

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 271 experiments for
(no ligands specified)

Metal : Zr++++

(no references specified)

(no experimental details specified)

e- HL Electron (442)
Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	oth	none	25°C	0.0	U				1952Lab	(1037) 1

K=-96.0(-1430 mV)

K: ZrO₂(s)+4H+4e=Zr(s)+2H₂O. From thermodynamic data

C03--	H2L	Carbonate	CAS	465-79-6	(268)
Carbonate;					

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	dis	oth/un	20°C	1.00M	U	I			1987JBb	(3454) 2

B4=39.95

When I=2.5 M: B4=39.59

Zr++++	gl	KCl	25°C	1.00M	U				1982KCc	(3455) 3
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K(Zr(OH)₂L+L)=11.2

Zr++++	gl	oth/un	20°C	1.0M	U				1980MCg	(3456) 4
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K4=10.9

Cl-	HL	Chloride	CAS	7647-01-0	(50)
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Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	cal	non-aq	25°C	100%	U	HM			1993DSb	(5964) 5

Medium: Toluene. DH(Zr(H-1A)2B2(s)+2HL=Zr(H-1L)2A2+2HB)=-373.5 kJ mol-1.
A:Cyclopentadiene. B:CH₃. Also for B=C4Ph4: DH=-300.5 kJ mol-1.

Zr++++	cal	non-aq	25°C	100%	U	HM			1993DSb	(5965) 6
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Medium: Toluene. DH([Zr(H-1A)2L]20(s)+2HL=2ZrL2(H-1A)2+H₂O)=-82.4 kJ mol-1.
A:Cyclopentadiene. In isopropylether, DH=-37.5 kJ mol-1.

Zr++++	dis	NaClO ₄	25°C	4.00M	U	M			1976TSa	(5966) 7
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K(ZrOH+Cl)=-0.52

Zr++++	sp	NaClO ₄	?	3.50M	U		K1=-0.5		1972TSa	(5967) 8
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Medium: HClO4

Zr++++ dis NaClO4 20°C 2.0M U K1=0.08 B2=-0.54 1970PHb (5968) 9
B3=-1.0

Medium: HClO4

Zr++++ ix NaClO4 ? 4.0M U I K1=0.04 B2=-0.68 1962MRc (5969) 10
B3=-1.30

Medium: HClO4. In 2 M HClO4: K1=-0.02, B2=-0.92, B3=-1.15, B4=-1.10

Zr++++ dis NaClO4 20°C 6.54M U K1=0.92 B2=1.32 1957S0b (5970) 11
K3=0.19
K4=-0.33

Zr++++ dis NaClO4 25°C 2.0M U K1=0.30 1949CMc (5971) 12

C1O4- HL Perchlorate CAS 7001-90-3 (287)
Perchlorate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ con oth/un ? dil U 1961MPb (6394) 13
K(ZrOL+L)=2.73?

F- HL Fluoride CAS 7644-39-3 (201)
Fluoride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ cal NaClO4 25°C 4.0M U H 1990AHa (7363) 14
DH(Zr+HF=ZrF+H)=17.5 kJ mol-1; DH(ZrF+HF=ZrF2+H)=16.8;
DH(ZrF2+HF=ZrF3+H)=11.2; DH(ZrF3+HF=ZrF4+H)=22

Zr++++ ISE NaNO3 25°C 0.10M U H 1987SMd (7364) 15
K(ZrA+L)=3.04

DH=-12.6 kJ mol-1, DS=16.4 J K-1 mol-1. H3A=HEDTA

Zr++++ cal none 25°C 0 U IH 1976VKb (7365) 16
DH(Zr+3L)=-25.1 kJ/mol
DH(Zr+4L)=25.5 kJ/mol
DH(Zr+5L)=7.5 kJ/mol
DH(Zr+6L)=23.4 kJ/mol

Zr++++ ISE NaClO4 25°C 4.0M U 1973N0a (7366) 17
*K1=5.97
*K2=4.4

Medium: (H,Na)ClO4. *Kn: ZrF(n-1)+HL=ZrFn+H

Zr++++ ix oth/un ? ? U 1972PAb (7367) 18
K6=3.84

Zr++++	ix	NaClO ₄	25°C	1.0M	U	I	1969KKe	(7368)	19
							*K1=5.32		
							*B2=9.11		
Medium:	HCLO ₄	; *K1=5.81(I=2); *B2=9.38(I=2).	In HNO ₃ , I=1: *K1=5.41, *B2=9.65						
I=2: *K1=5.79, *B2=9.53.	*Bn:	Zr+nHL=ZrLn+nH							
Zr++++	dis	NaClO ₄	20°C	4.0M	U		1969N0a	(7369)	20
							*K1=5.88		
							*K2=4.36		
							*K3=3.00		
							*K4=2.3		
Medium:	HCLO ₄ .	*Kn:	ZrF(n-1)+HF=ZrFn+F						
Zr++++	EMF	NaClO ₄	20°C	4.0M	U		1967N0a	(7370)	21
							K(ZrF+HF=ZrF ₂ +H)=4.42		
							K(ZrF ₂ +HF=ZrF ₃ +H)=3.00		
							K(ZrF ₃ +HF=ZrF ₄ +H)=2.28		
							K(ZrF ₄ +HF=ZrF ₅ +H)=1.53		
Medium:HCLO ₄ .	K(ZrF ₅ +HF=ZrF ₆ +H)=0.3								
Zr++++	EMF	oth/un	?	0.50M	U		1967PMa	(7371)	22
						K6=3.6			
Medium:	HCl								
Zr++++	EMF	NaClO ₄	25°C	1.0M	U		1966BFb	(7372)	23
						K4=2.8			
						K5=1.9			
						K6=1.35			
Method:quinhydrone electrode									
Zr++++	ix	NaClO ₄	20°C	4.00M	U		1963AKc	(7373)	24
						K(Zr+HF=ZrF+H)=5.96			
						K(ZrF+HF=ZrF ₂ +H)=4.54			
Method: cation exchange. Medium:HCLO ₄									
Zr++++	con	NaClO ₄	25°C	0.50M	U		1962BUb	(7374)	25
						K(ZrF ₂ +HF=ZrF ₃ +H)=2.7			
						K(ZrF ₃ +HF=ZrF ₄ +H)=1.83			
						K(ZrF ₄ +HF=ZrF ₅ +H)=1.51			
						K(ZrF ₅ +HF=ZrF ₆ +H)=0.86			
Zr++++	dis	none	25°C	0.0	U	K1=9.80	1955PAa	(7375)	26
Zr++++	dis	NaClO ₄	25°C	2.00M	U		1949CMc	(7376)	27
						K(Zr+HF=ZrF+H)=5.80			
						K(ZrF+HF=ZrF ₂ +H)=4.32			
						K(ZrF ₂ +HF=ZrF ₃ +H)=2.83			

MoO₄-- H2L Molybdate (443)

Molybdate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	kin	oth/un	25°C	var	U			K1=9.11 K(Zr+H2L=ZrL+2H)=4.18	1963YRa (8764)	28

NO3-		HL	Nitrate	CAS	7697-37-2	(288)
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	dis	NaClO4	20°C	2.0M	U			K1=-0.05 K(Zr0+2L)=1.91 K(Zr0+4L)=1.63	1970PHb (10015)	29

Medium:	HC1O4									
Zr++++	ix	NaClO4	?	4.0M	U	I		K1=-0.06 K1=-0.04, B2=-0.34	1962MRc (10016)	30

Zr++++	dis	NaClO4	25°C	2.50M	U				1962NPc (10017)	31
Medium:	HC1O4.	Kd(Zr0+2H+4L+2T(CC14))=2.37; T=(BuO)2BuPO								
Zr++++	dis	oth/un	23°C	0.0	U	M			1961UHb (10018)	32
K(Zr+4L+2TBP(CC14))=ZrL4(TBP)2(CC14))=-1.8										
Zr++++	ix	NaClO4	?	2.0M	U		B2=0.22		1958PSb (10019)	33
Zr++++	dis	NaClO4	20°C	4.0M	U		K1=0.34 K3=-0.37 K4=-0.56 B5=-1.5 B6=-1.7	B2=0.11	1957S0b (10020)	34

Zr++++	dis	NaClO4	25°C	2.0M	U		K1=0.3		1949CMc (10021)	35
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OH-		HL	Hydroxide	(57)
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Zr++++	gl	NaClO4	25°C	1.0M	C	T	H		2004EKa (12532)	36
								*K1=-0.87 *B(2)=-2.9 *B3=-4.9 *B4=-6.7		

Additional method extraction of 95Zr with acac into toluene. *B(2,6)=-2.42
*B(4,8)=5.23. Data for 15 and 35 C. DH(*K1)=63, DH(*B4)=250, DH(*B2,6)=150

Zr++++	sol	NaClO4	25°C	1.0M	C				2004EKa (12533)	37
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$K_{s4}=-7.3$
 $*K_{so}=-0.70$
 $K_{s}(Zr(OH)4=ZrO_3+H_2O+2H)=-33.26$
 $K(Zr+3H_2O=ZrO_3+6H)=-32.56$

$K_{s4}: Zr(OH)4(s)=Zr(OH)4(aq).$

Zr++++ gl KNO₃ 25°C 0.20M U I 1998VDa (12534) 38
 $*B_3=-2.02$
 $*B_4=-6.09$
 $*B(2,7)=-5.26$

At I=0.50 M: $*B_3=-2.18$, $*B_4=-6.7$, $*B(2,7)=-5.39$.

At I=2.0 M: $*B_3=-1.77$, $*B_4=-6.9$, $*B(2,7)=-4.35$.

Zr++++ dis oth/un 25°C 0.00 U K1=12.0 1975CCa (12535) 39
 $*B_2=-3.5$

Zr++++ dis NaClO₄ 20°C 4.00M U T 1973N0a (12536) 40
 $*K_1=0.3$

Medium: HClO₄; $*K_1=-0.55$ (25 C) determined with fluoride-ion selective electrode

Zr++++ sp NaClO₄ ? 3.50M U 1972TSa (12537) 41
 $*B(3Zr+2H_2O=Zr_3O_2+4H)=ca.6$
 $K(3ZrOH+H_2O=(ZrO)_3OH+4H)=ca.6$

Zr++++ sp NaClO₄ 25°C 0.10M U I K1=14.1 B2=27.8 1969NMB (12538) 42
 $B_3=41.4$
 $B_4=54.6$

K1=14.1, B2=27.9, B3=41.5, B4=54.7(I=0.3); K1=14.2, B2=28.0, B3=41.6, B4=54.8(I=0.5); K1=14.3, B2=28.2, B3=41.9, B4=55.2(I=1)

Zr++++ dis none 25°C 0.0 M 1967STe (12539) 43
 $*K_3=1.13$
 $*K_4=1.13$
 $*K_1=0.68$
 $*K_2=0.90$

Zr++++ oth NaClO₄ 20°C 1.00M U T 1966BBa (12540) 44
 $*K_{so}(Zr(OH)4(s)+4H)=3.8$
 $K(Zr(OH)4(s)=Zr(OH)4)=-4.36$
 $*K_{so}=4.6(20\text{ C}), 5.05(40\text{ C}); K(Zr(OH)4(s)=Zr(OH)4)=-3.9(20 - 40\text{ C})$

Zr++++ oth oth/un 20°C var U 1966BBe (12541) 45
 $K(Zr(OH)4(s)=Zr(OH)4)=-4.6$

Medium: sea water. Method: Tyndall scattering

Zr++++ oth oth/un 25°C ? U 1966BBe (12542) 46
 $K(Zr(OH)4(s)=Zr(OH)4)=-4.6$

Medium: 50% sea water

Zr++++	oth	oth/un	?	2.0M	U	K1=11.77	1966LJa (12543)	47
Method:Literature evaluated data								
Zr++++	dis	none	25°C	0.0	M	K1=14.58 B2=29.38 B3=43.72 B4=57.85	1966SID (12544)	48

Zr++++	sol	none	19°C	0.0	U	1961KBc (12545)	49	
						Kso(Zr(OH)4)=-53.96 ?		

Zr++++	dis	NaClO4	25°C	1.0M	U	K1=14.32 K3=13.65 K4=13.36	1961PMb (12546)	50

Zr++++	sol	NaCl	?	2.0M	U	*B2=0.3	1961PPb (12547)	51

Zr++++	sol	oth/un	?	1.0M	U	1960SPa (12548)	52	
						K(Zr(OH)4(s)+OH=Zr(OH)5)=-3.6		
Medium: NaOH								
Zr++++	sol	oth/un	?	var	U	1958STb (12549)	53	
						Kso(Zr(OH)4)=-52		

Zr++++	dis	NaClO4	25°C	2.0M	U	1957SOb (12550)	54	
						*K1=-0.22 *K2=-0.62 *K3=-1.05 *K4=-1.17		

Zr++++	sol	oth/un	?	var	U	1957SRC (12551)	55	
						Kso(Zr(OH)4)=-52		

Zr++++	sp	oth/un	25°C	2.0M	U	1956ZCa (12552)	56	
						*B(3,4)=5.38 *B(4,8)=8.30?		

Zr++++	sp	NaClO4	25°C	2.0M	U	1956ZCa (12553)	57	
						*B(3,4)=5.40 *B(4,8)=8.18?		

Zr++++	dis	NaClO4	25°C	4.0M	U	1953ZJa (12554)	58	
						*B(3,5)=6.60 *B(4,8)=9.15		

Zr++++	gl	oth/un	25°C	var	U	1950LGa (12555)	59	
						Ks(Zr(OH)4=Zr(OH)2+2OH)=-25.5		

Zr++++	gl	oth/un	25°C	dil	U	19380Ka (12556)	60	
						Kso(Zr(OH)4)=-48.2		

O2-- H2L Peroxide CAS 7772-84-1 (2813)
Peroxide; -0.0-

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	sp	oth/un	25°C	2.00M	U				1985THc (12760)	61
								*B(2,4)=-19.38		
Zr++++	sp	oth/un	25°C	var	U				1973KPF (12761)	62
								K(ZrO+H2L)=3.2 to 4.8		
Zr++++	sp	oth/un	20°C	1.0M	U				1970SKe (12762)	63
								K(ZrO+H2L)=3.10 (?)		
Zr++++	sp	oth/un	?	dil	U				1961BUb (12763)	64
								K(2ZrO+H2L)=5.6		

P04--- H3L Phosphate CAS 7664-38-2 (176)
Phosphate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	sp	NaClO4	20°C	1.00M	U				1972DSg (13390)	65
								K(Zr+3HL)=1.59		
Zr++++	oth	oth/un	?	?	U				1945BUa (13391)	66
								Ks(ZrO(H2L)2=ZrO+2H2L)=-17.64		

SCN- HL Thiocyanate CAS 463-56-9 (106)
Thiocyanate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	dis	NaClO4	25°C	3.50M	U	M			1976TSa (15344)	67
								K(ZrOH+SCN)=0.95		
Zr++++	sp	NaClO4	?	3.50M	U		K1=1.06		1972TSa (15345)	68
Zr++++	dis	NaClO4	?	3.0M	U		K1=1.08	B2=2.33	1971LFb (15346)	69
							B4=2.31			
Zr++++	sp	non-aq	20°C	100%	U	I	K1=2.17	B2=3.17	1970GLa (15347)	70
							B3=6.04			
							B4=7.79			
							B5=9.43			
							B6=10.96			

B7=12.33, B8=13.72; Medium: N,N-dimethylformamide; In acetonitrile, B6=18.6

Zr++++	sp	NaClO4	?	?	U	I	K1=3.8	B2=7.3	1966GSi (15348)	71
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			B3=10.8
			B4=14.0
			B5=15.1
			B6=21.0
B7=22.9, B8=25.6; constants for 0.8 M H+. Data also for 0.1 M H+: K1=2.0, B2=3.4, B3=4.7, B4=5.8, B5=6.9, B6=7.9, B7=8.9, B8=9.9			
Zr++++	sp oth/un	20°C var U	1963SCf (15349) 72 K(ZrO(OH)+L)=1.15
S04--	H2L Sulfate	CAS 7664-93-9 (15)	
Sulfate;			
Metal	Mtd Medium Temp Conc Cal Flags Lg K values	Reference	ExptNo
Zr++++	kin NaClO4 25°C 0.02M U	1979ABb (16683)	73
	K(Zr(OH)3+HSO4=Zr(OH)2SO4+H2O)=4.96		
Zr++++	dis NaClO4 25°C 4.00M U	K1=1.60 B2=2.72	1976TSa (16684) 74
Zr++++	dis NaClO4 20°C 2.0M U	1970PHb (16685) 75 K(Zr+HL=ZrL+H)=2.6 K(Zr+2HL=ZrL2+2H)=4.5 K(Zr+3HL=ZrL3+3H)=5.5	
Medium: 2 M HClO4			
Zr++++	dis NaClO4 20°C 4.0M U	1969NOa (16686) 76 *K1=2.8 *K2=0.9 *K3=0.3	
Medium: HClO4			
Zr++++	ix NaClO4 ? 2.30M U	1964RMd (16687) 77 *K1=2.67 *B2=3.54 *B3=6.59	
Medium: HClO4			
Zr++++	ix NaClO4 20°C 4.0M U	1963AKc (16688) 78 *K1=2.85 *K2=1.85	
Zr++++	ix NaClO4 ? 2.30M U	1962REb (16689) 79 *K1=2.56 *B2=3.34 *B3=5.61	
Zr++++	sol oth/un 20°C var U	1959SAb (16690) 80 K(2ZrO+L)=8.73	

$$K(Zr+2H2L=ZrL2+4H)=9.7$$

Medium: HClO₄, I=4.0, K(Zr+H₂L=ZrL+2H)=5.60

Zr++++ gl oth/un 25°C ? U 1961GAa (19168) 91
K₃(?)=4.0

C₂H4O₂ HL Acetic acid CAS 64-19-7 (36)
Ethanoic acid; CH₃.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Zr++++ kin none 25°C 0.00 U 1973VPa (20239) 92
K(Zr(OH)₂+L)=6.18

Zr++++ EMF oth/un 25°C 0.01M U 1969KPb (20240) 93
K(Zr(OH)₃+L)=3.35
K(Zr(OH)₃L+L)=1.83

Medium: ZrOCl₂

C₂H4O₃ HL Glycolic acid CAS 79-14-1 (33)
2-Hydroxyethanoic acid; HO.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Zr++++ kin none 25°C 0.00 U 1973KPg (20667) 94
K(Zr(OH)₂+L)=7.7

Zr++++ sp KCl ? 1.00M U 1970KKd (20668) 95
K(Zr(OH)₂+L)=6.48
K(Zr(OH)₃+L)=6.57

C₂H₈O₇P₂ H₄L HEDPA CAS 2809-21-4 (436)
1-Hydroxyethane-1,1-diphosphonic acid; CH₃.C(OH)(PO₃H₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Zr++++ gl KCl 25°C 0.10M U 1967KLa (23407) 96
K(ZrO+L)=15.18
K(2ZrO+H-1L)=26.04
K(2ZrO+L)=20.40
K(ZrO+2H-1L)=21.92

$$K(ZrO+2L)=18.63$$

C₂H₈O₇P₂ H₅L CAS 76267-75-9 (4226)
2-Hydroxyethylidenediphosphonic acid; HO.CH₂.CH(PO₃H₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Zr++++ sp oth/un 25°C 2.0M U 1999VKa (23411) 97
K(Zr+H₃L=ZrH₃L)=7.82

In 2.0 M HClO₄, T=room

C3H2O5 H2L Mesoxalic acid (3544)
Oxopropanedioic acid; HOOC.CO.COOH (Ketomalonic acid)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr+++ ix NaClO₄ ? 2.0M U K1=4 1960REa (23490) 98

C3H4O4 H2L Malonic acid CAS 141-82-2 (79)
Propanedioic acid; CH₂(COOH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr+++ kin oth/un 25°C var U 1961YRa (24602) 99
 $K(Zr(OH)_3 + H_2L = Zr(OH)_3L + 2H) = 1.46$

C3H6O2 HL Propionic acid CAS 79-09-4 (35)
Propanoic acid; CH₃.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr+++ kin oth/un 25°C 0.05M U M 1977VPa (25079) 100
 $B(Zr+2OH+L) = 36.40$

Zr+++ EMF oth/un 25°C 0.01M U M 1969KPb (25080) 101
 $K(Zr(OH)_3 + L) = 3.77$

Medium: ZrOCl₂

C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)
L-2-Hydroxypropanoic acid; CH₃.CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr+++ ix NaClO₄ ? 2.0M U 1964RMd (25582) 102
 $K(Zr+HL = ZrL+H) = 2.28$
 $K(Zr+2HL = ZrL₂+2H) = 2.5$

Zr+++ ix oth/un ? 2.0M U 1960REa (25583) 103
 $K(Zr+HL = ZrL+H) = 1.98$

C3H7N02S H2L Cysteine CAS 52-90-4 (96)
2-Amino-3-mercaptopropanoic acid; H₂N.CH(CH₂.SH)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr+++ gl NaNO₃ 15°C 0.10M U T K1=14.40 1984IDa (26854) 104
At 30 °C, K1=13.15.

C3H8O L isoPropanol CAS 67-63-0 (2024)

2-Propanol; CH₃.CH(OH).CH₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ cal non-aq 25°C 100% U HM 1993DSb (27647) 105

Medium: Toluene or iso-propyl ether. DH(Zr(H-1A)2BH(s)+L=Zr(H-1A)2B(H-1L)+H₂)=-69.5 kJ mol-1. A:Cyclopentadiene. B:Cl. Also data for L= MeOH, EtOH etc

C3H12N0P3 H6L NTPA CAS 6419-19-8 (2920)

Nitrilotris(methylenephosphonic acid); N(CH₂P03H₂)₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ sp oth/un 25°C 2.0M U 1999VKa (28601) 106

K(Zr+H3L=ZrH3L)=13.04

In 2.0 M HClO₄, T=room

C4H6O4 H2L Succinic acid CAS 110-15-6 (112)

1,4-Butanedioic acid; HOOC.CH₂.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ kin oth/un 25°C var U 1961YRa (30078) 107

K(Zr(OH)₃+H2L)=1.46

C4H6O4S H3L Thiomalic acid CAS 70-49-5 (109)

2-Mercaptosuccinic acid, 2-Sulfanyl-1,4-butanedioic acid; HOOC.CH(SH).CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ gl oth/un ? .005M U 1965SSNa (30380) 108

K(ZrO+L)=9.6

C4H6O5 H2L Malic acid CAS 617-48-1 (393)

2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH₂.CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ ix NaClO₄ ? 2.0M U 1964RMd (30768) 109

K(Zr+H2L=ZrHL+H)=2.24

Zr++++ ix NaClO₄ ? 2.0M U 1960REa (30769) 110

K(Zr+H2L=ZrHL+H)=1.94

Medium: HClO₄

C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)

L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ sp KCl 25°C 1.00M U 1978KKf (31405) 111
K(Zr(OH)+HL)=8.76

Zr++++ EMF oth/un 25°C dil U T 1973KKf (31406) 112
K(Zr(OH)3+L)=6.07
K(20 C)=6.15; K(30 C)=5.92; K(35 C)=5.84

Zr++++ ix oth/un ? ? U 1973ZGb (31407) 113
K(Zr(OH)2+H2L)=9.80

Zr++++ EMF oth/un 25°C ? U 1970KKb (31408) 114
K(Zr(OH)3+L)=6.09
K(Zr(OH)3L+L)=3.06

Zr++++ oth oth/un ? ? U 1968KOa (31409) 115
K(?)=5.31

Method: metal indicator

Zr++++ sp KCl 19°C 1.0M U 1966KSe (31410) 116
K(Zr+HL)=6.86

Medium: HCl. 18-20 C

Zr++++ ix NaClO4 ? 2.0M U 1964RMd (31411) 117
K(Zr+H2L=ZrHL+H)=2.49

Zr++++ kin oth/un 25°C var U 1961YRa (31412) 118
K(Zr(OH)3+H2L)=5.51
K(Zr(OH)L+H2L=ZrL2+H)=0.95

Zr++++ ix NaClO4 ? 1.0M U I 1960REa (31413) 119
K(Zr+H2L=ZrHL+H)=3.15

Medium: HClO4

Zr++++ dis NaClO4 ? 2.0M U 1959REa (31414) 120
K(Zr+H2L=ZrHL+H)=2.19

Medium: HClO4

C4H6O6 H2L meso-Tartaric CAS 147-73-9 (91)
meso-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ sp KCl 25°C 1.00M U 1978KKf (31433) 121
K(Zr(OH)+HL)=9.40

C4H7N04 H2L Aspartic acid CAS 56-84-8 (21)
Aminobutanedioic acid; H2N.CH(CH2.COOH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ gl NaClO₄ 25°C 0.10M U K1=9.70 B2=16.55 1972SSg (31985) 122
K3=3.50

C4H7N04 H2L IDA CAS 142-73-4 (118)
Iminodiethanoic acid; HN(CH₂.COOH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ ix oth/un ? 2.0M U 1964PVc (32409) 123
K(?)=3.45

C4H8N203 HL Asparagine CAS 70-47-3 (17)
2-Aminobutanedioic acid 4-amide; H2N.CH(CH₂.CO.NH₂).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ gl KNO₃ 25°C 0.10M U T K1=8.50 B2=16.05 1986SSe (32751) 124
Data for 25-45 C and 0-1.0 M KNO₃. DH and DS values reported.

Zr++++ gl NaClO₄ 25°C 0.10M U K1=8.80 B2=15.05 1973TSe (32752) 125

C4H8O2 HL CAS 107-92-6 (1118)
n-Butanoic acid; CH₃.CH₂.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ EMF oth/un 25°C 0.01M U 1969KPb (33357) 126
K(Zr(OH)₃+L)=3.78

C5H5N L Pyridine CAS 110-86-1 (31)
Pyridine, Azine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ oth none 25°C 0.0 U HM 1958ERb (36692) 127
DG(ZrCl₄(s)+2L(g)=ZrCl₄L₂(S))=-32.8 kJ mol⁻¹, DH=-71, DS=-134. Also for ZrBr₄

C5H8O2 HL Acetylacetone CAS 123-54-6 (164)
Pentane-2,4-dione; CH₃.CO.CH₂.CO.CH₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ dis NaClO₄ 25°C 1.0M C T H K1=11.5 B2=21.60 2004EKa (38148) 128
B3=30.6
B4=37.3

Extraction of 95Zr with acac into toluene. Data for 15 and 35 C.
DH(K1)=-40 kJ mol⁻¹, DH(B2)=230, DH(B3)=130, DH(B4)=140.

Zr++++ sp KCl 19°C 1.0M U K1=11.25 1966KSc (38149) 129
Medium: HCl

C5H8O7 H2L CAS 40120-71-6 (3022)
2,3,4-Trihydroxypentanedioic acid, Trihydroxyglutaric acid; HOOC.(CH(OH))3.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	sp	KCl	25°C	1.00M	U				1978KKf (38447)	130
								K(Zr(OH)+HL)=9.72		
Zr++++	ix	NaCl	?	2.25M	U				1973ZGb (38448)	131
								K(Zr(OH)2+L)=11.37		
Zr++++	EMF	oth/un	25°C	?	U				1970KKb (38449)	132
								K(Zr(OH)3+L)=6.86		
Zr++++	ix	NaClO4	?	2.0M	U				1964RMd (38450)	133
								K(Zr+H2L=ZrHL+H)=3.41		
								K(Zr+2H2L=Zr(HL)2+2H)=5.40		

Medium: HClO4

Zr++++	ix	NaClO4	?	2.0M	U				1960REa (38451)	134
								K(Zr+H2L=ZrHL+H)=3.09		
								K(Zr+2H2L=Zr(HL)2+2H)=4.14		

Medium: HClO4

C5H9NO4 H2L Glutamic acid CAS 56-86-0 (22)

2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	gl	NaClO4	25°C	0.10M	U			K1=9.60 B2=16.00 K3=3.32	1972SSg (39150)	135

C5H10N2O3 HL Glutamine CAS 56-85-9 (18)

2-Aminopentanedioic acid 5-amide; H2N.CH(CH2.CH2.CO.NH2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	gl	NaClO4	25°C	0.10M	U			K1=8.75 B2=14.85	1973TSe (39850)	136

C5H10O2 HL n-Valeric acid CAS 109-52-4 (3027)

Pentanoic acid; CH3(CH2)3.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	EMF	oth/un	25°C	0.01M	U	M			1969KPb (40205)	137
								K(Zr(OH)3+L)=3.94		

Medium: 0.01 ZrOCl2

C6H204C12 H2L Chloranilic acd CAS 87-88-7 (1281)

3,6-Dichloro-2,5-dihydroxy-1,4-benzoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	sp	NaClO4	25°C	2.0M	U				1951TVa (42062)	138
K(Zr+H2L=ZrL+2H)=5.76										
K(ZrL+H2L=ZrL2+2H)=3.78										

C6H5N02		HL	Nicotinic acid	CAS	59-67-6	(419)				
3-Pyridine-carboxylic acid; C5H4N.COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	gl	KNO3	25°C	0.10M	U			K1=10.18 B2=19.28	1988ZMa (42693)	139
Metal is ZrO++.										

C6H5N3O4		L	Dinitroaniline	CAS	618-87-1	(1938)				
3,5-Dinitroaminobenzene; H2N.C6H3(NO2)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	sp	diox/w	25°C	100%	U				1975BSb (42990)	140
K(ZrBr4+2L)=4.15										

C6H6N2O2		L	o-Nitroaniline	CAS	88-74-4	(463)				
2-Nitroaminobenzene; H2N.C6H4.NO2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	sp	diox/w	25°C	100%	U				1975BSb (43362)	141
K(ZrBr4+2L)=4.78										

C6H6O		HL	Phenol	CAS	108-95-2	(457)				
Hydroxybenzene, phenol; C6H5.OH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	cal	non-aq	25°C	100%	U	HM			1993DSb (43551)	142
Medium: Toluene or iso-propyl ether. DH(Zr(H-1A)2BH(s)+HL=ZrL(H-1A)2B+H2)=-80.6 kJ mol-1. A:Cyclopentadiene. B:Cl.										

C6H6O2		H2L	Catechol	CAS	120-80-9	(534)				
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	ix	NaClO4	?	1.0M	U			K1=22.63	1967EKb (43872)	143
Medium: HClO4										

C6H6O3		H3L	Pyrogallol	CAS	87-66-1	(696)				

1,2,3-Trihydroxybenzene; C6H3(OH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	sp	KCl	19°C	0.10M	U	I			1966PRc (44003)	144
$K(Zr(OH)_3 + H_3L = Zr(OH)_2HL + H) = 4.17 (I=0), 4.06 (I=0.1)$										
C6H608S2		H4L	Tiron				CAS	149-45-1	(104)	

4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)₂C₆H₂(SO₃H)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	ix	oth/un	?	1.0M	U	I		K1=24.15	1967EKb (44526)	145
$K(Zr+H_2L = ZrL+2H) = 3.89$										
Medium: HClO ₄ . K=3.95 (I=0.5), 3.00 (I=2.0)										
Zr+++	gl	oth/un	25°C	0.10M	U	M			1964IMa (44527)	146
$K(ZrY+H_2L = ZrYHL+H) = -12$										
$K(Zr_2Y_2L_2 + 2H = 2ZrYHL) = 3.70$										

H4Y=EDTA

C6H703As	H2L	Phenylarsonic	CAS	98-05-5	(3690)					
Benzeneearsonic acid, phenylarsonic acid; C ₆ H ₅ AsO ₃ H ₂										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	sol	KNO ₃	?	2.0M	U			B2=30.6	1967KPb (45180)	147
Medium: HNO ₃										

Zr+++	sol	oth/un	18°C	0.10M	U			K1=11.5	1960MIa (45181)	148
$K(Zr+HL) = -12$										
C6H806	H2L	Ascorbic acid	CAS	50-81-7	(285)					

Ascorbic acid (Vitamin C);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	sp	oth/un	?	?	U				1966SAb (45668)	149

$K(Zr+HL) = 9.5$

C6H807	H3L	Citric acid	CAS	77-92-9	(95)					
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH ₂ .CH(OH)(COOH).CH ₂ COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	sp	oth/un	?	?	U				1968KOa (46323)	150

$K(?) = 6.14$

Zr+++	sp	KCl	19°C	1.0M	U				1966KSc (46324)	151
$K(Zr+HL) = 10.78$										

Medium: HCl

Zr++++	ix	NaClO4	?	2.0M	U		1964RMd (46325) 152
						K(Zr+H3L=ZrH2L+H)=3.41	
Zr++++	ix	NaClO4	?	2.0M	U	I	1960REa (46326) 153
							K(Zr+H3L=ZrH2L+H)=3.12
Medium:	HC1O4.	K=3.87(I=1.0)					

C6H9N06		H3L	NTA		CAS 139-13-9 (191)		
Nitrilotriethanoic acid; N(CH2.COOH)3							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values Reference ExptNo
Zr++++	sp	oth/un	?	1.0M	C		K1=19.31 1999VKb (47123) 154
Medium:	1.0 M HC1O4;	room temperature					
Zr++++	cal	oth/un	25°C	0.0	U	TIH	1981VBa (47124) 155
DH(K1)=35.1 kJ mol-1							
Zr++++	ix	NaClO4	20°C	2.00M	U		K1=18.6 1970PHb (47125) 156
Zr++++	ix	NaClO4	?	0.23M	U	I	T K1=20.81 1966EMd (47126) 157
Medium:	HC1O4.	K1=19.51(I=1.0)					
Zr++++	sp	KCl	19°C	1.0M	U		K1=18.93 1966KSc (47127) 158
Medium:	HCl						
Zr++++	sp	NaClO4	?	2.00M	U		B2=7.8 1966LPa (47128) 159
Zr++++	ix	NaClO4	?	2.0M	U	I	
							K(Zr+H3L=ZrL+3H)=4.08
Medium:	HC1O4.	K=5.35(I=1)					
Zr++++	sp	oth/un	25°C	0.10M	U	T	K1=20.8 1964IMa (47130) 161
Zr++++	ix	oth/un	?	2.0M	U		
							K(?)=6.58 1964PVb (47131) 162

C6H1008		H2L	Mucic acid		CAS 526-99-8 (3650)		
2,3,4,5-Tetrahydroxyhexanedioic acid, Galactaric acid; H0OC.(CHOH)4.CO0H							
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values Reference ExptNo
Zr++++	sp	KCl	25°C	1.00M	U		
							K(Zr(OH)+HL)=9.49 1978KKF (48442) 163
Zr++++	EMF	oth/un	25°C	?	U		
							K(Zr(OH)3+L)=6.55 1970KKb (48443) 164

C6H1008		H2L	Saccharic acid		CAS 87-73-0 (1191)		

D-2,3,4,5-Tetrahydroxy-1,6-hexanedioic acid, Glucaric acid; HOOC.(CHOH)4.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	ix	NaClO ₄	20°C	2.0M	U				1963RCa (48493)	165

$$K(Zr+H_2L=ZrL+2H)=2.43$$

Medium: HClO₄

C6H11N05 H2L HIMDA CAS 93-62-9 (192)
N-(2-Hydroxyethyl)iminodiethanoic acid; HO.CH₂.CH₂.N(CH₂.COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	gl	KNO ₃	25°C	0.10M	U				1959CGa (48820)	166

$$K(ZrO(OH)L+H)=5.3$$

C6H12O7 HL Gluconic acid CAS 526-95-4 (904)
D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH₂(CHOH)₄.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	ix	NaCl	?	2.00M	U				1971ZGa (49772)	167

$$K(Zr(OH)2+L)=6.60$$

$$K(Zr(OH)2L+L)=5.95$$

Zr+++	sp	NaClO ₄	25°C	0.20M	U				1970CMD (49773)	168
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$$K_{eff}(ZrO_2+2L)=2.25 \text{ pH 3}$$

Zr+++	ix	NaClO ₄	20°C	2.0M	U				1963RCa (49774)	169
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$$K(Zr+HL=ZrL+H)=1.73$$

Medium: HClO₄

C6H13N04 HL Bicine CAS 150-25-4 (2124)
N,N-Bis(2-hydroxyethyl)glycine; (HO.CH₂.CH₂)₂N.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	gl	KNO ₃	25°C	0.10M	U				1959CGa (50422)	170

$$K(ZrO(OH)L+H)=3.9$$

$$K(ZrO(OH)2L+H)=8.5$$

C7H6O3 H2L Salicylic acid CAS 69-72-7 (14)
2-Hydroxybenzoic acid, Salicylic acid; HO.C₆H₄.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	sp	KCl	?	0.10M	U				1970KKd (54345)	171

$$K(Zr(OH)3+HL)=6.38$$

$$K(Zr(OH)2+HL)=6.05$$

Zr++++ kin oth/un 25°C ? U 1969KMF (54346) 172
 $K(Zr(OH)_3+HL)=6.1$

C7H6O4 H3L Resorcylic acid CAS 89-86-1 (876)
2,4-Dihydroxybenzoic acid, *b*-Resorcylic acid; C6H3(OH)2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ gl KN03 30°C 0.10M U M T 1978SDa (54554) 173
 $K(ZrO+HL)=16.55$

C7H6O6S H3L CAS 5965-83-3 (399)
5-Sulfosalicylic acid, 2-Hydroxy-5-sulfobenzoic; HO3S.C6H3(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ sp KCl ? 1.0M U 1970KKd (55084) 174
 $K(Zr(OH)_2+HL)=6.15$
 $K(Zr(OH)_3+HL)=6.24$

Zr++++ kin oth/un 25°C ? U 1969KMF (55085) 175
 $K(Zr(OH)_3+HL)=6.0$

Zr++++ sp KCl 19°C 1.0M U 1966KSc (55086) 176
 $K(?)=4.79$

C7H7N02 H2L Salicylaldoxime CAS 94-67-7 (1486)
2-Hydroxybenzaldehyde oxime; HO.C6H4.CH:N.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ gl KCl 25°C 0.10M U I K1=12.43 1968MDe (55315) 177
K1=17.9(I=0), 17.35(I=0.01), 16.45(I=0.025), 15.13(I=0.05), 13.90(I=0.075)

C7H7N02 HL CAS 495-18-1 (184)
Benzohydroxamic acid; C6H5.CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ gl diox/w 37°C 30% C M B2=10.58 1983MAd (55527) 178
 $B(Zr(bpy)L)=12.60$

Zr++++ dis NaClO4 25°C 1.0M U K1=12.43 B2=24.08 1965BGa (55528) 179
Medium: HClO4

C7H8N2O2 L CAS 89-62-3 (466)
2-Nitro-4-methylaminobenzene; CH3.C6H3(NO2).NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ sp diox/w 25°C 100% U 1975BSb (55889) 180
 $K(ZrBr_4+2L)=5.58$

C7H11N06P2 H4L CAS 4712-06-5 (4470)
Amino(phenyl)methylenediphosphonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	gl	KCl	25°C	0.10M	U			K1=17.08 B2=21.66 $K(ZrO+HL)=11.61$ $K(2ZrO+L)=23.18$ $K(ZrO+H+2L)=16.69$	1969DMd (56945)	181

C8H502F3S HL TTA CAS 326-91-0 (165)
4,4,4-Trifluoro-1-(2-thienyl)butane-1,3-dione; F3C.CO.CH2.CO.C4H3S

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	dis	oth/un	25°C	1.0M	U			K1=10.98 B2=21.88 K3=10.36 K4=9.93	1967M0c (58699)	182

Zr++++ sp NaClO4 25°C 4.0M U 1951MCa (58700) 183
 $K(Zr+HL=ZrL+H)=3.03$

C8H502F3Se HL CAS 713-15-5 (3842)
4,4,4-Trifluoro-1-(2'-selenoyl)-butane-1,3-dione; F3C.CO.CH2.CO.C4H3Se

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	dis	oth/un	25°C	1.0M	U			K1=11.35 B2=22.10 K3=10.15 K4=9.55	1963MPc (58705)	184

C8H703Br HL CAS 1878-91-7 (3819)
2-(4'-Bromophenyl)-2-hydroxyethanoic acid, p-bromomandelic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	sp	KCl	?	1.0M	U				1970KKd (59245)	185

$K(Zr(OH)_2+L)=6.27$
 $K(Zr(OH)_3+L)=6.21$

Zr++++ dis NaClO4 25°C 1.0M U K1=7.15 B2=13.43 1961AHa (59246) 186
K3=6.65
K4=5.52

Medium: HClO4

C8H8O2 HL Phenylacetic CAS 103-82-2 (1361)
Phenylethanoic acid; C6H5.CH2.COOH

$$\begin{aligned} K(ZrL+H2L=ZrL2+2H) &= -1.55 \\ K(Zr+2OH+L) &= 32.55 \\ K(Zr+2OH+2L) &= 32.44 \end{aligned}$$

Medium: 40% v/v 0.10 M NaClO4. $K(Zr(OH)2L+2H)=3.4$, $K(Zr(OH)2L2+2H)=5.6$.

C10H608Cl2S2 H4L CAS 6155-33-5 (4761)

2,7-Dichlorochromotropic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	sp	oth/un	?	?	U				1973DMb (68538)	194
								$K(Zr(OH)2+2HL)=12.55$		

C10H609S2 H3L CAS 58425-39-1 (2004)

8-Hydroxy-1,2-naphthoquinone-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	sp	NaClO4	20°C	0.10M	U				1975MDa (68541)	195
								$B(Zr(OH)2L2)=17.7$		

C10H7N02 HL CAS 131-91-9 (2668)

1-Nitroso-2-naphthol, alpha-Nitroso-beta-naphthol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	sp	alc/w	32°C	50%	U			$K1=3.6$	1954JMa (68601)	196

Medium: 50% EtOH, 0.1 M NaNO3

C10H7N02	HL	CAS 132-53-6 (2524)
2-Nitroso-1-naphthol;		

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	sp	alc/w	25°C	50%	U	I		$K1=3.7$	1952JMa (68667)	197
In 50% dioxan										
B4=11.7										

C10H7N02	HL	Quinaldic acid	CAS 93-10-7 (2209)
Quinoline-2-carboxylic acid;			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr+++	gl	KNO3	25°C	0.10M	U			$K1=11.25$ $B2=19.60$ $K3=7.15$	1988ZMa (68726)	198

Metal is ZrO++.

C10H7N08S2	H3L	Nitroso-R acid	CAS 525-05-3 (1811)
1-Nitroso-2-hydroxynaphthalene-3,6-disulfonic acid;			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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 Zr++++ sp KCl 25°C 0.30M U 1976GMd (69041) 199
 K1eff=5.1 at pH 1.5
 ****=
 C10H7O2F3 HL CAS 326-06-7 (196)
 3-Benzoyl-1,1,1-trifluoroacetone; CF₃.CO.CH₂.CO.C₆H₅

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Zr++++ dis oth/un 25°C 1.0M U K1=10.85 B2=21.20 19710Ma (69166) 200
 K3=9.85
 K4=9.35
 ****=
 C10H8N2 L 2,2'-Bipyridyl CAS 366-18-7 (25)
 2,2'-Bipyridine; (C₅H₄N)₂

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Zr++++ gl diox/w 37°C 30% C M B2=6.98 1983MAd (69665) 201
 B(Zr(bha)L)=12.60
 bha: benzohydroxamic acid

 Zr++++ sp oth/un 20°C 0.10M U 1969SHf (69666) 202
 K(Zr(OH)₂+L)=9.66
 ****=
 C10H8O8S2 H4L Chromotropic ac CAS 148-25-4 (1875)
 1,8-Dihydroxynaphthalene-3,6-disulfonic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Zr++++ sp KCl 20°C 0.10M U 1963SMa (69985) 203
 K(Zr(OH)₂+HL=ZrOHL)=18.68
 ****=
 C10H1002 HL Benzoylacetone CAS 93-91-4 (197)
 1-Phenylbutane-1,3-dione; C₆H₅.CO.CH₂.CO.CH₃

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Zr++++ dis oth/un 25°C 1.0M U K1=12.71 B2=24.57 1961PMa (70791) 204
 K3=11.34
 K4=11.08
 ****=
 C10H12O2 HL CAS 1946-74-3 (202)
 3-Isopropyltropolone;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Zr++++ sp alc/w 25°C 50% U 1961HSa (71617) 205
 B4=20.46
 Medium: 50% EtOH, 0.01 M

C10H16N208 H4L EDDS CAS 52759-67-8 (1100)
1,2-Diaminoethane-N,N'-di-1,4-butanedioic acid; (CH₂.NH.CH(COOH)CH₂.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ gl KNO₃ 30°C 0.10M U 1971TSf (73203) 206
K(ZrO+L)=11.20

C10H16N208 H4L EDTA CAS 60-00-4 (120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ sp oth/un ? 1.0M C I K1=29.93 1999VKb (74342) 207
Medium: 1.0 M HClO₄; room temperature

Zr++++ ISE KNO₃ 25°C 0.10M C M 1996YHa (74343) 208
K(ZrL+F)=4.62
K(ZrLF+F)=2.8

Method: Fluoride ISE.

Zr++++ ix NaClO₄ 20°C 2.0M U K1=27.9 1970PHb (74344) 209

Zr++++ vlt KNO₃ ? ? U K1=28.4 1969SKa (74345) 210
In 3.6 M HNO₃: K1=28.4

Zr++++ EMF NaClO₄ 20°C 1.0M U T K1=27.7 1967BAc (74346) 211
K(ZrL+OH)=7.9

Zr++++ dis KCl 20°C 1.20M U K1=28.96 1967TIb (74347) 212
Medium: HCl

Zr++++ ix NaClO₄ ? 1.0M U I K1=28.0 1966EMd (74348) 213
Medium: HClO₄. K1=29.0(I=0.23)

Zr++++ sp NaClO₄ ? ? U K2=7.9 1966LPa (74349) 214

Zr++++ ix oth/un ? 5.0M U I T K1=30.63 1964CKa (74350) 215
Medium: 1-5 HNO₃. K1=28.46(1),30.58(2),31.11(3),30.92(4)

Zr++++ sp KCl 25°C 0.10M U K1=19.9 1964IMa (74351) 216
K(ZrLOH+H)=6.2
K(2ZrLOH=Zr₂L₂(OH)₂)=3.5

Zr++++ ix oth/un ? 2.0M U K(?)=5.91 1964PVb (74352) 217

Zr++++ ix KNO₃ 20°C 0.10M U I K1=29.5 1963KCb (74353) 218
Medium HNO₃. K1=28.5(I=1),30.6(I=5)

Zr++++ gl NaCl04 25°C 0.10M U K1=19.40 1956MJa (74354) 219

C10H18N207 H3L HEDTA CAS 150-39-0 (392)
N-(Hydroxyethyl)diaminoethane-N,N',N'-triethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ ISE KN03 25°C 0.10M C M 1996YHa (75549) 220
K(ZrH-1L+H)=1.18
K(ZrL+F)=5.37
K(ZrLF+F)=4.11
K(ZrH-1L+F)=3.50

Method: Fluoride ISE. K(ZrH-1LF+F)=2.4.

C11H9N302 H2L PAR CAS 1141-59-9 (636)
4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ vlt alc/w 25°C 50% U 1975TBa (77608) 221
K(Zr(OH)3+HL)=16.7

Medium: 50% EtOH/H2O

Zr++++ sp oth/un 20°C 0.10M U 1968SHb (77609) 222
K(Zr(OH)3+HL)=16.44

C11H1209 H3L CAS 69065-58-3 (2714)
1,2,4-Trihydroxy-3,4,5-trimethoxycarbonylcyclopentadiene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ sp NaCl 20°C 0.10M U 1979BLa (78429) 223
K[Zr(OH)3+HL=Zr(OH)3HL]=11.68
K[Zr(OH)+HL=Zr(OH)HL]=12.45
K[Zr(OH)2+HL=Zr(OH)2HL]=12.75

C11H18N208 H4L CAS 4408-81-5 (923)
1,3-Diaminopropane-N,N,N',N'-tetraethanoic acid; ((HOOC.CH2)2N.CH2.)2.CH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ dis NaCl04 ? 1.0M U K1=28.33 1968EMa (79481) 224
K(Zr+2HL)=33.02

C11H18N209 H4L HDPTA CAS 3148-72-9 (431)
1,3-Diamino-2-hydroxypropane-N,N,N',N'-tetraethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ oth KCl 20°C 0.10M U K1=23.58 1967Tib (79582) 225
 Method: chromatography. Medium: HCl

C12H8N2 L Phenanthroline CAS 66-71-7 (144)
 1,10-Phenanthroline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	sp	oth/un	20°C	0.10M	U				1968SHa (80533) 226	
								$K(Zr(OH)_2+2L)=10.07$		

C12H10N2O2 H2L CAS 2050-14-8 (3378)
 2,2'-Dihydroxyazobenzene; HO.C6H4.N:N.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	sp	alc/w	32°C	50%	U				1954JMa (80703) 227	
								$K(ZrOL+2H=ZrO+H2L)=5.0$		

Medium: 50% EtOH 0.03 M NaClO4

C12H20N2O8 H4L CAS 40623-42-5 (1101)
 1,2-Diaminoethane-N,N'-di(2-pentane-1,5-dioic acid); (CH₂NHCH(COOH)CH₂CH₂COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	gl	KNO ₃	30°C	0.10M	U				1971TSf (82113) 228	
								$K(ZrO+L)=11.80$		

C12H20N2O8S H4L TEDTA CAS 923-74-0 (3394)
 2,2'-Thiobis(ethyliminodiethanoic acid); S(CH₂.CH₂.N(CH₂.COOH)₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	dis	oth/un	20°C	0.10M	U			K1=23.17	1967Tib (82481) 229	

C12H20N2O9 H4L EEDTA CAS 923-73-9 (2112)
 Oxa-bis(ethyleneimino)diethanoic acid; ((HOOC.CH₂)₂N.CH₂.CH₂)₂O

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	dis	oth/un	20°C	0.10M	U			K1=24.72	1967Tib (82576) 230	

C13H10N2O2 HL CAS 27147-03-1 (6307)
 2-Hydroxy-5-(phenylazo)benzaldehyde; C₆H₅.N:N.C₆H₃(CHO)(OH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	gl	KNO ₃	30°C	1.0M	U				1976JDa (84848) 231	
								$K(ZrO+L)=6.70$		
								$K(ZrO+2L)=12.70$		

Data also for 3'-nitro (5.66 and 11.91) and 4'-nitro (5.71, 11.77) analogues

C13H10N2O4 H3L CAS 18924-02-2 (4012)
2-Carboxy-3',4'-dihydroxyazobenzene; HOOC.C6H4.N:N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Zr++++	sp	KCl	19°C	1.0M	U			1966KSc (84866)	232
							K(Zr+2H3L=ZrH4L2+2H)=14.34		

C13H17N3O L Aminopyrine (2030)
1-Phenyl-2,3-dimethyl-4-dimethylamino-5-pyrazolone, Dimethylaminoantipyrine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Zr++++	sp	oth/un	20°C	0.10M	U			1968SHb (86001)	233
							K(Zr(OH)3+L)=6.58		

C14H8O7S H3L DASA CAS 83-61-4 (950)
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Zr++++	sp	oth/un	25°C	1.60M	U			1967ZFa (86773)	234
							B(Zr(OH)2L)=49.0		

C14H9N03 HL CAS 116-85-8 (1020)
1-Amino-4-hydroxyanthraquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Zr++++	sp	alc/w	20°C	40%	U			1988ISa (86799)	235
							K(Zr+H3L=ZrH2L+H)=0.498		
							K(ZrH2L+H3L=Zr(H2L)2+H)=-0.108		
							in 40% EtOH, 0.1 NaClO4		

C14H22N2O8 H4L CDTA CAS 482-54-2 (200)
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Zr++++	ISE	KNO3	25°C	0.10M	C	M		1996YHa (88825)	236
							K(ZrL+F)=4.61		
							K(ZrLF+F)=3.0		

Method: Fluoride ISE.

Zr++++	ix	NaClO4	20°C	2.0M	U		K1=29.9	1970PLa (88826)	237
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Zr++++	ISE	KNO3	30°C	0.10M	U T H		K1=20.64	1965HWa (88827)	238
							K1=20.85(10 °C), 20.74(20 °C). DH(K1)=-17.6 kJ mol-1, DS=339 J K-1 mol-1		

Zr++++ ix oth/un ? 2.10M U K2=4.36 1964PVb (88828) 239

C14H23N3O10 H5L DTPA CAS 67-43-6 (238)
Diethylenetriamine-pentaethanoic acid; HOOC.CH2.N(CH2.CH2.N(CH2.COOH)2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	EMF	NaClO ₄	20°C	1.0M	U			K1=36.9 K(ZrL+OH)=8.1	1967BAc (89449)	240

Zr++++ dis oth/un 20°C 0.39M U K1=33.96 1967Tib (89450) 241

Zr++++ ix NaClO₄ ? 1.0M U I K1=35.40 1966EMd (89451) 242
Medium: HClO₄. K1=35.81(I=0.23 M)

Zr++++ sp NaClO₄ ? 100% U 1966LPa (89452) 243
K(Zr+H5L=ZrL+5H)=4.43

Zr++++ ix oth/un ? 2.0M U I 1964EMd (89453) 244
K(Zr+H5L=ZrL+5H)=3.63

Medium: HClO₄. K=5.67(I=1)

Zr++++ ix oth/un ? 2.10M U K2=4.89 1964PVb (89454) 245

C15H1003 HL CAS 577-85-5 (3443)
3-Hydroxyflavone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Zr++++ sp alc/w 20°C 20% C M 2002TTc (90978) 246
Method: fluorescence spectrophotometric titration. Medium: 20% v/v
EtOH/H₂O. K(Zr(H₂O)₂edta+L=Zr(edta)L+H)=5.43.

Zr++++ sp oth/un ? 0.10M U K1=5.7 B2=8.00 1955Hha (90979) 247

C15H1006 H4L Kaempferol CAS 520-18-3 (5099)
3,5,7-Trihydroxy-2-(4'-hydroxyphenyl)-1-benzopyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Zr++++ sp alc/w ? 40% U 1968Gta (90994) 248
K(ZrO+H4L=ZrO(H₃L)+H)=4.6

Medium: 40% EtOH

C15H1007 H5L Melanoxetin CAS 27696-41-9 (4054)
3,3',4',7,8-Pentahydroxyflavone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Zr++++ sp alc/w 20°C 50% U 1965KGa (91008) 249
K(ZrO+H5L=ZrO(H4L)+H)=5.56

Medium: 50% EtOH, 1 M HCl

C15H11N3O HL PAN CAS 85-85-8 (572)
1-(2-Pyridylazo)-2-naphthol; C5H4N.N:N.C10H6.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ vlt alc/w 25°C 50% U 1975TBa (91247) 250
K(Zr(OH)3+HL)=14.3

Medium: 50% EtOH/H2O

C16H11N3O10S2 H4L (5174)
2-Hydroxy-1-(2'-hydroxy-4'-nitro)phenylazo-3,6-disulfonaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ sp oth/un 25°C ? U 1971RCd (92883) 251
K(?)=7.07

C16H13N2O10AsS2 H5L Thorin I CAS 3688-92-4 (2609)
1-((2-Arsenophenyl)azo)-2-hydroxy-3,6-naphthalyl disulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ sp KCl 25°C 0.10M U 1976GMd (93221) 252
K1eff=5.2 at pH 2.0

C17H14N2O2 L CAS 4551-69-3 (698)
4-Benzoyl-3-methyl-1-phenyl-2-pyrazolin-5-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ dis oth/un 25°C 1.0M U K1=10.14 B2=20.22 19710Ma (95910) 253
K3=10.02
K4=9.96

C17H14N2O5S H3L Calmagite CAS 3147-14-6 (2875)
1-(1-Hydroxy-4-methyl-2-phenylazo)-2-naphthol-4-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ sp KCl 25°C 0.10M U K1=49.0 1976GMd (95933) 254

C18H14N2O2 HL CAS 15017-21-7 (6859)
2-Hydroxynaphthalidene benzoyl hydrazone; C6H5.CO.NH.N:CH.C10H6.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Zr++++ gl diox/w 20°C 75% U T HM K1=8.65 B2=16.48 1994MCa (96913) 255
 B3=21.75
 K(Zr(edta)+L)=4.31
 K(Zr(Hedta)+L)=3.92
 K(Zr(nta)+L)=3.68

Medium: 75% v/v dioxane/H₂O, 0.10 M NaClO₄. Data for 30 and 40 C.

DH and DS values.

C18H30N4O12 H6L TTHA CAS 869-52-3 (694)
 Triethylenetetraaminehexaethanoic acid;((HOOC.CH₂)₂N.CH₂.CH₂.N(CH₂.COOH).CH₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	ISE	KNO ₃	25°C	0.10M	C	M			1996YHa (98109)	256
								K(ZrL+H)=2.77		
								K(ZrHL+F)=5.48		
								K(ZrL+F)=2.9		

Method: Fluoride ISE.

Zr++++ ix NaClO₄ ? 0.50M U I K1=19.74 1966EMc (98110) 257
 K(Zr(OH)+H6L=ZrH2L+3H)=4.76

Medium: HClO₄. K1=9.74(I=1); K(Zr+H6L=ZrH2L+4H)=4.08(I=1), 2.9(I=2)

C19H14O7S H4L Pyrocatechol Vi CAS 369596-29-2 (709)

Pyrocatechol Violet,
 3-[3,4-Dihydroxyphenyl-3-hydroxy-4-oxo-2,5-cyclohexadien-1-ylidenemethyl-b.;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	sp	non-aq	25°C	100%	U				1968DSb (99119)	258
								B12(?)=11.04		

Medium: 1-butanol. Ligand: Pyrocatechol sulfophthalein

C20H14N2O5S H3L Solochrome 6B CAS 3564-14-5 (3507)

1-(1-Hydroxy-2-naphthylazo)-2-naphthol-4-sulfonic acid, Mordant Black3, Eriochrome blue-black B;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	sp	KCl	25°C	0.10M	U			K1=46.2	1976GMd (99671)	259

C22H18N4O14As2S2 H8L Arsenazo III CAS 1668-00-4 (1148)

2,7-Bis(2'-arsonophenylazo)chromotropic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Zr++++	sp	NaClO ₄	20°C	3.0M	U				1964BUd (101667)	260
								B(Zr2H18L2)=87.2		

Medium: 3-6 M HClO₄

C22H37N5O14	H7L		CAS 3234-59-1 (2425)	
Tetraethylenepentamineheptaethanoic acid;				
<hr/>				
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference ExptNo
Zr++++	ix	NaClO4	? 0.50M U I	1966EMc (102350) 261 $K(Zr(OH)+H7L=ZrH4L+2H)=6.08$
Medium: HClO4. $K(Zr+H7L=ZrH4L+3H)=5.45(I=1)$, $4.51(I=2)$				
<hr/>				
C23H24N4O2	L	Trichachnine	CAS 1251-85-0 (2606)	
4,4'-Diantipyrylmethane,				
4,4'-phenylmethlene-bis-(1,2-dihydro-1,5-dimethyl-2-phenylpyrazol-3-one				
<hr/>				
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference ExptNo
Zr++++	sp	oth/un	25°C 0.10M U	K1=8.71 1979SPe (102682) 262
Zr++++	sp	oth/un	25°C 1.00M U	B2=12.58 1979SPe (102683) 263
Zr++++	sp	KCl	? 0.10M U	1972SCb (102684) 264 B4=16.92
Zr++++	sp	KCl	? 1.0M U I	B2=11.8 1963BSb (102685) 265
Medium: HCl. $B2=11.5(I=0.1)$				
<hr/>				
C26H25N09S	H4L	Semi-Xylenol O	(426)	
3-(N,N-Di(carboxymethyl)aminomethyl)-2-cresolsulfonephthalein;				
<hr/>				
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference ExptNo
Zr++++	gl	KNO3	25°C 0.10M C	1993YHa (103951) 266 $K(ZrL+F)=7.4$ $K(2ZrL+2OH=Zr2(OH)4(HL)2)=29.0$
Zr++++	gl	KNO3	25°C 1.00M U	K2=26.25 1978SYb (103952) 267 $B(ZrHL)=27.56$
<hr/>				
C31H32N2013S	H6L	Xylenol orange	CAS 63721-85-5 (432)	
5,5'-Bis-N,N-bis(carboxymethyl)aminomethyl-4'-hydroxy-3,3'-dimethylfuchsone-2"-sulfonic acid;				
<hr/>				
Metal	Mtd	Medium	Temp Conc Cal Flags Lg K values	Reference ExptNo
Zr++++	sp	NaClO4	20°C 0.48M C	1984MSe (105514) 268 $K_{eff}=4.58$ $B_{eff}(Zr2L)=11.59$
Medium: 0.48 M HClO4.				
<hr/>				
Zr++++	gl	KNO3	25°C 1.00M U	1978SYb (105515) 269 $B(ZrH3L)=37.80$

B(ZrH4L)=38.68

B(Zr2L)=43.47

Zr++++ sp NaNO₃ 20?°C 0.20M U 1963BGa (105516) 270
B(Zr2L2)=31.0

Zr++++ sp NaClO₄ ? 0.80M U 1959CHa (105517) 271
K(?)=7.60

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EXPLANATORY NOTES

DATA Flags are :-

- T Data at other TEMPERATURES
- I Data with various BACKGROUNDS
- H Data for THERMOCHEMICAL quantities
- M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC

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