

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 119 experiments for

(no ligands specified)

5 metals : W(0), W(III), W(IV), W(V), W(VI)

(no references specified)

(no experimental details specified)

CO L Carbon monoxide CAS 630-08-0 (551)

Carbon monoxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(0)	gl	oth/un	0°C	var	U	M			1959HEb	(2824) 1
								K(H ₃ W ₃ L ₉ (OH) ₂ (H ₂ O)+H)=1.5		
								K(H ₂ W ₃ L ₉ (OH) ₂ (H ₂ O)+H)=5.4		

C3H9O3P L CAS 121-45-9 (1786)

Trimethylphosphite; (CH₃O)₃P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(0)	cal	non-aq	25°C	100%	U	HM			1991ZGa	(28004) 2
Medium:	THF.	DH(Mo(CO) ₃ A ₂ +L)=-110.9	kJ mol-1					A=P(C ₆ H ₁₁) ₃		

C18H33P L CAS 2622-14-2 (169)

Tri-(cyclohexyl)phosphine; (C₆H₁₁)₃P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(0)	cal	non-aq	25°C	100%	U	HM			1991ZGa	(98316) 3
								K(W(CO) ₃ py ₂ +L)=-6.84		

Medium: THF. DH=-79.1 kJ mol-1

C3H9P L CAS 594-09-2 (1732)

Trimethyl phosphine; (CH₃)₃P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(III)	nmr	non-aq	30°C	100%	U	T	HM		1992RZa	(28058) 4
								K(WL ₆ =WL ₄ A(H)+L)=1.25		

Metal::W(0). Method:NMR. Medium:C6D6. T=30-70C. K=1.48(40C); 1.78(50C); 1.92(60C); 2.00(70C). DH=38.9 kJ mol-1; DS=155. A:P(CH₃)₂(CH₂).

C8H19P L (6822)

Di(t-Butyl)phosphine; ((CH₃)₃C)₂PH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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W(III) nmr non-aq ? 100% U T HM 1992BCc (63202) 5
 Metal:W++. Method:NMR. Medium:toluene. DH(1,2-W2L2(NMe₂)₄(anti-gauche
 isomerization)=-2.1 kJ mol⁻¹, DS=-1.3. Data also for other phosphides

CN- HL Cyanide CAS 74-90-8 (230)
 Cyanide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(IV)	kin	KNO ₃	25°C	1.00M	U				1995SPb (2775)	6
								K(WO(H ₂ O)L ₄ +L)=3.0		
								K(WO(H ₂ O)L ₄ +HL)=WO(HL)L ₄)=0.0		
								*K(WO(HL)L ₄)=-5.8		

W(IV)	EMF	KCl	20°C	0.10M	U	I			1973HKa (2776)	7
								K(WO ₂ (CN) ₄ +H)=11.7		
								K'(WO ₂ (CN) ₄ H+H)=8.25		
K=12.0(I=0.014); 12.1(I=0.04); 11.8(I=0.06); 11.6(I=0.2); 11.6(I=0.5); -12.1(I=0). I=0: K=12.1. K'=8.57(I=0.014); -7.76(I=0.5). I=0: K'=8.84										

W(IV)	gl	oth/un	20°C	0.10M	U				1971SKc (2777)	8
								K(H+W(CN) ₈)=1.6		

F- HL Fluoride CAS 7644-39-3 (201)
 Fluoride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(IV)	sp	KNO ₃	20°C	1.00M	U	M			1986LBa (7331)	9
								K(WO(H ₂ O)(CN) ₄ +F)=2.15		

OH- HL Hydroxide (57)
 Hydroxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(IV)	gl	KCl	25°C	1.2M	C				1998ARa (12484)	10
								*K(WO(H ₂ O)(CN) ₄)=-7.89		
								*K(WO(OH)(CN) ₄)=-14.5		

Medium: KCl/KNO₃

W(IV)	EMF	oth/un	16°C	var	U				1959LMa (12485)	11
								K(W(CN) ₄ (OH)+OH)=9.10		
								K(W(CN) ₄ (OH) ₂ +OH)=6.67		
								K(W(CN) ₄ (OH) ₃ +OH)=3.28		

Metal is W(IV). Method: Bi electrode

W(IV)	gl	oth/un	25°C	var	U				1956BAa (12486)	12
								K1(W(CN) ₈ +H) < 2		
								K1(W(CN) ₈ +H) < 2		

C2H4	L	Ethylene	CAS	74-85-1	(478)	Reference	ExptNo		
Ethene; H2C:CH2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference ExptNo
W(IV)	nmr	non-aq	24°C	100%	U	M			1992HMa (19431) 13
								$K(WO_3Cl_2+L=W_2Cl_2+A)=2.30$	
Method:NMR. Medium:C6D6. A=PMePh2. When A=PMe3, K=-3.02									

e-	HL	Electron					(442)		
Electron;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference ExptNo
W(V)	EMF	none	25°C	0.0	U				1955BTa (1029) 14
								$K(W(CN)_8+e)=7.73(457 \text{ mV})$	
W(V)	vlt	oth/un	25°C	12.0M	U				1952LAb (1030) 15
								$K(W+e=W(IV))=-5(-300 \text{ mV})$	
$K(W+2e=W(III))=-7(\text{red } WC15, -200 \text{ mV}), -3.4(W(III), -100 \text{ mV}), 3.4(\text{green } W2C19)$									
W(V)	EMF	KCl	0°C	1.0M	U	I			1924COa (1031) 16
								$K(W(CN)_8+e)=11.0(597 \text{ mV})$	
At I=0.5 M K=K=10.7(580 mV), I=0.25 M: K=10.5(568 mV)									

CN-	HL	Cyanide					(230)		
Cyanide;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference ExptNo
W(V)	nmr	KNO ₃	25°C	0.10M	C				1994RLa (2778) 17
								$*K(WO(CN)_4(H_2O))=-7.85$	
W=W(IV). Method: N.M.R.									
W(V)	gl	oth/un	20°C	0.10M	U				1971SKc (2779) 18
								$K(H+W(CN)_8)=2.35$	
								$K(H+HW(CN)_8)=1.7$	

Cl-	HL	Chloride					(50)		
Chloride;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference ExptNo
W(V)	oth	oth/un	20°C	var	U	T	H		1972JRa (5948) 19
								$K=6.48$	
K: 2WOCl ₅ +6H ₂ O=W ₂ Cl ₂ (OH) ₂ (H ₂ O) ₂ +6H+8Cl. K=6.61(1 C), 6.30(40 C).									
DH(K)=-14.6 kJ mol ⁻¹ , DS=74 J K ⁻¹ mol ⁻¹ . Method: magnetic susceptibility									
W(V)	oth	KCl	40°C	var	U	T			1967JRa (5949) 20

$$K(2WOCl_5 + 6H_2O = X + 8Cl + 6H) = 12.58$$

Method: magnetic susceptibility. Medium:HCl var. K=13.23(1 C),12.94(20 C)
 $X=W_2O_2(OH)_6Cl_2$

OH- HL Hydroxide (57)
Hydroxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(V)	kin	oth/un	25°C	2.00M	U				1993PSa (12487)	21

* $K(Mo_2W_4(H_2O)) = -0.013$
* $K(Mo_2W_4(NCS)) = -1.7$

Metals are W(IV) and Mo(IV). Medium: 2.0 M Li(tetrafluoromethane sulfonate).

SCN- HL Thiocyanate CAS 463-56-9 (106)
Thiocyanate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(V)	kin	NaClO ₄	25°C	2.00M	U				1993VSa (15338)	22

$K(W_3S_4(H_2O)9+L) = 3.18$
 $K(MoW_2S_4(H_2O)9+L) = 3.29$
 $K(Mo_2WS_4(H_2O)9+L) = 3.02$

Medium: 2.0 M HClO₄. Metals are W(IV) and Mo(IV). For mixed Mo/W species the data refer to binding of L to W.

C5H9N L t-Butylnitrile CAS 7188-38-7 (913)
t-Butylcyanide; (CH₃)₃C.CN

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(V)	con	non-aq	40°C	100%	U	M			1992LJa (38458)	23

$K(WL_6I+I) = 3.11$

Medium: MeCN, 0.0063 M Bu₄NClO₄, W++. Contradictory data in Tables and text

e- HL Electron (442)
Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	vlt	oth/un	25°C	0.0	U	I			1952LAb (1032)	24

$K(W_2Cl_3+e) = 4.4(260 \text{ mV})$
 $K' = -9(-90 \text{ mV})$
 $K'' = -107(1050 \text{ mV})$

$K': W_3O_3(s) + 6H + 6e = W(s) + 3H_2O$. $K'': WO_4 + 4H_2O + 6e = W(s) + 8H$. $K(W_2Cl_3+e)$ in 12M HCl

CN- HL Cyanide CAS 74-90-8 (230)
Cyanide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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W(VI)	sp	KCl	20°C	0.20M	U		1978SSc (2780) 25
						K(WL7+H)=4.82	
						K(WL7OH+H)=9.17	
*****	*****	*****	*****	*****	*****	*****	*****
F-		HL	Fluoride		CAS 7644-39-3 (201)		
Fluoride;							
-----	Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values	Reference ExptNo
-----	W(VI)	nmr	non-aq	-75°C	100%	U	K1=1.7 B2=2.40 1974SBc (7332) 26
Medium: vinyl chloride							
-----	W(VI)	kin	KCl	25°C	0.20M	U	1964YPa (7333) 27
						K1eff=3.0	
-----	W(VI)	con	non-aq	-5°C	100%	U	1960NVa (7334) 28
Medium: liquid HF, m units						K(WO2F2+4HF=WF6+2H2O)=-2.8	
*****	*****	*****	*****	*****	*****	*****	*****
MoO4--		H2L	Molybdate			(443)	
Molybdate;							
-----	Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values	Reference ExptNo
-----	W(VI)	nmr	oth/un	25°C	>6	U	1994AHA (8761) 29
						Beff(1,6)=1.00	
						Beff(2,5)=1.57	
						Beff(3,4)=1.83	
						Beff(4,3)=1.83	
Beff(5,2)=1.55, Beff(6,1)=0.98. Beff(q,r): pH+qMo7024+rW7024 at pH 6.0							
*****	*****	*****	*****	*****	*****	*****	*****
NH2SO3-		H2L	Sulfamate		CAS 5329-14-6 (452)		
Sulfamate;							
-----	Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values	Reference ExptNo
-----	W(VI)	sp	NaCl	25°C	1.0M	U	1958SAC (8804) 30
						K(2H+2L+W04=W03L2+H2O)=-8 ?	
*****	*****	*****	*****	*****	*****	*****	*****
OH-		HL	Hydroxide			(57)	
Hydroxide;							
-----	Metal	Mtd	Medium	Temp	Conc	Cal Flags Lg K values	Reference ExptNo
-----	W(VI)	sp	NaClO4	25°C	2.00M	U	1992RSb (12488) 31
Medium: 2.0 M LiClO4. Metal is W(IV).						*K(W3S4(H2O)9)=-0.59	
-----	W(VI)	sol	oth/un	300°C	var	M T H	1992WOa (12489) 32

K_s(W₀₃(s)+H₂O=H₂WO₄)=-4.0
300-600 C and P=1 kbar. DH(K)=41 kJ mol⁻¹. Constant at I=0

W(VI) kin NaCl 25°C 0.10M U I 1978KKc (12490) 33
K(PW12040+OH)=0.85

In NaCl: K=0.20; in LiCl: K=0.78

W(VI) sp alc/w 20°C 10% U I 1977NPb (12491) 34
K(WO₂+OH)=13.17
K(WO₂+2OH)=25.72
K(WO₂+3OH)=37.69

W(VI) gl NaClO₄ 25°C 3.00M C 1974ASa (12492) 35
B(2,1)=11.30
B(6,6)=52.46
B(7,6)=60.76
B(14,12)=123.24

B(p,q): pH+qWO₄=Hp(WO₄)q. A recalculation of data published by Y.Sasaki

W(VI) sp NaNO₃ 25°C 0.10M U I K1=13.18 B2=25.9 1969NPd (12493) 36
B3=38.2

W(VI)=WO₂++. K1=13.28, B2=26.2, B3=38.7(I=2.5)

O2-- H2L Peroxide CAS 7772-84-1 (2813)
Peroxide; -0.0-

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

W(VI) sp oth/un 24°C var U 1963DLa (12758) 37
K(WL4+H)=4.9 to 6.9
K(WL4+H2L=HWL4+HL)=-6.9 to -8

SCN- HL Thiocyanate CAS 463-56-9 (106)
Thiocyanate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

W(VI) sp non-aq ? 100% U I K1=3.5 B2=7.2 1967USA (15339) 38
Medium: Me₂CO. W added as WC15. Conductivity also used. In MeCOEt: K1=3.1,
B2=6.3, B4=13.25. In cyclohexanone: B6=20.6

WO₄-- H2L Tungstate CAS 13783-36-3 (445)
Tungstate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

W(VI) gl NaCl 25°C 1.00M U 1979IRa (17453) 39
K(7H+6WO₄=HW6O21+3H₂O)=56.42
K(9H+6WO₄=H3W6O21+3H₂O)=70.45

CH2O2 HL Formic acid CAS 64-18-6 (37)
Methanoic acid; H.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	ix	oth/un	?	0.05M	U			K1=0.30 B3=3.18 B4=4.90	1970SHa (17661)	40

Metal ion: WO₂⁺⁺. Medium: 0.01-0.05 HL, pH 2.5

CH40 L Methyl alcohol CAS 67-56-1 (597)
Methanol; CH₃.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	EMF	alc/w	20°C	100%	U				1964GUa (17911)	41

$$K(WO(L')_3+L'=WO(L')_4)=12.51$$
$$K(WO(L')_5+H=WO(L')_4+L)=4.09$$

Method: H electrode. Medium: MeOH, 1.0 M Me4NCl. L'=H-1L (i.e. CH₃O)

C2H2O4 H₂L Oxalic acid CAS 144-62-7 (24)
Ethanedioic acid; (COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KNO ₃	21°C	0.22M	C				1978MBc (19152)	42

$$K(WO_4+2H+L=WO_3L+H_2O)=13.97$$

Medium pH 5-7.

W(VI)	oth	oth/un	?	?	U		K1=1.48		1969SHd (19153)	43
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Metal ion: WO₂⁺⁺

W(VI)	vlt	oth/un	25°C	0.16M	U				1962YBa (19154)	44
							K(H ₂ WO ₄ +H ₂ L)=4.85			

$$K(H_2WO_3L+H_2L)=7.5$$

Medium: 0.08-0.24 H₂SO₄

W(VI)	kin	oth/un	25°C	?	U				1962YPb (19155)	45
							K(H ₂ WO ₄ +H ₂ L)=5.13			

C2H4O3 HL Glycolic acid CAS 79-14-1 (33)
2-Hydroxyethanoic acid; HO.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	0.10M	C				1995HCa (20656)	46

$$Keff(WO_4+2L+2H=WO_2L_2)=16.85$$

Medium: 0.1 M acetate buffer, pH 4.7

C3H4O4 H₂L Malonic acid CAS 141-82-2 (79)

Propanedioic acid; CH₂(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	kin	oth/un	25°C	0.05M	U				1962YPa (24594)	47
								$K(H_2W_04+H_2L)=3.09$		

C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)
L-2-Hydroxypropanoic acid; CH₃.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	0.10M	C				1995HCA (25570)	48
								$K_{eff}(W_04+2L+2H=W_02L_2)=18.15$		

Medium: 0.1 M acetate buffer, pH 4.7

W(VI)	gl	NaCl	25°C	1.00M	C	H			1993CKb (25571)	49
								$B(1,2,2)=17.47$		
								$B(1,2,3)=18.38$		
								$B(1,1,2)=13.03$		
								$B(1,1,3)=14.56$		

$B(p,q,r): pW_04 + qHL + rH = (W_04)pLqHq+r.$ $B(2,2,3)=25.47.$ DH by calorimetry:
 $DH(1,2,2)=-80$ kJ mol-1.

C3H7NO2S H2L Cysteine CAS 52-90-4 (96)
2-Amino-3-mercaptopropanoic acid; H2N.CH(CH₂.SH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	NaCl	18°C	1.00M	U				1990CJa (26853)	50
								$K(W_04+L+2H=W_03L+H_2O)=18.8$		

C3H9P L CAS 594-09-2 (1732)
Trimethyl phosphine; (CH₃)₃P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	nmr	non-aq	25°C	100%	U	T	HM		1992WWa (28059)	51
								$K(WABC_2+L)=2.73$		

Method:NMR. Medium:Toluene. T=-10-40. K=4.23(-10C);3.73(0);3.28(10);2.96(18);2.10(40). A:CHC(CH₃)₃. B:NC₆H₄(i-C₃H₇)₂ C:OC(CH₃)₃. DH=-65.7kJmol-1;DS=-170

C4H6O4 H2L Succinic acid CAS 110-15-6 (112)
1,4-Butanedioic acid; HOOC.CH₂.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	ix	oth/un	22°C	0.10M	U			$K_1=1.06$	1973SDa (30076)	52
Metal ion:	W ₀₂ ⁺⁺ ,	pH 2.5								

C4H6O5 H₂L Malic acid CAS 617-48-1 (393)
 2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH₂.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	NaCl	25°C	1.0M	C	H			1997CKa (30756)	53
								B(1,1,1)=8.85		
								B(1,1,2)=14.78		
								B(1,1,3)=17.26		
								B(1,2,2)=17.20		
								B(p,q,r): pW04+qHL+rH=(W04)p(HL)qHr. B(1,2,3)=21.70, B(1,2,4)=25.61, B(2,1,3)=23.07, B(2,2,2)=18.87, B(2,2,3)=25.72, B(2,2,4)=31.88. DH by calorimetry.		
W(VI)	oth	NaClO ₄	30°C	1.00M	U	M			1979CBa (30757)	54
								B((W04)H ₂ L ₂)=23.1		

Method: polarimetry

C4H6O6 H₂L DL-Tartaric acid CAS 133-37-9 (94)
 DL-Tartaric acid, DL-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	0.10M	C				1995HCa (31037)	55
								Keff(2W04+2L+4H=(W0)2L ₂)=33.55		

Medium: 0.1 M acetate buffer, pH 4.7

C4H6O6 H₂L L-Tartaric acid CAS 87-69-4 (92)
 L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	kin	oth/un	25°C	0.10M	U				1962YPb (31396)	56
								K(H ₂ W04+H ₂ L)=3.93		

C4H7N04 H₂L Aspartic acid CAS 56-84-8 (21)
 Aminobutanedioic acid; H₂N.CH(CH₂.COOH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	NaClO ₄	25°C	0.10M	U			K1=8.20 B2=14.07 K3=3.81	1972SSe (31978)	57

Metal ion is WO₂⁺⁺

C4H7N04 H₂L IDA CAS 142-73-4 (118)
 Iminodiethanoic acid; HN(CH₂.COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	NaClO ₄	25°C	3.0M	U				1979ZLa (32398)	58
								B(W04+L+2H=W03L)=18.14		

W(VI)	gl	oth/un	25°C	0.15M	U	1966KRa (32399) 59 K(WO4+L+2H=WO3L)=18.5		
*****						*****		
C4H8N203	HL	Asparagine	CAS	70-47-3	(17)	2-Aminobutanedioic acid 4-amide; H2N.CH(CH2.CO.NH2).COOH		
*****						*****		
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference	ExptNo
W(VI)	gl	NaClO4	25°C	0.10M	U	K1=5.84 B2=10.95 K3=3.30	1973TSe (32746)	60
*****						*****		
C4H8O3	HL		CAS	594-61-6	(81)	2-Hydroxy-2-methylpropanoic acid; (CH3)2C(OH).COOH		
*****						*****		
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	0.10M	C		1995HCa (33537) 61 Keff(WO4+2L+2H=WO2L2)=18.75	
Medium: 0.1 M acetate buffer, pH 4.7						*****		
C4H8O3	HL		CAS	965-70-8	(423)	2-Hydroxybutanoic acid; CH3.CH2.CH(OH).COOH		
*****						*****		
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	0.10M	C		1995HCa (33588) 62 Keff(WO4+2L+2H=WO2L2)=18.25	
Medium: 0.1 M acetate buffer, pH 4.7						*****		
C5H9N04	H2L	Glutamic acid	CAS	56-86-0	(22)	2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH		
*****						*****		
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference	ExptNo
W(VI)	gl	NaClO4	25°C	0.10M	U	K1=7.95 B2=13.40 K3=3.60	1972SSe (39147)	63
*****						*****		
C5H9N04	H2L	MIDA	CAS	4408-64-4	(190)	N-Methyliminodiethanoic acid; CH3.N(CH2.COOH)2		
*****						*****		
Metal	Mtd	Medium	Temp	Conc	Cal Flags	Lg K values	Reference	ExptNo
W(VI)	gl	oth/un	25°C	0.15M	U		1966KRa (39293) 64 K(WO4+L+2H=WO3L)=18.70	
*****						*****		
W(VI)	nmr	oth/un	35°C	1.00M	U		1966KRa (39294) 65 K(WO4+L+2H=WO3L)=18.6	
*****						*****		
C5H10N203	HL	Glutamine	CAS	56-85-9	(18)			

2-Aminopentanedioic acid 5-amide; H₂N.CH(CH₂.CH₂.CO.NH₂)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	NaClO ₄	25°C	0.10M	U			K1=5.76 K3=3.20	B2=10.85 1973TSe (39847)	66

C5H10O5 L CAS 1114-34-7 (6113)
D-Lyxose

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
W(VI)	g1	KCl	25°C	0.10M	C					1989VCA (40340)	67
									$B(2WO_4 + 1 + 2H) = 18.08$		

C5H12O5 L Xylitol CAS 87-99-0 (2139)
Xylitol: HO.CH₂.HCOH.HOCH.HCOH.CH₂.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KCl	RT	0.10M	M	I			1990VSc (41694)	68
								$K(2WO_4 + 2H^+ + L \rightarrow W_2O_7L + H_2O) = 18.50$		

Data for 0.01-1.0 M KCl and NaCl. In 0.01 M KCl, $K=19.65$.

C6H204C12 H2I Chloranilic acid CAS 87-88-7 (1281)

3,6-Dichloro-2,5-dihydroxy-1,4-benzoquinone:

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

W(VI) sp. oth/up 30°C ? II K1=5 23 1981BMd (42061) 69

C6H4N2O6 H2L CAS 106-13-0

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W(VI) sp NaNO₃ 25°C 0.10M U 1972PSb (42268) 70
K(Hg2+2L) 30-74

C6H5NO4 H2L 4-Nitrocatechol CAS 3316-09-4 (890)

1,2-Dihydroxy-4-nitrobenzene; O2N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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W(VI) sp oth/un 25°C .575M U 1980NKA (42946) 71
 $K(H_2WO_4L + H_2L \rightleftharpoons WO_2L_2 + 2H_2O) = 3.15$

Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅. pH 8

C6H5O2Cl H2L 4-Cl-Catechol CAS 2138-22-9 (1656)
1,2-Dihydroxy-4-chlorobenzene; Cl.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	.575M	U				1980NKA (43086)	72

$$K(H_2WO_4L + H_2L = WO_2L_2 + 2H_2O) = 3.41$$
$$K(WO_4 + H_2L) = 3.25$$

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

W(VI)	sp	KCl	25°C	0.10M	U				1963HAc (43087)	73
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$$K(WO_4 + 2H_2L) = 7.1$$

C6H6O2 H2L Catechol CAS 120-80-9 (534)
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	.575M	U				1980NKA (43864)	74

$$K(H_2WO_4L + H_2L = WO_2L_2 + 2H_2O) = 3.98$$
$$K(WO_4 + H_2L) = 2.59$$

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

W(VI)	sp	oth/un	20°C	0.10M	U				1964PCa (43865)	75
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$$K(WO_4 + 2H_2L = WO_2L_2) = 6.53$$

Medium: 0.1 M NaHSO3

W(VI)	sp	oth/un	20°C	?	U				1959HAa (43866)	76
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$$K(WO_4 + 2H_2L = WO_2L_2) = 6.53$$

C6H6O3 H3L Pyrogallol CAS 87-66-1 (696)
1,2,3-Trihydroxybenzene; C6H3(OH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	.575M	U				1980NKA (43998)	77

$$K(H_2WO_4L + H_2L = WO_2L_2 + 2H_2O) = 3.21$$
$$K(WO_4 + H_2L) = 2.98$$

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

W(VI)	sp	oth/un	20°C	?	U				1959HAa (43999)	78
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$$K(WO_4 + 2H_3L = WO_2(HL)_2) = 6.98$$

W(VI)	sp	oth/un	20°C	?	U				1958PIa (44000)	79
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$$K(WO_4 + 2H_3L = WO_2(HL)_2) = 7.34$$

C6H8O7 H3L Isocitric acid CAS 1637-73-6 (2527)
2-Hydroxy-3-carboxypentanedioic acid; HOOC.CH(OH).CH(COOH).CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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W(VI) sp oth/un 25°C 0.10M C 1995HCA (45736) 80
Keff(2W04+2HL+2H=(W02)202L2)=25.7. Medium: 0.1 M acetate buffer, pH 4.7

C6H8O7 H3L Citric acid CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	NaCl	25°C	1.00M	U	H			1995CRa (46309)	81
								B(111)=10.21		
								B(121)=17.03		
								B(131)=21.67		
								B(141)=22.82		

B(pqr): pW04 + qH + rL = (W04)pHqLr

W(VI)	gl	NaCl	25°C	1.00M	C	H			1991CKa (46310)	82
								B(1,1,1)=10.21		
								B(1,2,1)=17.03		
								B(1,3,1)=21.67		
								B(1,4,1)=22.82		

B(2,4,2)=34.89, B(2,5,2)=39.33, B(1,6,2)=34.51, B(2,4,1)=31.68.

B(p,q,r): pW04+qH+rL=W04pHqLr. Also DH by calorimetry. Ligand defined as H4L

C6H9N06 H3L NTA CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH2.COOH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	NaClO4	25°C	0.10M	C	I			2004MZa (47096)	83

Keff(W04+2H+L)=19.00

Data for 0.3-1.0M NaClO4. At I=1.0 M, Keff=19.40.

W(VI)	sp	NaClO4	25°C	0.5M	C				1976CLa (47097)	84
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K(W04+2H+L=W03L+H2O)=17.75

Method: stopped flow spectrophotometry

W(VI)	gl	oth/un	25°C	0.15M	U				1966KRa (47098)	85
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K(W04+L+2H=W03L)=18.86

W(VI)	nmr	oth/un	35°C	2.00M	U				1966KRa (47099)	86
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K(W04+L+2H=W03L)=19.1

C6H12N2O4 H2L EDDA CAS 5657-17-0 (119)
1,2-Diaminoethane-N,N'-diethanoic acid; HOOC.CH2.NH.CH2.CH2.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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W(VI)	gl	NaClO4	25°C	3.0M	U				1979ZLa (49284)	87
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B(W04+L+2H=W03L)=19.62

C6H1205 L L-Rhamnose CAS 634-74-2 (3659)
6-Deoxy-L-mannose;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KCl	25°C	0.10M	C				1989VCa (49509)	88
								B((W04)2H2L)=17.04		
								B((W04)2H3L)=20.54		
								K((W04)2H2L+H)=3.50		

C6H1206 L D-Mannose CAS 3458-28-4 (1562)
D-Mannose

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KCl	25°C	0.10M	C				1989VCa (49608)	89
								B((W04)2H2L)=17.50		

M=W04

C6H1406 L D-Mannitol CAS 69-65-8 (3664)
D-Mannitol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KNO ₃	21°C	0.10M	C				1978MBc (51090)	90
Medium pH 3-5.								K(2W04+2H+L=W205(H-4L)+3H ₂ O)=18.78		

C6H1406 L Glucitol CAS 50-70-4 (2878)
D-Sorbitol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KCl	RT	0.10M	M	I			1990VSc (51110)	91
								K(2W04+2H+L=W207L+H ₂ O)=19.15		

Data for 0.01-1.0 M KCl and NaCl. In 0.01 M KCl, K=19.30.

W(VI)	gl	KNO ₃	21°C	0.10M	C				1978MBc (51111)	92
Medium pH 3-5.								K(2W04+2H+L=W205(H-4L)+3H ₂ O)=19.26		

C7H6O3 H2L CAS 139-85-5 (881)
3,4-Dihydroxybenzaldehyde, protocatechuic aldehyde; C₆H₃(OH)₂.CHO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	0.575M	U				1980NKA (54359)	93
								K(H ₂ W04L+H2L=W02L2+2H ₂ O)=3.39		
								K(W04+H2L)=3.10		

Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅. pH 8

W(VI) sp oth/un 20°C ? U 1959HAa (54360) 94
K(WO₄+H₂L=WO₂L₂)=7.76

C7H6O₄ H₃L Protocatechuic CAS 99-50-3 (875)
3,4-Dihydroxybenzoic acid; C₆H₃(OH)₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

W(VI) sp oth/un 25°C .575M U 1980NKA (54709) 95
K(H₂WO₄L+H₂L=WO₂L₂+2H₂O)=3.18
K(WO₄+H₂L)=3.23

Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅. pH 8

W(VI) sp oth/un 20°C ? U 1959HAa (54710) 96
K(WO₄+2H₃L=WO₂H₂L₂)=7.30

C7H6O₅ H₄L Gallic acid CAS 149-91-7 (446)
3,4,5-Trihydroxybenzoic acid; C₆H₂(OH)₃.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

W(VI) sp oth/un 25°C .575M U 1980NKA (54772) 97
K(H₂WO₄L+H₂L=WO₂L₂+2H₂O)=3.39
K(WO₄+H₂L)=3.34

Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅. pH 8

W(VI) sp oth/un 20°C ? U 1959HAa (54773) 98
K(WO₄+H₃L=WO₃HL)=3.37

C7H7NO₂ HL CAS 495-18-1 (184)
Benzohydroxamic acid; C₆H₅.CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

W(VI) dis KCl ? 4.0M U 1967PNa (55526) 99
K(WO₂+2HL=WO₂L₂+2H)=7.11

C7H8O₂ H₂L CAS 488-17-5 (1657)
1,2-Dihydroxy-3-methylbenzene; CH₃.C₆H₃(OH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

W(VI) sp oth/un 25°C .575M U 1980NKA (56057) 100
K(H₂WO₄L+H₂L=WO₂L₂+2H₂O)=3.60
K(WO₄+H₂L)=3.22

Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅. pH 8

C7H8O₂ H₂L Methylcatechol CAS 452-86-8 (525)
1,2-Dihydroxy-4-methylbenzene; CH₃.C₆H₃(OH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	.575M	U				1980NKA (56082)	101
								$K(H_2W_04L+H_2L=W_02L_2+2H_2O)=3.76$		
								$K(W_04+H_2L)=2.42$		
Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8										
W(VI)	sp	KCl	25°C	0.10M	U				1963HAc (56083)	102
								$K(W_04+2H_2L)=6.31$		

C8H8O3		HL	Mandelic Acid	CAS 611-72-3	(80)					
2-Phenyl-2-hydroxyethanoic acid; C6H5.CH(OH).COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	0.10M	C				1995HCA (59888)	103
								$K_{eff}(W_04+2L+2H=W_02L_2)=18.15$		
Medium: 0.1 M acetate buffer, pH 4.7										

C9H7N03S2		H2L		CAS 58447-10-2	(4675)					
8-Mercaptoquinoline-5-sulfonic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	?	?	U				1968ABA (64433)	104
								$K(W_2O_5+L)=10.6$		
								$K(W_2O_5+2L)=19.7$		

C9H7N04S		H2L	Sulfoxine	CAS 84-88-8	(448)					
8-Hydroxyquinoline-5-sulfonic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KNO3	16°C	0.10M	U				1969GTA (64589)	105
								$K(W_04+L+2H=W_03L)=19.87$		

C10H8O2		H2L		CAS 92-44-4	(1658)					
2,3-Dihydroxynaphthalene;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	.575M	U				1980NKA (69784)	106
								$K(H_2W_04L+H_2L=W_02L_2+2H_2O)=3.47$		
								$K(W_04+H_2L)=3.57$		
Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8										

C10H16N208		H4L	EDTA	CAS 60-00-4	(120)					
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

W(VI) gl NaClO4 25°C 3.0M U 1979ZLa (74306) 107
B(WO4+L+2H=WO3L)=19.67
B(2WO4+L+2H=W2O6L)=36.22

W(VI) gl NaClO4 25°C 0.10M U K1=9.92 1975PPb (74307) 108
K(WO3L+H)=7.31
B((WO3)2L)=18.41

K1: WO3+L=WO3L

W(VI) gl oth/un 25°C 0.15M U 1966KRa (74308) 109
K(WO4+L+2H)=18.9
K(WO4+WO3L+2H)=16.9

W(VI) nmr oth/un 35°C 1.0M U 1966KRa (74309) 110
K(WO4+L+2H)=18.7
K(WO4+WO3L+2H)=16.7
K(WO3L+H)=7.5

C10H25N5 L 15-Ane-N5 CAS 295-64-7 (99)
1,4,7,10,13-Pentaazacyclopentadecane; cyclo(-(HN.CH2.CH2)5-)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

W(VI) vlt NaClO4 25°C 0.20M C 1999SSe (76739) 111
K(WO4+H3L)=2.11

Method: differential pulse polarography.

C12H30N6 L CAS 296-35-5 (143)
1,4,7,10,13,16-Hexaaazacyclooctadecane; cyclo(-(-(NH.CH2.CH2)6-))

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

W(VI) vlt NaClO4 25°C 0.20M C 1999SSe (84361) 112
K(WO4+H3L)=2.21

Method: differential pulse polarography.

C14H8O7S H3L DASA CAS 83-61-4 (950)
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

W(VI) sp oth/un 25°C ? U B2=7.8 1959DBb (86771) 113

C15H11N3O4S H2L (5130)
7-Phenylazo-8-hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

W(VI) gl KN03 16°C 0.10M U 1969GTa (91339) 114

$$B((W_04)H_2L)=18.34$$

C15H11N3O7S2 H3L CAS 17852-90-3 (5131)
7-(4-Sulfophenylazo)-8-hydroxyquinoline-5-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KNO ₃	16°C	0.10M	U				1969GTA (91352)	115

$$B((W_04)H_2L)=18.05$$

C19H13N3O7S2 H3L SNAZOXS CAS 117-87-3 (995)
8-Hydroxy-7-(4'-sulfo-1'-naphthylazo)-quinoline-5-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KNO ₃	16°C	0.10M	U				1969GTA (99051)	116

$$K(W_04+L+2H)=18.00$$

C22H20013 H5L Carminic acid CAS 1260-17-9 (714)
Carminic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	22°C	?	U				1966KWb (101707)	117

$$K(W_04+H_5L=W_03H_3L)=5.5(?)$$

C22H24N2O8 L Deoxycycline CAS 564-25-0 (2204)
Deoxycycline, 6-Deoxy-5-hydroxytetracycline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	none	20°C	0.0	C				1991JMa (101767)	118

$$K(W_04+H_3L=W_03HL)=8.39$$

$$K(W_04+2H_3L=W_03(H_2L))=8.26$$

C22H24N2O8 H2L Tetracycline CAS 60-54-8 (2201)
Tetracycline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	none	20°C	dil	C				1989VJa (101831)	119

$$K(W_04+HL)=7.86$$

REFERENCES

- 2004Mza K Majlesi, K Zare, F Teimouri; J.Chem.Eng.Data, 49, 439 (2004)
- 1999SSe A Salimi, M Shamsipur; J.Inclusion Phenom., 34, 455 (1999)
- 1998ARa A Abou-Hamdan, A Roodt, A Merbach; Inorg.Chem., 37, 1278 (1998)
- 1997CKa J Cruywagen, L Kruger, E Rohwer; J.Chem.Soc., Dalton Trans., 1925 (1997)
- 1995CRa J Cruywagen, E Rohwer, G Wessels; Polyhedron, 14, 3481 (1995)

- 1995HCa M Hlaibi,S Chapelle,M Benaissa et al; Inorg.Chem.,34,4434 (1995)
 1995SPb J Smit,W Purcell,A Roodt,J Leipoldt; J.Chem.Soc.,Dalton Trans.,1201
 (1995)
- 1994AHa I Andersson,J Hastings,O Howarth; J.Chem.Soc.,Dalton Trans.,1061 (1994)
 1994RLa A Roodt,J Leipoldt,L Helm et al; Inorg.Chem.,33,140 (1994)
 1993CKb J Cruywagen,L Kruger,E Rohwer; J.Chem.Soc.,Dalton Trans.,105 (1993)
 1993PSa A Patel,S Siddiqui,D Richens et al; J.Chem.Soc.,Dalton Trans.,767 (1993)
 1993VSA J Varey,A Sykes; J.Chem.Soc.,Dalton Trans.,3293 (1993)
 1992BCc W Buhro,M Chisholm et al; J.Am.Chem.Soc.,114,557 (1992)
 1992HMa K Hall,J Mayer; J.Am.Chem.Soc.,114,10402 (1992)
 1992LJa A Lindmark; Inorg.Chem.,31,3507 (1992)
 1992RSb C Routledge,A Sykes; J.Chem.Soc.,Dalton Trans.,325 (1992)
 1992RZa D Rabinovich,R Zelman et al; J.Am.Chem.Soc.,114,4611 (1992)
 1992WOa S Wood; Geochim.Cosmo.Acta,56,1827 (1992)
 1992WWa Z Wu,D Wheeler et al; J.Am.Chem.Soc.,114,146 (1992)
 1991CKa J Cruywagen,L Kruger,E Rohwer; J.Chem.Soc.,Dalton Trans.,1727 (1991)
 1991JMa M Jelikic-Stankov,D Malesev; Polyhedron,10,455 (1991)
 1991ZGa K Zhang,A Gonzalez et al; J.Am.Chem.Soc.,113,9170 (1991)
 1990CJa A Cavaleiro,J de Jesus,V Gil et al; Inorg.Chim.Acta,172,25 (1990)
 1990VSc J Verchere,J Sauvage,G Rapaumba; Analyst,115,637 (1990)
 1989VCa J Verchere,S Chapelle; Polyhedron,8,333 (1989)
 1989VJa D Veselinovic,M Jelikic-Stankov; Mikrochim.Acta,329 (1989)
 1986LBa J Leipoldt,S Basson,A Roodt et al; S.Afr.J.Chem.,39,179 (1986)
 1981BMD G Bianchi,C Marone; J.Inorg.Nucl.Chem.,43,2985 (1981)
 1980NKA S Natansohn,J Krugler et al; J.Phys.Chem.,84,2972 (1980)
 1979CBA A Cervilla,A Beltran,J Beltran; Can.J.Chem.,57,773 (1979)
 1979IRa A Ilyasova,H Rakhimbekova; Koord.Khim.,5,395 (1979)
 1979ZLa K Zare,P Lagrange et al; J.Chem.Soc.,Dalton Trans.,1372 (1979)
 1978KKc D Kepert,J Kyle; J.Chem.Soc.,Dalton Trans.,1781 (1978)
 1978MBC M Mikesova,M Bartusek; Coll.Czech.Chem.Comm.,43,1867 (1978)
 1978SSc A Samotus,B Sieklucka; J.Inorg.Nucl.Chem.,40,315 (1978)
 1977NPb V Nazarenko,E Poluektova,G Shitareva; Zh.Neorg.Khim.,22,998(551) (1977)
 1976CLa J Collin,P Lagrange; Bull.Soc.Chim.,France,1304 (1976)
 1975PPb B Pham,J Podlahova; Collect.Czech.Chem.Commun.40,347 (1975)
 1974ASA R Arnek,Y Sasaki; Acta Chem.Scand.,A28,20 (1974)
 1974SBC A Steigel,S Brownstein; J.Am.Chem.Soc.,96,6227 (1974)
 1973Hka E Hejmo,A Kanas et al; Bull.Acad.Polon.Sci.Chim.,21,311 (1973)
 1973SDa D Shishkov,H Doichinova; Dokl.Bulg.Akad.Nauk,26,927 (1973)
 1973TSe R Tewari,M Srivastava; Talanta,20,133;360 (1973)
 1972JRa B Jezowska-Trzebiatowska,M Rudolf; Trans.Roy.Inst.Tech.(Stockholm),258
 (1972)
- 1972PSb E Poluektova,G Shitareva; Zh.Anal.Khim.,27,7,1301 (1972)
 1972SSe M Singh,M Srivastava; J.Inorg.Nucl.Chem.,34,567;2067;2081 (1972)
 1971SKc A Samotus,B Kosowicz-Czajkowska; Roczn.Chem.,45,1623 (1971)
 1970SHa D Shishkov; God.Vissn.Khimikotekhnol.I.Sof.,15,415 (1970)
 1969Gta S Goyal,J Tandon; Talanta,16,106 (1969)
 1969NPd V Nazarenko,E Poluektova; Zh.Neorg.Khim.,14,204(E:105) (1969)
 1969SHd D Shishkov; Dokl.Bulg.Akad.Nauk,22,763 (1969)
 1968ABA Y Atoks,Y Bankovskii; Izv.Akad.Nauk Latv.SSR,Khim.,1,122 (1968)
 1967JRa B Jezowska-Trzebiatowska,M Rudolf; Roczn.Chem.,41,453;1879 (1967)

1967PNa E Poluektova,V Nazarenko; Zh.Anal.Khim.,22,5,746 (1967)
1967USa N Ulko,R Savchenko; Zh.Neorg.Khim.,12,328 (1967)
1966KRa R Kula,D Rabenstein; Anal.Chem.,38,1934 (1966)
1966Kwb G Kirkbright,T West,C Woodward; Talanta,13,1637 (1966)
1964GUa R Gut; Helv.Chim.Acta,47,2262 (1964)
1964PCa Personal Communication etc; Chem.Soc.Spec.Publ.,no.17 (1964)
1964YPa K Yatsimirskii,K Prik; Zh.Neorg.Khim.,9,178 (1964)
1963DLa A Dedman,T Lewis,D Richards; J.Chem.Soc.,5020 (1963)
1963HAc J Halmekoski; Suomen Kem.,B36,29;40;46;55 (1963)
1962YBa K Yatsimirskii,L Budarin; Zh.Neorg.Khim.,7,942 (1824) (1962)
1962YPa K Yatsimirskii,K Prik; Zh.Neorg.Khim.,1,821 (1589) (1962)
1962YPb K Yatsimirskii,K Prik; Zh.Neorg.Khim.,7,821 (1589) (1962)
1960NVA N Nikolaev,S Vlasov,Y Buslaev et al; Izv.Sib.Otd.Akad.Nauk SSR,47 (1960)
1959DBb A Dey,S Banerji; Proc.Symp.Chem.of Coord.Comp.,Agra,198 (1959)
1959HAa J Halmekoski; Ann.Acad.Sci.Fennicae,96 (1959)
1959HEb W Hieber,K Englert,K Rieger; Z.Anorg.Chem.,300,295;304;311 (1959)
1959LMa V Litvinchuk,K Mikhalevich; Ukr.Khim.Zh.,25,563 (1959)
1958PIa E Pisko; Chem.Zvesti,12,95 (1958)
1958SAC P Sakellaridis; Chimika Chronika,23,263 (1958)
1956BAa H Baadsgaard; Diss.Eid.Tech.Hochschule,Zurich (1956)
1955BTa H Baadsgaard,W Treadwell; Helv.Chim.Acta,38,1669 (1955)
1952LAb W Latimer; "Oxidation Potentials",Prentice Hall,NY (1952)
1924COa O Collenberg; Z.Phys.Chem.,109,353 (1924)

EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES
I Data with various BACKGROUNDS
H Data for THERMOCHEMICAL quantities
M Data for TERNARY Complexes

EVALUATION Flags are :-

END