745

V0++ gl NaClO4 25°C 0.10M U I K1=3.97 B2=6.39 1987GMa (59033) 746
B((VO)HL)=6.28
I=0.4: K1=3.68, B2=5.85, B((VO)HL)=6.10. I=0.7: K1=3.61, B((VO)HL)=6.97

V0++ gl NaClO4 25°C 0.50M C K1=3.73 1987NMb (59034) 747

V0++ gl KN03 35°C 0.10M U T HM 1978JKa (59035) 748
B(VO(bpy)L)=10.97

Data for 45 C. DH and DS values reported.

V0++ gl KN03 30°C 0.10M M K1=4.01 1978SVa (59036) 749

C8H6O6 H4L (6671)
2,3-Dihydroxybenzene-1,4-dicarboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KCl 25°C 0.20M C K1=3.2 1994KMa (59078) 750
B((VO)H-1L)=0.31
B((VO)2H-3L2)=-0.5
B((VO)3H-6L3)=-4.39
B((VO)H-3L2)=-10.86

Ligand as H2L

C8H7N02Cl2 HL CAS 13538-26-6 (6286)
3,5-Dichloro-2-hydroxyacetophenone oxime; Cl2(HO)C6H2.C(CH3):NOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl alc/w 27°C 75% U I K1=8.70 B2=16.65 1976LGa (59120) 751
Data in 75% EtOH. Data also in 75% acetone and 75% dioxan

C8H8O3 HL Mandelic Acid CAS 611-72-3 (80)
2-Phenyl-2-hydroxyethanoic acid; C6H5.CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KCl 25°C 0.20M C K1=2.60 1993MSa (59886) 752
B((VO)H-1L)=-1.04
B((VO)H-1L2)=1.52
B((VO)H-2L2)=-3.10

V0++ gl KN03 25°C 0.10M U M 1978JSb (59887) 753
B(VO(Iminodiethanoate)L)=17.52

C8H9N02S HL CAS 6310-11-8 (4576)
3-Mercaptoacetamidophenol; HS.CH2.CO.NH.C6H4.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl oth/un 17°C ? U K1=6.38 B2=12.22 1973KPd (60387) 754

C8H9N03 H2L CAS 26071-07-8 (209)
5-Methylsalicylhydroxamic acid; CH3.C6H3(OH).CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl diox/w 30°C 50% U K1=9.22 1977DJb (60440) 755

C8H11N02 H2L Dopamine CAS 579-59-9 (251)
2-(3',4'-Dihydroxyphenyl)ethylamine; (HO)2.C6H3.CH2.CH2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KCl 25°C 0.20M C K1=21.34 B2=31.75 1993BDc (61087) 756
B((VO)HL)=27.10
B((VO)H2L2)=52.65
B((VO)HL2)=42.42
B((VO)H5L3)=87.26

K(VO+2HL)=25.45, *K(VOH2L2)=-10.23, *K(VOHL2)=-10.67.

C8H11N03 H2L Noradrenaline CAS 138-65-8 (253)
Norepinephrine, 3,4-Dihydroxyphenylethanolamine; (HO)2C6H3.CH(CH2.NH2).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KCl 25°C 0.20M C B2=31.40 1993BDc (61169) 757
B((VO)HL)=26.09
B((VO)H2L2)=50.35

$B((V0)HL2)=41.09$
 $B((V0)H5L3)=83.60$
 $K(V0+2HL)=24.26, *K(V0H2L2)=-9.26, *K(V0HL2)=-9.69.$

V0++ sp NaCl 25°C 0.20M U 1978CFa (61170) 758
 K1eff=2.45
 B2eff=7.43

Medium: tris buffer, pH 7.4; 0.1 M NaCl, 0.02 M KCl

C8H12O7P2 H4L (7244)
 1-Hydroxy-2-phenylethane-1,1-diphosphonic acid; HO.C(PO(OH)2)2.CH2C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KN03 25°C 0.10M C K1=7.02 1998DKa (61740) 759

C8H16N204 H2L (267)
 1,2-Diaminoethane-N,N'-di(2-propanoic acid); ((CH3)(COOH).CH.NH.CH2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KN03 25°C 0.10M U K1=13.34 1983FSa (62477) 760

C9H7NO HL Oxine CAS 148-24-3 (504)
 8-Hydroxyquinoline (8-quinolinol);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ sp NaClO4 25°C 1.00M U K1=12.4 B2=21.0 1979YYa (64372) 761

V0++ gl NaClO4 25°C 0.10M U 1966KFc (64373) 762
 $K(V0(OH)L+H=VOL)=5.3$

V0++ gl oth/un 25°C .085M U K1=10.97 B2=20.19 1957TBa (64374) 763

C9H7N03S2 H2L CAS 58447-10-2 (4675)

8-Mercaptoquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ sp oth/un ? ? U B2=7.8 1968ABA (64432) 764

C9H7N04S H2L Sulfoxine CAS 84-88-8 (448)

8-Hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KN03 30°C 0.10M M M K1=10.48 1981VSd (64585) 765

$K(V0+H2L=VOL+2H)=-2.07$

$K(V0+HL+A)=12.55$

$$K(VOL+A)=2.07$$

HA is hippuric acid

V0++ sp none 25°C 0.0 U K1=11.8 1980D0a (64586) 766

V0++ gl KN03 30°C 0.10M U M K1=10.48 1975STb (64587) 767
 $K(VO(salicylate)+L)=12.7$
 $K(VO(sulfosalicylate)+L)=10.7$
 $K(VO(maleate)+L)=5.5$
 $K(VO(phthalate)+L)=3.4$

V0++ gl KN03 25°C 0.10M U T H K1=11.79 1966MMb (64588) 768
 $K(VO(OH)L+H=VOL)=6.45$
 $K(2VO(OH)L=(VO((OH)L)2)=4.84$

$K1=11.32(35\text{ }^{\circ}\text{C})$. $DH(K1)=-7.1\text{ kJ mol}^{-1}$, $DS=188\text{ J K}^{-1}\text{ mol}^{-1}$

C9H7N3O2S H2L TAR CAS 2246-46-0 (707)
4-(2'-Thiazolylazo)-resorcinol; C3H2NS.N:N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	alc/w	25°C	50%	U				1967NPb (64738)	769
								$K(VO+HL)=11.2$		
								$K(VO(HL)+HL)=9.8$		

Medium: 50% MeOH, 0.1 M NaClO4

C9H8O4 H3L Caffeic acid CAS 331-39-5 (6037)
3-(3,4-Dihydroxyphenyl)propenoic acid; (HO)2C6H3.CH:CH.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaCl	25°C	0.15M	U			K1=16.52 B2=29.42	2002WBb (64921)	770
								$B(VOH-1L)=11.10$		
								$B(VOH-2L)=0.79$		
								$B((VO)3L2)=39.67$		
								$B((VO)2L)=20.30$		

C9H9N03 HL Hippuric acid CAS 495-69-2 (1184)
Benzoylaminooethanoic acid, N-benzoylglycine; C6H5.CO.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KN03	30°C	0.10M	M				1975STc (65058)	771
								$K(VO+H2L=VO(OH)L+H)=-4.47$		

C9H9N04 H2L Salicylglycine CAS 487-54-7 (3869)
N-(2-Hydroxybenzoyl)glycine, 2-hydroxyhippuric acid; HO.C6H4.CO.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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VO++ gl KCl 25°C 0.20M C K1=7.05 1998KPa (65096) 772
 B(VOHL)=10.24
 B(VOH-1L)=2.29
 B(VOH-2L)=-5.28
 B(VOH-1L2)=5.55

C9H1008 H4L CAS 3724-52-5 (1264)
cis-1,2,3,4-Cyclopentanetetracarboxylic acid; C5H6.(COOH)4

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
VO++	gl	NaClO4	30°C	0.19M	U			K1=6.75 B2=11.85	1985MSb (65653)	773

C9H11N04 H3L DOPA CAS 59-92-7 (5)
 2-Amino-3-(3,4-dihydroxyphenyl)propanoic acid; H2NCH(CH₂C₆H₃(OH)2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
VO++	gl	KCl	25°C	0.20M	C			K1=21.34 B2=31.79	1993BDc (66404)	774

B((VO)HL)=26.88
 B((VO)H2L2)=51.67
 B((VO)HL2)=42.18
 B((VO)H5L3)=86.24
 B(VOH2L)=30.05, K(VO+2HL)=24.79, *K(VOH2L2)=-9.49, *K(VOHL2)=-10.39.

C9H12N205 L CAS 951-78-0 (6537)
2'-Deoxy-uracil-1-beta-D-ribofuranoside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
VO++	nmr	KCl	25°C	1.00M	U				1990TJa (66629)	775

At pH 7 in 30mM HEPES. Data also for methyl-beta-D-ribofuranoside: K=0.52,
 B=7.46

C9H12N206 HL Uridine CAS 58-96-8 (828)
 Uracil-1-beta-D-ribofuranoside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
VO++	nmr	KCl	25°C	1.00M	U				1990TJa (66716)	776

At pH 7 in 30mM HEPES
 Keff(VO+L)=0.20
 Beff((VO)2L2)=7.61

C9H13N03 H2L (-)Adrenaline CAS 51-43-4 (252)
 4-(1-Hydroxy-2-(methylamino)ethyl)-1,2-dihydroxybenzene,
 Epinephrine; CH₃NHCH(OH)C₆H₃(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++	gl	KCl	25°C	0.20M	C	B2=31.82 B((VO)HL)=26.64 B((VO)H2L2)=51.59 B((VO)HL2)=41.92 B((VO)H5L3)=85.42	1993BDc (66870)	777
						K(VO+2HL)=24.95, *K(VOH2L2)=-9.67, *K(VOHL2)=-10.10.		

V0++	gl	KNO ₃	25°C	0.10M	U	K1=21.42 B2=32.01 B((VO)HL)=26.81 B((VO)H2L2)=51.63 B((VO)HL2)=42.01	1986JKa (66871)	778
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C9H13N3O5 L Cytidine CAS 65-46-3 (2152)
Cytidine, Cytosine-1-beta-D-ribofuranoside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	nmr	KCl	25°C	1.00M	U				1990TJa (67086)	779

At pH 7 in 30mM HEPES

C9H14N4O3 HL Carnosine CAS 305-84-0 (272)
3-Alanyl-histidine; H₂N.CH₂.CH₂.CO.NH.CH(CH₂.C₃H₃N₂).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.15M	C			B2=11.2 K(VO+H3L+H2L)=ca.4 K(VO+2H2L)=ca.4 K(VO+H2L+HL)=6.5 K(VO+2HL)=8.6	2002AMc (67327)	780

K(VO+HL+L)=9.5, *K(VOH5L2)=ca.-3, *K(VOH4L2)=ca.-4, *K(VOH3L2)=-4.5,
*K(VOH2L2)=-8.5, *K(VOHL2)=-8, *K(VOL2)=-8.7.

C10H7N05S H2L CAS 3682-32-4 (1812)
2-Nitroso-1-hydroxynaphthalene-4-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++	sp	KCl	25°C	0.10M	C	K1=5.96			1975MLa (68896)	781
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V0++	gl	KNO ₃	35°C	0.10M	U	K1=5.47			1974LSa (68897)	782
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C10H7N05S H2L CAS 23525-13-6 (1813)
2-Nitroso-1-hydroxynaphthalene-5-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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V0++	sp	KCl	25°C 0.10M C	K1=6.19	1975MLa (68913) 783

C10H7N05S		H2L		CAS 31005-79-9	(1814)
2-Nitroso-1-hydroxynaphthalene-8-sulfonic acid;					
Metal	Mtd	Medium	Temp Conc	Cal Flags Lg K values	Reference ExptNo

V0++	sp	KCl	25°C 0.10M C	K1=7.19	1975MLa (68954) 784

C10H7N08S2		H3L	Nitroso-R acid	CAS 525-05-3	(1811)
1-Nitroso-2-hydroxynaphthalene-3,6-disulfonic acid;					
Metal	Mtd	Medium	Temp Conc	Cal Flags Lg K values	Reference ExptNo

V0++	sp	KCl	25°C 0.10M C	K1=6.71	1975MLa (69037) 785

C10H7N08S2		H3L		CAS 52664-45-6	(1627)
2-Nitroso-1-hydroxynaphthalene-4,6-disulfonic acid;					
Metal	Mtd	Medium	Temp Conc	Cal Flags Lg K values	Reference ExptNo

V0++	sp	KCl	25°C 0.10M C	K1=5.68	1975MLa (69056) 786

C10H8N2		L	2,2'-Bipyridyl	CAS 366-18-7	(25)
2,2'-Bipyridine; (C5H4N)2					
Metal	Mtd	Medium	Temp Conc	Cal Flags Lg K values	Reference ExptNo

V0++	gl	KCl	25°C 0.20M C	K1=4.91 B2= 8.46 B(VOH-1L)=0.2 B((VO)2H-2L2)=4.35 B(VOH-1L2)=3.84	2002JBa (69662) 787

V0++	gl	NaNO3	25°C 0.10M C	K1=5.08 B2=8.65 K(VOL2+OH)=1.23	1994DHa (69663) 788

C10H805S		H3L	DHNSA		(877)
2,3-Dihydroxynaphthalene-6-sulfonic acid;					
Metal	Mtd	Medium	Temp Conc	Cal Flags Lg K values	Reference ExptNo

V0++	gl	NaClO4	25°C 0.10M C	I K1=16.21 B2=30.64	1979LPb (69868) 789

C10H808S2		H4L	Chromotropic ac	CAS 148-25-4	(1875)
1,8-Dihydroxynaphthalene-3,6-disulfonic acid;					
Metal	Mtd	Medium	Temp Conc	Cal Flags Lg K values	Reference ExptNo

V0++	gl	KCl	30°C 0.10M U	M K1=17.17	1967LAd (69984) 790
K(VO(phen)+L)=18.09					

C10H9N03S2 HL (7206)
6-Methyl-5-sulfo-8-mercaptopquinoline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	oth/un	20°C	0.10M	U		K1=8.0		1985DAb	(70181) 791

C10H9N3		L	Dipyridylamine	CAS	1202-34-2	(2428)				
(2,2'-Dipyridyl)amine; C5H4N.NH.C5H4N										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaNO3	25°C	0.10M	C		K1=5.48	B2=9.37	1994DHa	(70342) 792

C10H10O2		HL	Benzoylacetone	CAS	93-91-4	(197)				
1-Phenylbutane-1,3-dione; C6H5.CO.CH2.CO.CH3										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	dis	oth/un	25°C	0.10M	U		K1=10.52	B2=20.55	1970MKh	(70783) 793

C10H12N203S		HL		CAS	93100-65-3	(6199)				
2-(2-Pyrrolideneamino)benzene sulfonic acid; C4H7N:N.C6H4.HSO3										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO4	25°C	0.10M	U T H		K1=19.08		1987RDb	(71212) 794
35 C:K=19.77, 45 C:20.34. DH=114.31 kJ mol-1, DS=740 J K-1 mol-1										

C10H12N204		H2L		CAS	16598-05-3	(967)				
2-Pyridylmethylenimino diethanoic acid; C5H4N.CH2.N(CH2.COOH)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	EMF	NaClO4	25°C	0.10M	U				1985NSa	(71280) 795
K((V0)H-1L+H)=6.5										
V0++	gl	KNO3	25°C	0.10M	U		K1=11.3		1983FSa	(71281) 796
V0++	gl	NaClO4	25°C	0.10M	U				1966KFc	(71282) 797
K(VOL(OH)+H=VOL)=6.45										

C10H12N405		HL	Inosine	CAS	58-63-9	(2344)				
Hypoxanthine-9-beta-D-ribofuranoside;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	nmr	KCl	25°C	1.00M	U I				1990TJa	(71409) 798

$$K_{eff}(V_0+L)=0.58$$

$$B_{eff}((V_0)2L2)=7.85$$

At pH 7 in 30mM HEPES. In 0.035M KCl: $K_{eff}(V_0+L)=0.59$, $B_{eff}((V_0)2L2)=7.81$

C10H13N5O4 L Adenosine CAS 58-61-7 (2154)
Adenosine, Adenine-9-beta-D-ribofuranoside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	nmr	KCl	25°C	1.00M	U				1990TJa (71958)	799
								$K_{eff}(V_0+L)=0.76$		
								$B_{eff}((V_0)2L2)=7.61$		

At pH 7 in 30mM HEPES

C10H13N5O5 HL Guanosine CAS 118-00-3 (1402)
2-Aminopurin-6-one-9-riboside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	nmr	KCl	25°C	1.00M	U				1990TJa (72025)	800
								$B_{eff}((V_0)2L2)=7.56$		

At pH 7 in 30mM HEPES

C10H15N5O10P2 H3L ADP CAS 20398-34-9 (2181)
Adenosine-5'-diphosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			$K1=6.68$ $B2=10.80$	1995ADb (73021)	801
								$B((V_0)HL)=10.09$		
								$B((V_0)H-2L2)=-4.45$		
								$B((V_0)H-4L2)=-22.34$		
								$B((V_0)2H-2L2)=4.96$		

$B((V_0)2H-4L2)=-10.53$, $B((V_0)2H-6L2)=-28.03$.

C10H16N2O8 H4L EDDS CAS 52759-67-8 (1100)
1,2-Diaminoethane-N,N'-di-1,4-butanedioic acid; ($\text{CH}_2\text{NH}\text{CH}(\text{COOH})\text{CH}_2\text{COOH}$)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	oth	KNO ₃	25°C	0.10M	U				1972TSd (73196)	802
								$K(V_0+H2L=VOL+2H)=12.89$		
								$K(VOL+H2O=V_0(OH)L+H)=-9.60$		
								$K(VOL+2H2O=V_0(OH)2L+2H)=-18.0$		

$K(2VOL+2H2O=(V_0)2(OH)2L2+2H)=-15.08$

V0++	gl	KNO ₃	25°C	0.10M	U			$K1=12.89$	1972TSd (73197)	803
								$K(V_0(OH)L+H)=9.60$		
								$K(V_0(OH)2L+2H)=18.0$		
								$K((V_0(OH)L)2+2H)=15.08$		

C10H16N208 H4L EDTA CAS 60-00-4 (120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	cal	KNO ₃	25°C	0.10M	C	H		K1=18.63	1987AKa (74300)	804
DH(K1)=-10.5 kJ mol ⁻¹ , DS=326 J K ⁻¹ mol ⁻¹										
V0++	EMF	NaClO ₄	25°C	1.00M	U	H			1985SKa (74301)	805
K(VOL+H)=3.22										
V0++	gl	NaClO ₄	25°C	1.00M	C				1983AHa (74302)	806
K(VOL+H)=3.16										
V0++	gl	KNO ₃	25°C	0.10M	U	T		K1=18.77	1983FSa (74303)	807
V0++	sp	none	?	0.0	U			K1=18.0	1958RIa (74304)	808
K(VO+HL)=11.4										
K(VOL+H)=3.65										

V0++ vlt KCl 20°C 0.10M U T K1=18.77 1954SGa (74305) 809

C10H16N5O13P3 H4L ATP CAS 56-65-5 (403)
Adenosine-5'-triphosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C			K1=6.49 B2=10.42	1995ADb (74838)	810
B((VO)HL)=10.00										
B((VO)H-2L2)=-3.76										
B((VO)H-4L2)=-22.57										
B((VO)2H-2L2)=3.82										
B((VO)2H-4L2)=-9.27, B((VO)2H-6L2)=-28.84.										

V0++ gl NaClO₄ 25°C 0.10M U K1=6.67 B2=10.32 1989CGb (74839) 811
B((VO)2L)=10.21

C10H17N08S HL (1735)
2-(5-Carboxy-1,2,3,4-tetrahydroxypentyl)4-carboxythiazolidine,
Galactocarboxythiazolidine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO ₄	25°C	0.10M	C			K1=5.73 B2=9.88	1992GNa (75014)	812
B((VO)HL)=8.33										
B((VO)H-1L)=2.14										
B((VO)H-1L2)=4.34										
B((VO)H-2L2)=-2.40										
B((VO)H-3L2)=-11.89, B((VO)2H-3L2)=1.46, B((VO)2H-4L2)=4.96,										

$B((VO)2H-5L2)=-13.68$, $B((VO)2H-6L2)=-23.88$

C10H17N3O6S H3L Glutathione CAS 70-18-8 (333)
Glutamyl-cysteinyl-glycine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	1.5M	C				2001AMB (75149)	813
								$K(VO+2H3L)=4.7$		
								$K(VO+2H3L=VO(H2L)2+2H)=3.2$		
								$K(VO+2H3L=VO(HL)H2L+3H)=>-8.4$		
								$K(VO+2H3L=VO(HL)2+4H)=-13.7$		

C10H18N2O7 H3L HEDTA CAS 150-39-0 (392)
N-(Hydroxyethyl)diaminoethane-N,N',N'-triethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO ₃	25°C	0.10M	U			K1=17.12	1983FSa (75534)	814

C10H20N2O4 H2L CAS 7532-84-5 (1027)
1,2-Diaminoethane-N,N'-di(2-(2-methyl)propanoic acid)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO ₃	25°C	0.10M	U			K1=12.23	1983FSa (75766)	815

C10H20N2O6 H2L CAS 5616-21-7 (570)
N,N'-Bis(2-hydroxyethyl)diaminoethane-N,N'-diethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO ₃	25°C	0.10M	U				1959CGa (75860)	816

C11H8O4 HL CAS 6724-42-1 (6183)
8-Formyl-7-hydroxy-4-methyl-2H-1-benzopyran-2-one; CHO.C9H30(:O)(CH₃)(OH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
V0++	gl	alc/w	35°C	70%	U			K1=6.98	B2=13.52	1988Krc (77209)	817

C11H9N3O2 H2L PAR CAS 1141-59-9 (636)
4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo	
V0++	sp	oth/un	?	?	U					1973KLb (77602)	818
								$K(VOL+H)=4.8$			
								$K(VO2L+H)=4.5$			

V0++	sp	oth/un	?	?	U	1970BBg (77603) 819 K(VO(OH)+L)=16.47
*****						*****
C11H11N06		H3L				CAS 1147-65-5 (425) N-(2'-Carboxyphenyl)iminodiethanoic acid; HOOC.C6H4.N(CH2.COOH)2
*****						*****
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values Reference ExptNo
V0++	gl	KNO3	25°C	0.10M	U	K1=9.49 1983FSa (77840) 820
*****						*****
C11H12N202		HL				CAS 103314-23-4 (6182) 2-(N-2-Pyrrolidimino)benzoic acid; C4H7N:N.C6H4.COOH
*****						*****
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values Reference ExptNo
V0++	gl	NaClO4	25°C	0.10M	U TIH	B2=24.99 1988GRb (78025) 821 35 C:B2=25.10, 45 C:25.25. DH(B2)=23.6 kJ mol-1, DS=557.1 J K-1 mol-1
*****						*****
C11H12N202		HL	Tryptophan			CAS 73-22-3 (3) 2-Amino-3-(3-indolyl)propanoic acid; H2N.CH(CH2.C8H6N)COOH
*****						*****
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values Reference ExptNo
V0++	gl	NaClO4	20°C	0.10M	U T H	K1=7.58 B2=14.81 1981SSh (78239) 822 Also data for 30 and 40C. DH(B2)=111 kJ mol-1, DS(B2)=663 J K-1 mol-1
*****						*****
C11H14N204		H2L				CAS 511537-84-1 (8567) N-[(2-Hydroxyphenyl)methyl]glycyl-glycine;
*****						*****
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values Reference ExptNo
V0++	gl	KCl	25°C	0.20M	C	K1=11.60 B2=18.47 2002PCa (78903) 823 B((VO)H2L)=19.2 B((VO)H-1L)=6.17 B((VO)H-2L)=-2.09
*****						*****
C11H18N208		H4L				CAS 4408-81-5 (923) 1,3-Diaminopropane-N,N,N',N'-tetraethanoic acid; ((HOOC.CH2)2N.CH2.)2.CH2
*****						*****
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values Reference ExptNo
V0++	gl	KNO3	20°C	0.10M	U	K1=18.97 1977SJJa (79477) 824
*****						*****
C12H8N2		L	Phenanthroline			CAS 66-71-7 (144) 1,10-Phenanthroline;
*****						*****
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values Reference ExptNo

V0++ gl NaNO₃ 25°C 0.10M C K1=5.48 B2=10.25 1994DHa (80529) 825
K(VO₂+OH)=5.12

V0++ gl KCl 30°C 0.10M U M K1=5.88 1967LAd (80530) 826
K(VO(OH)L+H)=3.04

Ternary complexes with catechol, tiron and chromotropic acid

V0++ gl oth/un 25°C 0.08M U K1=5.47 B2=9.69 1957TBa (80531) 827

C12H9N03 HL CAS 63098-85-1 (6279)
2-(N-2'-Furfuralideneimino)benzoic acid; C4H3O.CH:N.C6H4.CO0H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl NaClO₄ 25°C 0.10M U TI K1=4.43 B2=8.02 1978SKg (80583) 828

C12H10N2O2 H2L CAS 2050-14-8 (3378)
2,2'-Dihydroxyazobenzene; HO.C6H4.N:N.C6H4.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl diox/w 25°C 75% U K1=22.2 1998FHa (80702) 829
Medium: 75% (v/v) dioxane/water; 0.1 M KNO₃.

C12H20N2O8 H4L CAS 40623-42-5 (1101)
1,2-Diaminoethane-N,N'-di(2-pentane-1,5-dioic acid); (CH₂NHCH(COOH)CH₂CH₂COOH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KNO₃ 25°C 0.10M U K1=12.49 1972TSd (82108) 830
K(VO(OH)L+H)=8.31
K(VO(OH)₂L+2H)=16.90
K((VO(OH)L)₂+2H)=14.71

C12H22O12 HL Lactobionic acid CAS 96-82-2 (2487)
4-O-Beta-D-Galactopyranosyl-D-gluconic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KNO₃ 25°C 0.10M C B2=6.07 1991KBa (82934) 831
B((VO)H-1L2)=2.32
B((VO)H-2L2)=-1.92
B((VO)H-3L2)=-10.31
B((VO)H-4L2)=-18.56

C13H11NOS H2L (7306)
2-(Salicylideneamino)thiophenol, Salicylaldehyde-2-mercaptoanil; HO.C6H4.CH:N.C6H4.SH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 V0++ gl diox/w 25°C 75% U K1=19.4 1998FHa (85049) 832

 Medium: 75% (v/v) dioxane/water; 0.1 M KNO₃.

 C13H11N02 HL CAS 1761-56-4 (3408)

 2-(Salicylideneamino)phenol, Salicylaldehyde-2-hydroxyanil; HO.C6H4.CH:N.C6H4.OH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 V0++ gl diox/w 25°C 75% U K1=17.7 1998FHa (85070) 833

 Medium: 75% (v/v) dioxane/water; 0.1 M KNO₃.

 C13H11N203F3 HL (5563)

 3-(2-Acetylphenylhydrazone)-1,1,1-trifluoropentane-2,4-dione;
 CF₃.CO.C(CO.CH₃):N.HN.C6H4.COCH₃

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 V0++ gl diox/w 25°C 75% U K1=10.68 B2=18.68 1990ASb (85256) 834

 C13H12N40 L Diphenylcarbaz. CAS 538-62-5 (1195)

 Diphenylcarbazone; C6H₅.NH.NH.CO.N:N.C6H₅

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 V0++ gl diox/w 25°C 50% U K1=8.5 B2=16.10 1986MHb (85422) 835

 C13H14N203 HL (4940)

 3-(2-Acetylphenylhydrazone)pentane-2,4-dione; (CH₃.CO)2C:N.NH.C6H4(CO.CH₃)

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 V0++ gl diox/w 25°C 75% U K1=12.75 B2=24.22 1990ASb (85619) 836

 C13H17N305 H2L CAS 511537-86-3 (8568)

 N-[(2-Hydroxyphenyl)methyl]glycylglycylglycine;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 V0++ gl KCl 25°C 0.20M C K1=11.18 B2=18.29 2002PCa (86019) 837

 B((VO)H2L)=19.1

 B((VO)H2L) determined by spectrophotometry.

 C14H807S H3L DASA CAS 83-61-4 (950)

 1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 V0++ sp oth/un 25°C ? U 1966SMa (86770) 838

 K(VO₃+H2L=V02L)=8.4(?)

C14H12N20S H2L (7309)
Salicylaldehyde thiobenzoylhydrazone; HO.C6H4.CH:N.N:C(SH).C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ gl diox/w 25°C 75% U K1=20.5 1998FHa (87164) 839
Medium: 75% (v/v) dioxane/water; 0.1 M KNO3.

C14H12N202 H2L (7307)
Salicylaldehyde benzoylhydrazone; HO.C6H4.CH:N.N:C(OH).C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ gl diox/w 25°C 75% U K1=23.8 1998FHa (87191) 840
Medium: 75% (v/v) dioxane/water; 0.1 M KNO3.

C14H12N203 H3L (7308)
Salicylaldehyde salicylhydrazone; HO.C6H4.CH:N.N:C(OH).C6H4OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ gl diox/w 25°C 75% U K1=17.6 1998FHa (87233) 841
Medium: 75% (v/v) dioxane/water; 0.1 M KNO3.

C14H12O3 HL Benzilic acid CAS 76-93-7 (710)
Diphenylglycolic acid, (benzilic acid); (C6H5)2C(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ gl KCl 25°C 0.20M C K1=2.29 1993MSa (87353) 842
B((VO)H-1L)=-0.92
B((VO)H-1L2)=1.63
B((VO)H-2L2)=-1.90

C14H22N208 H4L CDTA CAS 482-54-2 (200)
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ gl KNO3 25°C 0.10M U K1=20.1 1983FSa (88816) 843

VO++ vlt KNO3 20°C 0.10M U K1=19.40 1954SGa (88817) 844

C14H22N4010 H3L CAS 29725-87-9 (5074)
Ethylenedinitrilo-N,N'-bis(methylenecarbonyliminoethanoic)-N,N'-diethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

VO++ gl KNO3 25°C 0.10M U K1=12.46 1970MMc (88935) 845

$$K(VOL+H)=3.64$$

$$K(VOHL+H)=2.70$$

C14H22N4O10 H4L DGENTA CAS 29725-86-8 (2371)
N,N-Diglycyldiaminoethane-tetraethanoic acid; (-CH₂.HNCOCH₂N(CH₂COOH)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KN03 25°C 0.10M U K1=10.3 1970MMc (88953) 846
K(VOL+H)=5.39

C14H23N3O10 H5L DTPA CAS 67-43-6 (238)
Diethylenetriamine-pentaethanoic acid; HOOC.CH₂.N(CH₂.CH₂.N(CH₂.COOH)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl NaClO₄ 25°C 0.50M U K1=16.31 B2=23.3 1975NAb (89432) 847
K(VO+HL)=13.84
K(VOL+H)=7.00

C14H24N2O10 EGTA CAS 67-42-5 (349)
Ethyleneglycol-O,O'-bis(2-aminoethyl ether)-N,N,N',N'-tetraethanoic acid; H4L

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl KN03 25°C 0.10M U 1983FSa (89961) 848
K(VO+HL)=10.48

V0++ gl NaClO₄ 25°C 0.50M U K1=14.02 1975NAb (89962) 849
K(VO+HL)=10.33
K(VOL+H)=5.20

C14H32N2O4 L CAS 102-60-3 (2678)
Tetra(2-hydroxypropyl)-N,N,N',N'-diaminoethane; (-CH₂.N(CH₂.CH(OH).CH₃)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl NaNO₃ 25°C 0.10M C K1=6.57 1994DHa (90751) 850

C15H11N02 H2L (430)
2-(2'-Hydroxyphenyl)-8-hydroxyquinoline; HO.C6H₄.C9H₅N.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

V0++ gl diox/w 25°C 75% U K1=26.0 1998FHa (91057) 851
Medium: 75% (v/v) dioxane/water; 0.1 M KN03.

C15H11N2O2F3S H2L (7312)
1-(2-Thenoyl)-3,3,3-trifluoracetone benzoylhydrazone;
C4H3S.C(OH):CH.C(CF₃):N.N:C(OH).C6H5

$$\begin{aligned}B(VOHL) &= 13.86 \\B(VOH-4L2) &= -22.93 \\B((VO)2H-2L2) &= 10.52\end{aligned}$$

$$B((VO)2H-3L2) = 3.26, \quad B((VO)2H-4L2) = -5.89.$$

C15H24N2O17P2 H3L CAS 133-89-1 (7986)

Uridine-5'-diphosphoglucose;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KNO3	25°C	0.10M	C		B2=15.91 B(VOHL)=14.66 B(VOH2L2)=26.32 B(VOH-1L2)=7.98 B(VOH-2L2)=-0.46	2001GLa (92334)	859	

$$B(VOH-3L2) = -10.96, \quad B(VOH-4L2) = -22.17, \quad B((VO)2L2) = 22.17,$$

$$B((VO)2H-2L2) = 12.95, \quad B((VO)2H-3L2) = 4.88.$$

C16H9N3O6Cl2S H3L (6683)
7-((3,5-Dichloro-2-carboxyphenyl)azo)-8-hydroxyquinoline-5-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	KNO3	25°C	0.10M	U		K1=15.94	1993HKc (92674)	860	

C16H11N2O8Cl2 H4L (7166)
2-((3-Chlorophenyl)azo)1,8-dihydroxynaphthalene-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	NaClO4	25°C	0.10M	C			1994HKb (92775)	861	
								K(VO+H2L=VOL+2H)=33.72		

C16H11N2O9Cl2 H4L Plasmocorinth CAS 1058-92-0 (5203)
3-(5-Chloro-2-hydroxyphenylazo)chromotropic acid (Eriochrome Blue SE)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	NaClO4	25°C	0.10M	C			1994HKb (92787)	862	
								K(VO+H3L=VOHL+2H)=30.28		
								K(VOHL+H3L=VOH2L2+2H)=29.35		

C16H12N2O2 H2L CAS 9486-98-2 (3462)
1-(2-Hydroxyphenylazo)-2-hydroxynaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	diox/w	25°C	75%	U		K1=22.6	1998FHa (92960)	863	

Medium: 75% (v/v) dioxane/water; 0.1 M KNO3.

C16H12N2O8S2 H4L (6676)
1-((2-Hydroxy-5-sulfophenyl)azo)-2-hydroxynaphthalene-6-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	NaClO4	25°C	0.10M	U			K1=34.28	1993HKb (93040)	864

C16H12N2O8S2 H4L (6675)
2-((2-Hydroxy-5-sulfophenyl)azo)-1-hydroxynaphthalene-4-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	NaClO4	25°C	0.10M	U			K1=32.27	1993HKb (93045)	865

C16H18N2O3 HL (5564)
2-(2-Acetylphenylhydrazone)-5,5-dimethyl-1,3-cyclohexanedione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	diox/w	25°C	75%	U			K1=12.80 B2=23.20	1990ASb (93789)	866

C16H20N2O6 H2L CAS 488827-72-1 (8831)
N,N'-Bis(3-hydroxy-6-methyl-2-methylene-4-pyrone)ethylenediamine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaCl	25°C	0.16M	C			K1=15.57 K(VO+HL)=11.77 K(VO+H2L)=7.6 K(VO+H3L)=4.2 *K(VOL)=-8.95	2002SSb (94009)	867

C17H10N2O6Cl2S H3L (6684)
2-((3,5-Dichloro-2-carboxyphenyl)azo)-1-hydroxynaphthalene-4-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	KNO3	25°C	0.10M	U			K1=14.68	1993HKc (95684)	868

For 1-....-2-hydroxynaphthalene K1=19.17

C17H14N2O2 L CAS 4551-69-3 (698)
4-Benzoyl-3-methyl-1-phenyl-2-pyrazolin-5-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	dis	NaClO4	25°C	0.10M	U			B2=14.11	1972KEc (95906)	869

C17H16N2O2 H2L (7310)
Benzoylacetone benzoylhydrazone; C6H5.C(OH):CH.C(CH3):N.N:C(OH).C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO4	25°C	0.50M	C				1976NAa (98104)	876

C19H14O7S		H4L		Pyrocatechol	Vi	CAS	369596-29-2	(709)		
Pyrocatechol Violet, 3-[3,4-Dihydroxyphenyl-3-hydroxy-4-oxo-2,5-cyclohexadien-1-ylidenemethyl-b.;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	oth/un	25°C	0.0	U				1967MPc (99117)	877

C19H19N07		H2L						(7003)		
3-Methoxy-5-(N,N-dicarboxymethyl)aminomethyl-4-hydroxybenzophenone;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	20°C	0.10M	U		K1=17.0		1981SYa (99258)	878

C20H15N03		H2L						(2120)		
2-(alpha-Phenyl-2-hydroxybenzylideneimino)benzoic acid; HO.C6H4.C(C6H5):N.C6H4.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	NaClO4	25°C	0.10M	U	TIH	K1=11.19	B2=20.68	1986SGb (99750)	879
35 C: K1=11.55, K2=9.82; 45 C:K1=11.80, K2=10.17										
DH(K1)=-58.07 kJ mol-1, DS=248 J K-1 mol-1										

C20H32N6012S2		H4L		GSSG		CAS	27025-41-8	(1241)		
Glutathione oxidized; (HOOC.CH(NH2)C2H4.CO.NH.CH(CO.NH.CH2.COONH)CH2.S)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	25°C	0.20M	C		K1=11.39		2001PTa (100488)	880

B((VO)H-1L)=3.27, B((VO)H-2L)=-6.65.										

C21H22N40		HL				CAS	56932-30-0	(5308)		
1-Hydroxy-2-(2-N-methylanabasinyl-alpha-azo)naphthalene;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

V0++ sp oth/un ? ? U B2=10.49 1966APa (101204) 881

C22H17N3O3 HL CAS 53855-37-1 (4154)
8-Hydroxy-7-(3'-nitroanilinobenzyl)-quinoline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	diox/w	25°C	70%	U			K1=10.08 B2=19.47	1978SPf (101572)	882
Medium: 70% v/v dioxane/H ₂ O, 0.10 M KNO ₃ .										

C23H16O9Cl2S	H4L	Chrome azurol S	CAS	1667-99-8	(711)					
Chromazurol S;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	oth/un	25°C	?	U				1967SSb (102578)	883
K _{1eff} =4.6 (pH 4.0)										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	diox/w	25°C	75%	U			K1=11.42 B2=20.60	1990ASb (102604)	884

C25H28N2O13	H6L	CAS	42281-29-8	(5335)						
(Carbonylbis((6-hydroxy-5-methoxy-3-phenylene)methylene)trilo))tetraethanoic acid;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	gl	KCl	20°C	0.10M	U			K1=20.8 K(VO+HL)=17.9 K(VO+H2L)=14.9 K(VO+H3L)=9.9	1973VIB (103666)	885

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
V0++	sp	NaClO ₄	25°C	2.00M	U				1991BBC (103824)	886
K(VO+H4L=VOH3L+H)=1.27 K(VOH3L=VOH2L+H)=-1.70 K(VHL+H2O=VOH2L+H)=4.22										

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EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES
I Data with various BACKGROUNDS
H Data for THERMOCHEMICAL quantities
M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC

END